## Oxidation of SO<sub>2</sub> and NO by epoxy groups on graphene oxides: The role of the hydroxyl group

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Figure S1 Topviews of MEPs for  $SO_2$  oxidation on (a) HO\_OGP1 and (b) HO\_OGP2. Since the FS1 is energetic favorable to FS2 by 0.03 eV, only the FS1 is reported as the final state in Figure 2 in the manuscript.



Figure S2 Topviews of MEPs for NO oxidation on (a) HO\_OGP1 and (b) HO\_OGP2.



Figure S3 Top and side views of an alternative MEP for NO oxidation on HO\_OGP2. The charge difference plot of the IS is shown in Figure 6d in the manuscript. The adsorption energies for IS3 and FS3 are -0.14 and -0.64 eV, respectively. The oxidation barrier is 0.13 eV. For the MEP for NO oxidation on HO\_OGP2 discussed in Figure 4 and Figure S3, the adsorption energies of IS2 and FS2 are -0.25 (see Table 1 in the manuscript) and -0.80 eV (see Table S2), respectively. Subsequently, only the more stable configurations and the oxidation process related are reported in the manuscript.



Figure S4 Orbit-projected density of states (PDOS) of adsorbed NO molecule. Each of the  $\sigma$  orbits is filled with 2e. The  $1\pi$  is filled with 4e. Because the  $2\pi$  is filled with 1e, three fourth of which is unoccupied, it is located at the Fermi level.



Figure S5 Topviews of MEPs for CO oxidation on (a) HO\_OGP1 and (b) HO\_OGP2. Since the FS1 is energetic favorable to FS2 by 0.1 eV, only the FS1 is reported as the final state in Figure 3 in the manuscript.

| Entries -                | non-spin polarized |            | spin polarized |            |
|--------------------------|--------------------|------------|----------------|------------|
|                          | D2                 | optB88-vdW | D2             | $m(\mu_B)$ |
| SO <sub>2</sub> /OGP     | -0.30              | -0.46      | -0.30          | -0.0003    |
| SO <sub>2</sub> /HO_OGP1 | -0.40              | -0.55      | -0.40          | +0.0084    |
| SO <sub>2</sub> /HO_OGP2 | -0.45              | -0.61      | -0.45          | +0.0000    |
| SO <sub>2</sub> /2HO_OGP | -0.58              | -0.72      | -0.58          | -0.0001    |
| NO/OGP                   | -0.10              | /          | -0.10          | +1.0053    |
| NO/HO_OGP1               | -0.54              | /          | -0.29          | +0.2030    |
| NO/HO_OGP2               | -0.25              | /          | -0.20          | +0.9714    |

Table S1 Adsorption energy (in eV) recheck for optB88-vdW and spin polarization.

Table S2 Oxidation barrier (in eV) recheck for spin polarization.

| Reactions                                | non-spin polarized | spin polarized |
|------------------------------------------|--------------------|----------------|
| SO <sub>2</sub> /OGP→SO <sub>3</sub> /GP | -0.21              | -0.207         |
| NO/HO_OGP1→NO <sub>2</sub> /HO_GP        | -0.11              | -0.119         |

Table S3 Adsorption energy and charge transfer for the adsorption of SO<sub>3</sub>, NO<sub>2</sub> and CO<sub>2</sub> on GP,

| Conf.                   | $\Delta E_{\rm ads}$ , eV | $\Delta q$ , e |
|-------------------------|---------------------------|----------------|
| SO <sub>3</sub> /GP     | -0.31                     | 0.080          |
| SO <sub>3</sub> /HO_GP  | -0.46                     | 0.290          |
| SO <sub>3</sub> /2HO_GP | -0.51                     | 0.400          |
| NO <sub>2</sub> /GP     | -0.49                     | 0.344          |
| NO <sub>2</sub> /HO_GP  | -0.80                     | 0.407          |
| CO <sub>2</sub> /GP     | -0.15                     | 0.015          |
| CO <sub>2</sub> /HO_GP  | -0.20                     | 0.010          |

HO\_GP and 2HO\_GP surfaces. Only the values of the most stable configurations are collected.