

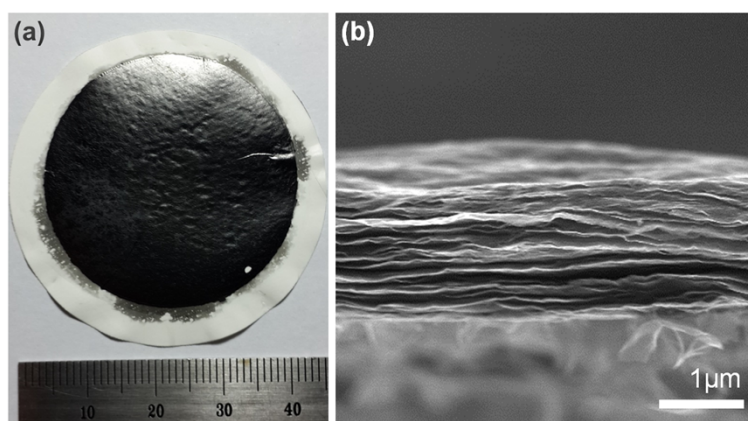
[Electronic Supplementary Information]

## Facile Fabrication of Highly Flexible Graphene Paper for High-Performance Flexible Lithium Ion Battery Anode

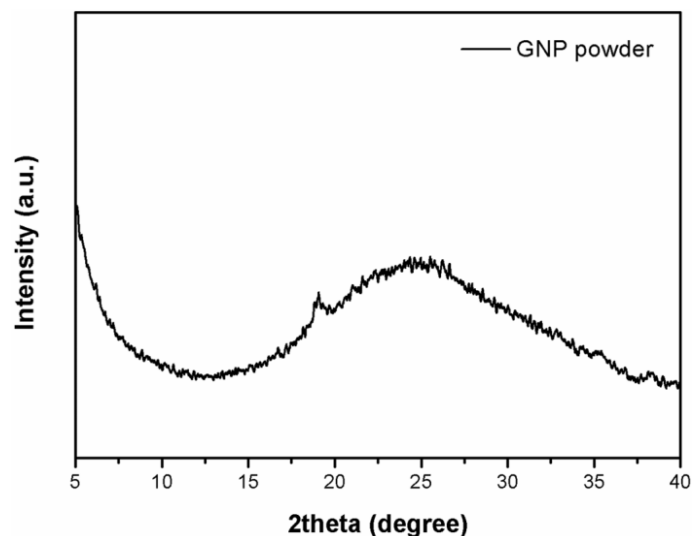
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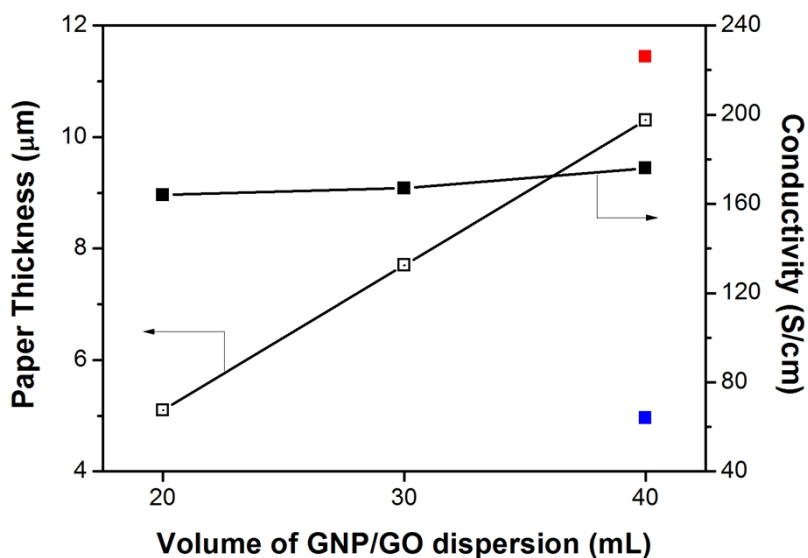
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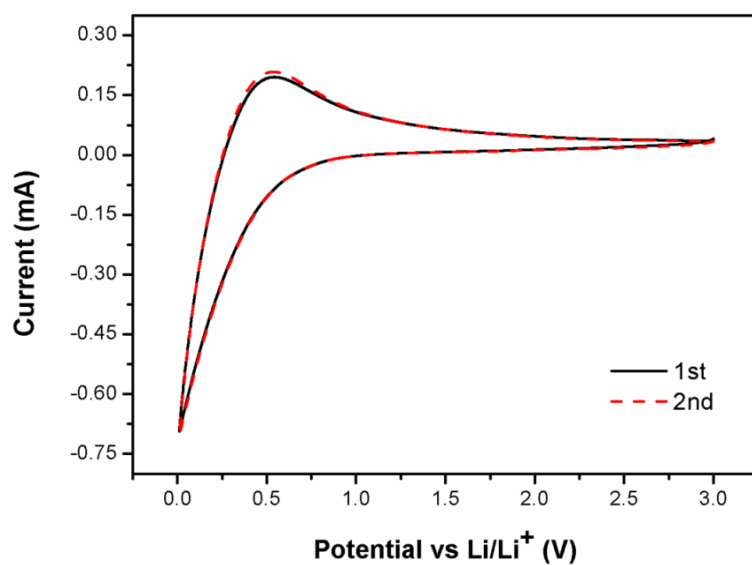
**Fig. S1** a) Optical and b) SEM images of reduced graphene oxide (rGO) paper with the same amount of graphene derivatives for the GNP/GO paper. The rGO paper was fabricated by a vacuum filtration using GO dispersion after chemical reduction by hydrazine solution, showing highly stacked layer structure with a thickness of approximately 2 μm due to GOs' high dispersion stability in water, which is consistent with what have been found by others.<sup>S1,S2</sup>



**Fig. S2** XRD pattern of GNP powder used in this study, showing only a broad peak around  $2\theta = 25^\circ$ , which indicates poor long-range ordering of graphene sheets like general graphene aerogels. In addition, there was no sharp peak centered at  $2\theta = 26.5^\circ$ , which corresponds to the folded structure of graphene layers. Peak centered at  $2\theta = 26.5^\circ$  was observed only in the case of the GNP/GO paper, indicating that local folded and corrugated morphology of graphene layers has been developed by the effective control of assembling process with intrinsically wrinkled GNP sheets.



**Fig. S3** Thickness of the GNP/GO paper can be controlled by varying the volume of GNP/GO dispersion for a vacuum filtration (filtering diameter: *ca.* 37 mm). Conductivity of the GNP/GO paper was slightly increased as the thickness of the paper was increased. Blue and red squares in the graph indicate the conductivity of papers with 40 mL of GNP/GO dispersion after thermal annealing at 350 °C for 1 h in an air, and with GNP/GO dispersion with a weight ratio of 2/1, respectively.



**Fig. S4** CV curves for the GNP/GO paper that was thermally annealed at 350 °C for 1 h in an air.

**Table S1.** BET surface area of the rGO paper, the GNP/GO paper, and pristine GNP powder.

	rGO paper	GNP/GO paper	pristine GNP powder
surface area [m <sup>2</sup> g <sup>-1</sup> ] <sup>[a]</sup>	52.8	278.9	> 400 <sup>[b]</sup>

<sup>[a]</sup> BET surface areas were determined with a gas sorption analyzer (Belsorp-max).

<sup>[b]</sup> This value was acquired from a technical data sheet provided from the Angstrom Materials.

## References

- S1 H. Chen, M. B. Müller, K. J. Gilmore, G. G. Wallace, D. Li, *Adv. Mater.*, 2008, **20**, 3557–3561.  
 S2 C. Wang, D. Li, C. O. Too, G. G. Wallace, *Chem. Mater.*, 2009, **21**, 2604–2606.