Cu(OTf)₂ catalyzed three component strategy for the synthesis of thienopyridine containing spirooxindoles and their cytotoxic evaluation

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IR, ¹H NMR, ¹³C NMR and mass spectra of selected compounds



¹H NMR spectrum of 5a

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¹³C NMR spectrum of 5a



REF:NMR/1580/64/24

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Ph.D. KP/S-II/03/5-CliSATRITHIMCA

¹H NMR spectrum of 5c

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¹³C NMR spectrum of 5c



Mass spectrum of 5c

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¹H NMR spectrum of 5i



¹³C NMR spectrum of 5i



¹H NMR spectrum of 5j



¹³C NMR spectrum of 5j



Mass spectrum of 5j



¹H NMR spectrum of 5k



¹³C NMR spectrum of 5k



Mass spectrum of 5k



IR spectrum of 5k



¹H NMR spectrum of 5l



¹³C NMR spectrum of 5l



Mass spectrum of 51

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IR spectrum of 51



¹H NMR spectrum of 50



¹³C NMR spectrum of 50



Mass spectrum of 50



¹H NMR spectrum of 5t



Expanded ¹H NMR spectrum of 5t



¹³C NMR spectrum of 5t



DEPT-135¹³C NMR of compound 5t



COSY spectrum of compound 5t



Expanded COSY spectrum of compound 5t







KEF:WPHK/161V/52/19

TOWARTER REMINISTER VELOCITION

HSQC spectrum of compound 5t



Expanded HSQC spectrum of compound 5t



Expanded HSQC spectrum of compound 5t



HMBC spectrum of compound 5t







Expanded HMBC spectrum of compound 5t



Expanded HMBC spectrum of compound 5t



IR spectrum of 5t



¹H NMR spectrum of 5u



¹³C NMR spectrum of 5u



Mass spectrum of 5u



¹H NMR spectrum of 5w



Expanded ¹H NMR spectrum of 5w



Expanded ¹H NMR spectrum of 5w

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¹³C NMR spectrum of 5w





IR spectrum of 5w



Mass spectrum of 5w



Numbering of compound 5t for NMR assignments

Chemical Shift δ (ppm)	Multiplicity	Number of Hydrogen(s)	Coupling constant ${}^{2}J_{\rm HH} \& {}^{3}J_{\rm HH}$ (Hz)	Assignment
2.08 & 2.69	Two doublets	2	14.8	24-CH ₂
2.64	Multiplet	4	-	16- & 17-CH ₂
3.44	Singlet	2	-	19-CH ₂
6.85-6.89	Multiplet	2	-	5- & 7-CH
6.95	Doublet	1	7.2	4-CH
6.99	Triplet	2	7.2	27- & 29-CH
7.17	Triplet	1	7.6	6-CH
7.26	Broad signal (D ₂ O exchangeable)	2	-	22-NH ₂
7.34-7.41	Multiplet	1	-	28-CH
10.67	Broad signal (D ₂ O exchangeable)	1	-	1-NH

¹H NMR characteristics of compound **5t** recorded in DMSO- d_6

Chemical shift at δ 2.50 ppm and at δ 3.35 ppm are due to DMSO- d_6 and solvent water respectively.

Additional signals at δ 1.04, δ 3.41 and δ 4.35 ppm are due to ethanol used for recrystallization. The numbering is given for NMR assignments only.

Group/Carbon	Chemical shift
Gloup/Carbon	δ (ppm)
C-17 & C-19	24.6 & 49.9
C-24	47.6
C-16	48.1
C-3	49.2
C-10	55.7
C-7	110.1
C-21	110.9
*C-27 & C-29	111.2 & 111.4
*C-25	111.7, 111.9 & 112.1
C-23	117.9
C-15	121.6
C-5	122.2
C-4	124.4
C-20	127.9
C-6	129.2
*C-28	129.7, 129.8 & 129.9
C-9	131.9
C-8	141.2
C-13	148.8
*C-26 & C-30	159.7, 159.8, 162.1 & 162.2
C-11	160.8
2-C=O	177.1

¹³ C NMR characteristics of compound 5t recorded in D	$MSO-d_6$
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Chemical shift around δ 39.5 ppm is due to the solvent, DMSO- d_6 . *Splitting of signals is due to ¹³C-¹⁹F coupling.

Empirical formula	$C_{52}H_{38}F_4N_8O_6S_2$
Formula weight	1011.02
Temperature	298(2) K
Wavelength	0.71073 Å
Crystal system, space group	Triclinic, P-1
Unit cell dimensions	a = 12.8051(5) Å; $b = 12.8767(5)$ Å; c
	= 15.6983(6) Å; α = 66.206(2)°; β =
	$88.780(2)^{\circ}; \gamma = 81.531(2)^{\circ}$
Volume	$2340.53(16) \text{ Å}^3$
Z, calculated density	2, 1.435 mg m ⁻³
Absorption coefficient	0.192 mm^{-1}
F(000)	1044
Crystal size	0.25 x 0.22 x 0.12 mm
θ range for data collection	$1.42 - 25.00^{\circ}$
Limiting indices	$-15 \le h \le 15, -15 \le k \le 15, -18 \le l \le 18$
Reflections collected /unique	26180/7948 [R(int) = 0.0367]
Completeness to $\theta = 25.00$	96.4 %
Absorption correction	Multi-scan
Refinement method	Full-matrix least-squares on F ²
Data/restraints/parameters	7948/3/664
Goodness-of-fit on F^2	1.349
Final R indices [I ² sigma(I)]	R1 = 0.0628, $wR2 = 0.1751$
R indices (all data)	R1 = 0.0977, $wR2 = 0.2157$
Largest diff. peak and hole	0.971 and -0.486 e.Å ⁻³
CCDC	936,526

Crystal data, data collection and refinement parameters of compound $\mathbf{5t}$