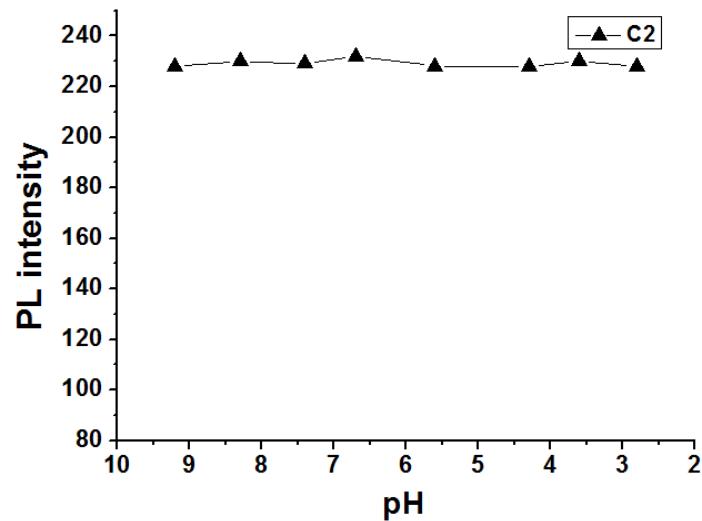


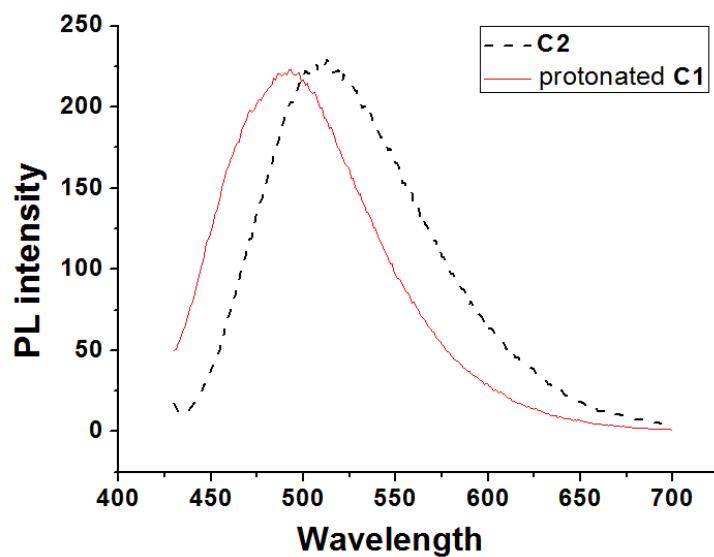
## ELECTRONIC SUPPLEMENTARY INFORMATION

Novel fluorescent sensors based on benzimidazo[2,1-a]benz[de]isoquinoline-7-one-12-carboxylic acid fo Cu<sup>2+</sup>

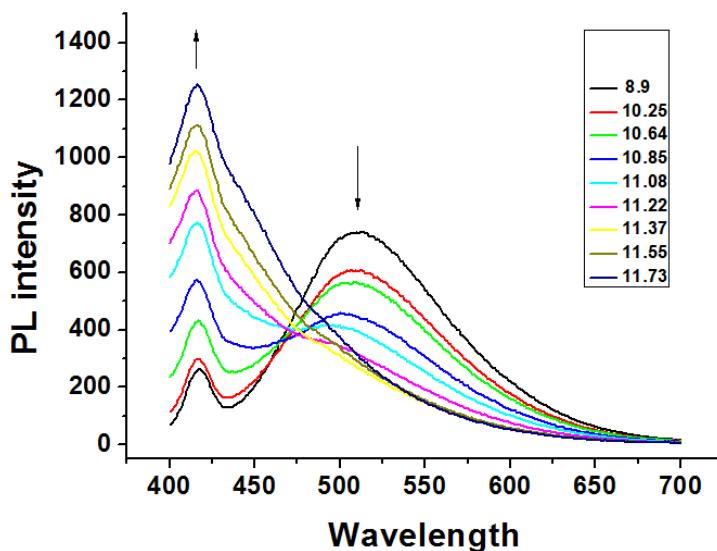
Zheng Liu, Yuhua Qi, Chaoxia Guo, Yingying Zhao, Xiaofeng Yang, Meishan Pei and  
Guangyou Zhang \*



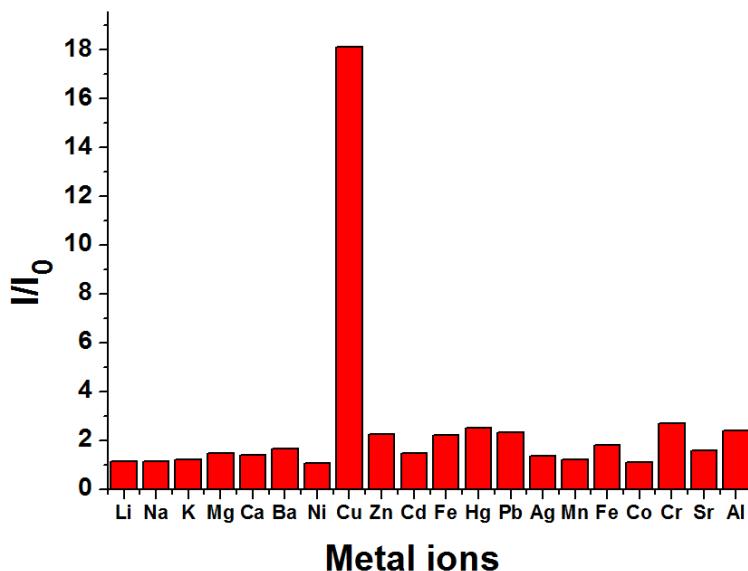
**Figure S1.** Emission intensity of **C2** versus the different pH values. Emission wavelength is at 510 nm.



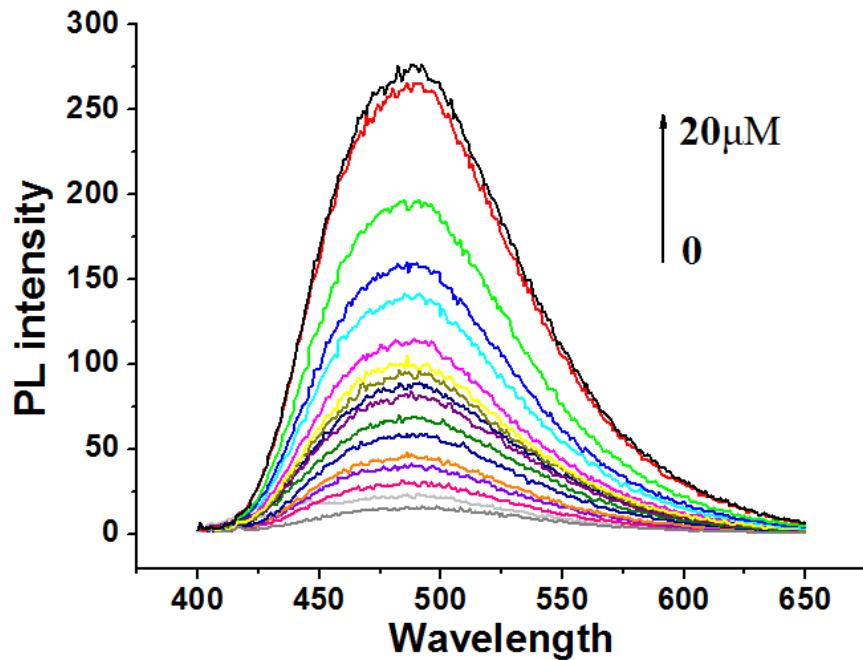
**Figure S2.** Emission spectra of protonated **C1** at pH 2.8 and that of **C2** in acetonitrile-H<sub>2</sub>O (9:1) solution.



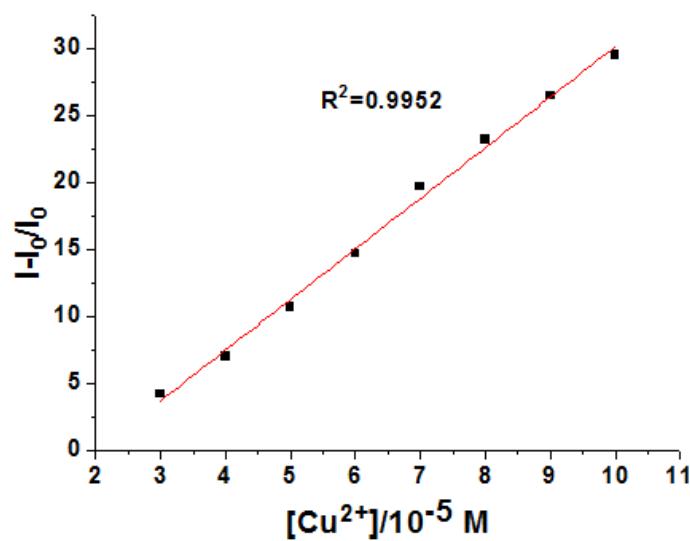
**Figure S3.** pH-dependence of the emission spectra of sensor **C2** ( $1 \times 10^{-5}$ M) in pure water. The insert indicates the change of pH increases from 8.9 to 11.73 with the titration of NaOH. Excitation is at 370 nm.



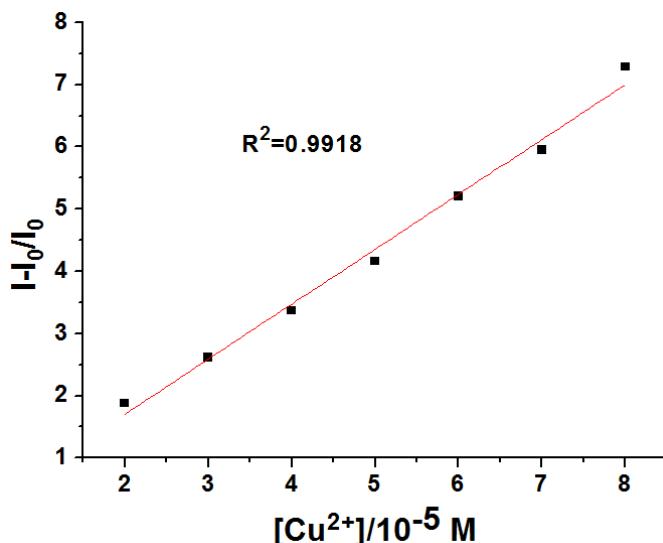
**Figure S4.** The relative PL intensity ( $I/I_0$ ) of **C2** ( $1 \times 10^{-5}$ M) in the presence of 20 equiv of  $\text{Cu}^{2+}$  ( $1 \times 10^{-5}$ M) and 40 equiv of various other metal ions ( $\text{Li}^+$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Pb}^{2+}$ ,  $\text{Ag}^+$ ,  $\text{Mn}^{2+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Co}^{2+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Sr}^{3+}$  and  $\text{Al}^{3+}$ ) ( $1 \times 10^{-5}$ M) in acetonitrile- $\text{H}_2\text{O}$  (9:1) containing HEPES (5mM, pH=7.4) at 25 °C, respectively. Excitation is at 370 nm, and emission is monitored at 494 nm.



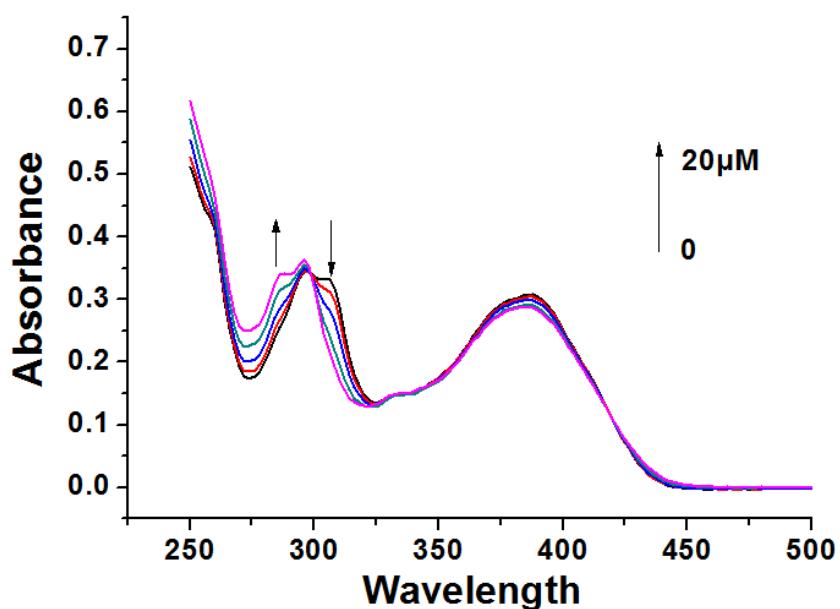
**Figure S5.** Changes in the PL intensity of **C2** ( $1 \times 10^{-5}$ M) in acetonitrile-H<sub>2</sub>O (9:1) containing HEPES (5mM, pH=7.4) upon titration with Cu<sup>2+</sup> ( $1 \times 10^{-5}$ M). Excitation is at 370 nm.



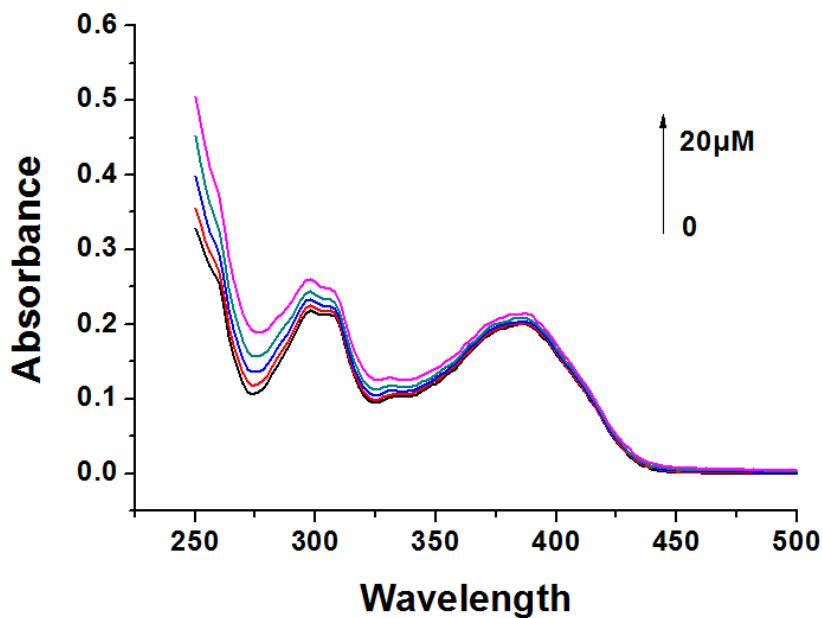
**Figure S6.** Change ratio of fluorescence of **C1** ( $1 \times 10^{-5}$ M) upon addition of Cu<sup>2+</sup> ( $1 \times 10^{-5}$ M) in acetonitrile-H<sub>2</sub>O (9:1) containing HEPES (5mM, pH=7.4) at 25 °C. Excitation is at 370 nm, and emission is monitored at 494 nm.



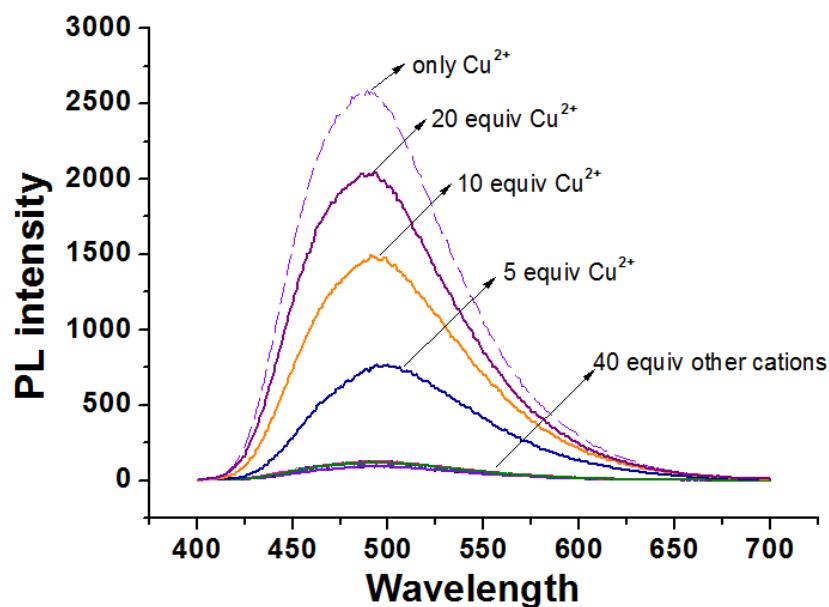
**Figure S7.** Change ratio of fluorescence of **C2** ( $1 \times 10^{-5}$ M) upon addition of  $\text{Cu}^{2+}$  ( $1 \times 10^{-5}$ M) in acetonitrile- $\text{H}_2\text{O}$  (9:1) containing HEPES (5mM, pH=7.4) at 25 °C. Excitation is at 370 nm, and emission is monitored at 494 nm.



**Figure S8.** Changes in the absorption spectra of **C1** ( $1 \times 10^{-5}$ M) in acetonitrile- $\text{H}_2\text{O}$  (9:1) containing HEPES (5mM, pH=7.4) upon titration with  $\text{Cu}^{2+}$  ( $1 \times 10^{-5}$ M). Excitation is at 370 nm.



**Figure S9.** Changes in the absorption spectra of **C2** ( $1\times 10^{-5}$ M) in acetonitrile-H<sub>2</sub>O (9:1) containing HEPES (5mM, pH=7.4) upon titration with Cu<sup>2+</sup> ( $1\times 10^{-5}$ M). Excitation is at 370 nm.

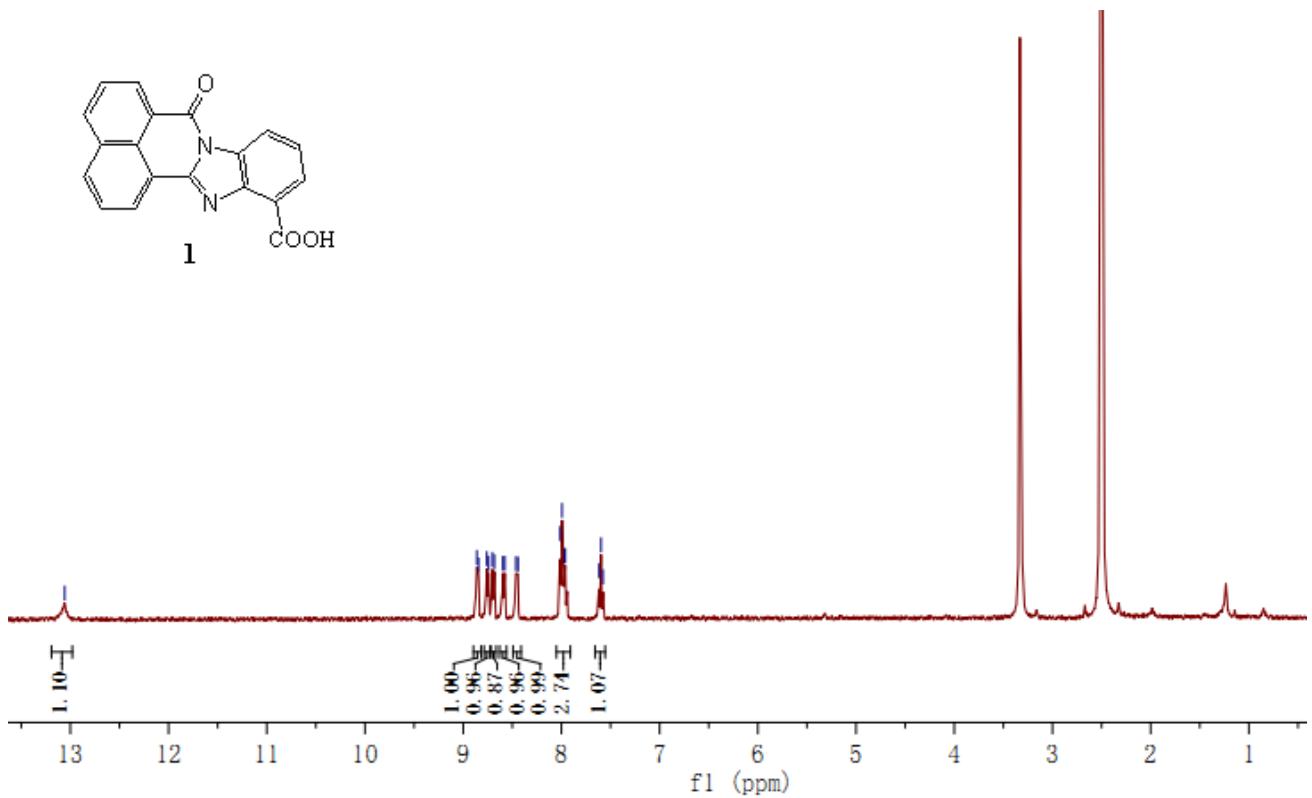


**Figure S10.** Competition experiments: fluorescence emission spectra of **C1** with addition of 40 equiv of (Li<sup>+</sup>, Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, Ba<sup>2+</sup>, Ni<sup>2+</sup>, Zn<sup>2+</sup>, Cd<sup>2+</sup>, Fe<sup>2+</sup>, Hg<sup>2+</sup>, Pb<sup>2+</sup>, Ag<sup>+</sup>, Mn<sup>2+</sup>, Fe<sup>3+</sup>, Co<sup>2+</sup>, Cr<sup>3+</sup>, Sr<sup>3+</sup> and Al<sup>3+</sup>) ( $1\times 10^{-5}$ M) and Cu<sup>2+</sup> (5equiv, 10 equiv, 20 equiv respectively) in acetonitrile-H<sub>2</sub>O

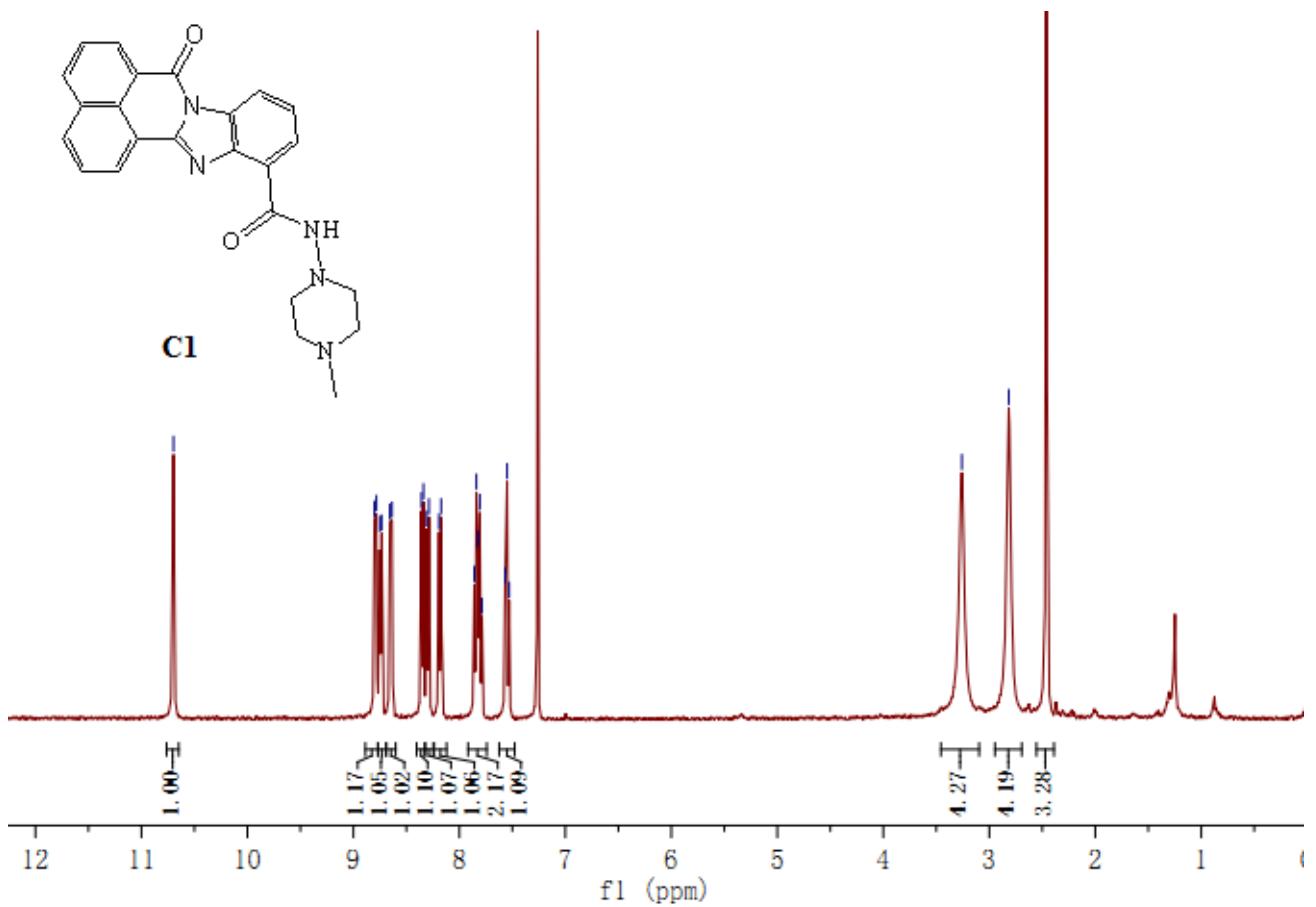
(9:1) containing HEPES (5mM, pH=7.4). Excitation is at 370 nm, and emission is monitored at 494 nm.

	orbital transition	E (eV)	$\lambda$ (nm)	<i>f</i>
<b>C2</b>	HOMO-LUMO	3.0027	412.91	0.2226
<b>C2+Cu<sup>2+</sup></b>	HOMO( $\alpha$ )-LUMO( $\alpha$ )	3.0816	402.33	0.3086
	HOMO( $\beta$ )-LUMO+1( $\beta$ )			

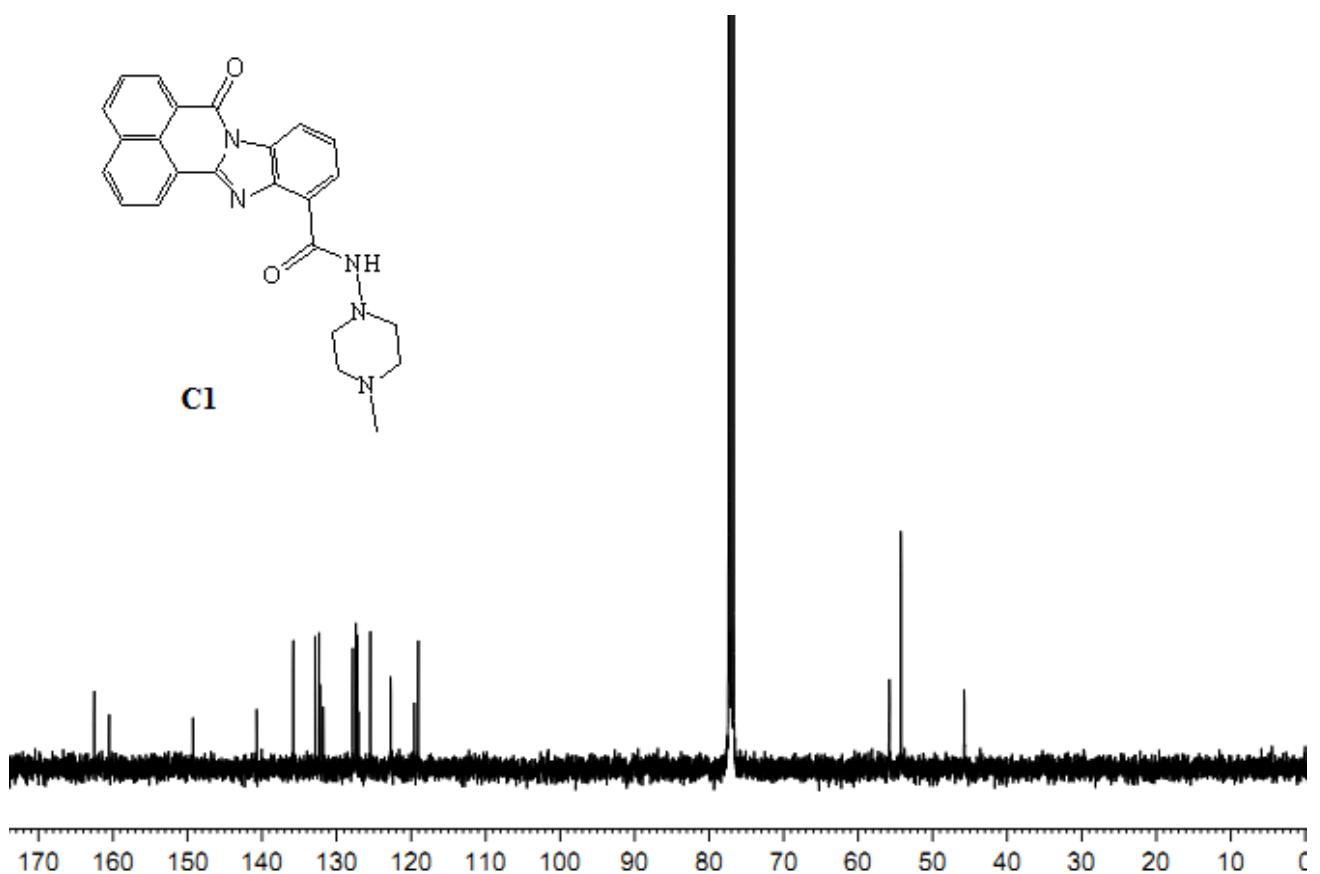
**Table S1.** Calculated excitation energy (E), wavelength ( $\lambda$ ), and oscillator strength (f) for the lowest singlet electronic transition of C2 and doublet electronic transitions of **C1+Cu<sup>2+</sup>** complex.



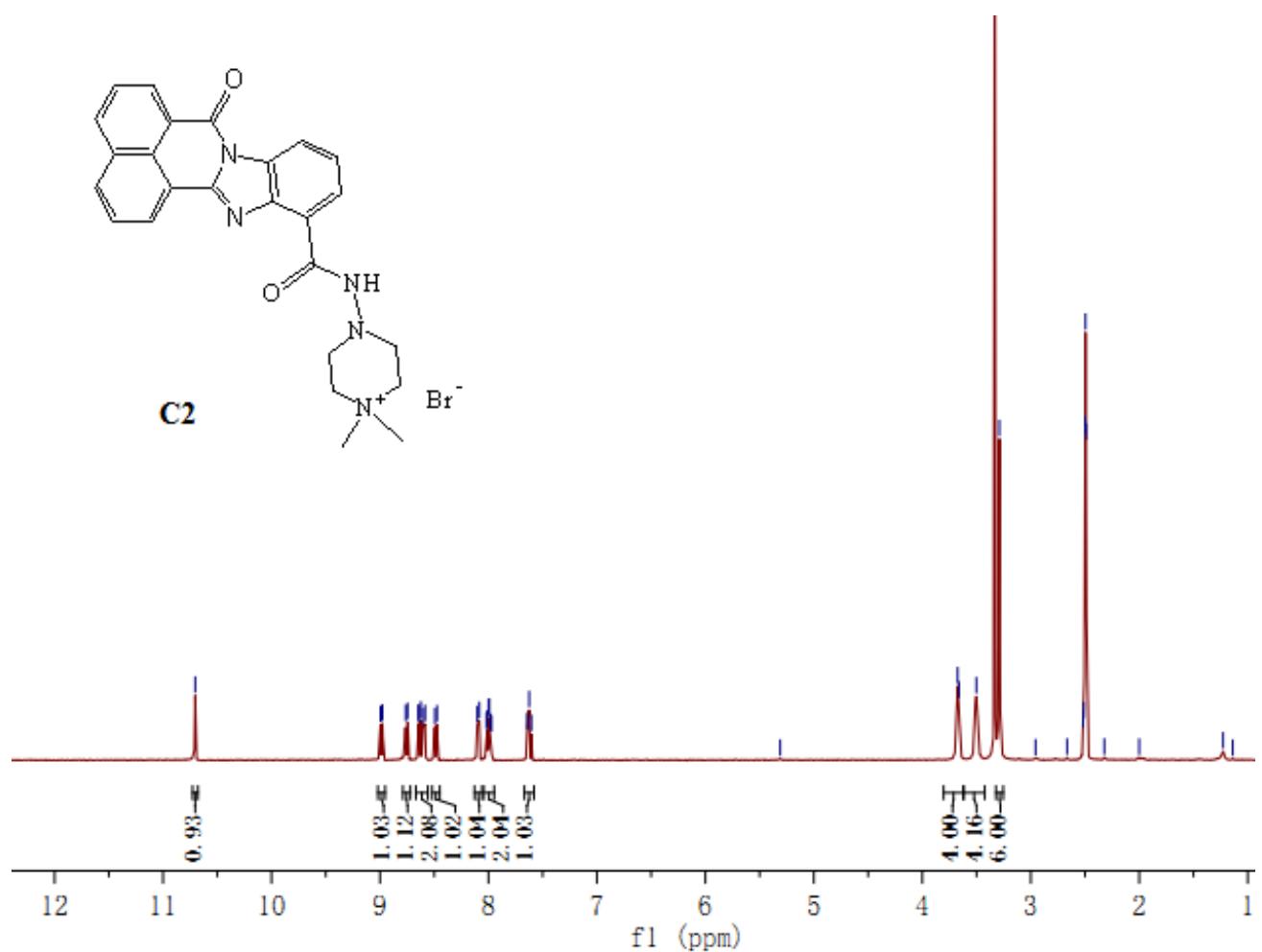
**Figure S11.** <sup>1</sup>H NMR spectrum of compound **1** (benzimidazo[2,1-a]benz[de]isoquinoline- 7-one-10-carboxylic acid) in CDCl<sub>3</sub>.



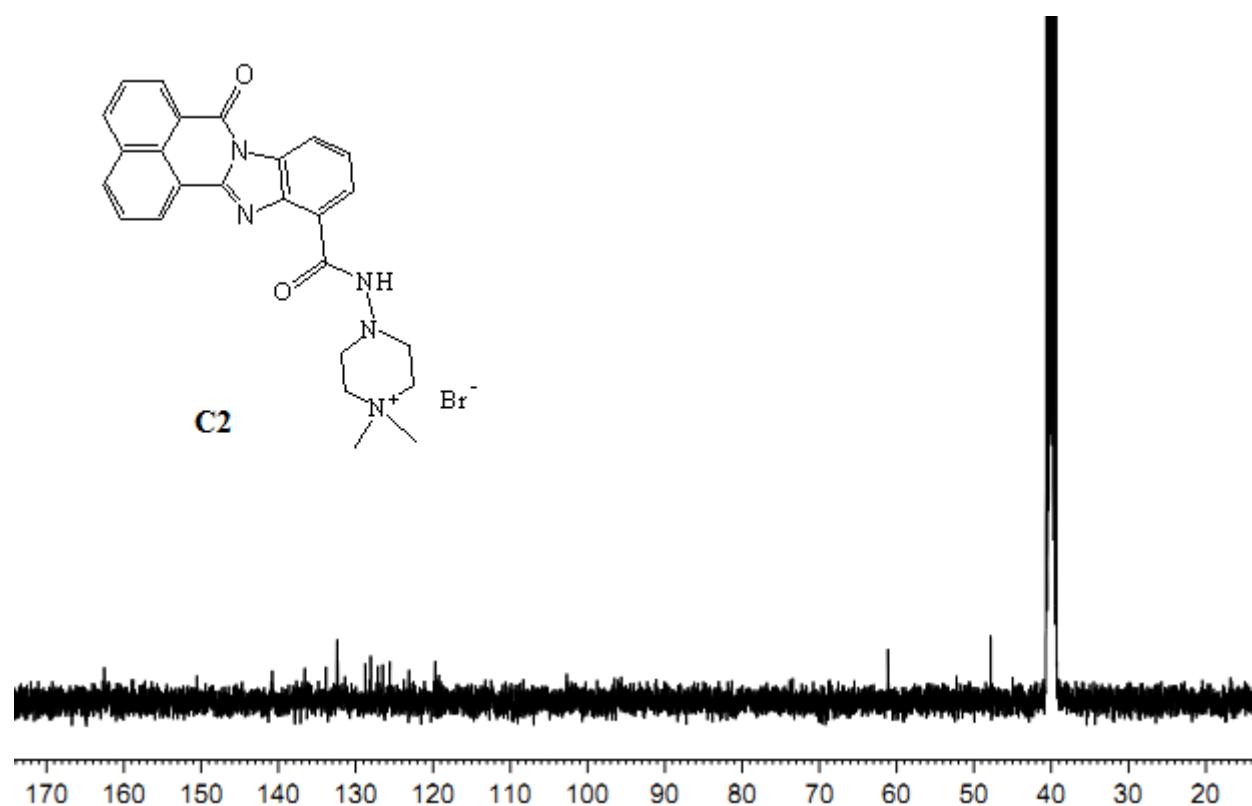
**Figure S12.** <sup>1</sup>H NMR spectra of C1 in CDCl<sub>3</sub>.



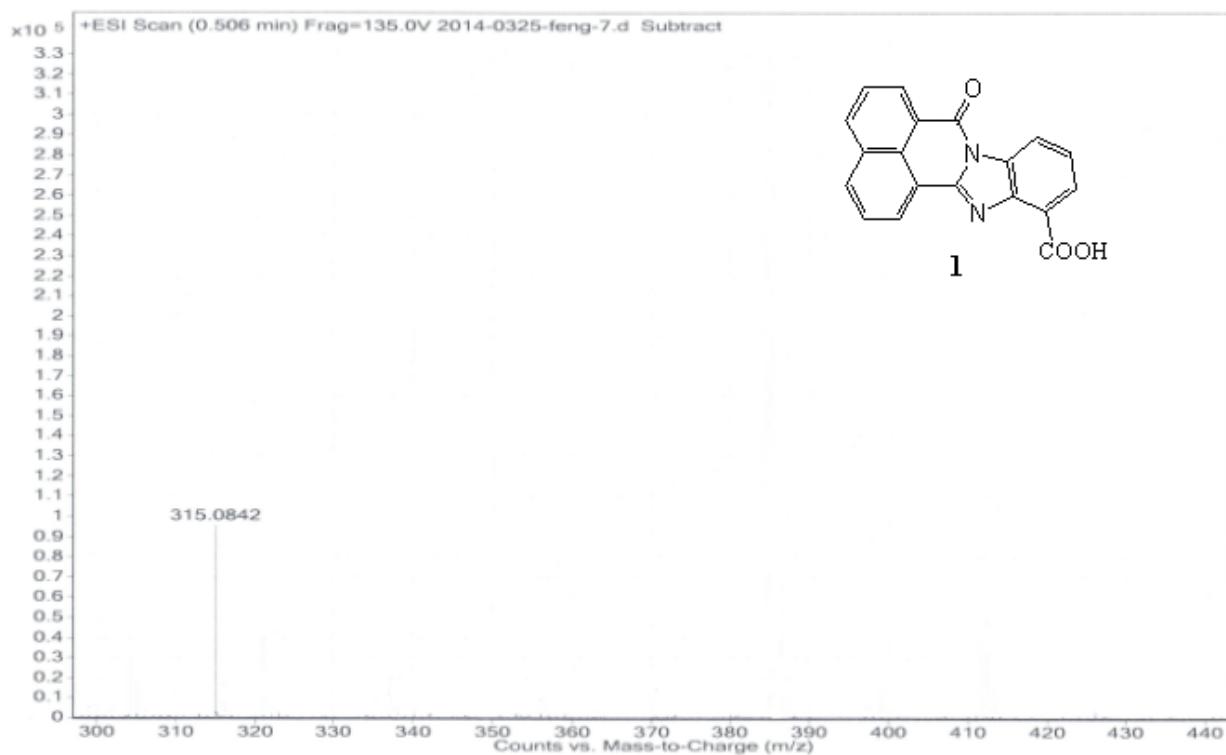
**Figure S13.**  $^{13}\text{C}$  NMR spectra of **C1** in  $\text{CDCl}_3$ .



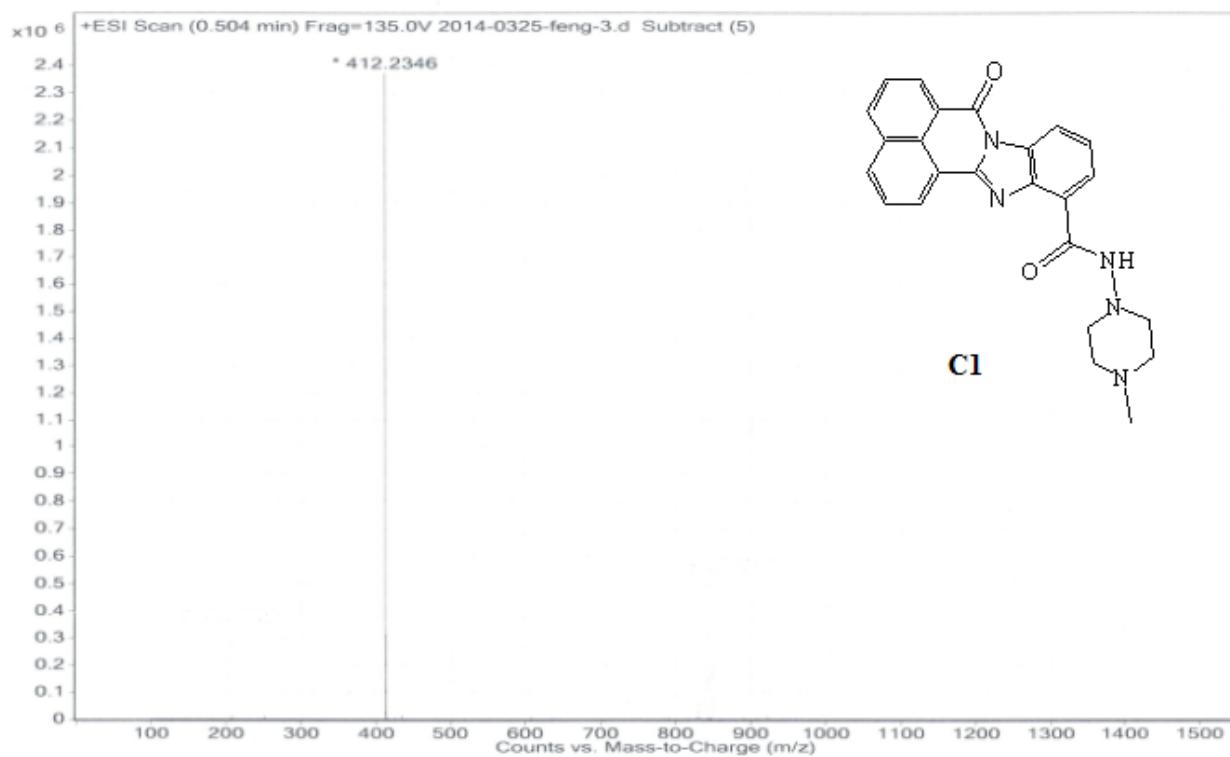
**Figure S14.**  $^1\text{H}$  NMR spectra of C2 in DMSO.



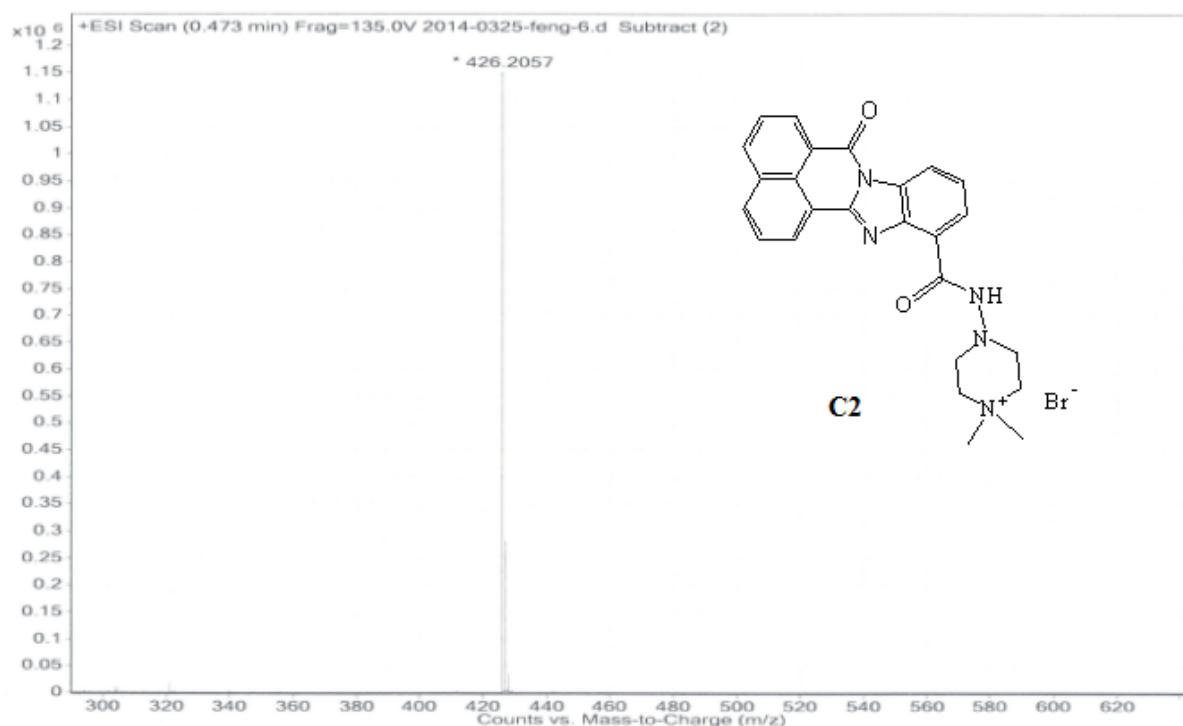
**Figure S15.**  $^{13}\text{C}$  NMR spectra of **C2** in DMSO.



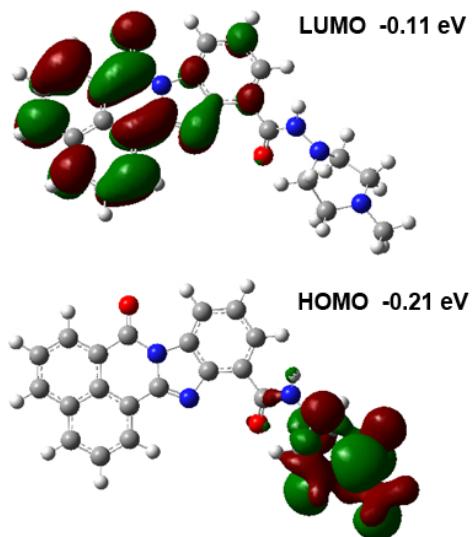
**Figure S16.** The mass spectrum of compound **1** (benzimidazo[2,1-a]benz[de]isoquinoline-7-one-10-carboxylic acid).



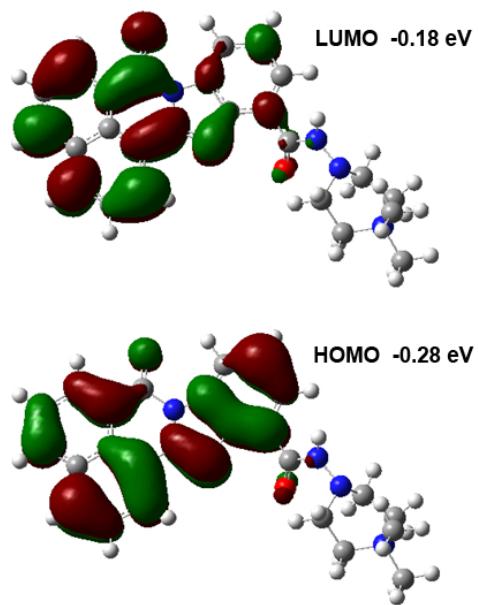
**Figure S17.** The mass spectrum of **C1**.



**Figure S18.** The mass spectrum of **C2**.

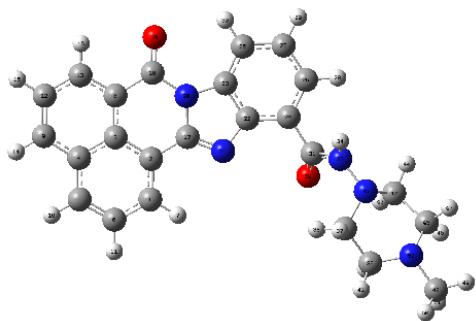


**Figure S19.** Frontier molecular orbitals (MO) of sensor **C1** calculated with Time-dependent density functional theory (TDDFT) at the B3LYP/6-31+G(d) level using Gaussian 03.



**Figure S20.** Frontier molecular orbitals (MO) of neutral sensor **C2** calculated with Time-dependent density functional theory (TDDFT) at the B3LYP/6-31+G(d) level using Gaussian 03.

**Table S2.** XYZ coordinate of the optimized structure of **C1**.



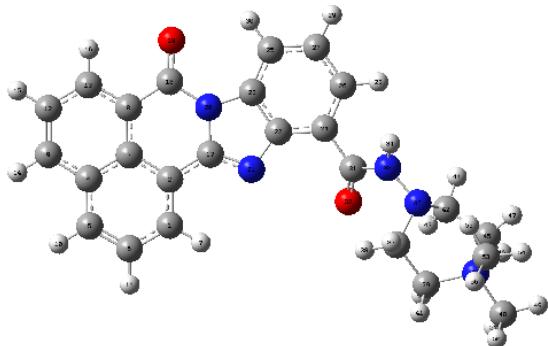
Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.282861	-2.565300	0.275230
2	6	0	2.734694	-1.260921	0.131758
3	6	0	4.127543	-0.993159	-0.009708
4	6	0	5.051685	-2.083752	-0.006606
5	6	0	4.552618	-3.406455	0.139340
6	6	0	3.197675	-3.638578	0.277851
7	1	0	1.217764	-2.742526	0.387765
8	6	0	4.619456	0.338296	-0.152330
9	6	0	6.437514	-1.804976	-0.148190
10	1	0	5.255973	-4.235751	0.142252
11	1	0	2.827799	-4.653633	0.391676
12	6	0	6.892194	-0.507699	-0.285560
13	6	0	5.980331	0.568133	-0.287269
14	1	0	7.140854	-2.634589	-0.146085
15	1	0	7.954922	-0.309860	-0.392189
16	1	0	6.327775	1.591155	-0.393045
17	6	0	1.823211	-0.131881	0.123216
18	6	0	3.696464	1.501420	-0.158665
19	8	0	4.050826	2.665232	-0.279543
20	7	0	2.339549	1.170034	-0.012575
21	7	0	0.519997	-0.154339	0.230420
22	6	0	0.119138	1.172814	0.175323
23	6	0	1.233850	2.029734	0.013564
24	6	0	-1.179423	1.709840	0.227729
25	6	0	1.115071	3.414774	-0.096077
26	6	0	-1.303662	3.097833	0.104948
27	6	0	-0.180829	3.931990	-0.047282
28	1	0	-2.292179	3.546848	0.157252

29	1	0	-0.325980	5.006016	-0.122937
30	1	0	1.985173	4.048092	-0.213799
31	6	0	-2.376528	0.825622	0.463498
32	8	0	-2.483236	0.092136	1.439604
33	7	0	-3.351244	0.947803	-0.498464
34	1	0	-3.153977	1.480460	-1.337187
35	7	0	-4.540763	0.211845	-0.527692
36	6	0	-4.323866	-1.239447	-0.729409
37	1	0	-4.202778	-1.409211	-1.806520
38	1	0	-3.419375	-1.581278	-0.211382
39	6	0	-5.534143	-2.029625	-0.208920
40	1	0	-5.425124	-2.234780	0.874743
41	1	0	-5.561114	-3.002739	-0.711738
42	6	0	-5.465664	0.522324	0.567736
43	1	0	-5.107862	0.127162	1.528981
44	1	0	-5.524158	1.610789	0.661175
45	6	0	-6.856809	-0.042714	0.225752
46	1	0	-7.408984	-0.172037	1.177211
47	1	0	-7.422540	0.669994	-0.388550
48	6	0	-7.966249	-2.113965	-0.366092
49	1	0	-8.839421	-1.527723	-0.676075
50	1	0	-7.906168	-2.997634	-1.011293
51	1	0	-8.138196	-2.454152	0.676372
52	7	0	-6.765527	-1.310970	-0.512566

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**Table S3.** XYZ coordinate of the optimized structure of **C2**.



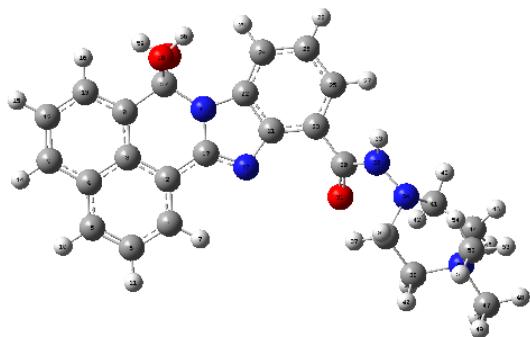
Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-2.682942	-2.607442	-0.186727
2	6	0	-3.103780	-1.285810	-0.085109
3	6	0	-4.494342	-0.976364	0.023269
4	6	0	-5.446778	-2.044788	0.029616
5	6	0	-4.978283	-3.384589	-0.072384
6	6	0	-3.625232	-3.657214	-0.178839
7	1	0	-1.622378	-2.812857	-0.275578
8	6	0	-4.957734	0.371802	0.122308
9	6	0	-6.830285	-1.729661	0.135802
10	1	0	-5.702348	-4.193355	-0.067569
11	1	0	-3.282071	-4.682525	-0.259354
12	6	0	-7.256935	-0.416498	0.230524
13	6	0	-6.318936	0.637135	0.223420
14	1	0	-7.551532	-2.541149	0.140246
15	1	0	-8.314127	-0.190645	0.309827
16	1	0	-6.642599	1.669346	0.295518
17	6	0	-2.167364	-0.186036	-0.089470
18	6	0	-4.012824	1.503217	0.118407
19	8	0	-4.337948	2.702629	0.206002
20	7	0	-2.654104	1.137731	0.006009
21	7	0	-0.844794	-0.229523	-0.175595
22	6	0	-0.420663	1.104477	-0.150267
23	6	0	-1.532122	1.981086	-0.023654
24	6	0	0.883482	1.632231	-0.193922
25	6	0	-1.398710	3.366522	0.058645
26	6	0	1.019107	3.029105	-0.101099
27	6	0	-0.096149	3.876064	0.017389

28	1	0	2.007225	3.473282	-0.169038
29	1	0	0.055422	4.948348	0.066555
30	1	0	-2.265686	4.005685	0.148150
31	6	0	2.064468	0.755537	-0.368371
32	8	0	2.126635	-0.231798	-1.135112
33	7	0	3.180389	1.107809	0.390315
34	1	0	3.136688	1.814671	1.109584
35	7	0	4.382535	0.378145	0.303450
36	6	0	4.248433	-1.023232	0.798773
37	1	0	4.247937	-1.002428	1.892548
38	1	0	3.311152	-1.474942	0.454242
39	6	0	5.399668	-1.875148	0.256655
40	1	0	5.196687	-2.222310	-0.759004
41	1	0	5.584634	-2.744671	0.891768
42	6	0	5.055053	0.472175	-1.011274
43	1	0	4.559143	-0.143014	-1.772256
44	1	0	5.002225	1.511028	-1.344082
45	6	0	6.531587	0.091792	-0.838125
46	1	0	6.960902	-0.234151	-1.788012
47	1	0	7.121648	0.921164	-0.444839
48	6	0	7.826821	-1.994117	-0.298386
49	1	0	8.732309	-1.402362	-0.439953
50	1	0	7.999240	-2.762497	0.456593
51	1	0	7.531493	-2.456196	-1.241146
52	7	0	6.712249	-1.078103	0.174960
53	6	0	7.077362	-0.520399	1.545299
54	1	0	8.064655	-0.061244	1.476956
55	1	0	6.331191	0.223419	1.818370
56	1	0	7.097370	-1.340740	2.264493

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**Table S4.** XYZ coordinate of the optimized structure of C2+OH<sup>-</sup>.



Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.353324	2.708566	-0.297819
2	6	0	2.842318	1.419042	-0.137740
3	6	0	4.247832	1.184975	-0.004473
4	6	0	5.141952	2.307034	-0.027626
5	6	0	4.602917	3.614684	-0.199524
6	6	0	3.239420	3.809377	-0.331677
7	1	0	1.284104	2.853782	-0.403046
8	6	0	4.778709	-0.124741	0.174418
9	6	0	6.537171	2.082288	0.134591
10	1	0	5.283899	4.460354	-0.221937
11	1	0	2.842371	4.810386	-0.462688
12	6	0	7.026313	0.804839	0.331894
13	6	0	6.141647	-0.299341	0.361097
14	1	0	7.211827	2.932943	0.113160
15	1	0	8.088696	0.640067	0.475153
16	1	0	6.539927	-1.288406	0.566174
17	6	0	1.961375	0.272933	-0.115440
18	8	0	4.094423	-1.976805	1.479771
19	7	0	2.485989	-1.017991	-0.015591
20	7	0	0.632754	0.272445	-0.220005
21	6	0	0.257929	-1.068836	-0.209465
22	6	0	1.405678	-1.903090	-0.086012
23	6	0	-1.025249	-1.645827	-0.277738
24	6	0	1.314953	-3.297220	-0.066631
25	6	0	-1.115728	-3.046443	-0.216321
26	6	0	0.033691	-3.855703	-0.126324
27	1	0	-2.087354	-3.524506	-0.297152

28	1	0	-0.076458	-4.934669	-0.126582
29	1	0	2.194855	-3.928733	-0.053064
30	6	0	-2.234406	-0.803633	-0.446737
31	8	0	-2.360446	0.118131	-1.284996
32	7	0	-3.294426	-1.112757	0.401429
33	1	0	-3.195390	-1.773258	1.160043
34	7	0	-4.509324	-0.397985	0.349364
35	6	0	-4.360626	1.029810	0.758537
36	1	0	-4.336408	1.073824	1.852148
37	1	0	-3.427614	1.457266	0.371907
38	6	0	-5.516035	1.855365	0.189337
39	1	0	-5.346879	2.108598	-0.859714
40	1	0	-5.656523	2.781257	0.752597
41	6	0	-5.270302	-0.582855	-0.906482
42	1	0	-4.828974	-0.025639	-1.742726
43	1	0	-5.239734	-1.643609	-1.166216
44	6	0	-6.730652	-0.186634	-0.661182
45	1	0	-7.233763	0.034671	-1.605316
46	1	0	-7.279376	-0.972011	-0.138420
47	6	0	-7.957508	1.976898	-0.291302
48	1	0	-8.882793	1.399960	-0.330566
49	1	0	-8.077508	2.827126	0.381968
50	1	0	-7.692248	2.325900	-1.290150
51	7	0	-6.846245	1.084705	0.232784
52	6	0	-7.172689	0.681509	1.664806
53	1	0	-8.165732	0.229610	1.676375
54	1	0	-6.428025	-0.038448	1.999799
55	1	0	-7.163349	1.574583	2.292106
56	1	0	3.639278	-2.844972	1.516712
57	6	0	3.882543	-1.348665	0.194672
58	8	0	4.178501	-2.315541	-0.844995
59	1	0	5.144656	-2.447281	-0.935834

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