

Largest Discrete Supertetrahedral Clusters Synthesized in Ionic Liquids

Wei-Wei Xiong, Jian-Rong Li, Bing Hu, Bin Tan, Ren-Fu Li, and Xiao-Ying Huang*

Supporting information

1. More structure figures and tables

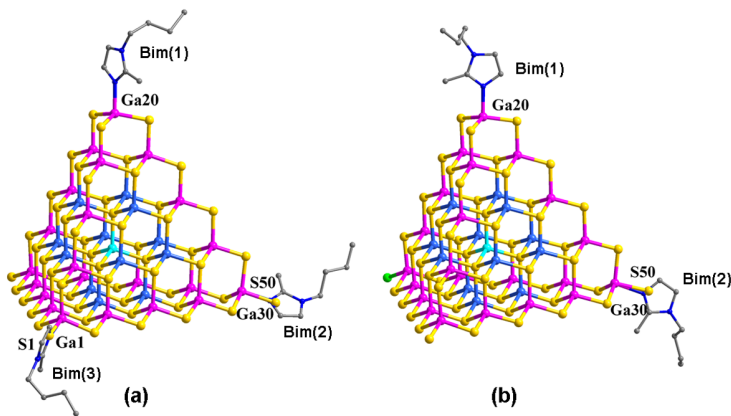


Figure S1. The T5 $[\text{Cu}_5\text{Ga}_{30}\text{S}_{52}(\text{SH})_2(\text{Bim})_2]^{11-}$ and $[\text{Cu}_5\text{Ga}_{30}\text{S}_{52}(\text{SH})_{1.5}\text{Cl}(\text{Bim})_{1.5}]^{11.5-}$ clusters in compounds **3** (a) and **4** (b) with numbering scheme of the terminal ligands. Hydrogen atoms are omitted for clarity.

Table S1. The terminal M-SH and M-Cl (M = Ga, In) bond lengths (Å) for compounds **1-4**.

Compound 1		Compound 2		Compound 3		Compound 4	
In(1)-X(1)	2.447(2)	Ga(1)-S(1)	2.313(5)	Ga(1)-S(1)	2.304(10)	Ga(1)-Cl(1)	2.256(4)
(X(1) = S(54)/Cl(1))							
In(20)-X(2)	2.449(3)	Ga(9)-S(16)	2.311(2)	Ga(30)-S(50)	2.288(5)	Ga(30)-S(50)	2.288(7)
(X(2) = S(55)/Cl(2))							
In(30)-X(3)	2.462(2)			Ga(34)-S(55)	2.293(3)	Ga(34)-S(55)	2.274(3)
(X(3) = S(49)/Cl(3))							
In(34)-X(4)	2.447(4)						
(X(4) = S(56)/Cl(4))							

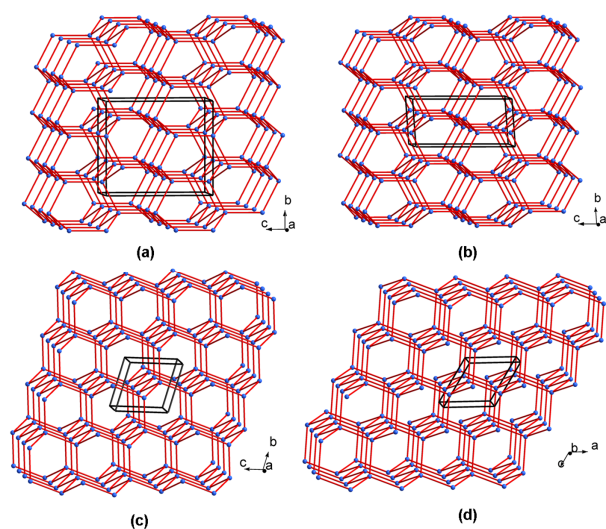


Figure S2. By connecting the barycenters of the discrete T5 clusters in compounds **1-4**, the topological type of the arrangements of **1** (a) and **2** (b) is distorted hexagonal diamond superlattice, while that of **3** (c) and **4** (d) is distorted cubic diamond lattice.

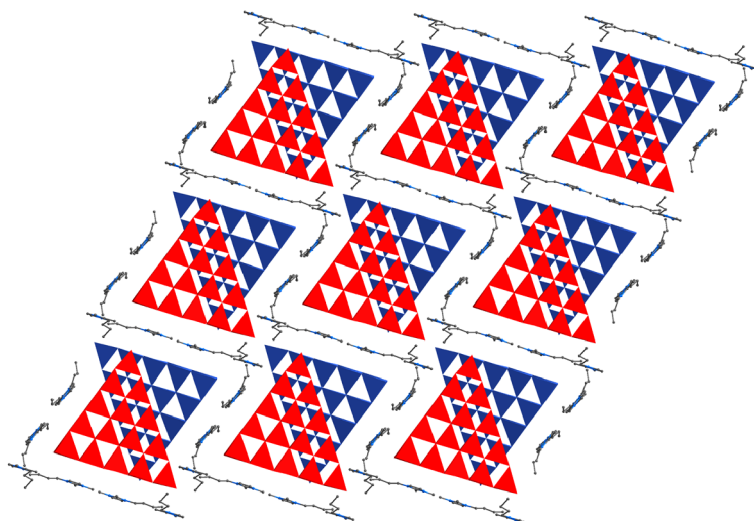


Figure S3. Taking the structure of compound **1** as an example, most of the $[\text{Bmmim}]^+$ cations are located between the tetrahedral faces of two T5 clusters, and the imidazolium rings of $[\text{Bmmim}]^+$ cations are parallel to the tetrahedral faces of T5 clusters.

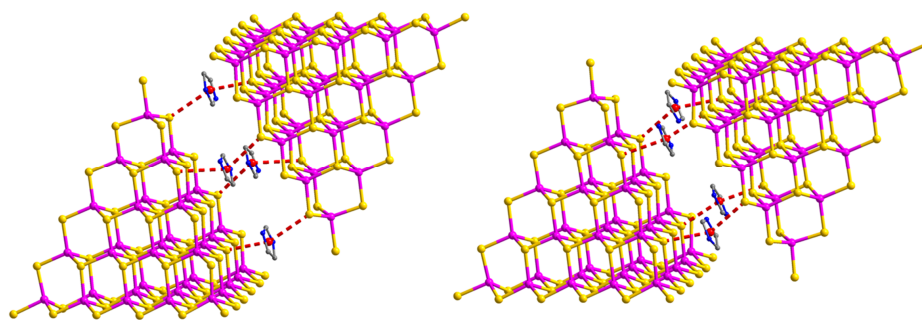


Figure S4. Taking the structure of compound **1** as an example, the anion- π interactions between the anions on the surfaces of T5 clusters and the imidazolium rings are showed as dotted lines.

2. More characterizations

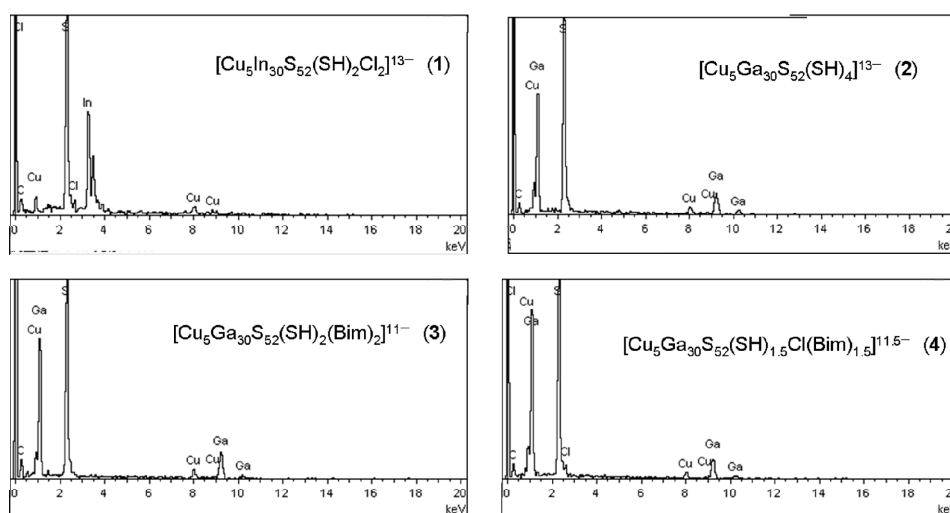


Figure S5. The energy dispersive X-ray (EDX) spectroscopy of compounds **1-4**. The energy dispersive X-ray (EDX) spectroscopy was performed on an Oxford INCA instrument.

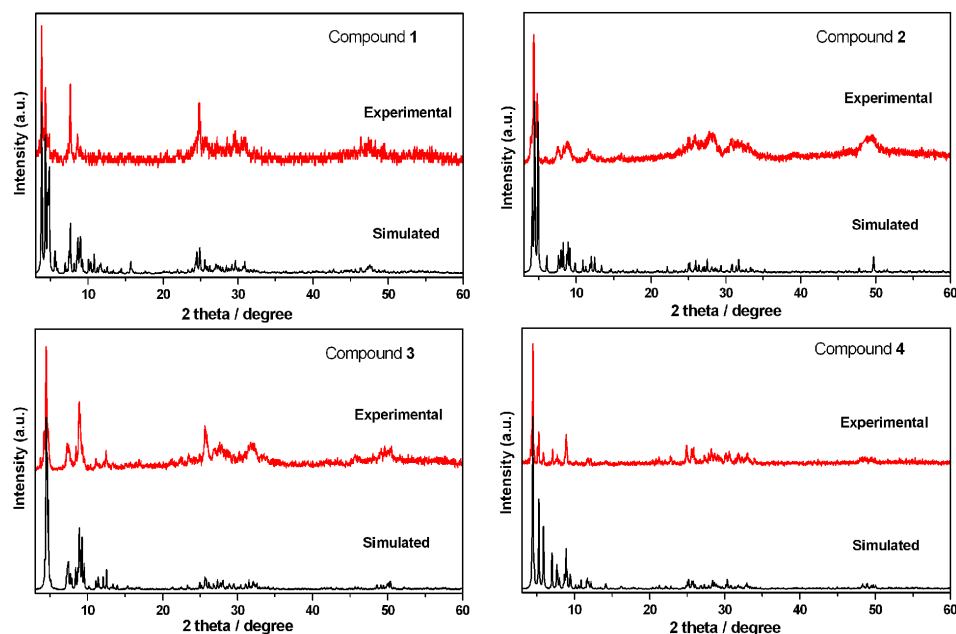


Figure S6. The experimental (red) and simulated (black) PXRD patterns (red) of compounds **1-4**. Powder X-ray diffraction (PXRD) patterns were recorded on a Rigaku MiniFlex II using CuK α radiation.

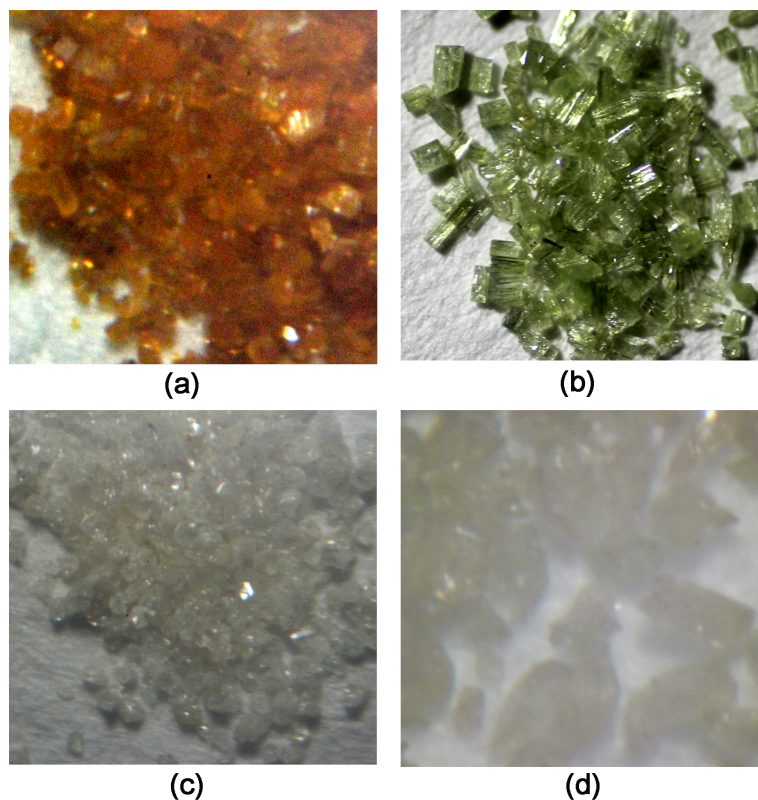


Figure S7. Photographs of the crystals of compounds **1** (a), **2** (b), **3** (c), **4** (d).