

## Supplementary Information

### **Theoretical Calculations Based Synthesis of Poly(p-phenylenediamine)-Fe<sub>3</sub>O<sub>4</sub> Composite: A Magnetically Recyclable Photocatalyst with Highly Selectivity for Acid Dyes**

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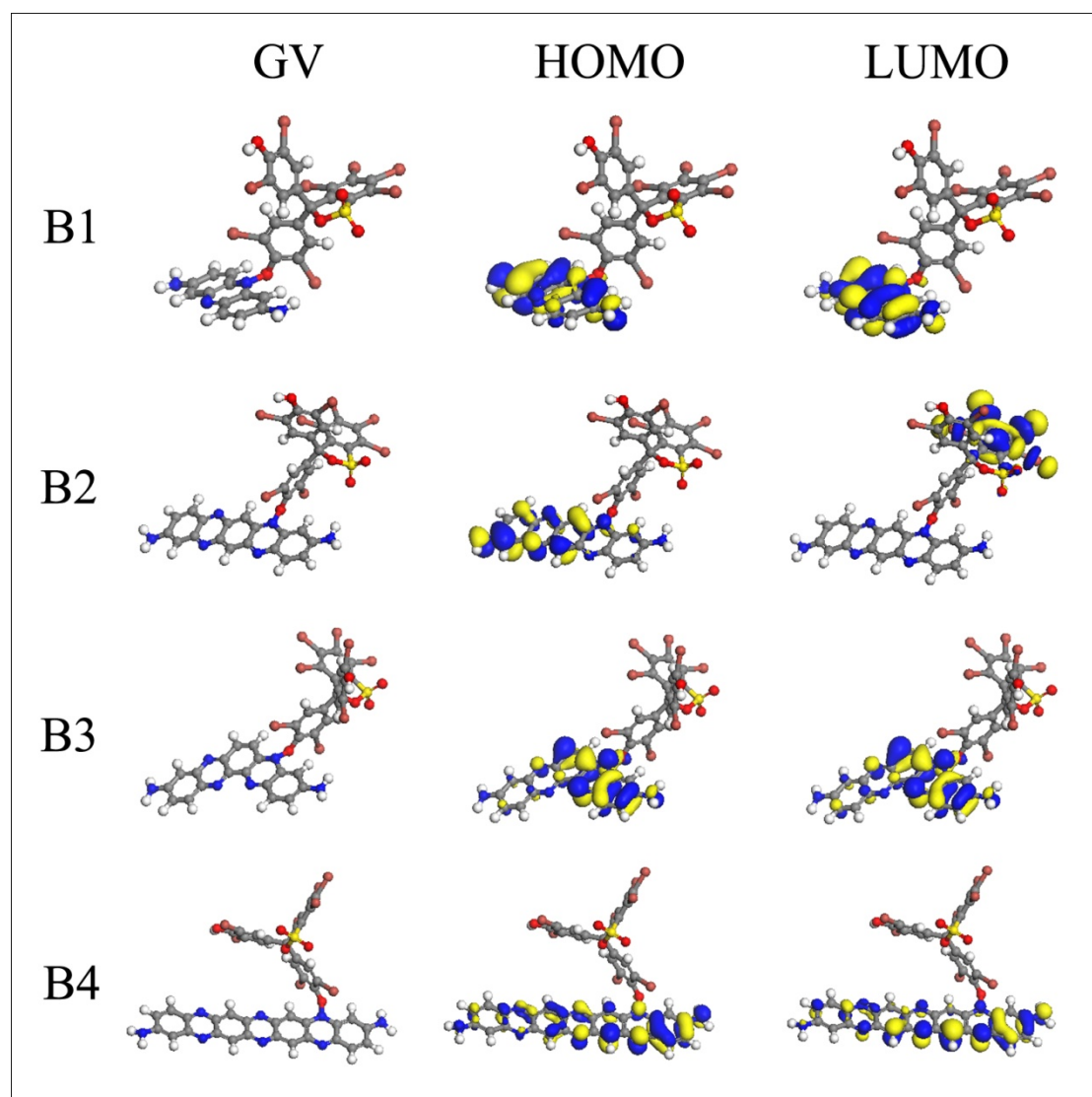
Dr. Siwei Yang: E-mail: yangsiwei@mail.sim.ac.cn, Tel. /fax: +86 021 62511070 420.

**Table S1** The surface area and pore structure parameters of Fe<sub>3</sub>O<sub>4</sub>-PpPD composite

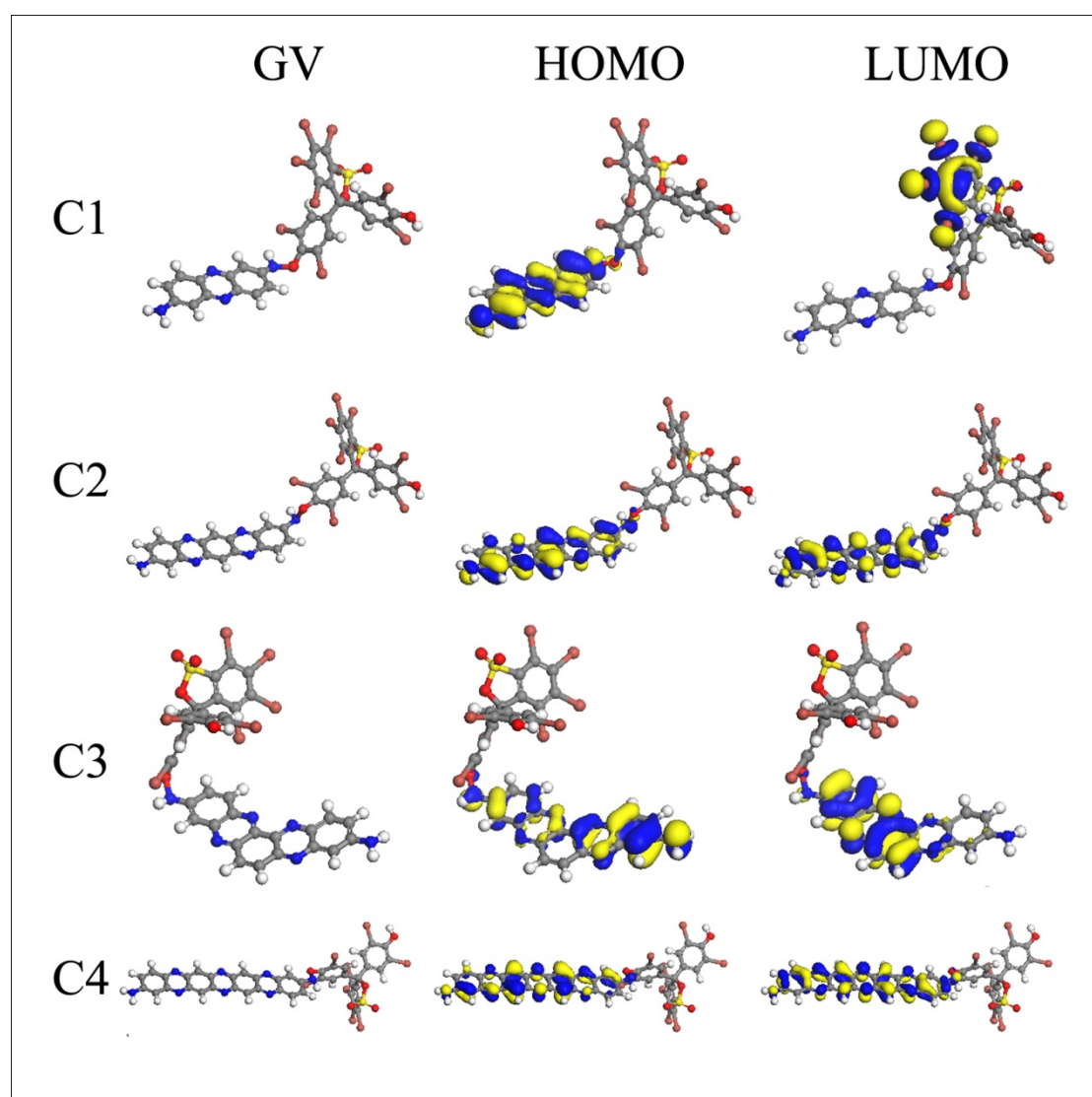
|      | BET (m <sup>2</sup> /g) | Pore volume<br>(mm <sup>3</sup> /g) | Pore size (nm) |
|------|-------------------------|-------------------------------------|----------------|
| PdAP | 25.58                   | 77.27                               | 12.06          |

**Table S2** The total energy ( $E_0$ , Ha), band gap ( $BG$ , eV) and the energy (eV) of HOMO-1, HOMO, LUMO, LUMO+1 of PANI (A1) and PpPD (A2-5).

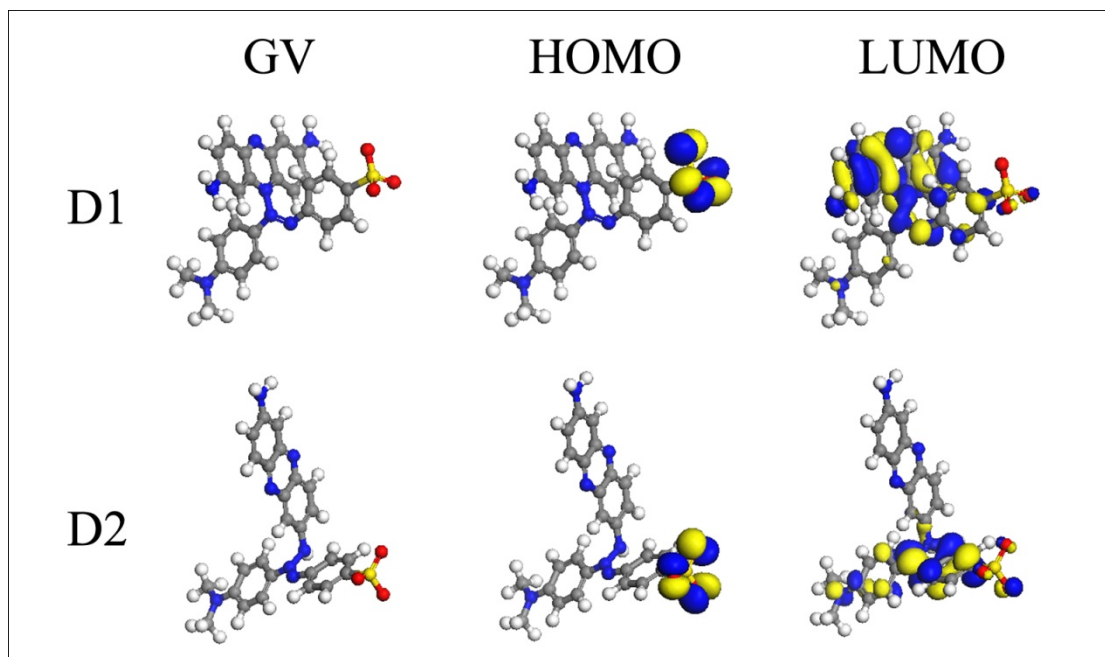
| Molecule | total energy<br>( $E_0$ , Ha) | Band<br>Gap<br>( $BG$ , eV) | HOMO-1<br>(eV) | HOMO<br>(eV) | LUMO<br>(eV) | LUMO+1<br>(eV) |
|----------|-------------------------------|-----------------------------|----------------|--------------|--------------|----------------|
| A1       | -1144.301492                  | 1.18745                     | -5.08965       | -5.08965     | -3.90220     | -3.90220       |
| A2       | -682.332765                   | 1.85440                     | -5.29433       | -4.63269     | -2.77828     | -1.05109       |
| A3       | -1021.688234                  | 1.08584                     | -5.43265       | -4.68689     | -3.60105     | -2.38560       |
| A4       | -1021.701963                  | 2.05174                     | -5.38037       | -4.96839     | -2.91665     | -2.83377       |
| A5       | -1361.038991                  | 0.58921                     | -5.32628       | -4.68491     | -4.09570     | -3.13796       |



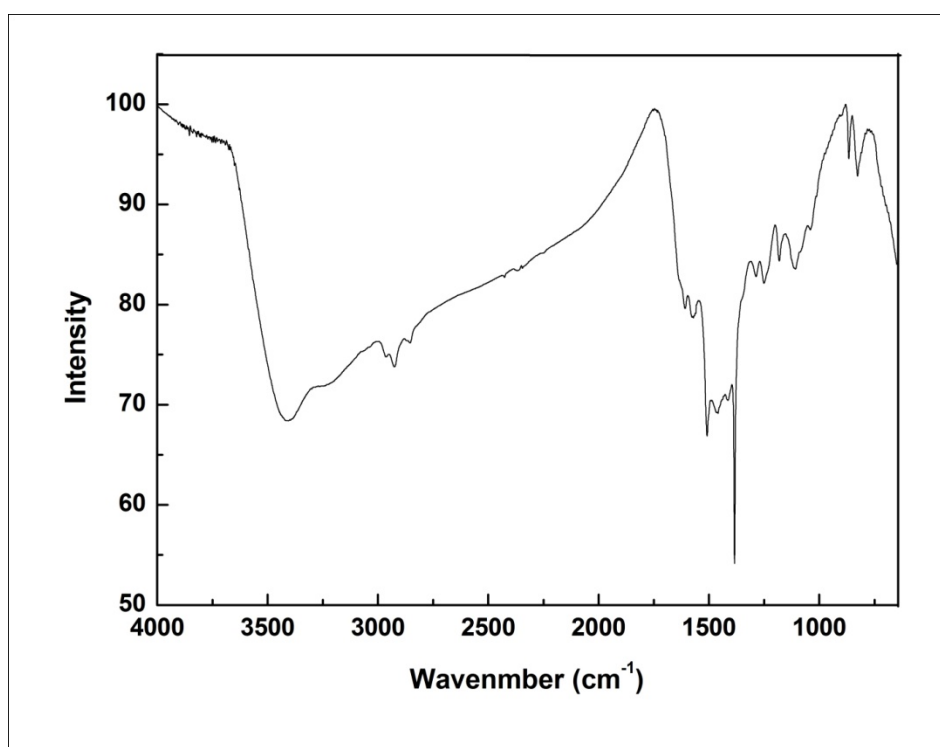
**Fig. S1** Ball-and-stick model and theoretical electron distribution of the HOMO–LUMO energy states of B1-4.



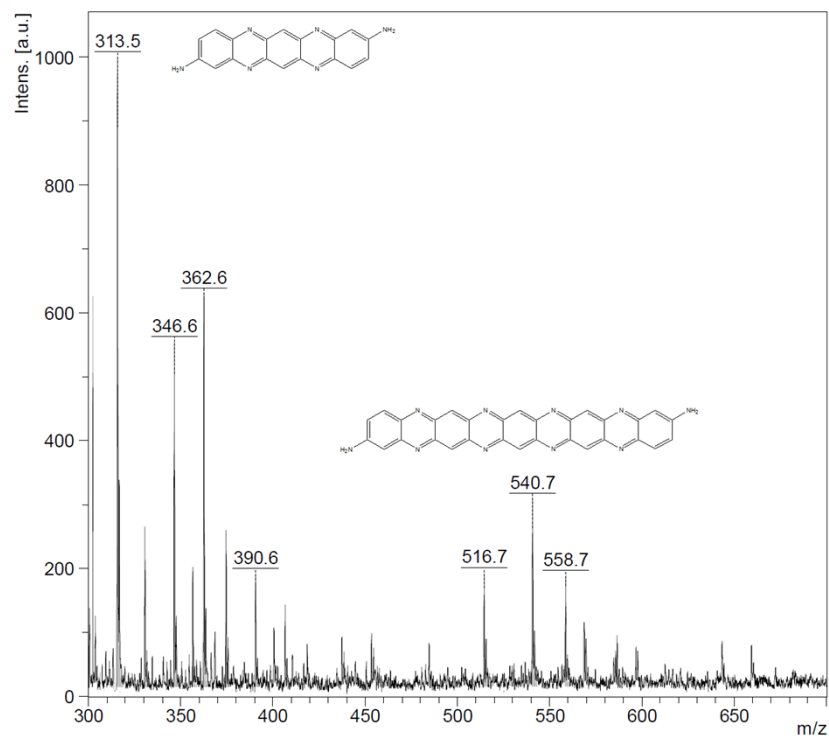
**Fig. S2** Ball-and-stick model and theoretical electron distribution of the HOMO-LUMO energy states of C1-4.



**Fig. S3** Ball-and-stick model and theoretical electron distribution of the HOMO–LUMO energy states of D1 and D2.



**Fig. S4** FT-IR spectra of  $\text{Fe}_3\text{O}_4$ -PpPD composite.



| m/z     | S/N  | Res. | Intens. |
|---------|------|------|---------|
| 302.528 | 21.4 | 844  | 576.75  |
| 313.547 | 34.3 | 679  | 887.93  |
| 319.484 | 7.6  | 1178 | 210.25  |
| 330.567 | 8.7  | 883  | 238.63  |
| 346.560 | 19.2 | 928  | 502.24  |
| 356.598 | 6.9  | 1018 | 188.60  |
| 362.563 | 23.4 | 1115 | 600.47  |
| 374.601 | 9.4  | 1029 | 245.48  |
| 390.611 | 6.7  | 960  | 178.09  |
| 516.720 | 7.0  | 1206 | 166.55  |
| 540.748 | 12.0 | 1195 | 260.72  |
| 558.654 | 7.6  | 1392 | 169.41  |

#### Target

Position C4

#### Laser

Laser beam attenuation 98.611  
Laser repetition rate 60 Hz  
Number of shots 100

#### Spectrometer

positive voltage polarity POS  
PIE delay 0 ns  
Ion source voltage 1 19 kV  
Ion source voltage 2 16.85 kV  
Lens voltage 5.8 kV  
Linear detector voltage 2.667 kV  
Deflection on false  
Deflection mass 0 Da  
SampleRate 0.5 ns

#### Instrument

Instrument type microflex  
Serial instrument number 269944.00022  
Name of computer FLEX-PC  
Operator ID or name BDAL@US  
flexControl version flexControl 3.3.99.0  
flexAnalysis version

Date of Acquisition 2011-12-23T03:46:25.531-05:00  
Acquisition method D:\Methods\flexControlMethods\LP\_SmallMolecules-0-700-110711.par  
Processing method  
File Name D:\Data\luhua\maldi-tof-luh-20111223-w4\0\_C4\1

|              |             |
|--------------|-------------|
| Performed by | Viewed by   |
| _____        | _____       |
| Date / Sign  | Date / Sign |

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Fig. S5 MALDI-TOF-MASS spectrum of PpPD

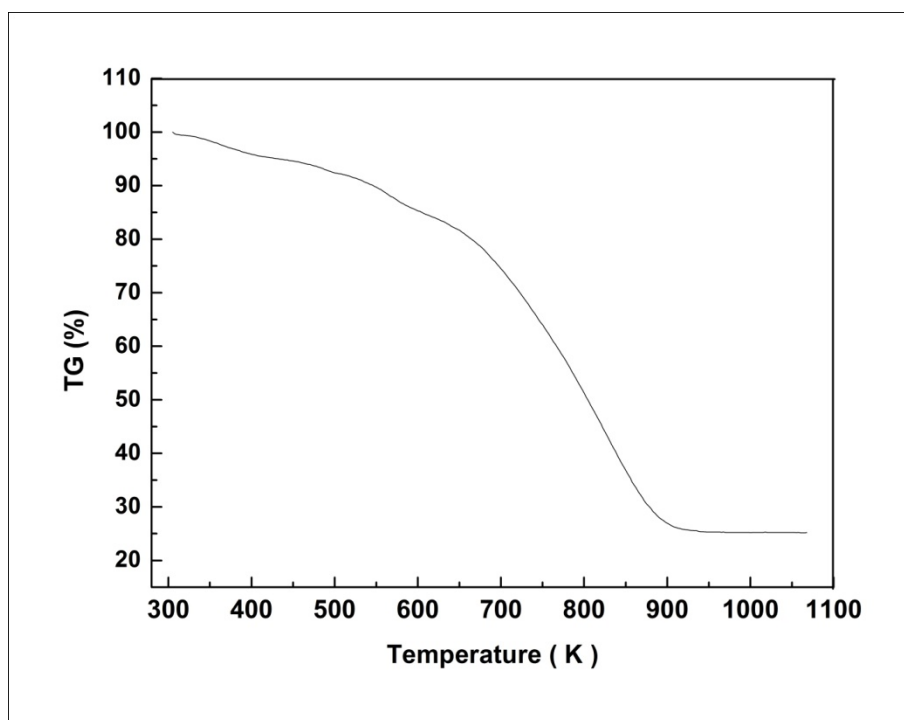


Fig. S6 pectrum of Fe<sub>3</sub>O<sub>4</sub>-PpPD composite.

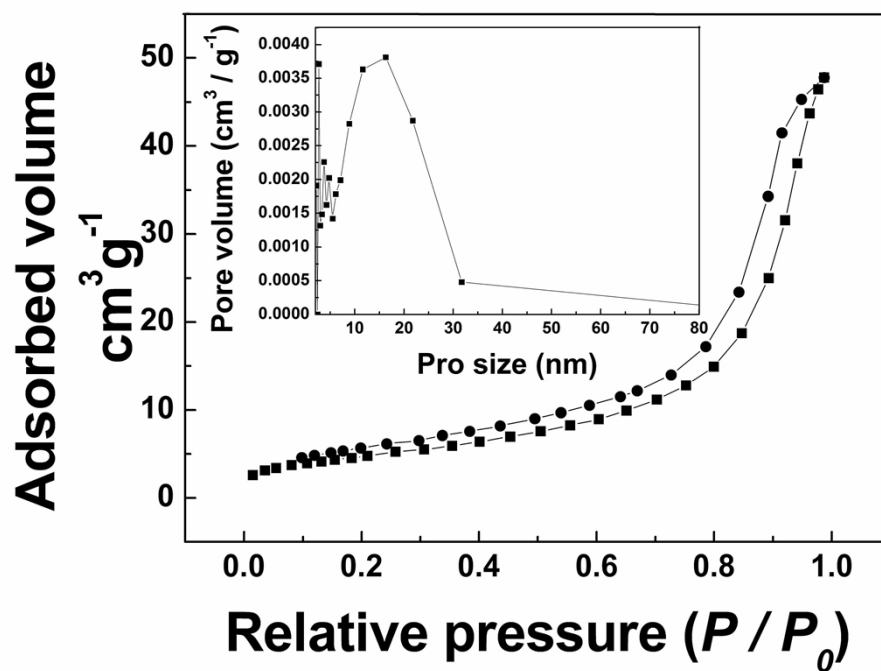
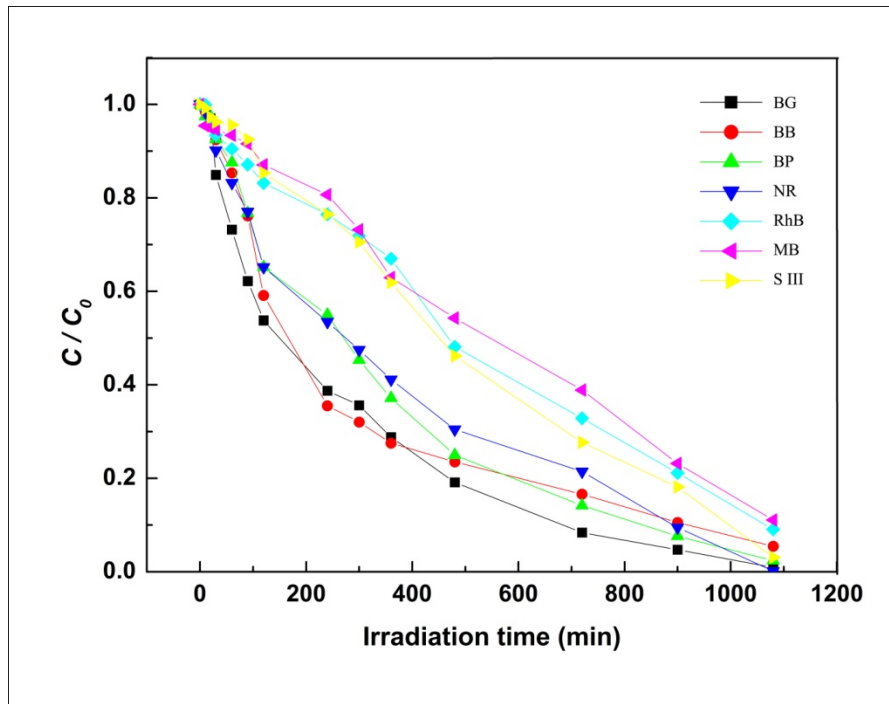
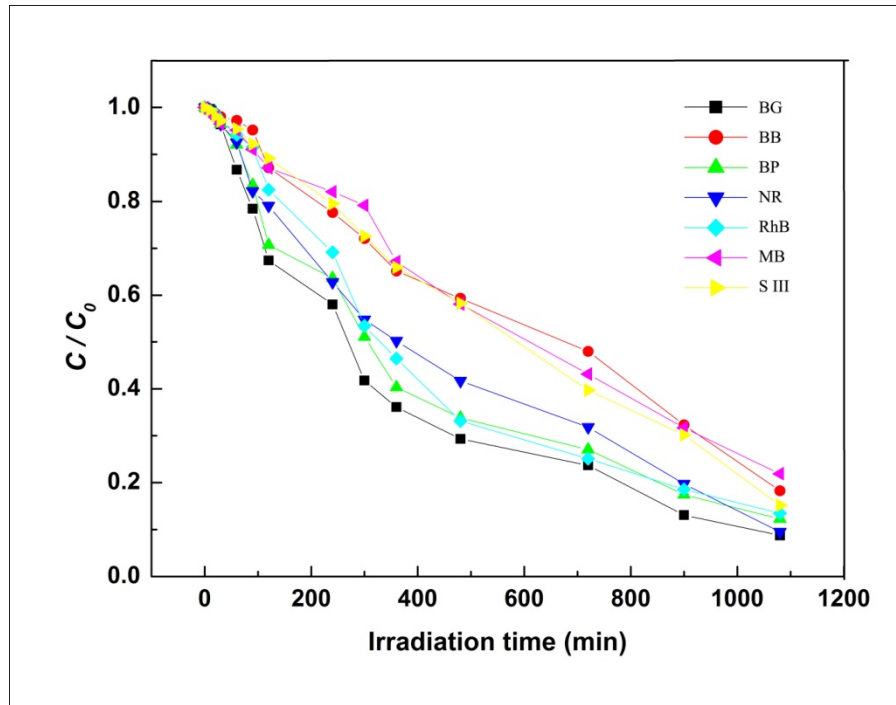


Fig. S7 N<sub>2</sub> adsorption-desorption isotherms of PpPD-Fe<sub>3</sub>O<sub>4</sub>, the inset shows the pore size distributions of PpPD-Fe<sub>3</sub>O<sub>4</sub>.

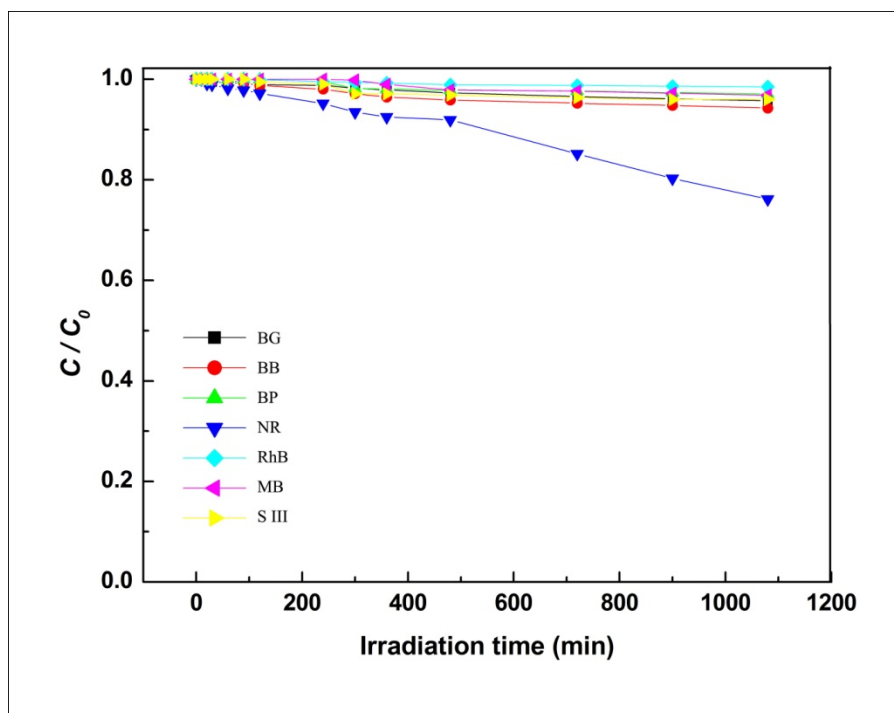


**Fig. S8** The photodegradation of dyes under UV ight irradiation.

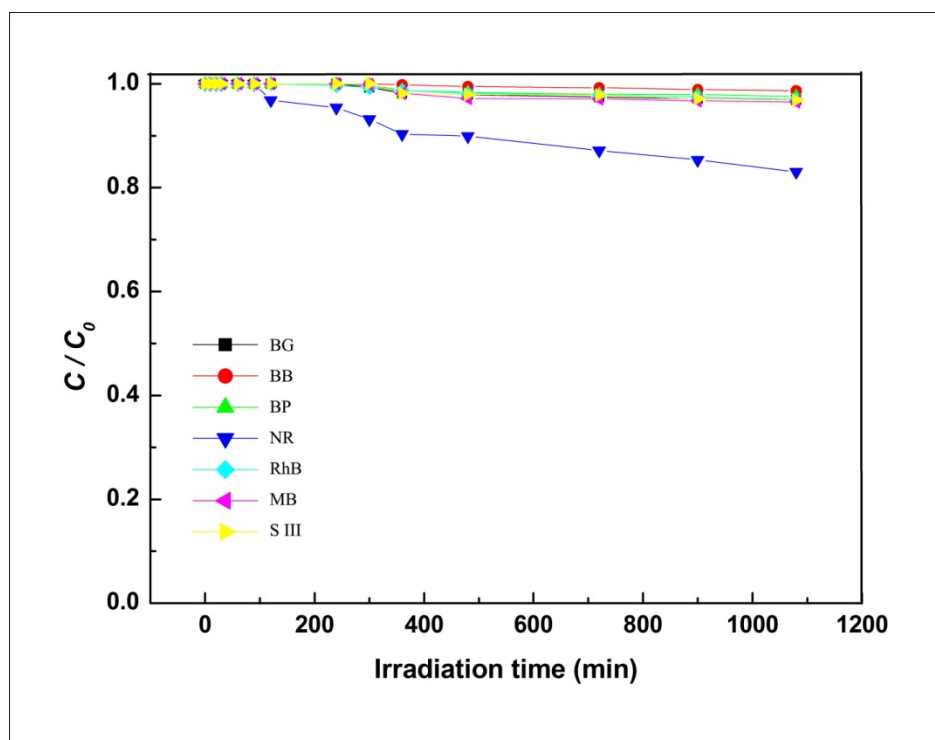


**Fig. S9** The photodegradation of dyes under isible light irradiation

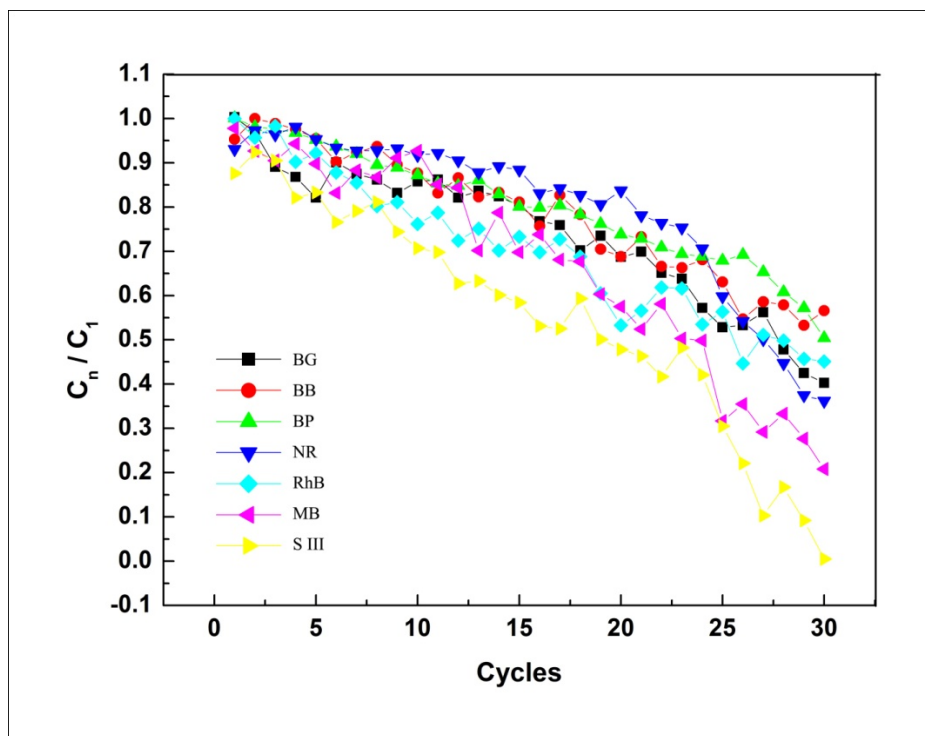




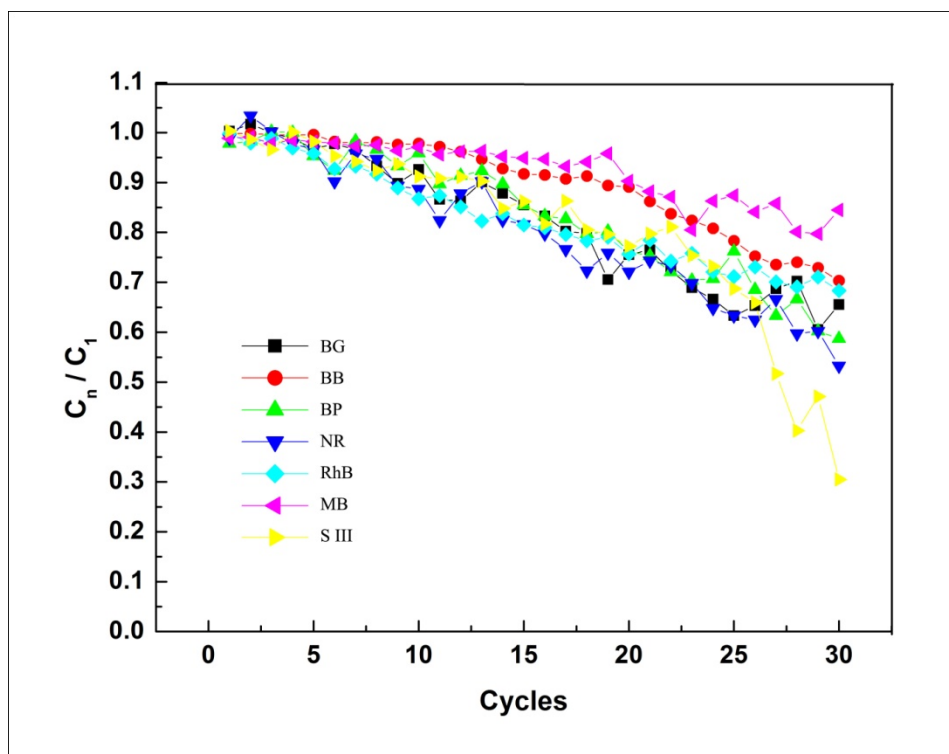
**Fig. S10** The photodegradation of dyes in the absent of photocatalysts under UV light irradiation



**Fig. S11** The photodegradation of dyes in the absent of photocatalysts under visible light irradiation



**Fig. S12** Photodegradation of dyes in the presence of  $\text{Fe}_3\text{O}_4$ -PpPD under UV light irradiation in 30 cycles



**Fig. S13** Photodegradation of dyes in the presence of  $\text{Fe}_3\text{O}_4$ -PpPD under visible light irradiation in 30 cycles