## Electronic supplementary information (ESI)

## Pt-Cu nanoctahedra: synthesis and comparative study with nanocubes on their electrochemical catalytic performance

Jun Zhang,<sup>a</sup> Hongzhou Yang,<sup>b</sup> Benjamin Martens,<sup>c</sup> Zhiping Luo,<sup>d</sup> Dan Xu,<sup>a</sup> Yuxuan Wang,<sup>c</sup> Shouzhong Zou,<sup>\*b</sup> and Jiye Fang<sup>\*a,c</sup>

 <sup>a</sup>Department of Chemistry, State University of New York at Binghamton, Binghamton, New York 13902, USA
<sup>b</sup>Department of Chemistry and Biochemistry, Miami University, Oxford, Ohio 45056, USA
<sup>c</sup>Materials Science and Engineering Program, State University of New York at Binghamton, Binghamton, New York 13902, USA
<sup>d</sup>Microscopy and Imaging Center and Materials Science and Engineering Program, Texas A&M
University, College Station, Texas 77843, USA; present address: Department of Chemistry and Physics, Fayetteville State University, Fayetteville, NC 28301, USA

jfang@binghamton.edu; zous@muohio.edu





Figure S1. Typical TEM image of Pt-Cu nanoctahedron in a multilayer-assembly pattern.

**Figure S2.** Typical EDS spectra of Pt-Cu nanoctahedra at different locations on a Ni-grid and their analysis results.













Processing option: All elements analyzed (Normalized)

All results in atomic%

Spectrum	In stats.	Cu	Pt
Spectrum 1	Yes	33.15	66.85
Spectrum 2	Yes	34.68	65.32
Spectrum 3	Yes	33.83	66.17
Spectrum 4	Yes	32.16	67.84
Spectrum 5	Yes	33.41	66.59
Spectrum 6	Yes	32.80	67.20
Mean		33.34	66.66
Std. deviation		0.87	0.87
Max.		34.68	67.84
Min.		32.16	65.32

**Figure S3.** TEM image of Pt-Cu spherical nanoparticles prepared using  $Cu(II)(acac)_2$  as the Cu-source under otherwise identical conditions for Pt-Cu nanoctahedron synthesis..



**Figure S4.** TEM images of Pt-Cu nanoparticles for tuning the particle morphology as a function of input  $Cu^{2+}/Cu^{+}$  molar ratio while other synthesis conditions were kept the same: (a), 8:1; (b), 6:1; (c), 5:1; (d), 4:1; (e), 1:1; (f), 1:6; (g), 1:10; (h), 0:1. Data bar: 100 nm.



Figure S5. TEM image of Pt-Cu nanocrystals prepared using oleylamine as the reaction solvent without oleic acid, while the other conditions were same as those of typical Pt-Cu nanoctahedron synthesis.





Figure S6. TEM image of Pt<sub>60</sub>Cu<sub>40</sub> nanocubes used for electrochemical study.

Figure S7. TEM image of the Pt-Cu nanoctahedra after an electrochemical measurement.

