Electronic Supplementary Information for

A Fluorescent Probe for the Rapid Detection of Hydrogen Sulfide in Blood Plasma and Brain Tissues in Mice

Yong Qian, ^a Ling Zhang, ^{ab} Shuting Ding, ^a Xin Deng, ^c Chuan He, ^c Hai-Liang Zhu, ^a and Jing Zhao*^a

^aInstitute of Chemistry and Biomedical Sciences, State Key Laboratory of Pharmaceutical Biotechnology, School of Life Sciences, Nanjing University, Nanjing 210093, China.

^bSchool of Pharmacy, Xuzhou Medical College, 84 West Huaihai Road, Xuzhou, Jiangsu 221002, China.

^cDepartment of Chemistry and Institute for Biophysical Dynamics, The University of Chicago, 929 East 57th Street, Chicago, Illinois 60637, USA.

Table of Content:

General Information

Figure S1. HRMS identification of SFP-3 probe (8) and the sulfide addition product of SFP-3 probe 8a

Figure S2. Fluorescence spectra of SFP-3 incubated with Na₂S in PBS after different incubation times

Figure S3. Fluorescence spectra of SFP-3 in PBS buffer incubated with different concentrations of Na₂S

Figure S4. Fluorescence spectra of SFP-3 probe incubated with various sulfur sources or amino acid

Figure S5. Fluorescence spectra of SFP-3 measured in the presence H₂S

Figure S6-S7. Fluorescence spectra of the SFP-3 probe incubated with Na₂S in bovine serum

Figure S8. Determination of H₂S in spiked mouse blood plasma

Figure S9. Determination of H₂S in fresh blood plasma

Figure S10. Determination of H₂S in fresh brain tissues

Synthesis and Characterization of compounds

Quantum Yields

Detection of H₂S protocols

References

NMR spectrum of compounds

General Information: All reagents were purchased from Sigma-Aldrich. All solvents were distilled before use unless otherwise noted. Eight-week old male C57BL6/J mice were purchased from The Jackson Laboratory (Bar Harbor, ME) and Yangzhou University. Phosphate-Buffered Saline PBS, Cat# BM-220 was purchased from Boston Bioproducts. Chromatographic purification of products was accomplished by using forced-flow chromatography on EM Science Geduran silica gel 60 (35-75 µm). Thin layer chromatography was performed on EM Science silica gel 60 F254 plates (250 µm). Visualization of the developed chromatogram was accomplished by UV lamp. Nuclear magnetic resonance (NMR) spectra were acquired on Bruker DRX-500/400 operating at 125/100 MHz for ¹H NMR and ¹³C NMR, respectively; residual protio solvent signals were used as internal standards for calibration purposes. Data for ¹H NMR are reported as follows: chemical shift (ppm), multiplicity (s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet), integration, coupling constant (Hz). High-Resolution Mass was performed by Mass Spectrometry Facility at University of Norte Dame. All fluorescence measurements were recorded on a Varian Cary Eclipse Fluorescence Spectrophotometer and Hitachi F4600 Fluorescence Spectrophotometer.



found 483.2054) prior to the addition of sulfide.



This material is based upon work supported by the National Science Foundation under CHE-0741793

Figure S1b. HRMS identification for the sulfide addition product of SFP-3 probe **8a** (yield 40%) (calculated for $C_{29}H_{27}BF_2N_2NaO_3S$ (M+Na)⁺ 539.1752; found 539.1748).



Figure S2. SFP-3 probe (10 μ M) incubated with 100 μ M Na₂S in PBS buffer (20 mM, pH 7.0, 40% CH₃CN) at 25 °C after 0-5-10-15-20-25-30-min. The data represent the average of three independent experiments. Excitation: 500nm, slit, 5nm; emission: 505-600 nm, slit, 5nm, 450V.



Figure S3. SFP-3 probe (10 μ M) incubated with different concentrations of Na₂S (10, 20, 30, 40, 50, 60, 80 and 100 μ M) at 25 °C for 20 min (20 mM PBS, PH 7.0, 40% CH₃CN), excitation: 500 nm, slit, 5 nm; emission: 505-600 nm, slit, 5 nm, 450V.



Figure S4. SFP-3 probe (10 μ M) incubated with Na₂S and various thiols or amino acid in 20 mM PBS (PH 7.0, 40% CH₃CN) at 25 °C for 20 min. Excitation:500 nm, slit, 5 nm; emission: 505-600 nm, slit, 5 nm, 450V.



Figure S5. Fluorescence spectra of 10 μ M SFP-3 measured in the presence of 1 μ l H₂S buffers that H₂S gas had been bubbled through for varying lengths of time. SFP-3 probe was allowed to incubate in the buffers for 30 min at 25 °C. Measurements performed in 20 mM PBS buffer (pH 7.0, 40% CH₃CN). Excitation: 500 nm, emission: 505–600 nm. All H₂S buffers were prepared by adding 10 ml DI H₂O to a 25 ml Schlenk tube and subsequently bubbling nitrogen through for 30 min. H₂S was then bubbled through for various lengths of time using a 0.8 mm needle. Ventilation rate was maintained at 1 bubble per second.



Figure S6. Fluorescence spectra of the SFP-3 probe (100 μ M) incubated with 100 μ M Na₂S after 30 s, 60 s, 90 s, 120 s, 180 s, 240 s, 300 s in bovine serum (40% CH₃CN) at 25 °C. Excitation: 500 nm, emission: 505–600 nm. The data represent the average of four independent experiments.



Figure S7. Fluorescence spectra of the SFP-3 probe (100 μ M) incubated with 0, 20, 40, 60, 80, 100, 150, 200 μ M Na₂S after 3 min in bovine serum (40% CH₃CN) at 25 °C. Excitation: 500 nm, emission: 505–600 nm. The data represent the average of four independent experiments.



Figure S8. Determination of sulfide concentration in Spiked mouse blood plasma using Na₂S as internal standard (final concentrations 0, X, X+5, X+15 μ M). Four eppendorf tubes containing 100 μ L blood were centrifuged at 6000 rpm for 2 min. 1 μ L ZnCl₂ (76 mM, final concentration 1 mM. as 0 point), DI H₂O (as X point), Na₂S (375 μ M, final concentration 5 μ M, as X+5 μ M point), Na₂S (1.125 mM, final concentration 15 μ M, as X+15 μ M point) were spiked into 45 μ L blood plasma respectively, then 30 μ L CH₃CN was added respectively to precipitate the protein in plasma¹. The samples were centrifuged at 6000 rpm for 1min, 70 μ L supernatant liquid was added into 140 μ L CH₃CN/H₂O buffer (40% CH₃CN) to precipitate proteins again, then centrifuged at 6000 rpm for 5 min, get supernatant liquid samples which will be used to detect in PBS buffer. 50 μ L sample was added into 30 μ L 40 mM PBS buffer (final concentration 12 mM, pH 7.0), after that, 20 μ L 100 μ M SFP-3 probe (final concentration 20 μ M, 40% CH₃CN) was added. Emission spectra (λ ex = 500 nm, emission at 517 nm) were collected after the mixture were incubated at 37 °C for 30 min. The data represent the average of three independent experiments.



Figure S9. Determination of sulfide concentration in the fresh mouse blood plasma with SFP-3 probe (see the following H₂S detection protocols). Sulfide concentrations in mouse blood plasma 1, 2, 3, 4 were 55.41 μ M, 57.93 μ M, 52.64 μ M, 57.98 μ M, respectively. The average sulfide concentration is 55.99 ± 2.54 μ M.



Figure S10. Determination of sulfide concentrations in mice brain tissues with SFP-3 probe (see the following H₂S detection protocols). Sulfide concentrations in mice brain homogenates (2 %, w/v) were 10.70 μ M, 8.49 μ M, 8.67 μ M, 5.33 μ M, respectively. Total protein concentrations in mice brain homogenates (1 %, w/v) were 0.80 g/L, 0.50 g/L, 0.56 g/L, 0.50 g/L, respectively. The H₂S concentrations in mice brain tissues were expressed as μ mol/g protein. The H₂S concentrations in mice brain mice brain were 6.69 μ mol/g protein, 8.49 μ mol/g protein, 7.74 μ mol/g protein, 5.33 μ mol/g protein, 7.74 μ mol/g protein. Excitation: 400 nm, slit, 5 nm; emission: 505-600 nm, slit, 5 nm, 500V.



Figure S11. SFP-3 probe (10 μ M) incubated with 50 μ M Na₂S and GSH (5, 10 mM) in 20 mM PBS (PH 7.4, 40% CH₃CN) at 25 °C for 60 min. Excitation:496 nm, slit, 5 nm; emission: 500-600 nm, slit, 5 nm, 300V, at 240 nm/min. Fluorescence measurements were carried out at room temperature on a Hitachi Fluorescence Spectrophotometer F-7000. The data represent the average of three independent experiments.

Supplementary Methods

Synthetic protocols of compounds 1-3



((3-bromo-4-(1,3-dioxan-2-yl)benzyl)oxy)(tert-butyl)dimethylsilane (2)

A solution of TBSCl (331 mg, 2.21 mmol) in dry DMF (2 mL) was added by using a syringe to a solution of compound **1** (500 mg, 1.84 mmol) and imidazole (461 mg, 3.68 mmol) in dry DMF (2 mL) in a 25 mL Schlenk tube. The reaction mixture was stirred at room temperature for 6 h. The reaction was quenched with deionized water (100 mL) and extracted with EtOAc (3 x 100 mL). The combined organic layers were washed with water and brine and then dried with Na₂SO₄. The solvent was evaporated and the crude product was purified by column chromatography on SiO₂ to give the purified product, colorless oil. Yield 82 %. TLC (silica, hexane:EtOAc, 4:1 v/v): $R_f = 0.6$; ¹H NMR (500 MHz, CDCl₃): δ 7.63 (d, *J* = 8.0 Hz, 1 H), 7.50 (s, 1 H), 7.26 (d, *J* = 8.0 Hz, 1 H), 5.74 (s, 1 H), 4.96 (s, 2 H), 4.25 (dd, *J* = 5.0, 11.0 Hz, 2 H), 3.99–4.04 (m, 2 H), 2.01–2.25 (m, 1 H), 1.3 (dd, *J* = 1.0, 13.5 Hz, 1 H), 0.92 (s, 9 H), 0.08 (s, 6 H) ; ¹³C NMR (125 MHz, CDCl₃): δ 144.2, 136.1, 130.1, 127.9, 125.1, 122.3, 101.0, 67.7, 64.1, 26.0, 25.8, 18.5, -5.2; HRMS (ESI⁺): (M+H)⁺ calcd. for C₁₇H₂₈BrO₃Si, 387.0991; found, 387.0968 .

5-(((tert-butyldimethylsilyl)oxy)methyl)-2-(1,3-dioxan-2-yl)benzaldehyde (3)

To a solution of compound **2** (500 mg, 1.30 mmol) in THF (4 mL) cooled to -78°C under nitrogen was added *n*-BuLi (240 µL, 2.55 mmol) in hexane dropwise with the temperature kept below -70°C. After 1 h, dry DMF (812 µL, 5.2 mmol) was added dropwise. The resulting solution was allowed to gradually warm to room temperature and stirring was continued for 2h. The reaction was quenched with saturated aqueous NH₄Cl (5 mL) and extracted with EtOAc (3 x 100 mL). The combined organic layers were washed with water and brine and then dried with Na₂SO₄. The solvent was evaporated and the crude product was purified by column chromatography on SiO₂ to give the purified product, white solid. Yield 67 %. TLC (silica, hexane:EtOAc, 4:1 v/v): $R_f = 0.4$; ¹H NMR (500 MHz, CDCl₃): δ 10.54 (s, 1 H), 7.88 (s, 1 H), 7.60–7.68 (m, 2 H), 6.02 (s, 1 H), 4.80 (s, 2 H), 4.27–4.29 (m, 2 H), 4.02–4.07 (m, 2 H), 2.25–2.28 (m, 1 H), 1.48–1.50 (m, 1 H), 0.95 (s, 9 H),

0.11 (s, 6 H); ¹³C NMR (125 MHz, CDCl₃): δ 192.4, 142.9, 138.4, 133.8, 131.0, 127.4, 127.2, 100.2, 67.7, 64.2, 26.0, 25.7, 18.4, -5.2; HRMS (ESI⁺): (M+H)⁺ calcd. for C₁₈H₂₈O₄Si, 337.1835; found, 337.1819.



(E)-3-(5-(((tert-butyldimethylsilyl)oxy)methyl)-2-(1,3-dioxan-2-yl)phenyl)-1-phenylprop-2-en-1-one (4)

5 N NaOH (2.1 mL) was added dropwise into a stirred solution of aldehyde **3** (359 mg, 1.07 mmol) and acetophenone (125 μ L, 1.07 mmol) in 5 mL MeOH. The reaction mixture was continuously stirred at room temperature for 2 h. After, the mixture was filtered to collect the solid, which was then washed with water to obtain the crude product. The crude product was then purified by column chromatography on SiO₂ to give the purified product as a yellow solid. Yield 74 %. TLC (silica, hexane:EtOAc, 4:1 v/v): R_f = 0.45; ¹H NMR (400 MHz, CDCl₃): δ 8.36 (d, *J* = 15.6 Hz, 1 H), 8.05 (d, *J* = 7.2 Hz, 2 H), 7.76 (s, 1 H), 7.58–7.64 (m, 2 H), 7.51–7.54 (m, 2 H), 7.45 (d, *J* = 16.0 Hz, 1 H), 7.38 (d, *J* = 8.0Hz, 1 H), 5.75 (s, 1 H), 4.80 (s, 2 H), 4.27–4.30 (m, 2 H), 3.99–4.05 (m, 2 H), 2.22–2.26 (m, 1 H), 1.46 (d, *J* = 13.2Hz, 1 H), 0.99 (s, 9 H), 0.13 (s, 6 H); ¹³C NMR (125 MHz, CDCl₃): δ 190.8, 142.6, 142.4, 138.2, 136.4, 133.1, 132.6, 128.6, 128.5, 127.5, 127.2, 124.0, 123.8, 100.3, 67.5, 64.3, 25.9, 25.6, 18.3, -5.3; HRMS (ESI⁺): (M+H)⁺ calcd. for C₂₆H₃₅O₄Si, 439.2304; found, 439.2301.

(E)-3-(2-(1,3-dioxan-2-yl)-5-(hydroxymethyl)phenyl)-1-phenylprop-2-en-1-one (5)

A solution of **4** (394 mg, 0.9 mmol) and TBAF (1.0M in THF, 856 μ L, 0.9 mmol) in THF (8 mL) was stirred at RT for 30 min, then it was quenched with aq. sat. NaHCO₃ (6 mL) and extracted with EtOAc (3 x 100 mL), dried over Na₂SO₄, filtered, and evaporated. The crude product was purified by column chromatography on SiO₂ to give the purified product, a yellow solid. Yield 82 %. TLC (silica, hexane:EtOAc, 1:2 v/v): R_f = 0.6; ¹H NMR (500 MHz, CDCl₃): δ 8.28 (d, *J* = 15.5 Hz, 1 H), 7.96–8.02 (m, 2 H), 7.71 (s, 1 H), 7.55–7.60 (m, 2 H), 7.47–7.50 (m, 2 H), 7.35–7.43 (m, 2 H), 5.71 (s, 1 H), 4.70 (s, 2 H), 4.24–4.27 (m, 2 H), 3.97–4.02 (m, 2 H), 2.20–2.25 (m, 1 H), 1.44(d, *J* = 13.5 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃): δ 191.2, 142.5, 142.1, 138.2, 136.9, 133.4, 132.9, 128.8, 128.7, 128.4, 127.4, 124.9, 124.0, 100.3, 67.6, 64.6, 25.7; HRMS (ESI⁺): (M+H)⁺ calcd. for C₂₀H₂₁O₄, 325.1440; found, 325.1431.



(E)-4-(1,3-dioxan-2-yl)-3-(3-oxo-3-phenylprop-1-en-1-yl)benzaldehyde (6)

A mixture of compound **5** (330 mg, 1.02 mmol), PCC (332 mg, 1.53 mmol) and Celite (300 mg) in CH₂Cl₂ (10 ml) was stirred at room temperature for 1 h. The reaction mixture was filtered through Celite and a silica gel pad and then evaporated to obtain the crude product. This was purified by column chromatography on SiO₂ to give a white solid. Yield 93 %. TLC (silica, hexane:EtOAc, 1:2 v/v): $R_f = 0.8$; ¹H NMR (500 MHz, CDCl₃): δ 10.04 (s, 1 H), 8.25 (d, J = 15.5 Hz, 1 H), 8.20 (s, 1 H), 8.03 (d, J = 7.0 Hz, 2 H), 7.87–7.95 (m, 2 H), 7.56–7.58 (m, 1 H), 7.43–7.53 (m, 3 H), 5.75 (s, 1 H), 4.26–4.28 (m, 2 H), 3.98–4.03 (m, 2 H), 2.21–2.24 (m, 1 H), 1.47 (d, J = 13.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃): δ 191.6, 190.3, 143.0, 140.8, 137.8, 136.7, 134.5, 133.1, 130.9, 128.7, 128.6, 128.0, 127.6, 125.2, 99.4, 67.6, 25.6; HRMS (ESI⁺): (M+H)⁺ calcd. for C₂₀H₁₉O₄, 323.1283; found, 323.1281.

(E)-10-(4-(1,3-dioxan-2-yl)-3-(3-oxo-3-phenylprop-1-en-1-yl)phenyl)-5,5-difluoro-1,3,7,9-tetramethyl-5H-dipyrrolo[1,2-c:2',1'-f][1,3,2]diazaborinin-4-ium-5-uide (7)

Compound **6** (85mg, 0.264 mmol) and 2,4-dimethylpyrrole (54.4 μ L, 0.53 mmol) were dissolved in 100 mL of dry CH₂Cl₂ under N₂. One drop of TFA was added, after the solution was stirred at room temperature for 12 h, DDQ (66 mg, 0.29 mmol) was added, and stirring was continued for 1.5 h. Then 184 μ L of Et₃N and 163 μ L of BF₃·Et₂O were added, and the solution was stirred at room temperature for 2 h. The reaction mixture was washed with water three times and brine once, dried over Na₂SO₄, filtered, and evaporated. The crude product was purified by column chromatography on SiO₂ to give the purified product, a red foam. Yield 22 %. TLC (silica, CH₂Cl₂): R_f = 0.3; ¹H NMR (500 MHz, CDCl₃): δ 8.33 (d, *J* = 16.0 Hz, 1 H), 8.02 (d, *J* = 7.5 Hz, 2 H), 7.82 (d, *J* = 7.5 Hz, 1 H), 7.73 (s, 1 H), 7.57–7.60 (m, 1 H), 7.48–7.52 (m, 1 H), 7.46 (d, *J* = 15.5 Hz, 1 H), 7.38 (dd, *J* = 1.5, 8.0 Hz, 1 H), 6.00 (s, 2 H), 5.83 (s, 1 H), 4.33 (dd, *J* = 4.5, 11.0 Hz, 2 H), 4.05–4.10 (m, 2 H), 2.56 (s, 6 H), 2.26–2.34 (m, 1 H), 1.51 (d, *J* = 13.5 Hz, 1 H), 1.42 (s, 6 H); ¹³C NMR (125 MHz, CDCl₃): δ 190.2, 156.0, 143.1, 141.0, 140.4, 138.7, 137.9, 136.0, 134.5, 133.1, 131.4, 129.5, 128.8, 128.7, 128.2, 126.3, 124.4, 121.6, 99.9, 67.8, 25.7, 15.0, 14.7; HRMS (ESI⁺): (M+H)⁺ calcd. for C₃₂H₃₂ BF₂N₂O₃, 541.2474; found, 541.2506.

(E)-5,5-difluoro-10-(4-formyl-3-(3-oxo-3-phenylprop-1-en-1-yl)phenyl)-1,3,7,9-tetramethyl-5Hdipyrrolo[1,2-c:2',1'-f][1,3,2]diazaborinin-4-ium-5-uide (8)

10 N HCl (708 μ L) was added to a solution of compound **7** (85 mg, 0.16 mmol) in acetone (4 mL) and stirred at room temperature for 2 h. The reaction was quenched with water (100 mL) and extracted with EtOAc (3 x 100 mL). The combined organic layers were washed with water, a saturated NaHCO₃ solution, and brine, and then dried with Na₂SO₄. The crude product was purified by column chromatography on SiO₂ to give the purified product, a red foam. Yield 81 %. TLC (silica, CH₂Cl₂): $R_f = 0.7$; ¹H NMR (500 MHz, CDCl₃): δ 10.44 (s, 1 H), 8.62 (d, *J* = 16.0 Hz, 1 H), 8.05 (d, *J* = 8.0 Hz, 1 H), 8.01 (d, *J* = 7.5 Hz, 2 H), 7.78 (s, 1 H), 7.54–7.60 (m, 2 H), 7.48–7.51 (m, 2 H), 7.45 (d, *J* = 16.0 Hz, 1 H), 6.02 (s, 2 H), 2.56 (s, 6 H), 1.41 (s, 6 H); ¹³C NMR (125 MHz, CDCl₃): δ 190.9, 189.8, 156.7, 142.7, 141.1, 139.5, 138.9, 138.2, 137.5, 134.5, 133.4, 132.8, 130.9, 130.0, 128.9, 128.8, 127.9, 127.6, 121.9, 15.0, 14.8; HRMS (ESI⁺): (M+H)⁺ calcd. for C₂₉H₂₆ BF₂N₂O₂, 483.2055; found, 483.2054.

Quantum Yields

Quantum yields were determined using fluorescein as a standard according to a published method². For SFP-3 and fluorescein, the absorbance spectra were measured within an absorbance range of 0.01 to 0.1. The quantum yield was calculated according to the equation: $\Phi_{sample} = \Phi_{standard}$ (Grad_{sample}/Grad_{standard})($\eta^2_{sample}/\eta^2_{standard}$); where Φ is the quantum yield, $\Phi_{fluorescein} = 0.79$ in 0.1 M NaOH, Grad is the slope of the plot of absorbance versus integrated emission intensity, and η is the refractive index of the solvent.

SFP-3, $\varepsilon = 1.12^* 10^5 \text{ M}^{-1} \text{ cm}^{-1}$, $\Phi = 0.10$.

Protocols for H₂S Measurement

Fluorometric analysis.

All fluorescence measurements were carried out at room temperature on a Varian Cary Eclipse Fluorescence Spectrophotometer. The samples were excited at 500 nm with the excitation and emission slit widths set at 5 nm. The emission spectrum was scanned from 505 nm to 600 nm at 120 nm/min. The photomultiplier voltage was set at 450 V. The probe was dissolved in CH_3CN to make a 1.0 mM stock solution, which was diluted to the required concentration for measurement.

SFP-3 stock solution preparation

SFP-3 (1.3 mg, 0.002697 mmol) was dissolved into CH3CN (1.08 mL) to get 2.5 mM stock solution, which was then diluted to 1.0 mM stock solution for general use.

Na₂S stock solution (20 mM) preparation³

5 mg EDTA was dissolved in 10 mL DI H₂O in a 25 mL Schlenk tube. The solution was purged vigorously with nitrogen for 15 min. Then 48 mg sodium sulfide (Na₂S·9H₂O) was dissolved in the solution under nitrogen. The resulting solution was 20 mM Na₂S, which was then diluted to 1.0 - 2.0 mM stock solution for general use.

SFP-3 Fluorescence responses to H₂S

Fig 2a. 1 µL of SFP-3 stock solution (1.0 mM) (for a final concentration: 10 µM) was dissolved into a mixed solution of 50 µL PBS buffer (pH 7.0, 40 mM) and 39 µL CH₃CN in a 1.5 mL eppendorf tube. 10 µL Na₂S stock solution (1.0 mM) was added (for a final concentration of 100 µM). The mixture was then incubated at room temperature. Emission spectra ($\lambda_{ex} = 500$ nm) were collected at 0, 5, 10, 15, 20, 25, and 30 min.

Fig 2b. 1 μ L of SFP-3 stock solution (1.0 mM) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM), 39 μ L CH₃CN and DI H₂O (9, 8, 7, 6, 5, 4, 2, and 0 μ L respectively) in a 1.5 mL eppendorf tube. Na₂S stock solution (1, 2, 3, 4, 5, 6, 8, and 10 μ L, 1.0 mM, respectively) was added (for a final concentration of 10, 20, 30, 40, 50, 60, 80, and 100 μ M, respectively). Emission spectra ($\lambda_{ex} = 500$ nm) were collected after the mixture was incubated at room temperature for 20 min.

SFP-3 selectivity tests to other thiols and amino acids

Fig 2c. Selectivities for SFP-3 probe (10 μ M) were measured by fluorescence responses at 20 min in PBS buffer (20 mM, pH 7.0, 40% CH₃CN) at 25 °C. Excitation: 500 nm, emission: 505–600 nm. The data represent the average of three independent experiments.

Na₂S (0 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. Then 10 μ L DI H₂O was added.

Na₂S (50 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 5 μ L DI H₂O and 5 μ L Na₂S stock solution (1.0 mM) was then added.

Cystine (100 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 10 μ L cystine (1.0 mM) was then added.

Lys (100 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 10 μ L Lys (1.0 mM) was then added.

Ala (100 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 10 μ L Ala (1.0 mM) was then added.

Cys (100 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 10 μ L Cys (1.0 mM) was then added.

GSH (100 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 10 μ L GSH (1.0 mM) was then added.

GSH (100 μ M) + **Na₂S** (50 μ M): 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM) and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. 4 μ L H₂O, 1 μ L GSH (10 mM), and 5 μ L Na₂S stock solution (1.0 mM) were then added.

SFP-3 Fluorescence responses to H₂S gas buffer

Fig 2d. 10 μ L of SFP-3 stock solution (100 μ M) (for a final concentration: 10 μ M) was dissolved into a mixed solution of 50 μ L PBS buffer (pH 7.0, 40 mM), 9 μ L DI H₂O, and 30 μ L CH₃CN in a 1.5 mL eppendorf tube. Then the mixture was incubated with 1 μ l H₂S buffer (bubbling H₂S 10 min–saturated solution) at 25 °C. Emission spectra (λ ex = 500 nm) were collected from 5 s–300 s.

Fluorescence response of SFP-3 probe in Bovine plasma

Figure 3a: Fluorescence spectra of the SFP-3 probe (100 μ M) incubated with 100 μ M Na₂S in bovine serum (40% CH₃CN) at 25 °C. Excitation: 500 nm, emission: 505–600 nm. The data represent the average of four independent experiments.

20 μ L Na₂S stock solution (1.0 mM) (for a final concentration: 100 μ M) was spiked into a 1.5 mL eppendorf tube containing 100 μ L Bovine plasma. Then, 80 μ L of SFP-3 probe stock solution (250 μ M) (for a final concentration 100 μ M) was added, the mixture were incubated at 25 °C. The emission spectra were collected at 30, 60, 90, 120, 180, 240 and 300 s, respectively.

Figure 3b: 1.0 mM Na₂S stock solution (0, 4, 8, 12, 16, and 20 μ L, respectively) were spiked into a 1.5 mL eppendorf tube containing 100 μ L Bovine plasma and DI H₂O (20, 16, 12, 8, 4, and 0 μ L, respectively) (for Na₂S final concentration 0, 20, 40, 60, 80, 100 μ M, respectively). Then, 80 μ L of SFP-3 probe stock solution (250 μ M) (for a final concentration 100 μ M) was added, Emission spectra (λ ex = 500 nm) were collected after the mixture were incubated at 25 °C for 3min.

2.0 mM Na₂S stock solution (15 and 20 μ L, respectively) were spiked into a 1.5 mL eppendorf tube containing 100 μ L Bovine plasma and DI H₂O (5 and 0 μ L, respectively) (for a final concentration 150 and 200 μ M, respectively). Then, 80 μ L of SFP-3 probe stock solution (250 μ M) (for a final concentration 100 μ M) was added, Emission spectra (λ ex = 500 nm) were collected after the mixture were incubated at 25 °C for 3min.

Detection of H₂S in mouse blood plasma

Table 1 and Figure S9: The animal protocol is approved by the Institutional Animal Care and Use Committee at the University of Chicago. Eight-week old male C57BL6/J mice were purchased from The Jackson Laboratory (Bar Harbor, ME). Mice were anesthetized with ketamine/xylazine and blood was drawn with a heparinized (~50 USP/ml) syringe from the inferior vena cava of male C57BL6/J mice. Mouse blood (500 µL) was centrifuged in a eppendorf tube at 6000 rpm for 5min. 10 µL blood plasma was added into 50 µL bovine plasma, then 0, 1, 3, 5 µL Na₂S stock solution (100 µM) were spiked into the sample as internal standard (a final H₂S concentration X, X+1, X+3, X+5 µM, respectively), after that, 40 µL 50 µM SFP-3 probe (final concentration 20 µM. 40% CH₃CN) was added. Emission spectra ($\lambda ex = 500$ nm, emission at 516 nm) were collected after the mixture were incubated at 37 °C for 3 min. Zero point was obtained by adding 1 µL, 100 mM (final concentration 1 mM) ZnCl₂ to trap H₂S in the sample. The data represent the average of three independent experiments.

Detection of H₂S in mouse brain tissues

Table 1 and Figure S10: All animal procedures are approved by the Institutional Animal Care and Use Committee at Nanjing University. Eight-week old male C57BL/6 mice were purchased from Yangzhou University. For the measurement of H₂S, mice were sacrificed and brains were immediately removed, homogenized with a 9 volumes (w/v) of ice-cold 100 mM PBS buffer (pH 7.4) and centrifuged at 10,000 g for 10 min at 4 °C. All the above operations were performed in ice bath, and the homogenate supernatants were immediately used for the following determination. All fluorescence measurements were recorded on a Hitachi F4600 Fluorescence Spectrophotometer. Total protein concentrations in mice brain were determined using the Coomassie Blue method.

Determination of sulfide concentration in spiked brain homogenates using Na₂S as internal standard (X, X+5, X+10, X+15, X+20, X+30 μ M). 20 μ L of 10% homogenates supernatant (final concentration 2%, w/v) was added into Eppendorf tubes with 30 μ L PBS buffer (100 mM, pH 7.4), 30 μ L CH₃CN and DI H₂O (10, 9.5, 9, 8.5, 8, and 7 μ L respectively). Then 0, 0.5, 1, 1.5, 2, 3 μ L Na₂S stock solution (1.0 mM) were spiked into the sample as internal standard, after that, 10 μ L 1.0 mM SFP-3 probe (final concentration 100 μ M) was added. Emission spectra ($\lambda_{ex} = 400$ nm, emission at 520 nm) were collected after the mixture were incubated at 37 °C for 20 min. Zero point was obtained by adding 1 μ L, 100 mM (final concentration 1 mM) ZnCl₂ to trap H₂S in the sample. The H₂S concentration of each sample was calculated by a calibration curve of Na₂S and results were expressed as μ mol/g protein. The data represent the average of three independent experiments.

References

- 1. S. Sreejith, K. P. Divya, A. Ajayaghosh. Angew. Chem. Int. Ed. 2008, 120, 8001-8005.
- 2. A. T. R. Williams, S. A. Winfield, J. N. Miller. Analyst. 1983, 108, 1067-1071.
- 3. Y. Zhao, H. Wang, M. Xian. J. Am. Chem. Soc., 2011, 133, 15–17.

Electronic Supplementary Material (ESI) for Chemical Science This journal is O The Royal Society of Chemistry 2012









Electronic Supplementary Material (ESI) for Chemical Science This journal is O The Royal Society of Chemistry 2012





Electronic Supplementary Material (ESI) for Chemical Science This journal is \circledcirc The Royal Society of Chemistry 2012

0 OTBS





Electronic Supplementary Material (ESI) for Chemical Science This journal is \circledcirc The Royal Society of Chemistry 2012

0





Electronic Supplementary Material (ESI) for Chemical Science This journal is O The Royal Society of Chemistry 2012













Electronic Supplementary Material (ESI) for Chemical Science This journal is \circledcirc The Royal Society of Chemistry 2012

СНО 0 В



