

Supplementary Information

Synthesis of yolk/shell $\text{Fe}_3\text{O}_4@\text{poly(ionic liquid)s}$ -derived nitrogen doped graphitic porous carbon materials and its application as support for nickel catalysts

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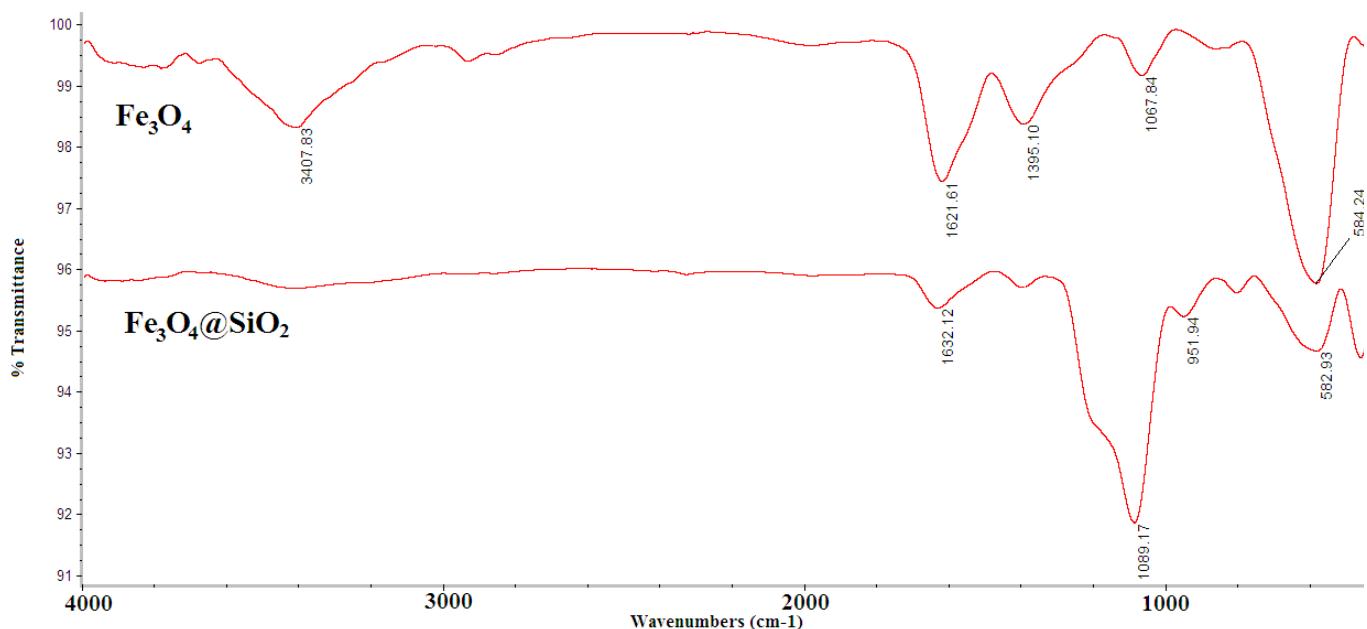


Fig. S1 FT-IR spectra of Fe_3O_4 , and $\text{Fe}_3\text{O}_4@\text{SiO}_2$

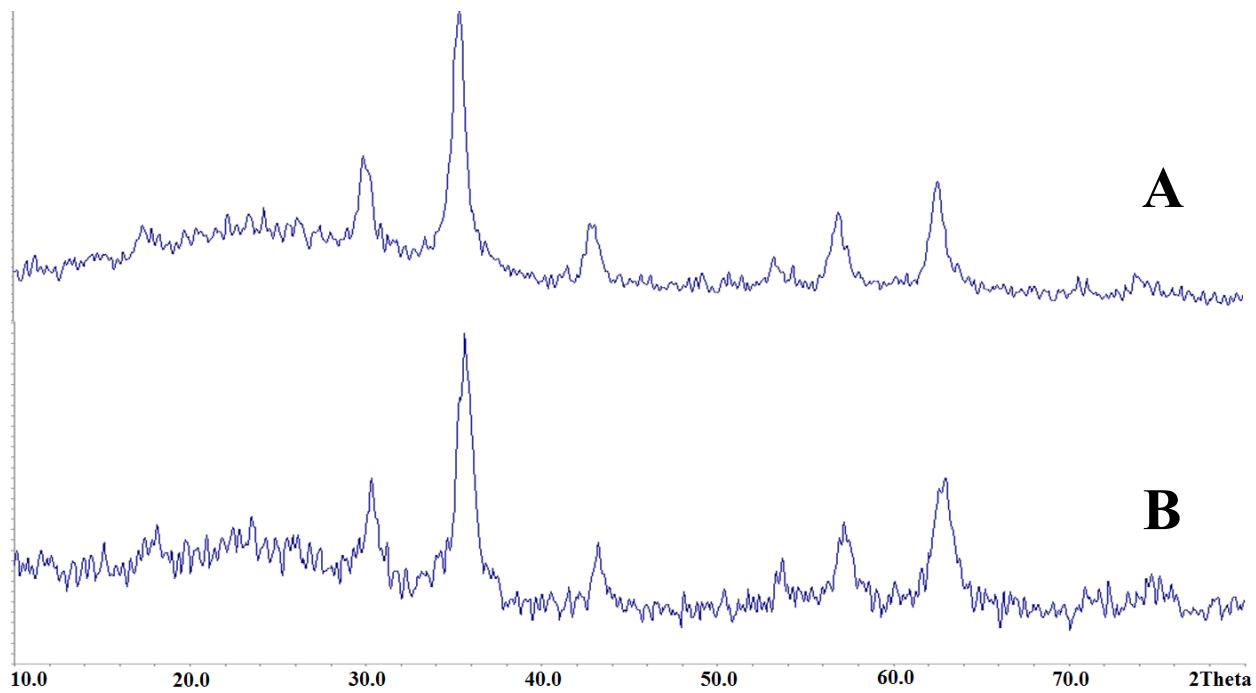


Fig. S2 XRD patterns of $\text{Fe}_3\text{O}_4@{\text{SiO}}_2$ (A), and $\text{Fe}_3\text{O}_4@{\text{SiO}}_2@{\text{PCMVImCl}}$ (B)

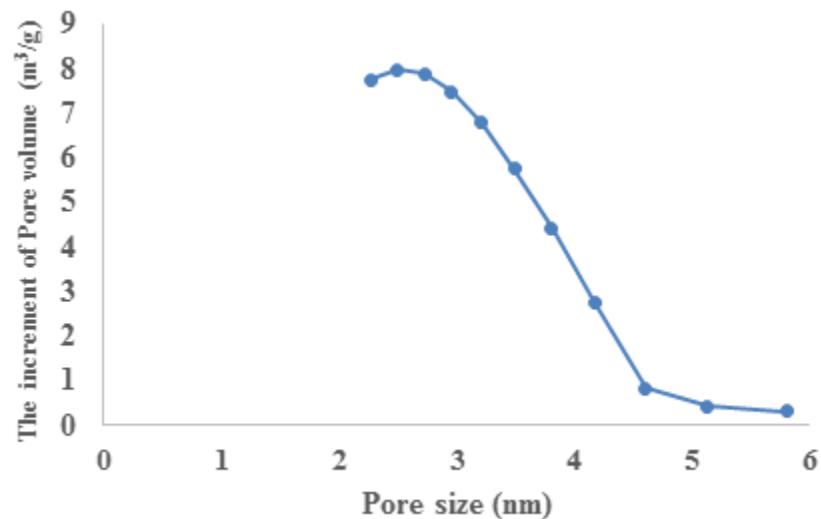


Fig. S3 Pore size distribution plot obtained from the adsorption branch of the isotherm for YS $\text{Fe}_3\text{O}_4@{\text{PIL-d-(N)GPC}}$.

Table S1 Influence of solvent, and the amount of catalyst on reduction reaction of 4-nitrotoluene.^a

Entry	Amount of catalyst (Ni(0) content/mol %)	Solvent	Isolated Yield (%)
1	1	DMSO	80
2	1	MeOH	77
3	1	EtOH	83
4	1	H ₂ O	90
6	1.2	H ₂ O	92
7	0.8	H ₂ O	81
10	0	H ₂ O	-

^aReaction conditions: 1 mmol 4-nitrotoluene, 0.5 mmol NaBH₄, 5 mL solvent, NiNPs@YS Fe₃O₄@PIL-d-(N)GPC as catalyst.

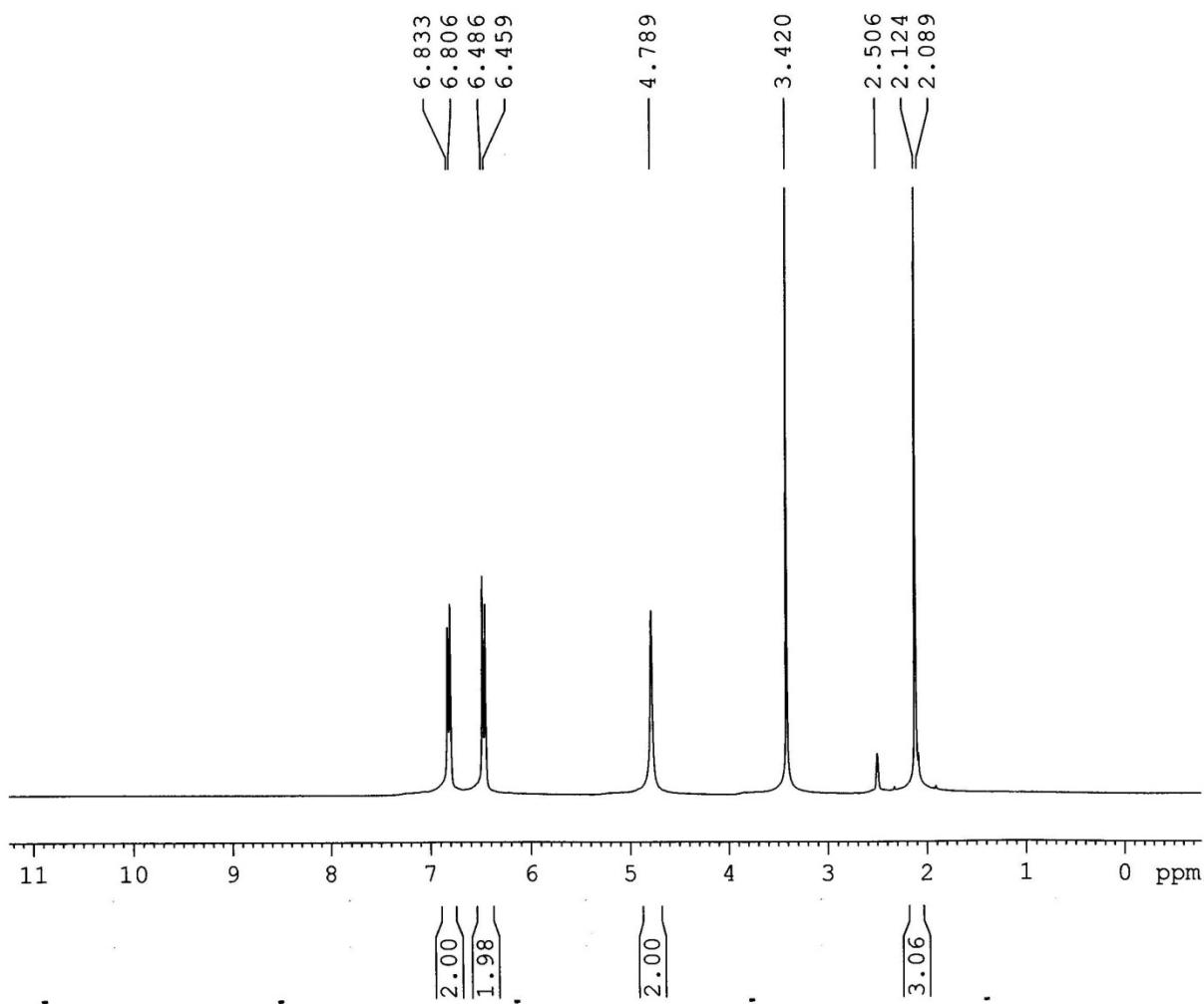


Fig. S4 ^1H NMR spectrum of p-toluidine in $\text{DMSO}-d_6$.

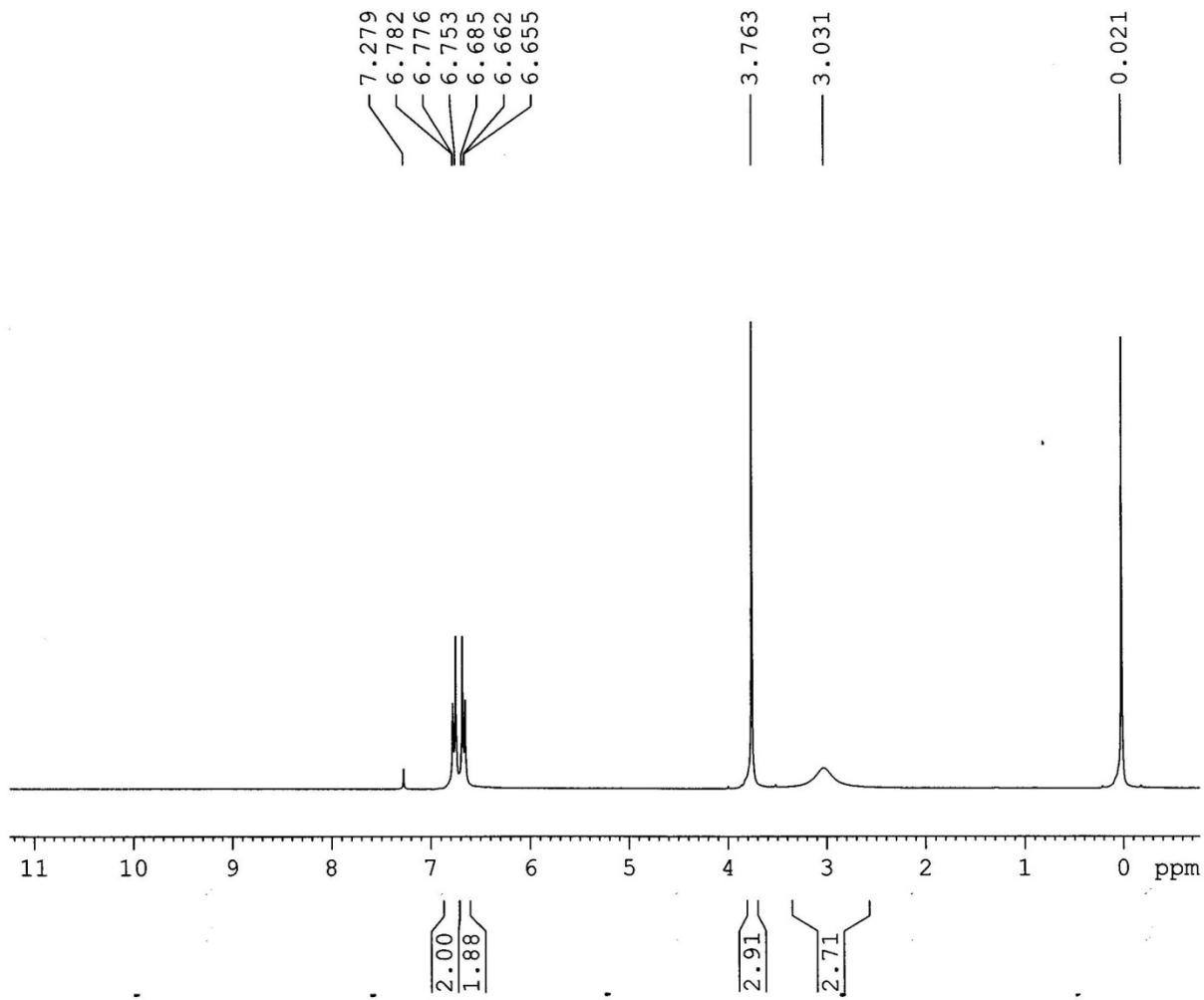


Fig. S5 ^1H NMR spectrum of 4-methoxyaniline in CDCl_3 .

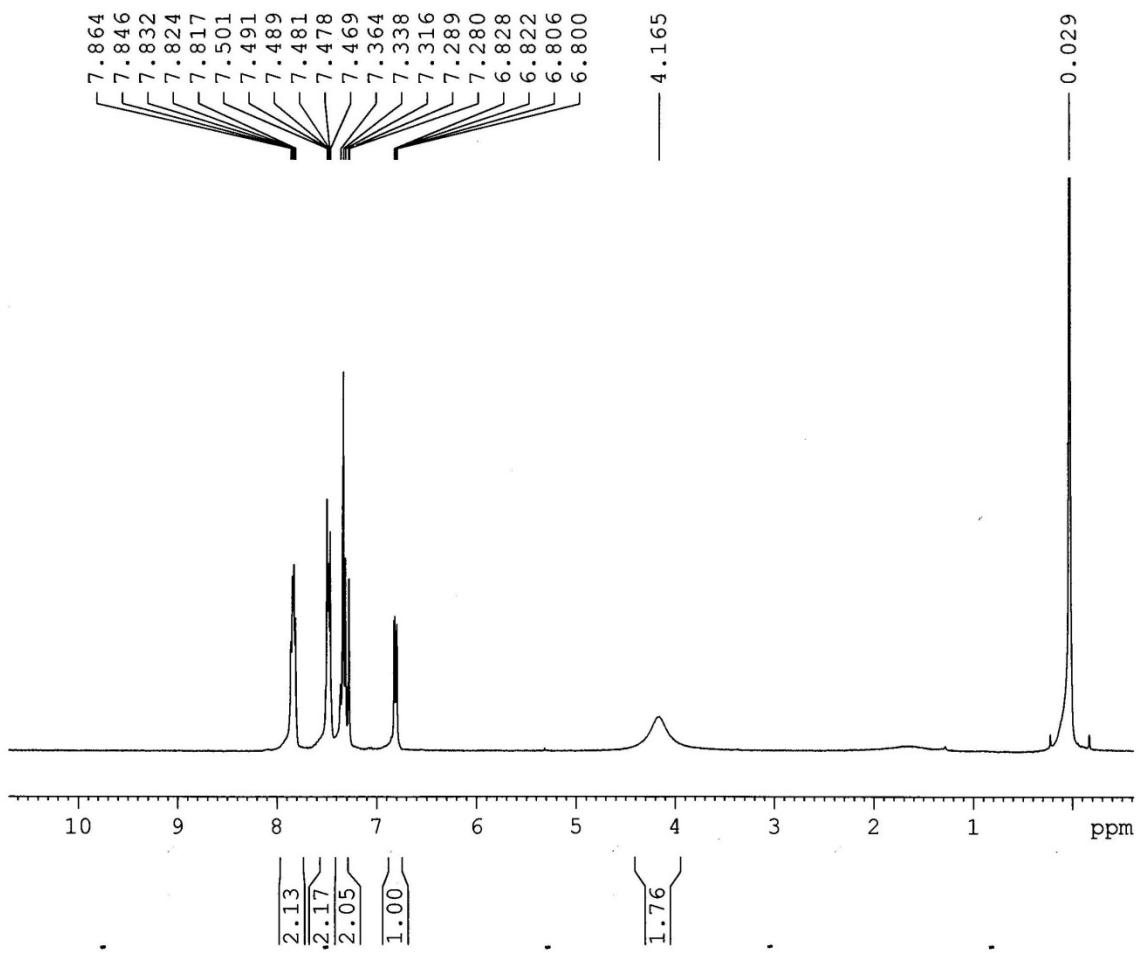


Fig. S6 ^1H NMR spectrum of naphthalen-1-amine in CDCl_3 .

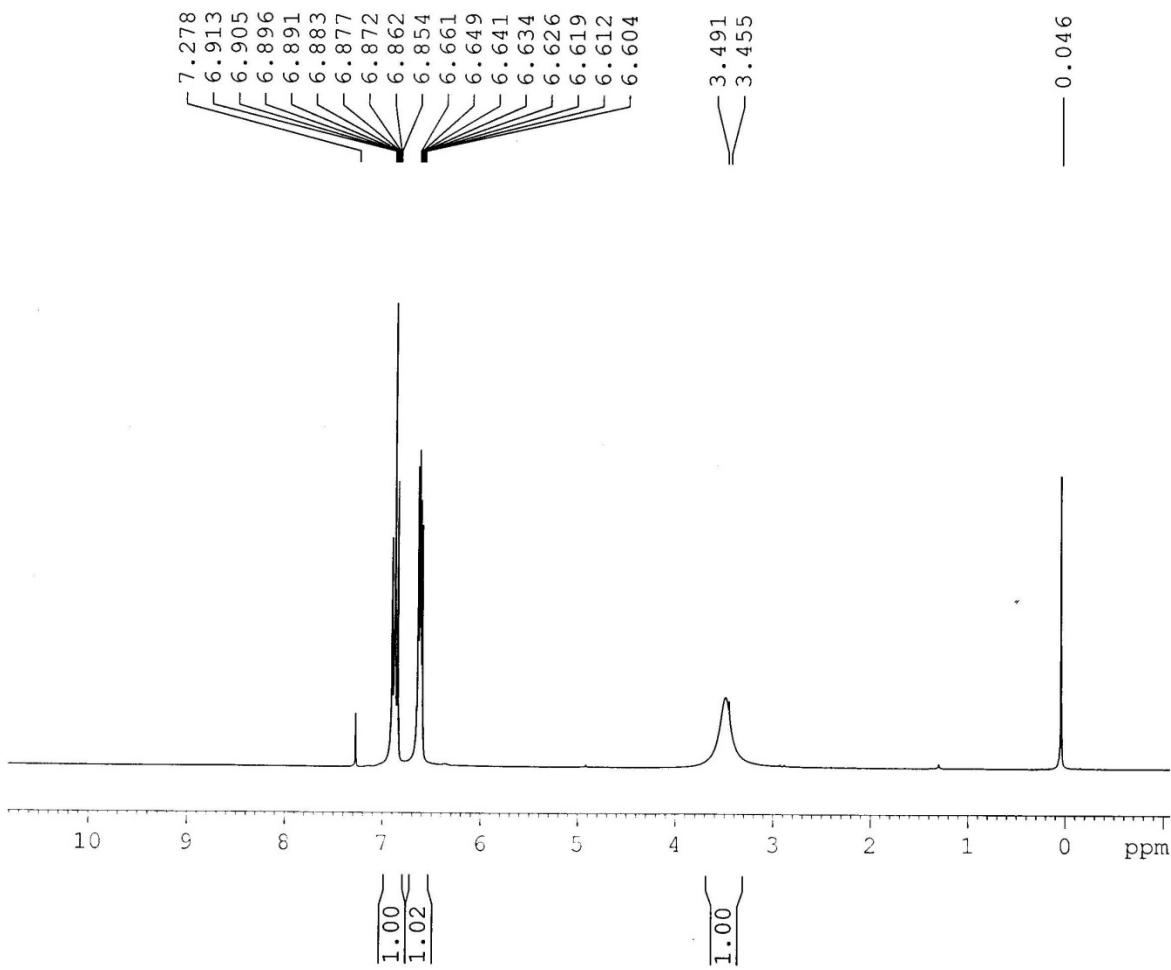


Fig. S7 ^1H NMR spectrum of 4-fluoroaniline in CDCl_3 .

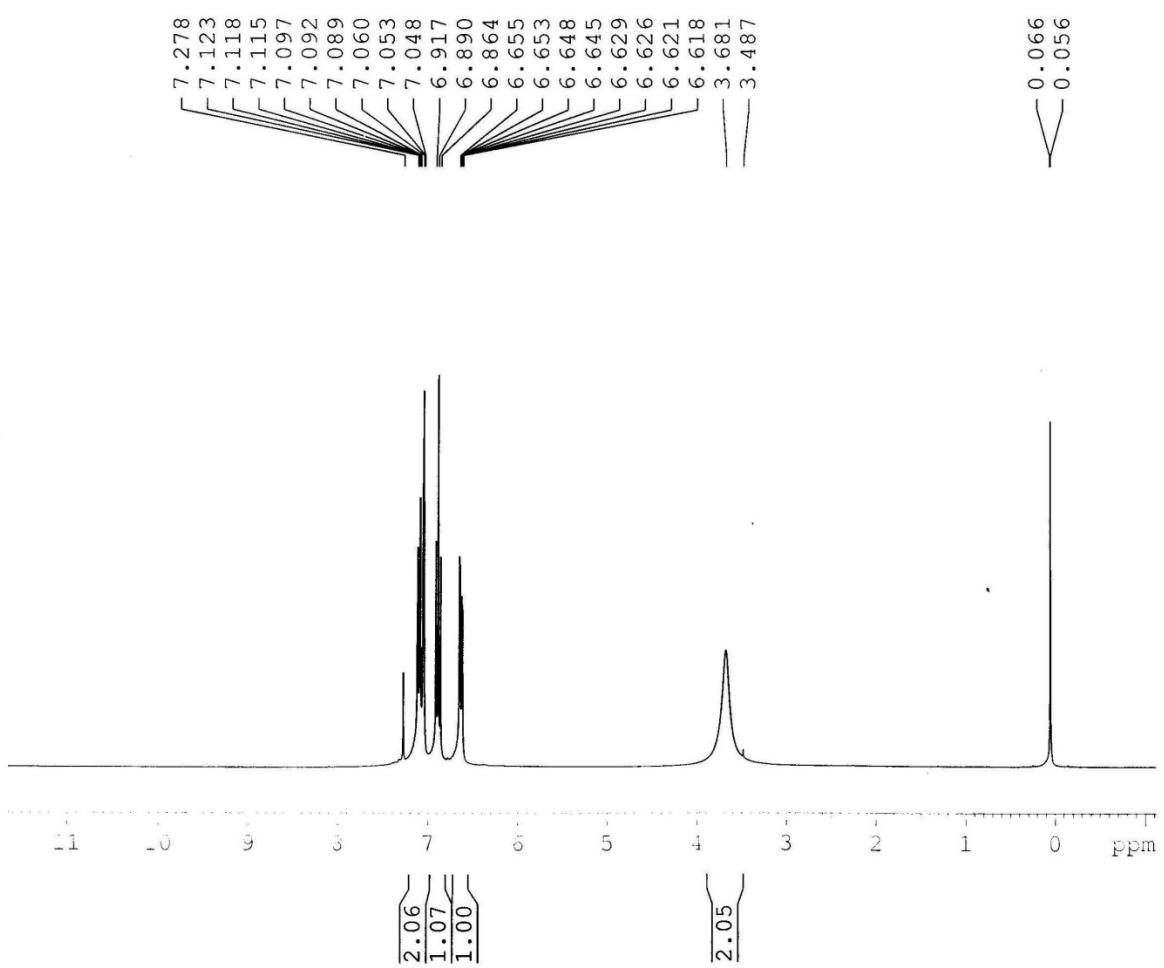


Fig. S8 ^1H NMR spectrum of 3-iodoaniline in CDCl_3 .

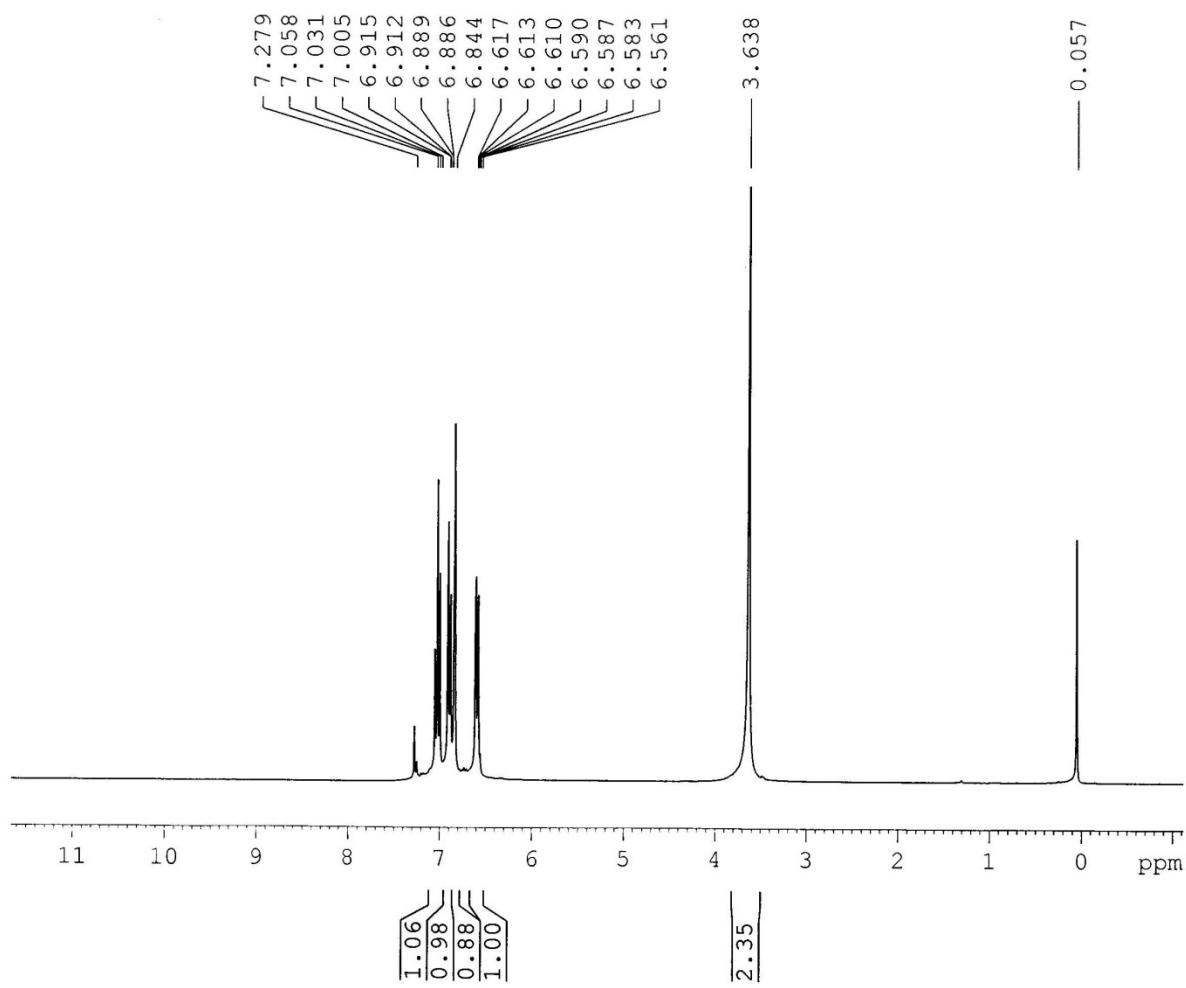


Fig. S9 ^1H NMR spectrum of 4-bromoaniline in CDCl_3 .

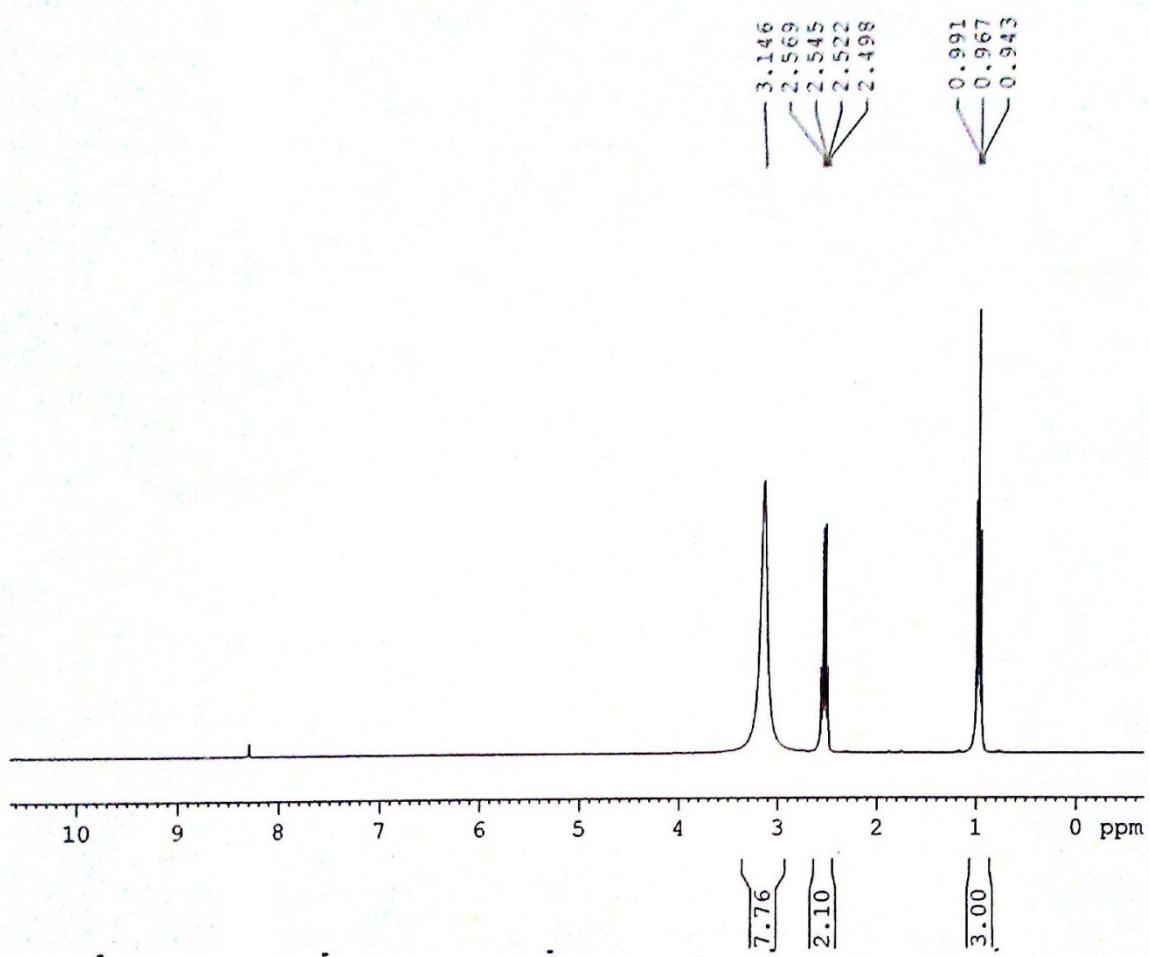


Fig. S10 ${}^1\text{H}$ NMR spectrum of ethanamine in $\text{DMSO}-d_6$.

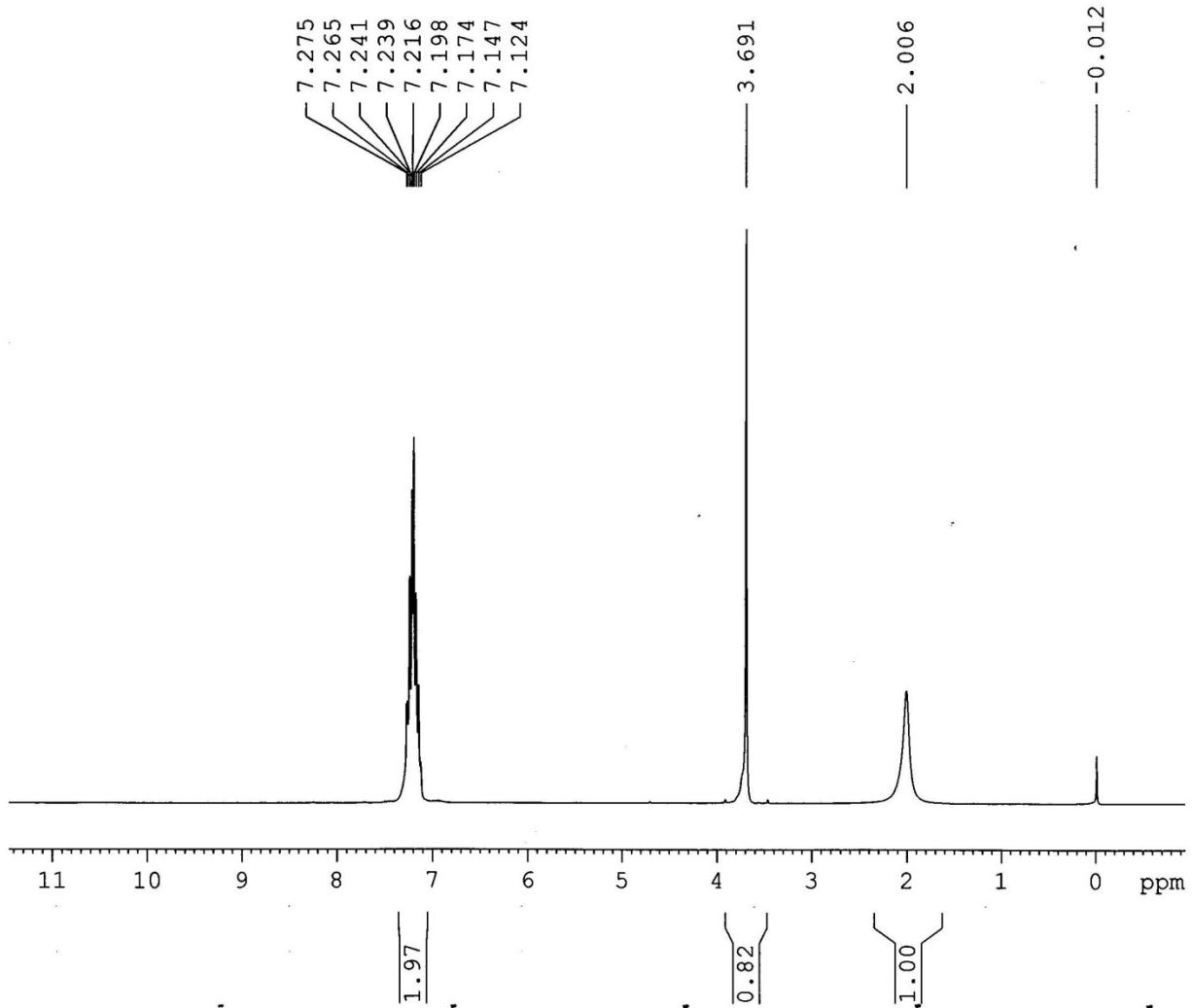


Fig. S11 ${}^1\text{H}$ NMR spectrum of phenylmethanamine in CDCl_3 .

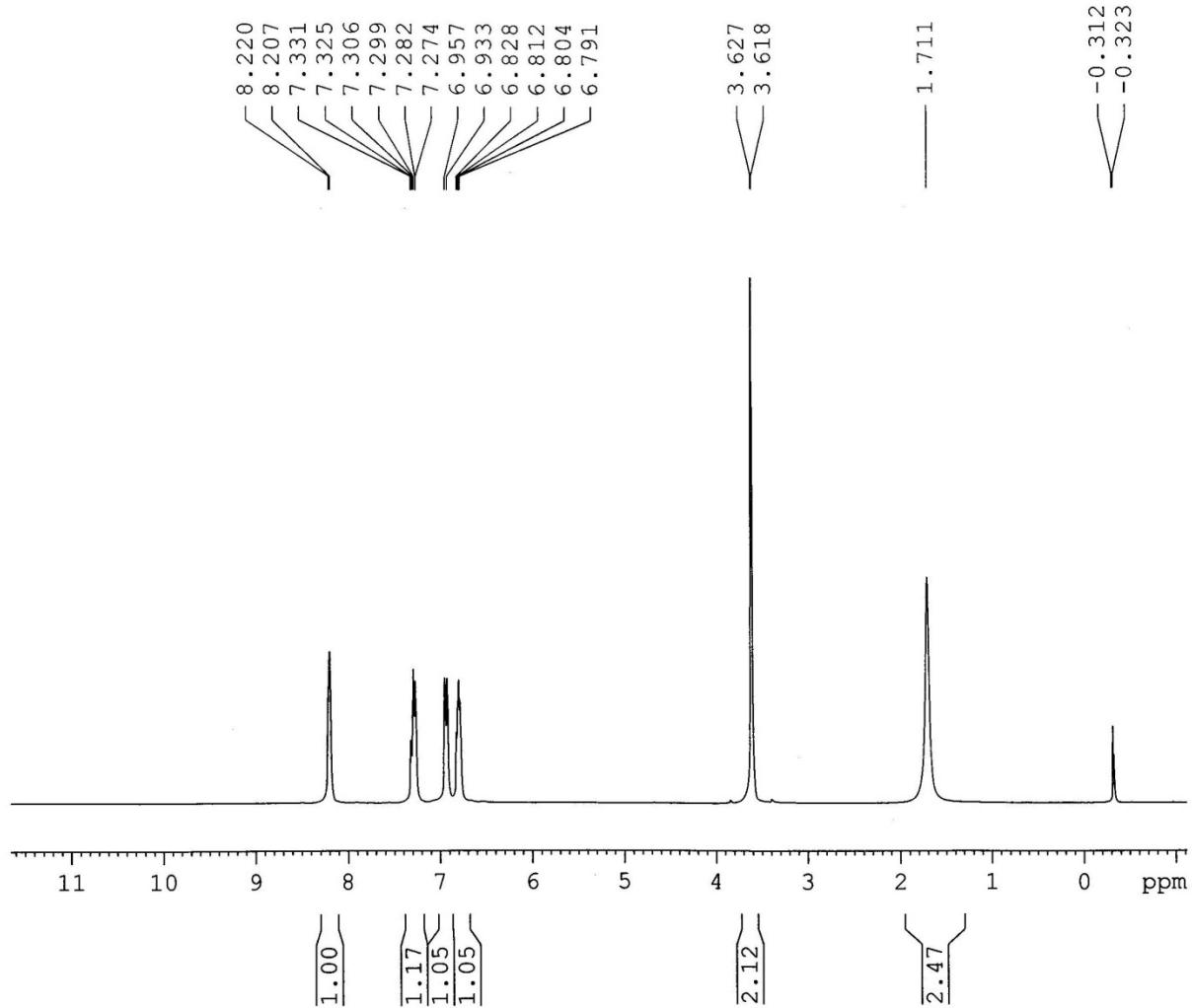


Fig. S12 ^1H NMR spectrum of pyridin-2-ylmethanamine in CDCl_3 .

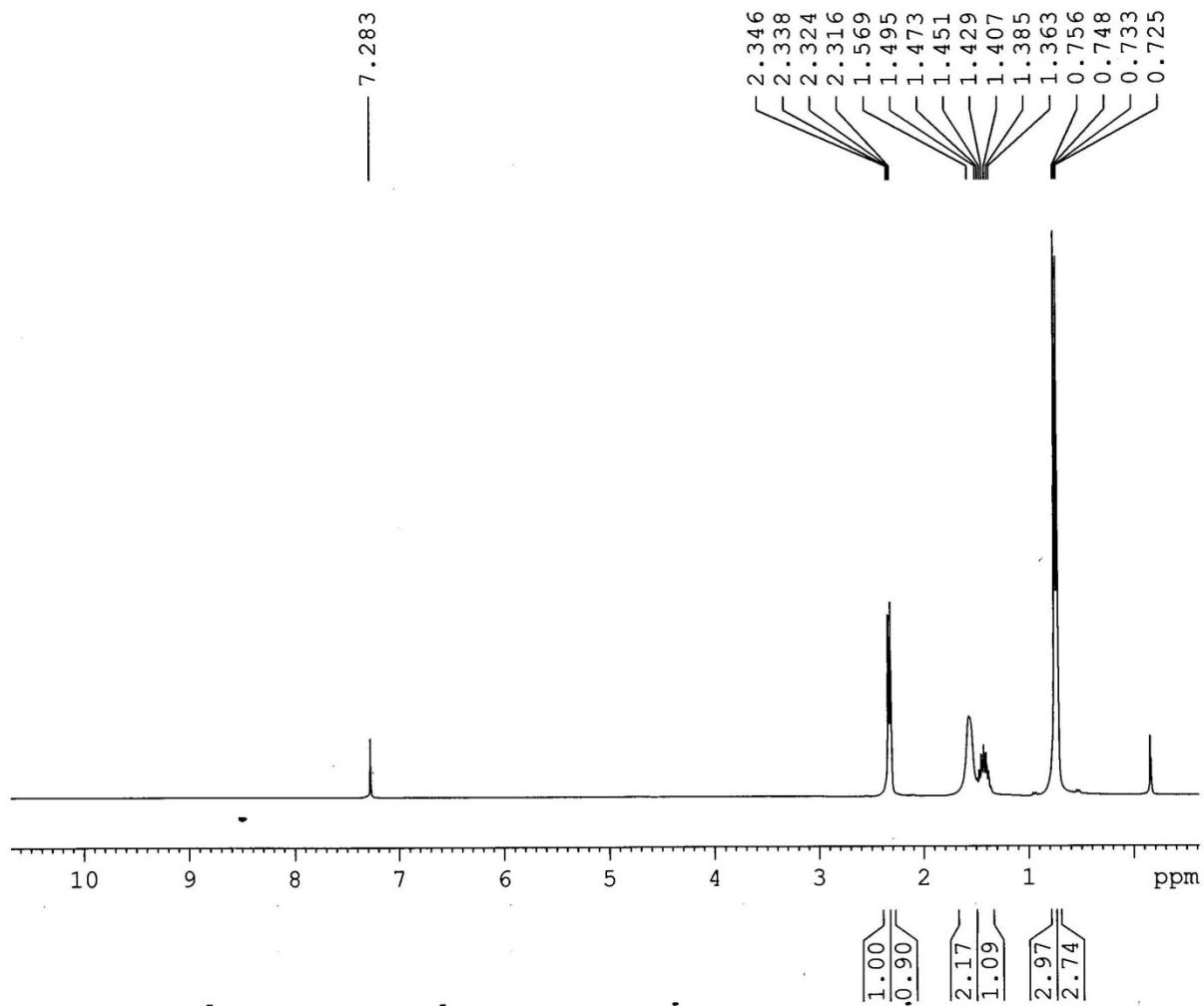


Fig. S13 ^1H NMR spectrum of 2-methylpropan-1-amine in CDCl_3 .

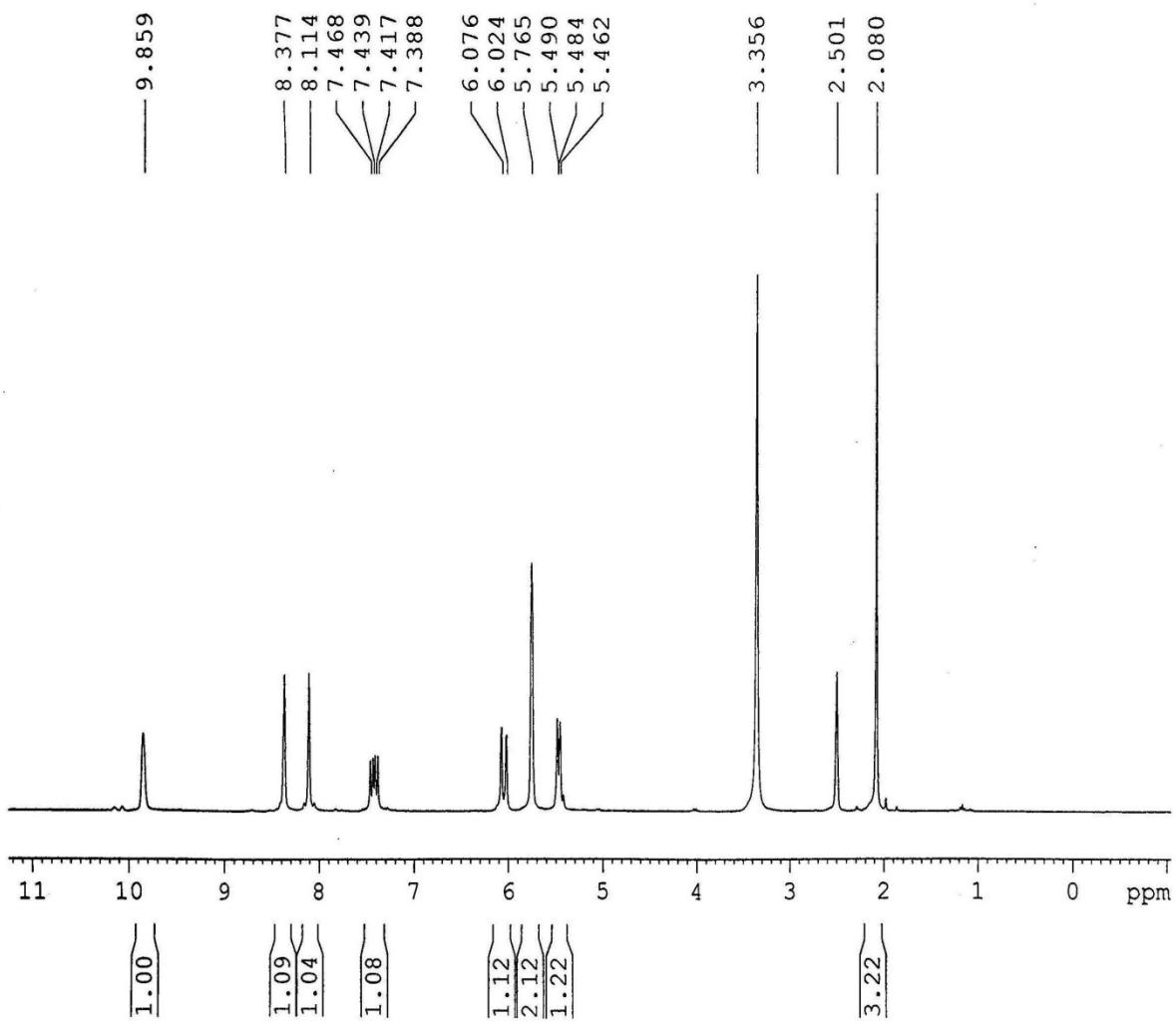


Fig. S14 ^1H NMR spectrum of 3-Cyanomethyl-1-vinylimidazolium chloride in $\text{DMSO}-d_6$.