Electronic Supplementary Information

A rapid-response electrochromic device with significantly enhanced electrochromic performance

Experimental Section

 $Na_2WO_4 \cdot 2H_2O_5$, sulfuric acid, H_2O_2 , acetone, ethanol, LiClO₄, propylene carbonate (PC), 3,4ethylenedioxythiophene (EDOT), 1-butyl-3-methylimidazolium hexafluorophosphate ([Bmim]PF₆) (various chemical suppliers) used in this study were reagent grade.

Electrodeposition of three-dimensionally ordered macroporous WO₃ films. Indium-tin oxide (ITO) coated glass substrates (~9 Ω cm⁻², 1 cm×4 cm) and double-sided ITO-coated glass substrates (~60 Ω cm⁻², 1 cm×4 cm) were ultrasonically cleaned in acetone, ethanol and then distilled water for 20 min, separately. Monodispersed polystyrene (PS) latex spheres (diameters of 450 nm) were obtained using an emulsier-free emulsion polymerization technique.¹ PS colloidal crystal templates were grown using a controlled vertical drying method.¹ The cleaned ITO-coated glass substrates were vertically placed into cylindrical glass vessels. The PS sphere suspension diluted to 0.5 wt% was added into glass vessels and then evaporated in an incubator at a stable temperature of 60 °C to obtain polystyrene colloidal crystal templates. Cathodic electrodeposition of WO₃ into polystyrene colloidal crystal templates was performed at a constant current of 2 mA in an ice-water-bath mixture of 10 mL of 0.025 mol/L Na₂WO₄, 0.05 mol/L H₂SO₄ and one drop of 30% H₂O₂ for 5 min with Pt foil as the counter electrode. After electrodeposition, samples were immersed in the toluene for 24 h to remove the polystyrene templates. Finally, the as-prepared samples were dried at 70 °C for 30 min.

Electropolymerization of PEDOT films. [Bmim]PF₆ was used after drying under vacuum condition at 100 $^{\circ}$ C until the water content below 1 ppm. EDOT was used as delivered. PEDOT films were electropolymerized at a constant voltage of 1.4 V versus Ag (1.0 mm diameter wire) from [Bmim]PF₆ containing 0.1M EDOT monomer and Pt ring as the counter electrode. Then the films were immersed in ethanol to remove the residue monomers.

Characterization. The morphologies of 3DOM WO₃ films and PEDOT films were characterized by scanning electron microscopy (SEM, Hitachi S-4800). Crystalline structures of the 3DOM WO₃ films were investigated using a Rigaku D/mol/Lax-3C equipment with a sweep rate of 5°/min through a diffraction angle range of 20–70°. The Raman spectra were determined on a Renishaw in Via micro Raman spectroscopy system, using the TE air-cooled 576×400 CCD array in a confocal Raman system (wavelength: 633 nm). The incident laser power was kept at 0.1 mW, ×100 objective and an integration times of 5 s were employed. In situ visible and near-IR (NIR) electrochromic measurements of ECDs and NECDs were performed using an experimental setup produced in-house² in combination with a CHI 660E electrochemical workstation (Shanghai Chenhua Instrument Co. Ltd.). The experimental setup was sealed in an argon-filled glove box (Vigor Glove Box from Suzhou, China) before testing. One side of the setup was connected to a white lamp (DT-mini-2-GS, Ocean Optics) by an optical fiber; the other side was connected to an optic spectrometer (MAYA 2000-Pro, Ocean Optics). The transmittances of the ECDs and NECDs without electrochromic layers and electrolyte were used as a reference for 100% transmittance. Chronoamperometry measurements were done by applying +3 V for 1000 s to de-intercalate all Li⁺ ions present, then setting the potential back to -3 V for 20 s followed by +3 V for 20 s, and monitoring the current as a function of time during testing.

Fig. S1 shows the coloration/bleaching transmittance response curves of the WO₃-NECD and the PEDOT-NECD for the first, 150th and 300th cycles with alternately applied potentials of ± 3 V. The NECDs retain the electrochromic properties after 300 switching cycles, with little changes of optical contrast and switching time.



Fig. S1. Change in optical transmittance versus time for (a) the WO₃-NECD and (b) the PEDOT-NECD at initial cycle, after 150 cycles and after 300 cycles with alternately applied potentials of ± 3 V.

References

- 1 M. A. McLachlan, N. P. Johnson, M. Richard and D. W. McComb, Journal of Materials Chemistry, 2004, 14, 144.
- 2 G. Bar, N. Larina, L. Grinis, V. Lokshin, R. Gvishi, I. Kiryuschev, A. Zaban and V. Khodorkovsky, *Solar Energy Materials and Solar Cells*, 2012, **99**, 123.