

Supporting Information

Ring-opening of Enantiomerically Pure Oxa-containing Heterocycles with Phosphorus Nucleophiles

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(A) Single crystal X-ray structure determinations

X-Ray Data: Single crystals of enantiopure compounds **2a**, **ent-2b**, **2c**, **2d**, **5**, and **11** suitable for X-ray diffraction analysis were grown by slow diffusion of *n*-hexane into EtOAc solutions of each compound at room temperature. Single crystals of enantiopure compound **9** suitable for X-ray diffraction analysis were grown by slow diffusion of *n*-hexane into DCM solutions of each compound at room temperature. The measured crystals were prepared under inert conditions immersed in perfluoropolyether as protecting oil for manipulation.

Data collection: Crystal structure determinations of **2c**, **2d**, **5**, **9**, and **11** were carried out using a Bruker-Nonius diffractometer equipped with an APEX 2 4K CCD area detector, a FR591 rotating anode with Mo_K α radiation, Montel mirrors as monochromator and a Oxford Cryosystems low temperature device Cryostream 700 plus (T = –173 °C). Crystal structure determinations for **2a**, and **ent-2b** was carried out using a Apex DUO Kappa 4-axis goniometer equipped with an APEX 2 4K CCD area detector, a Microfocus Source E025 IuS using Mo_K α radiation, Quazar MX multilayer Optics as monochromator and a Oxford Cryosystems low temperature device Cryostream 700 plus (T = –173 °C). Full-sphere data collection was used with ω and ϕ scans. Programs used: Data collection APEX-2,¹ data reduction Bruker Saint,² and absorption correction SADABS.³

Structure Solution and Refinement: Crystal structure solution was achieved using direct methods as implemented in SHELXTL⁴ and visualized using the program XP. Absolute configuration was determined based on the Flack parameter.⁵ Missing atoms were subsequently located from difference Fourier synthesis and added to the atom list. Least-squares refinement on F² using all measured intensities was carried out using the program SHELXTL. All non-hydrogen atoms were refined including anisotropic displacement parameters.

¹ Data collection with: *APEX II* versions v1.0-22, v2009.1-0 and v2009.1-02, Bruker AXS Inc., Madison, Wisconsin, USA, 2007.

² Data reduction with: *SAINT* versions V.2.10, V/.60A and V7.60A, Bruker AXS Inc., Madison, Wisconsin, USA, 2003/2007.

³ *SADABS*, V.2.10, V2008 and V2008/1, Bruker AXS Inc., Madison, Wisconsin, USA, 2003/2001, see: R. H. Blessing, *Acta Crystallogr., Sect. A*, 1995, **51**, 33.

⁴ *SHELXTL*, versions V6.12 and 6.14, see: G. M. Sheldrick, *Acta Crystallogr., Sect. A*, 2008, **64**, 112.

⁵ a) H. D. Flack, *Acta Crystallogr., Sect. A*, 1983, **39**, 876; b) H. D. Flack and G. J. Bernardelli, *J. Appl. Crystallogr.*, 2000, **33**, 1143.

Crystal data for 2a: $C_{22}H_{26}B_1O_2P_1$, Mr = 364.21; monoclinic; space group $P2_1$, $a = 9.3550(7)$ Å, $b = 33.043(3)$ Å, $c = 13.5151(11)$ Å, $\beta = 107.899(2)^\circ$, $V = 3975.6(5)$ Å³, $Z = 8$, $\rho_{cal} = 1.217$ Mg/m³, $\mu = 0.151$ mm⁻¹, 52980 reflections were collected of which 20660 are unique ($R_{int} = 0.0572$), 18650 $F_o > 4\sigma(F_o)$, 1003 refined parameters, $R1 [I > 2\sigma(I)] = 0.0554$, $wR2 [I > 2\sigma(I)] = 0.1304$, Flack parameter: 0.06(6), Goodness of fit on $F2 = 1.077$, maximum residual electron density 0.553 (−0.326) e Å³.

Crystal data for ent-2b: $C_{40}H_{38}B_1O_2P_1$, Mr = 592.48; monoclinic; space group $P2_1$, $a = 9.3160(15)$ Å, $b = 14.835(3)$ Å, $c = 12.0350(19)$ Å, $\beta = 102.299(3)^\circ$, $V = 1625.1(5)$ Å³, $Z = 2$, $\rho_{cal} = 1.211$ Mg/m³, $\mu = 0.119$ mm⁻¹, 10052 reflections were collected of which 6869 are unique ($R_{int} = 0.0210$), 6204 $F_o > 4\sigma(F_o)$, 688 refined parameters, $R1 [I > 2\sigma(I)] = 0.0400$, $wR2 [I > 2\sigma(I)] = 0.0907$, Flack parameter: 0.00(7), Goodness of fit on $F2 = 1.037$, maximum residual electron density 0.224 (−0.239) e Å³.

Crystal data for 2c: $C_{22}H_{38}B_1O_2P_1$, Mr = 376.30; orthorhombic; space group $P2_12_12_1$, $a = 9.8800(6)$ Å, $b = 10.5070(6)$ Å, $c = 20.9290(13)$ Å, $V = 2172.6(2)$ Å³, $Z = 4$, $\rho_{cal} = 1.150$ Mg/m³, $\mu = 0.140$ mm⁻¹, 28028 reflections were collected of which 10164 are unique ($R_{int} = 0.0249$), 9515 $F_o > 4\sigma(F_o)$, 238 refined parameters, $R1 [I > 2\sigma(I)] = 0.0327$, $wR2 [I > 2\sigma(I)] = 0.0878$, Flack parameter: −0.01(4), Goodness of fit on $F2 = 1.073$, maximum residual electron density 0.427 (−0.207) e Å³.

Crystal data for 2d: $C_{40}H_{50}B_1O_2P_1$, Mr = 604.58; monoclinic; space group $P2_1$, $a = 9.6592(6)$ Å, $b = 12.9354(11)$ Å, $c = 14.2059(9)$ Å, $\beta = 106.664(2)^\circ$, $V = 604.58$ Å³, $Z = 2$, $\rho_{cal} = 1.181$ Mg/m³, $\mu = 0.114$ mm⁻¹, 23001 reflections were collected of which 12939 are unique ($R_{int} = 0.0249$), 12301 $F_o > 4\sigma(F_o)$, 399 refined parameters, $R1 [I > 2\sigma(I)] = 0.0552$, $wR2 [I > 2\sigma(I)] = 0.1408$, Flack parameter: −0.04(5), Goodness of fit on $F2 = 1.056$, maximum residual electron density 0.857 (−0.806) e Å³.

Crystal data for 5: $C_{22}H_{26}B_1O_2P_1$, Mr = 364.21; orthorhombic; space group $P2_12_12_1$, $a = 14.456(3)$ Å, $b = 15.0089(18)$ Å, $c = 28.280(4)$ Å, $V = 6135.8(17)$ Å³, $Z = 12$, $\rho_{cal} = 1.183$ Mg/m³, $\mu = 0.147$ mm⁻¹, 28811 reflections were collected of which 16496 are unique ($R_{int} = 0.0249$), 11016 $F_o > 4\sigma(F_o)$, 706 refined parameters, $R1 [I > 2\sigma(I)] = 0.0514$, $wR2 [I > 2\sigma(I)] = 0.1031$, Flack parameter: −0.07(7), Goodness of fit on $F2 = 1.007$, maximum residual electron density 0.277 (−0.314) e Å³.

Crystal data for 9: $C_{22}H_{32}B_1O_1P_1$, Mr = 354.26; monoclinic; space group $P2_1$, $a = 10.2789(7)$ Å, $b = 12.6688(7)$ Å, $c = 16.3394(12)$ Å, $\beta = 101.241(3)^\circ$, $V = 2086.9(2)$ Å³,

$Z = 4$, $\rho_{\text{cal}} = 1.128 \text{ Mg/m}^3$, $\mu = 0.138 \text{ mm}^{-1}$, 15818 reflections were collected of which 10592 are unique ($R_{\text{int}} = 0.0449$), 8724 $F_o > 4\sigma(F_o)$, 459 refined parameters, $R_1 [I > 2\sigma(I)] = 0.0516$, $wR_2 [I > 2\sigma(I)] = 0.1270$, Flack parameter: $-0.06(8)$, Goodness of fit on $F^2 = 1.015$, maximum residual electron density $0.521 (-0.334) \text{ e } \text{\AA}^{-3}$.

Crystal data for 11: $\text{C}_{20}\text{H}_{22}\text{B}_1\text{O}_1\text{P}_1$, $M_r = 320.16$; monoclinic; space group $P2_1$, $a = 9.1144(6) \text{ \AA}$, $b = 10.1755(7) \text{ \AA}$, $c = 9.5868(9) \text{ \AA}$, $\beta = 97.133(4)^\circ$, $V = 882.23(12) \text{ \AA}^3$, $Z = 2$, $\rho_{\text{cal}} = 1.205 \text{ Mg/m}^3$, $\mu = 0.157 \text{ mm}^{-1}$, 14246 reflections were collected of which 6941 are unique ($R_{\text{int}} = 0.0258$), 6195 $F_o > 4\sigma(F_o)$, 215 refined parameters, $R_1 [I > 2\sigma(I)] = 0.0382$, $wR_2 [I > 2\sigma(I)] = 0.0937$, Flack parameter: $0.14(5)$, Goodness of fit on $F^2 = 1.052$, maximum residual electron density $0.526 (-0.204) \text{ e } \text{\AA}^{-3}$.

CCDC 931316–931319, CCDC 931321, CCDC 931323 and CCDC 931325 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

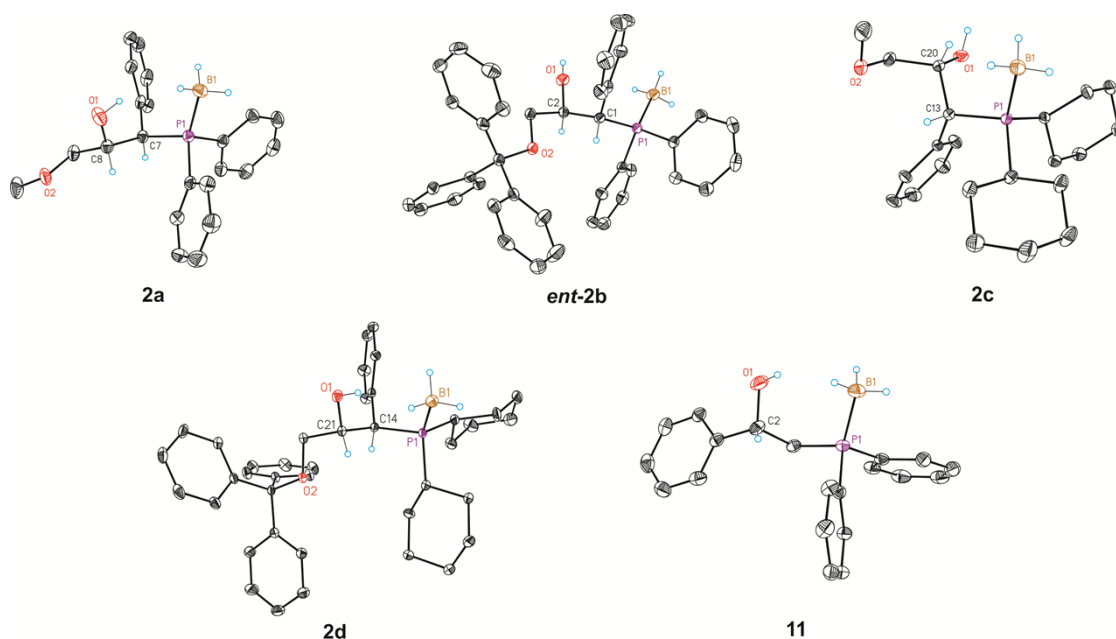
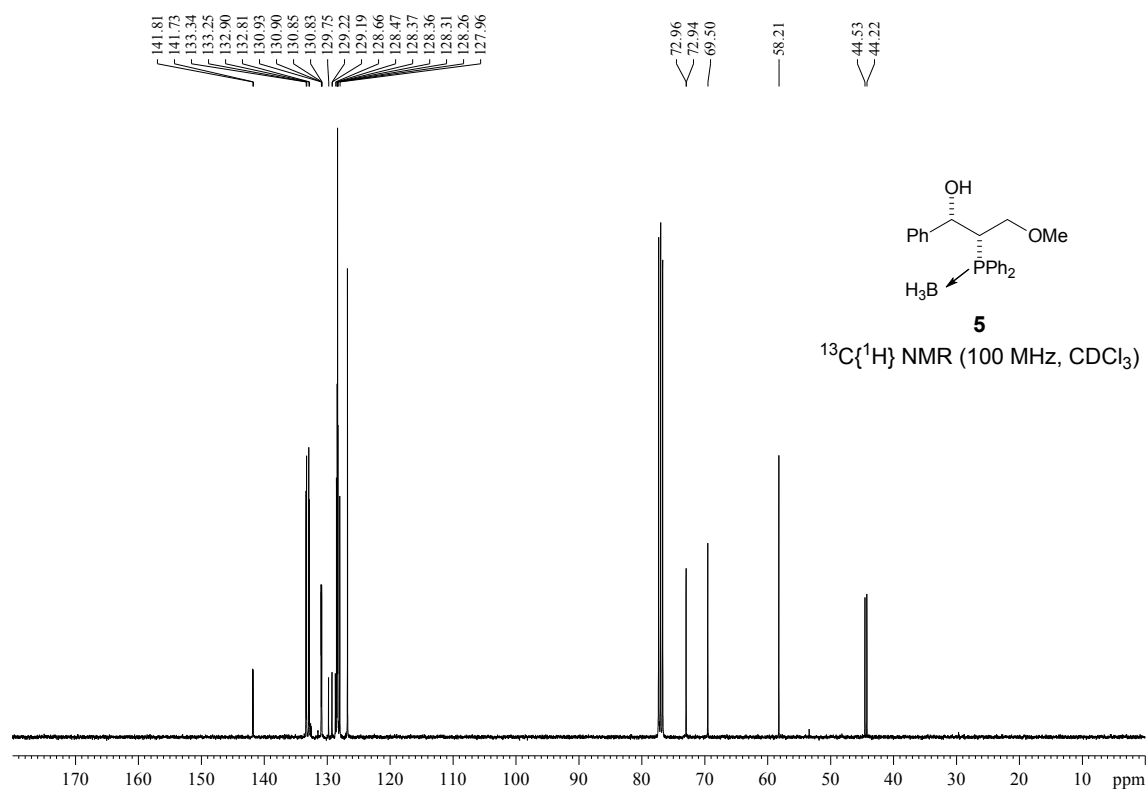
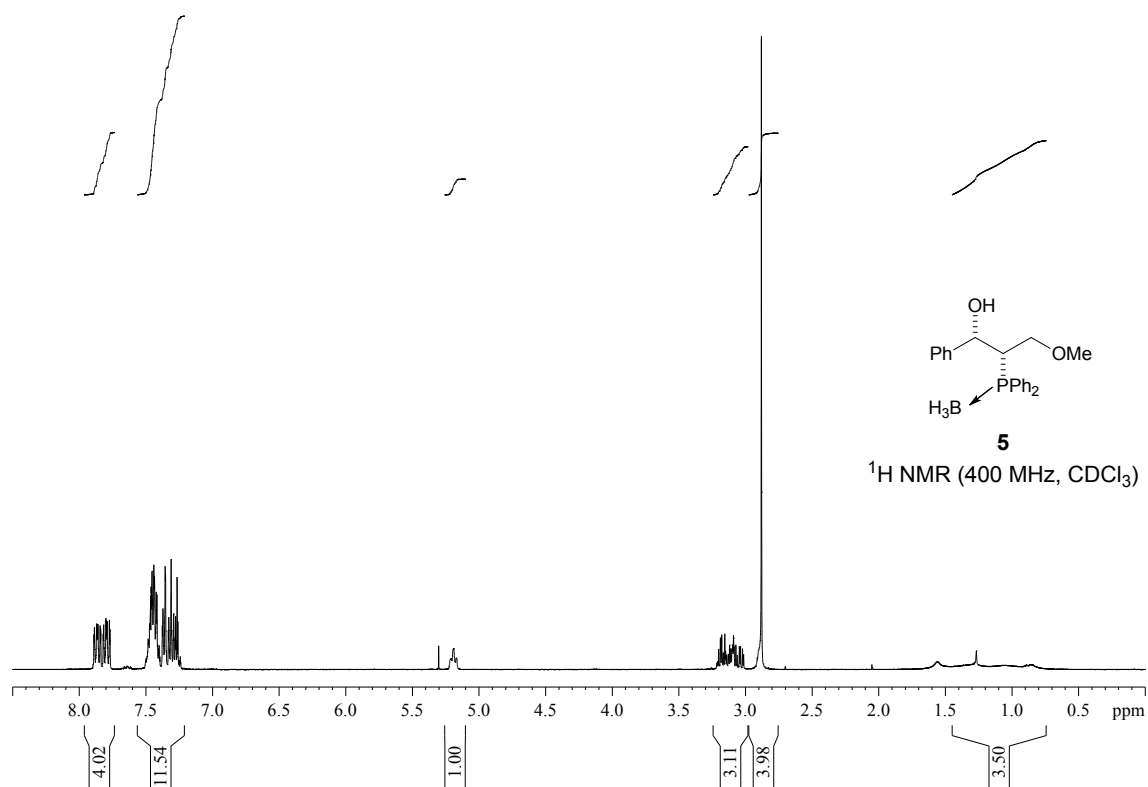
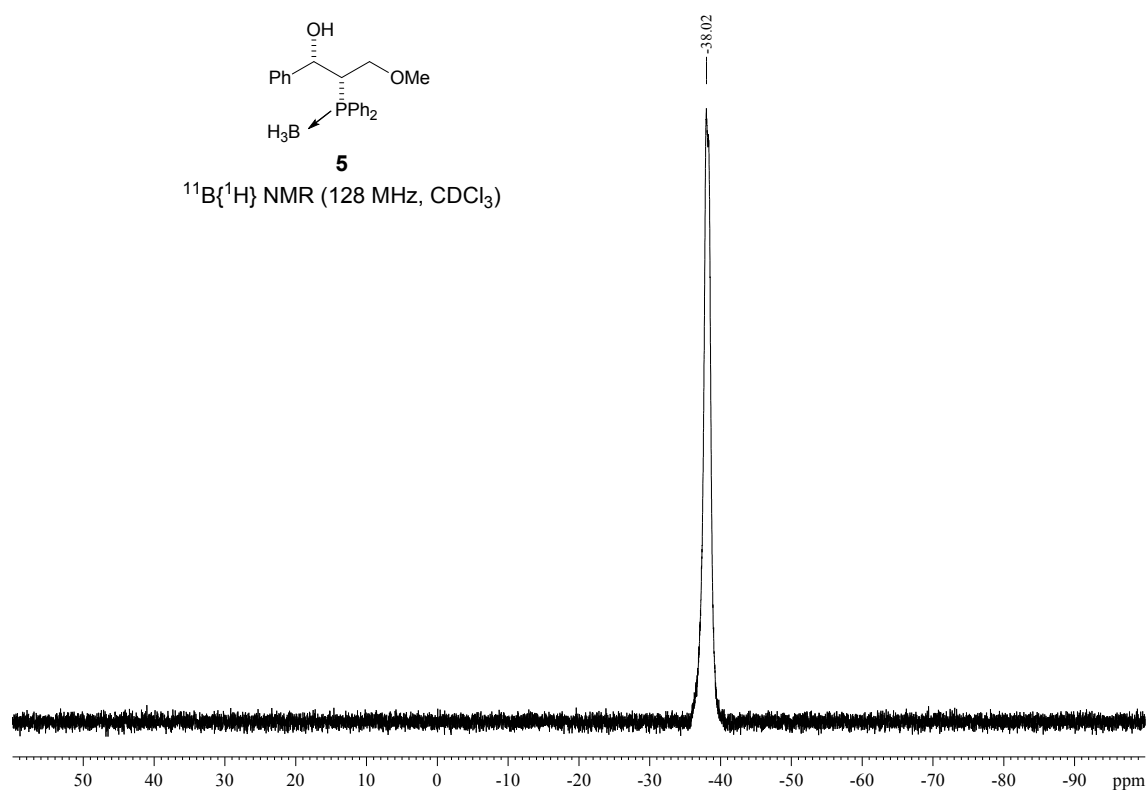
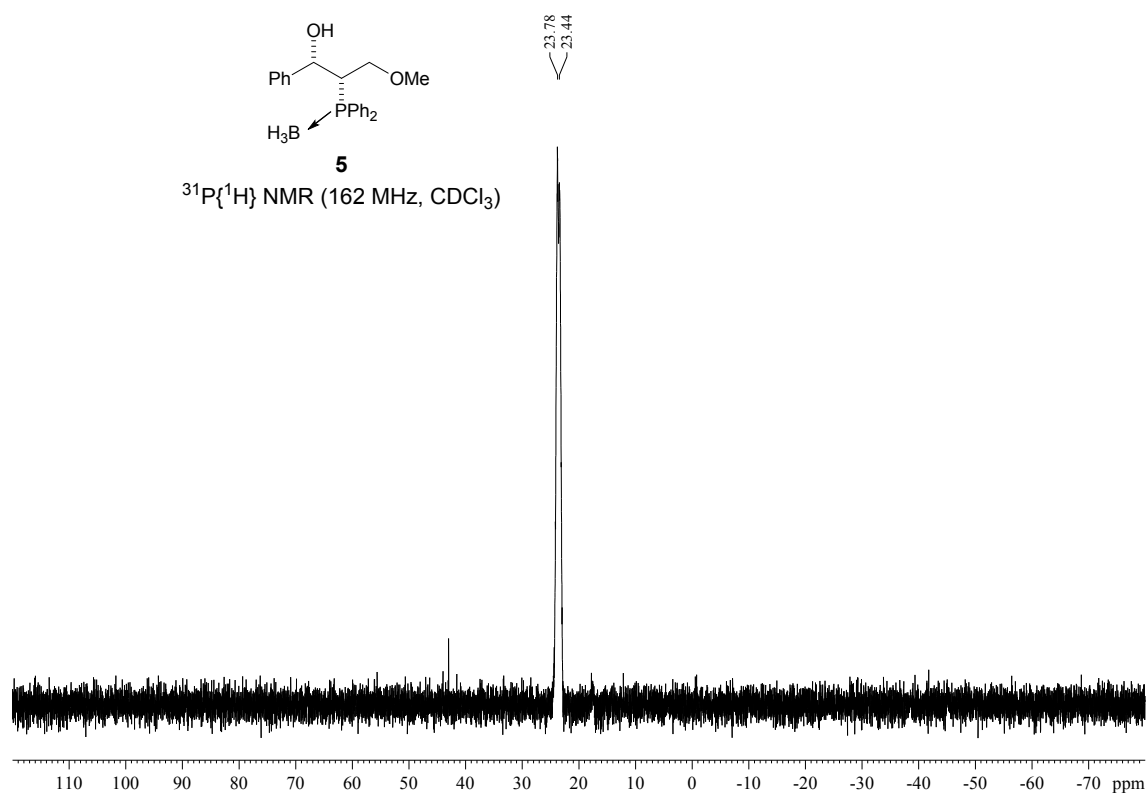
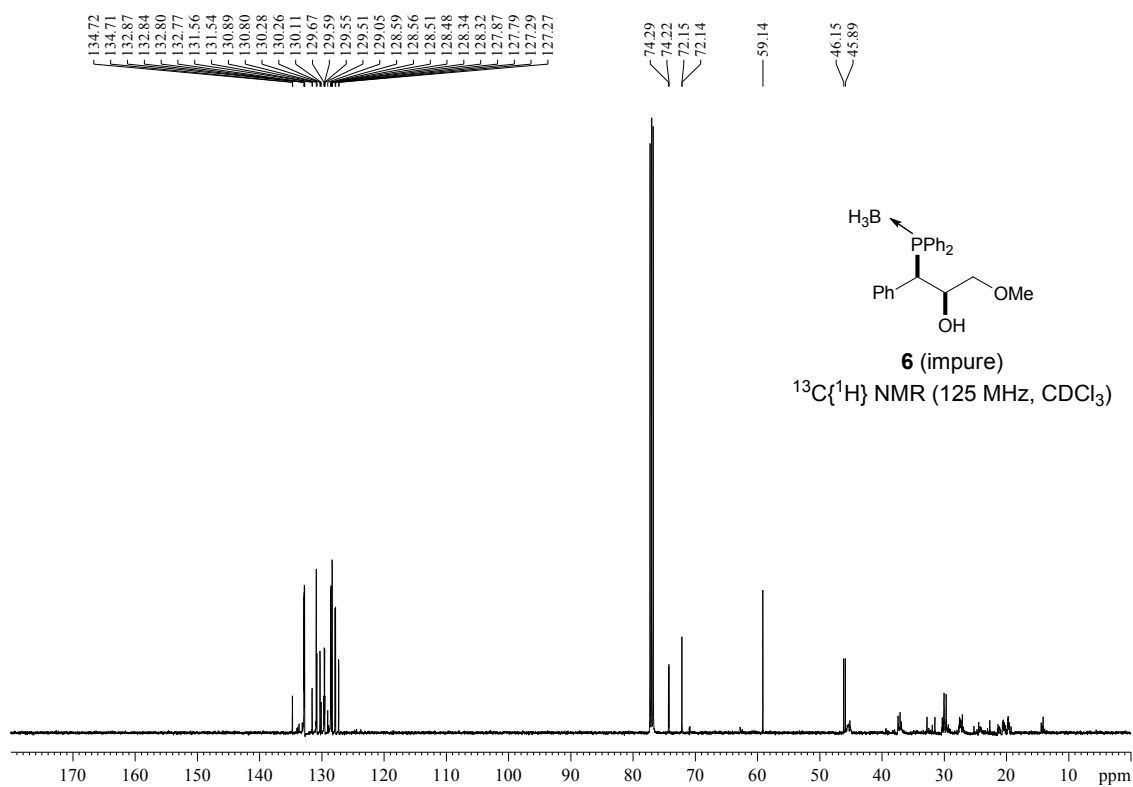
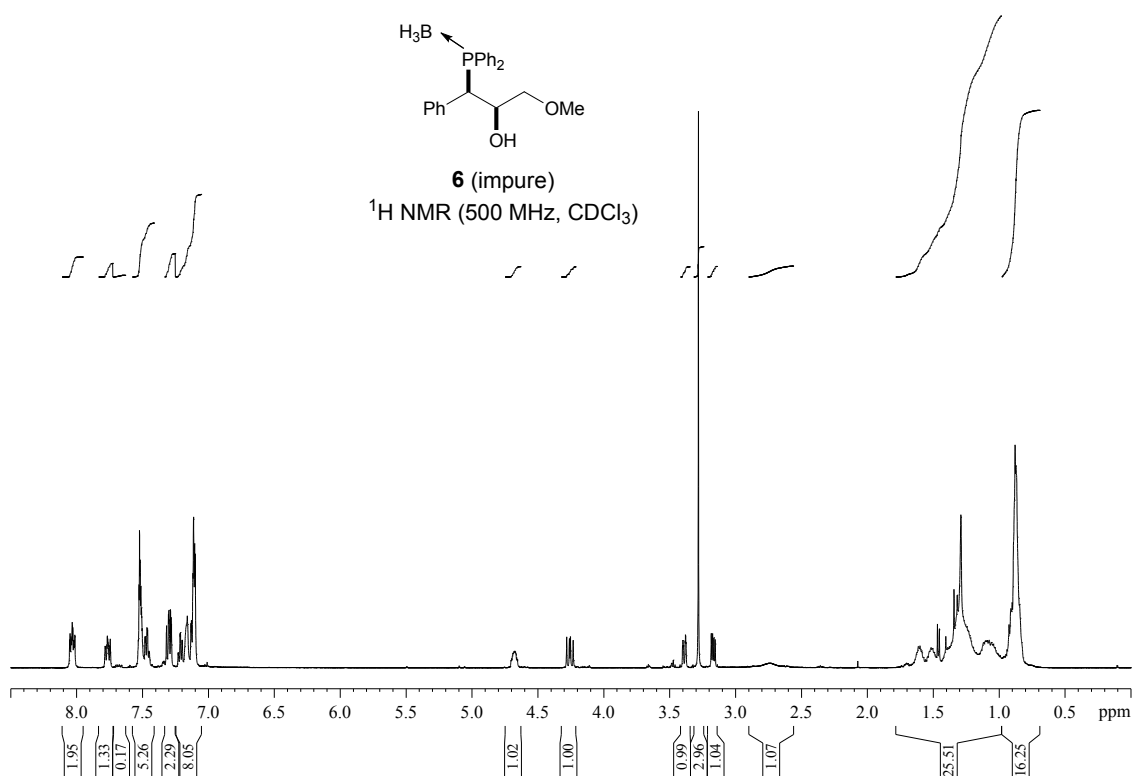


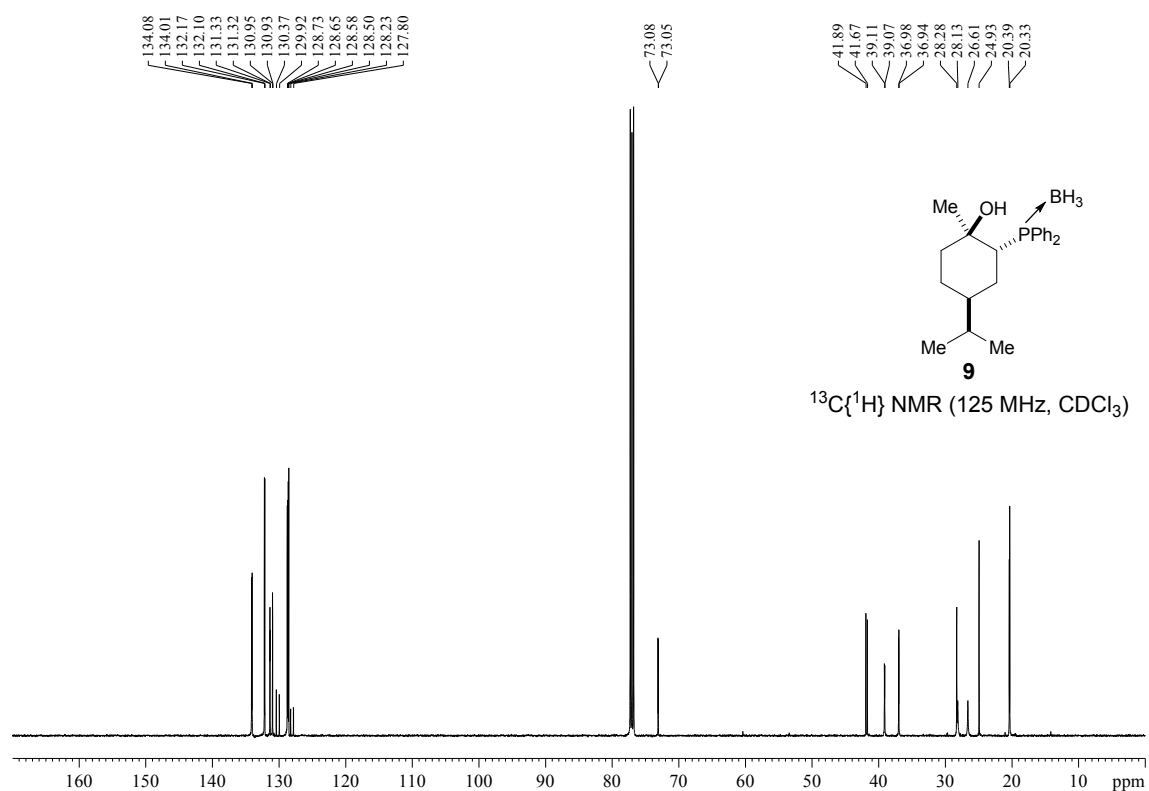
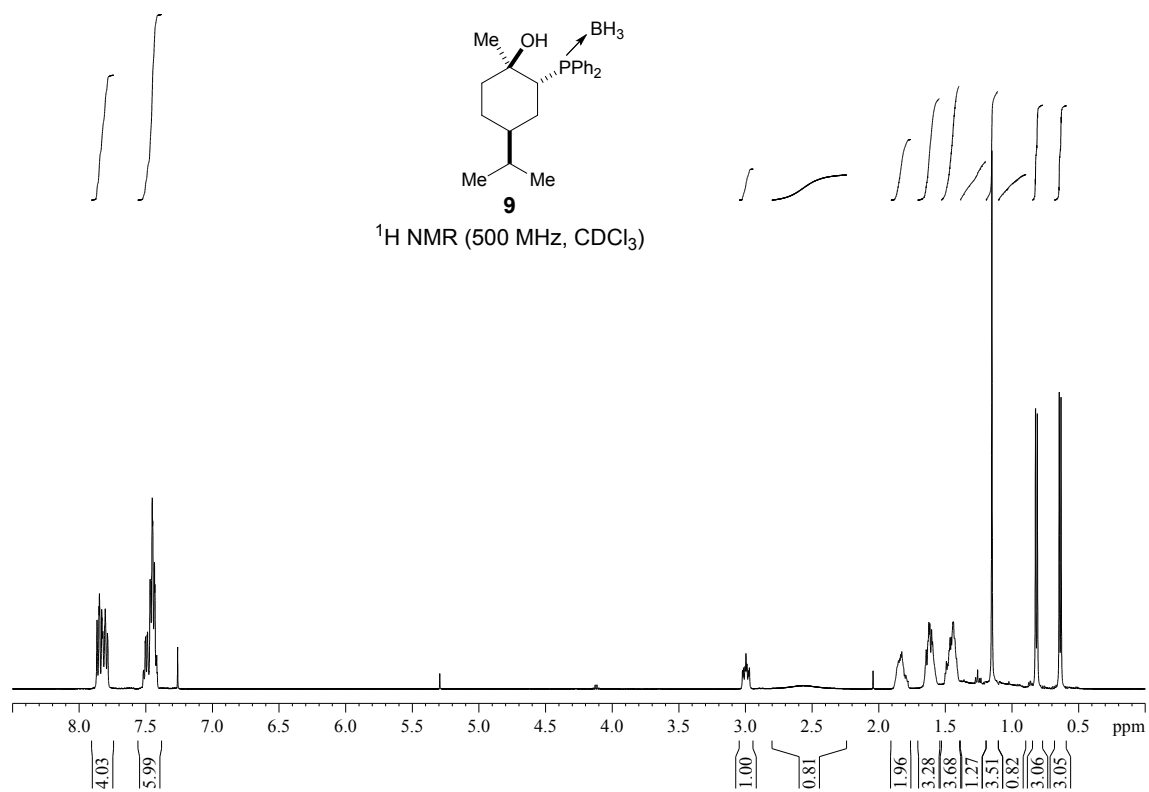
Fig. SI 1. Crystal structures of **2a**, **ent-2b**, **2c**, **2d**, and **11** (ORTEP drawing showing thermal ellipsoids at 50% probability). Non-relevant hydrogen atoms have been omitted for clarity.

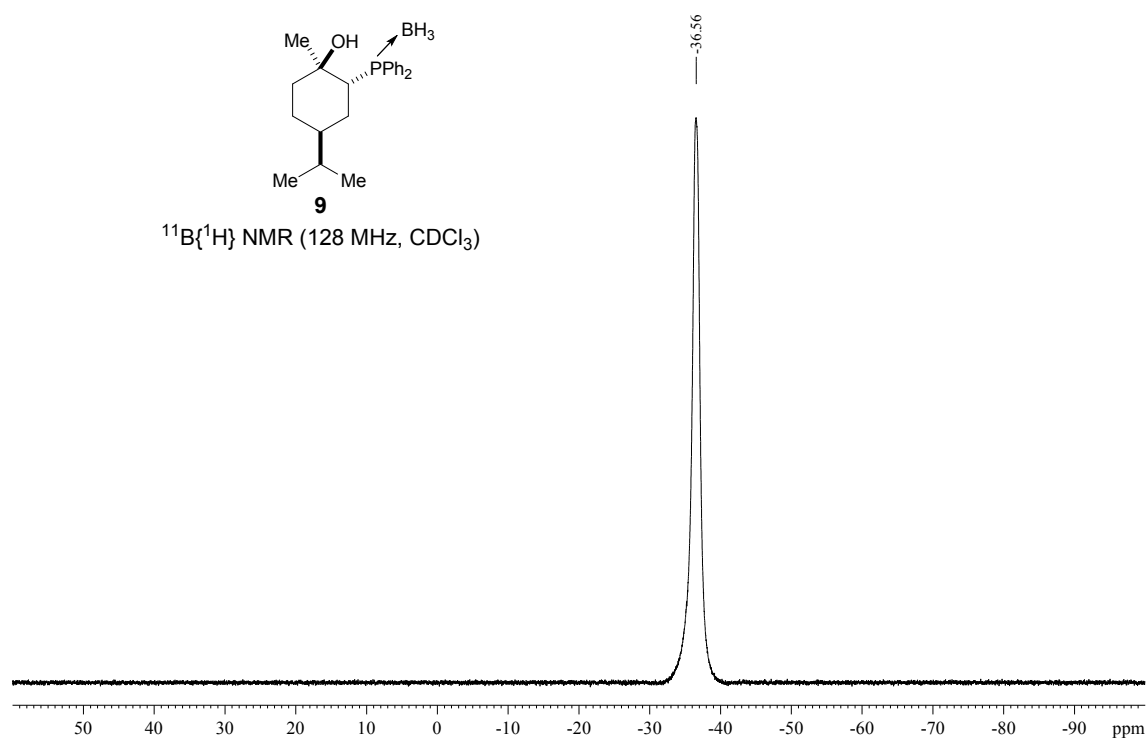
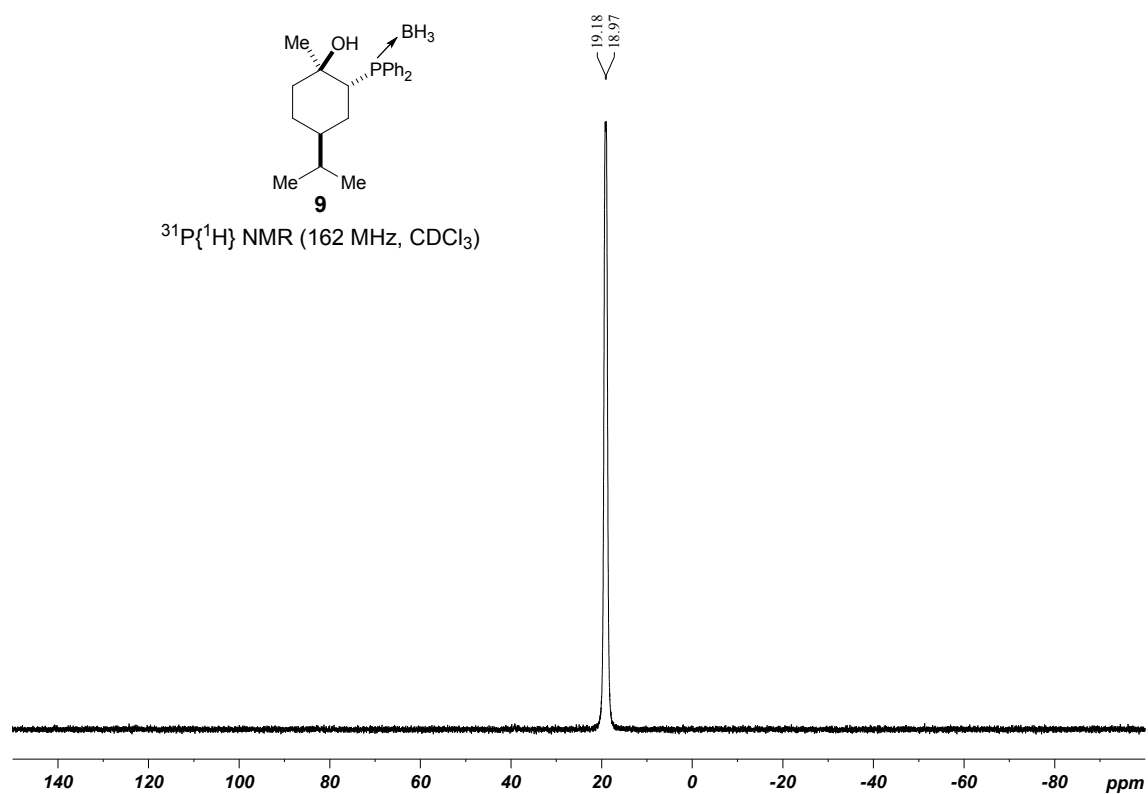
(B) NMR spectra of new compounds

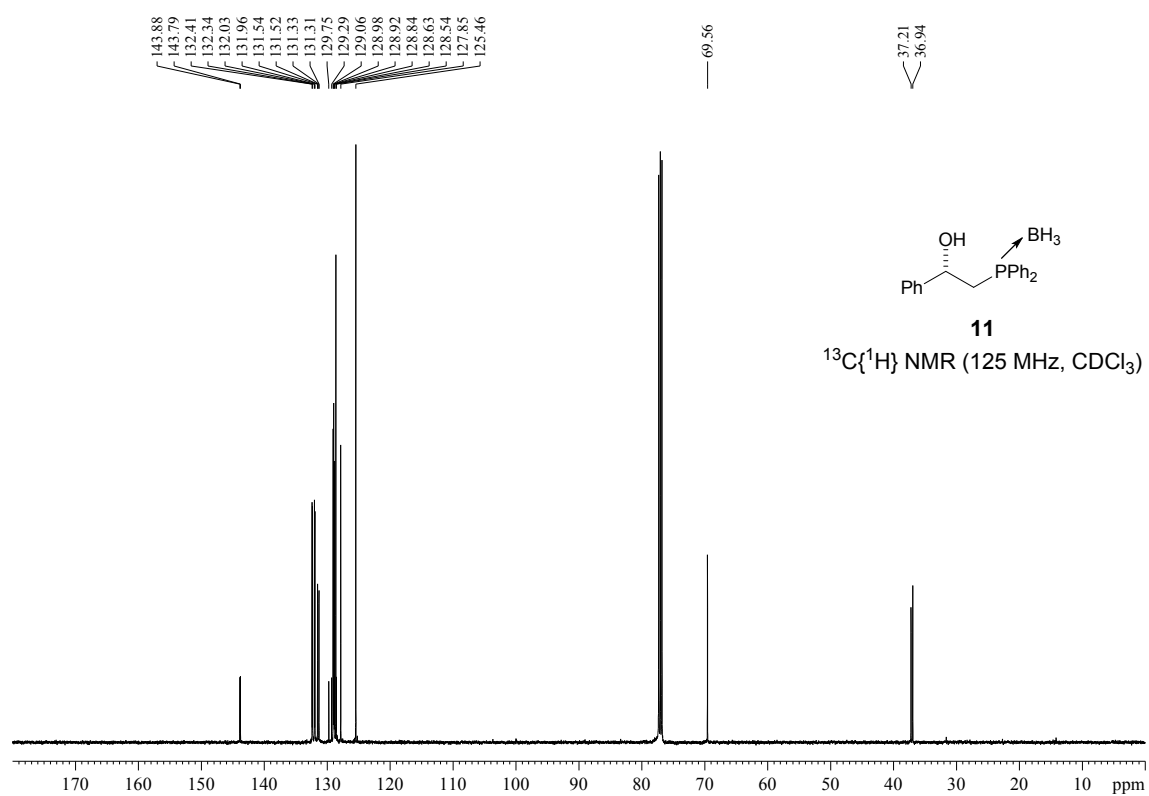
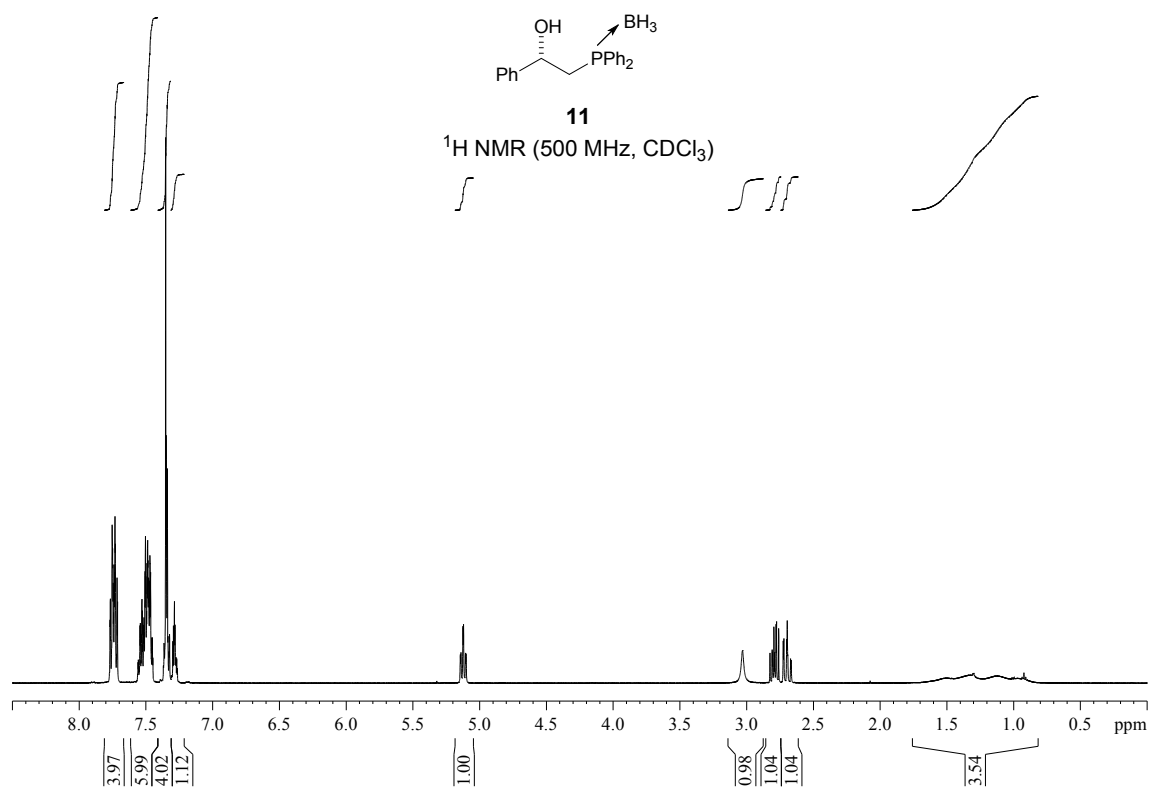


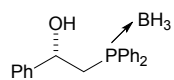






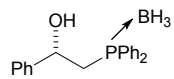
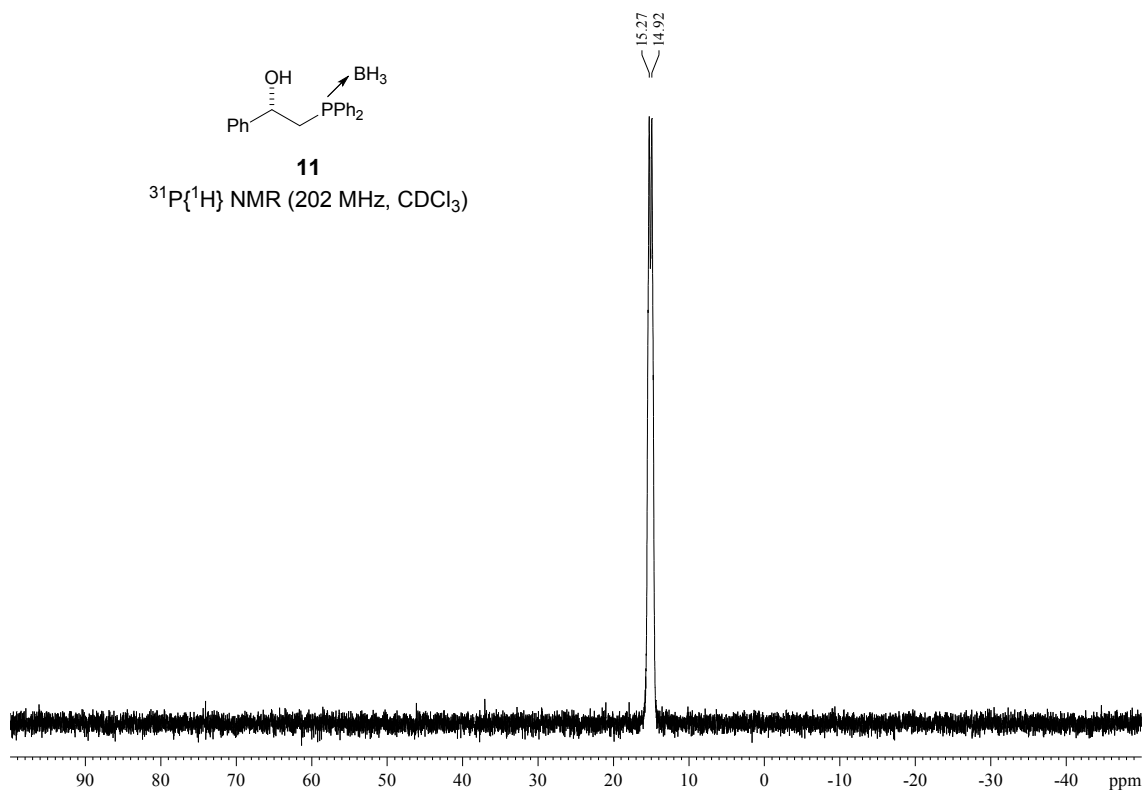






11

$^{31}\text{P}\{^1\text{H}\}$ NMR (202 MHz, CDCl_3)



11

$^{11}\text{B}\{^1\text{H}\}$ NMR (160 MHz, CDCl_3)

