

SUPPORTING INFORMATION

Atomistic insight into the oxidation of monolayer transition metal dichalcogenides: from structures to electronic properties

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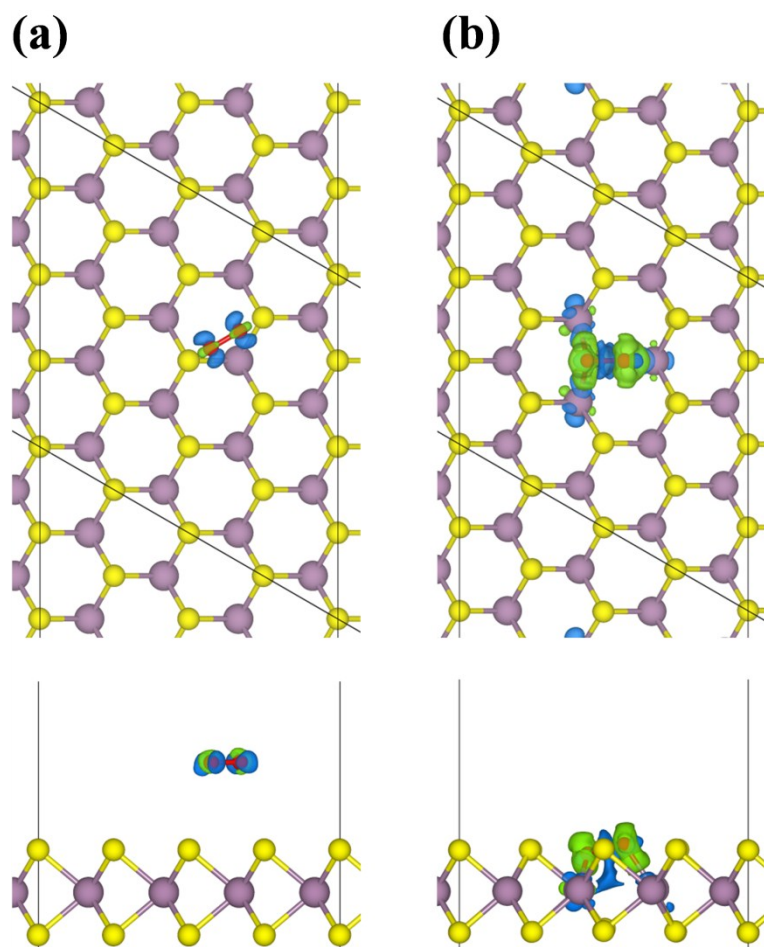


Figure S1. Charge density difference for O₂ adsorbed on (a) perfect and (b) defective MoS₂ monolayer. The iso-surface for electron density is $2 \times 10^{-4} \text{ e}/\text{\AA}^3$ for (a) and $6 \times 10^{-3} \text{ e}/\text{\AA}^3$ for (b). Green is charge accumulation and blue is charge depletion. The charge density difference is calculated as $\text{Charge}_{\text{total}} - \text{Charge}_{\text{O}_2} - \text{Charge}_{\text{sub}}$, where $\text{Charge}_{\text{total}}$ is the total charge density of the systems, $\text{Charge}_{\text{O}_2}$ is the charge density of O₂ and $\text{Charge}_{\text{sub}}$ is the charge density of perfect or defective MoS₂ monolayer.

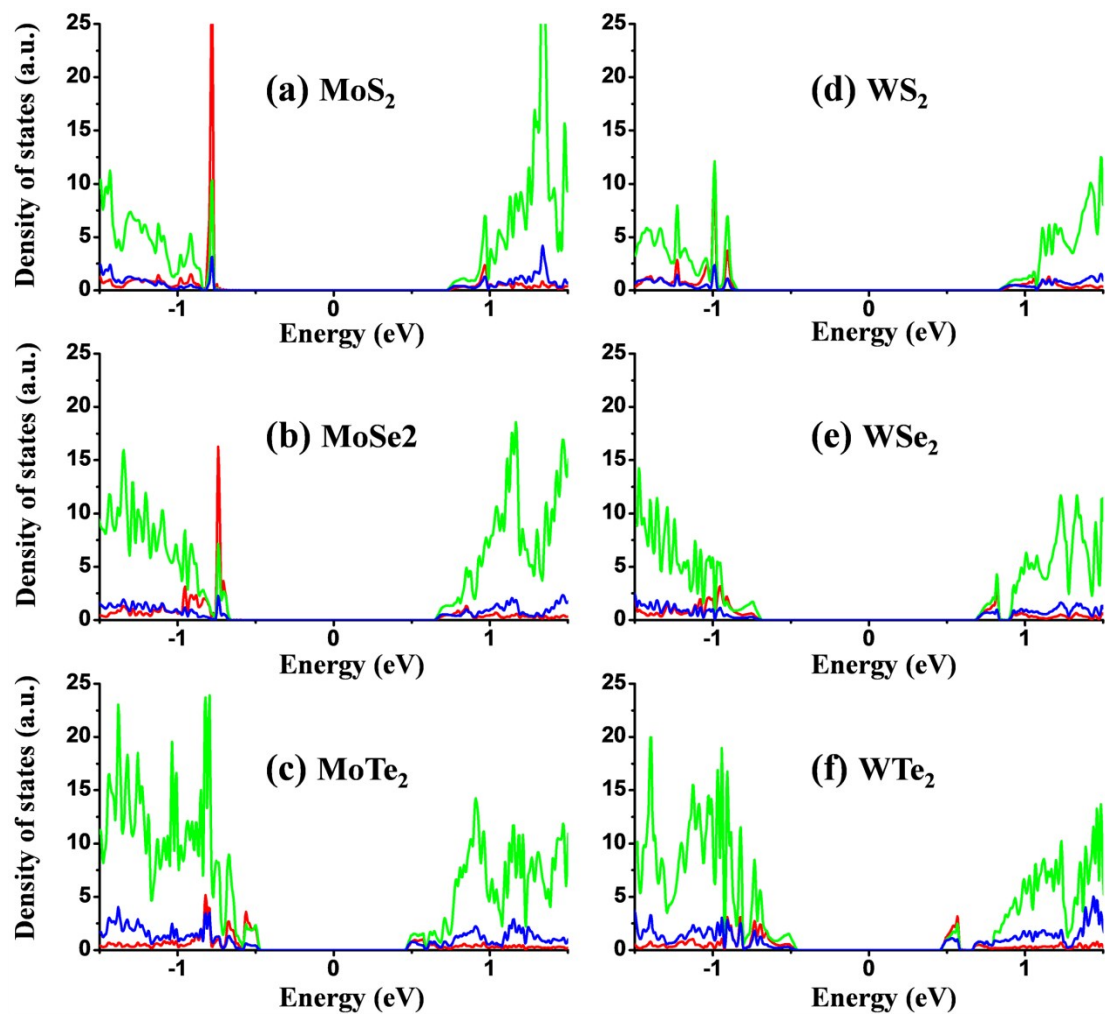


Figure S2. Partial density of states (PDOS) for all the six defective TMD monolayers with O_2 adsorption at vacancy sites. Red lines represent states from O_2 . Green lines represent states from chalcogen (S, Se or Te). Blue lines represent states from transition metal atoms.