

Supplementary Information

Cu(II) immobilized on aminated epichlorohydrin activated silica (CAES): as a new, green and efficient nanocatalyst for preparation of 5-substituted-1*H*-tetrazoles

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Experimental

General

The purity determinations of the products were accomplished by TLC on silica gel polygram STL G/UV 254 plates. The melting points of products were determined with an Electrothermal Type 9100 melting point apparatus. The FT-IR spectra were recorded on an Avatar 370 FT-IR Thermo Nicolet spectrometer. The NMR spectra were provided on Bruker Avance 100 and 400 MHz instruments in acetone-*d*₆, DMSO-*d*₆ and CD₃CN. Mass spectra were recorded with Agilent Technologies (HP) 5973 Network Mass Selective Detector and Shimadzu GC-MS-QP5050 instruments at 70 eV. Thermogravimetric analysis (TGA) was performed on a Shimadzu Thermogravimetric Analyzer (TG-50) under air atmosphere. BET surface area and pore size distribution were measured on a Belsorp mini II system at -196 °C using N₂ as adsorbate. Inductively coupled plasma (ICP) was carried out on a Varian, VISTA-PRO, CCD, Australia. All of the products were known compounds and characterized by the FT-IR and comparison of their melting points with known compounds. The structure of selected products was further confirmed by ¹HNMR, ¹³CNMR spectroscopy, and mass spectrometry.

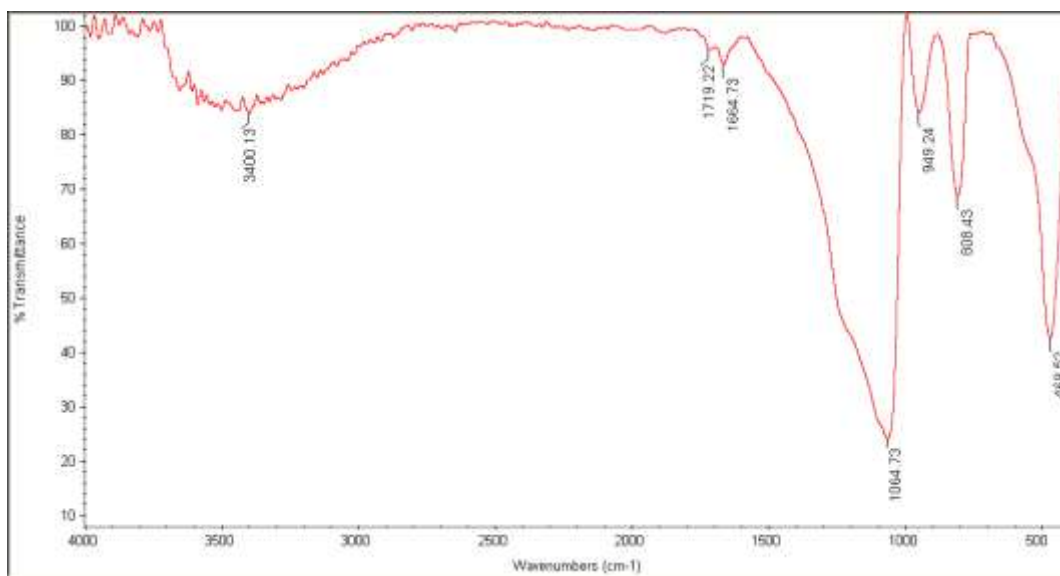


Figure 1: FT-IR (KBr) of activated silica with HCl.

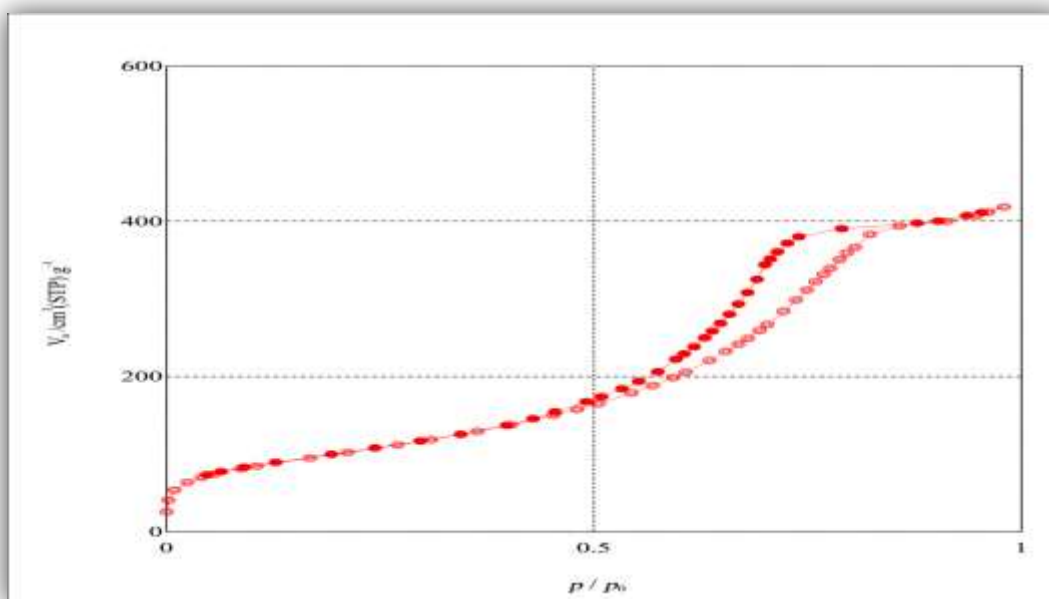


Figure 2: Adsorption/desorption isotherm of activated silica with HCl.

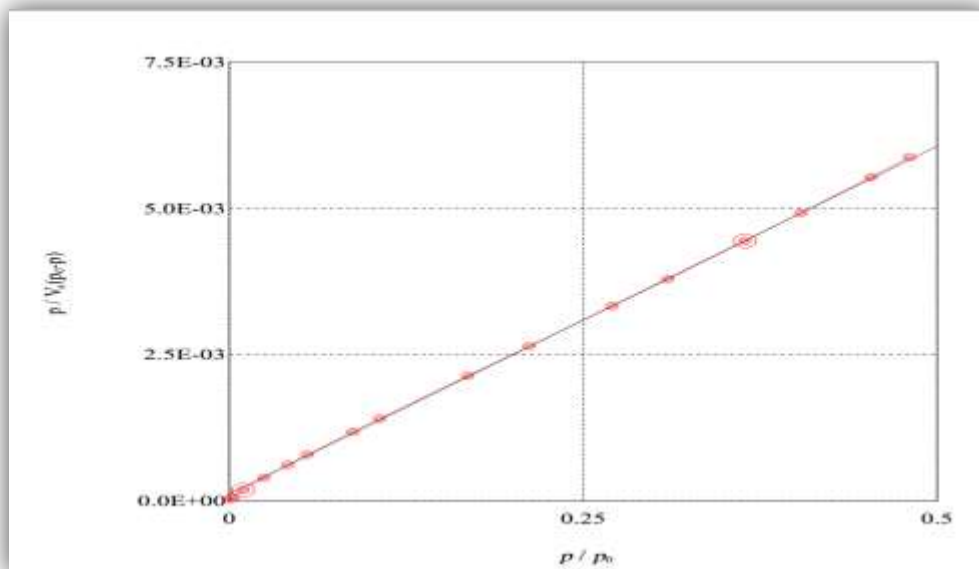


Figure 3: BET-Plot of activated silica with HCl.

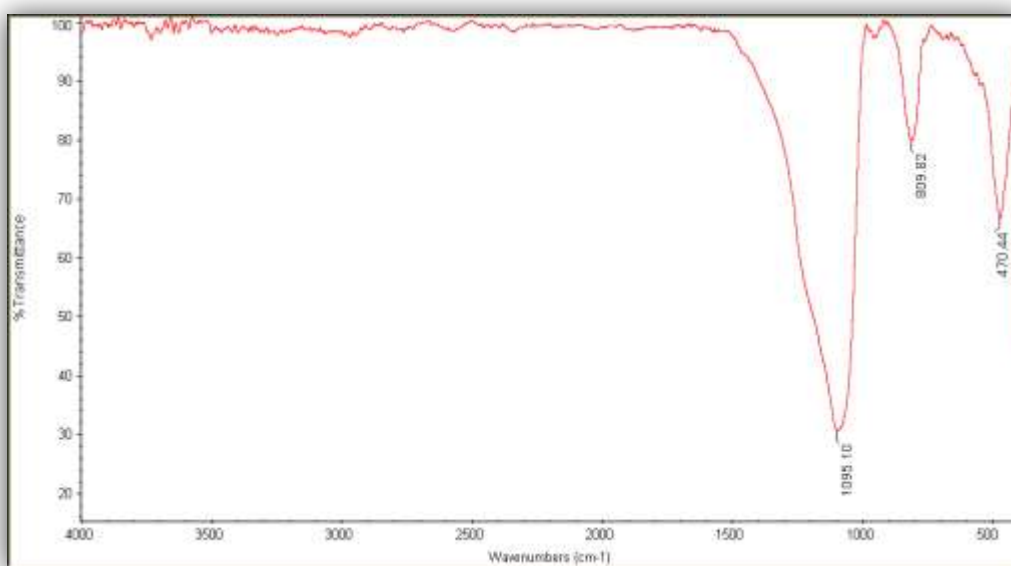


Figure 4: FT-IR (KBr) of activated silica with NaOH solution.

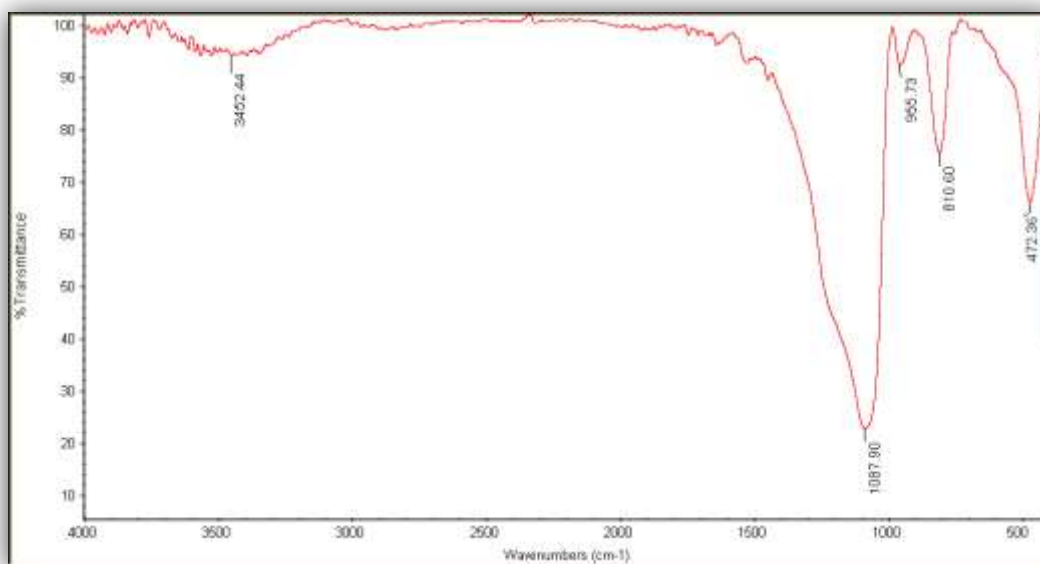


Figure 5: FT-IR (KBr) of epichlorohydrin-SiO₂.

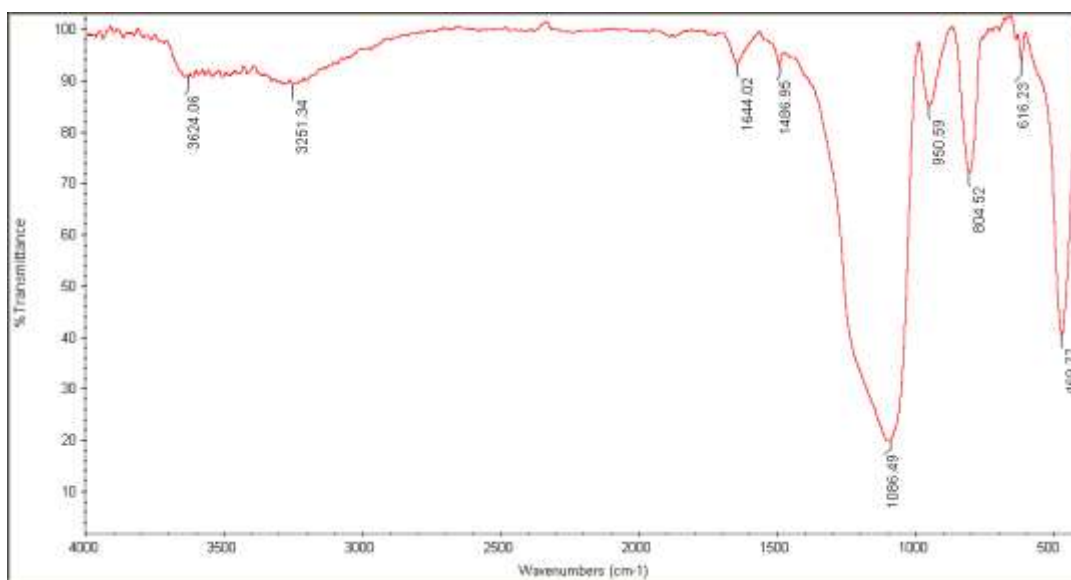


Figure 6: FT-IR (KBr) of aminated epichlorohydrin-SiO₂.

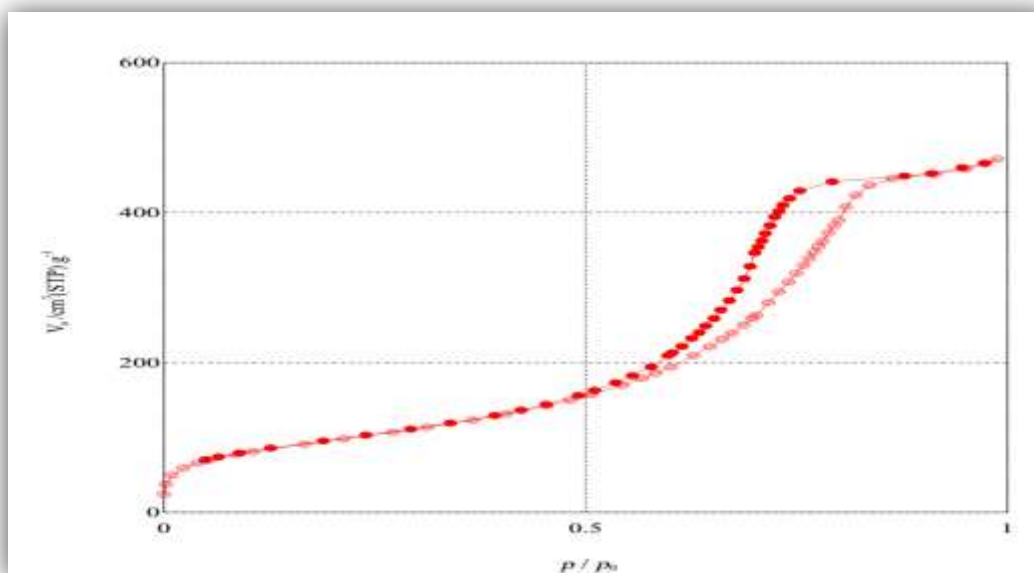


Figure 7: Adsorption/desorption isotherm of aminated epichlorohydrin-SiO₂.

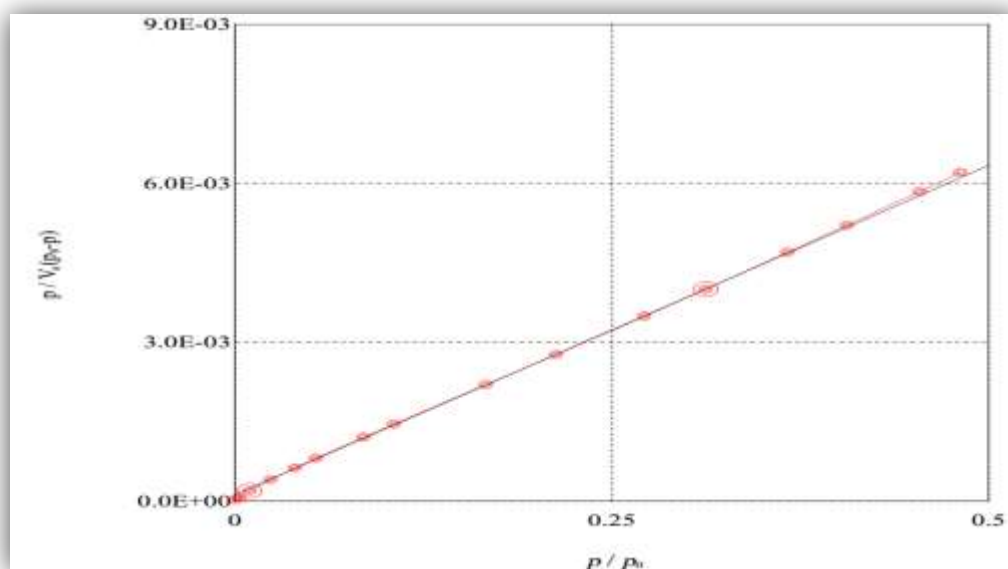


Figure 8: BET-Plot of aminated epichlorohydrin-SiO₂.

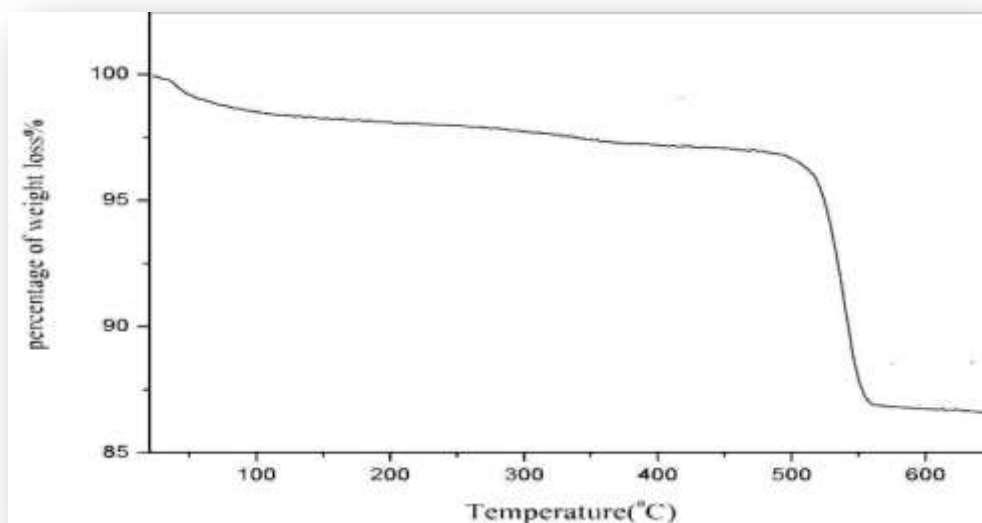


Figure 9: Thermogravimetric analysis (TGA) of aminated epichlorohydrin-SiO₂.

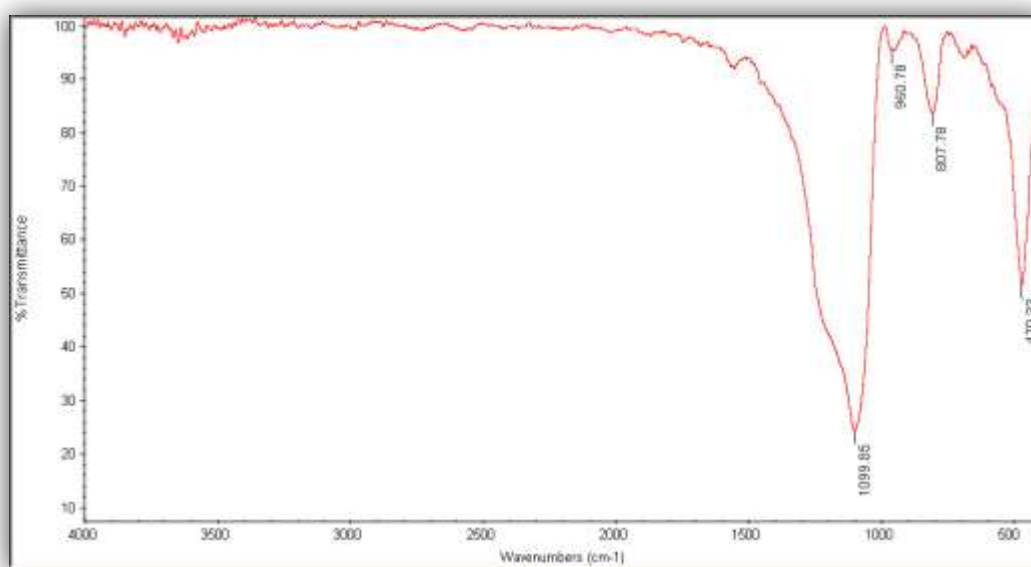


Figure 10: FT-IR (KBr) of Cu supported on aminated epichlorohydrin-SiO₂.

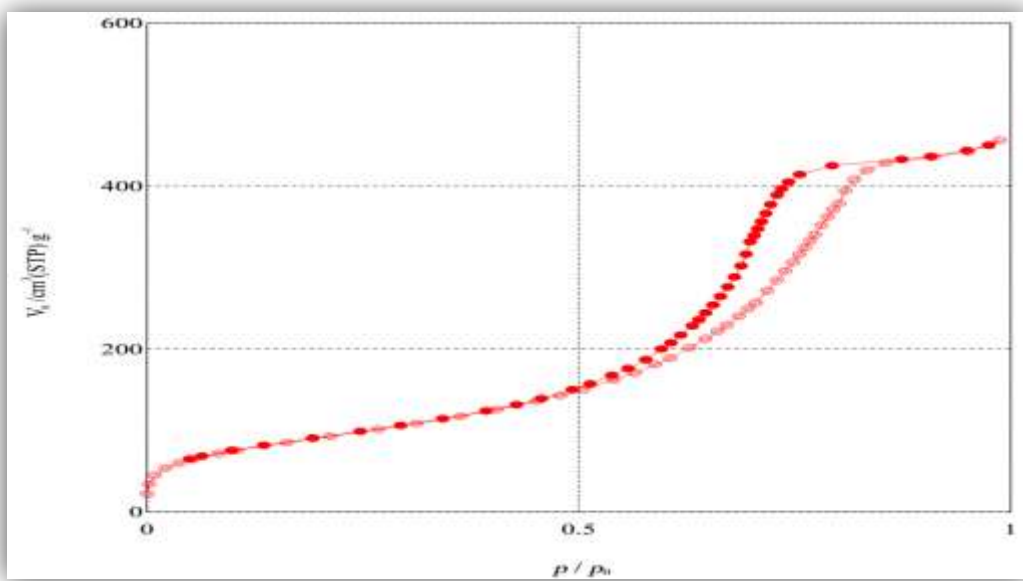


Figure 11: Adsorption/desorption isotherm of Cu supported on aminated epichlorohydrin-SiO₂.

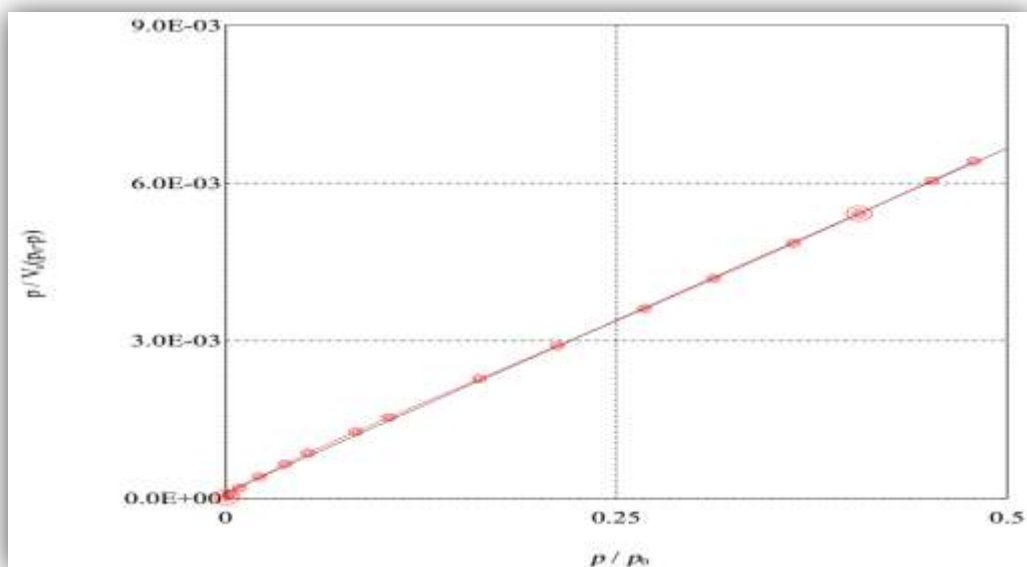


Figure 12: BET-Plot of Cu supported on aminated epichlorohydrin-SiO₂.

Sample (Samp)		12/31/2014, 9:12:09 AM				Tube 61	
Weight: 0.0617		Volume: 25				Dilution: 1	
El	Wavelen.	Sol'n Conc.	Units	SD	%RSD	Int. (c/s)	Calc Conc.
Cu	324.754	39.716	ppm	0.21826	0.5	104924	16092 ppm

Figure 13: ICP of Cu supported on aminated epichlorohydrin-SiO₂.

Cu (Samp)		2/21/2015, 5:33:15 AM				Tube 33	
Weight: 0.0444		Volume: 20				Dilution: 5	
El	Wavelen.	Sol'n Conc.	Units	SD	%RSD	Int. (c/s)	Calc Conc.
Cu	324.754	4.6406	ppm	0.016139	0.3	11199	12452 ppm

Figure 14: ICP of fifth reused Cu supported on aminated epichlorohydrin-SiO₂.

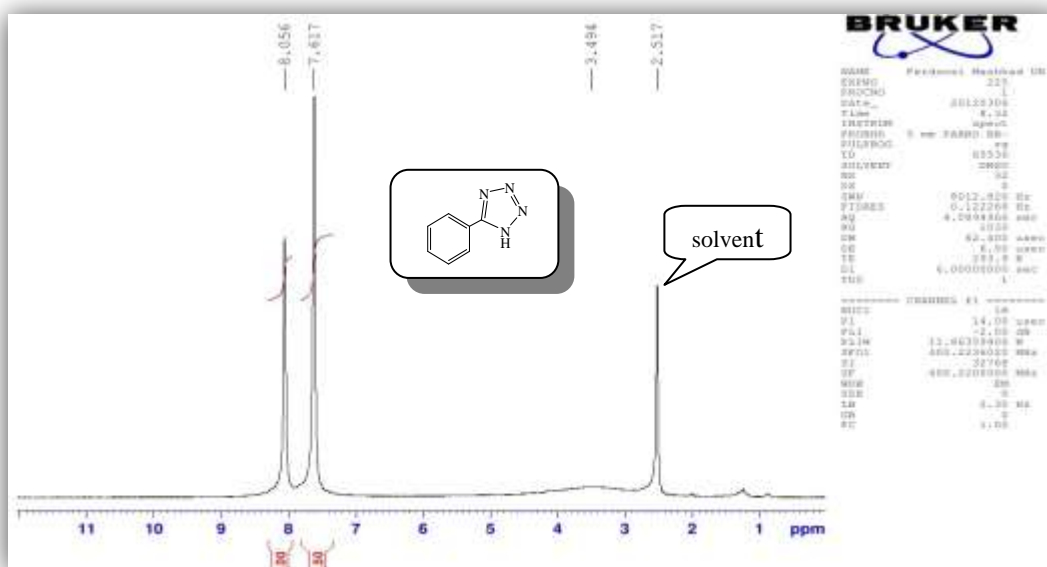


Figure 15: ^1H NMR (400 MHz, DMSO-d_6) of 5-Phenyl-1H-tetrazole (Table 2, entry 1).

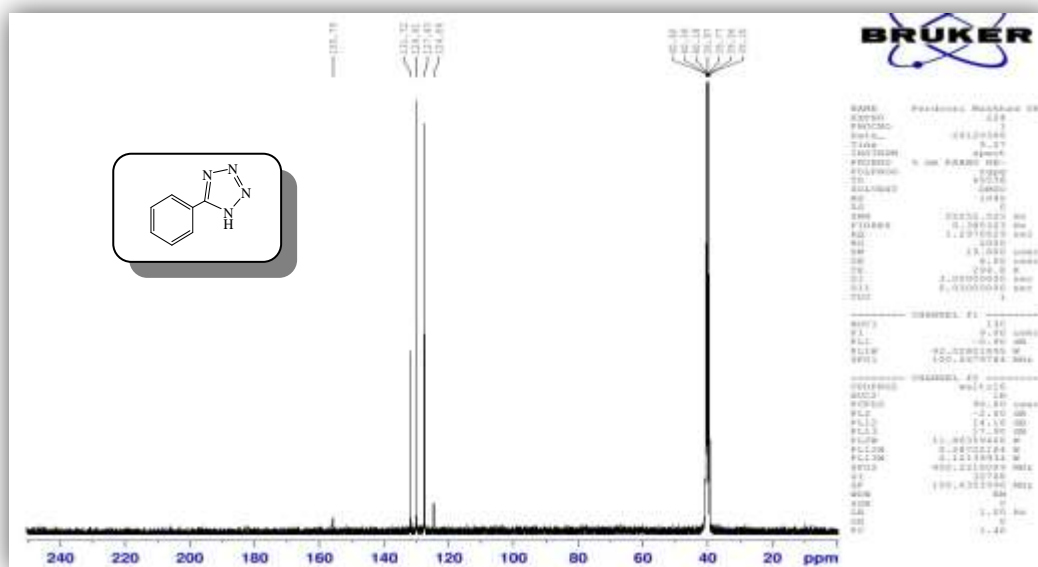
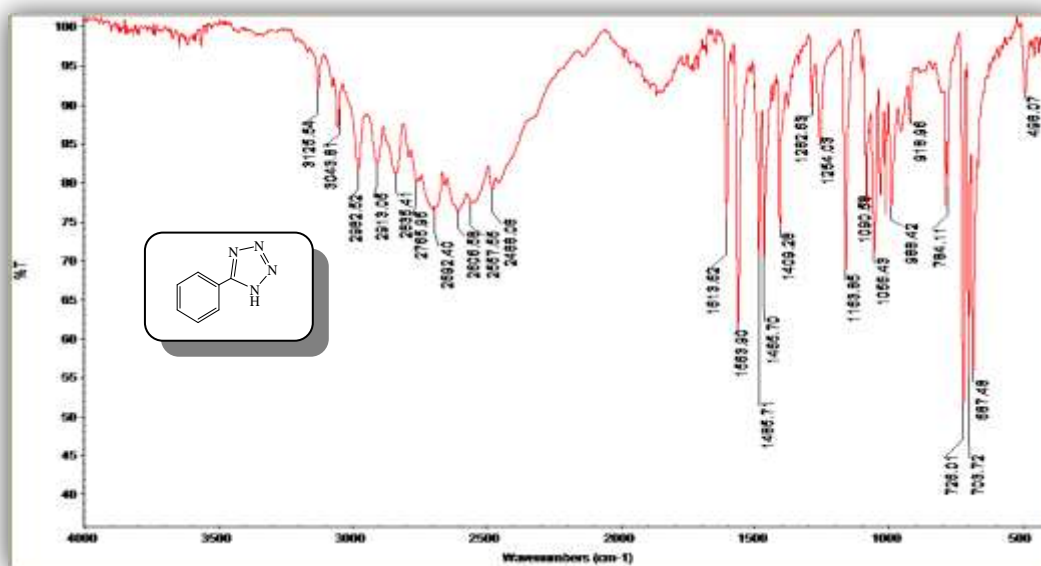
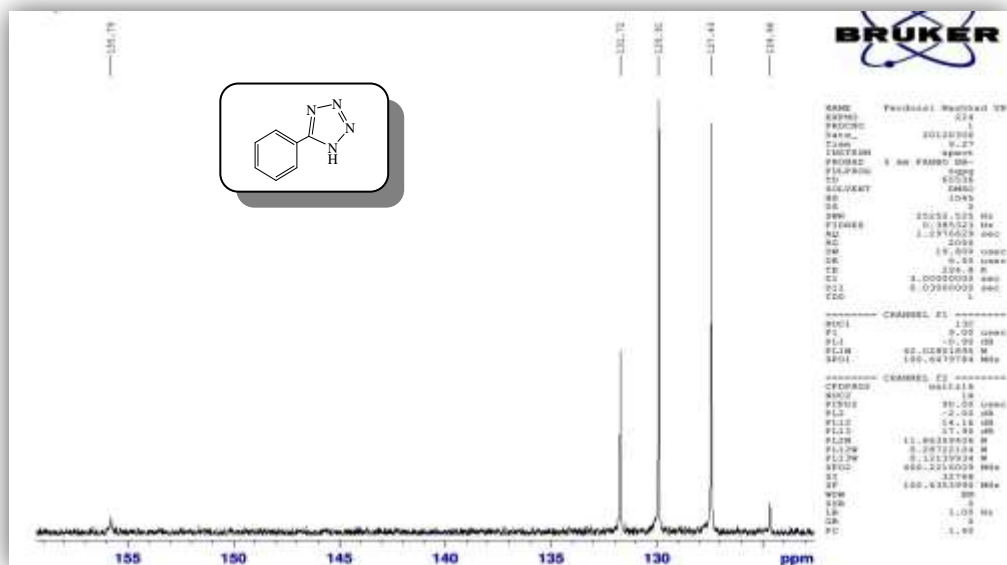


Figure 16: ^{13}C NMR (100 MHz, DMSO-d_6) of 5-Phenyl-1H-tetrazole (Table 2, entry 1).



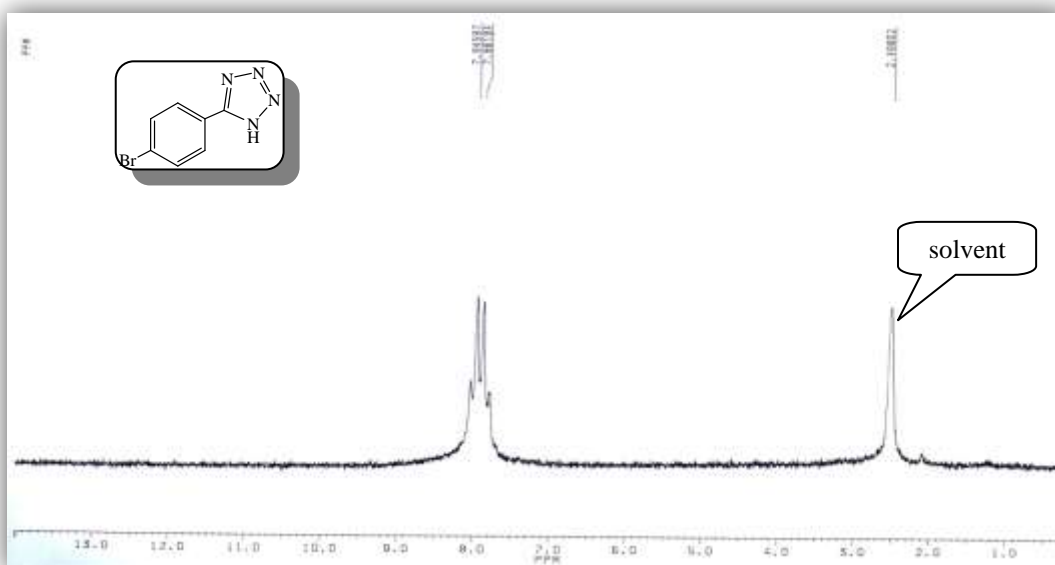


Figure 19: ¹H NMR (100 MHz, DMSO-d₆) of 5-(4-Bromophenyl)-1*H*-tetrazole (Table 2, entry 2).

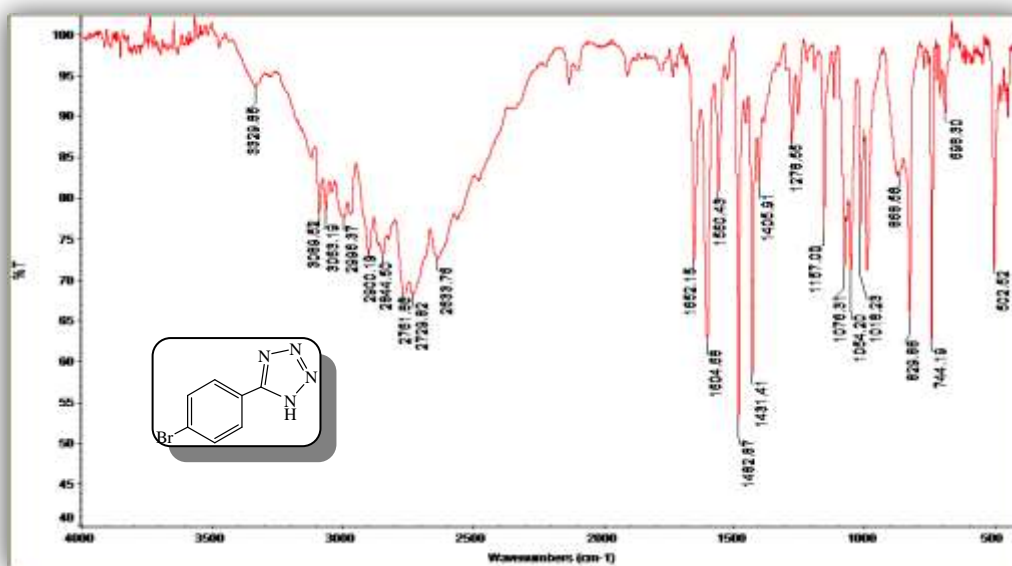


Figure 20: FT-IR (KBr) of 5-(4-Bromophenyl)-1*H*-tetrazole (Table 2, entry 2).

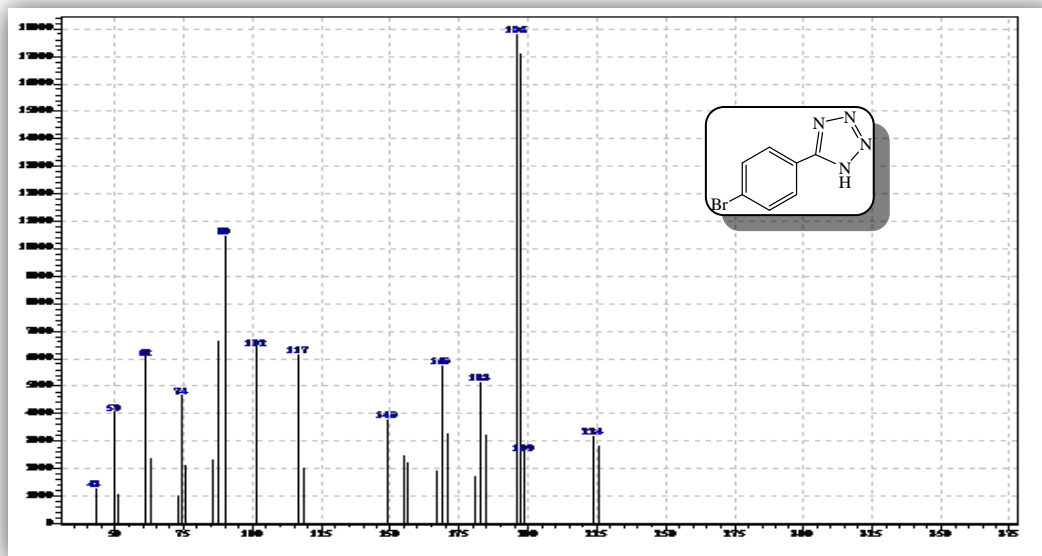


Figure 21: Mass spectrum of 5-(4-Bromophenyl)-1H-tetrazole (Table 2, entry 2).

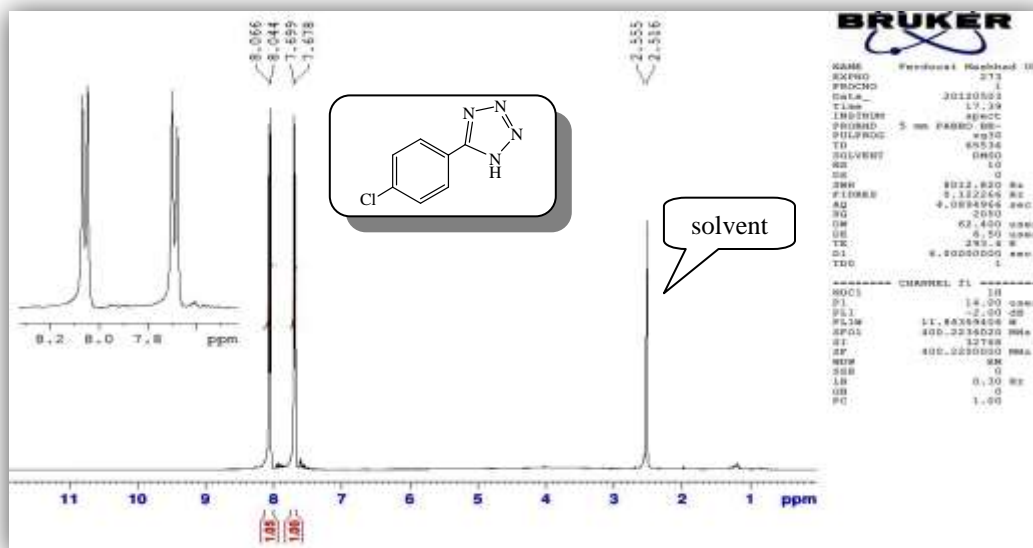


Figure 22: ¹H NMR (400 MHz, DMSO-d₆) of 5-(4-Chlorophenyl)-1H-tetrazole (Table 2, entry 3).

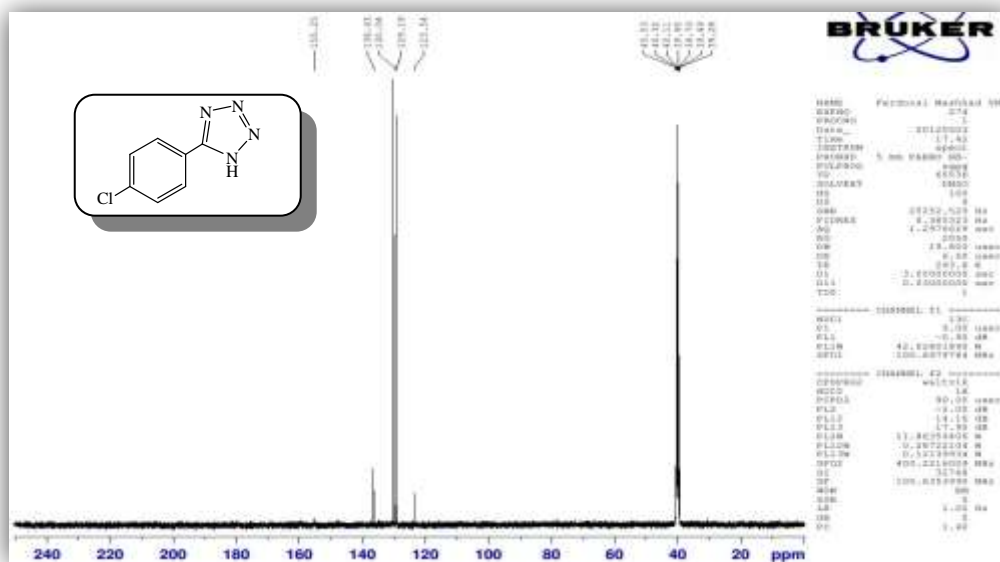


Figure 23: ¹³C NMR (100 MHz, DMSO-d₆) of 5-(4-Chlorophenyl)-1*H*-tetrazole (Table 2, entry 3).

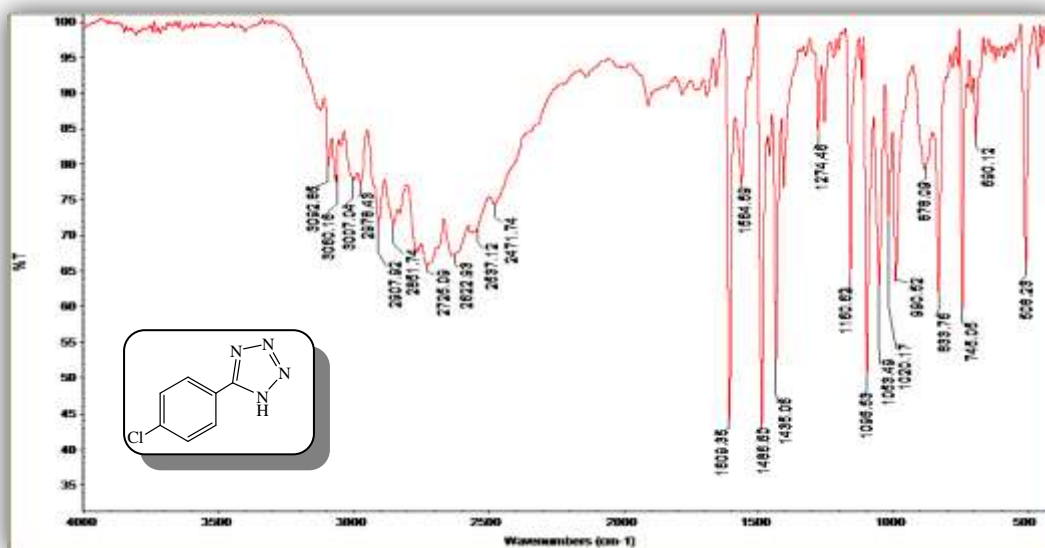


Figure 24: FT-IR (KBr) of 5-(4-Chlorophenyl)-1*H*-tetrazole (Table 2, entry 3).

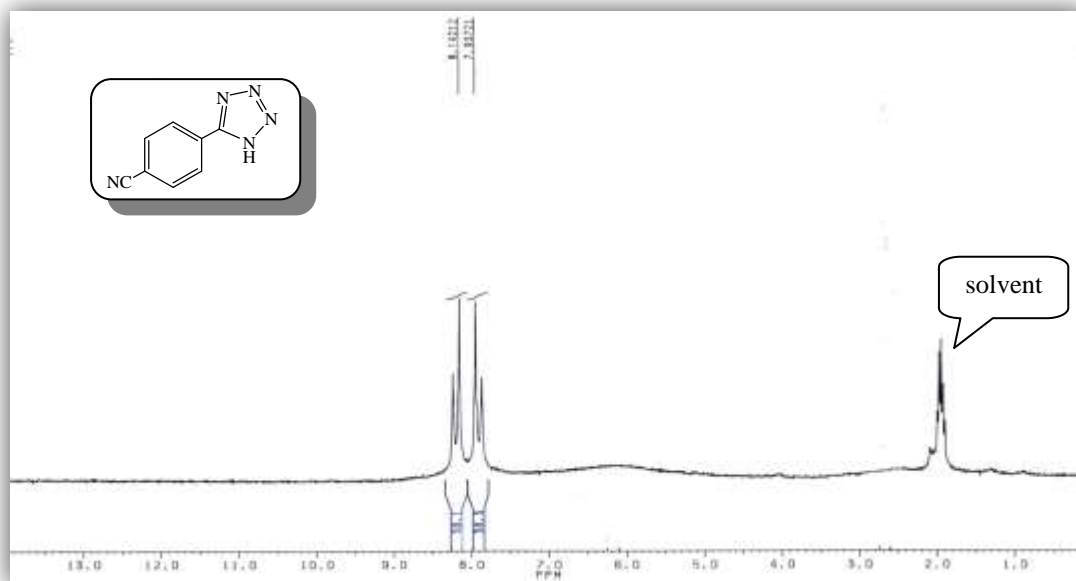


Figure 25: ¹H NMR (100 MHz, CD₃CN) of 4-(1*H*-tetrazol-5-yl)benzonitrile (Table 2, entry 4).

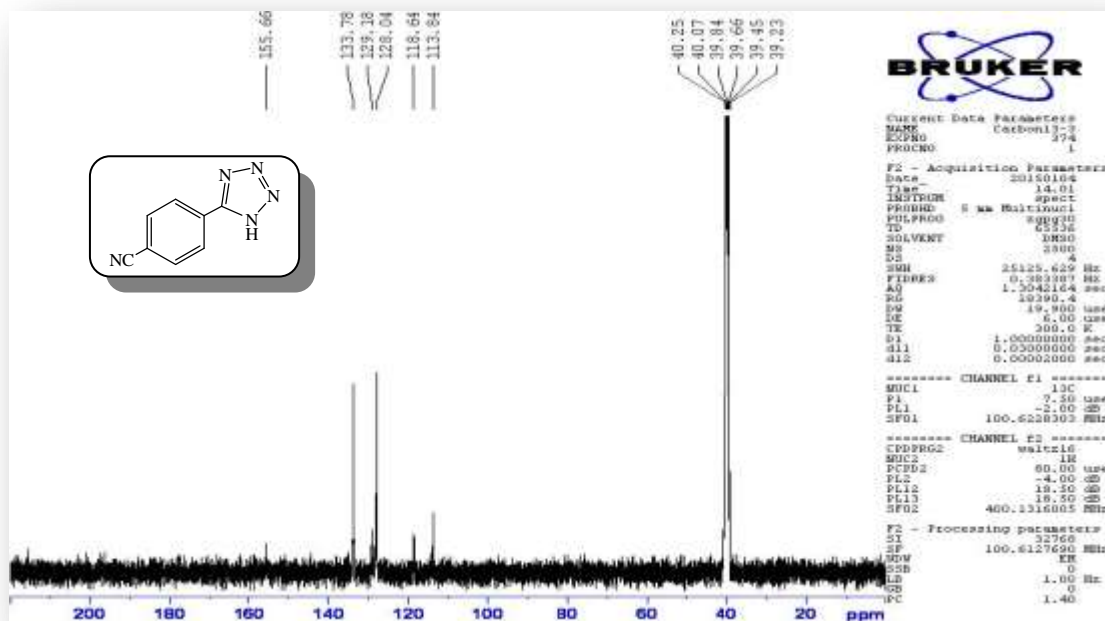


Figure 26: ¹³C NMR (400 MHz, DMSO-*d*₆) of 4-(1*H*-tetrazol-5-yl)benzonitrile (Table 2, entry 4).

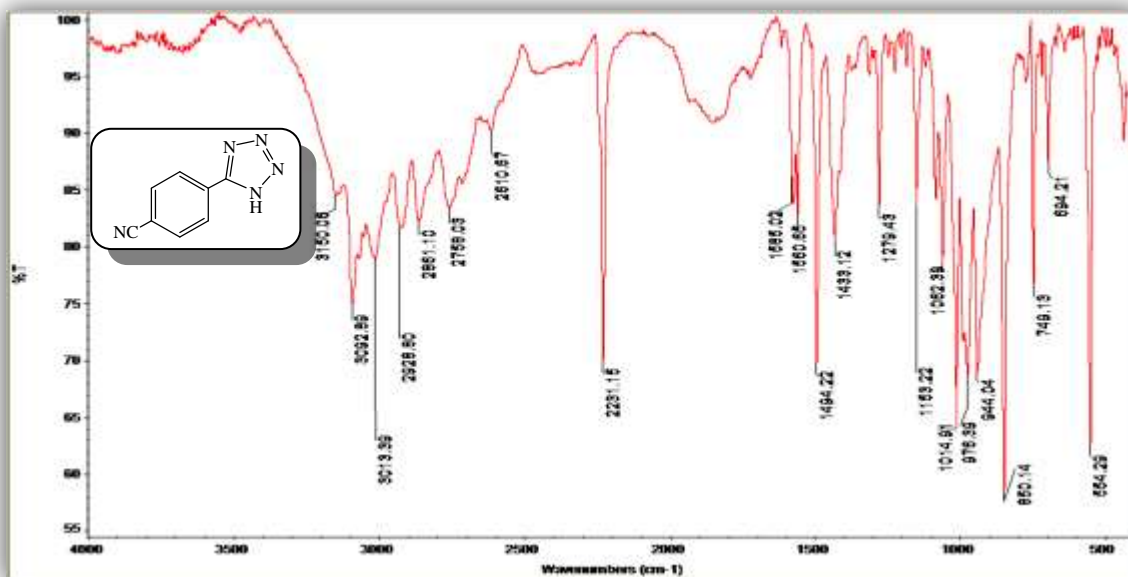


Figure 27: FT-IR (KBr) of 4-(1H-tetrazol-5-yl)benzonitrile (Table 2, entry 4).

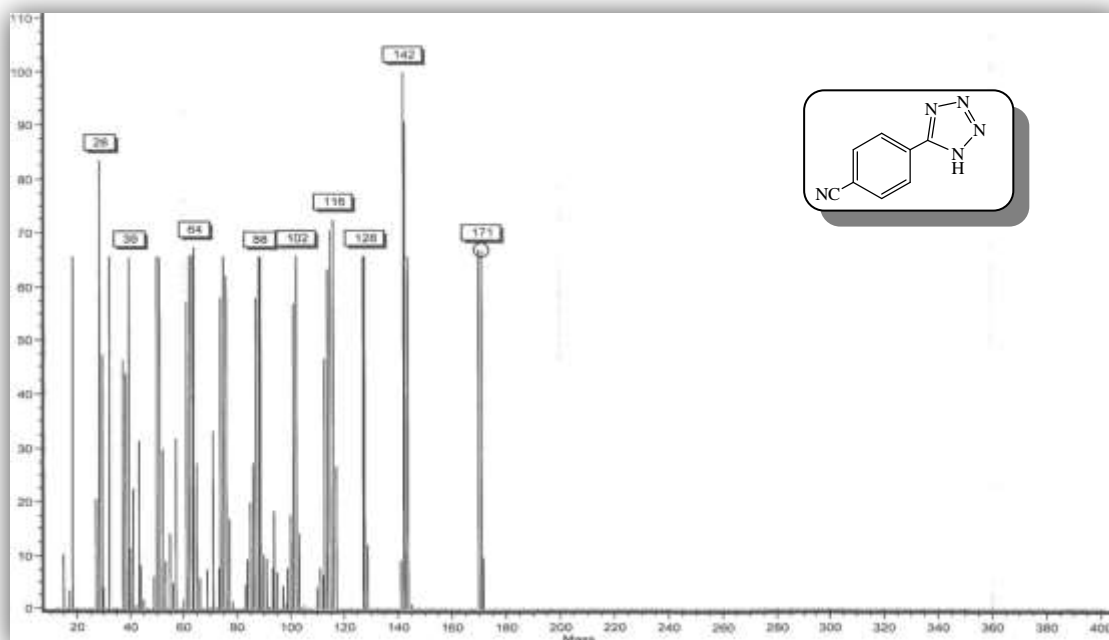


Figure 28: Mass spectrum of 4-(1H-tetrazol-5-yl)benzonitrile (Table 2, entry 4).

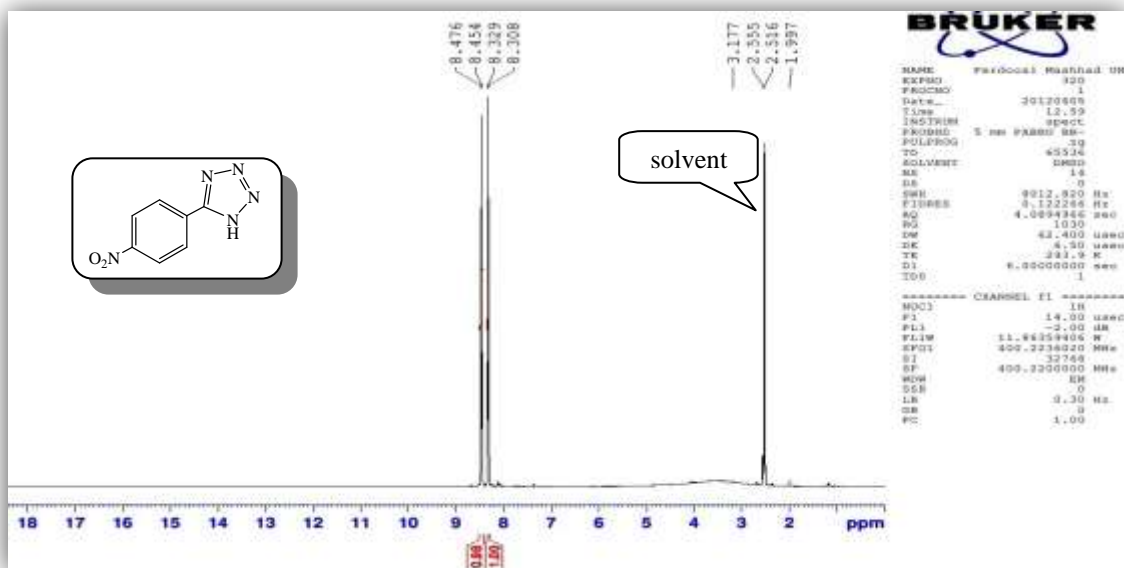


Figure 29: ^1H NMR (400 MHz, DMSO-d_6) of 5-(4-Nitrophenyl)-1H-tetrazole (Table 2, entry 5).

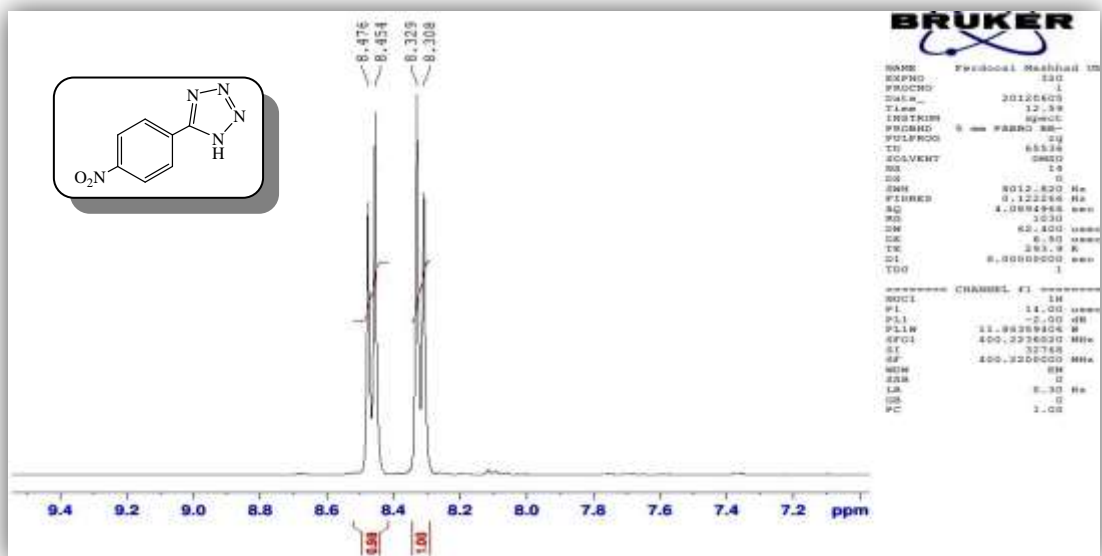
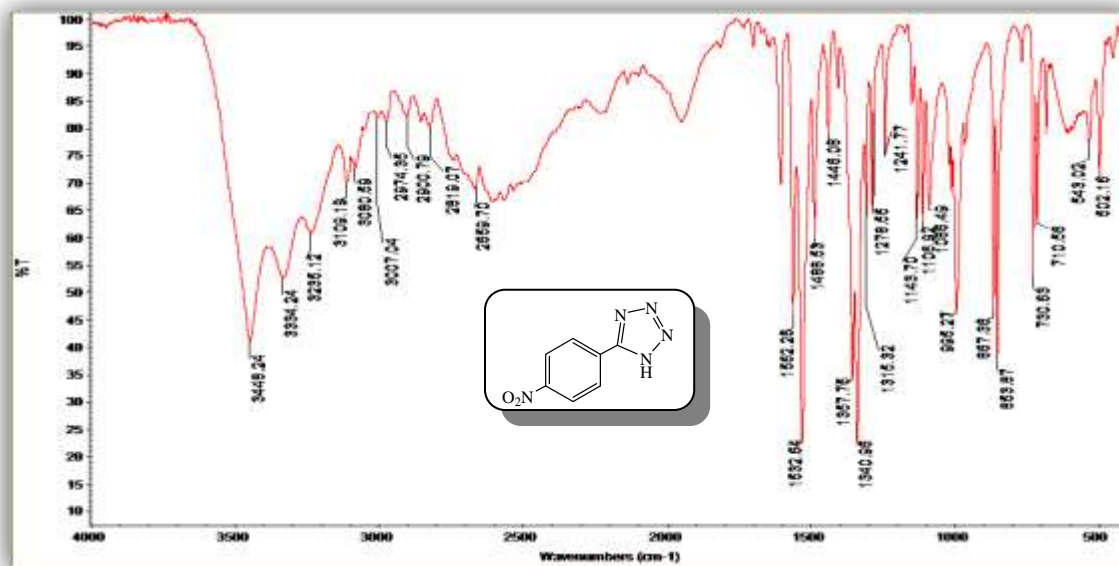
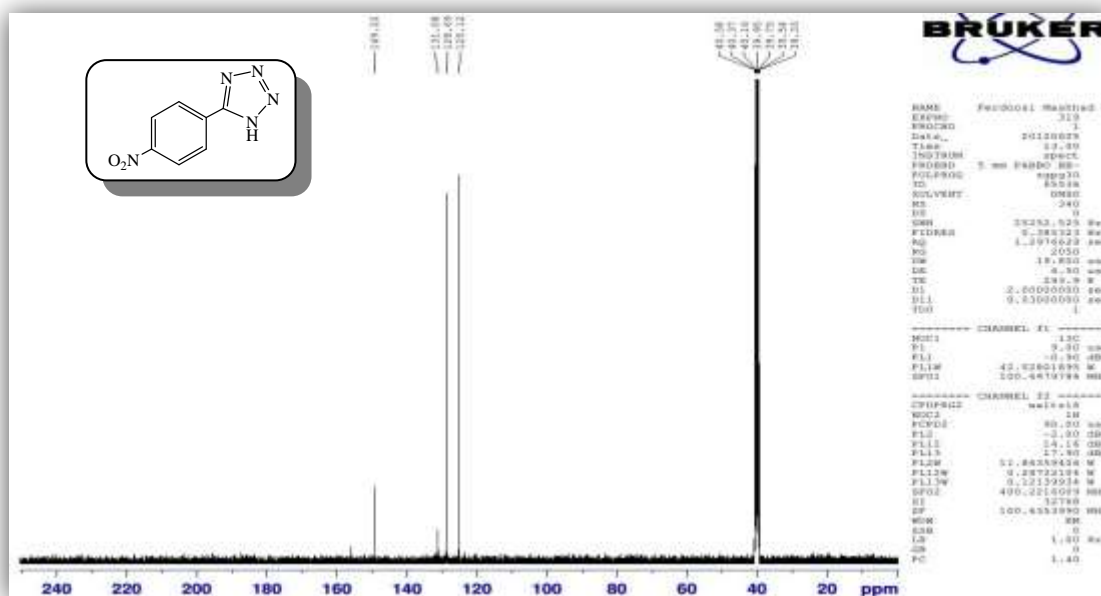


Figure 30: ^1H NMR (400 MHz, DMSO-d_6) of 5-(4-Nitrophenyl)-1H-tetrazole (Table 2, entry 5) expanded.



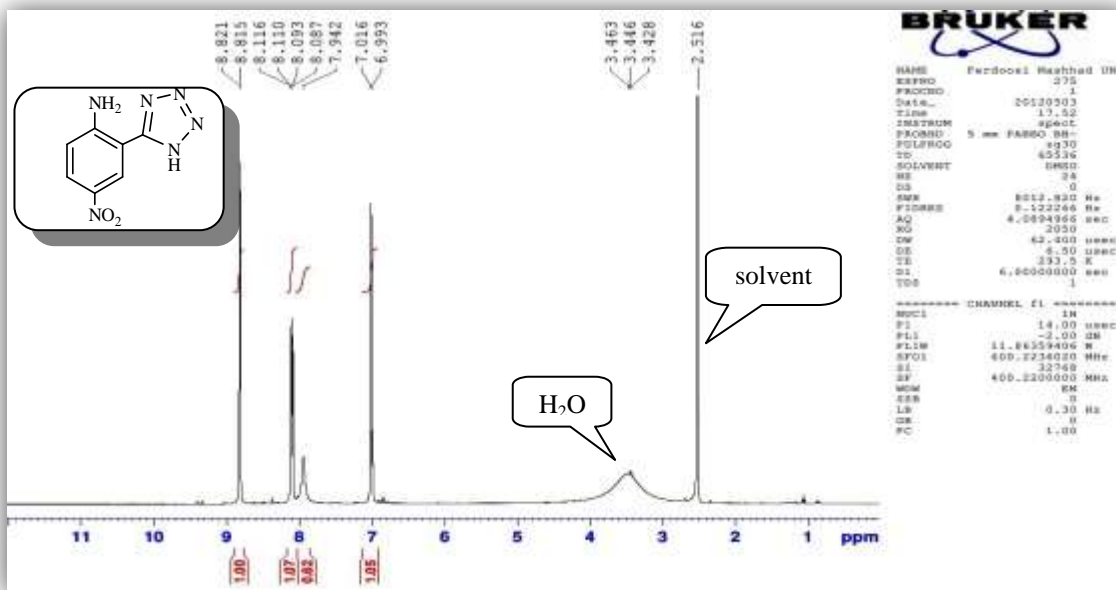


Figure 33: ^1H NMR (400 MHz, DMSO- d_6) of 4-Nitro-2-(1*H*-tetrazol-5-yl)benzenamine (Table 2, entry 6).

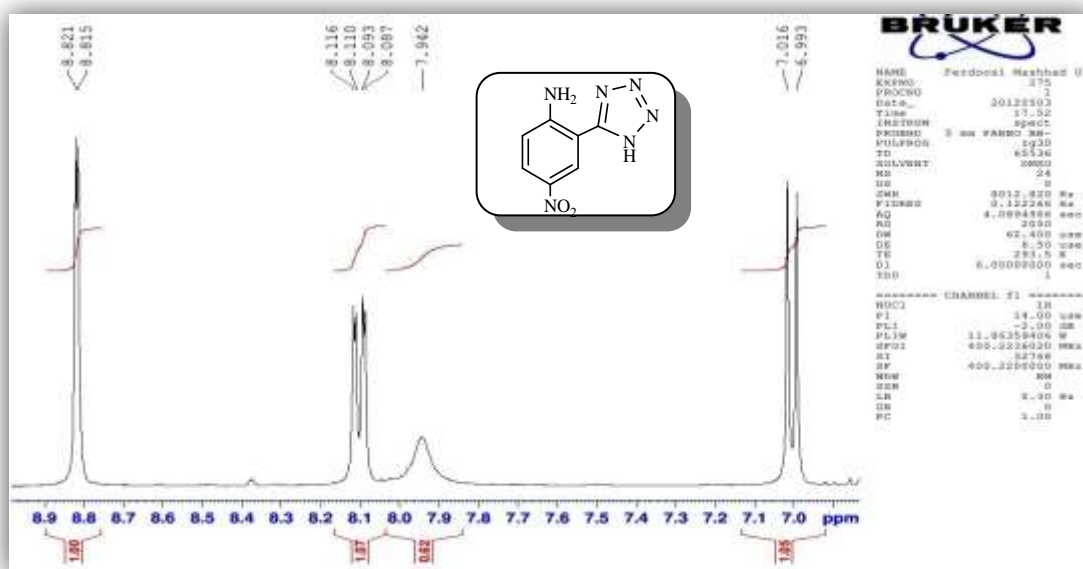


Figure 34: ^1H NMR (400 MHz, DMSO- d_6) of 4-Nitro-2-(1*H*-tetrazol-5-yl)benzenamine (Table 2, entry 6) expanded.

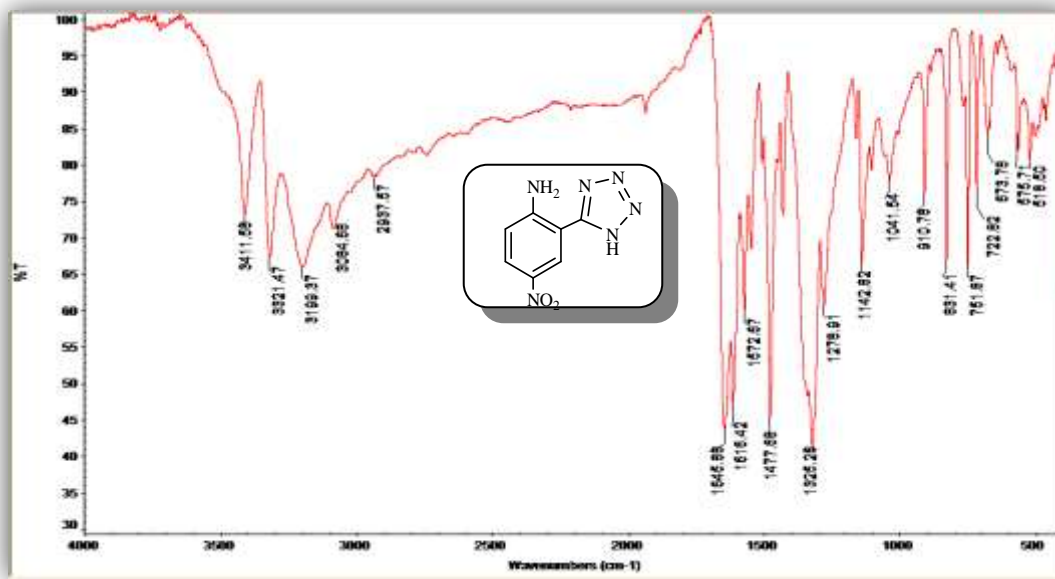


Figure 37: FT-IR (KBr) of 4-Nitro-2-(1*H*-tetrazol-5-yl)benzenamine (Table 2, entry 6).

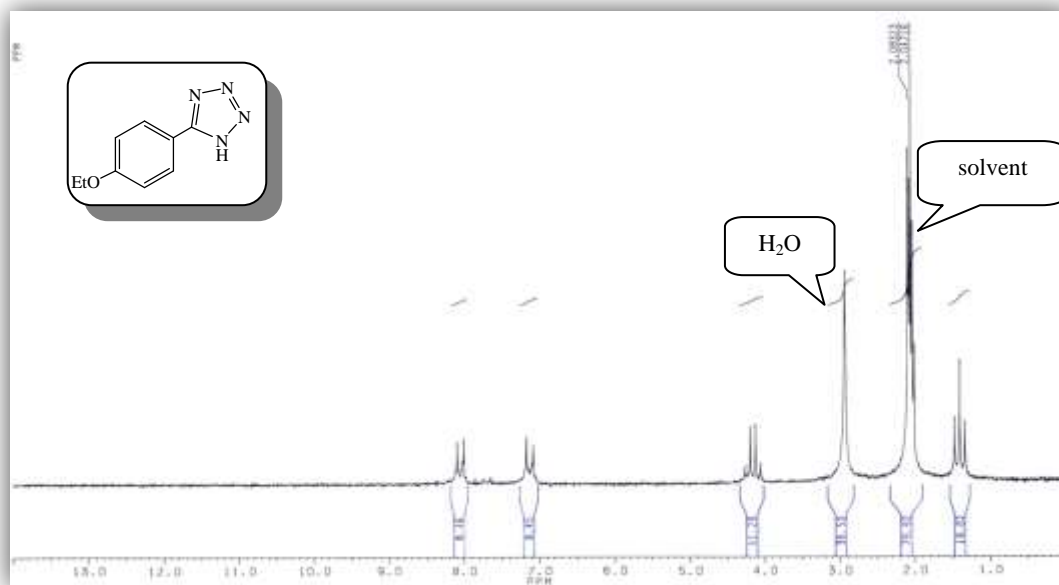


Figure 38: ^1H NMR (100 MHz, acetone- d_6) of 5-(4-Ethoxyphenyl)-1*H*-tetrazole (Table 2, entry 7).

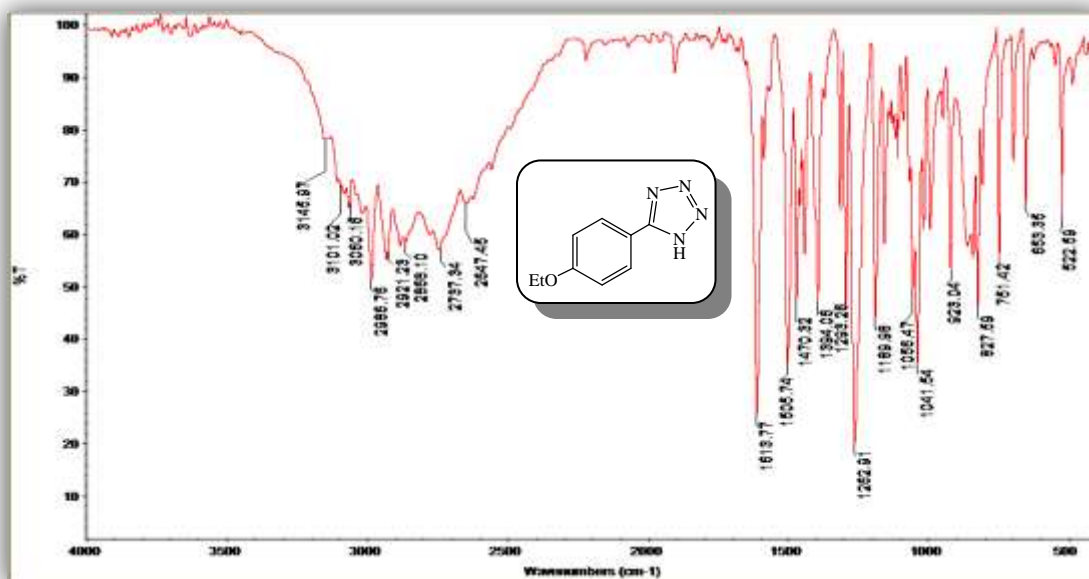


Figure 39: FT-IR (KBr) of 5-(4-Ethoxyphenyl)-1*H*-tetrazole (Table 2, entry 7).

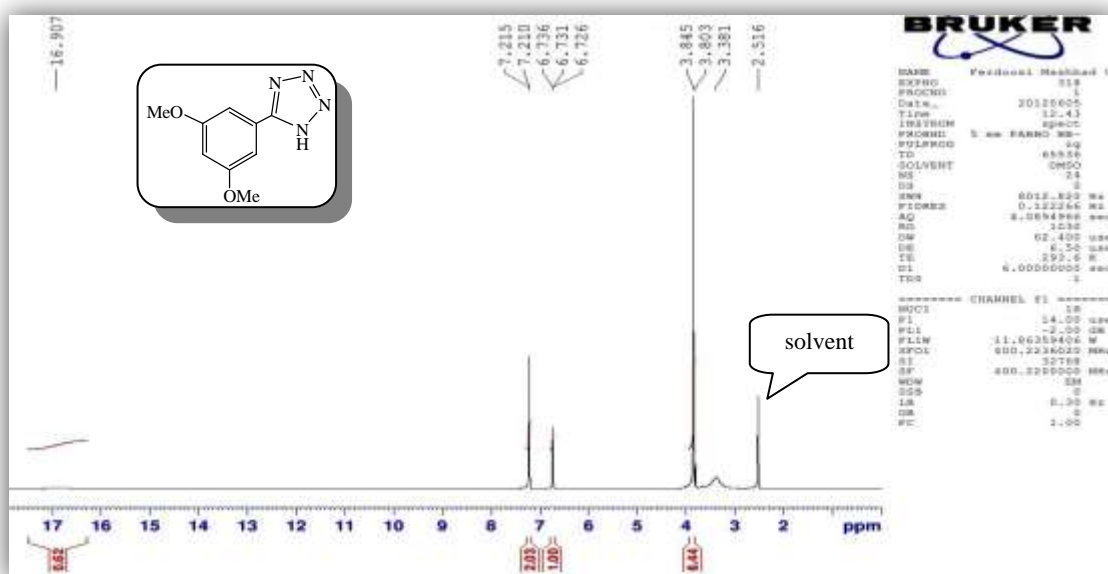


Figure 40: ^1H NMR (400 MHz, DMSO-d_6) of 5-(3,5-Dimethoxyphenyl)-1*H*-tetrazole (Table 2, entry 8).

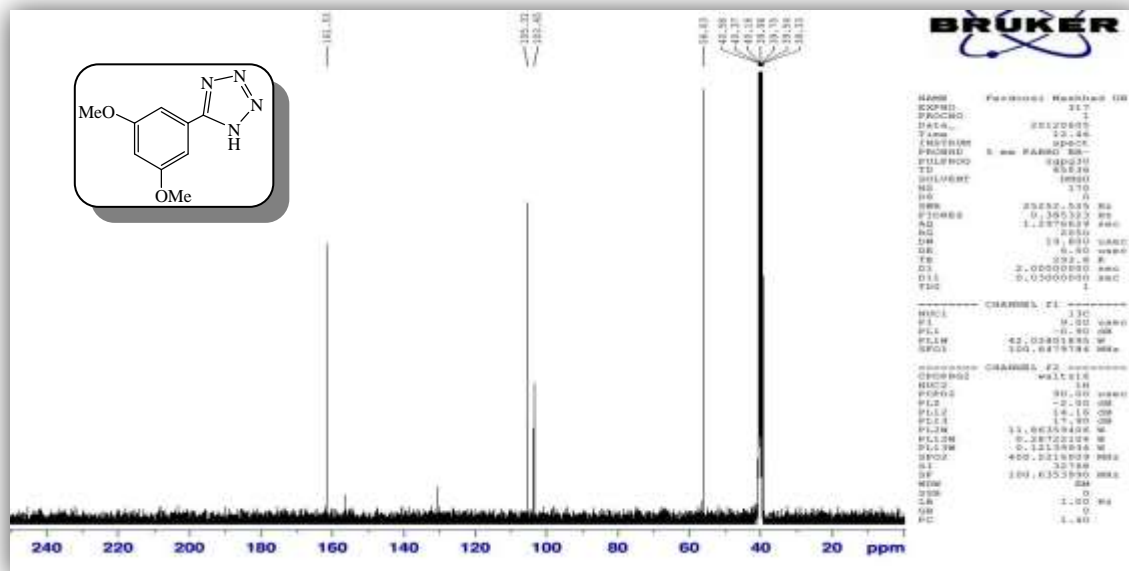


Figure 41: ^{13}C NMR (100 MHz, DMSO- d_6) of 5-(3,5-Dimethoxyphenyl)-1H-tetrazole (Table 2, entry 8).

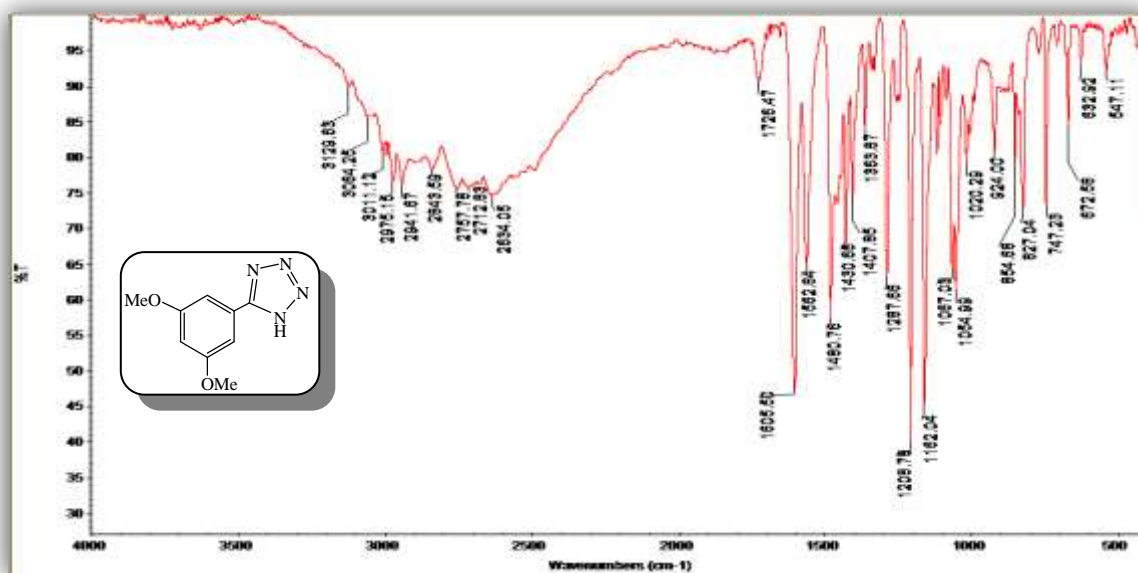


Figure 42: FT-IR (KBr) of 5-(3,5-Dimethoxyphenyl)-1H-tetrazole (Table 2, entry 8).

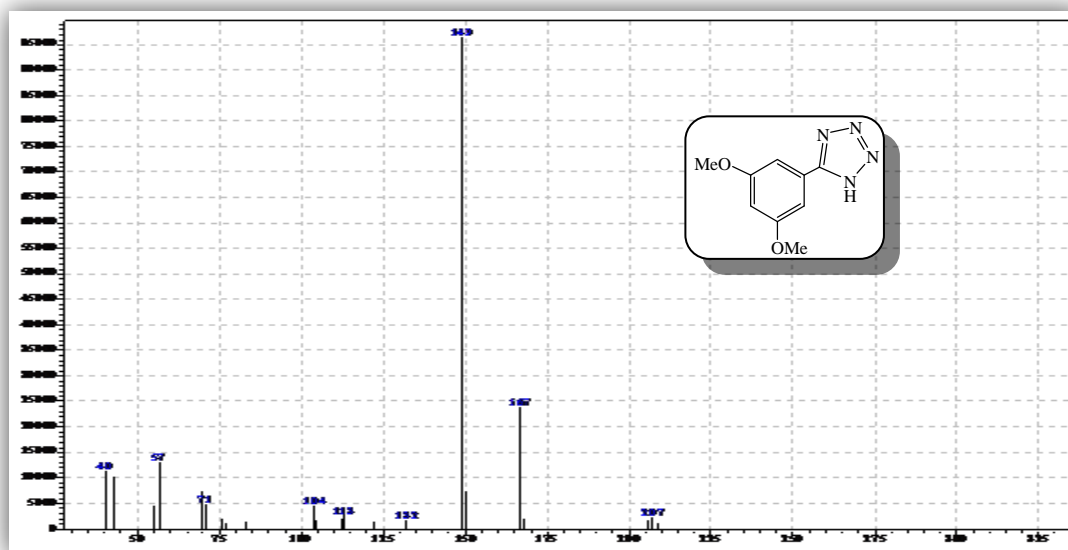


Figure 43: Mass spectrum of 5-(3,5-Dimethoxyphenyl)-1H-tetrazole (Table 2, entry 8).

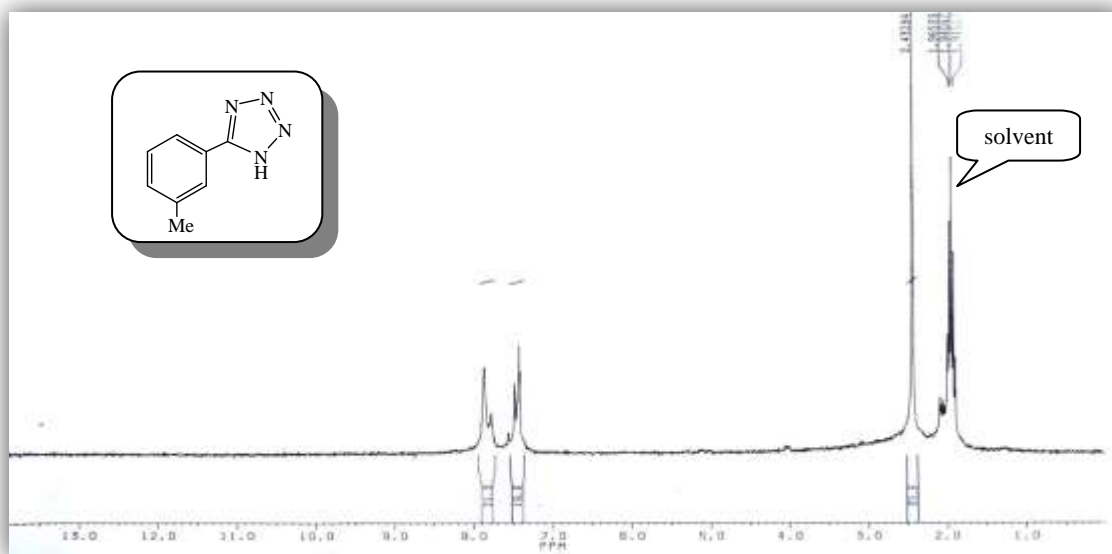


Figure 44: ¹H NMR (100 MHz, CD₃CN) of 5-m-Tolyl-1H-tetrazole (Table 2, entry 9).

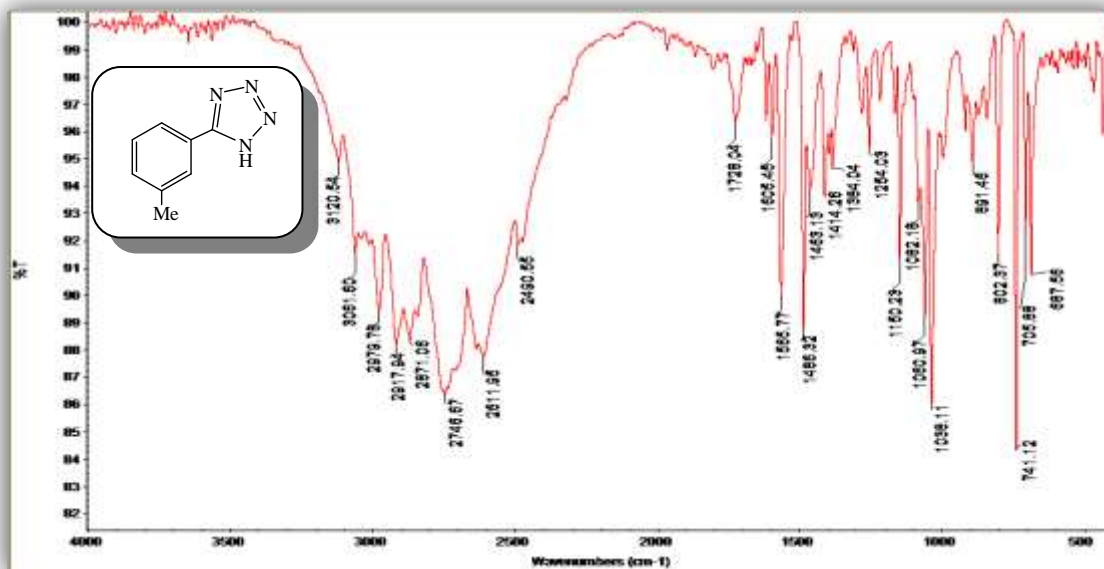


Figure 45: FT-IR (KBr) of 5-*m*-Tolyl-1*H*-tetrazole (Table 2, entry 9).

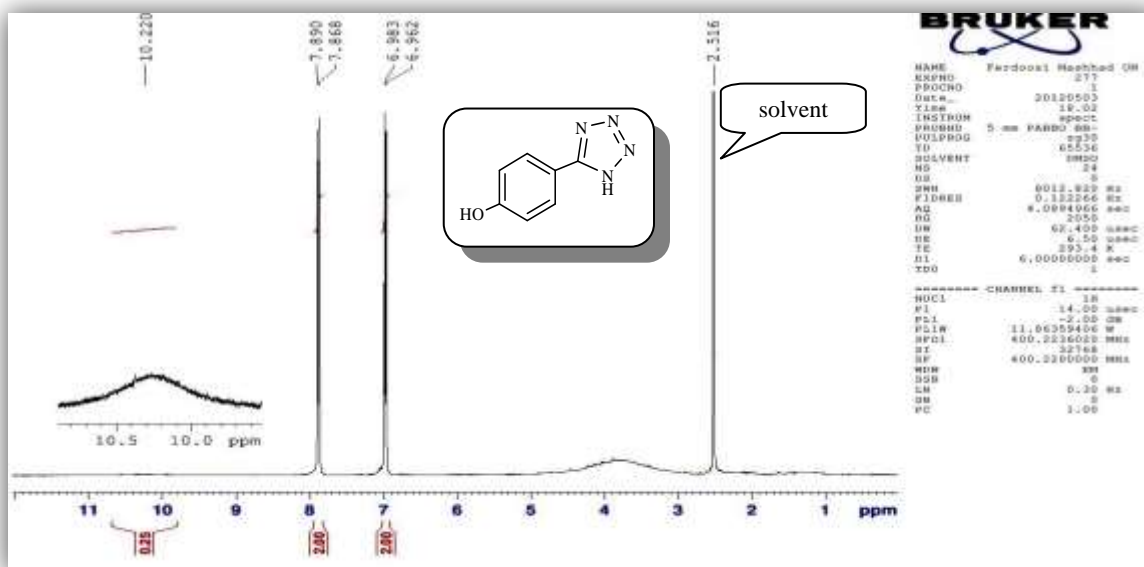


Figure 46: ^1H NMR (400 MHz, $\text{DMSO}-d_6$) of 4-(1*H*-tetrazol-5-yl) phenol (Table 2, entry 10).

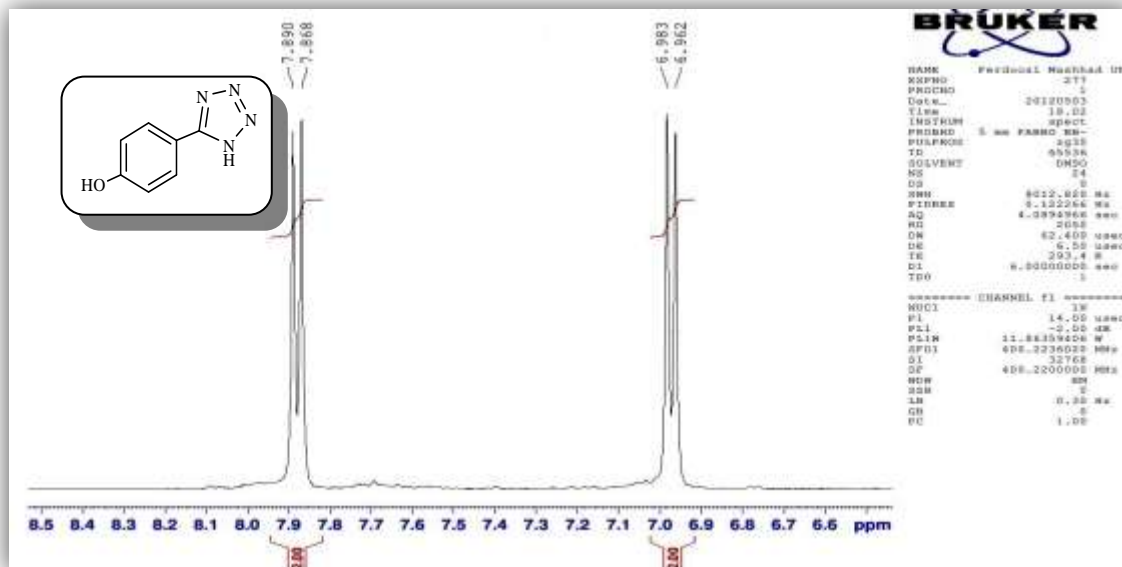


Figure 47: ^1H NMR (400 MHz, DMSO-d_6) of 4-(1*H*-tetrazol-5-yl)phenol (Table 2, entry 10) expanded.

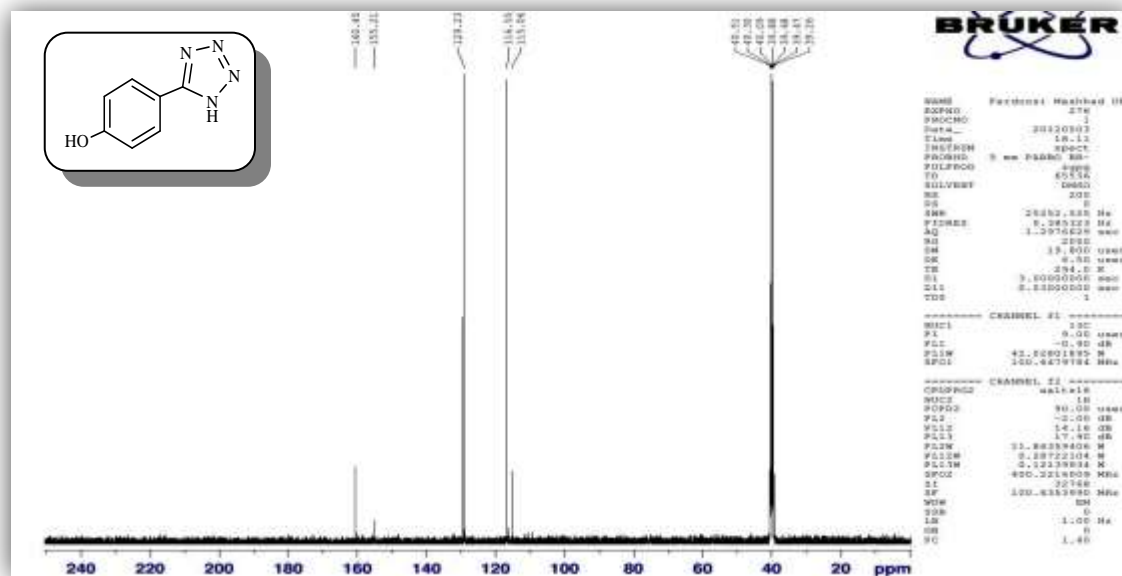


Figure 48: ^{13}C NMR (100 MHz, DMSO-d_6) of 4-(1*H*-tetrazol-5-yl)phenol (Table 2, entry 10).

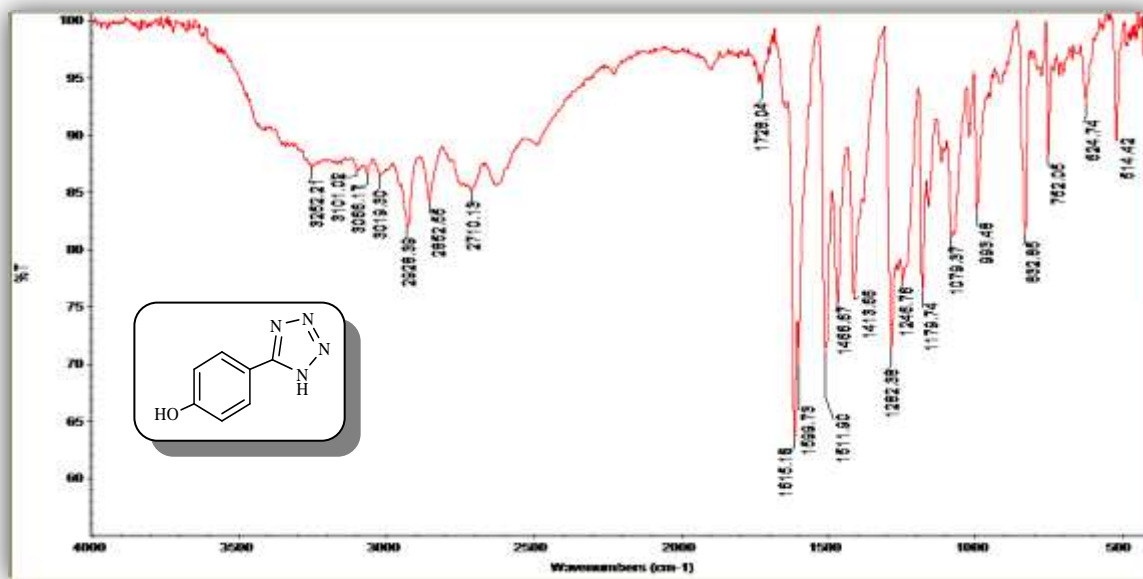


Figure 49: FT-IR (KBr) of 4-(1*H*-tetrazol-5-yl) phenol (Table 2, entry 10).

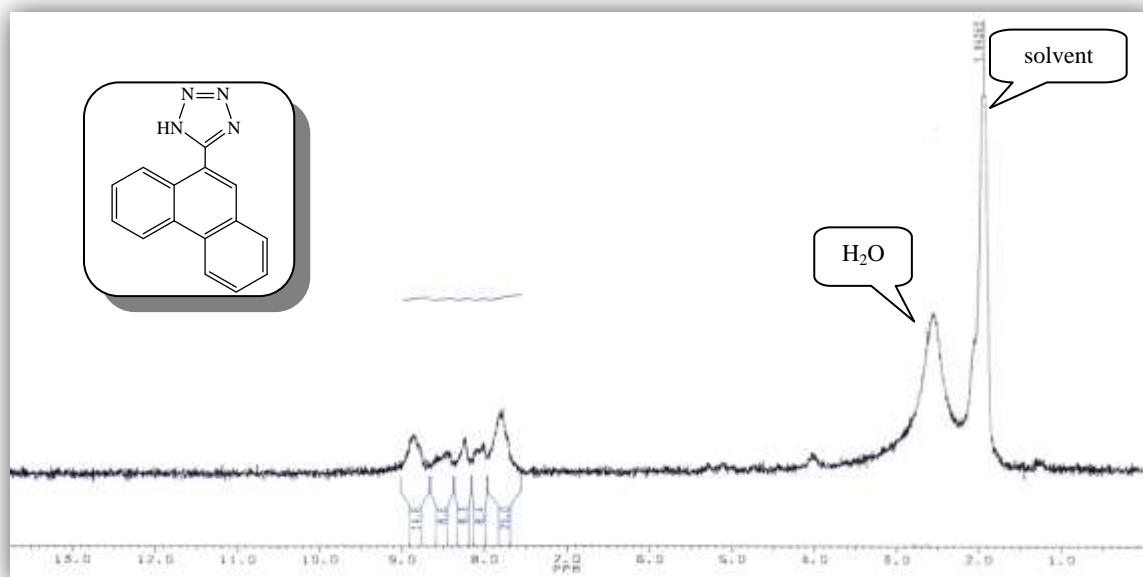


Figure 50: ¹H NMR (100 MHz, DMSO-*d*₆) of 5-(Phenanthren-9-yl)-1*H*-tetrazole (Table 2, entry 11).

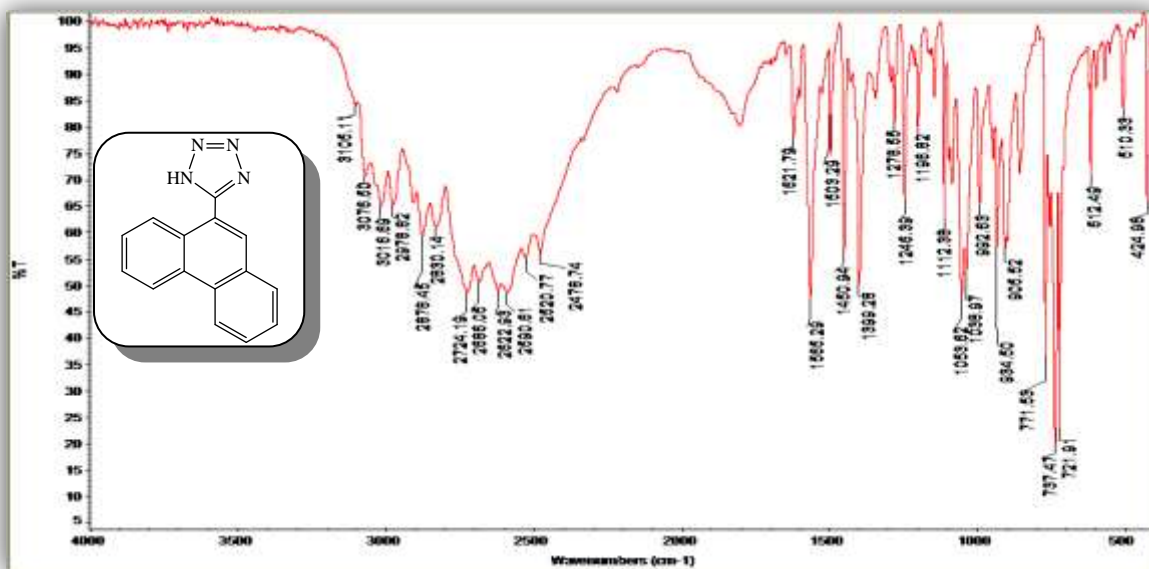


Figure 51: FT-IR (KBr) of 5-(Phenanthren-9-yl)-1H-tetrazole (Table 2, entry 11).

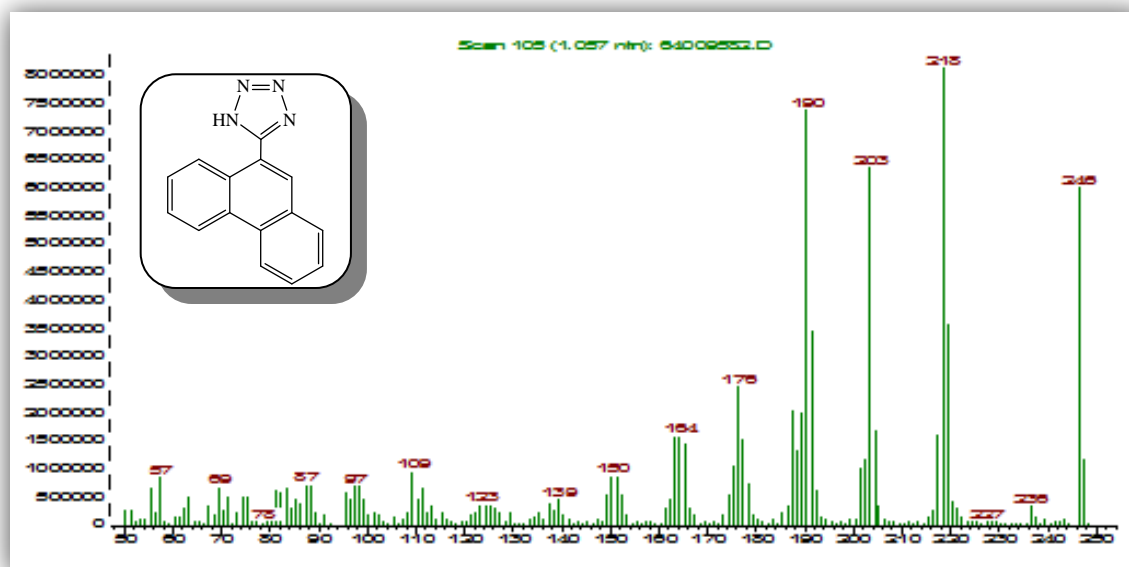


Figure 52: Mass spectrum of 5-(Phenanthren-9-yl)-1H-tetrazole (Table 2, entry 11).

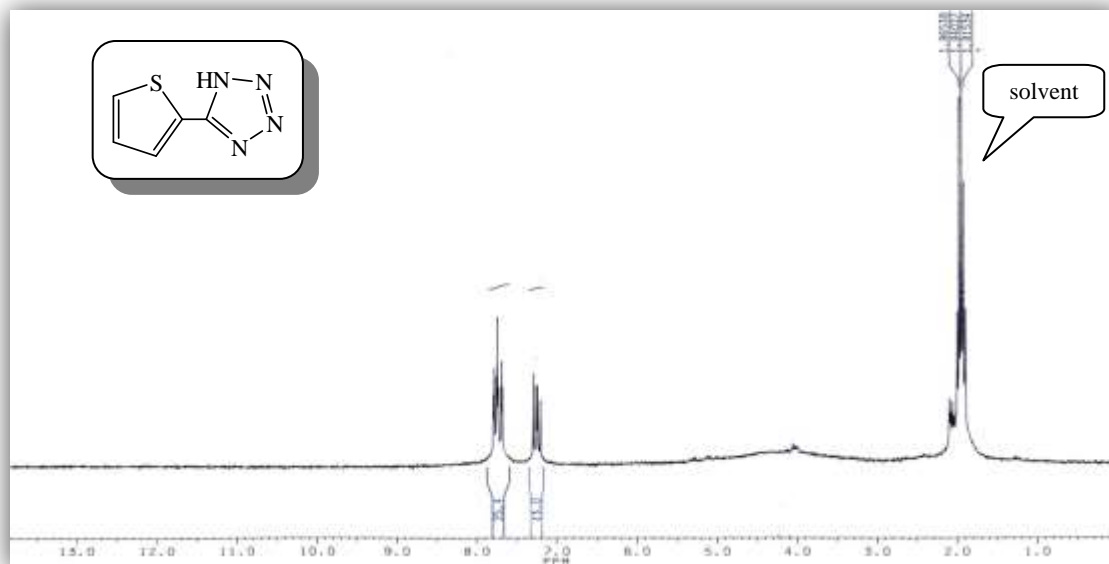


Figure 53: ^1H NMR (100 MHz, CD_3CN) of 5-(Thiophen-2-yl)-1*H*-tetrazole (Table 2, entry 12).

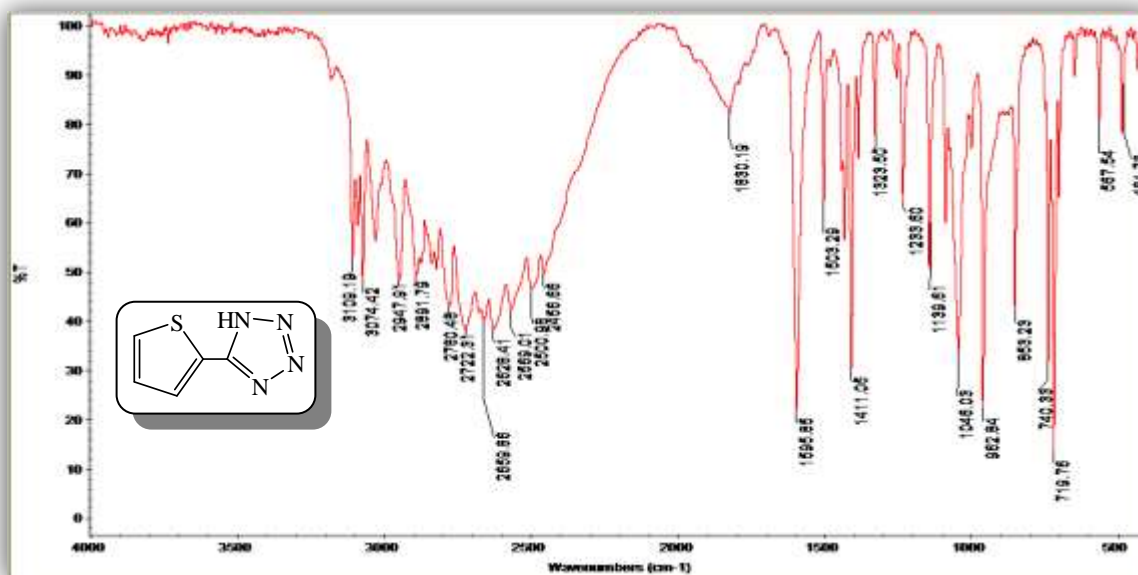


Figure 54: FT-IR (KBr) of 5-(Thiophen-2-yl)-1*H*-tetrazole (Table 2, entry 12).

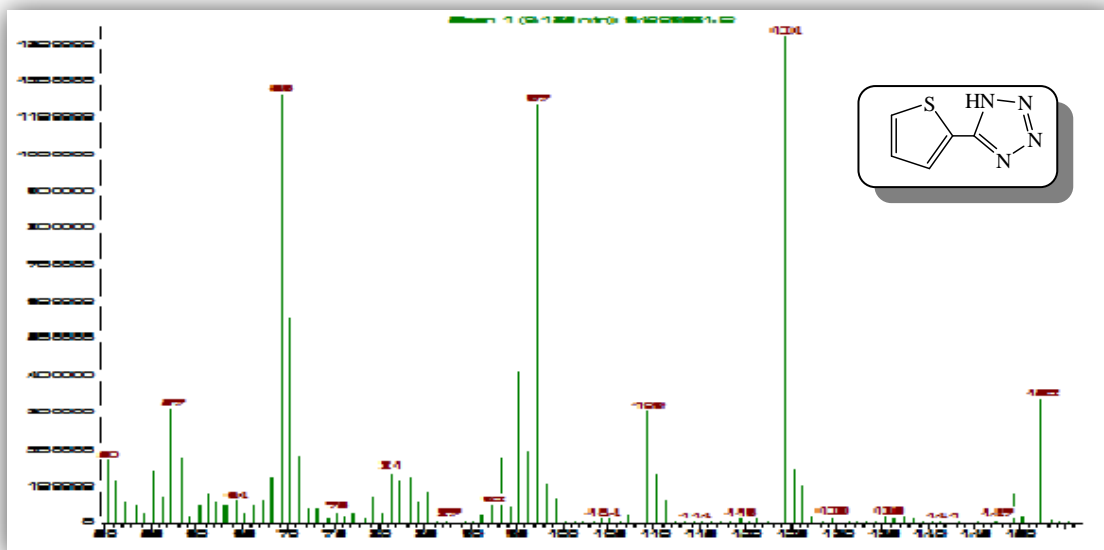


Figure 55: Mass spectrum of 5-(Thiophen-2-yl)-1H-tetrazole (Table 2, entry 12).

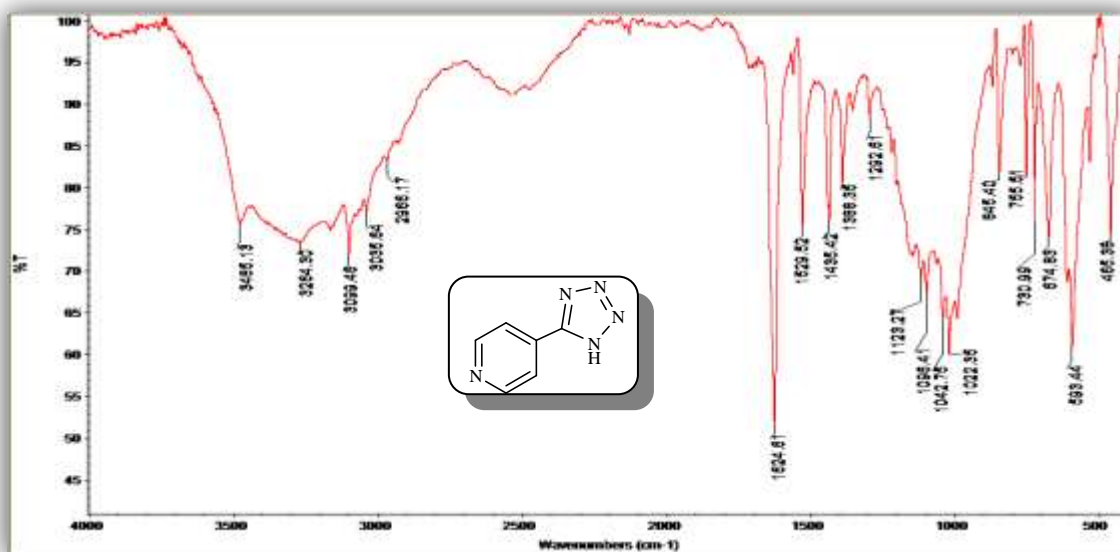


Figure 56: FT-IR (KBr) of 4-(1H-tetrazol-5-yl)pyridine (Table 2, entry 13).

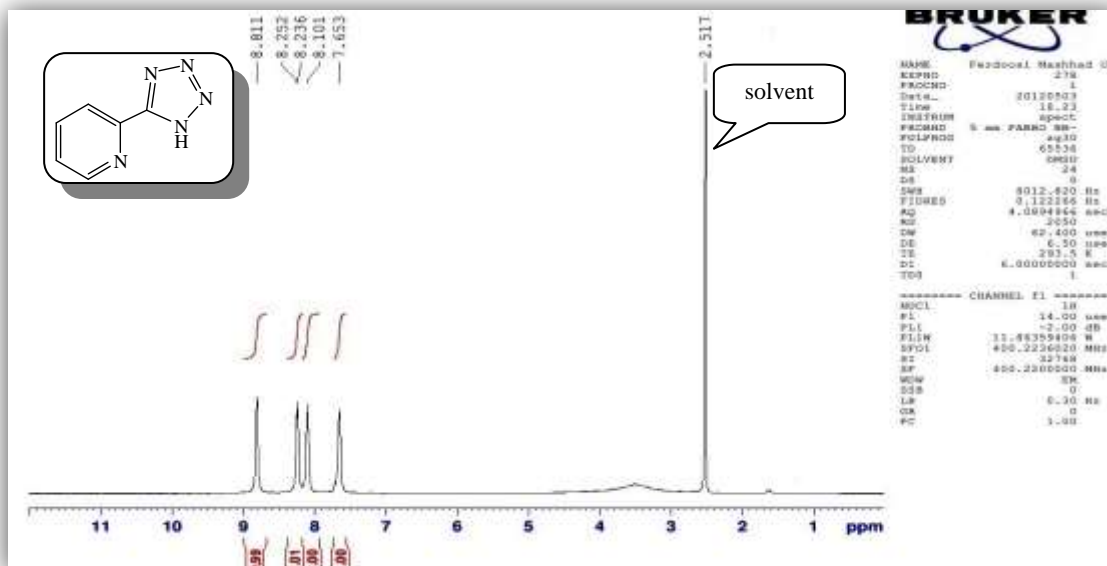


Figure 57: ¹H NMR (400 MHz, DMSO-d₆) of 2-(1*H*-tetrazol-5-yl)pyridine (Table 2, entry 14).

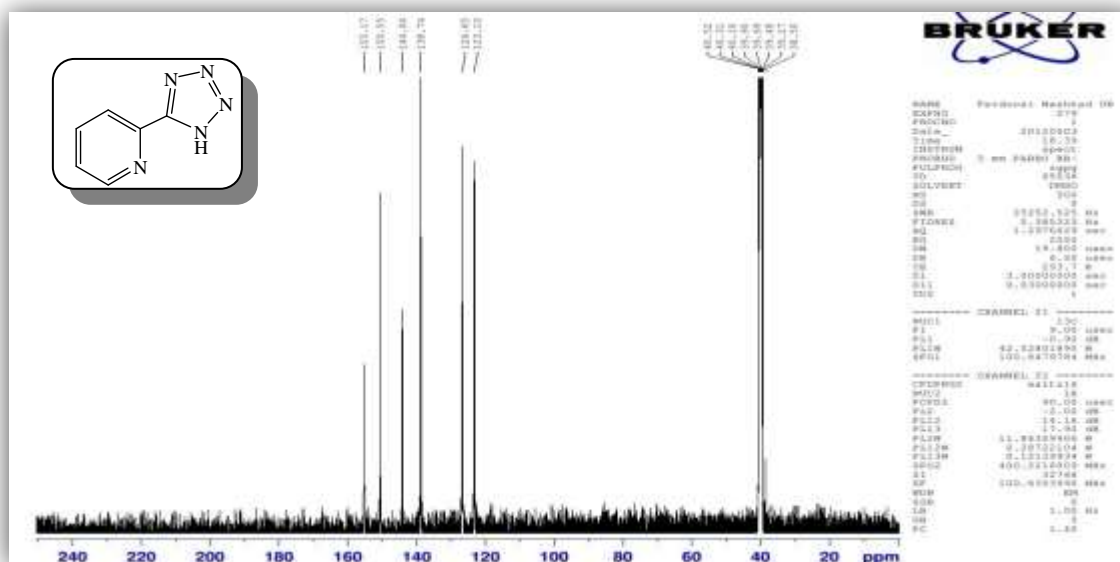


Figure 58: ¹³C NMR (100 MHz, DMSO-d₆) of 2-(1*H*-tetrazol-5-yl) pyridine (Table 2, entry 14).

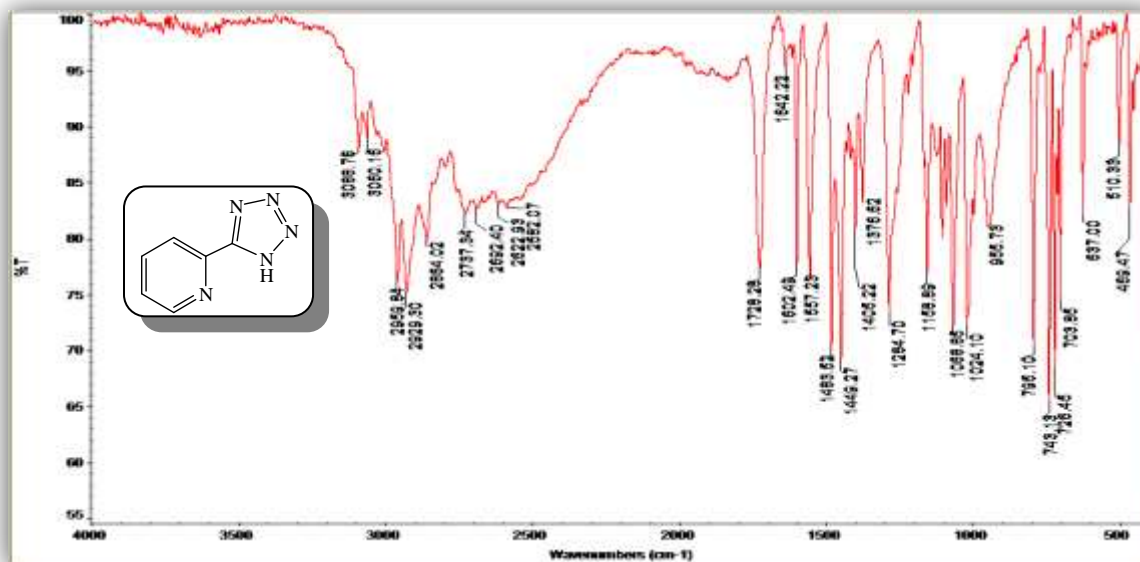


Figure 59: FT-IR of 2-(1*H*-tetrazol-5-yl) pyridine (Table 2, entry 14).

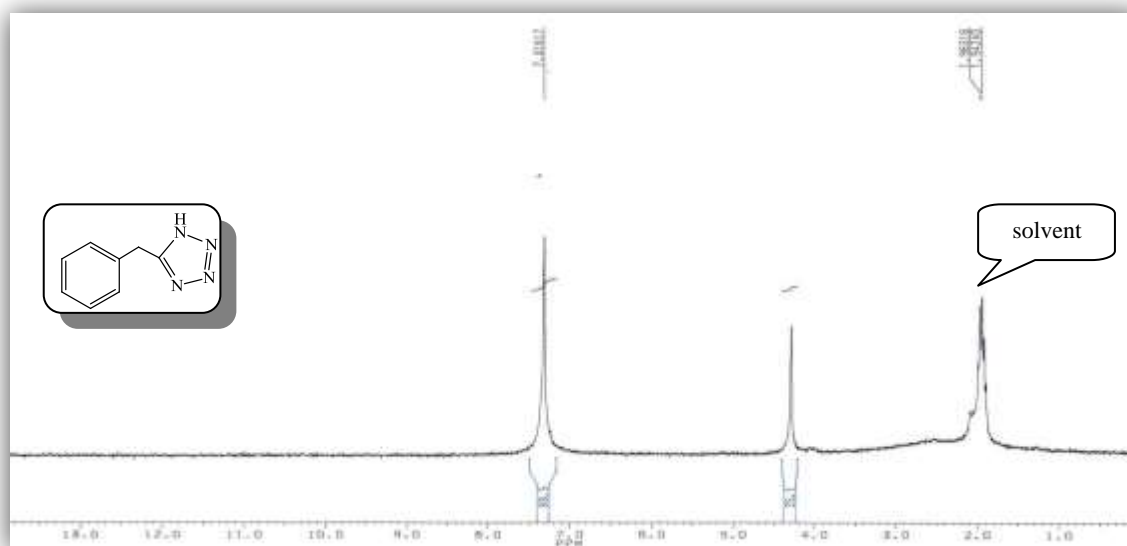


Figure 60: ^1H NMR (100 MHz, CD_3CN) of 5-Benzyl-1*H*-tetrazole (Table 2, entry 15).

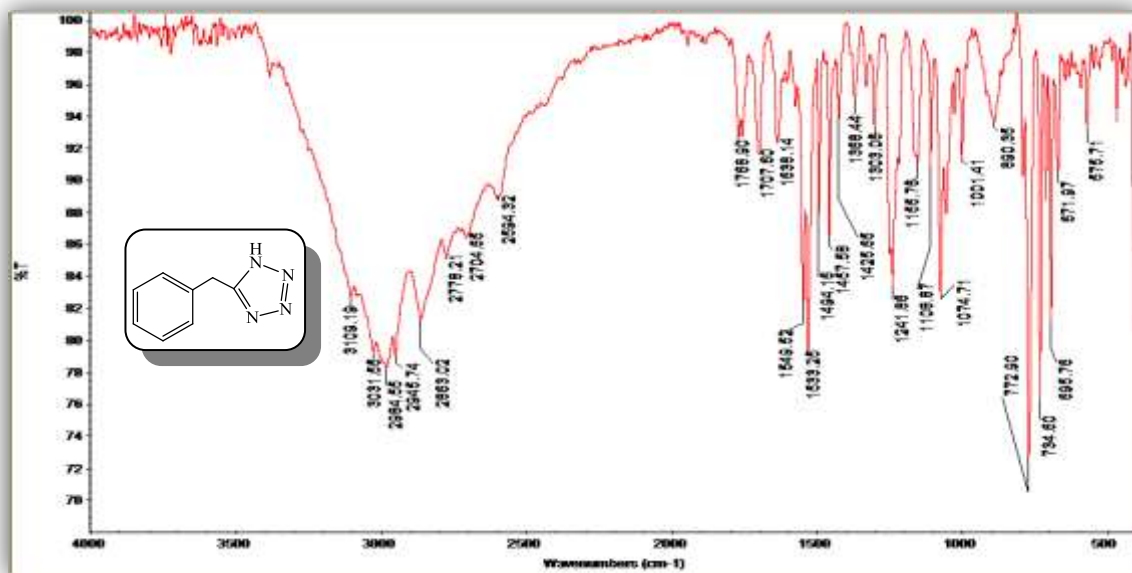


Figure 61: FT-IR of 5-Benzyl-1*H*-tetrazole (Table 2, entry 15).

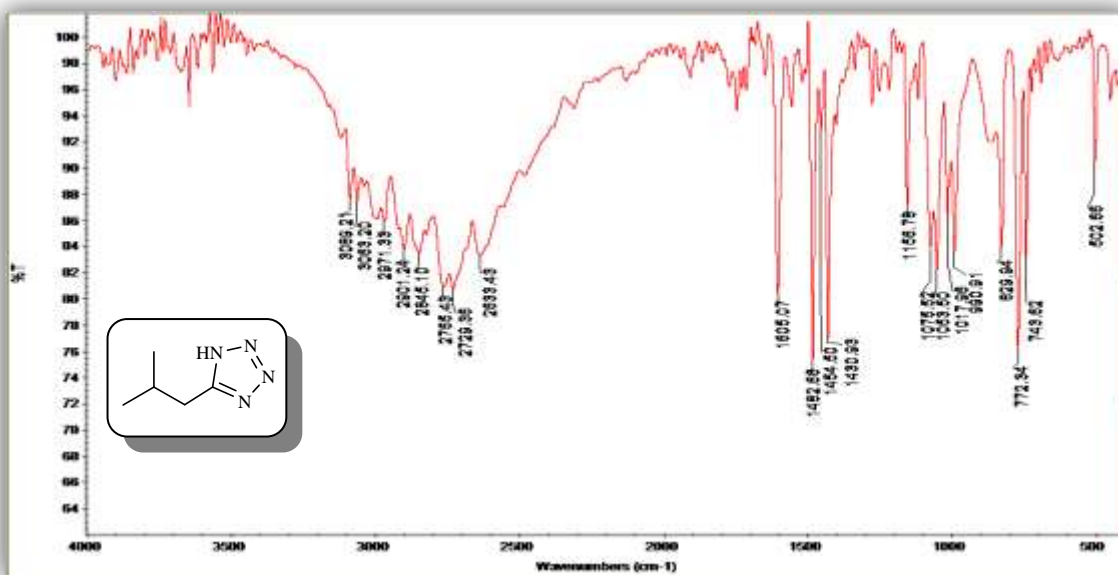


Figure 62: FT-IR of 5-Isobutyl-1*H*-tetrazole (Table 2, entry 16).

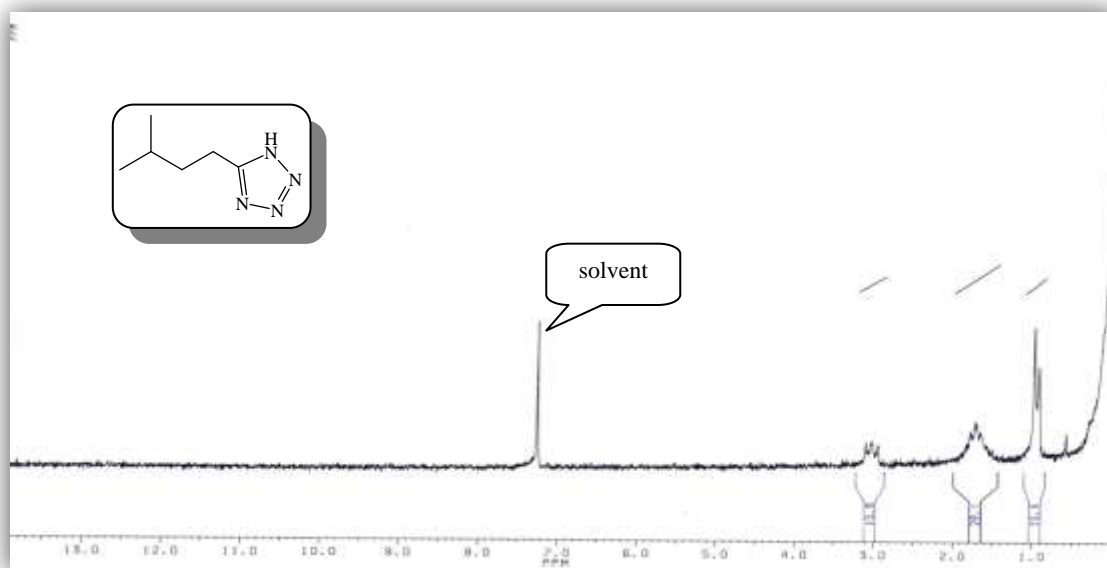


Figure 63: ¹H NMR (100 MHz, CDCl₃) of 5-Isopentyl-1*H*-tetrazole (Table 2, entry 17).

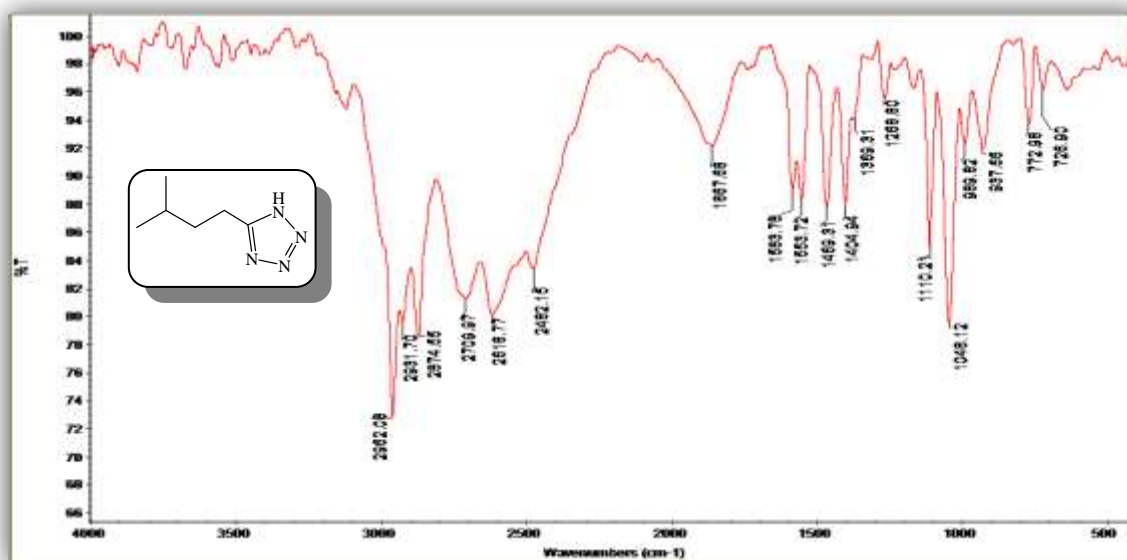


Figure 64: FT-IR of 5-Isopentyl-1*H*-tetrazole (Table 2, entry 17).