

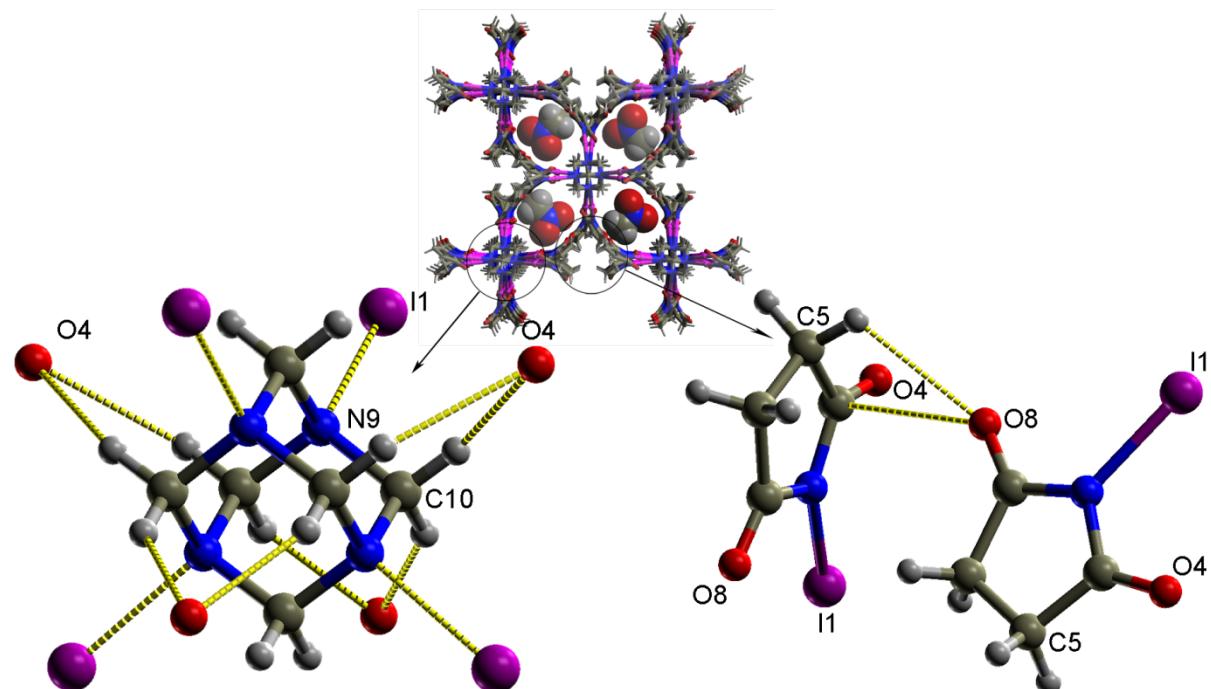
SUPPLEMENTARY INFORMATION

Breathing Molecular Crystals: Halogen and Hydrogen Bonded Porous Molecular Crystals with Solvent Induced Adaptation of the Nano-sized Channels

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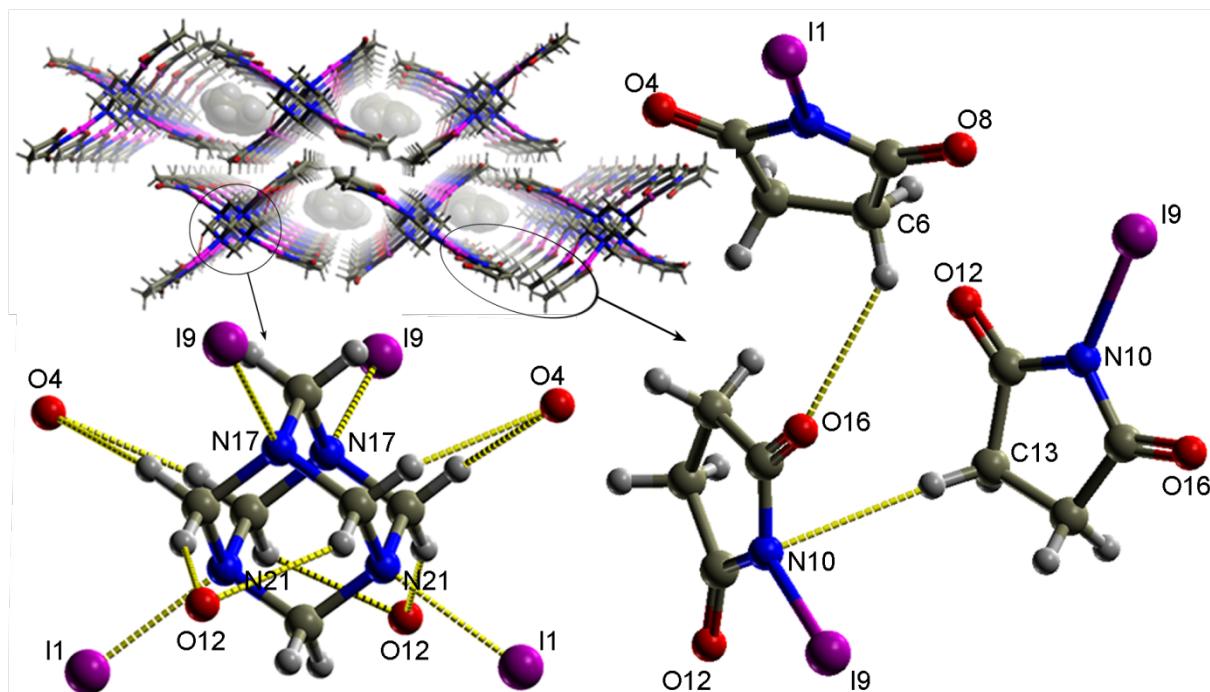
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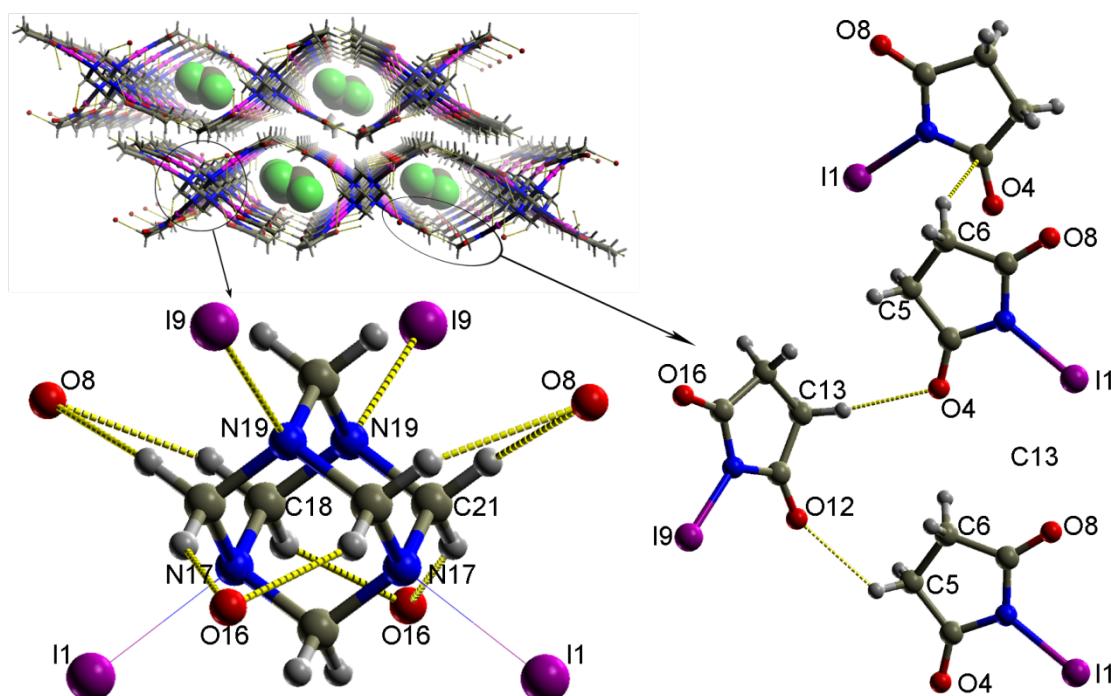
H-bond	d [Å]	C–H ... O [°]	X-bond	d [Å]	C–H ... O [°]
C10–H ... O4	3.189	152.2	N2–I1...N9	2.541	175.5
C10–H ... O4	3.333	144.8			
C5–H ... O8	3.134	114.3			

Scheme S1. Hydrogen and halogen bonding in **II@CH₃NO₂**



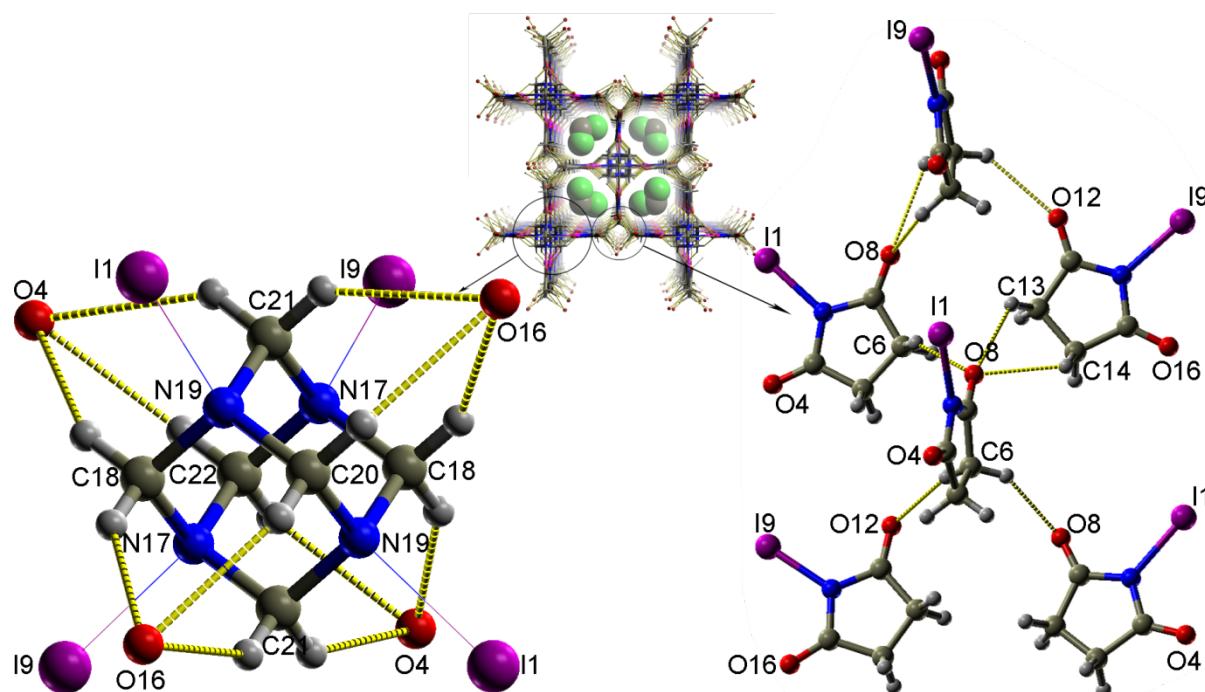
H-bond	d [Å]	C–H ••• O [°]	X-bond	d [Å]	C–H ••• O [°]
C19–H ••• O4	3.345	139.8	N10–I9•••N17	2.541	176.8
C20–H ••• O4	3.062	155.1	N2–I1•••N21	2.563	174.6
C19–H ••• O12	3.314	148.2			
C20–H ••• O12	3.335	148.1			
C13–H ••• N10	3.689	169.2			
C6–H ••• O16	3.408	136.0			

Scheme S2. Hydrogen and halogen bonding in $\mathbf{I}@\mathrm{C}_6\mathrm{H}_5\mathrm{CH}_3$



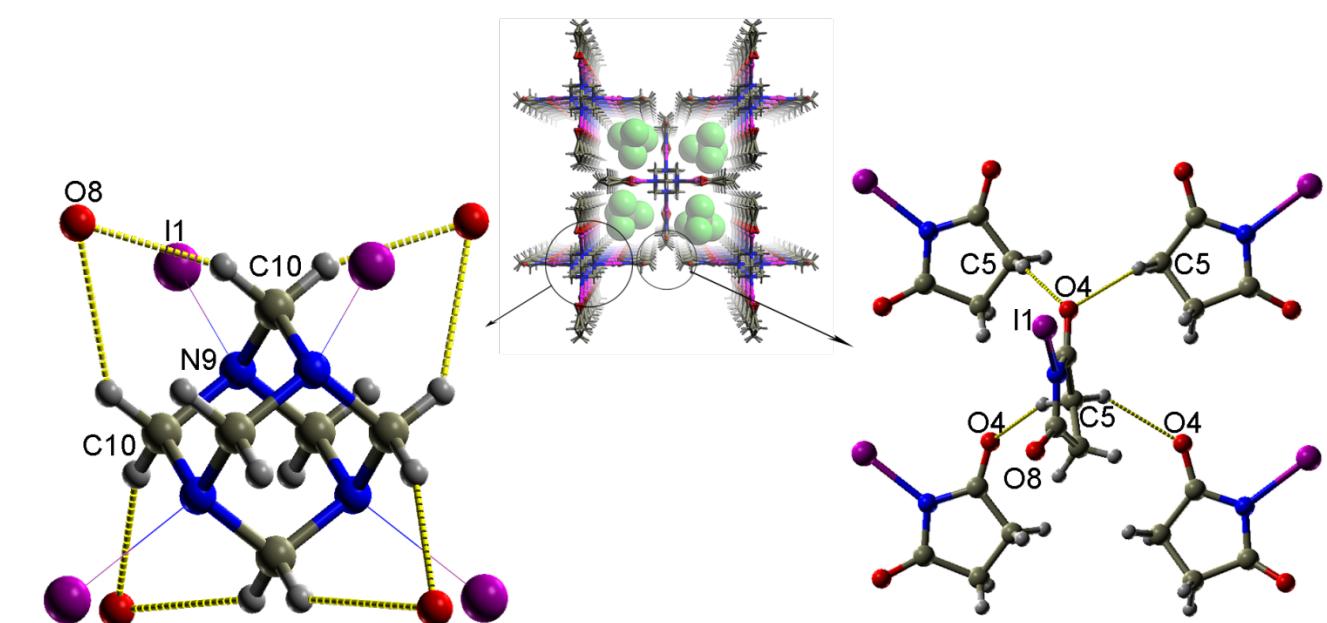
H-bond	d [Å]	C–H ••• O [°]	X-bond	d [Å]	C–H ••• O [°]
C21–H ••• O8	3.223	148.8	N2–I1•••N17	2.523	176.7
C18–H ••• O8	3.290	145.5	N10–I9•••N19	2.586	176.7
C21–H ••• O16	3.185	144.6			
C18–H ••• O16	3.135	146.9			
C13–H ••• O4	3.323	172.0			
C5–H ••• O12	3.088	120.1			

Scheme S3. Hydrogen and halogen bonding in I@CHCl₃



H-bond	d [Å]	C–H •••O [°]	X-bond	d [Å]	C–H •••O [°]
C22–H •••O4	3.240	146.3	N2–I1•••N19	2.516	175.2
C21–H •••O4	3.427	138.2	N10–I9•••N17	2.486	179.5
C18–H •••O4	3.223	147.0			
C18–H •••O16	3.223	146.6			
C20–H •••O16	3.267	145.3			
C21–H •••O16	3.383	140.4			
C6–H •••O8	3.275	146.6			
C6–H •••O12	3.364	162.8			
C13–H •••O8	3.270	126.2			
C14–H •••O8	3.261	126.0			

Scheme S4. Hydrogen and halogen bonding in **III**@CH₂Cl₂



<i>H</i> -bond	<i>d</i> [\AA]	<i>C</i> – <i>H</i> \cdots <i>O</i> [\square]	<i>X</i> -bond	<i>d</i> [\AA]	<i>C</i> – <i>H</i> \cdots <i>O</i> [\square]
C10–H \cdots O8	3.079	153.7	N2–I1 \cdots N9	2.520	179.6
C10–H \cdots O8	3.329	140.2			
C5–H \cdots O4	3.394				

Scheme S5. Hydrogen and halogen bonding in **III**@ CCl_4