

## Supplementary Information †

### Structure, magnetism and catecholase activity of the first dicopper(II) complex having a single $\mu$ -alkoxo bridge

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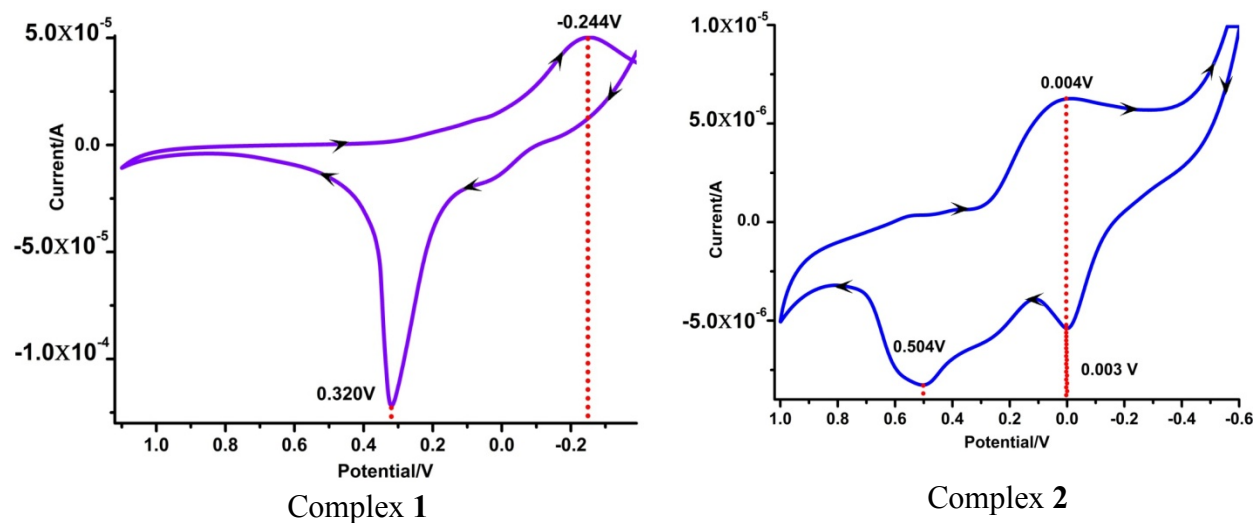
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**Table S1.** Dependence of initial rates on the concentration of 3,5-DTBC, [conc. of Cu(II) complexes is  $1 \times 10^{-4}$  mol dm<sup>-3</sup>].

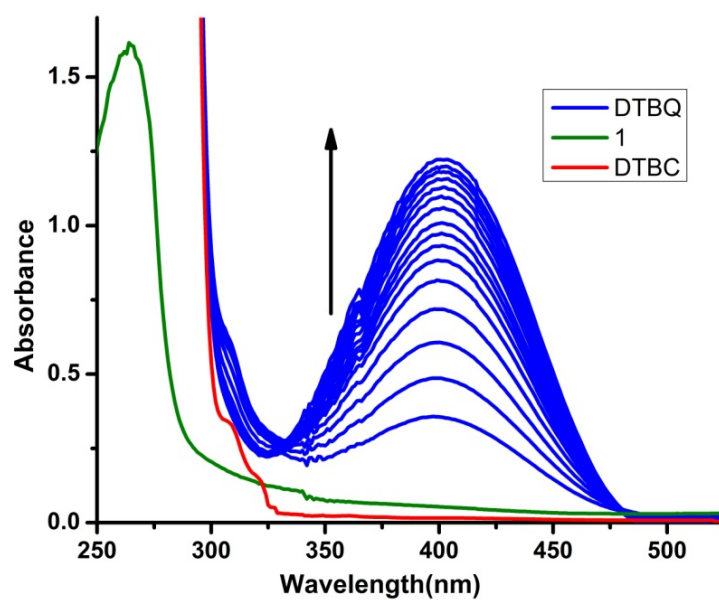
[3,5-DTBC] (mol dm <sup>-3</sup> )	V (mol dm <sup>-3</sup> s <sup>-1</sup> ) complex <b>1</b>	V (mol dm <sup>-3</sup> s <sup>-1</sup> ) complex <b>2</b>
1E-3	7E-5	2.6049E-4
2E-3		3.564E-4
3E-3	1.34E-4	4.2946E-4
5E-3	1.68E-4	4.73E-4
7E-3	1.8949E-4	4.82E-4
1E-2	1.954E-4	5.013E-4

**Table S2.** Crystal data and structure refinement for **1** and **2**

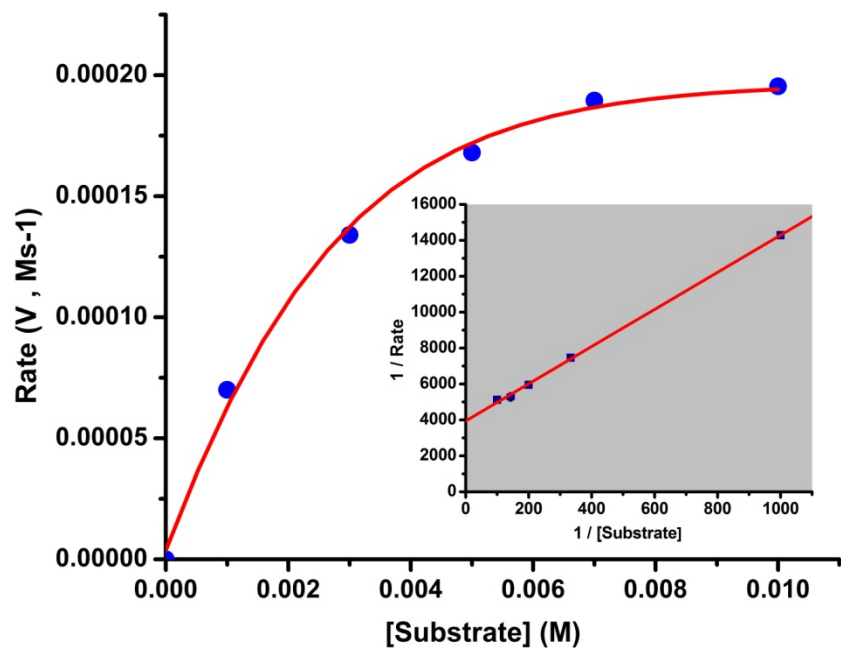
	<b>1</b>	<b>2</b>
empirical formula	C <sub>14</sub> H <sub>17</sub> CuN <sub>2</sub> O <sub>4</sub> ·ClO <sub>4</sub>	C <sub>35</sub> H <sub>41</sub> Cl <sub>2</sub> Cu <sub>3</sub> N <sub>5</sub> O <sub>18</sub>
formula weight	440.29	1081.25
crystal system	Orthorhombic	Monoclinic
space group	<i>Fdd2</i>	<i>P2</i> <sub>1</sub>
<i>T</i> (K)	100(2)	100(2)
<i>a</i> (Å)	27.107 (2)	11.1962(5)
<i>b</i> (Å)	42.977 (2)	15.1146(7)
<i>c</i> (Å)	11.7228 (8)	11.9446(5)
$\alpha$ (°)	90	90
$\beta$ (°)	90	94.215(2)
$\gamma$ (°)	90	90
Volume (Å <sup>3</sup> )	13656.8 (16)	2015.87(15)
<i>Z</i>	32	2
$\mu$ (mm <sup>-1</sup> )	1.48	1.787
reflections collected	41818	19831
independent reflections	6835 [ <i>R</i> <sub>int</sub> = 0.05]	19831 [ <i>R</i> <sub>int</sub> = 0.09]
data/restraints/parameters	41818/ 9 / 539	19831/ 1497 / 578
final <i>R</i> index [ <i>I</i> > 2σ( <i>I</i> )]	<i>R</i> <sub>1</sub> = 0.0337, <i>wR</i> <sub>2</sub> = 0.0635	<i>R</i> <sub>1</sub> = 0.0525, <i>wR</i> <sub>2</sub> = 0.1168
final <i>R</i> index [ <i>all data</i> ]	<i>R</i> <sub>1</sub> = 0.0418, <i>wR</i> <sub>2</sub> = 0.0665	<i>R</i> <sub>1</sub> = 0.0698, <i>wR</i> <sub>2</sub> = 0.1314



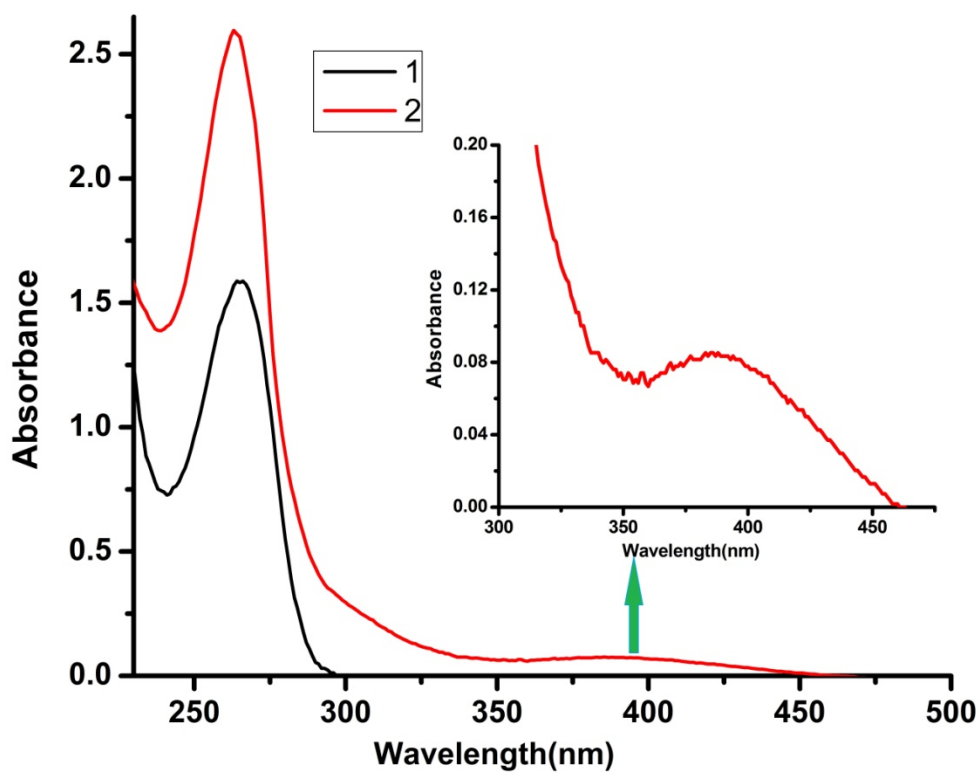
**Fig. S1** Cyclic voltammogram of complexes **1** and **2** in MeOH using 0.1 M TBAP as supporting electrolyte and platinum working electrode (scan rate  $100 \text{ mVs}^{-1}$ ).



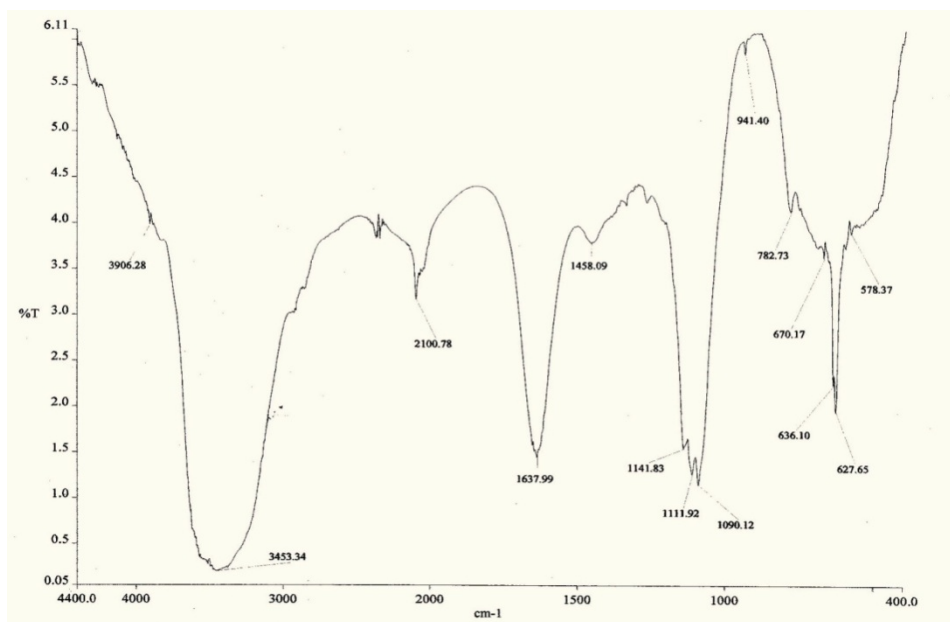
**Fig. S2** Increase in the quinone band at 400 nm after addition of 100 fold 3,5-DTBC ( $1 \times 10^{-2} \text{ M}$ ) to the methanol solution of complex **2** ( $1 \times 10^{-4} \text{ M}$ ). The spectra are recorded at an interval of 5 min.



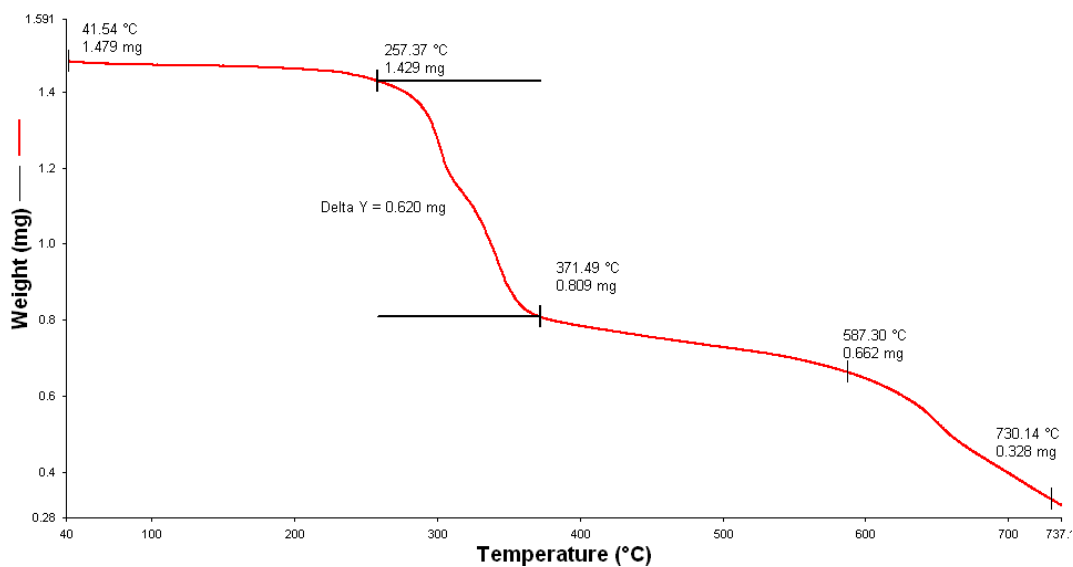
**Fig. S3** Plot of rate vs. substrate concentration for complex **1**. Inset shows Lineweaver–Burk plot.



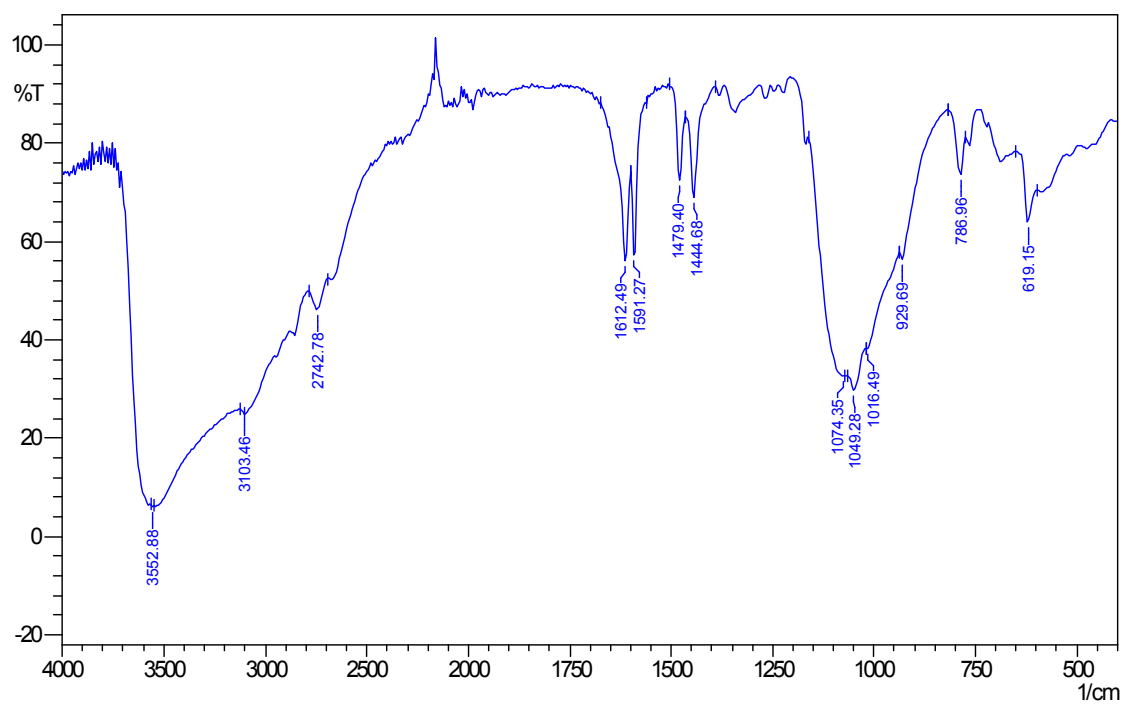
**Fig. S4** UV-Vis spectra of complex **1** and **2** in MeOH ( $[1] = 1 \times 10^{-4}$  M)



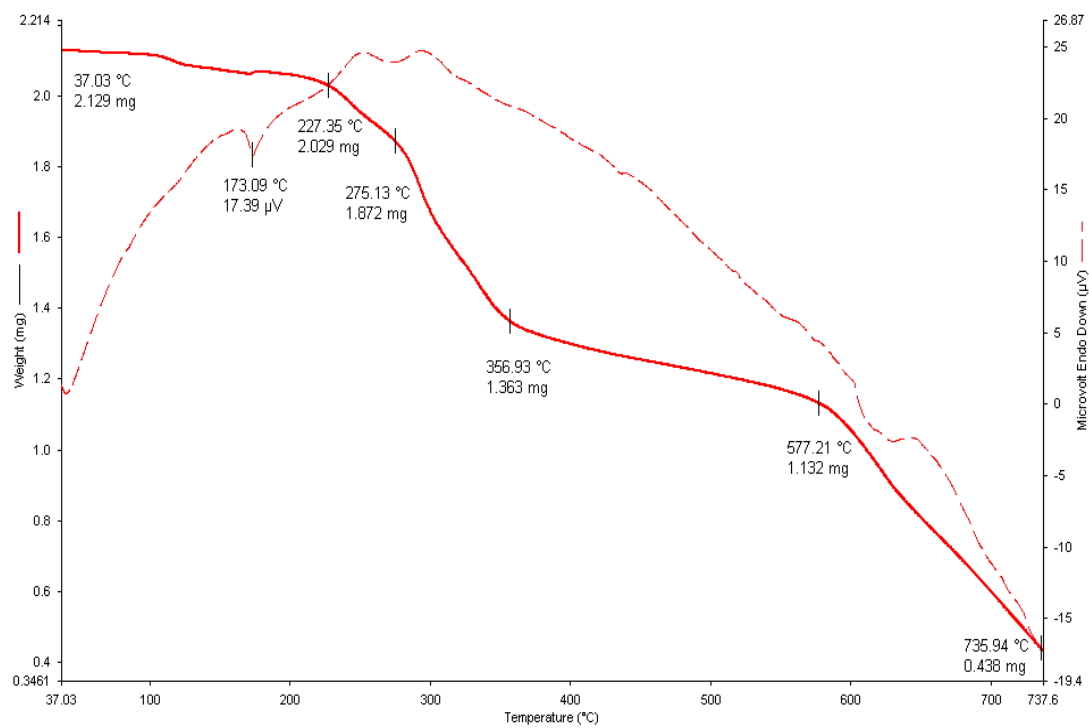
**Fig. S5** FTIR spectrum of **1**



**Fig. S6** Thermogram of **1**



**Fig. S7** FTIR spectrum of **2**



**Figure S8** Thermogram of **2**