Supplementary Information

Exploring the catalytic activity of new water soluble dinuclear copper(II) complexes towards the glycoside hydrolysis

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Figures with captions



Fig. S1 FTIR spectrum of the ligand H₃phpda in the region of 4000-450 cm⁻¹.



Fig. S2 ¹H NMR spectrum of the ligand H_3 phpda in D_2O .



Fig. S3 ¹³C NMR spectrum of the ligand H_3 phpda in D_2O .

Fig. S4 ESI mass spectrum (positive ion mode) of the ligand H₃phpda in methanol solution.

Fig. S5 FTIR spectrum of complex 1 in the region of 4000-450 cm⁻¹.

Fig. S6 UV-vis spectra of complex 1 at (a) 10^{-3} (M) and (b) 10^{-4} (M) in aqueous solution at pH~7.2.

(a)

(b)

Fig. S7 UV-vis spectra of complex 2 at (a) 10^{-3} (M) and (b) 10^{-4} (M) in aqueous solution at pH~7.2.

Fig. S8 UV-vis spectra of (a) complex 1 at 10^{-3} (M) and (b) complex 2 at 10^{-3} (M) in aqueous solution at pH~10.5.

Fig. S9 ESI mass spectrum (positive ion mode) of complex 1 in aqueous solution at pH~7.2.

Fig. S10 ESI mass spectrum (positive ion mode) of complex 2 in aqueous solution at pH~7.2.

Fig. S11 ESI mass spectrum (positive ion mode) of complex 1 in aqueous solution at pH~10.5.

Fig. S12 ESI mass spectrum (positive ion mode) of complex 2 in aqueous solution at pH~10.5.