

Conventional polymer solar cells with power conversion efficiencies increased to >9% by a combination of methanol treatment and an anionic conjugated polyelectrolyte interface layer

Tao Yuan,^a Dong Yang,^b Xiaoguang Zhu,^a Lingyu Zhou,^b Jian Zhang,^{*b} Guoli Tu^{*a} and Can Li^b

^aWuhan National Laboratory for Optoelectronics, Huazhong University of Science and Technology, Wuhan, 430074, People's Republic of China. E-mail: tgl@hust.edu.cn

^bState Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian National Laboratory of Clean Energy, Dalian 116023, People's Republic of China. E-mail: jianzhang@dicp.ac.cn

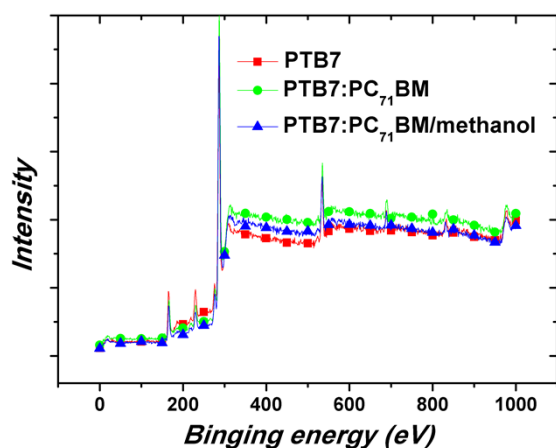


Figure S1. XPS survey of the top surfaces at various conditions.

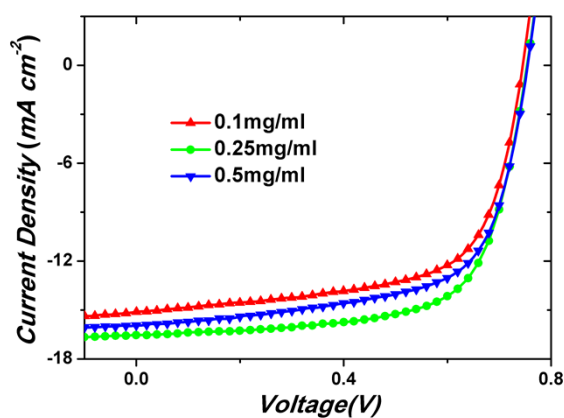


Figure S2. The current density-voltage (J - V) characteristics of the PSCs with $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ interlayer spin-coated at different solutions under AM 1.5G irradiation (100 mW cm^{-2}).

The $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ was dissolved in methanol, and the thickness of $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ layer was optimized by varying the concentrations of its methanol solutions from 0.1 mg/ml to 0.5 mg/ml with fixed spin-coating speed and was measured by AFM. The optimized concentration of spin-coated $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ interlayer was 0.25 mg/mL. The J_{sc} and FF of the resulting PSCs were highly dependent on the thickness of $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$. When spin-coated from 0.1 mg/mL solution, the $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ film could only cover a small part of the active layer, which led to a moderate enhancement. When the concentration of $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ was further increased to 0.5 mg/mL, the performance was reduced because the $\text{PF}_{\text{EO}}\text{SO}_3\text{Na}$ film was too thick and acted as an insulating layer. The detailed devices characteristics with different $[\text{PF}_{\text{EO}}\text{SO}_3\text{Na}]$ were summarized in Table S1.

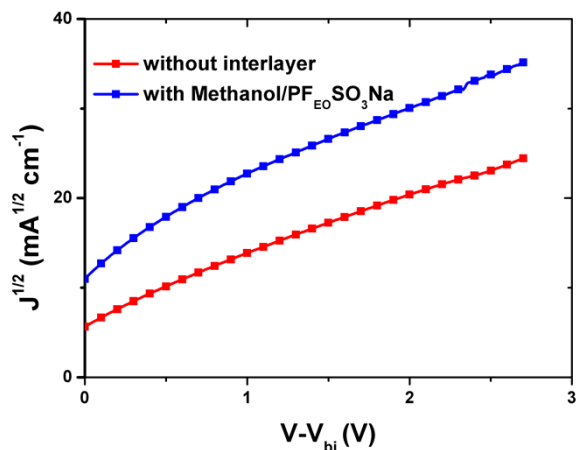


Figure S3. J-V curves of the hole mobility devices with methanol and PF_{EO}SO₃Na cathode interface layer or without interlayer.

To confirm the effect of methanol treatment and PF_{EO}SO₃Na layer on charge transport property, hole only devices with configuration of ITO/PEDOT:PSS/PTB7:PC₇₁BM/MoO₃/Au with or without methanol treatment and PF_{EO}SO₃Na layer were constructed and their hole mobility was calculated from the space charge limited current (SCLC) J-V curves obtained in the dark (Figure S3). SCLC can be

characterized by the Mott-Gurney square law: $J = \frac{9}{8} \epsilon_r \epsilon_0 \mu_0 \frac{V^2}{L^3}$, in which ϵ_0 is the vacuum permittivity, ϵ_r is the dielectric permittivity of the active layer, L is the thickness of the active layer and μ_0 is the hole mobility. In pristine PTB7:PC₇₁BM devices, the hole mobility was $1.7 \times 10^{-4} \text{ cm}^2 \text{ V}^{-1} \text{ s}$. After combination of methanol treatment and PF_{EO}SO₃Na layer, the hole mobility was increased to $4.0 \times 10^{-4} \text{ cm}^2 \text{ V}^{-1} \text{ s}$, which suggested that the hole transport in devices with methanol treatment and PF_{EO}SO₃Na layer was improved.

Table S1. Device performance of PSCs obtained with different [PF_{EO}SO₃Na].

| Cathode | V_{oc} [V] | J_{sc} [mA cm ⁻²] | FF [%] | PCE [%] |
|-----------|-----------------|------------------------------------|-------------|------------|
| 0.1mg/mL | 0.74 | 15.1 | 65.8 | 7.36 |
| 0.25mg/mL | 0.74 | 16.5 | 69.5 | 8.50 |
| 0.5mg/mL | 0.75 | 15.9 | 65.5 | 7.83 |