

Supporting Information

Effects of Deposited Ions on the Photocatalytic Activity of TiO₂-Au Nanospheres

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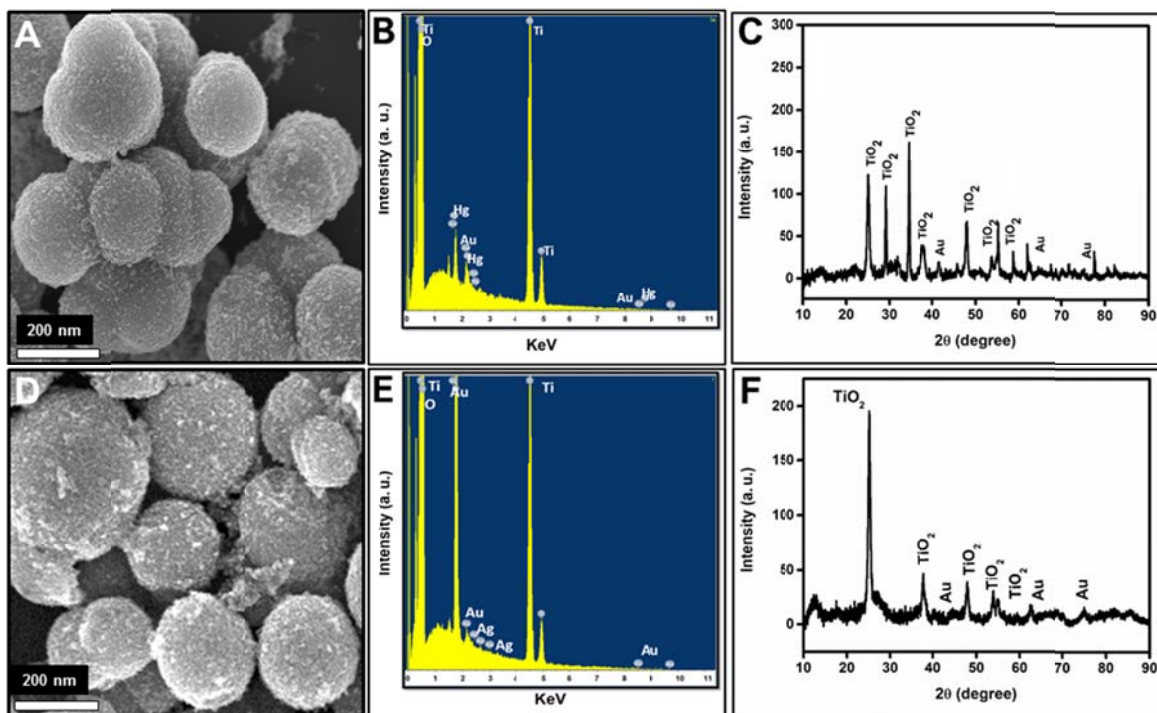


Figure S1. (A) SEM image (B) EDX and (C) XRD pattern of TiO₂-Au/Hg NSs; (D) SEM image, (E) EDX and (F) XRD pattern of TiO₂-Au/Ag NSs.

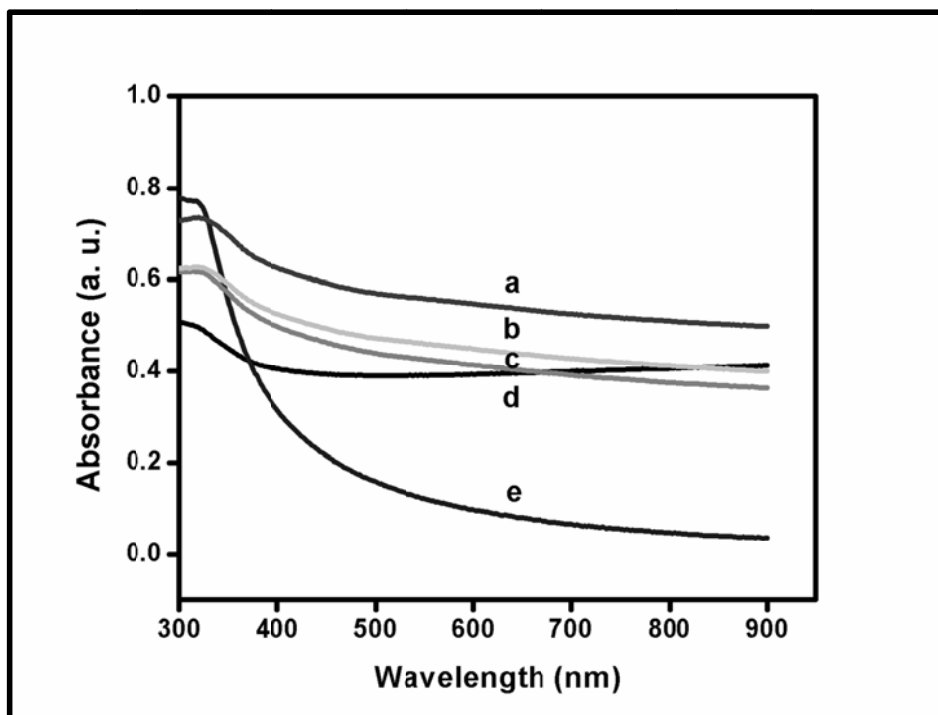


Figure S2. UV-Vis absorption spectra of (a) TiO_2 - Au NSs, (b) TiO_2 -Au/Ag NSs, (c) TiO_2 -Au/Hg NSs, (d) TiO_2 NSs and (e) P25.

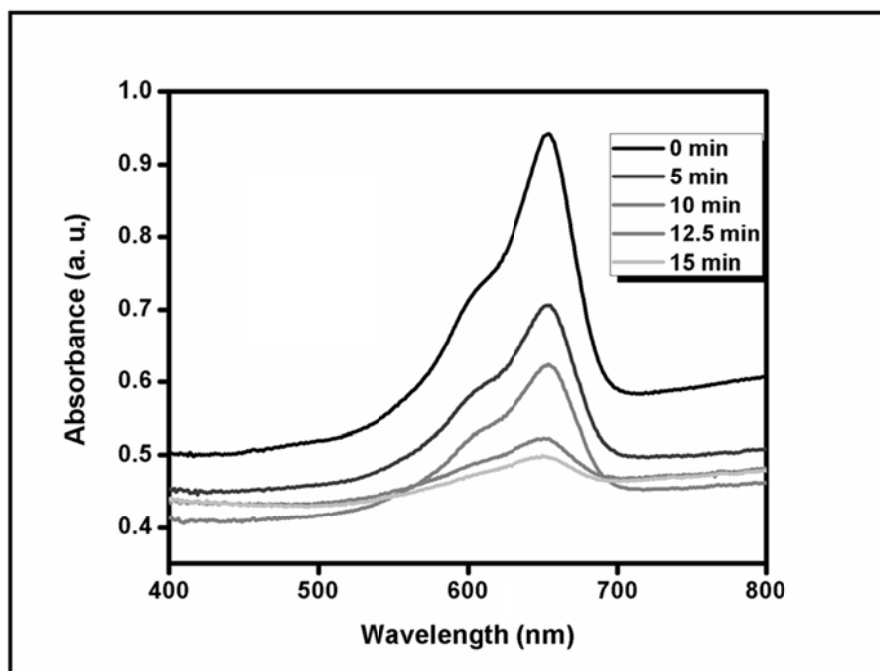


Figure S3. UV-Vis absorption spectra of MB (10 μM) that had been subjected to photocatalytic degradation for various times using $\text{TiO}_2\text{-Au}$ NSs (final concentration: 10 mg mL^{-1}).

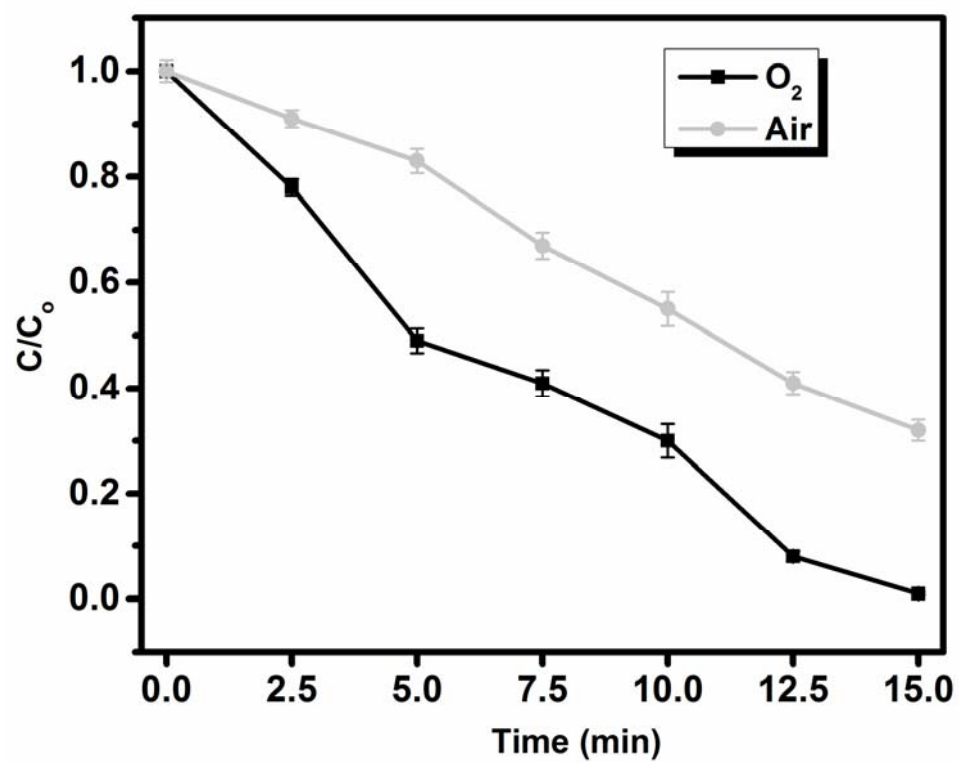


Figure S4. Effect of oxygen on the photocatalytic degradation of MB (10 μ M) using TiO_2 -Au NSs (final concentration: 10 mg mL^{-1}). Other conditions are the same as that in Figure 3.

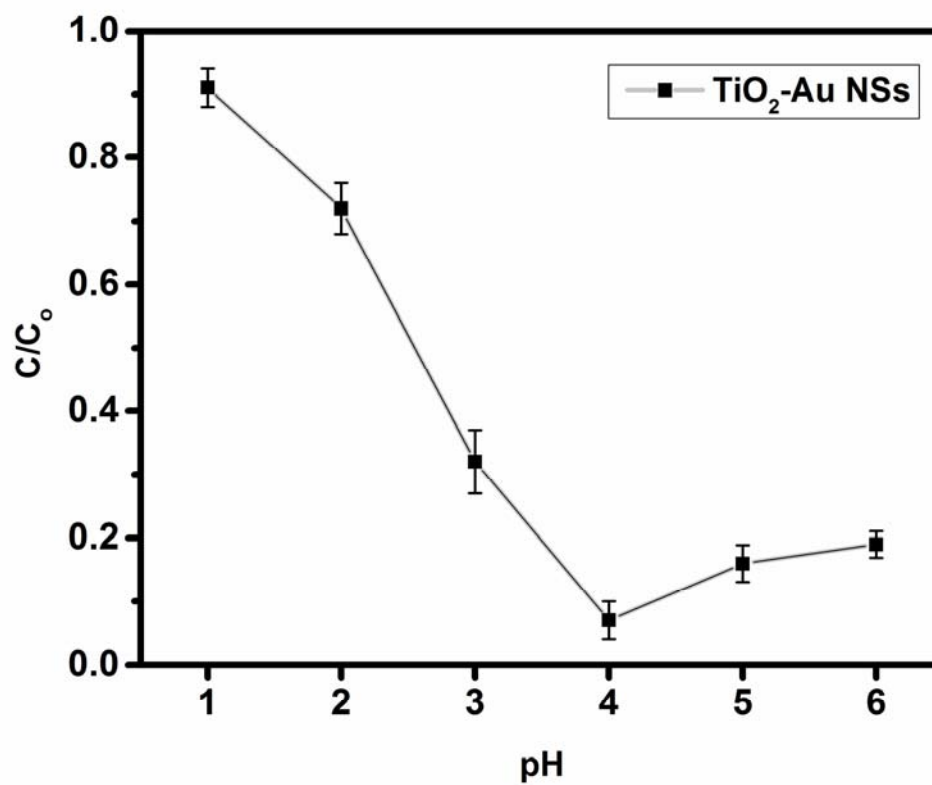


Figure S5. Effect of pH on the reduction of Cr^{6+} ($34\ \mu\text{M}$) by $\text{TiO}_2\text{-Au NSs}$ (final concentration: $10\ \text{mg mL}^{-1}$). Buffer: $10\ \text{mM}$ acetate solution. Other conditions are the same as that in Figure 4.

Table 1: Usability of TiO₂-Au, TiO₂-Au/Hg and TiO₂-Au/Ag NSs as photocatalysts

Cycles	MB			Cr ⁶⁺
	TiO ₂ -Au NSs	TiO ₂ -Au NSs/Hg	TiO ₂ -Au NSs/Ag	TiO ₂ -Au NSs
	(%)	(%)	(%)	(%)
1	100	100	100	100
2	99	97	97	99
3	97	96	96	99
4	96	94	94	98
5	92	93	93	97

The efficiency in each case is set at 100% for the first cycle.