

Electronic Supplementary Information

Optimization of the time and temperature of the microwave-assisted amination of the phenylene-PMO

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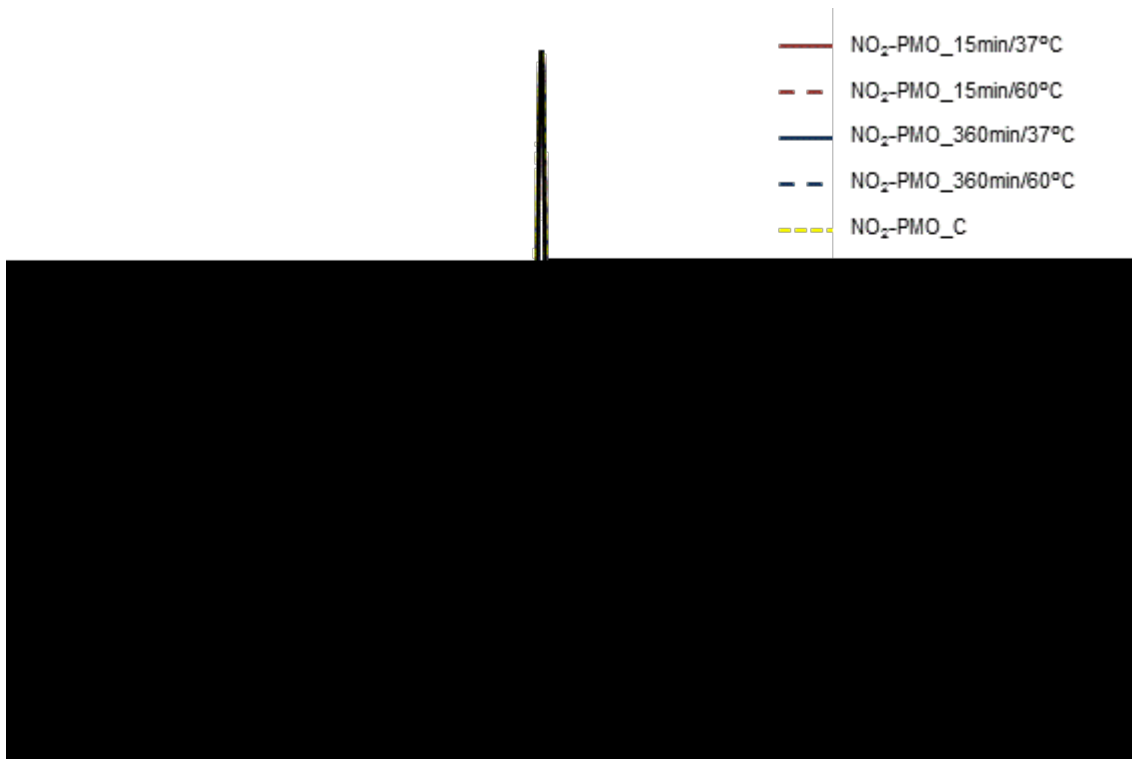
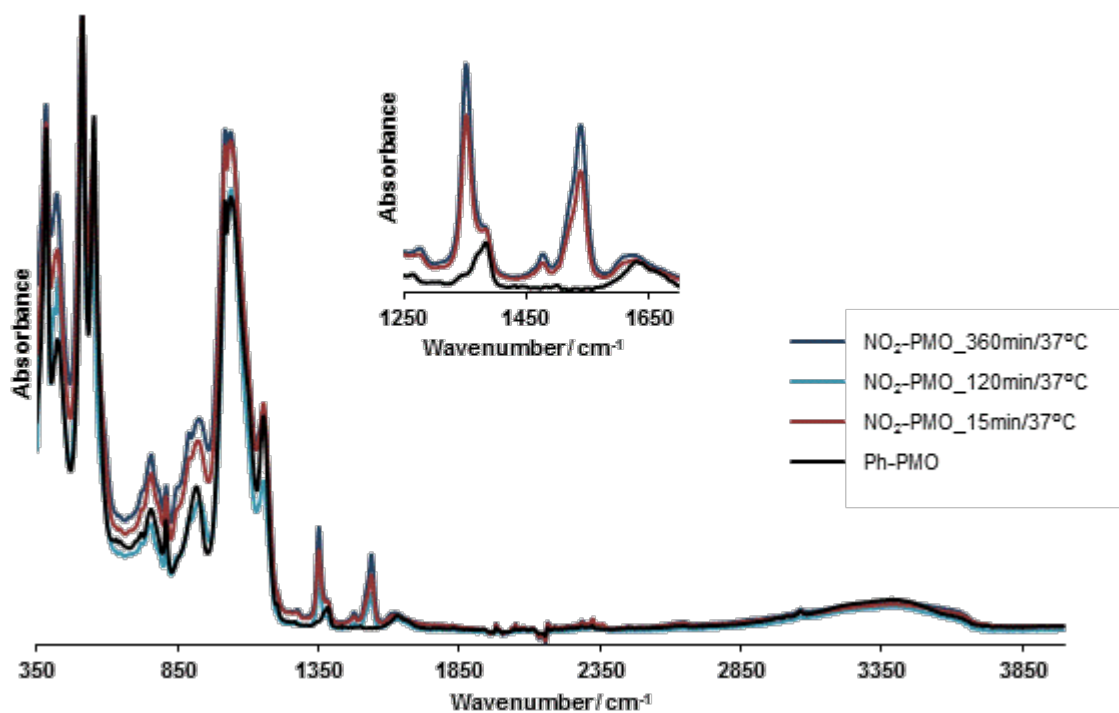
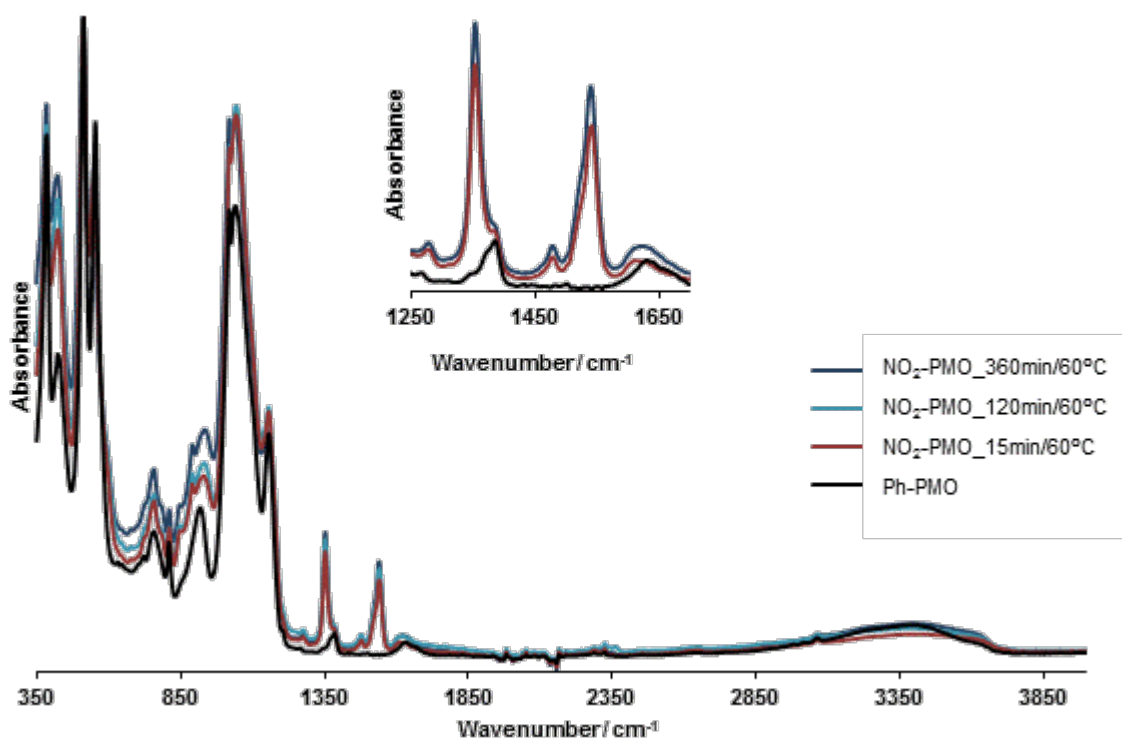


Figure S1. ^{13}C CP MAS NMR spectra for PMO and $\text{NO}_2\text{-PMO}_{x/y}$ samples obtained under different synthetic conditions. * denotes spinning sidebands.

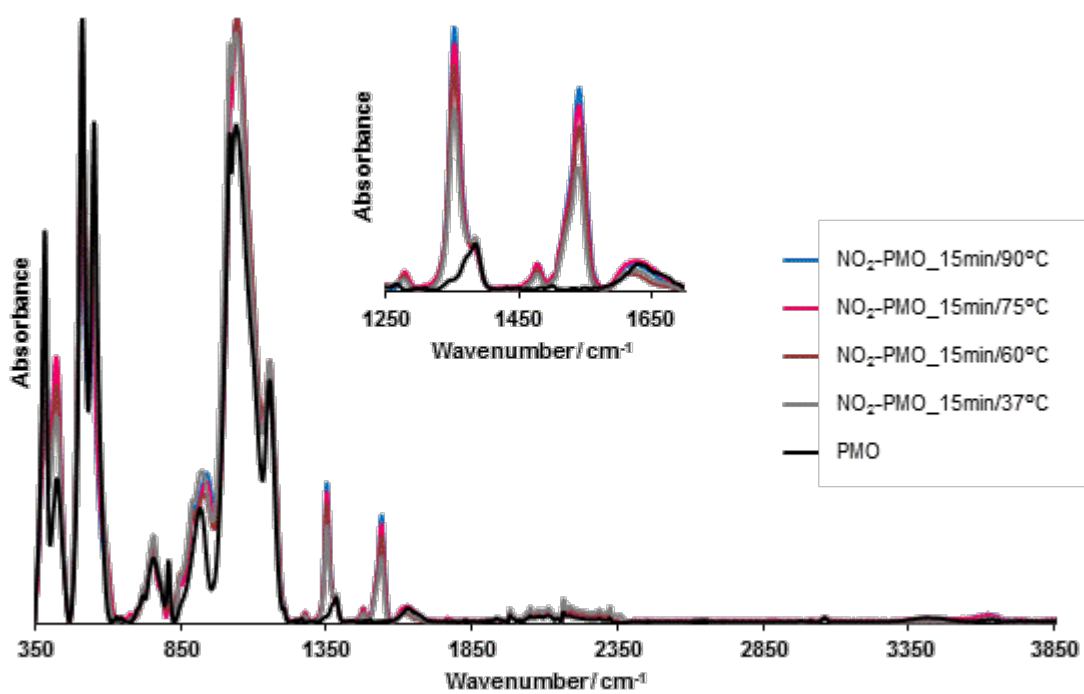
a)



b)



c)



d)

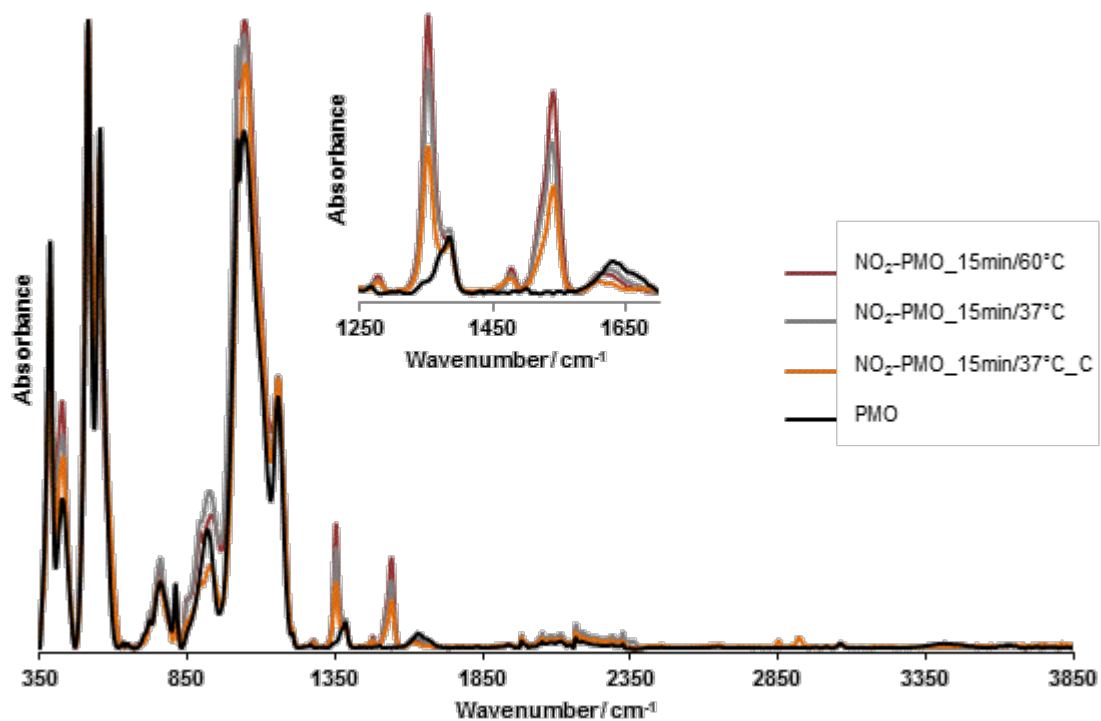


Figure S2. FTIR (ATR) spectra of PMO, NO₂-PMO_{x/y} at a) 37 °C and b) 60 °C for different reaction times, c) NO₂-PMO_{15min/y} at different temperatures and d) NO₂-PMO_{15min/y} at different temperatures using microwave and conventional heating. The insets display the region 1250 – 1700 cm^{-1} magnified.

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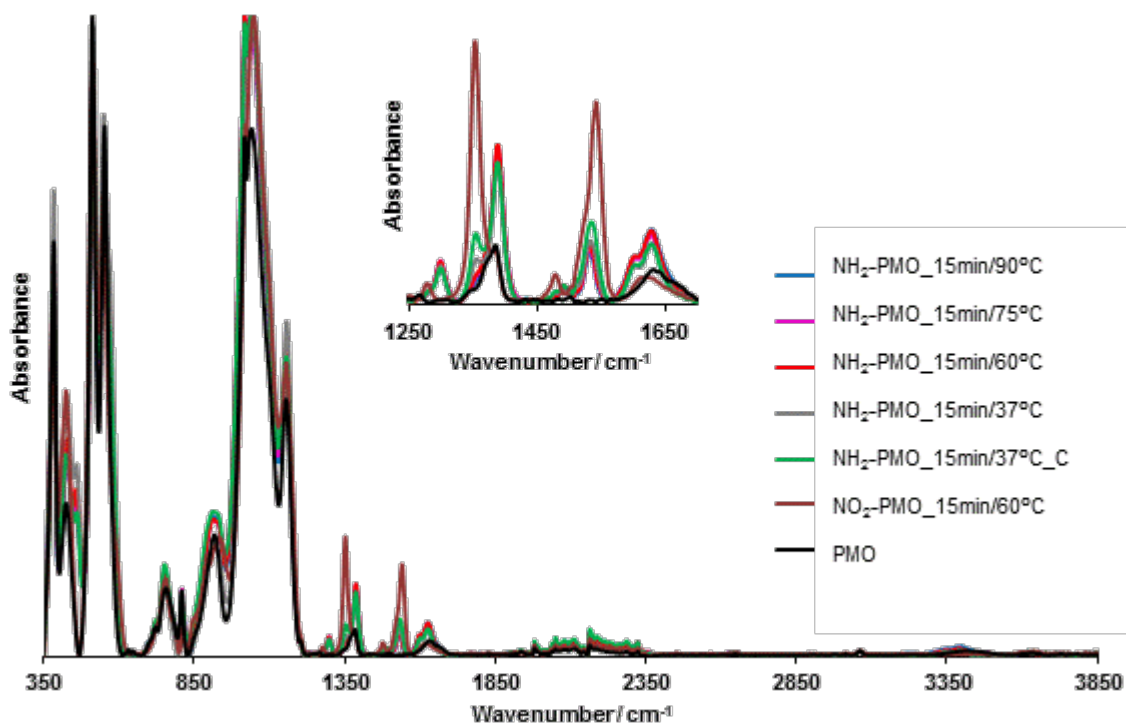


Figure S3. FTIR (ATR) spectra of PMO, NO₂-PMO_15min/60°C and NH₂-PMO_15min/y at different temperatures. The insets display the region 1250 – 1700 cm⁻¹ magnified.

Table S1. Textural parameters and physical properties of PMO and NO₂-PMOs_x/y.

Sample	d_{100} / nm	a / nm^a	$S_{\text{BET}} / \text{m}^2 \text{g}^{-1}$	$V_p / \text{cm}^3 \text{g}^{-1}$	d_p / nm^b	b / nm^c
PMO	4.55	5.25	920	0.66	3.58	1.67
NO ₂ -PMO_15min/37°C	4.70	5.42	788	0.62	3.58	1.84
NO ₂ -PMO_120min/37°C	4.50	5.20	757	0.58	3.51	1.69
NO ₂ -PMO_360min/37°C	4.65	5.36	626	0.49	3.51	1.85
NO ₂ -PMO_15min/60°C	4.70	5.42	698	0.57	3.54	1.88
NO ₂ -PMO_120min/60°C	4.55	5.25	743	0.57	3.51	1.74
NO ₂ -PMO_360min/60°C	4.55	5.25	808	0.63	3.51	1.74
NO ₂ -PMO_15min/75°C	4.82	5.57	730	0.61	3.69	1.88
NO ₂ -PMO_15min/90°C	4.82	5.57	776	0.65	3.55	2.02

^aUnit cell parameter calculated as $(2d_{100}/\sqrt{3})$. ^bPore width obtained from the BJH method with the corrected Kelvin equation, i.e. KJS–BJH method at the maximum of pore size distribution calculated on the basis of adsorption data. ^cPore wall thickness calculated as $(2d_{100}/\sqrt{3} - d_p)$, where the first term is the unit cell parameter.

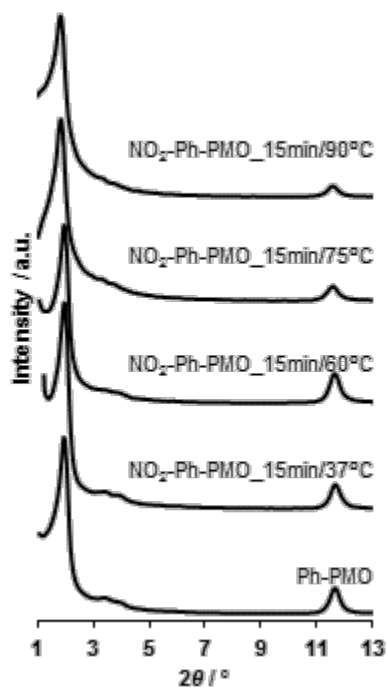


Figure S4. X-ray diffraction patterns of Ph-PMO and NO₂-Ph-PMO_{15min} synthesized at different reaction temperatures during 15 minutes.

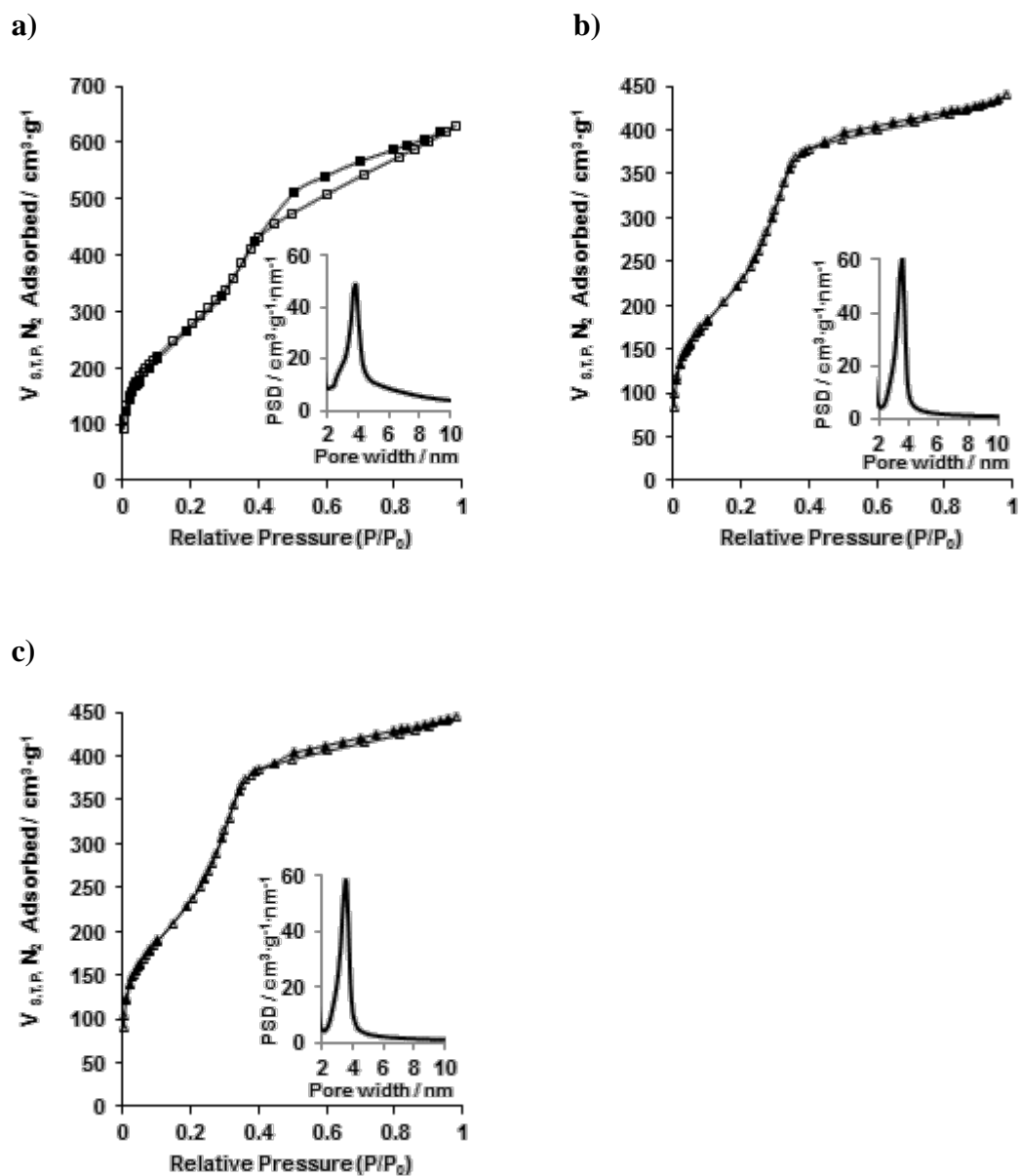


Figure S5. -196 °C nitrogen type IV isotherms of: a) PMO; b) NO_2 -PMO_15min/37°C and c) NO_2 -PMO_360min/60°C. Empty/filled symbols correspond to adsorption/desorption. The insets display the characteristic narrow PSD curves of PMO materials.

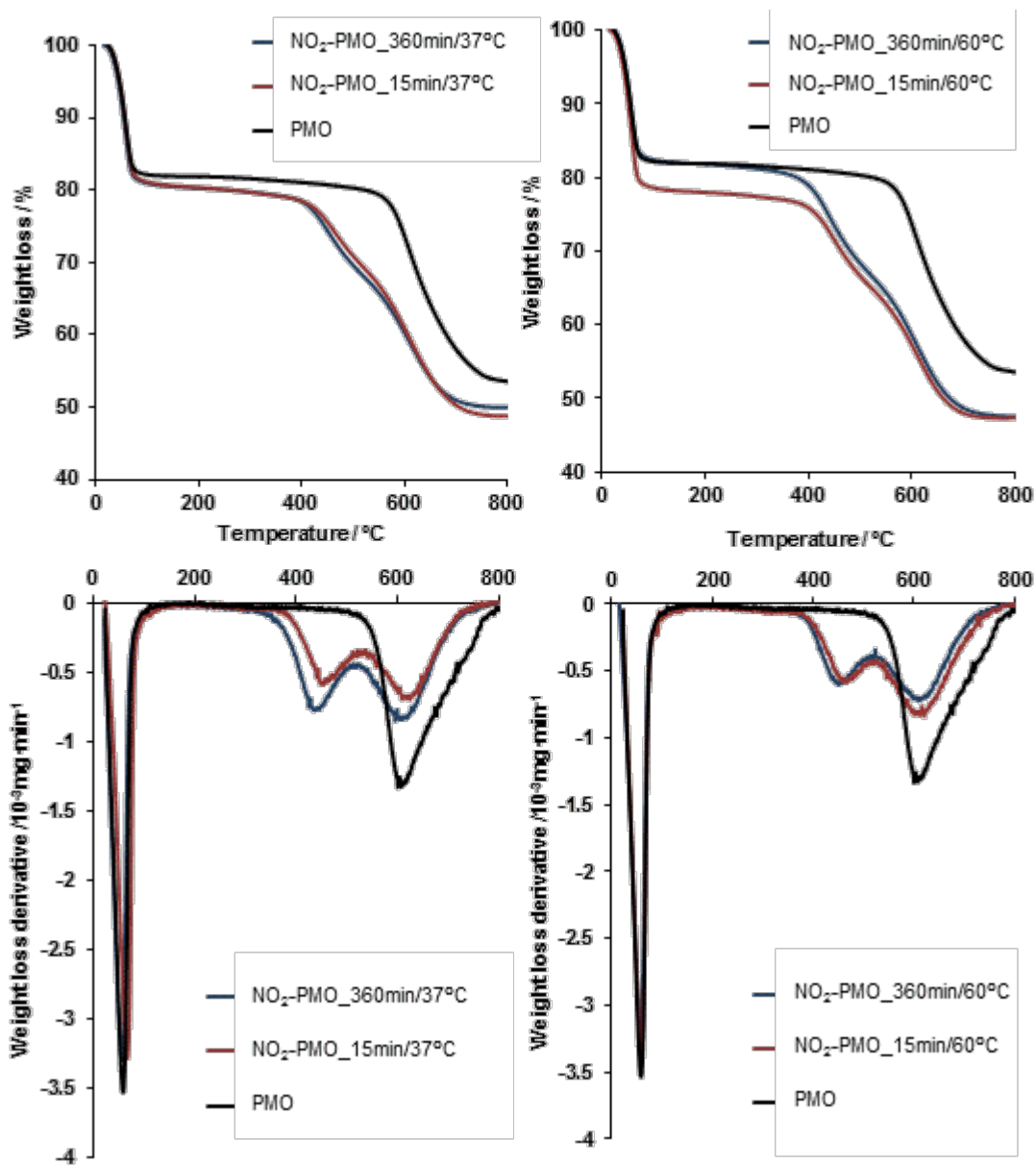


Figure S6. Weight loss curves (up) and weight loss derivatives (down) for PMO and NO₂-PMO.

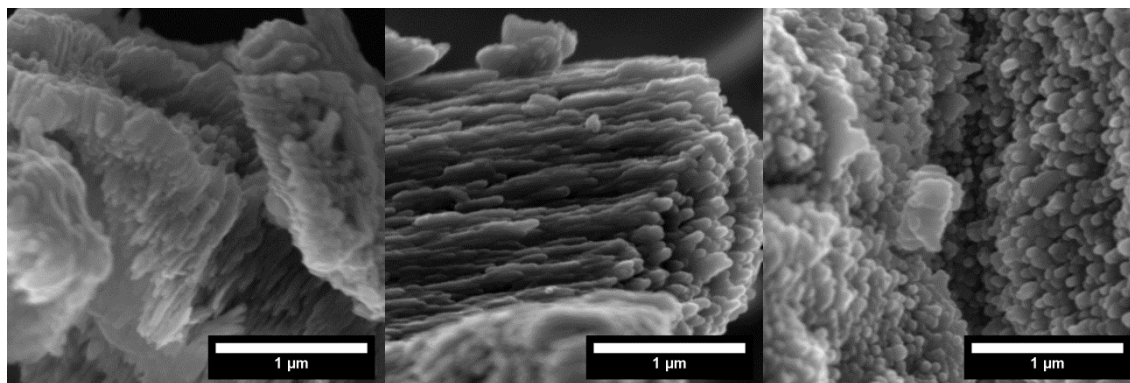


Figure S7. Scanning electron microscopy images for NO₂-PMO_C (left), NO₂-PMO_15min/37°C (middle) and NO₂-PMO_360min/60°C (right).

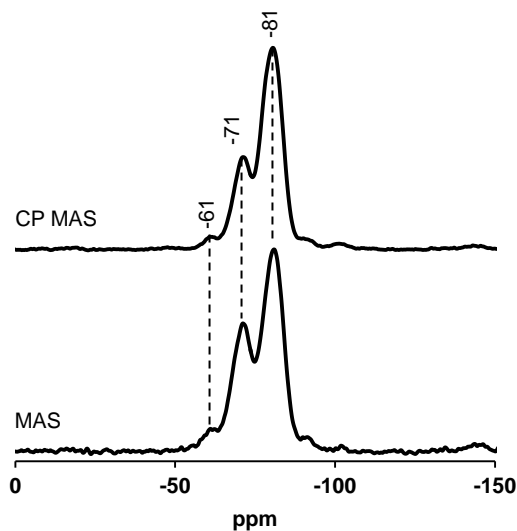


Figure S8. ²⁹Si MAS and CP MAS NMR spectra of the NH₂-PMO_15min/60°C.

Table S2. Physical properties of PMO, NO₂-PMO_15min/60°C and NH₂-PMO_15min/y materials.

Sample	d_{100} / nm	a / nm^a	$S_{\text{BET}} / \text{m}^2 \text{g}^{-1}$	$V_p / \text{cm}^3 \text{g}^{-1}$	d_p / nm^b	b / nm^c
PMO	4.55	5.25	920	0.66	3.58	1.67
NO ₂ -PMO_15min/60°C	4.70	5.42	698	0.57	3.54	1.88
NH ₂ -PMO_15min/37°C_C	4.46	5.15	752	0.44	3.27	1.88
NH ₂ -PMO_15min/37°C	4.37	5.05	722	0.44	3.40	1.65
NH ₂ -PMO_15min/60°C	4.41	5.10	699	0.44	3.40	1.70
NH ₂ -PMO_15min/75°C	4.37	5.05	688	0.43	3.40	1.65
NH ₂ -PMO_15min/90°C	4.46	5.15	686	0.44	3.54	1.61

^aUnit cell parameter calculated as $(2d_{100}/\sqrt{3})$. ^bPore width obtained from the BJH method with the corrected Kelvin equation, i.e. KJS-BJH method at the maximum of pore size distribution calculated on the basis of adsorption data. ^cPore wall thickness calculated as $(2d_{100}/\sqrt{3} - d_p)$, where the first term is the unit cell parameter.

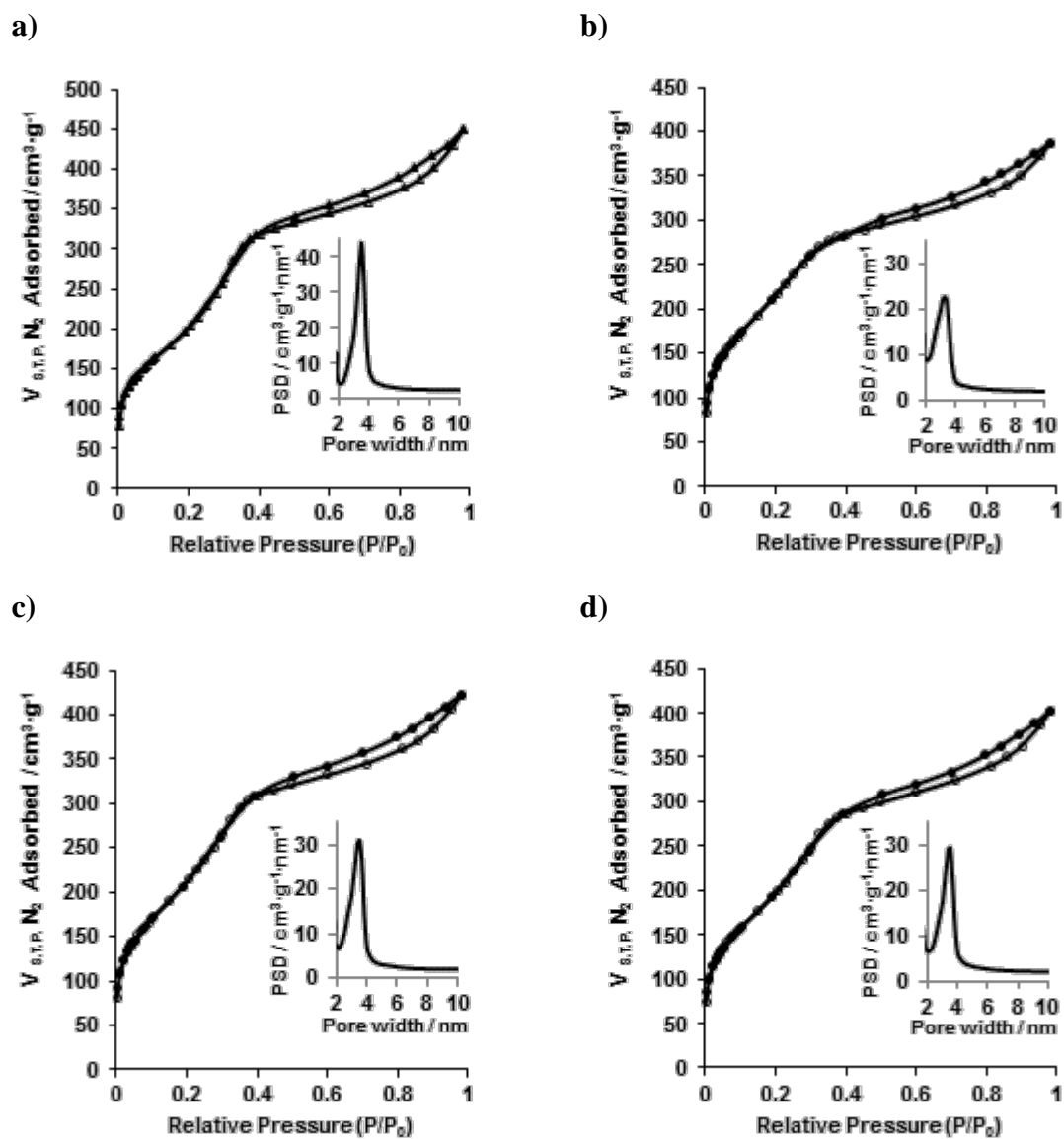


Figure S9. -196°C N_2 isotherms of a) $\text{NO}_2\text{-PMO}_{15\text{min}/60^\circ\text{C}}$ (triangles); b) $\text{NH}_2\text{-PMO}_{15\text{min}/37^\circ\text{C}_C}$ (circles); c) $\text{NH}_2\text{-PMO}_{15\text{min}/37^\circ\text{C}}$ (circles) and d) $\text{NH}_2\text{-PMO}_{15\text{min}/90^\circ\text{C}}$ (circles). Empty/full symbols correspond to adsorption/desorption. The insets display PSD curves.

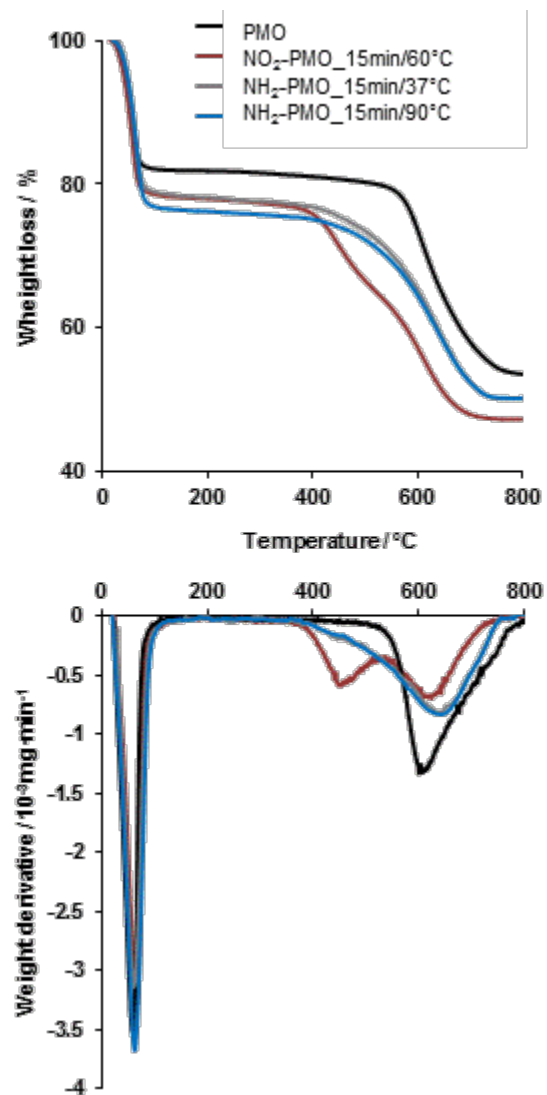


Figure S10. Weight loss curves (up) and weight loss derivatives (down) for PMO, NO₂-PMO_15min/60°C, NH₂-PMO_15min/37°C and NH₂-PMO_15min/90°C.