SUPPORTING INFORMATION

Fe-Catalysed Kumada-type Alkyl-Alkyl Cross-Coupling. Evidence for the intermediacy of Fe(I) complexes

Manuel Guisán-Ceinos, Francisco Tato, Elena Buñuel, Paloma Calle, Diego J. Cárdenas

Departamento de Química Orgánica, Universidad Autónoma de Madrid, Cantoblanco, 28049-Madrid, Spain

Table of Contents

1. Materials and Methods	S2
2. Optimization of Reaction Parameters	S2
3. Determination of Iron Oxidation State by GC Monitoring	S4
4. Compound Characterization	S6
5. References	S14
6. Copy of ¹ H RMN and ¹³ C NMR	S16

1. Materials and Methods

All reactions were carried out under argon in anhydrous THF dried using a Solvent Purification System (SPS) and/or distillation over Na/benzophenone. Thin layer chromatography was carried out using pre-coated TLC-aluminum sheets with 0.20 mm of silica gel and fluorescent indicator UV₂₅₄ (ALUGRAM[®] Xtra SIL G/UV₂₅₄) and visualized with UV light and/or phosphomolybdic acid ethanol solution and potassium permanganate aqueous solution. Chromatography purifications were carried out using flash grade silica gel (Chromatogel 60 ACC, 40-60 μm). NMR spectra were recorded at 23 °C on a Bruker Avance 300 MHz. The ¹H and the ¹³C {1H} NMR chemical shifts are given in parts per million (ppm) relative to the residual signals of the deuterated solvents.¹ Carbon types were determined from DEPT - ¹³C NMR. The following abbreviations are used to indicate the multiplicity: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; bs, broad signal. Mass spectra were recorded on a Waters VG AutoSpec and gas chromatography was recorded on a Varian 3800 GC. EPR spectra were recorded on a Varian E-12 ((modulation frequency 100 kHz; modulation amplitude 5 G; power 6.32 kW).

Starting iodides such as (3-iodopropyl)benzene **1**,² 1-(3-iodopropyl)-4-methoxybenzene,³ methyl 5iodopentanoate,⁴ ethyl 4-iodocyclohexanecarboxylate,⁵ 2-(iodomethyl)tetrahydro-*2H*-pyran,⁶ 4iodo-2,2-dimethyltetrahydro-*2H*-pyran,⁷ 2-(3-iodopropoxy)tetrahydro-2H-pyran,⁸ tert-butyl 4iodopiperidine-1-carboxylate,⁹ 1-benzyl-4-iodopiperidine,¹⁰ 2-(2-iodoethyl)-*N*-tosylindol,¹¹ and cholesteryl iodide¹² have been synthesized following the procedures reported in the literature and using starting materials commercially available. Other chemicals were used as received without further purifications. Iron(II) acetate was purchased from Aldrich and Strem Chemicals with 99.99% and 97% purity respectively and no change was observed concerning the reactivity. However the catalytic effectiveness diminished over time. All other commercially available compounds were used as received.

2. Optimization of Reaction Parameters

Electrophiles:

Various alkyl halides and tosylate were tested in the alkyl-alkyl cross-coupling. As we can see in Table S1, the more reactive alkyl iodide (Table S1, entry 1), gave the higher yield (55%). With the bromide derivative, the yield dropped to 39% (Table S1, entry 2) and with the alkyl chloride (Table S1, entry 3), the reaction did not take place. On the other hand, when the alkyl tosylate was used as starting electrophile the reaction afforded a 20% yield (Table S1, entry 4).

x	Ph	BrM +		Fe(OAc) ₂ (5 mol%) IMes·HCl (12 mol%)	Ph 🔨		
		_		THF, RT		2a	0
		_	Entry ^[a]	Alkyl-X	Yield ^[b]		
			1	X = I	55%		
			2	X = Br	39%		
			3	X = Cl	NR		
			4	X = OTs	20%		

Table S1. Alkyl electrophile investigations of the alkyl-alkyl cross-coupling reaction.

[a] Reaction was carried out by the slow addition of Grignard reagent (2 equiv) to a THF solution of alkyl halide or tosylate (1 equiv), Fe(OAc)₂ (5 mol%) and IMes HCl (12 mol%) under Argon at RT. [b] Yield of isolated product.

Iron Source:

Both iron(II) and iron(III) fluorides, chlorides, and bromides, iron(II) iodide and iron(III) acetylacetonate were tested in the reaction and surprisingly both iron(II) and iron(III) chlorides, bromides, and Fe(acac)₃ gave a similar yields (see Table S2, entries 3-6, 8) which were indicating the no influence of the iron counterion and the plausible formation of a common intermediate responsible of the cross-coupling reaction. Iron(II) and iron(III) fluorides and iron(II) iodide (see Table S2, entries 1-2, 7) possess some handling difficulties with regard to solubility and hygroscopy respectively which could explain the different behavior of those with respect to the aforementioned.

Table S2. Iron source investigations of the alkyl-alkyl cross-coupling reaction.



3	$\operatorname{FeCl}_{2}(5)$	IMes·HCl (12)	70%
4	$\operatorname{FeCl}_{3}(5)$	IMes·HCl (12)	73%
5	$\operatorname{FeBr}_{2}(5)$	IMes·HCl (12)	72%
6	$\operatorname{FeBr}_{3}(5)$	IMes·HCl (12)	69%
7	$\operatorname{FeI}_{2}(5)$	IMes·HCl (12)	32%
8	$Fe(acac)_3(5)$	IMes·HCl (12)	70%

[a] Reaction was carried out following the general procedure. [b] Yield of isolated product.

Flow Rate:

A study on the flow addition rate was carried out using iodocyclohexane as starting material. As we and other authors noted, the presence of a large excess of Grignard reagent in the reaction media provoked a reducing or even the inhibition of the cross-coupling. That means a control on the flow rate is required for the success of the reaction. As we can see in the Table S4, lower yields were obtained when faster additions rates were used.

Table S3. Flow rates investigations of the alkyl-alkyl cross-coupling reaction.



[a] Reaction was carried out following the general procedure. [b] Yield of isolated product.

3. Determination of Iron Oxidation State by GC Monitoring

Compounds 2-ethyl-1,3-dioxane I,¹³ 2-vinyl-1,3-dioxane II,¹⁴ and 1,4-di(1,3-dioxanyl)butane III were prepared according to the modified procedures described in the literature. These compounds were synthesized in order to obtain the calibration curves in which dodecane was used as internal standard. The acquired data were collected from the experiment described as follows: Iron(II) acetate (17.4 mg, 0.1 mmol) and 1,3-dimesityl-1H-imidazol-3-ium chloride (82 mg, 0.24 mmol) were placed in Schlenk flask and dried under vacuum. Then, dry THF (20 mL) was added and the mixture was heated at 50-60 °C under Ar. A 0.5 M solution of alkylmagnesium bromide (in THF) was added and reactions were stirred for 20 min. Internal standard (dodecane) was added and reactions were hydrolyzed by passing the mixtures through a short pad of silica-gel. Data were collected for the addition of 0.1, 0.2, 0.3, 0.4, 0.6, 0.8, 1.0, 1.2, 1.4, and 1.6 mmol of Grignard reagent (Table S4). 2-Vinyl-1,3-dioxane II was not detected in any experiment and 2-ethyl-1,3dioxane I showed as a highly volatile compound and values of its concentration did not correspond with those used in the iron activation reaction. Line of purple points represents the uncorrected data collected directly from the experiment. However in order to obtain a reliable value of the homocoupling III, the amount of this homocoupling product III already present in the Grignard reagent must be subtracted from the original line giving the line of blue points (Figure S1). A more representative graphic could be obtained by representation of the electrons that are intervening in the reaction per mmol of iron and the equivalent of Grignard reagent (Figure S2).

Table S4. Collected data recorded from GC monitoring in order to determine the iron oxidation state.

6

F&(OA&) ₂	≠ IMes∶HCI -	THF, 60 °C			● ≠ [₣€ ^(ñ)]
nard reagent	internal standa	ard homocoupled	homocoupled	homocoupled	e⁻ / Fe
(equiv.)	(area)	(area)	(mmol)	(corrected mmol)	

0

(equiv.)	(area)	(area)	(mmol)	(corrected mmol)	e ⁻ / Fe
1	723028	5538	0,006	0,0034	0,07
2	689675	7917	0,008	0,0040	0,08
3	715996	9959	0,010	0,0035	0,07
4	697583	19883	0,021	0,0120	0,24
6	661095	40442	0,045	0,0314	0,63
8	641311	47069	0,053	0,0359	0,72
10	83633	7320	0,064	0,0418	0,84
12	565112	49397	0,064	0,0373	0,75
14	515489	46597	0,066	0,0351	0,70
16	456949	46401	0,074	0,0388	0,78



Figure S1. Formation of Grignard homocoupled product **III** (mmol) with respect to the added equivalent of Grignard reagent.

Figure S2. Electrons that are intervening in the reaction per mmol of iron and the equivalent of Grignard reagent



4. Compound Characterization

trans- and cis-Ethyl 4-iodocyclohexanecarboxylate: Products were synthesized following the described procedure in the literature⁵ however spectroscopy data did not match EtO₂C with those reported in the reference 5. 1H-imidazole (1.0 g, 14.91 mmol), cis / trans triphenylphosphine (3.91 14.91 g, mmol). and ethyl 4hydroxycyclohexanecarboxylate (2.0 ml, 12.43 mmol) were combined in THF (10 ml) and cooled in an ice bath to 2-3 °C under argon. A freshly prepared solution of I₂ (3.78 g, 14.91 mmol) in THF (10 ml) was added dropwise maintaining the internal temperature below 10 °C. After the addition was complete, the reaction was allowed to warm to RT and stirred overnight. Then, 10 wt % aqueous solution of NaHSO₃ (5 mL) was added in one portion and Et₂O (5 mL) was added. The aqueous layer was extracted with Et_2O (3 × 5 mL) and the combined organic layer were washed with water and dried over MgSO₄. Solvent was evaporated under vacuum and products were purified by column chromatography in hexane/Et₂O 97:3 affording a 72:28 mixture of *trans/cis* isomers with a 50% combined yield and as colorless oil. The spectroscopy data for the *trans*isomer were obtained from a pure sample: ¹H-NMR (300 MHz, CDCl₃) δ : 4.63 (bs, 1H), 4.13 (q, J = 7.1 Hz, 2H), 2.45-2.32 (m, 1H), 2.18-1.88 (m, 4H), 1.84-1.64 (m, 4H), 1.24 (q, J = 7.1 Hz, 3H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ: 174.8 (C), 60.4 (CH₂), 41.6 (CH), 36.0 (2 × CH₂), 32.7 (CH), $26.3 (2 \times CH_2)$, 14.3 (CH₃) ppm; MS (CI) m/z 283 (M+H), 155, 41. The spectroscopy data for the cis-isomer were extracted from a 34:66 mixture of the cis/trans-isomers. ¹H-NMR (300 MHz, CDCl₃) δ: 4.63 (bs, 1H), 4.06 (q, J = 7.1 Hz, 2H), 2.45-2.32 (m, 1H), 2.18-1.63 (m, 7H), 1.60-1.41 (m, 1H), 1.17 (q, J = 7.1 Hz, 3H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 175.3 (C), 60.4 (CH₂), 41.6 (CH), 39.0 (2 × CH₂), 30.8 (2 × CH₂), 28.4 (CH), 14.2 (CH₃) ppm; MS (EI) *m/z* 283 (M+H), 155, 41.

General procedure for the sp³-sp³ cross coupling reaction catalyzed by Iron:

Iron(II) acetate (2.5 mol%) and 1,3-dimesityl-1H-imidazol-3-ium chloride (6 mol%) were placed in Schlenk flask and dried under vacuum. Then dry THF (2 mL) was added and the mixture was heated at 50-60 °C under Ar. A 0.5 M solution of alkylmagnesium bromide (30 mol%) was slowly added and the reaction mixture was stirred at 50-60 °C for 20 min. After cooling at RT, alkyl iodide (0.407 mmol) was added followed by the slow addition of 0.5 M solution of alkylmagnesium bromide (1.5 mmol) with syringe pump (flow rate = 0.16 mL/h). After the addition of the Grignard reagent, the reaction was stirred for 8 h and then, hydrolyzed with saturated aqueous NH4Cl solution (5 mL). The aqueous phase was extracted with DCM (3×5 mL) and the combined organic

phases were dried over MgSO4. The solvent was evaporated under vacuum and the product was purified by column chromatography in silica-gel.

2-(5-Phenylpentyl)-1,3-dioxane (2a): Product was obtained following the general procedure and using (3-iodopropyl)benzene as starting iodide. The compound 2a was purified by column chromatography (hexane/EtOAc 95:5) as colorless oil with a 70% yield. ¹H-NMR (300 MHz, CDCl₃) δ: 7.32-7.24 (m, 2H), 7.21-7.13 (m, 3H), 4.51 (t, *J* = 5.1 Hz, 1H), 4.10 (ddd, *J* = 10.7, 5.0 and 1.1 Hz, 2H), 3.76 (dt, *J* = 12.2 and 2.2 Hz, 2H), 2.61 (t, *J* = 7.3 Hz, 2H), 2.18-1.98 (m, 1H), 1.71-1.54 (m, 4H), 1.49-1.25 (m, 5H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ: 142.9 (C), 128.5 (2 × CH), 128.3 (2 × CH), 125.7 (CH), 102.5 (CH), 67.0 (2 × CH₂), 35.9 (CH₂), 35.3 (CH₂), 31.5 (CH₂), 29.2 (CH₂), 26.0 (CH₂), 23.9 (CH₂) ppm; HRMS (EI) calcd. for C₁₅H₂₁O₂ [M-H]: 233.1542; Found: 233.1545; Anal. calcd. for C₁₅H₂₁O₂: C, 76.88; H, 9.46; Found: C, 76.79; H, 9.39.

2-[5-(4-Methoxyphenyl)pentyl]-1,3-dioxane (2b): Product was obtained following the general



procedure and using 1-(3-iodopropyl)-4-methoxybenzene as starting iodide. The compound was purified by column chromatography in silica-gel (hexane/EtOAc 90:10) as colorless

oil with 67% yield. ¹H NMR (300 MHz, CDCl₃) δ 7.08 (d, *J* = 8.7 Hz, 2H), 6.82 (d, *J* = 8.7 Hz, 2H), 4.50 (t, *J* = 5.1 Hz, 1H), 4.09 (dd, *J* = 10.7 and 4.7 Hz, 2H), 3.78 (s, 3H), 3.75 (td, *J* = 12.3 and 2.5 Hz, 2H), 2.54 (t, *J* = 7.5 Hz, 2H), 2.16-1.97 (m, 1H), 1.65-1.51 (m, 4H), 1.45-1.25 (m, 5H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 157.8 (C), 135.0 (C), 129.4 (2 × CH), 113.8 (2 × CH), 102.5 (CH), 67.1 (2 × CH₂), 55.4 (CH₃), 35.3 (CH₂), 35.0 (CH₂), 31.7 (CH₂), 29.2 (CH₂), 26.0 (CH₂), 23.9 (CH₂) ppm; HRMS (EI) calcd. for C₁₆H₂₄O₃: 264.1725; Found: 264.1726; Anal. calcd. for C₁₆H₂₄O₃: C, 72.69; H, 9.15; Found C, 72.70; H, 9.01.

2-(4-Phenylbuty)-1,3-dioxane (2c): Product was obtained following the general procedure and using (2-iodoethyl)benzene as starting iodide. The compound **2c** was purified by column chromatography (hexane/EtOAc 95:5) as colorless oil with a 66% yield. ¹H-NMR (300 MHz, CDCl₃) δ : 7.32-7.24 (m, 2H), 7.21-7.14 (m, 3H), 4.52 (t, *J* = 5.1 Hz, 1H), 4.10 (ddd, *J* = 12.0, 5.0 and 1.1 Hz, 2H), 3.75 (dt, *J* = 12.0 and 2.5 Hz, 2H), 2.62 (t, *J* = 7.6 Hz, 2H), 2.19-1.98 (m, 1H), 1.70-1.58 (m, 4H), 1.52-1.39 (m, 2H), 1.38-1.26 (m, 1H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 142.7 (C), 128.5 (2 × CH), 128.4 (2 × CH), 125.7 (CH), 102.5 (CH), 67.0 (2 × CH₂), 36.0 (CH₂), 35.2 (CH₂), 31.5 (CH₂), 26.0 (CH₂), 23.9 (CH₂) ppm; HRMS (EI) calcd. for C₁₄H₁₉O₂ [M-H]: 219.1385; Found: 219.1386; Anal. calcd. for C₁₄H₂₀O₂: C, 76.33; H, 9.15; Found: C, 76.12; H, 9.13.

2-Butyl-1,3-dioxane (2d): Product was obtained following the general procedure and using 1-iodoethane as starting iodide. The compound **2d** was purified by column chromatography (pentane/Et₂O 95:5) as volatile colorless oil with a 59% yield. ¹H-NMR (300 MHz, CD₂Cl₂) δ : 4.47 (t, J = 5.6 Hz, 1H), 4.04 (ddd, J = 11.8, 4.9)

and 1.3 Hz, 2H), 3.72 (dt, J = 12.3 and 2.4 Hz, 2H), 2.08-1.90 (m, 1H), 1.59-1.47 (m, 2H), 1.39-1.22 (m, 5H), 0.89 (t, J = 6.8 Hz, 3H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 103.0 (CH), 67.4 (2 × CH₂), 35.5 (CH₂), 26.8 (CH₂), 26.6 (CH₂), 23.2 (CH₂), 14.3 (CH₃) ppm; HRMS (EI) calcd. for C₈H₁₅O₂ [M-H]: 143.1072; Found: 143.1071.

2-Hexyl-1,3-dioxane (2e): Product was obtained following the general procedure and using 1-iodobutane as starting iodide. The compound **2e** was purified by column chromatography (hexane/EtOAc 95:5) as colorless oil with a 60% yield. ¹H-NMR (300 MHz, CD₂Cl₂) δ : 4.50 (t, *J* = 5.3 Hz, 1H), 4.09 (ddd, *J* = 11.9, 4.7 and 1.1 Hz, 2H), 3.75 (dt, *J* = 12.4 and 2.6 Hz, 2H), 2.16-1.97 (m, 1H), 1.64-1.52 (m, 2H), 1.43-1.19 (m, 9H), 0.87 (t, *J* = 7.1 Hz, 3H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 102.6 (CH), 67.0 (2 × CH₂), 35.4 (CH₂), 31.9 (CH₂), 29.3 (CH₂), 26.0 (CH₂), 24.1 (CH₂), 22.7 (CH₂), 14.2 (CH₃) ppm; HRMS (EI) calcd. for C₁₀H₁₉O₂ [M-H]: 172.1385; Found: 172.1392; Anal. calcd. for C₁₀H₂₀O₂: C, 69.72; H, 11.70; Found: C, 69.61; H, 11.34.

2-Tetradecyl-1,3-dioxane (2f): Product was obtained following the general procedure and using **2f 1**-iodododecane as starting iodide. The compound was purified by column chromatography in silica-gel (hexane/EtOAc 97:3) as pale yellow oil (solid below 25 °C) with 60% yield. ¹H NMR (300 MHz, CDCl₃) δ 4.50 (t, *J* = 5.1 Hz, 1H), 4.09 (dd, *J* = 11.3 and 4.8 Hz, 2H), 3.75 (td, *J* = 12.3 and 2.5 Hz, 2H), 2.17-1.97 (m, 1H), 1.63-1.53 (m, 2H), 1.42-1.17 (m, 25H), 0.88 (t, *J* = 6.7 Hz, 3H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 102.7 (CH), 67.1 (2 × CH₂), 35.4 (CH₂), 32.1 (CH₂), 29.8 (5 × CH₂), 29.7 (3 × CH₂), 29.5 (CH₂), 26.0 (CH₂), 24.1 (CH₂), 22.8 (CH₂), 14.3 (CH₃) ppm; HRMS (EI) calcd. for C₁₈H₃₆O₂: 284.2715; Found: 284.2710; Anal. calcd. for C₁₈H₃₆O₂: C, 76.00; H, 12.76; Found C, 76.36; H, 12.26.

2-(2-Cyclohexylethyl)-1,3-dioxane (2g): Although this compound was previously described in the



literature,¹⁶ some controversial spectroscopy data were found. Product was obtained following the general procedure using iodocyclohexane as starting iodide. The compound was purified by column chromatography in silica-gel (hexane/EtOAc 95:5) as colorless oil with 88% yield. ¹H NMR (300 MHz,

CDCl₃) δ 4.48 (t, J = 5.2 Hz, 1H), 4.09 (dd, J = 11.8 and 5.0 Hz, 2H), 3.75 (td, J = 12.3 and 2.3 Hz, 2H), 2.16-1.98 (m, 1H), 1.76-1.52 (m, 7H), 1.38-1.07 (m, 7H), 0.95-0.77 (m, 2H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 103.0 (CH), 67.1 (2 × CH₂), 37.7 (CH), 33.4 (2 × CH₂), 32.8 (CH₂), 31.6 (CH₂), 26.8 (CH₂), 26.5 (2 × CH₂), 26.0 (CH₂) ppm; HRMS (EI) calcd. for C₁₂H₂₂O₂: 198.1620; Found: 198.1579; Anal. calcd. for C₁₂H₂₂O₂: C, 72.68; H, 11.18; Found C, 72.63; H, 11.22.

Methyl 7-(1,3-dioxan-2-yl)heptanoate (2h): Product was obtained following the general procedure and using methyl 5-iodovalerate as starting iodide. The compound was purified by column chromatography in silica-gel (hexane/EtOAc 90:10) as colorless oil with 57% yield. ¹H NMR (300 MHz, CDCl₃) δ 4.49 (t, J = 5.1 Hz, 1H), 4.09 (dd, J = 10.7 and 5.0 Hz, 2H), 3.75 (td, J = 12.4 and 2.4 Hz, 2H), 3.65 (s, 3H), 2.29 (t, J = 7.5 Hz, 2H), 2.14-1.98 (m, 1H), 1.70-1.50 (m, 4H), 1.45-1.23 (m, 7H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 174.4 (C), 102.5 (CH), 67.0 (2 × CH₂), 51.6 (CH₃), 35.3 (CH₂), 34.2 (CH₂), 29.2 (CH₂), 29.2 (CH₂), 26.0 (CH₂), 25.0 (CH₂), 23.9 (CH₂) ppm; HRMS (EI) calcd. for C₁₂H₂₂O₄: 230.1518; Found: 230.1506; Anal. calcd. for C₁₂H₂₂O₄: C, 62.58; H, 9.63; Found C, 62.99; H, 6.56.

cis- and *trans-*Ethyl 4-[2-(1,3-dioxan-2-yl)ethyl]cyclohexanecarboxylate 2i: Products were obtained following the general procedure and using a 34:66 mixture of both isomers of ethyl 4-iodocyclohexanecarboxylate as starting iodides. Compounds were purified by column chromatography in silica-gel (hexane/EtOAc 90:10) as colorless oil with 65% combined

yield (22:78 mixture of both isomers). ¹H NMR (300 MHz, CDCl₃, signals of major isomer) δ : 4.46 (t, J = 5.1 Hz, 1H), 4.14-4.03 (m, 4H), 3.73 (td, J = 12.4 and 1.8 Hz, 2H), 2.25-1.87 (m, 4H), 1.86-1.73 (m, 2H), 1.66-1.14 (m, 11H), 1.00-0.80 (m, 2H) ppm; ¹³C NMR (75 MHz, CDCl₃, signals of major isomer) δ : 176.4 (C), 102.7 (CH), 67.1 (2 × CH₂), 60.2 (CH₂), 43.7 (CH₂), 36.9 (CH₂), 32.8 (CH₂), 32.3 (2 × CH₂), 31.3 (CH₂), 29.1 (2 × CH₂), 26.0 (CH₂), 14.4 (CH₃) ppm; HRMS (EI) calcd. for C₁₅H₂₅O₄ [M-H]: 269.1753; Found: 269.1744.

2-[3-(Tetrahydro-2H-pyran-2-yl)propyl]-1,3-dioxane (2j): Product was obtained following the general procedure and using 2-(iodomethyl)tetrahydro-2H-pyran as starting iodide. The compound was purified by column chromatography (hexane/EtOAc 80:20) as colorless oil with a 63% yield. ¹H-NMR (300 MHz,

CDCl₃) δ : 4.49 (t, J = 5.0 Hz, 1H), 4.07 (ddd, J = 11.8, 5.0 and 1.2 Hz, 2H), 3.97-3.88 (m, 1H), 3.73 (dt, J = 12.5 and 2.6 Hz, 2H), 3.37 (dt, J = 11.3 and 2.6 Hz, 1H), 3.25-3.12 (m, 1H), 2.14-1.95 (m, 1H), 1.82-1.73 (m, 1H), 1.62-1.13 (m, 12H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 102.5 (CH),

77.8 (CH), 68.6 (CH₂), 67.0 (2 × CH₂), 36.5 (CH₂), 35.3 (CH₂), 32.0 (CH₂), 26.3 (CH₂), 26.0 (CH₂), 23.7 (CH₂), 20.2 (CH₂) ppm; HRMS (EI) calcd. for $C_{12}H_{21}O_3$ [M-H]: 213.1491; Found: 213.1484; Anal. Calcd. for $C_{13}H_{24}O_3$: C, 68.38; H, 10.59; Found: C, 68.25; H, 10.45.

2-[2-(2,2-Dimethyltetrahydro-2H-pyran-4-yl)ethyl]-1,3-dioxane (2k): Product was obtained following the general procedure and using 4-iodo-2,2-dimethyltetrahydro-2H-pyran as starting iodide. The compound was purified by column chromatography (hexane/EtOAc 80:20) as colorless oil with an 83% yield. ¹H-NMR (300 MHz, CDCl₃) δ : 4.49 (t, *J* = 5.0 Hz, 1H), 4.10 (dd, *J* = 11.4 and 5.0 Hz, 2H), 3.82-3.56 (m, 4H), 2.18-1.97 (m, 1H), 1.69-1.47 (m, 5H), 1.39-1.23 (m, 3H), 1.92-0.97 (m, 2H), 1.19 (s, 3H), 1.18 (s, 3H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 102.5 (CH), 71.7 (CH), 67.0 (2 × CH₂), 61.7 (CH₂), 43.7 (CH₂), 32.8 (CH₂), 32.3 (CH), 31.8 (CH₂), 31.4 (CH₃), 31.1 (CH₂), 25.9 (CH₂), 22.0 (CH₂) ppm; HRMS (EI) calcd. for C₁₃H₂₃O₃ [M-H]: 227.1647; Found: 227.1657; Anal. calcd. for C₁₂H₂₂O₃: C, 67.26; H, 10.35; Found: C, 67.16; H, 10.01.

2-[5-(Tetrahydro-2H-pyran-2-yloxy)pentyl]-1,3-dioxane (21): Product was obtained following the general procedure and using 2-(3-iodopropoxy)tetrahydro-2H-pyran as starting iodide. The compound was purified by column chromatography in silica-gel (hexane/EtOAc 80:20) as colorless oil with 54% yield. ¹H NMR (300 MHz, CDCl₃) δ 4.56 (t, *J* = 3.5 Hz, 1H), 4.51 (t, *J* = 5.1 Hz, 1H), 4.09 (dd, *J* = 11.0 and 4.9 Hz, 2H), 3.91-3.81 (m, 1H), 3.80-3.67 (m, 3H), 3.54-3.44 (m, 1H), 3.38 (dt, *J* = 9.6 and 6.6 Hz, 1H), 2.15-1.99 (m, 1H), 1.86-1.30 (m, 15H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 102.5 (CH), 99.0 (CH), 67.7 (CH₂), 67.0 (2 × CH₂), 62.5 (CH₂), 35.3 (CH₂), 30.9 (CH₂), 29.8 (CH₂), 26.3 (CH₂), 26.0 (CH₂), 25.7 (CH₂), 24.0 (CH₂), 19.8 (CH₂) ppm; HRMS (EI) calcd. for C₁₄H₂₅O₄ [M-H]: 257.1753; Found: 257.1743; Anal. calcd. for C₁₄H₂₆O₄: C, 65.09; H, 10.14; Found C, 64.73; H, 9.88.

tert-Butyl 4-[2-(1,3-dioxan-2-yl)ethyl]piperidine-1-carboxylate (2m): Product was obtained following the general procedure and using *tert*-butyl 4-iodopiperidine-1carboxylate as starting iodide. The compound was purified by column chromatography (hexane/EtOAc 80:20) as colorless oil with a 67% yield. ¹H-NMR (300 MHz, CDCl₃) δ : 4.49 (t, J = 5.0 Hz, 1H), 4.15-3.95 (m, 4H), 3.75 (dt, J = 12.3 and

2.3 Hz, 2H), 2.65 (t, J = 12.3 Hz, 2H), 2.19-1.97 (m, 1H), 1.71-1.52 (m, 5H), 1.51-1.23 (m, 3H), 1.44 (s, 9H), 1.16-0.97 (m, 2H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 154.9 (C), 102.5 (CH), 79.2

(C), 67.0 (2 × CH₂), 44.1 (2 × CH₂), 36.0 (CH), 32.5 (CH₂), 32.1 (CH₂), 30.6 (CH₂), 28.5 (2 × CH₃), 25.9 (CH₂) ppm; HRMS (EI) calcd. for $C_{11}H_{20}NO_2$ [M-CO₂tBu]: 198.1494; Found: 198.1497; Anal. calcd. for $C_{16}H_{29}NO_4$: C, 64.18; H, 9.76; N, 4.68; Found: C, 64.09; H, 9.51; N, 4.33.

4-[2-(1,3-Dioxan-2-yl)ethyl]-1-benzylpiperidine (2n): Product was obtained following the general procedure and using 1-benzyl-4-iodopiperidine as starting iodide. Ph/ `N^ compound was purified by column chromatography The 2n (hexane/EtOAc/Et₃N 80:20:1) as colorless oil with a 78% yield. ¹H-NMR $(300 \text{ MHz}, \text{CDCl}_3)$ δ : 7.34-7.17 (m, 5H), 4.48 (t, J = 5.0 Hz, 1H), 4.09 (dd, J = 11.2 and 4.7 Hz, 2H), 3.74 (dt, J = 12.1 and 2.1 Hz, 2H), 3.47 (s, 2H), 2.85 (d, J = 11.2 Hz, 2H), 2.16-1.97 (m, 1H), 1.91 (t, J = 10.6 Hz, 2H), 1.72-1.50 (m, 4H), 1.40-1.12 (m, 6H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ: 138.7 (C), 129.3 (2 × CH), 128.2 (2 × CH), 126.9 (CH), 102.7 (CH), 67.0 (2 × CH₂), 63.6 (CH_2) , 54.0 (2 × CH₂), 35.8 (CH), 32.8 (CH₂), 32.4 (2 × CH₂), 30.8 (CH₂), 25.9 (CH₂) ppm; HRMS (EI) calcd. for C₁₈H₂₆NO₂ [M-H]: 288.1964; Found: 288.1963.

2-[4-(1,3-Dioxan-2-yl)butyl]-1-tosyl-1H-indole (20): Product was obtained following the general

procedure and using 2-(2-iodoethyl)-1-tosyl-1H-indole as starting iodide. The compound was purified by column chromatography (hexane/Acetone 90:10) as pale yellow oil with a 36% yield. ¹H-NMR (300 MHz, CDCl₃) δ : 8.16 (d, J = 8.3 Hz, 1H), 7.60 (d, J = 8.3 Hz, 2H), 7.40 (d, J = 8.3 Hz, 1H), 7.28-7.13 (m, 4H), 6.38 (s, 1H), 4.54 (t, J = 5.2 Hz, 1H), 4.11 (dd, J = 11.4 and 5.2 Hz, 2H), 3.76 (dt, J = 12.1 and 1.7 Hz, 2H), 2.98 (t, J = 7.6 Hz 2H), 2.33 (s, 3H), 2.10-1.99 (m, 1H), 1.84-1.44 (m, 6H), 1.39-1.23 (m, 1H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 144.7 C), 142.4 (d), 137.4 (C), 136.4 (C), 130.0 (C), 129.9 (2 × CH), 126.4 (2 × CH), 123.9 (CH), 123.6 (CH), 120.2 (CH), 115.0 (CH), 108.8 (CH), 102.3 (CH), 67.0 (2 × CH₂), 35.1 (CH₂), 29.1 (CH₂), 28.9 (CH₂), 26.0 (CH₂), 23.9 (CH₂), 21.7 (CH₃) ppm; HRMS (EI) calcd. for C₂₃H₂₇NO₄S [M]: 413.1661; Found: 413.1677.

2-(2-Cholesterylethyl)-1,3-dioxane (2p): Product was obtained following the general procedure



and using cholesteryl iodide as starting iodide. The compound was purified by column chromatography in silica-gel (hexane/EtOAc 97:3) as white solid with 75% yield. Mp: 114-116 °C (acetone); $[\alpha]_D^{22} = -19$ (*c* 0.66, CHCl₃); ¹H NMR (300 MHz, CDCL₃) δ 5.30-5.23 (m, 1H), 4.49 (t, *J* = 5.2 Hz, 1H), 4.10 (dd, *J* = 10.7 and 5.0 Hz,

Ме Ме

2H), 3.75 (td, J = 12.4 and 2.5 Hz, 2H), 2.17-0.82 (m, 45H), 0.67 (s, 3H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 143.5 (C), 119.4 (CH), 102.8 (CH), 67.1 (2 × CH₂), 57.0 (CH), 56.3 (CH), 50.7 (CH), 42.5 (C), 40.0 (CH₂), 39.7 (CH₂), 39.7 (2 × CH₂), 39.4 (CH), 37.4 (C), 36.4 (CH₂), 36.0 (CH), 32.8 (CH₂), 32.1 (CH and CH₂), 31.4 (CH₂), 29.2 (CH₂), 28.4 (CH₂), 28.2 (CH), 26.0 (CH₂), 24.4 (CH₂), 24.0 (CH₂), 23.0 (CH₃), 22.7 (CH₃), 21.1 (CH₂), 19.6 (CH₃), 18.9 (CH₃), 12.0 (CH₃) ppm; HRMS (EI) calcd. for C₃₃H₅₆O₂: 484.4280; Found: 484.4267; Anal. calcd. for C₃₃H₅₆O₂: C, 81.76; H, 11.64; Found C, 82.18; H, 11.21.

4-Benzyl-2,2-dimethyltetrahydro-2H-pyran (2q): Product was obtained following the general procedure and using 4-iodo-2,2-dimethyltetrahydro-2H-pyran as starting iodide and benzylmagnesium chloride (0.5 M in THF) as Grignard reagent. The compound was purified by column chromatography in silica-gel 2q (hexane/EtOAc 97:3) as colorless oil with 52% yield. ¹H-NMR (300 MHz,

 $CDCl_3$) δ : 7.33-7.25 (m, 2H), 7.24-7.11 (m, 3H), 3.72 (ddd, J = 11.9, 5.3 and 1.6 Hz, 1H), 3.60 (dt, J = 11.9 and 1.9 Hz, 1H), 2.49 (d, J = 7.4 Hz, 2H), 2.01-1.83 (m, 1H), 1.63-1.45 (m, 2H), 1.29-1.06 (m, 2H), 1.20 (s, 3H), 1.16 (s, 3H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ: 140.3, 129.2, 128.3, 126.0, 71.8, 61.7, 44.0, 43.6, 33.4, 32.7, 31.9, 22.0 ppm; HRMS (EI) calcd. for C₁₄H₂₀O [M]: 204.1514; Found: 204.1518.

2-(3-Cyclopentylpropyl)-1,3-dioxane (3a) and 2-(oct-7-enyl)-1,3-dioxane (3b): Products were

obtained following the general procedure as a 1:1.4 mixture of 3a and 3b using 6-iodohex-1-ene as starting iodide. The compounds

were purified by column chromatography in silica-gel (hexane/EtOAc 95:5) as colorless oil with 36% combined yield. ¹H NMR (300 MHz, CDCl₃, 3a + 3b) δ 5.80 (ddt, J = 16.9, 10.2 and 6.7 Hz, 1H), 4.95 (m, 2H), 4.50 (t, J = 5.1 Hz, 2H), 4.10 (dd, J = 11.0 and 4.9 Hz, 4H), 3.75 (td, J = 12.3and 2.1 Hz, 4H), 2.19-0.98 (m, 31H) ppm; ¹³C NMR (75 MHz, CDCl₃, **3a** + **3b**) δ 139.2 (CH), 114.1 (CH₂), 102.5 (CH), 102.5 (CH), 66.9 ($4 \times CH_2$), 40.1(CH), 36.1 (CH₂), 35.5 (CH₂), 35.2 (CH₂), 33.7 (CH₂), 32.7 (2 × CH₂), 29.3(CH₂), 29.0 (CH₂), 28.8 (CH₂), 25.9 (2 × CH₂), 25.2 (2 × CH₂), 23.9 (CH₂), 23.2(CH₂) ppm; HRMS (EI) calcd. for C₁₂H₂₂O₂ (**3a**): 198,1620; Found: 198.1560 HRMS (EI) calcd. for C₁₂H₂₂O₂ (**3b**): 198.1620; Found: 198.1610.

2-(Hex-5-en-1-vl)-1.3-dioxane 4a: Product was obtained following the general procedure and using (iodomethyl)cyclopropane as starting iodide. The compound was purified by column chromatography (pentane/Et₂O 95:5) as volatile colorless 4a

oil with a 63% yield. ¹H-NMR (300 MHz, acetone- d_6) δ : 5.94-5.74 (m, 1H), 5.03 (dd, J = 17.2 and 1.8 Hz, 1H), 4.95 (dd, J = 10.2 and 1.0 Hz, 1H), 4.57-4.50 (m, 1H), 4.04 (ddd, J = 11.4, 4.8 and 1.0 Hz, 2H), 3.76 (dt, J = 12.0 and 1.9 Hz, 2H), 2.14-1.86 (m, 1H), 1.62-1.28 (m, 7H) ppm; ¹³C-NMR (75 MHz, acetone- d_6) δ : 139.7 (CH), 114.8 (CH₂), 102.8 (CH), 67.2 (2 × CH₂), 35.9 (CH₂), 34.4 (CH₂), 29.6 (CH₂), 26.7 (CH₂), 24.2 (CH₂) ppm; HRMS (EI) calcd. for C₁₀H₁₇O₂ [M-H]: 169.1229; Found: 169.1226.

2-Ethyl-1,3-dioxane¹² I: MgSO₄ (2 g) and 4-methylbenzenesulfonic acid monohydrate (26 mg, 0.13 mmol) were added to a solution of acetaldehyde (0.8 g, 13.77 mmol) and 1,3propanediol (1.05 g, 13.77 mmol) in CH₂Cl₂ (10 mL). Reaction was stirred for 48 h at RT and filtered through a short pad of Celite. The filtrate was carefully evaporated and

product was purified by column chromatography (pentane/Et₂O 95:5) affording the product as highly volatile colorless oil with a 50% yield. ¹H-NMR (300 MHz, acetone- d_6) δ : 4.44 (t, J = 4.8 Hz, 1H), 4.02 (ddd, J = 11.9, 5.0 and 1.3 Hz, 2H), 3.73 (dt, J = 11.9 and 2.1 Hz, 2H), 2.03-1.84 (m, 1H), 1.52 (dq, J = 7.8 and 5.2 Hz, 2H), 0.88 (t, J = 7.8 Hz, 3H) ppm; ¹³C-NMR (75 MHz, acetone- d_6) δ : 103.9 (CH), 67.5 (2 × CH₂), 29.3 (CH₂), 27.0 (CH₂), 8.8 (CH₃) ppm; MS (EI) *m/z* 115 (M-H), 87 (100%).

2-Vinyl-1,3-dioxane¹³ II: MgSO₄ (4 g) and 4-methylbenzenesulfonic acid monohydrate (71 mg, 0.37 mmol) were added to a solution of acryaldehyde (2.1 g, 37.5 mmol) and 1,3propanediol (2.85 g, 37.5 mmol) in CH₂Cl₂ (20 mL). Reaction was stirred for 48 h at RT and filtered through a short pad of Celite. The filtrate was carefully evaporated and product was purified by column chromatography (pentane/Et₂O 95:5) affording the product as highly volatile colorless oil with a 32% yield. ¹H-NMR (300 MHz, CDCl₃) δ : 5.92-5.77 (m, 1H), 5.46 (d, *J* = 17.2 Hz, 1H), 5.29 (d, *J* = 10.7 Hz, 1H), 4.96 (d, *J* = 4.3 Hz, 1H), 4.16 (dd, *J* = 11.7 and 4.3 Hz, 2H), 3.84 (t, *J* = 12.0 Hz, 2H), 2.23-2.03 (m, 1H), 1.42-1.32 (m, 1H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 135.1 (CH), 118.5 (CH₂), 100.8 (CH), 67.1 (2 × CH₂), 25.9 (CH₂) ppm; MS (EI) *m/z* 113 (M-H), 87 (100%).

1,4-Di(1,3-dioxanyl)butane III: According to the procedure reported in literature,¹⁵ compound **III** was synthesized as follows: A round bottom flask containing silver triflate (4.2 mg, 0.015 mmol) was purged three times with an argon/vacuum cycle. Dry THF (2 mL) was added to the flask followed by the addition of 1,2-

dibromoethane (338 mg, 1.8 mmol). Reaction was stirred at RT under argon and then (1,3-dioxan-2-ylethyl)magnesium bromide (3 mL. 1.5 mmol) was slowly added (about 1 h). After stirring for 30 min, the reaction was hydrolyzed with a saturated solution of NH_4Cl (2 mL). Phases were

separated and the aqueous phase was extracted with CH_2Cl_2 (3 × 5 mL). The combined organic phases were drier over MgSO₄ and solvent was evaporated under vacuum. The residue was purified by column chromatography (hexane/EtOAc 70:30) affording the product as colorless oil with a 38% yield. ¹H-NMR (300 MHz, CDCl₃) δ : 4.50 (t, *J* = 4.7 Hz, 2H), 4.09 (dd, *J* = 10.9 and 5.2 Hz, 4H), 3.74 (dt, *J* = 12. and 2.1 Hz, 4H), 2.16-1.95 (m, 2H), 1.65-1.51 (m, 4H), 1.44-1.25 (m, 6H) ppm; ¹³C-NMR (75 MHz, CDCl₃) δ : 102.2 (2 × CH), 66.8 (4 × CH₂), 35.1 (2 × CH₂), 25.8 (2 × CH₂), 23.9 (2 × CH₂) ppm; MS (EI) *m/z* 229 (M-H), 171, 127, 87 (100%).

5. References

[1] G. R. Fulmer, A. J. M. Miller, N.H. Sherden, H. E. Gottlieb, A. Nudelman, B. M. Stoltz, J. E. Bercaw, K. I. Goldberg, *Organometallics* **2010**, *29*, 2176.

[2] S. M. Smith, J. M. Takacs, J. Am. Chem. Soc. 2010, 132, 1740.

[3] C. Dai, J. M. R. Narayanam, C. R. J. Stephenson, Nat. Chem. 2011, 3, 140.

[4] J. P. Collman, Y. Yang, R. A. Decréau, Org. Lett. 2007, 9, 2855.

[5] A. Amjad, N. Ravi, L. M. Man-Chu, Y. Lin, H. Pei, F. Christopher, PCT Int. Appl., 2011053519, 2011.

[6] H. Firouzabadi, N. Iranpoor, S. Kazemi, Can. J. Chem. 2009, 87, 1675.

[7] G. Sabitha, K. B. Reddy, M. Bhikshapathi, J. S. Yadav, Tetrahedron Lett. 2006, 47, 2807.

[8] F. Louafi, J. Moreau, S. Shahane, S. Golhen, T. Roisnel, S. Sinbandhit, J.-P. Hurvois, *J. Org. Chem.* 2011, 76, 9720.

[9] E. G. Corley, K. Conrad, J. A. Murry, C. Savarin, J. Holko, G. Boice, *J. Org. Chem.* 2004, 69, 5120.

[10] S. F. McCann, L. E. Overman, J. Am. Chem. Soc. 1987, 109, 6107.

[11] H. Adachi, K. K. Palaniappan, A. A. Ivanov, N. Bergman, Z.-G. Gao, K. A. Jacobson, *J. Med. Chem.* **2007**, *50*, 1810.

[12] N. Ortega, A. Feher-Voelger, M. Brovetto, J. I. Padrón, V. S. Martín, T. Martín, *Adv. Synth. Catal.* 2011, 353, 963.

[13] J. Hao, H. Liu, D. Liu, Ing. Eng. Chem. Res. 2005, 44, 4380.

[14] J. Cheng, R. Ji, S.-J. Gao, F.-S. Du, Z.-C. Li, Biomacromolecules 2012, 13, 173.

[15] T. Nagano, T. Hayashi, Chem. Lett. 2005, 34, 1152.

[16] M. P. Muñoz, M. C. de la Torre, M. A. Sierra, Adv. Synth. Catal. 2010, 352, 2189; Y.-S. Lee,

L. del Valle, G. L. Larson, Synth. Commun. 1987, 17, 385.

6. Copy of ¹H RMN and ¹³C NMR.





Electronic Supplementary Material (ESI) for Chemical Science This journal is The Royal Society of Chemistry 2013





Electronic Supplementary Material (ESI) for Chemical Science This journal is O The Royal Society of Chemistry 2013



Electronic Supplementary Material (ESI) for Chemical Science This journal is The Royal Society of Chemistry 2013











s \$1\$ 11 x x 1 2j



T

and the second second







T

1





Boc_N 2m



T

1.6.1

T



1

I

I.





Electronic Supplementary Material (ESI) for Chemical Science This journal is The Royal Society of Chemistry 2013







Electronic Supplementary Material (ESI) for Chemical Science This journal is O The Royal Society of Chemistry 2013

4a / / / /



