Electronic supplementary information

Enhanced photocatalytic activities of visible-light driven green synthesis in water and

environmental remediation on Au/Bi₂WO₆ hybrid nanostructures

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Fig. S1A Energy-dispersive X-ray (EDX) analysis of 3.0% Au/Bi₂WO₆ hybrid nanoparticles; **S1B** Effect of the added Au³⁺/Bi₂WO₆ weight ratio on the obtained Au/Bi₂WO₆ weight ratio in the nanostructures measured by EDX analysis: Au/Bi₂WO₆ ratio from EDX *vs* Au³⁺/Bi₂WO₆ ratio.

Entry	Substrate	Product	Conversion (%)	Selectivity of aldehyde (%)
1	СН2ОН	СНО	65.4	82
2	H ₃ CO-CH ₂ OH	Н₃СО-√СНО	69.5	80
3	H ₃ C-CH ₂ OH	Н ₃ С-СНО	62.7	78
4	O ₂ N-CH ₂ OH	О2N-СНО	64.5	71
5	CI-CH ₂ OH	сі— Сно	56.6	67

Table S1 Photocatalytic selective oxidation of various benzylic alcohols over 2.0% Au/Bi₂WO₆ in waterunder visible light irradiation for 8 h.



Fig. S2 Electron spin resonance (ESR) spectra of superoxide radicals trapped by DMPO over 2.0% Au/Bi₂WO₆ photocatalyst as a function of irradiation time in methanol solution.



Fig. S3 Mott-Schottky plot for Bi₂WO₆ nanosheets in 0.5 mol/L Na₂SO₄ aqueous solution (pH=6.8).



Fig. S4 XRD patterns of fresh 1.0% Au/Bi₂WO₆ and used 1.0% Au/Bi₂WO₆ sample after five cycles of photocatalytic reduction of Cr(VI).



Fig. S5 The influence of incident light intensity (A) and light wavelength range (B) on the photocatalytic reduction of Cr(VI) over 1.0% Au/Bi₂WO₆ hybrid nanostructures.