

Facile synthesis of Fe₃O₄/hierarchical-Mn₃O₄/graphene oxide as a synergistic catalyst for activation of peroxymonosulfate for degradation of organic pollutants

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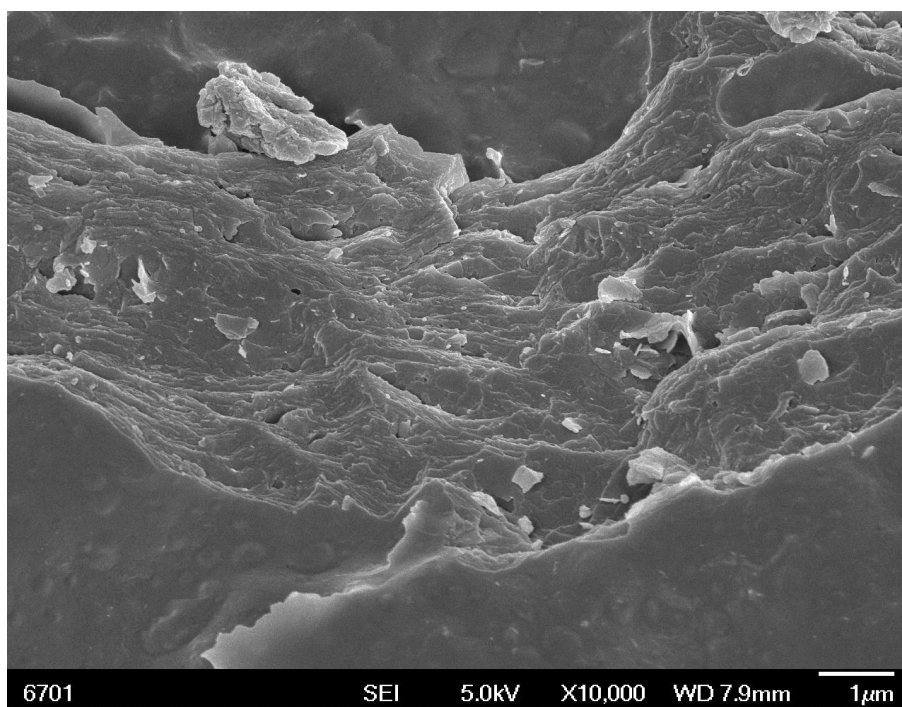


Fig. S1 SEM image of Mn₃O₄/rGO

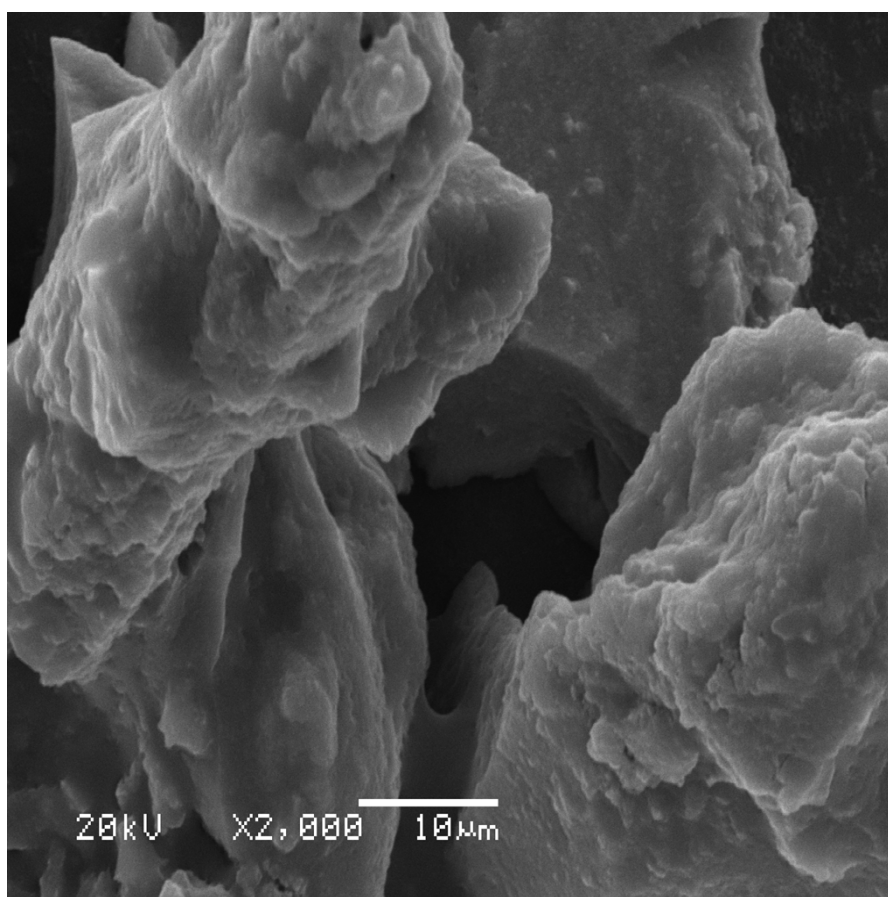


Fig. S2 SEM image of Fe₃O₄/Mn₃O₄/rGO

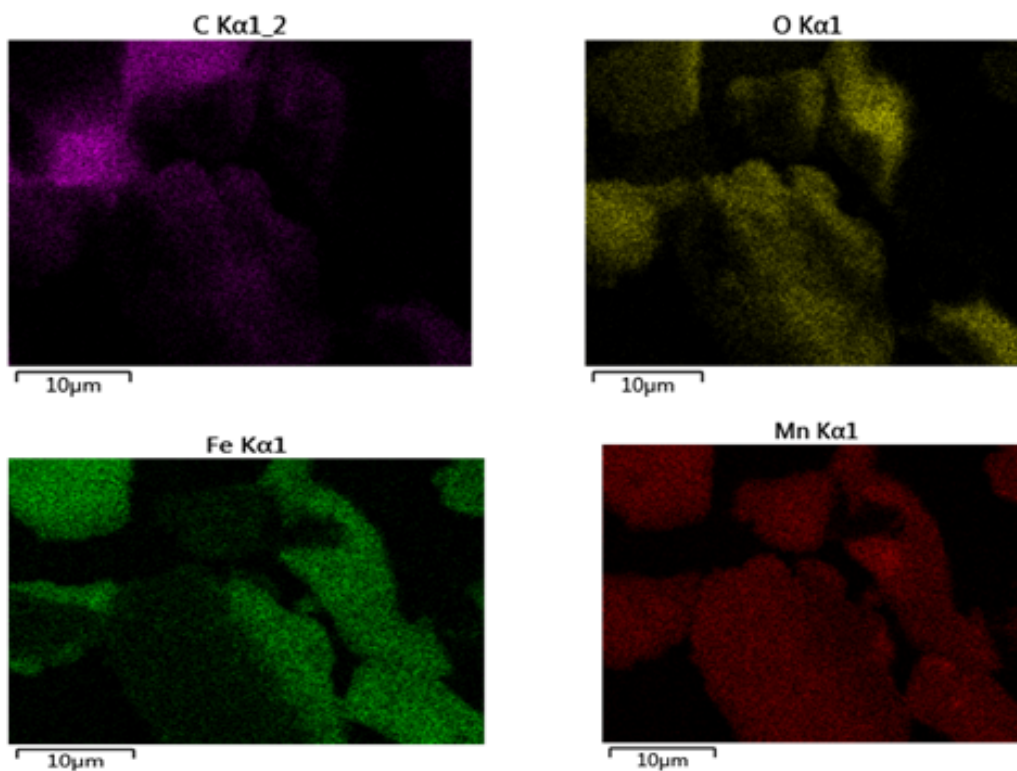


Fig. S3 Elemental distribution maps of C, O, Fe, and Mn of $\text{Fe}_3\text{O}_4/\text{Mn}_3\text{O}_4/\text{rGO}$ hybrids

Table S1

EDXS results of $\text{Fe}_3\text{O}_4/\text{Mn}_3\text{O}_4/\text{rGO}$ nanocomposite

Element	C	O	Mn	Fe	Total
Wt%	37.02	25.87	23.96	13.16	100
Mol%	57.39	30.11	8.12	4.39	100

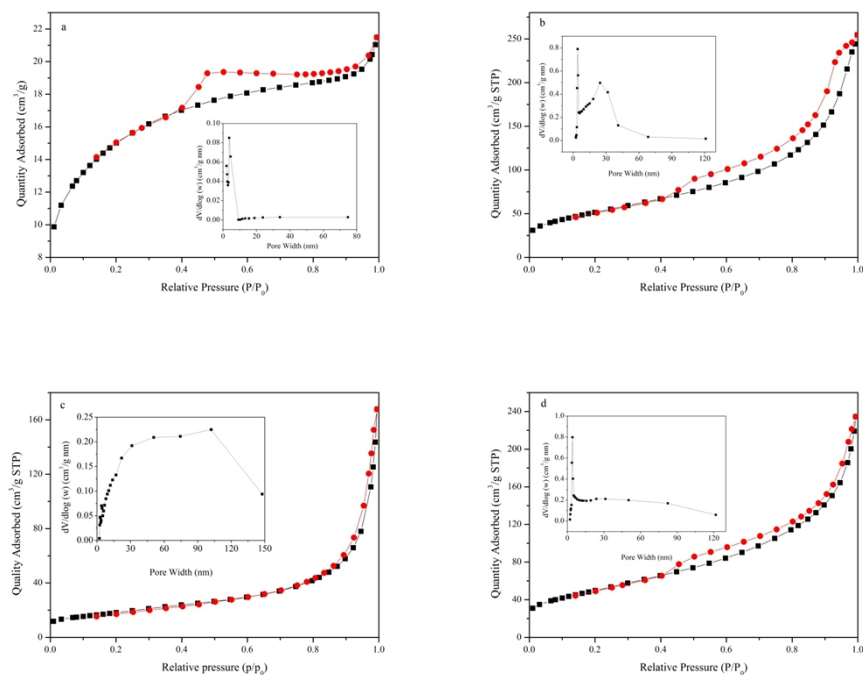


Fig. S4 Nitrogen adsorption-desorption isotherm and BJH pore size distribution (inset) of GO (a), Fe₃O₄/rGO (b), Mn₃O₄/rGO (c), Fe₃O₄/Mn₃O₄/rGO (d)

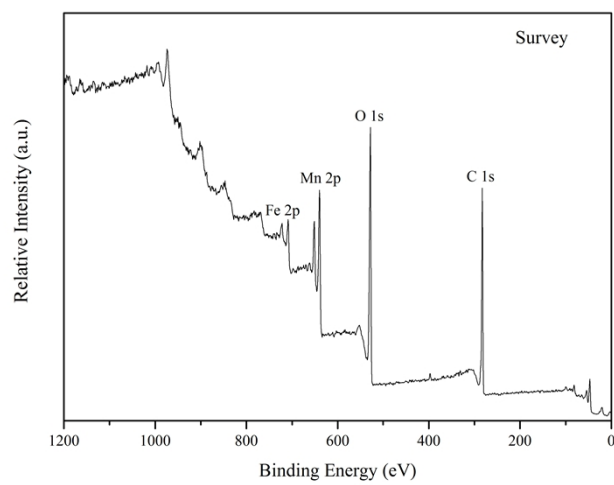


Fig. S5 XPS spectra of $\text{Fe}_3\text{O}_4/\text{Mn}_3\text{O}_4/\text{rGO}$ composites (Survey scan)

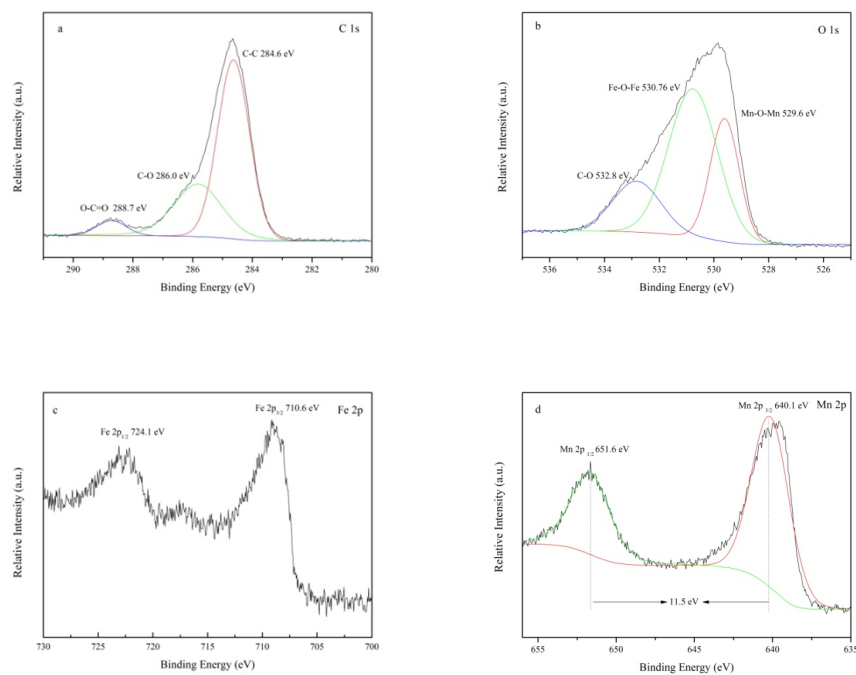


Fig. S6 XPS spectra of $\text{Fe}_3\text{O}_4/\text{Mn}_3\text{O}_4/\text{rGO}$, C 1s region (a), O 1s region (b), Fe2p region (c), and Mn 2p region (d)

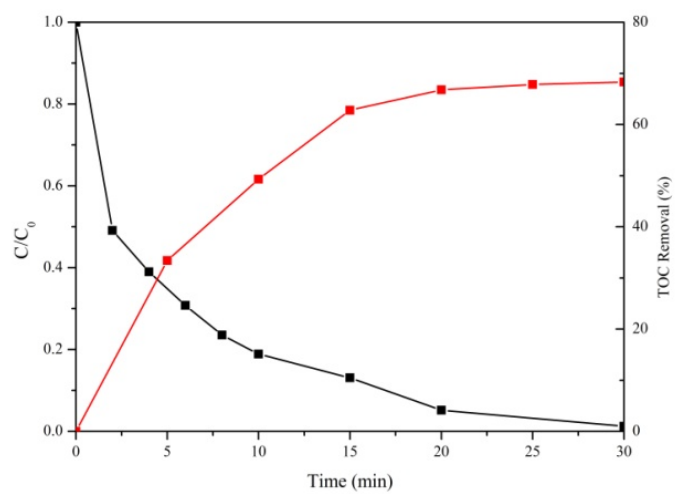


Fig. S7 discoloration and TOC removal during the degradation of MB, (Reaction conditions: catalyst = 100 mg/L, Oxone dose = 0.3 g/L, initial MB concentration = 50 mg/L, pH=7.0, T = 25 °C)