

A POCl₃-mediated synthesis of substituted fused azoacridones derivatives

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Synthesis of 2-(pyridin-2-ylamino)benzoic acid (**1b**)

2-chlorobenzoic acid (626 mg, 4.0 mmol) and 2-Aminopyrimidine (188 mg, 2.0 mmol) were dissolved in DMF (10 mL), and then K₂CO₃ (552 mg, 4.0 mmol) and Cu (64 mg, 1.0 mmol) were added. The mixture was stirred overnight at 130 °C, which was then cooled to room temperature. The mixture was pooled into ice water. Diluted hydrochloric acid was added and precipitate occurred, which was filtered, washed with water, and dried to give crude compound **1b**, which was used without further purification.

Synthesis of 2-((4-(phenylamino)pyrimidin-2-yl)amino)benzoic acid (**1e**)¹

2,4-dichloropyrimidines (290 mg, 2.0 mmol) and aniline (465 mg, 5.0 mmol) were added into H₂O (10 mL), then 6N HCl (0.5 mL) was added. The mixture was refluxed for 2-3 h, which was then cooled to room temperature and filtered. The solid was directly added into EtOH (10 mL), then anthranilic acid (301 mg, 2.2 mmol) and 6N HCl (0.5 mL) was added. The mixture was refluxed for 2-3 h, which was then cooled to room temperature and filtered, washed the precipitate with EtOH and dried to give white powder **1e** (412 mg, 67%).

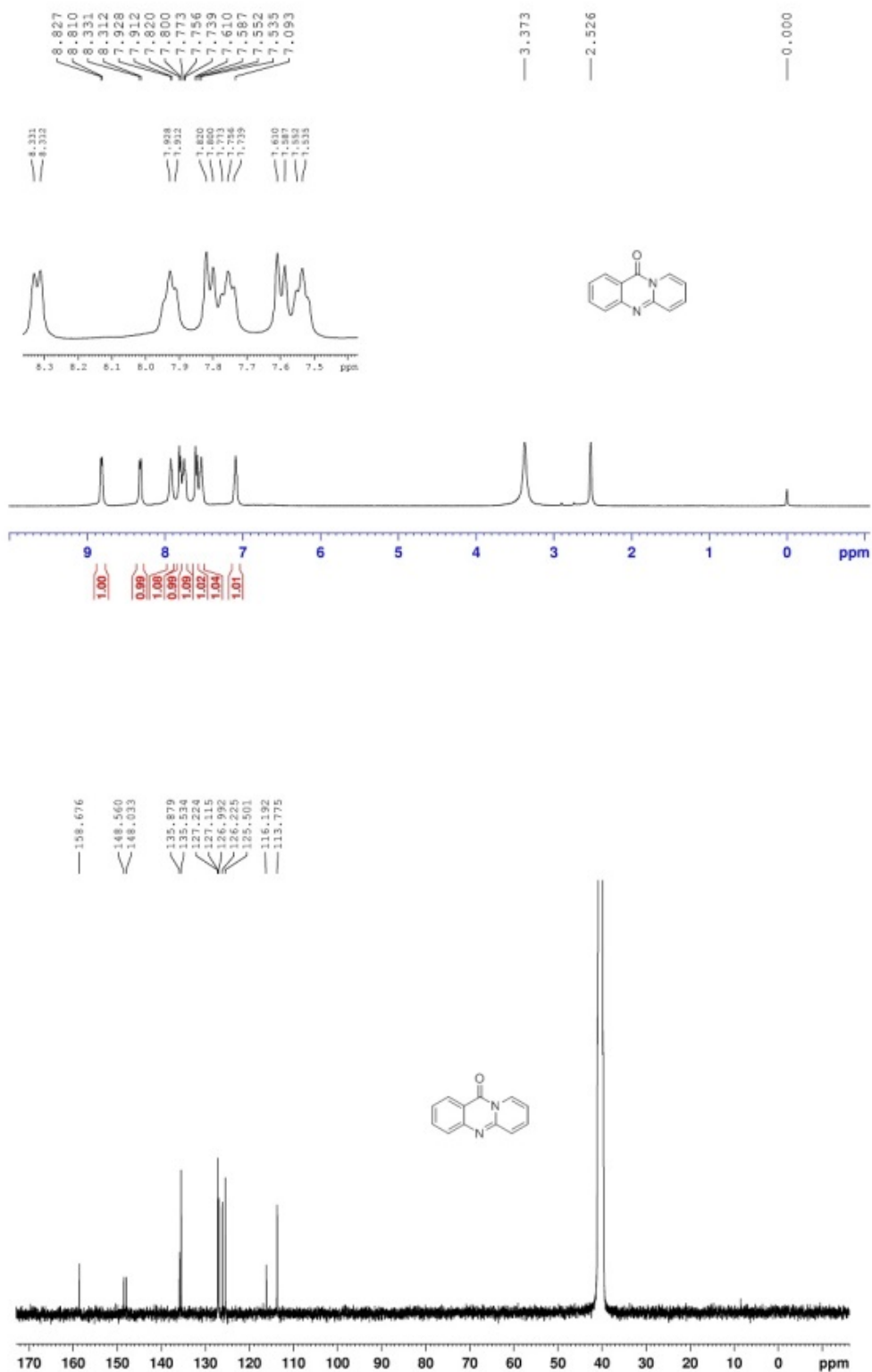
The synthesis method of **1e** was also used to synthesize compounds **1f-1h**, **5q-5t**.

5a-5t: 2,4-dichloropyrimidines (2.0 mmol) and anthranilic acids (5.0 mmol) were dissolved in EtOH (10 mL), then 6N HCl (0.5 mL) was added. The mixture was refluxed for **2-3 h**, which was then cooled to room temperature and filtered, washed the precipitate with EtOH and dried to give pure products **5a-5t**.

1. H. R. Lawrence et al. *J. Med. Chem.*, 2012, **55**, 7392-7416.

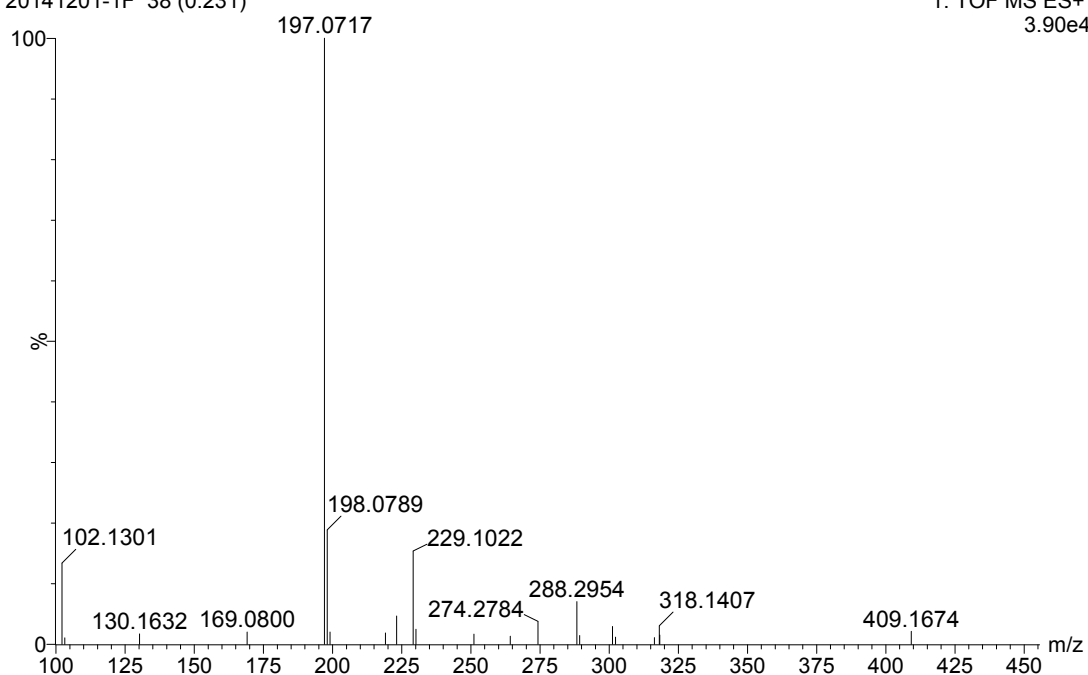
Charts of ¹H-NMR, ¹³C-NMR and HRMS of products

2d

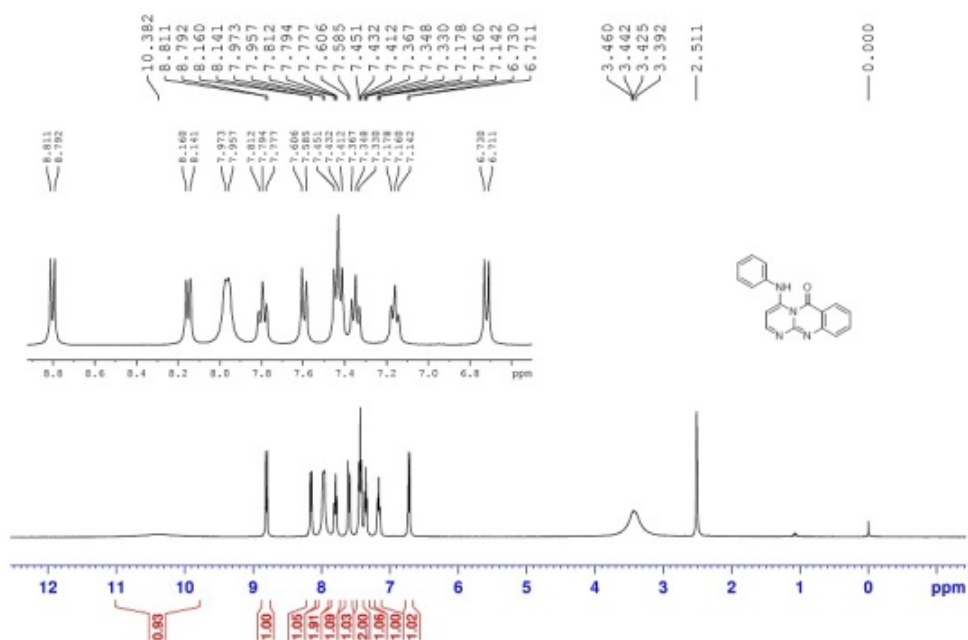


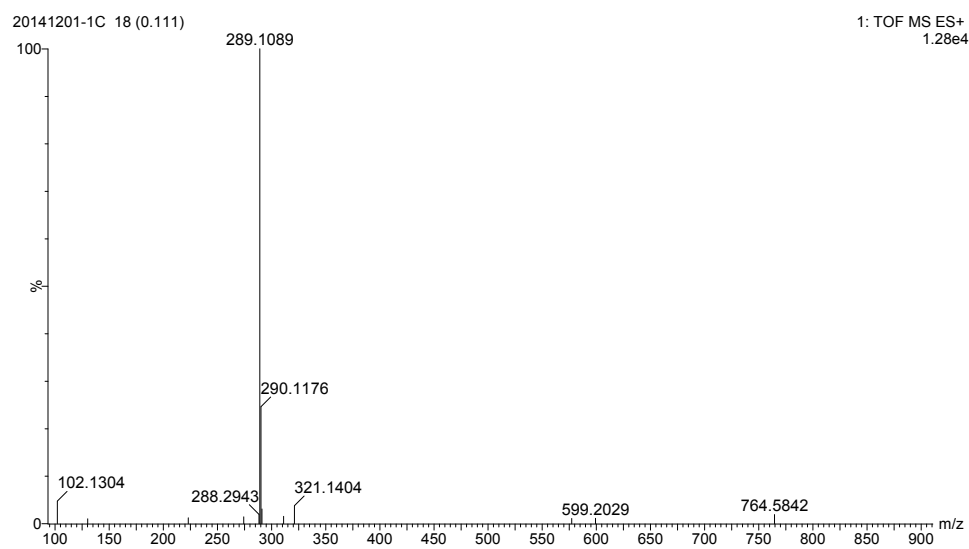
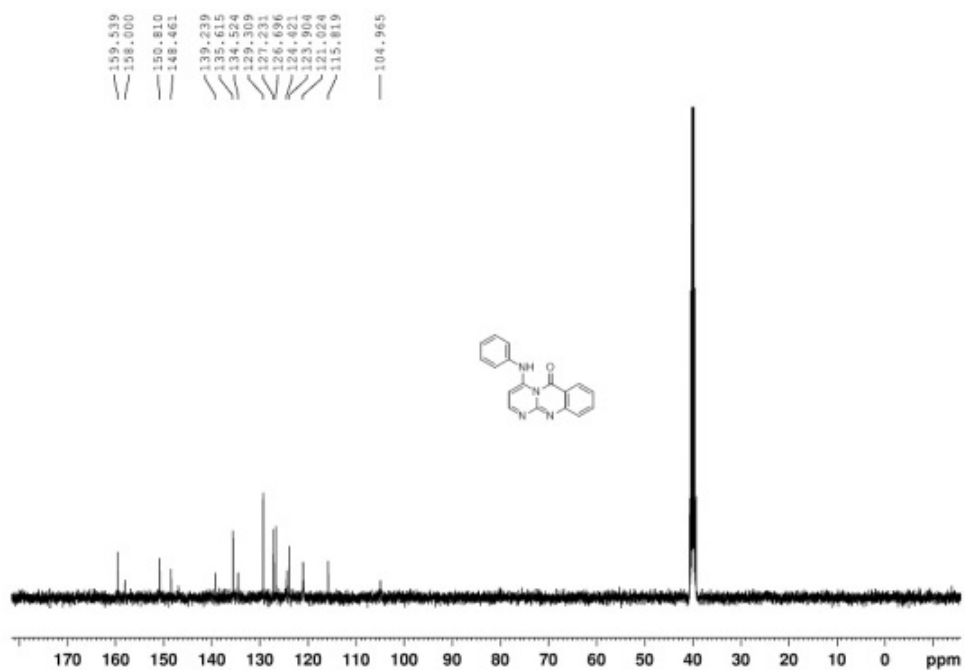
20141201-1F 38 (0.231)

1: TOF MS ES+
3.90e4

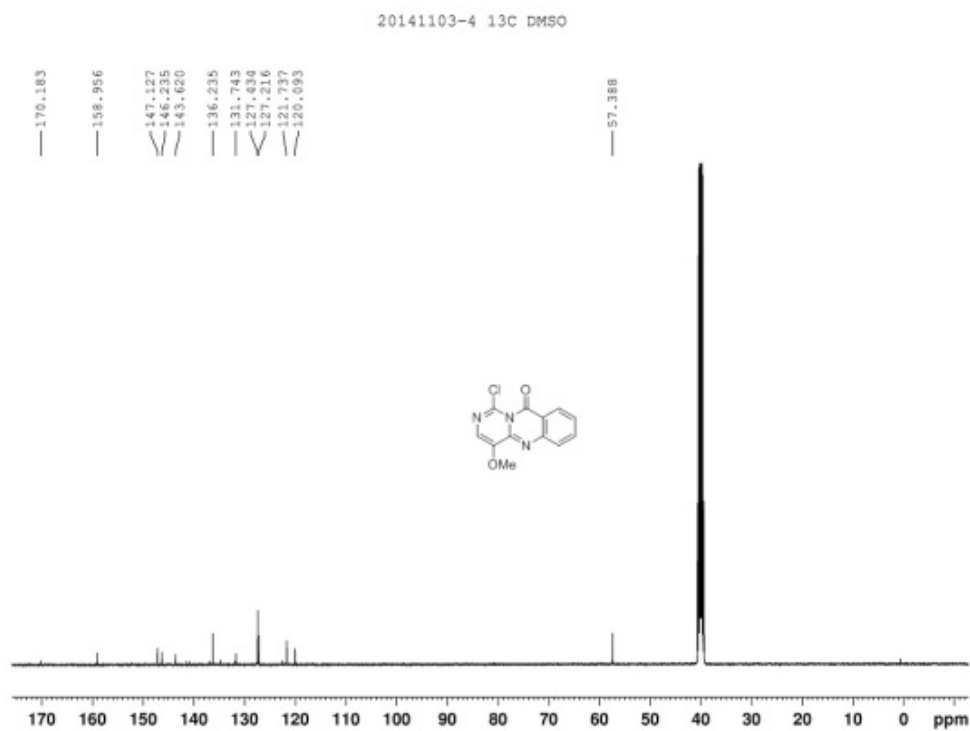
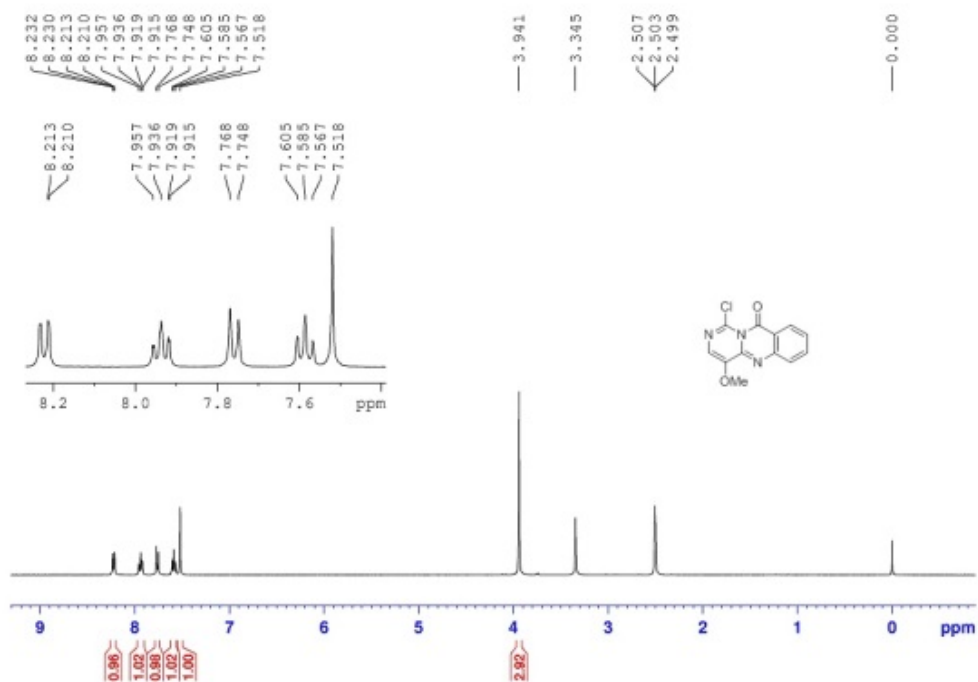


2ga



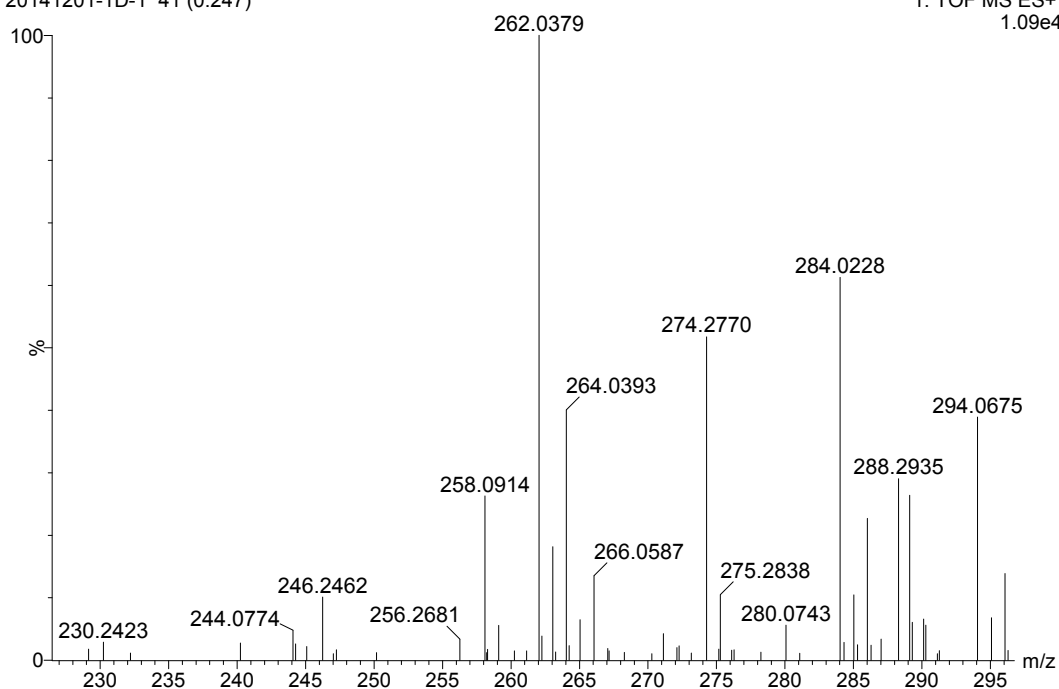


2i

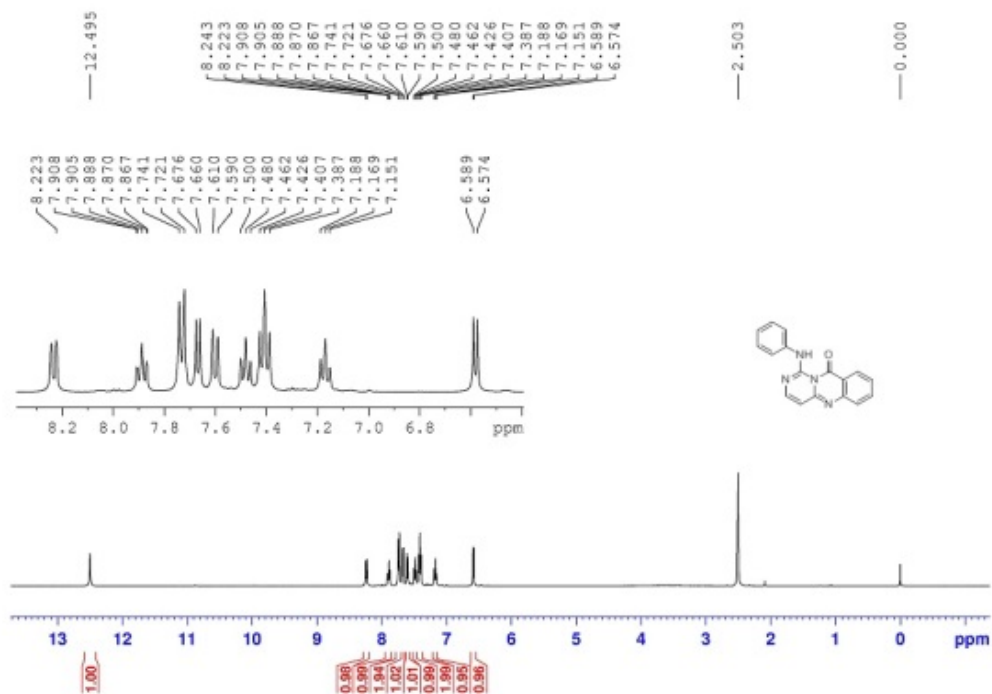


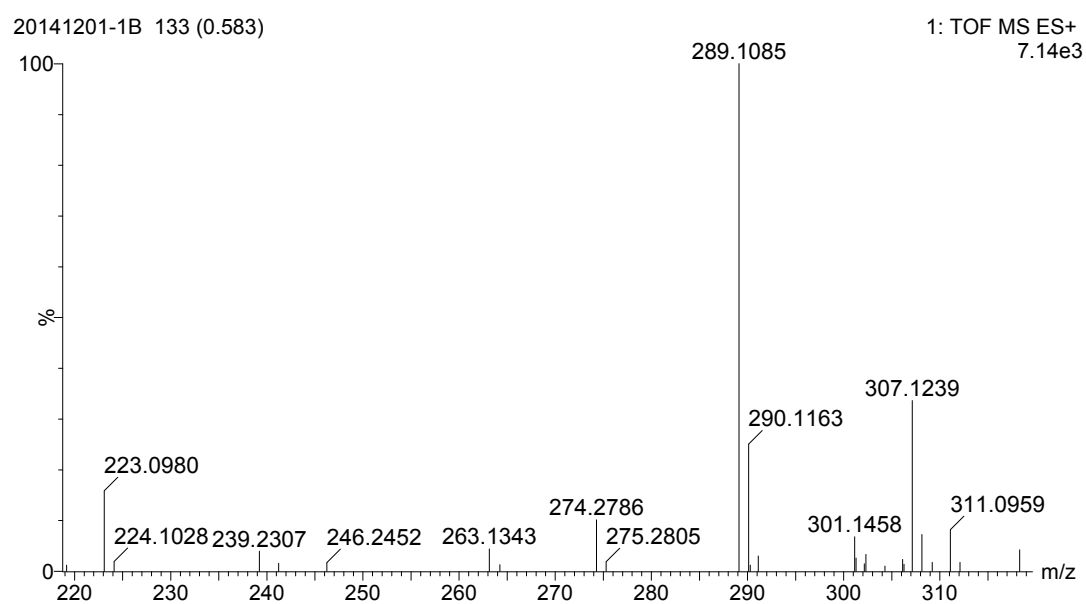
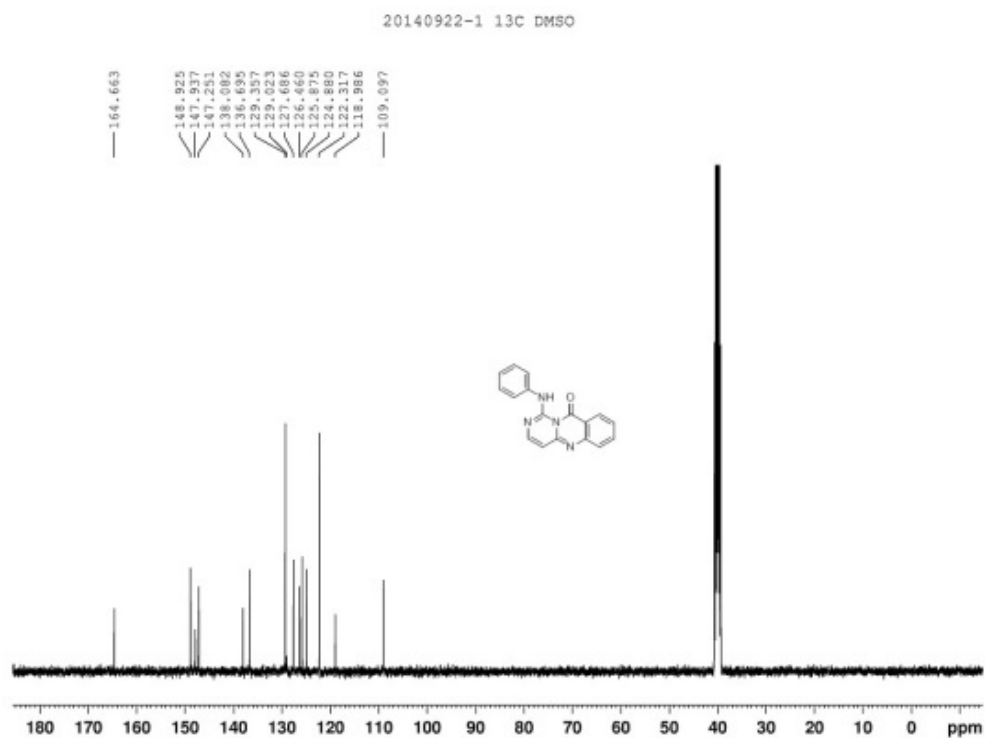
20141201-1D-1 41 (0.247)

1: TOF MS ES+
1.09e4

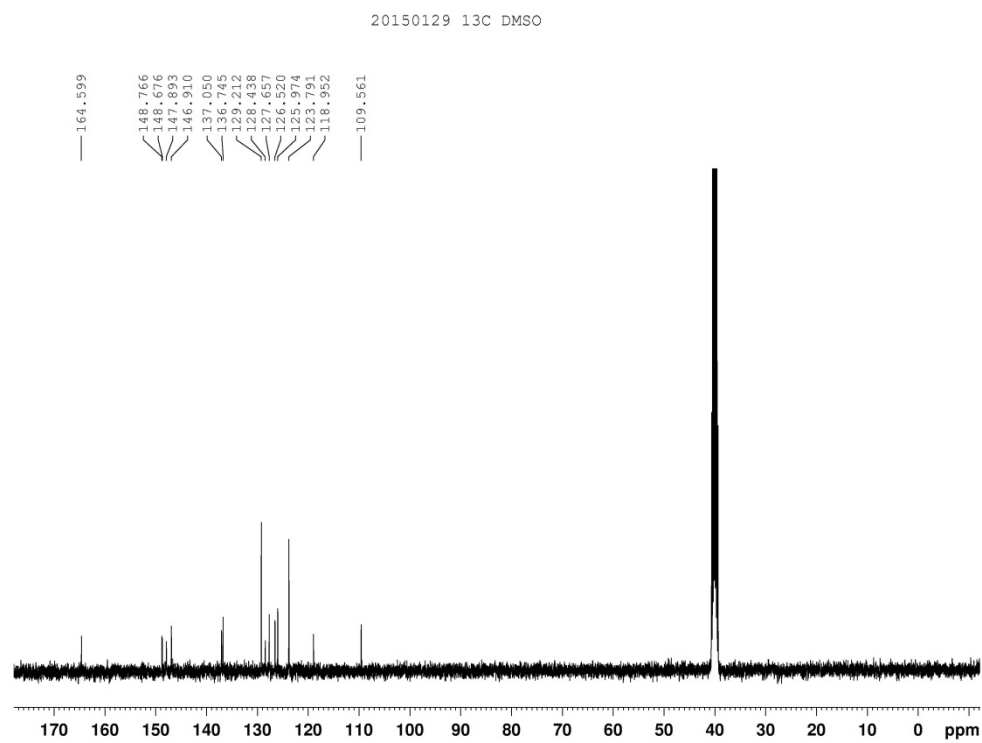
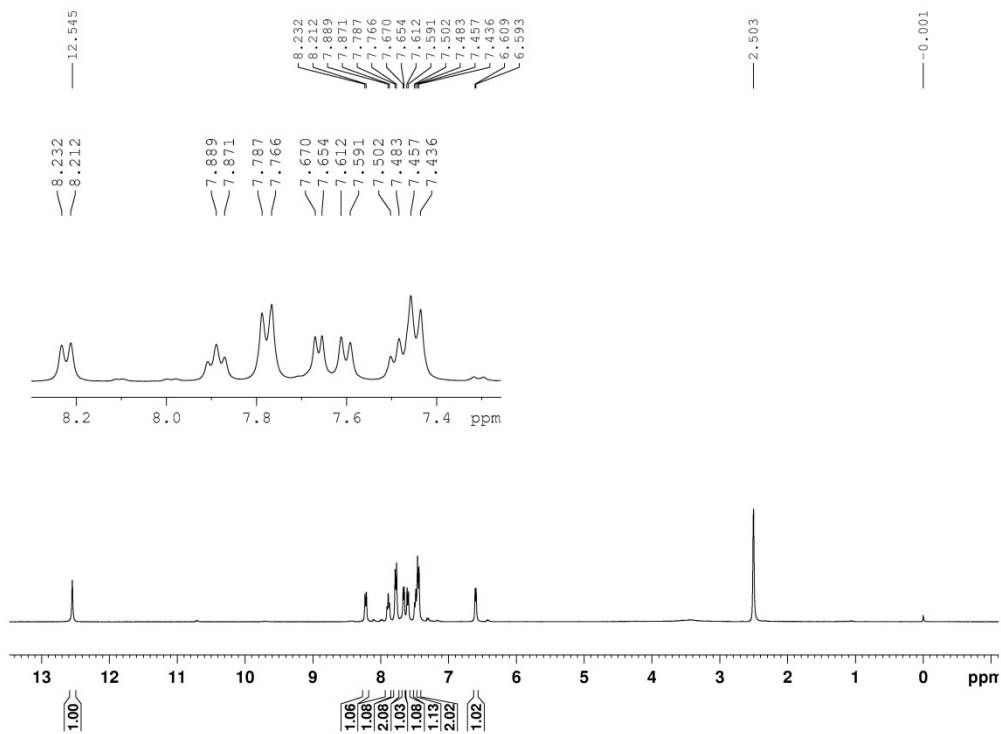


2j



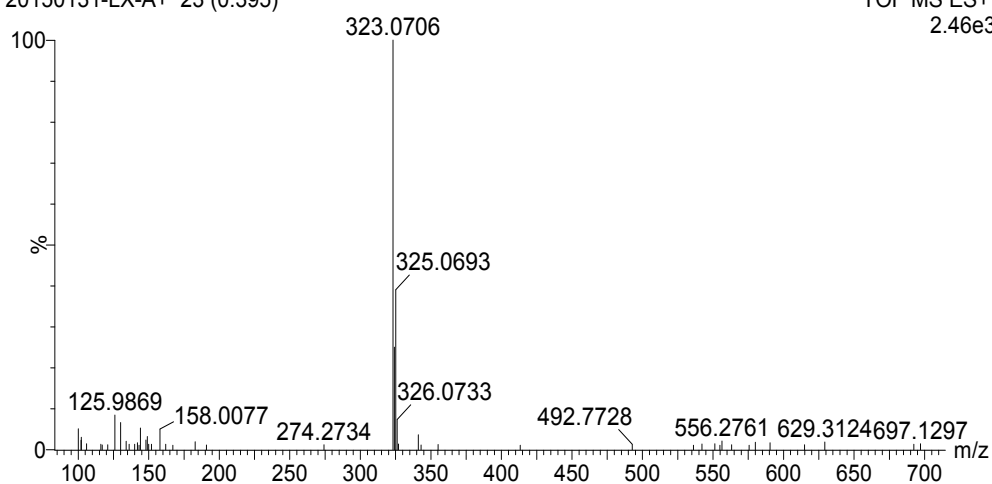


2k



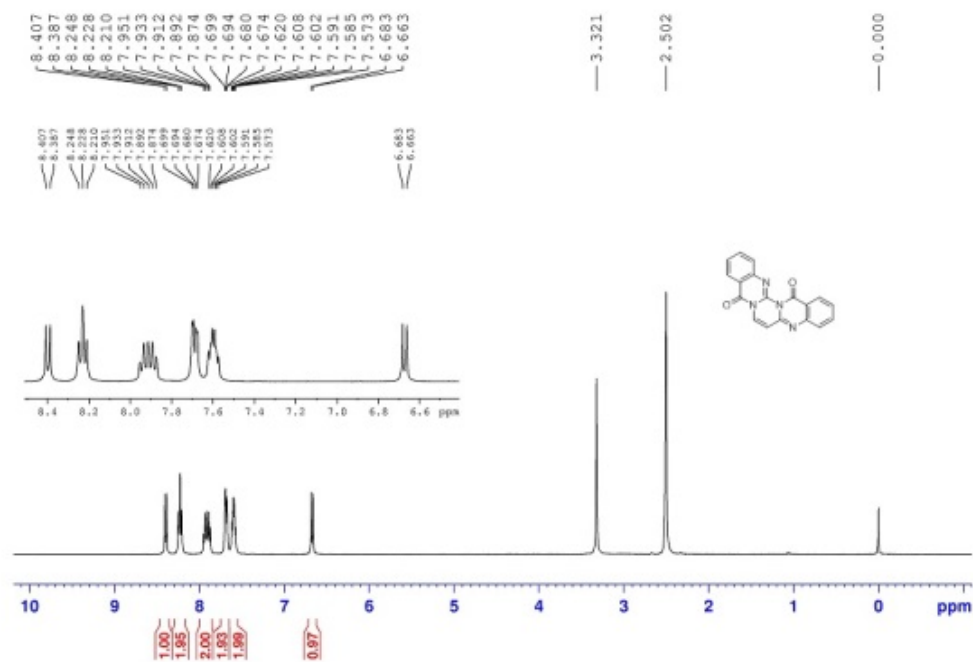
20150131-LX-A+ 23 (0.395)

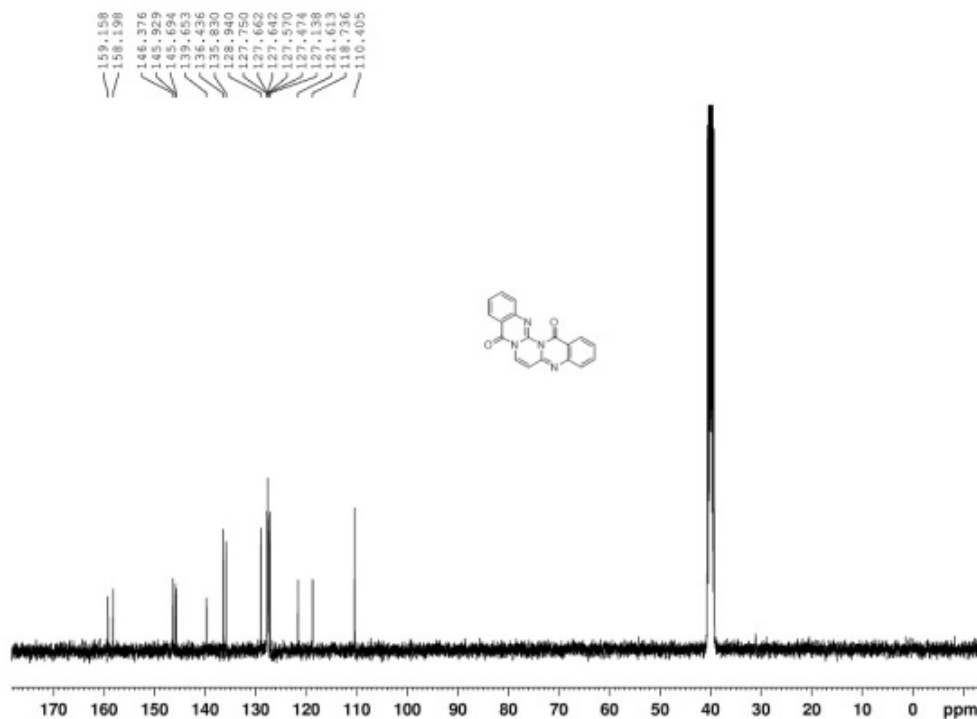
TOF MS ES+
2.46e3



6a

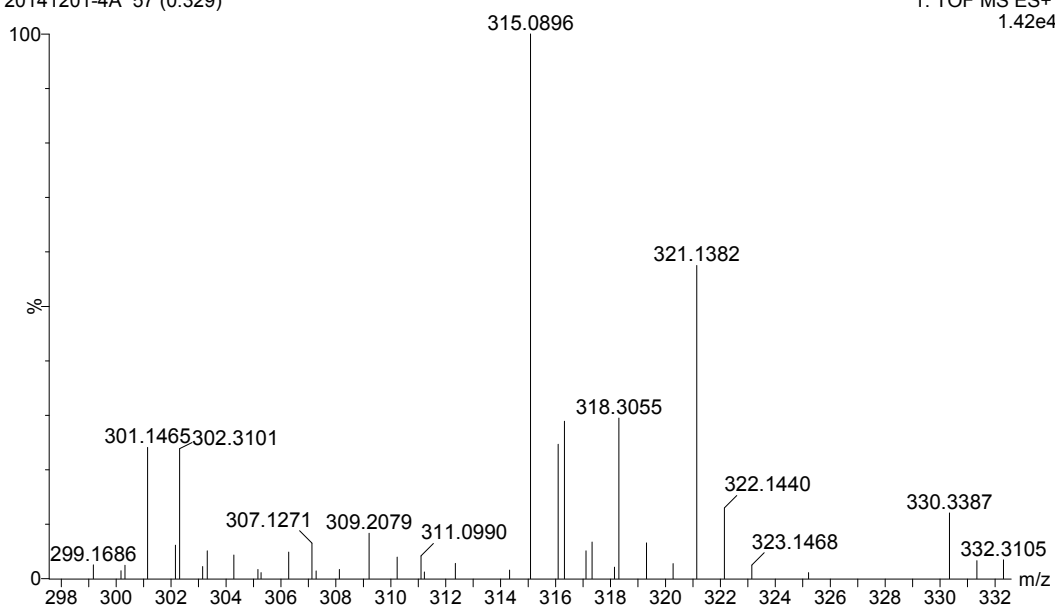
20140919-4 DMSO



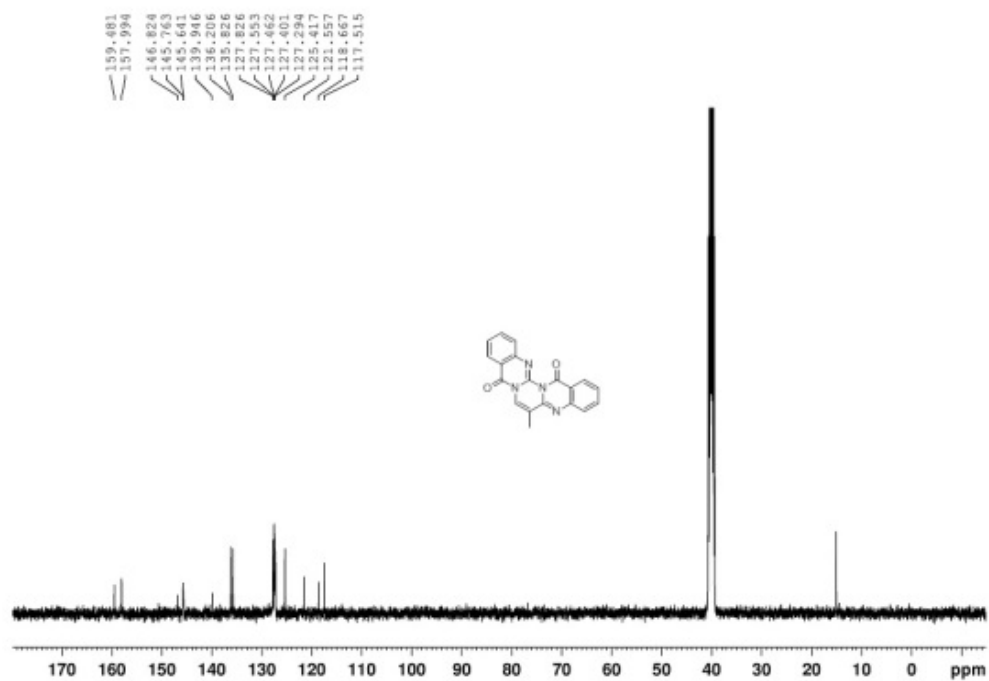
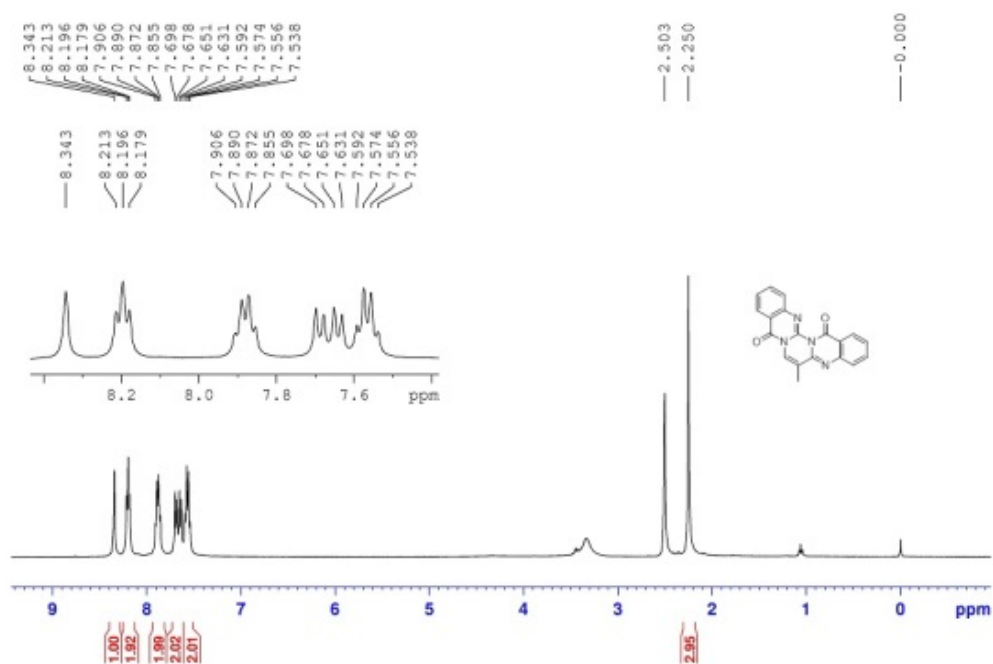


20141201-4A 57 (0.329)

1: TOF MS ES+
1.42e4

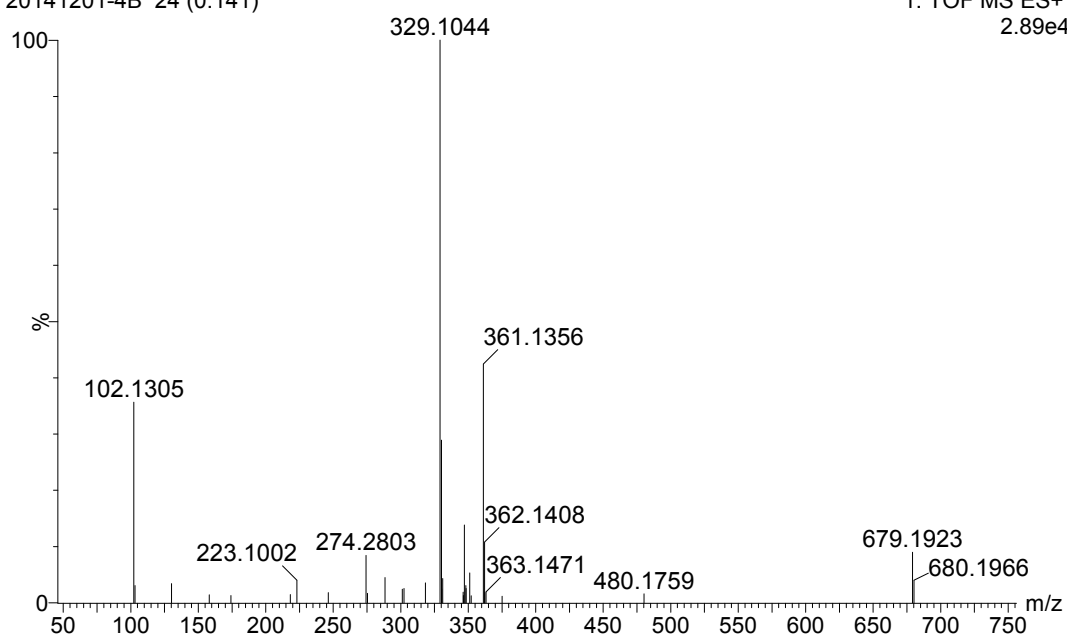


6b

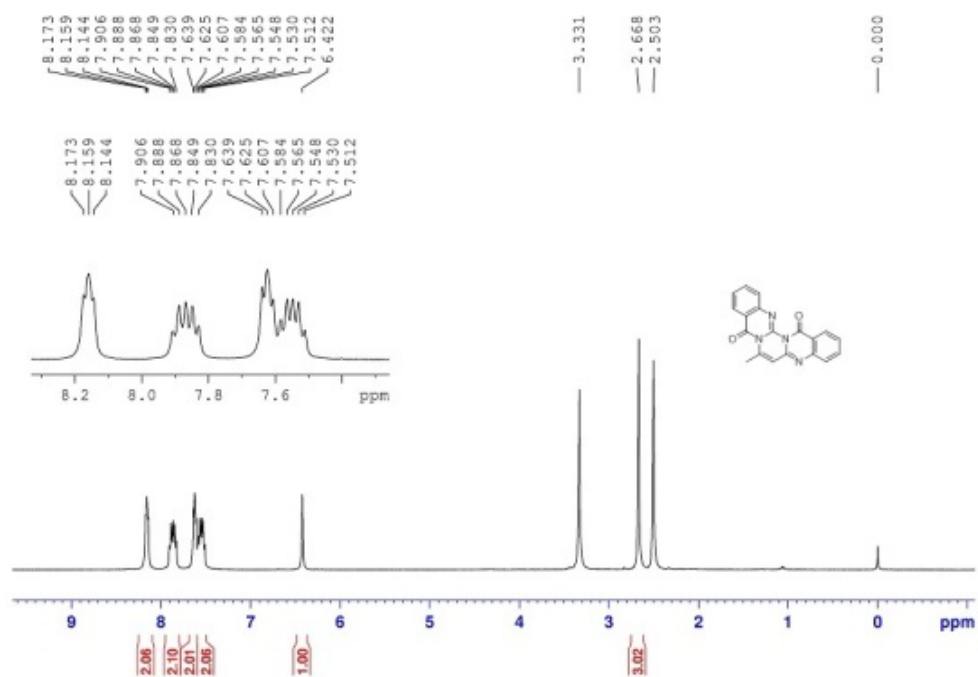


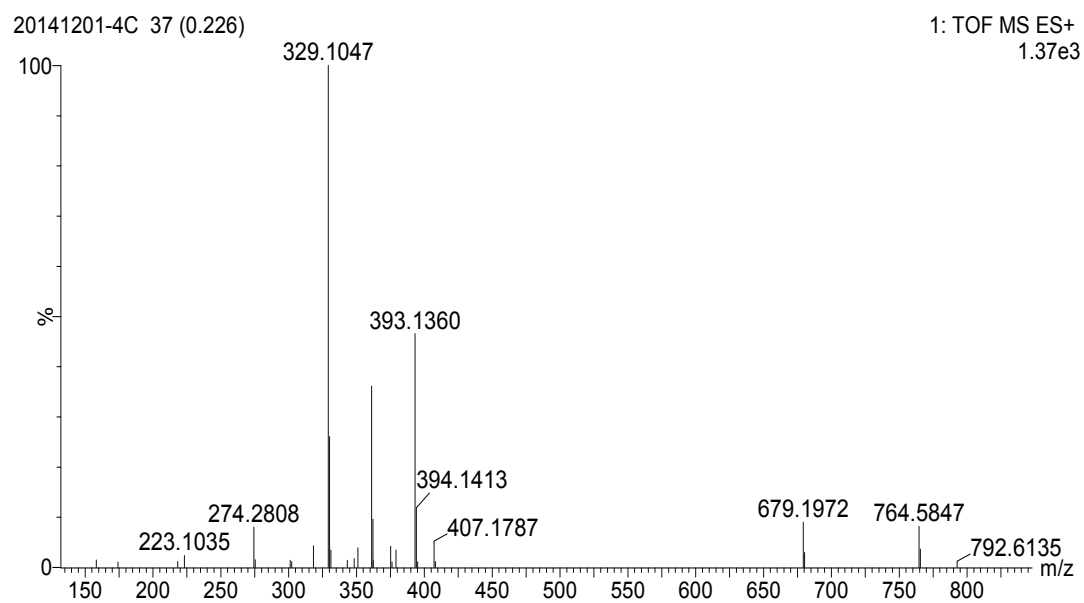
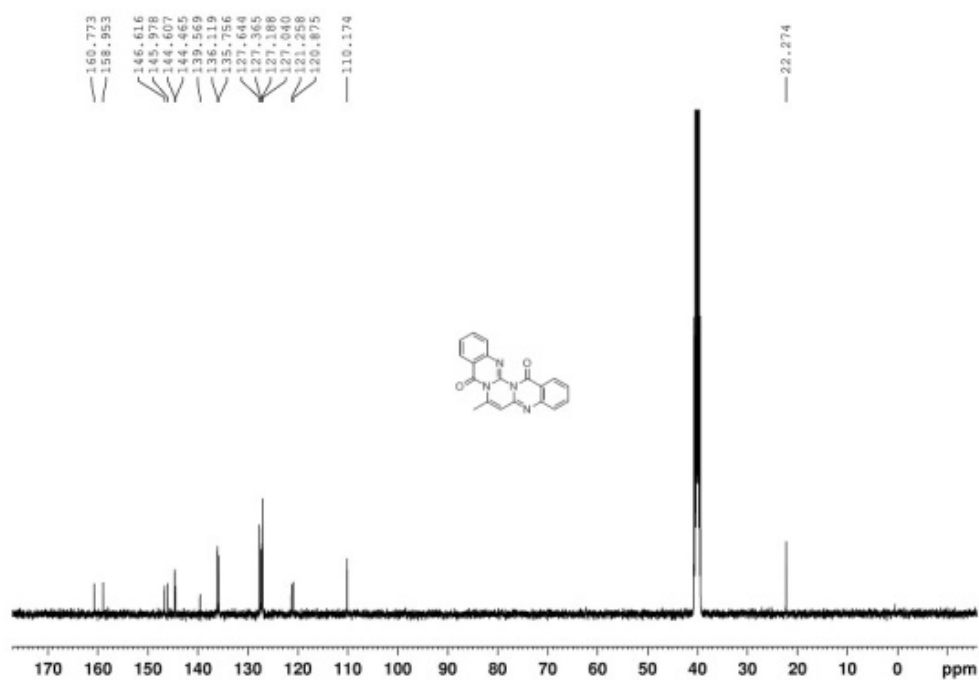
20141201-4B 24 (0.141)

1: TOF MS ES+
2.89e4

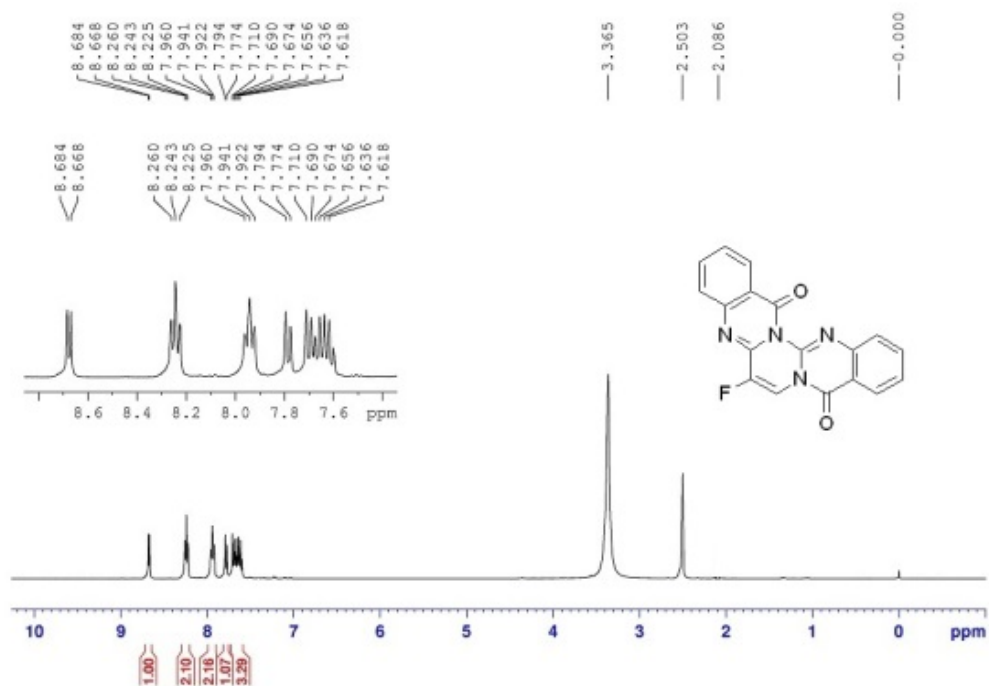


6c

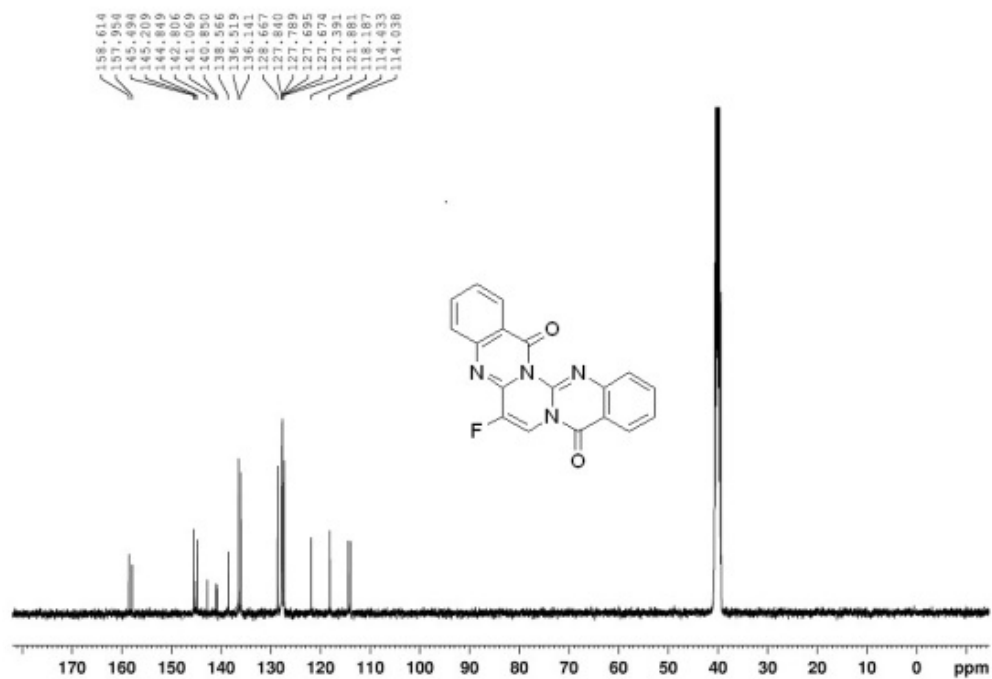




6d

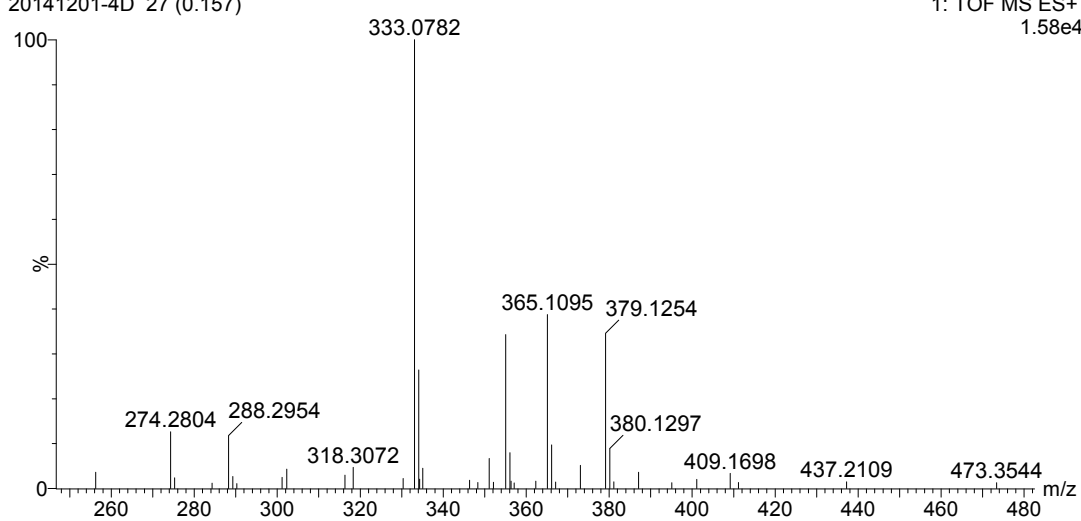


20141011-2 ¹³C DMSO

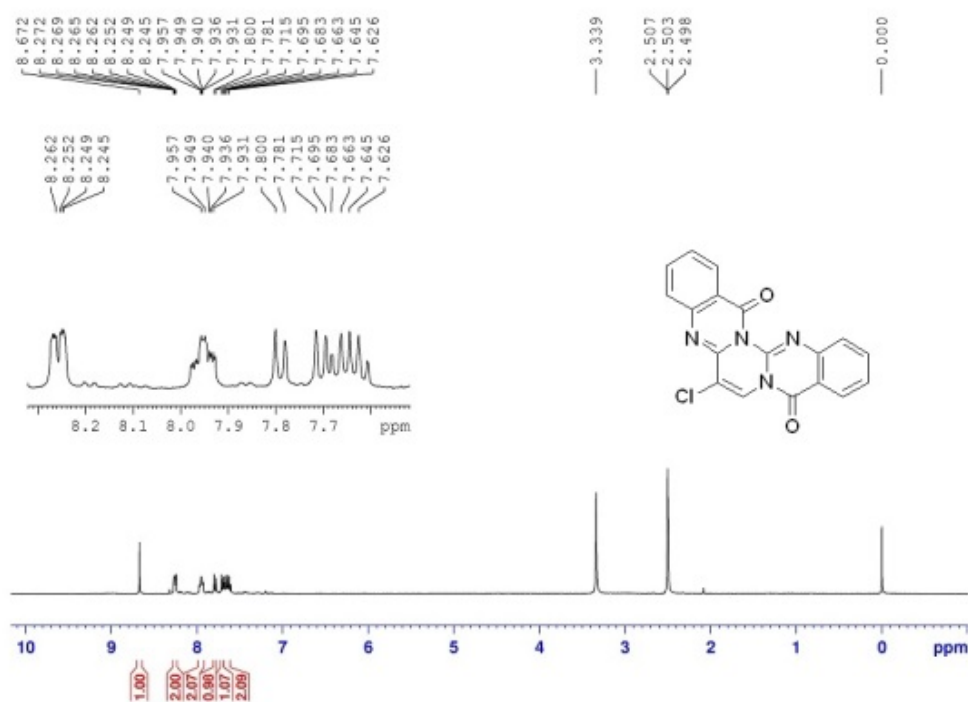


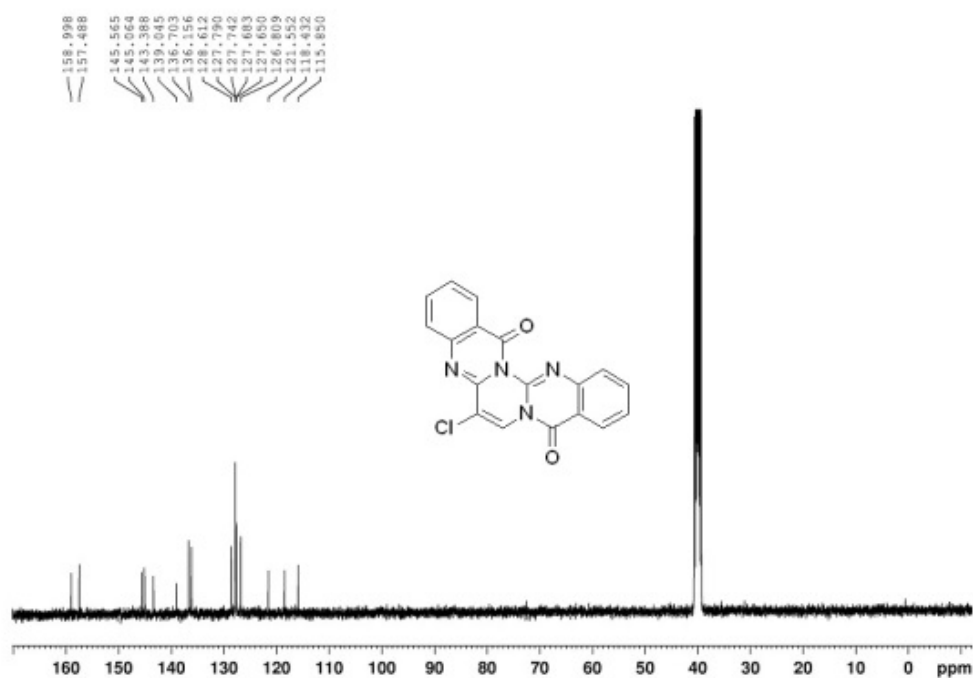
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1: TOF MS ES+
1.58e4



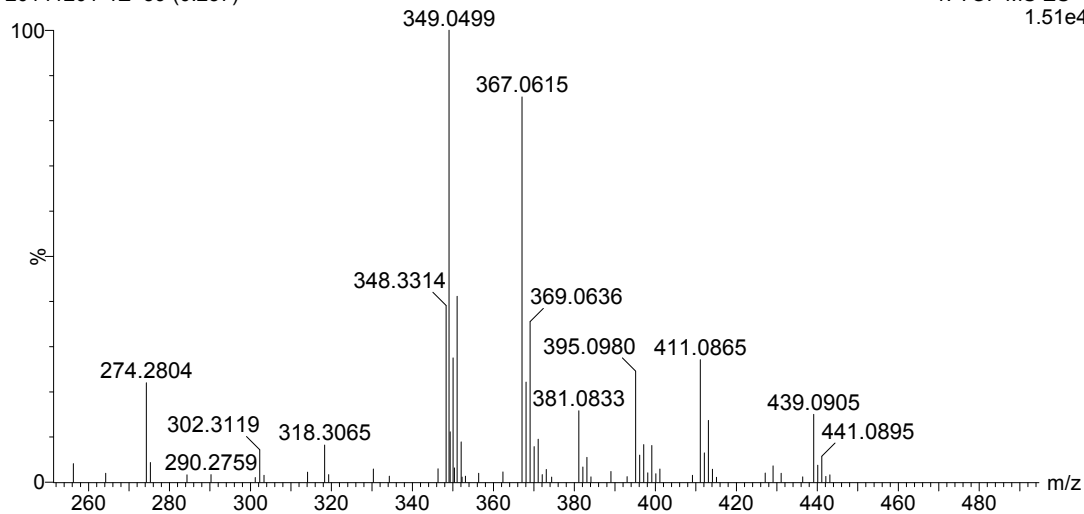
6e



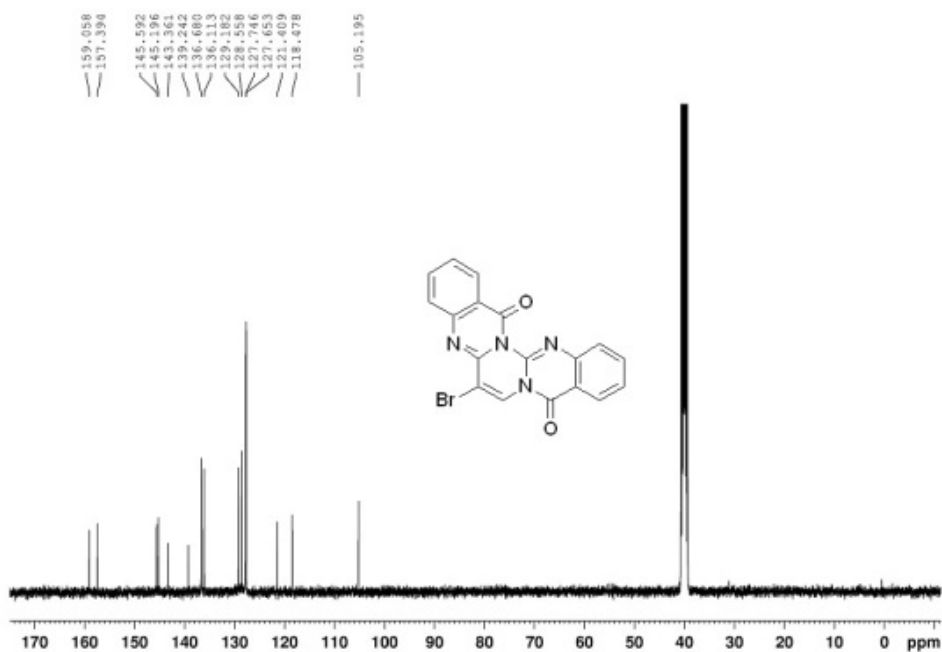
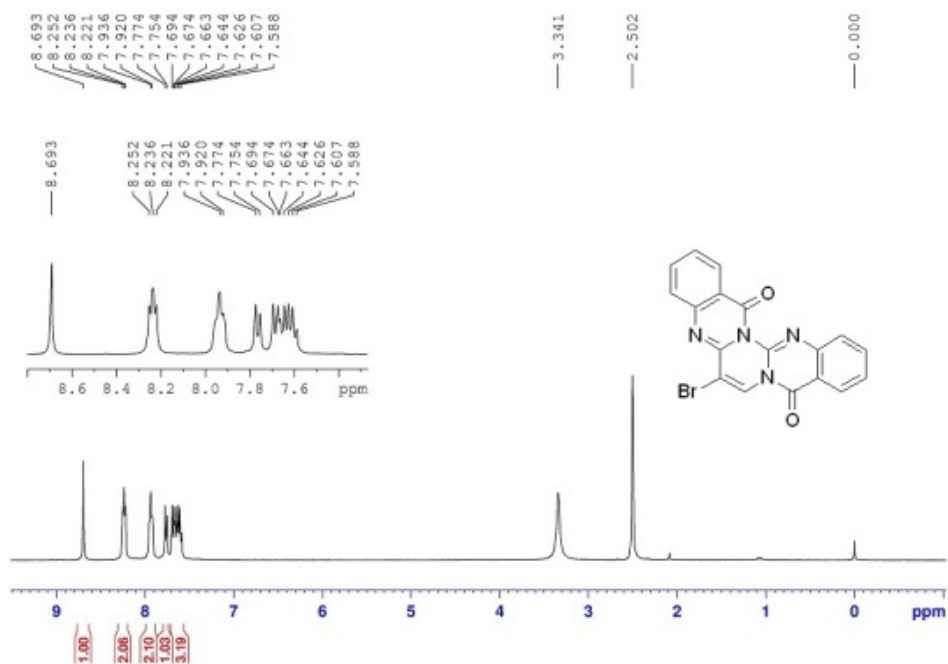


20141201-4E 39 (0.237)

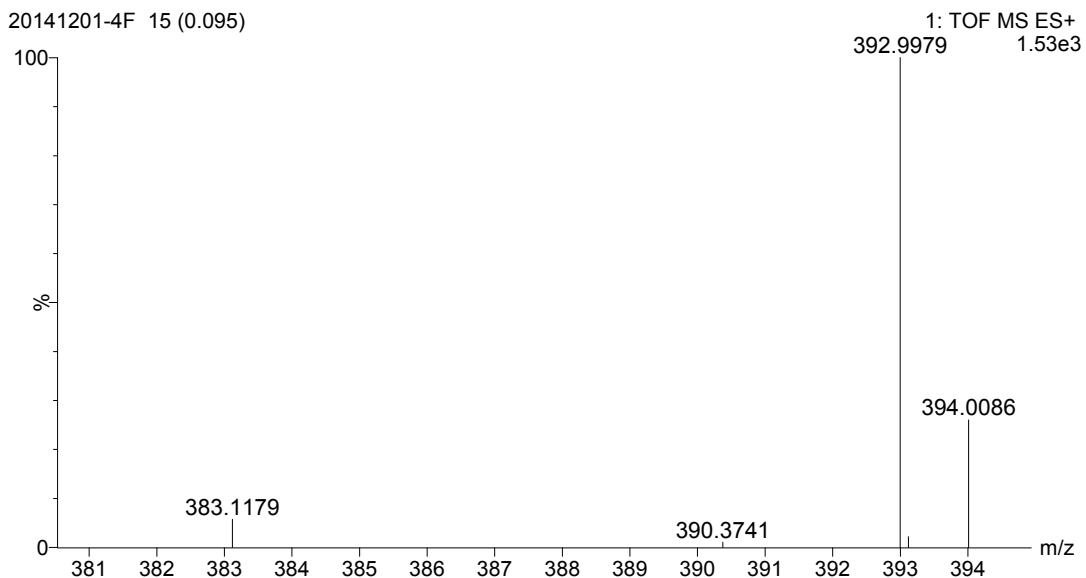
1: TOF MS ES+
1.51e4



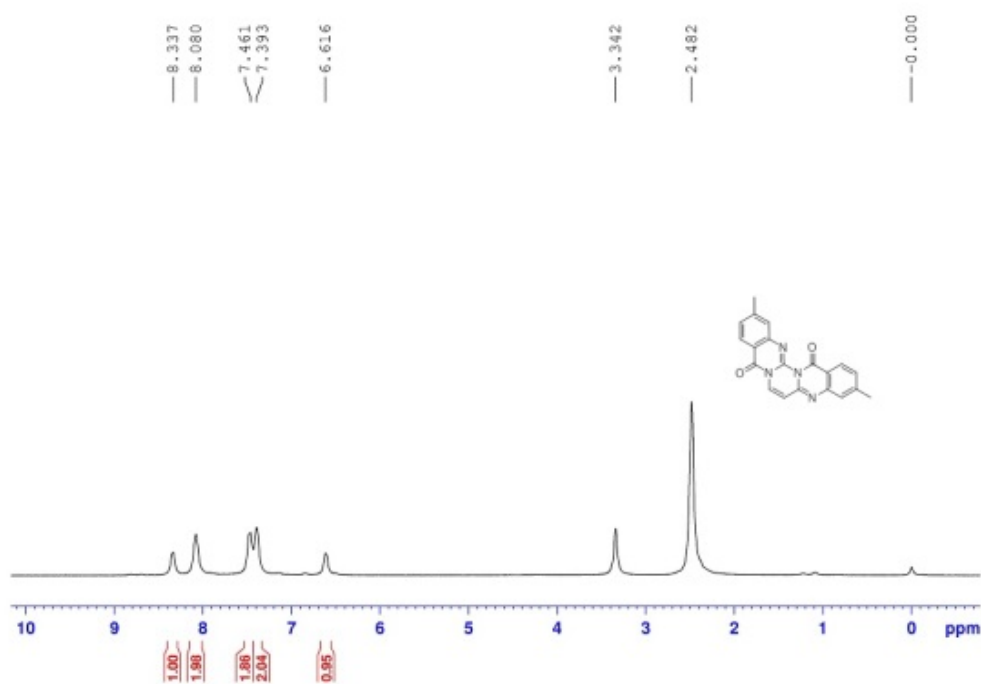
6f

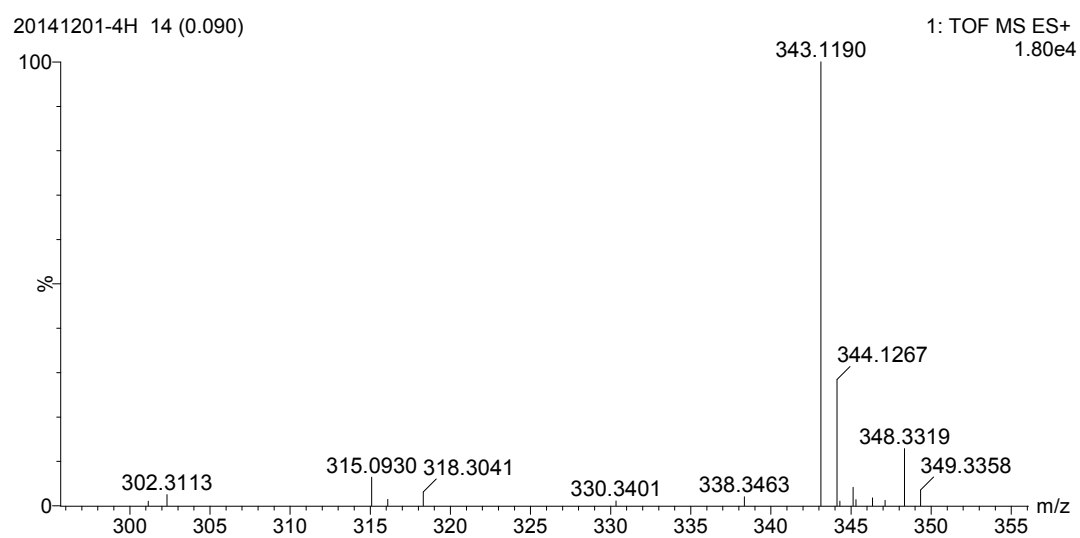
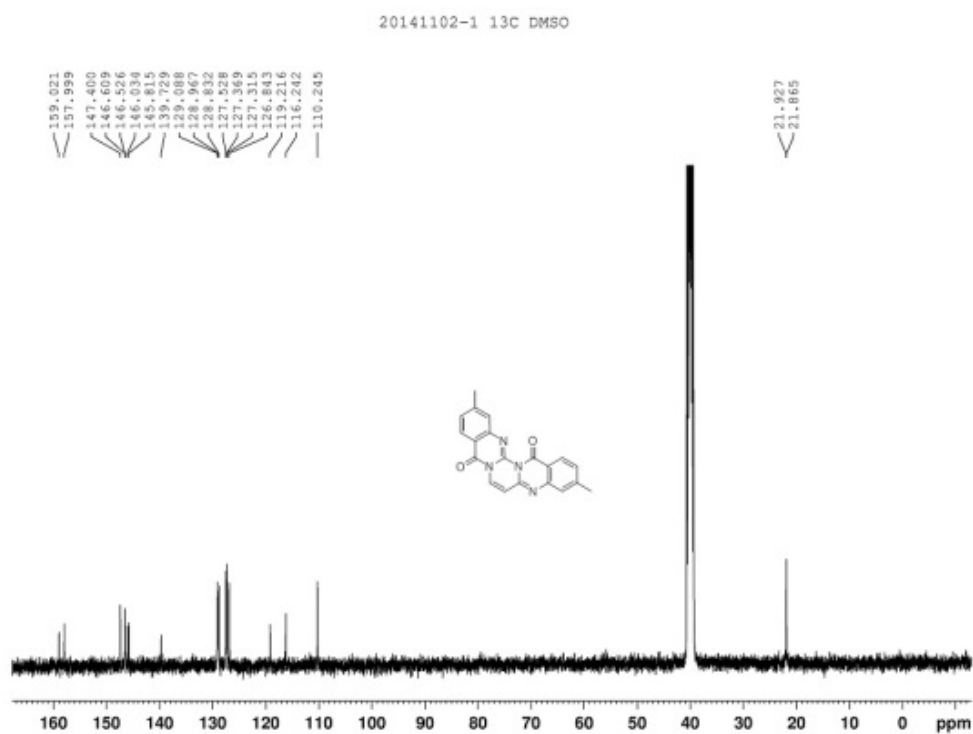


20141201-4F 15 (0.095)

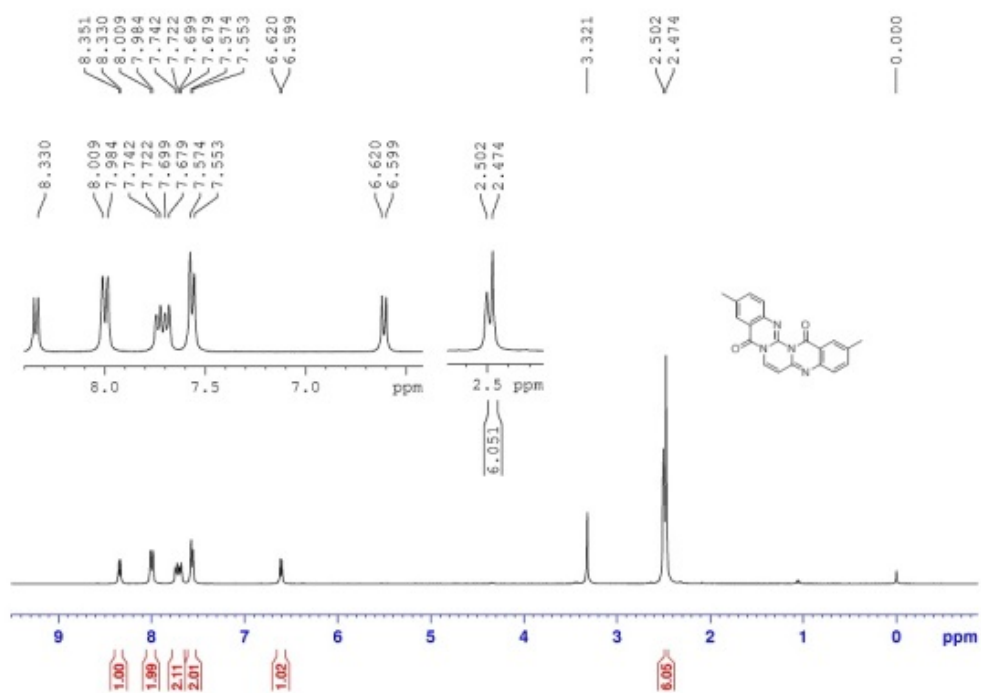


6h

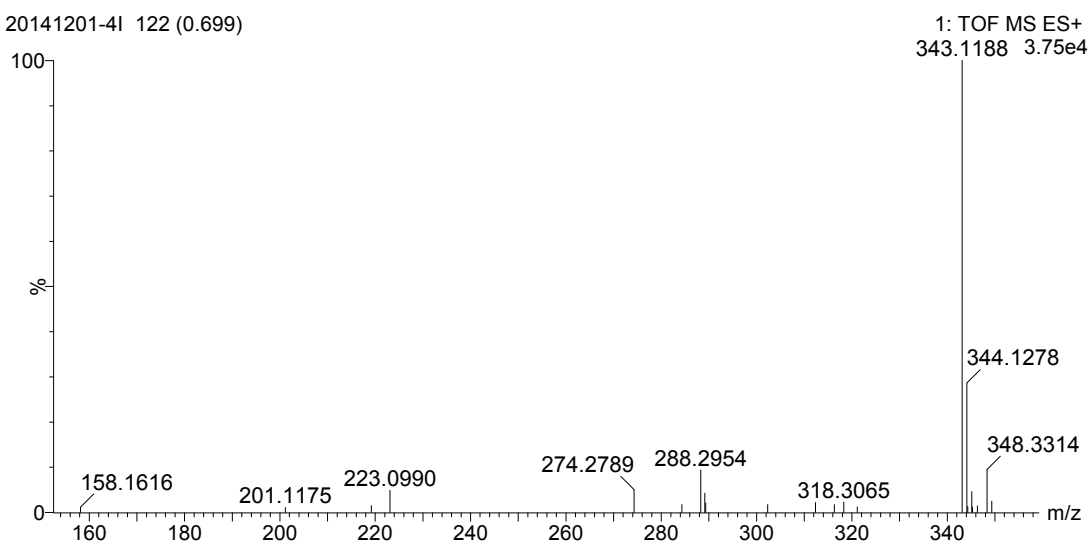




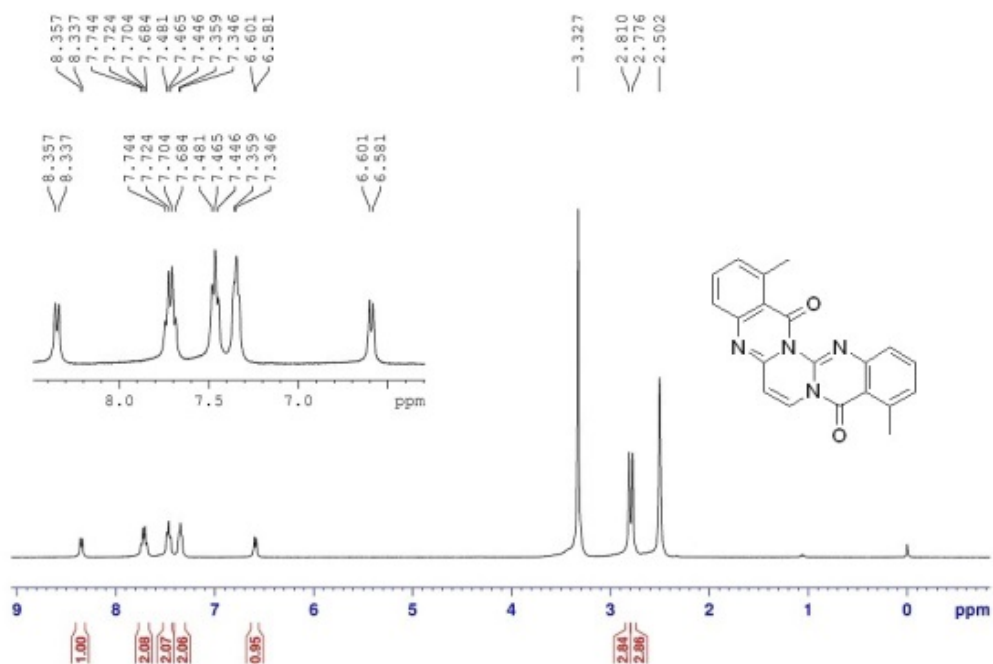
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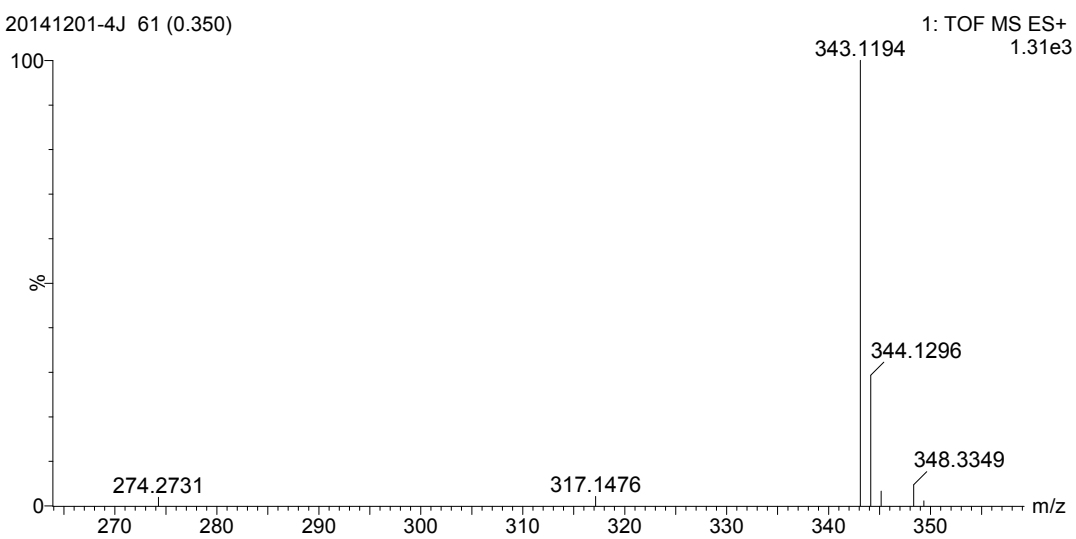
20141201-4I 122 (0.699)



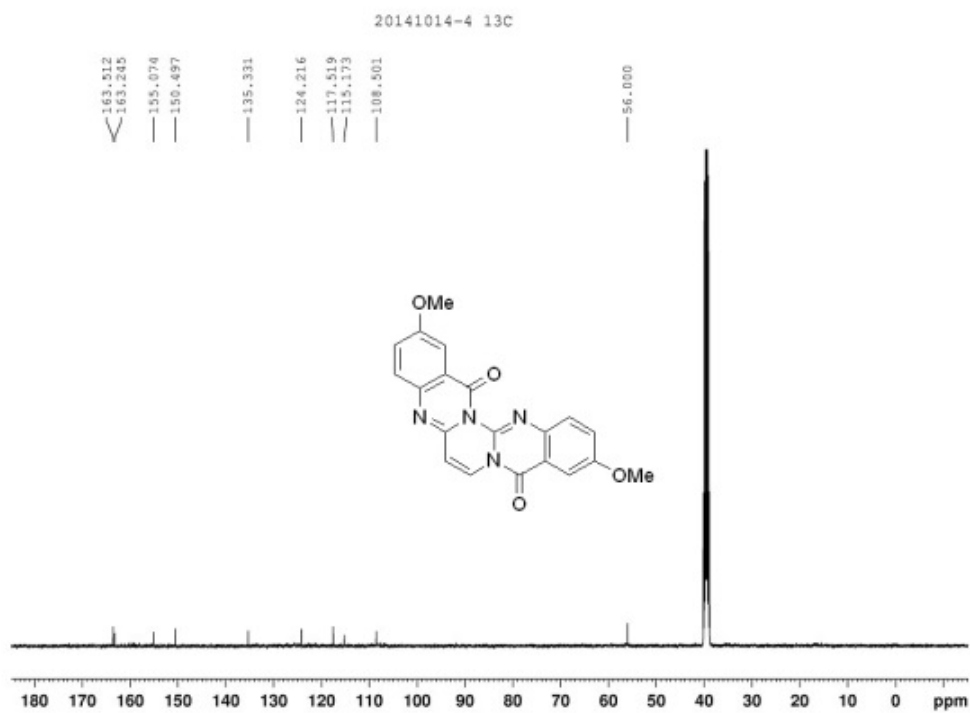
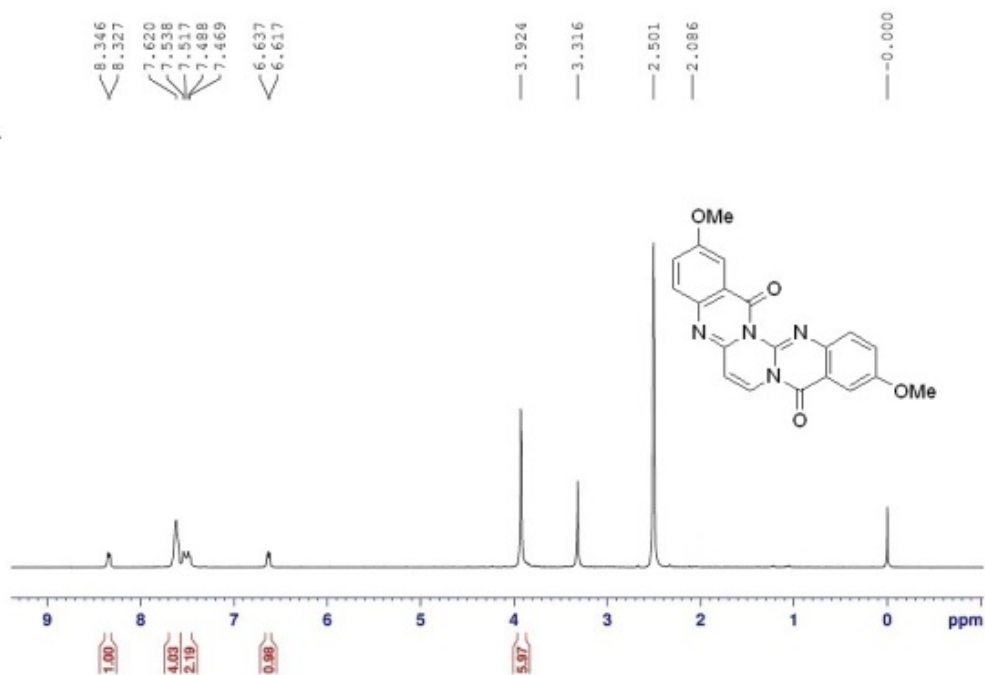
6j



20141201-4J 61 (0.350)

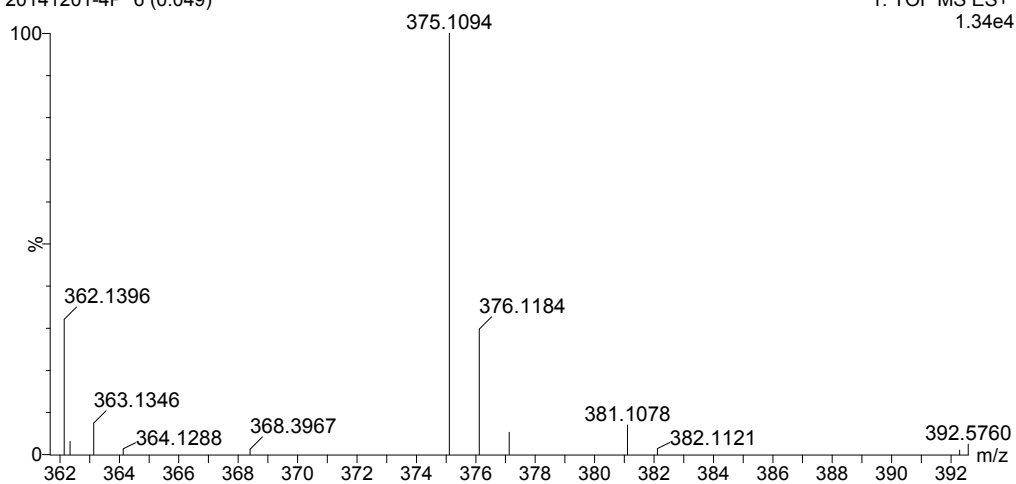


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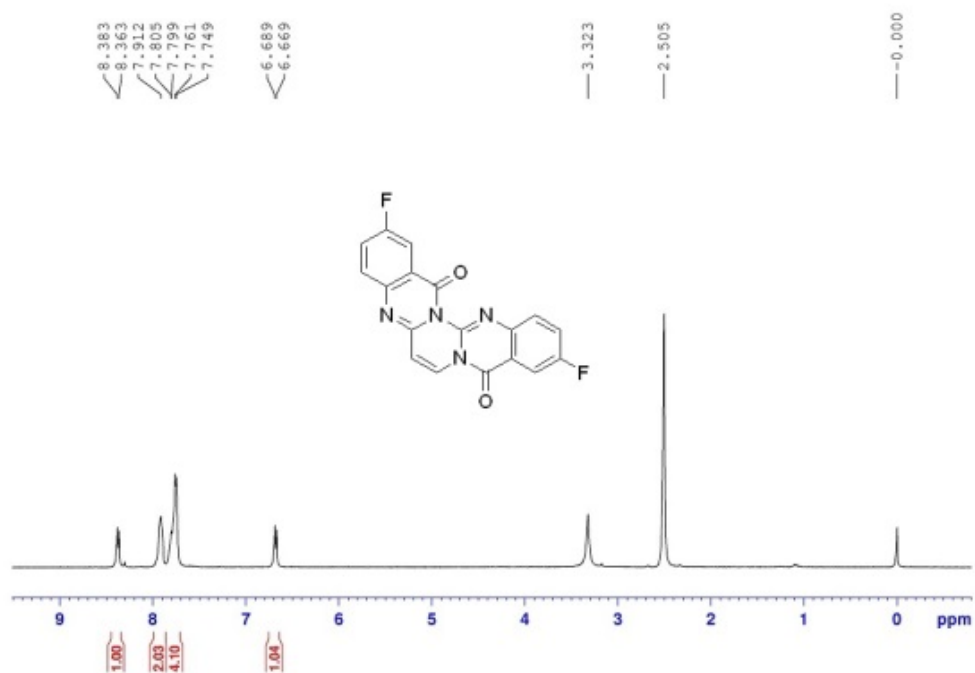


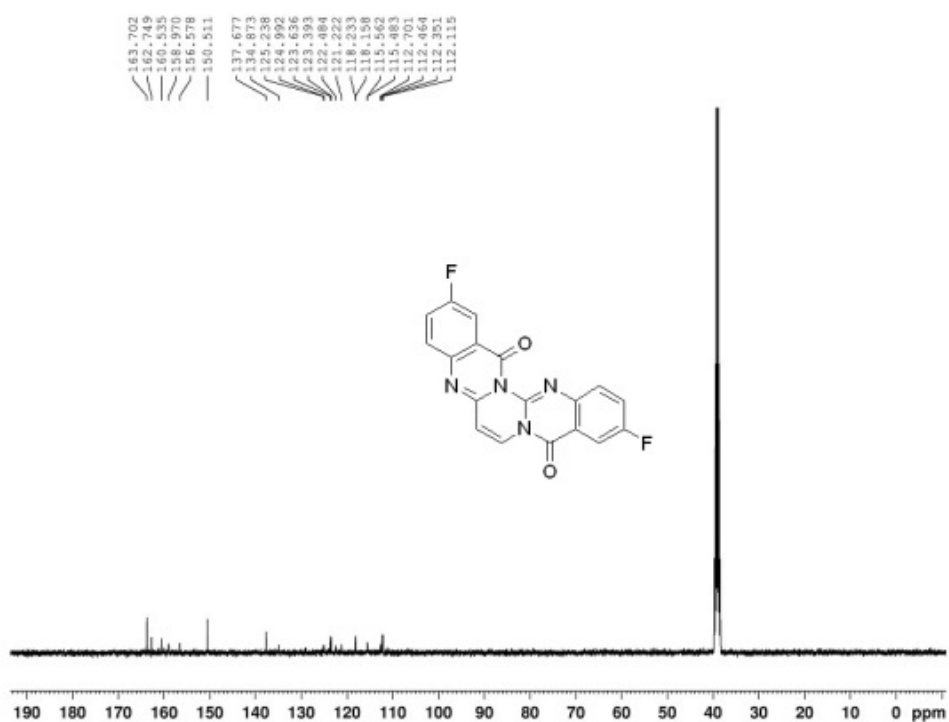
20141201-4P 6 (0.049)

1: TOF MS ES+
1.34e4



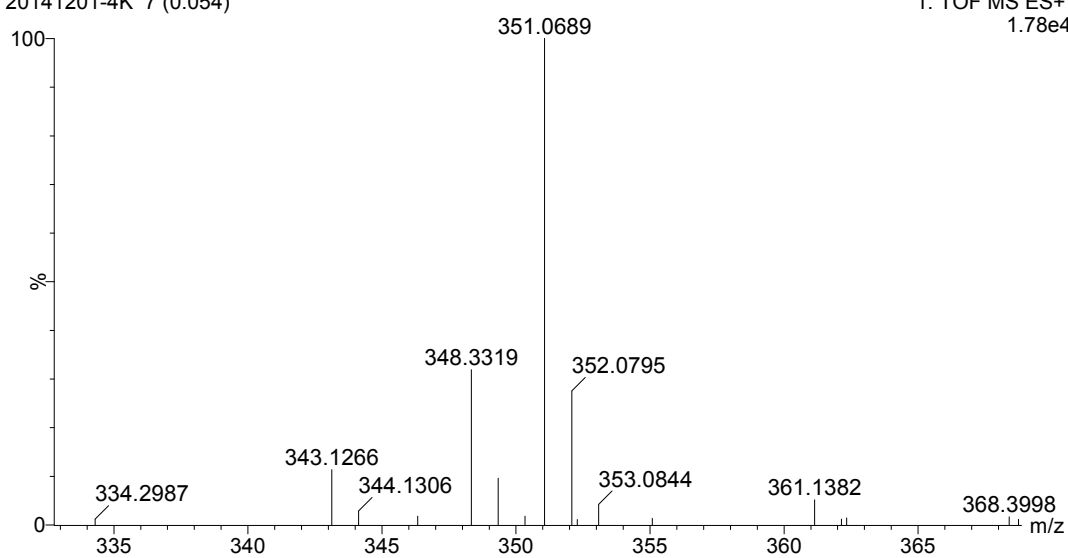
61



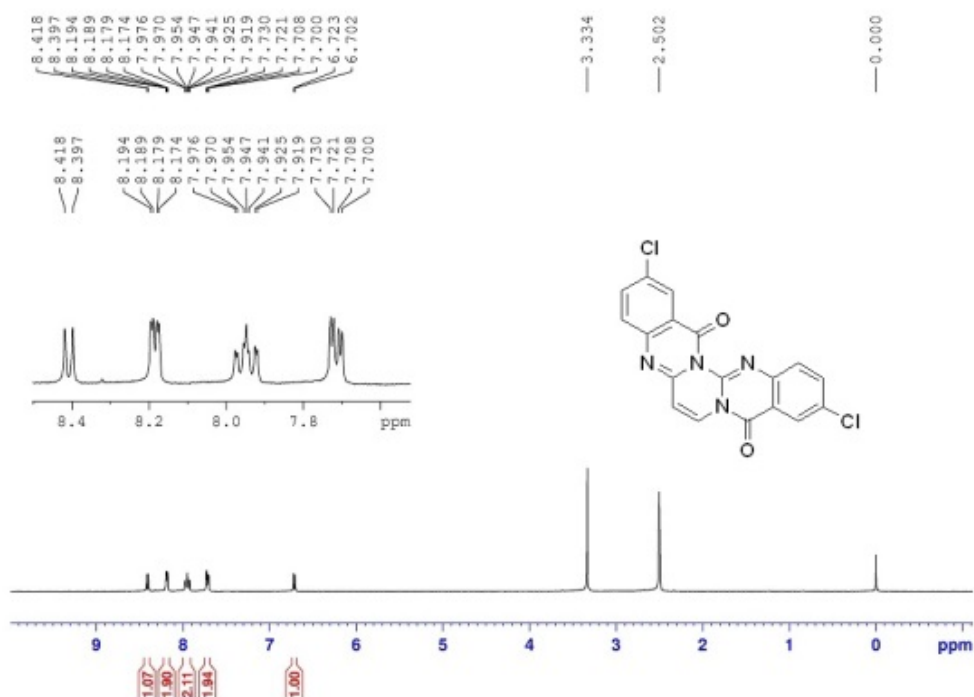


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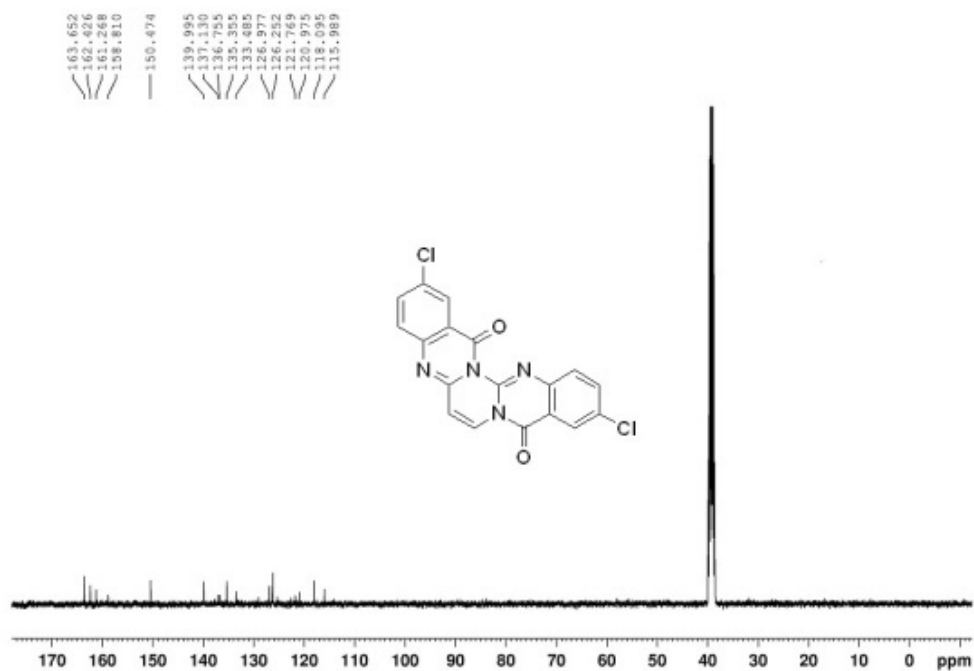
1: TOF MS ES+
1.78e4



6m

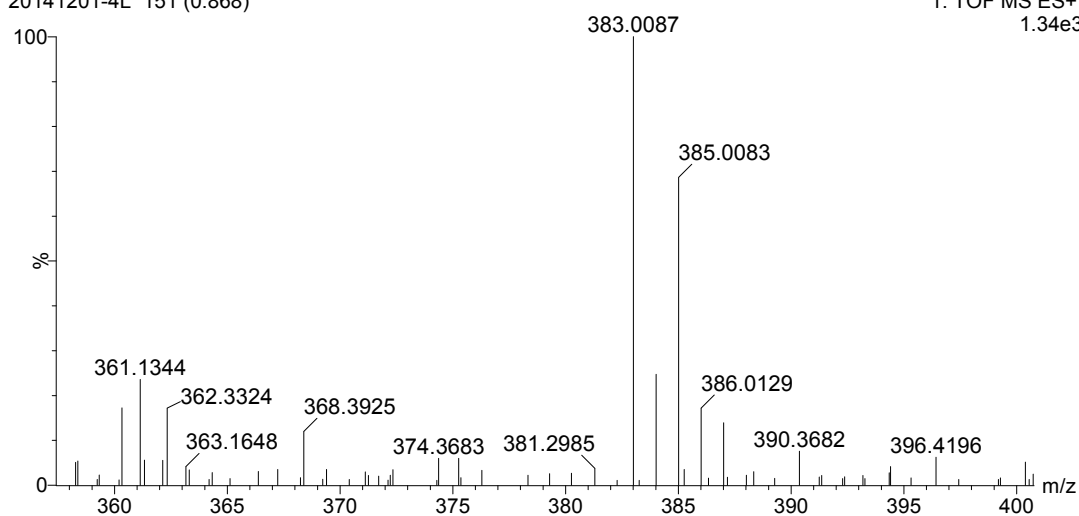


20141014-1 ¹³C DMSO

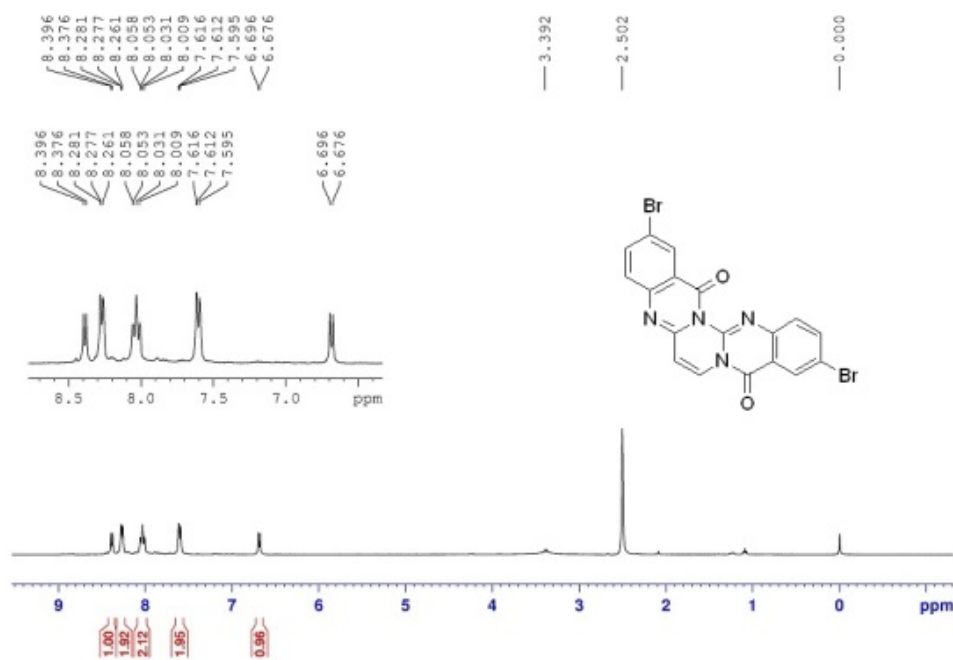


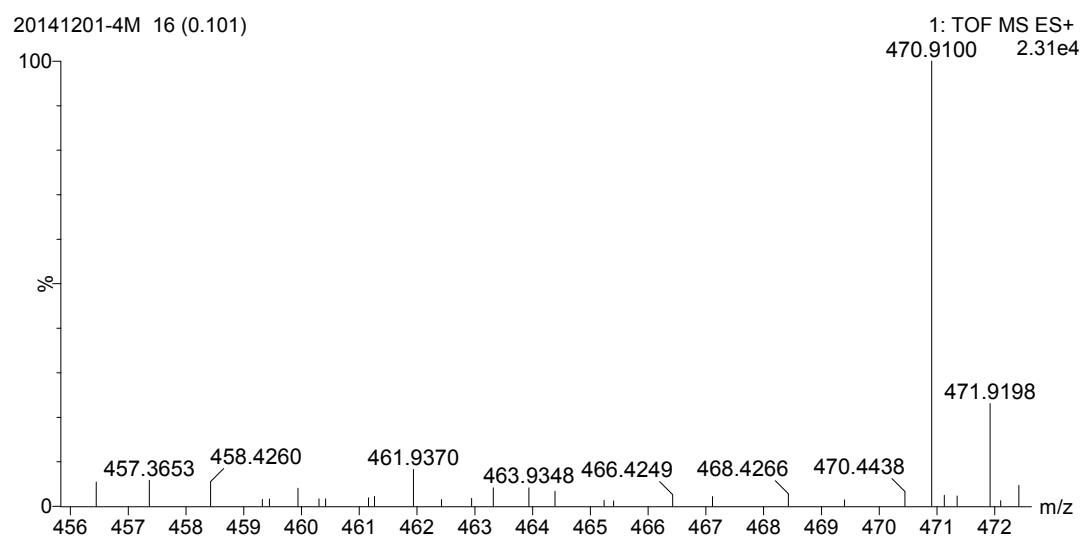
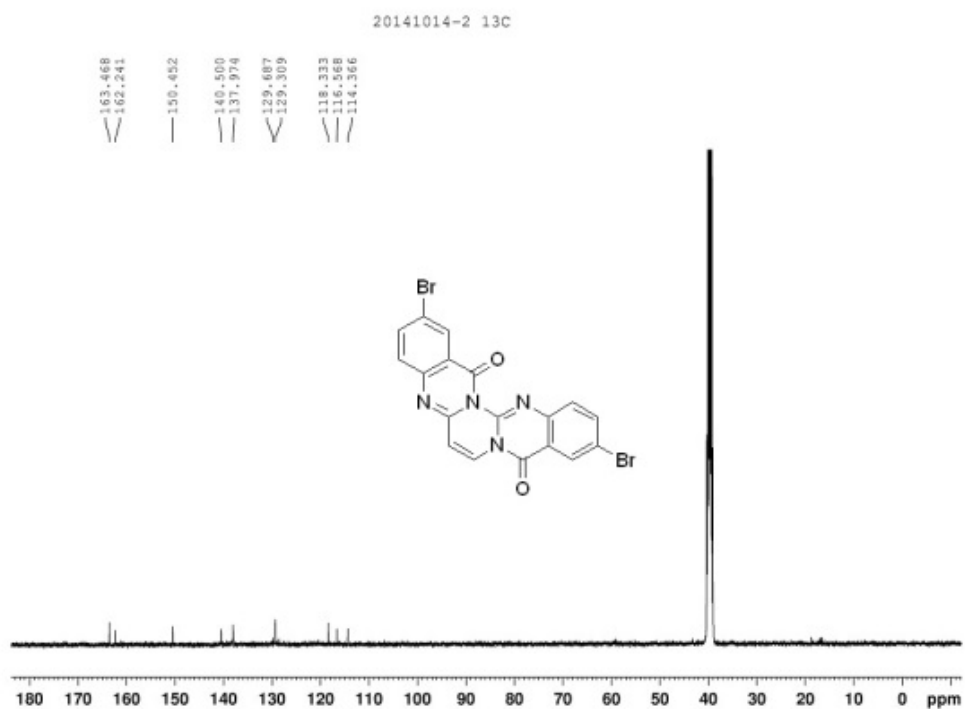
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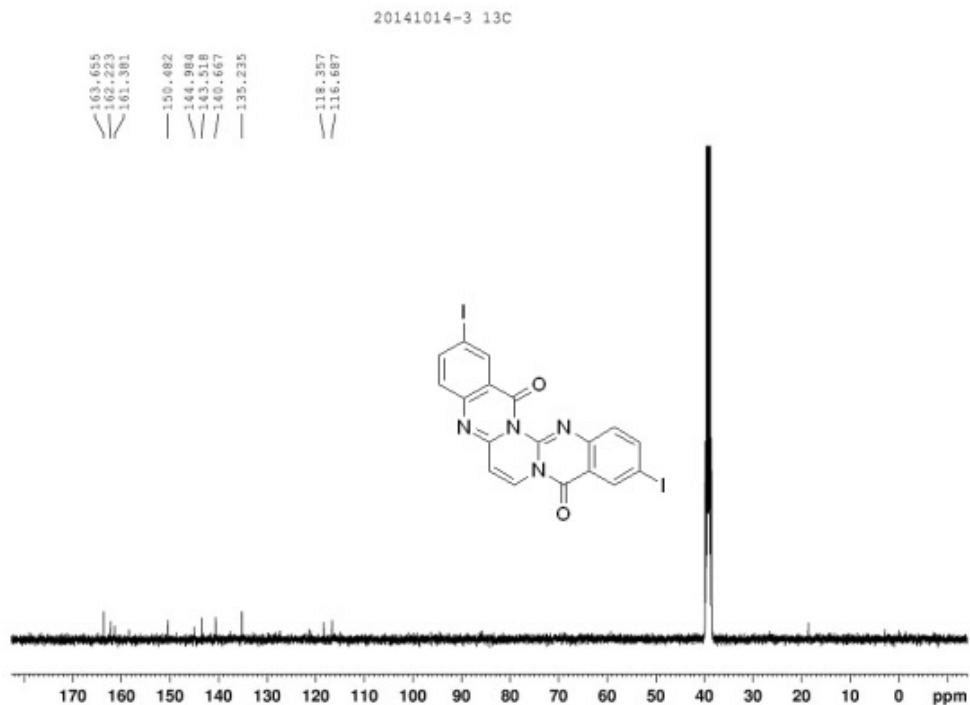
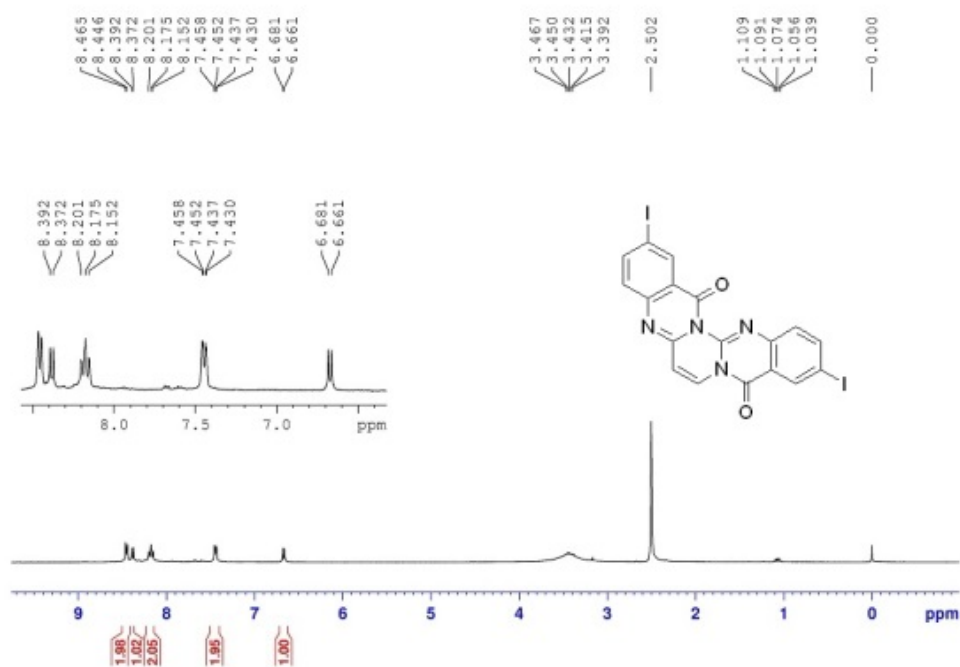
1: TOF MS ES+
1.34e3



6n

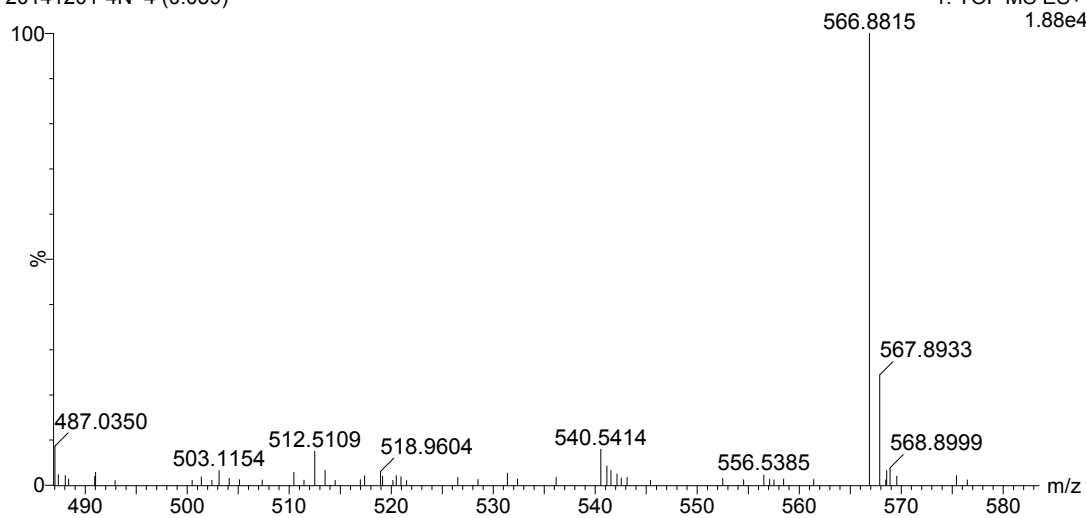




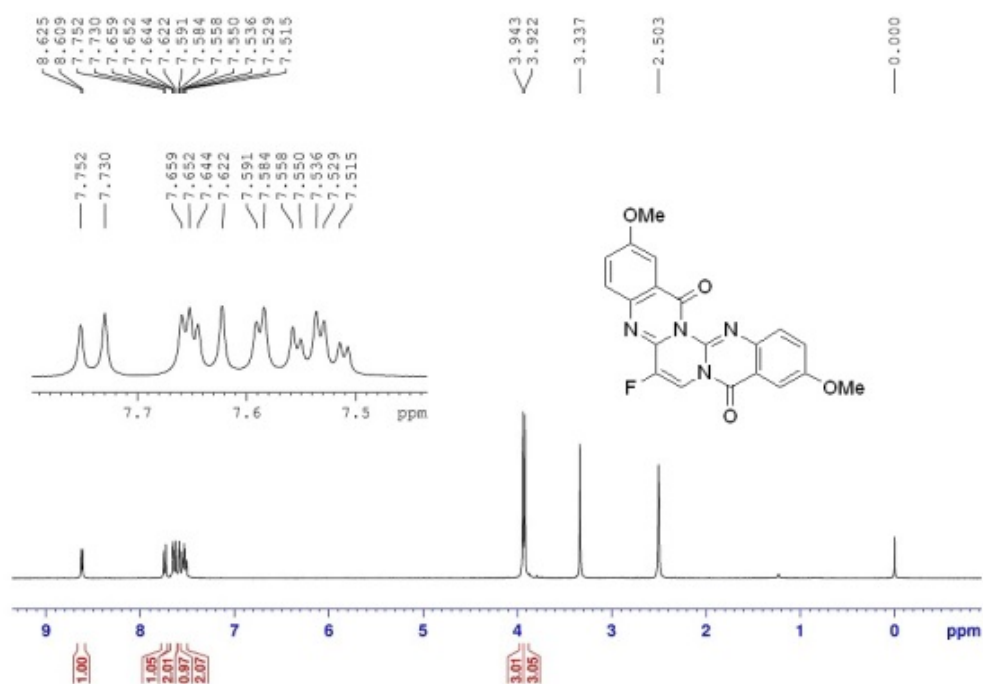


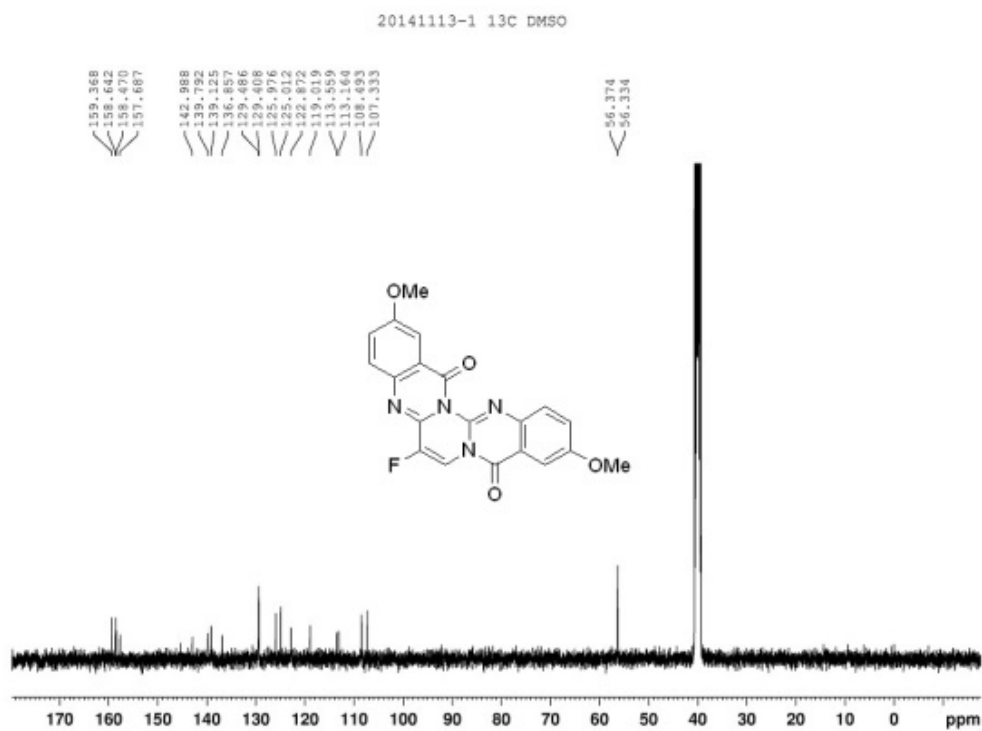
20141014-3 13C

1: TOF MS ES+
1.88e4

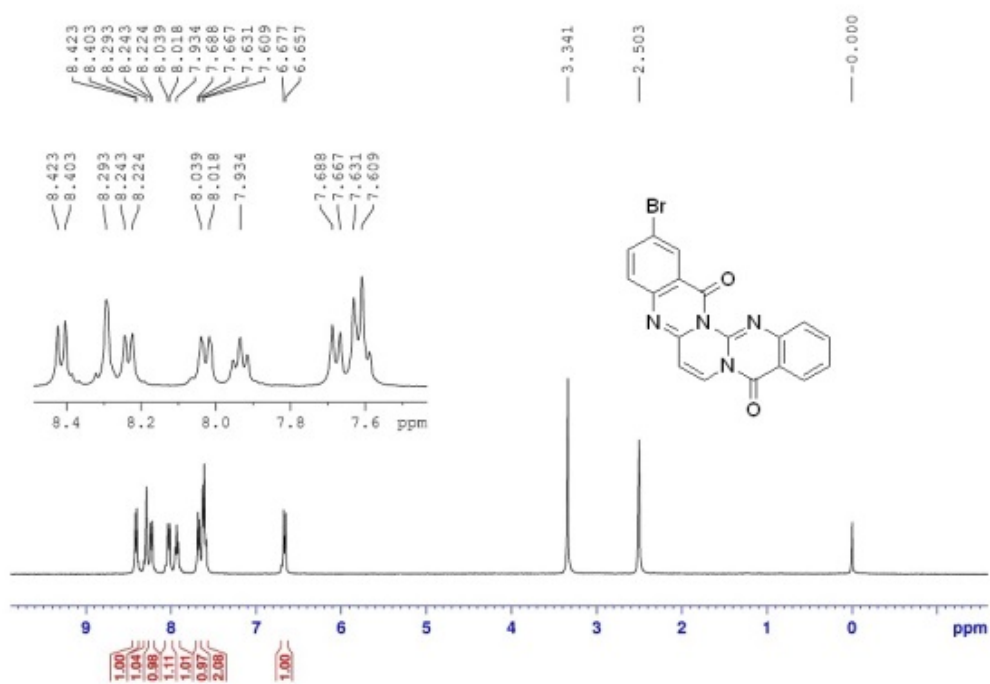


6p

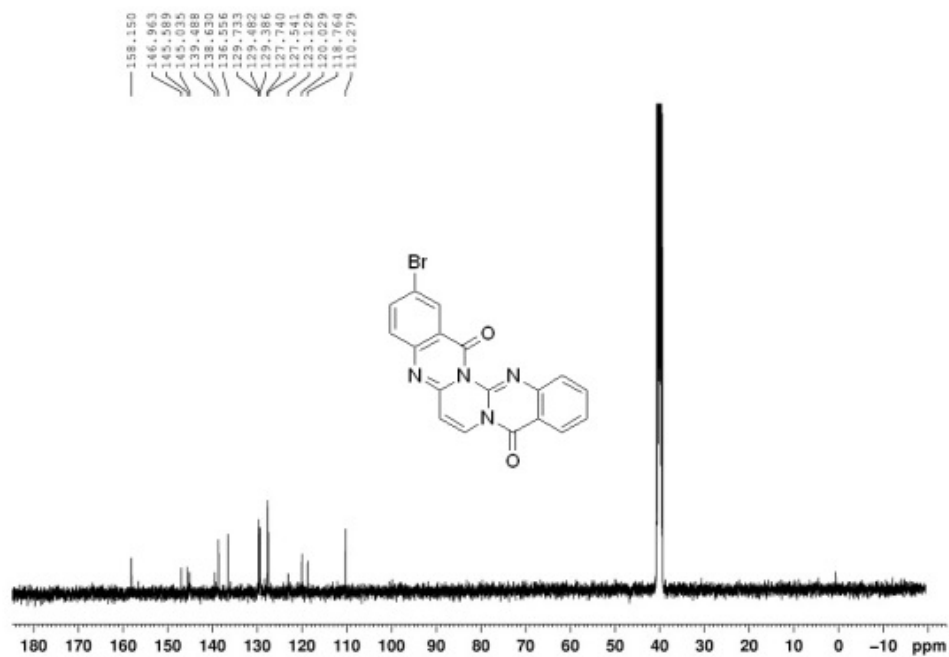




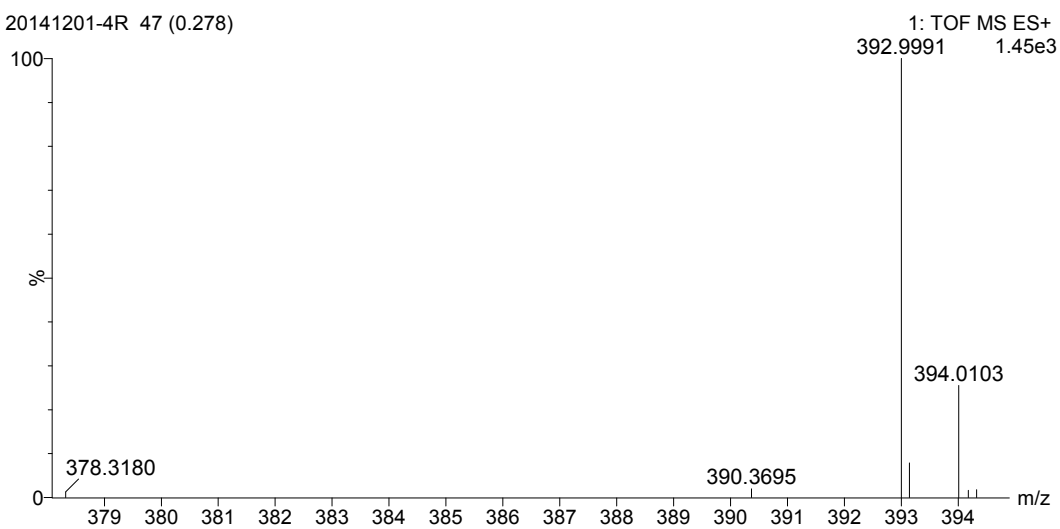
6qa



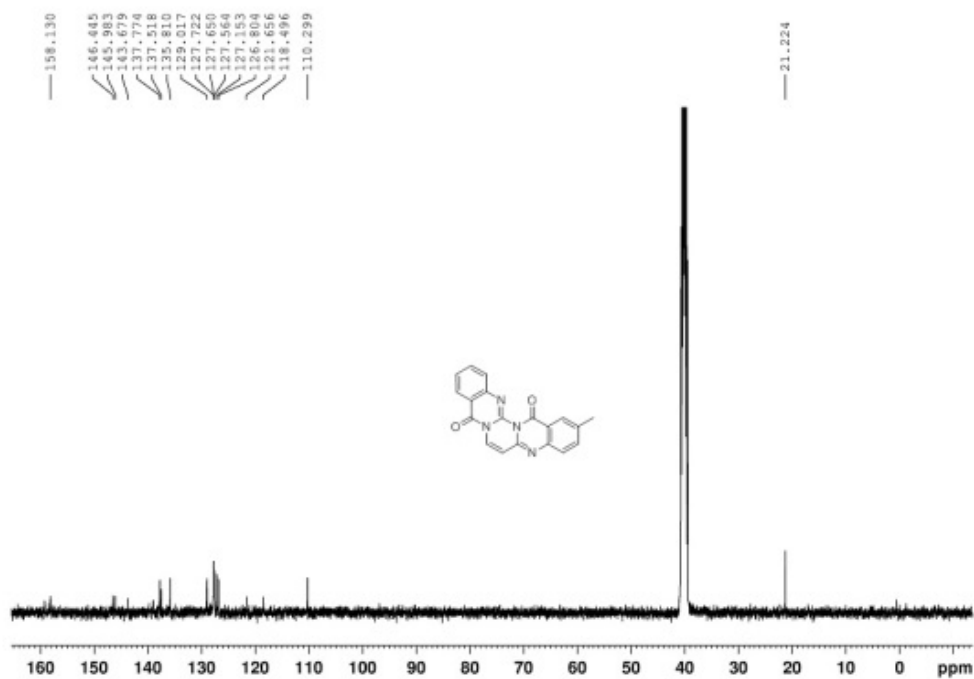
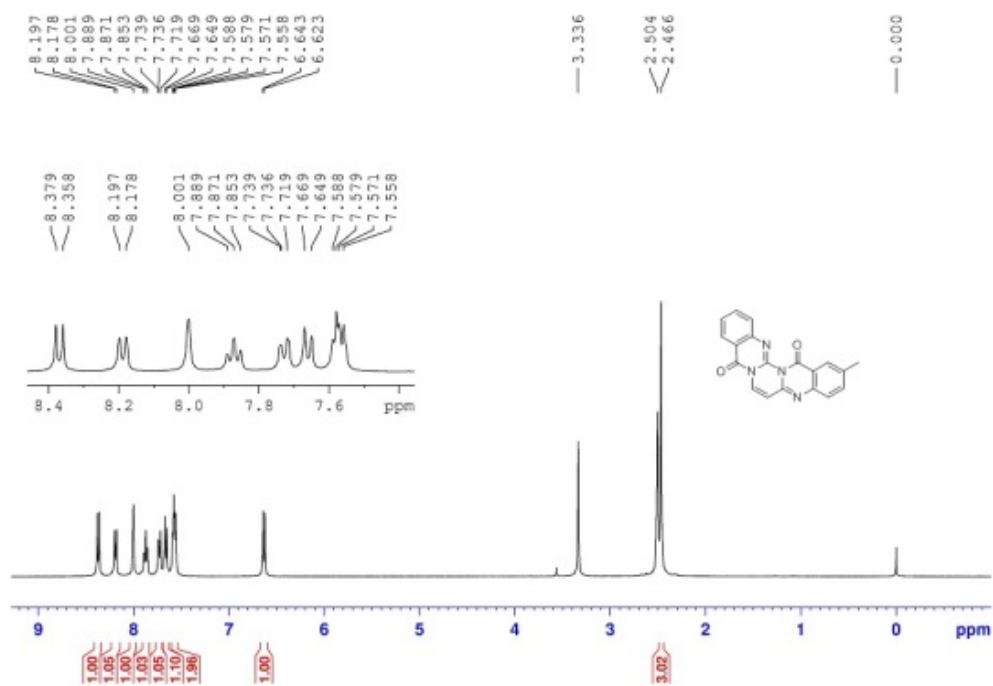
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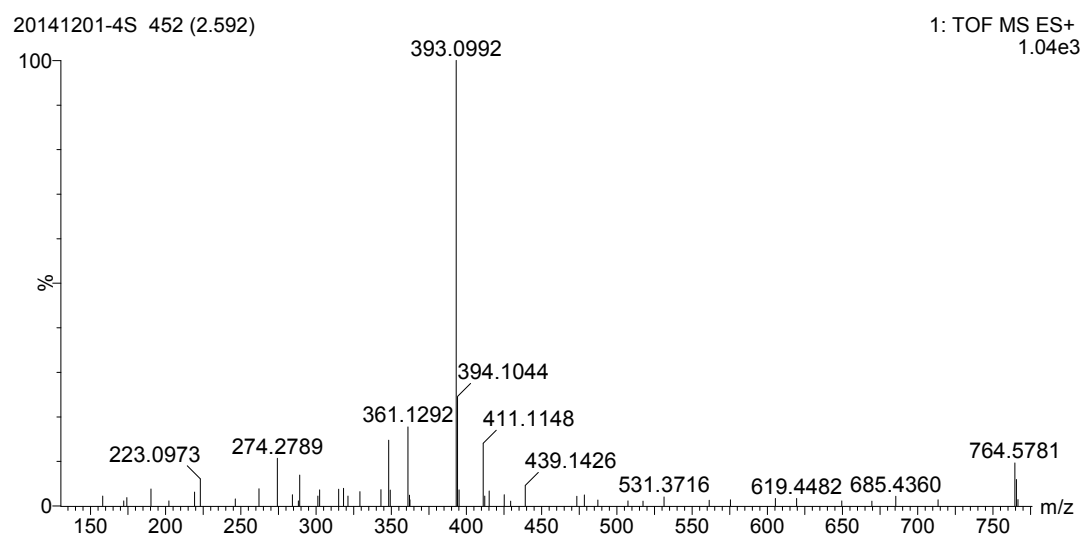
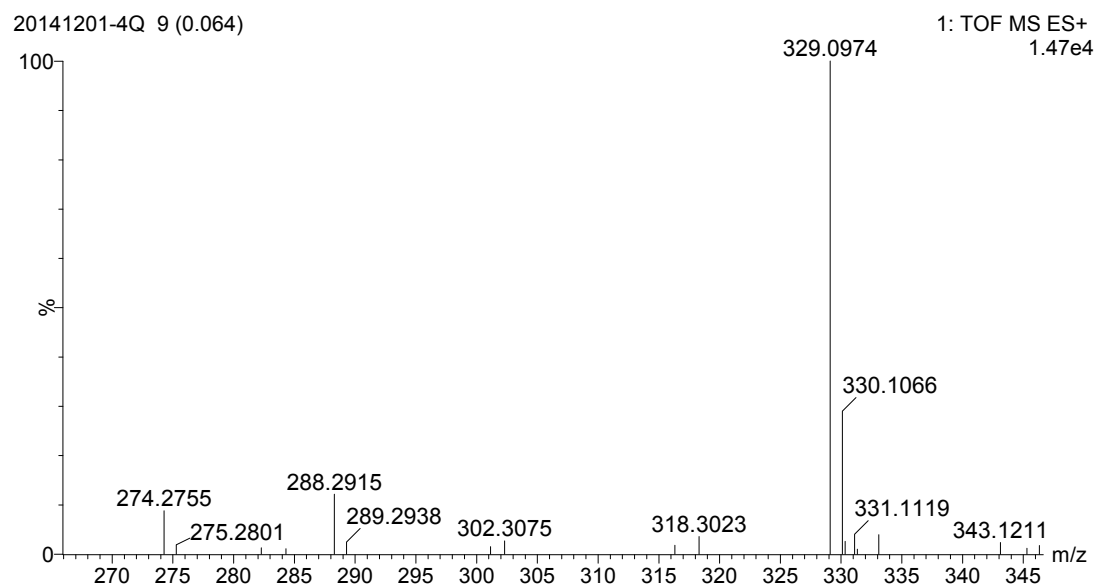


20141201-4R 47 (0.278)

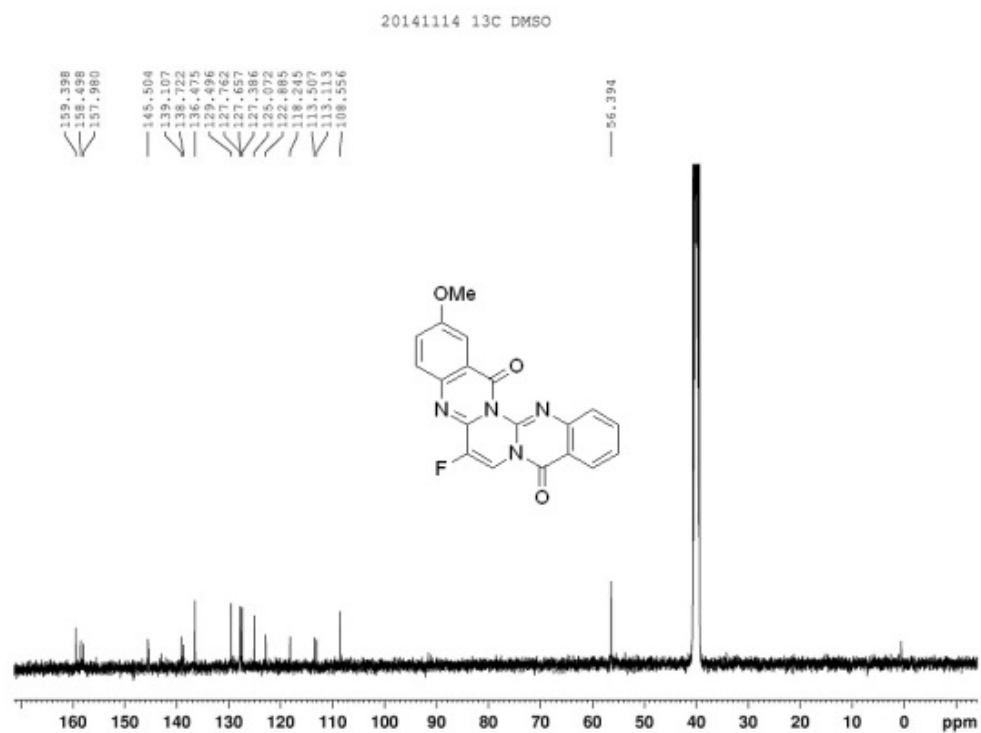
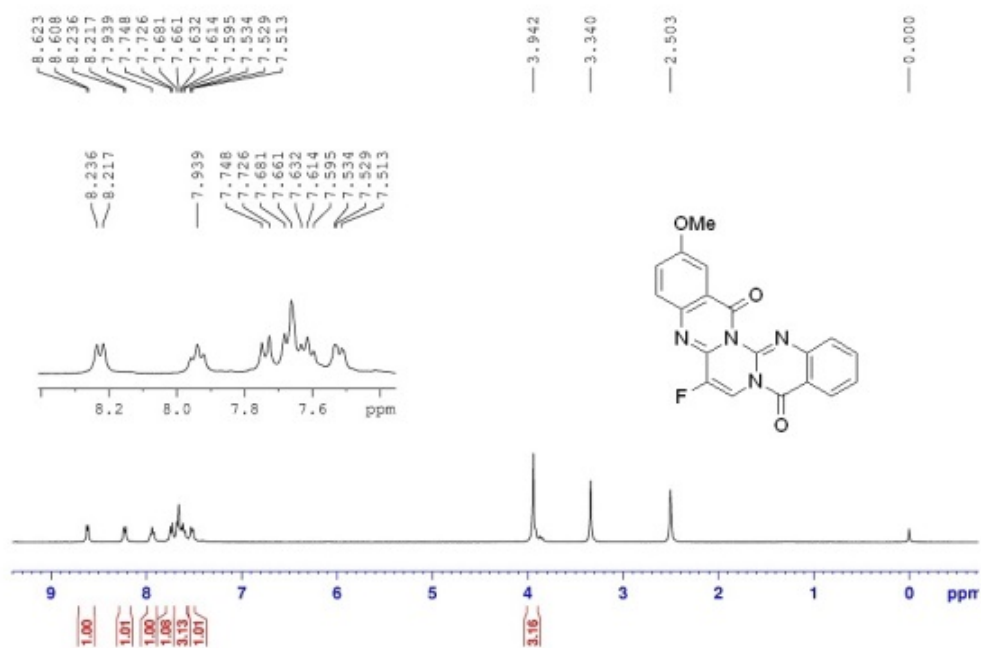


6r



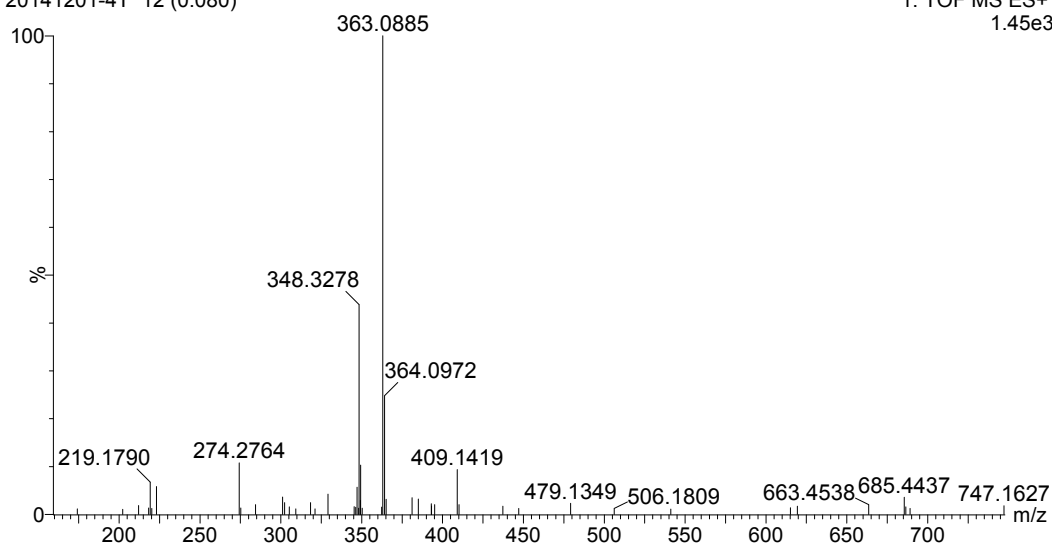


6s



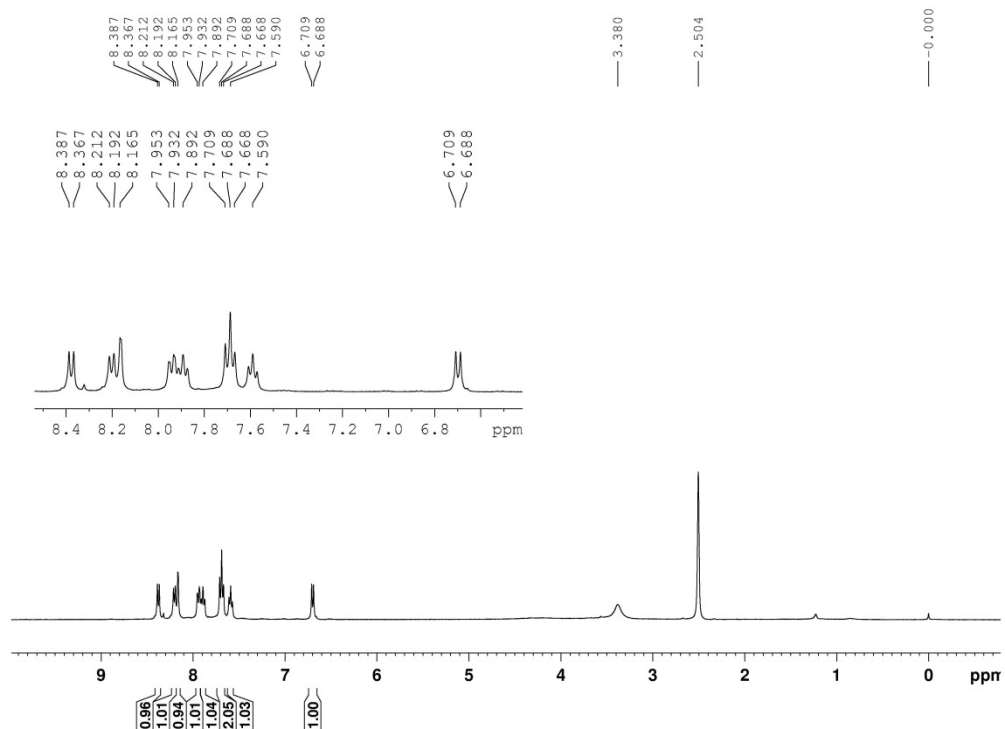
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1: TOF MS ES+
1.45e3

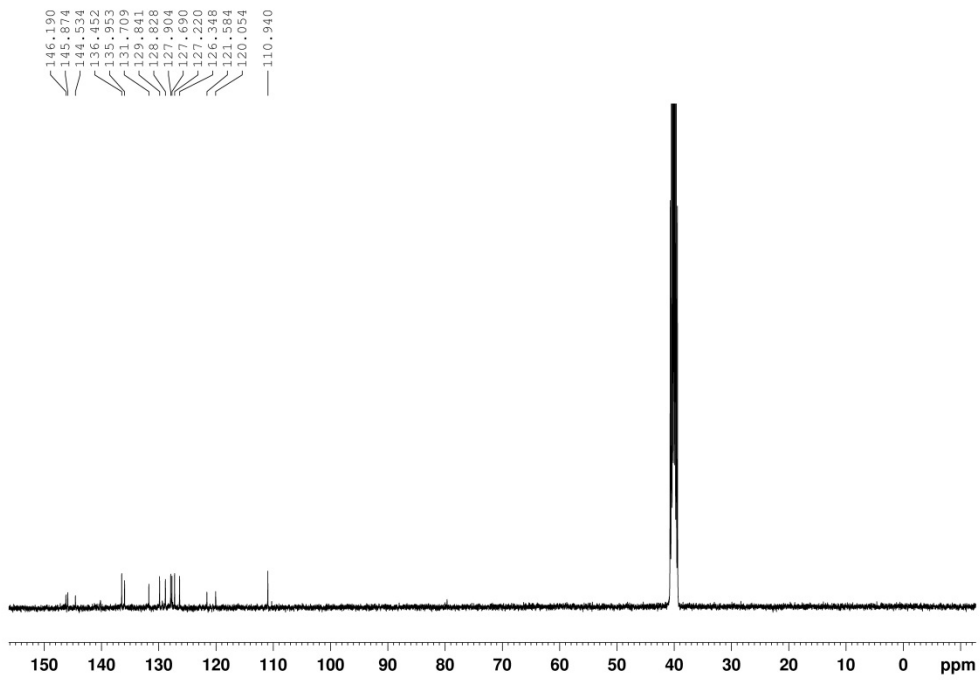


6t

20150128 DMSO



20150128 13C DMSO



20150131-LX-B+ 19 (0.136)

TOF MS ES+
1.43e3

