Electronic Supplementary Information

Impact of Linker in Polypyrrole/Quinone Conducting Redox Polymers

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Figure S1. UV/vis/NIR spectra during an in situ oxidation sweep of a) **P4** and c) **P5**. Difference spectra over the pendant group redox processes for b) **P4** and d) **P5**. The energy scale is the same in all plots.



Figure S2. Electropolymerization CVs of P3 - P5 on EQCM Au electrodes, and the concurrent mass. The potential scale is the same in all plots.



Figure S3. Mass changes during EQCM CV oxidation sweeps of P3 in various electrolytes.



Figure S4. Mass change vs number of charges in CV of P3 in various electrolytes.



Figure S5. Mass changes during EQCM CV oxidation sweeps of P4 in various electrolytes.



Figure S6. Mass change vs number of charges in CV of P4 in various electrolytes.



Figure S7. Mass changes during EQCM CV oxidation sweeps of P5 in various electrolytes.



Figure S8. Mass change vs number of charges in CV of P5 in various electrolytes.

		Q	Q	PPy	PPy	PPy	PPy
		n-π*	π-π*	4	3	2	1
РРу	OX			325	430		>700
РРу	red			328	412	590	
	ox Q	250 ^a	396 ^a	300 ^b	415		>700
P1	ox HQ	312 ^a	503 ^a	306	400	550 ^b	>700
	red			307	398	598	
	ox Q	250 ^a	417 ^a	306 ^b	400	622	(>700)
P2	ox HQ	315 ^a	~500 ^{a,b}	310	394	622	(>700)
	red			311	394	622	
	ox Q	250	323 ^a	293 ^b	420		645
P3	ox HQ	296 ^a	511 ^a	293	430	560 ^b	690
	red			293	390	560 ^b	
	ox Q	250 ^a	433 ^a	301	407		870
P4	ox HQ	356 ^a (295 ^a)	518 ^a	299	407		900
	red			299	387	570 ^b	
	ox Q	250 ^a	514 ^a				>700
P5	ox HQ	333 ^a	318	285		590 ^b	>700
	red			285	425 ^b	650 ^b	

Table S1. UV/vis/NIR absorbance maxima (nm) of the polymers in the reduced (red) and oxidized (ox) states and with reduced (HQ) and oxidized (Q) pendant groups.

^a From difference spectra, ^b Shoulder