## Multicomponent synthesis of Pyridines *via* diamine functionalized mesoporous ZrO<sub>2</sub> domino Intramolecular tandem Michael type addition

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## Materials and methods Apparatus and analysis

All chemicals used were reagent grade and were used as received without further purification. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded at 25 °C at 400 MHz and 100 MHz (Bruker Avance) respectively, using TMS as internal standard. Chemical shifts are given in parts per million (ppm). FT-IR spectroscopy was performed on a Perkin-Elmer Precisely 100 FT-IR spectrometer in the 400-40000 cm-1 region. ESI-MS spectra were determined on a LCQ ion trap mass spectrometer (Thermo Fisher, San Jose, CA, USA), equipped with an ESI source. Melting points were recorded on a hot stage melting point apparatus Ernst Leitz Wetzlar, Germany and were uncorrected. All the reactions and the purity of products were monitored using thin layer chromatography (TLC) on aluminum-backed plates coated with Merck Kieselgel 60 F254 silica gel, visualizing the spots under ultraviolet light and iodine chamber.



Figure S1 (a) N<sub>2</sub>-adsoption-desoption isotherm of AAPTMS/m-ZrO<sub>2</sub> and (b) pore size distribution of AAPTMS/m-ZrO<sub>2</sub>MAS NMR spectra of DF/m-ZrO<sub>2</sub>



Figure S2. Small angle XRD spectra of DF/m-ZrO<sub>2</sub>



Figure S3. Scanning electron micrograph of AAPTMS/m-ZrO<sub>2</sub> sample



Figure S4. TEM image of AAPTMS/m-ZrO<sub>2</sub> sample



Figure S5. The schematic diagram for the synthesis of AAPTMS/m-ZrO $_2$  catalyst





FTIR spectra of compound 5a





FTIR spectra of compound 5b





FTIR spectra of compound 5c





FTIR spectra of compound **5d** S11





FTIR spectra of compound 5e





FTIR spectra of compound 5f





FTIR spectra of compound 5g







FTIR spectra of compound 5h

