Supporting Information

C–H Activation Route to Dibenzo[*a*,*e*]pentalenes: Annulation of Arylacetylenes Promoted by PdCl₂/AgOTf/*o*-chloranil

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1. General

Unless otherwise noted, all reagents including anhydrous solvents were obtained from commercial suppliers and used without further purification. Substituted diphenylacetylenes 1b, 1c, 1d, 1e were prepared according to literature procedure.^[1] 2,6,2',6'-Tetradeuteriodiphenyletyne **1a**-*d* was prepared from deuteriobromobenzene, which was prepared according to literature procedure.^[2,3] AgOTf was stored in a glove box filled by argon prior to use. o-chloranil was recrystallized from benzene prior to use. Column chromatography was performed with silica gel 60 (230-400 mesh). Preparative thin-layer chromatography (PTLC) was performed using silica-coated plates prepared in our laboratory (325 mesh, 5% binder, 0.75 mm thickness). Synthetic manipulations that required an inert atmosphere (where noted) were carried out under nitrogen using standard Schlenk techniques or in an inert-atmosphere glove box. Analythical thin-layer chromatograghy (TLC) was performed using E. Merck silica gel 60 F₂₅₄ precoated plates (0.25 mm). The developed chromatogram was analyzed by UV lamp (254 nm and 365 nm), ethanolic phosphomolybdic acid. Gas chromatography (GC) was conducted on a Shimadzu GC-2010 instrument equipped with a HP-5 column $(30 \text{ m} \times 0.25 \text{ mm}, \text{Hewlett-Packard})$. GC yields are expressed vs. *n*-undecane as an internal standard. High-resolution mass spectra (HRMS) were obtained from a JMS-T100TD instrument (DART). Elemental analysis were recorded on Yanaco MT-6. Nuclear magnetic resonance (NMR) spectra were recorded on a JEOL ECA-600 (¹H NMR 600 MHz, ¹³C NMR 150 MHz) spectrometers. Chemical shifts for ¹H NMR are expressed in parts per million (ppm) relative to solvent signal ¹H (CHCl₃: 7.26 ppm; CHDCl₂: 5.30 ppm) and 13 C (CDCl₃: 77.0 ppm; CD₂Cl₂: 53.8 ppm).

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2. General procedure



To a heat-dried screw-capped test tube were added a magnetic stirring bar, **1** (400 μ mol), PdCl₂ (10.6 mg, 60 μ mol) and *o*-chloranil (97.2 mg, 400 μ mol). The screw-capped test tube was introduced inside an argon atmosphere glovebox. To the tube was added AgOTf (102 mg, 400 μ mol). The screw-capped tube was taken out of the glovebox, and to it was added anhydrous DMAc (6 mL). The test tube was sealed and the resulting mixture was stirred at 60 °C for 3 days. After cooling to room temperature, contents were diluted with CHCl₃. The organic layer was passed on a short silica plug, eluted with CHCl₃ and the solvents were evaporated under reduced pressure to afford the crude reaction mixture. The crude reaction mixture was extracted by 20 ml CHCl₃ in two times to remove residual DMAc and purified by silica gel column chromatography to afford product. For most products, the reaction could be performed by low catalyst loading with slight loss of product yield. In those cases, PdCl₂ (3.54 mg, 20 μ mol, 5 mol%) was used and the reaction mixture was stirred at 80 °C for 18 h.



Following general procedure with **1a** (71 mg, 400 μ mol). The crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/CHCl₃ as the mobile phase to afford **2a** as a red solid (36 mg, 52%). The product was recrystallized with CHCl₃/pentane

^{2a} at room temperature to obtain red needle crystal for X-ray crystallography. ¹H-NMR (600 MHz, CDCl₃): $\delta = 6.85$ (1H, td, J = 8, 1 Hz), 6.90 (1H, td, J = 8, 1 Hz), 7.03 (1H, d, J = 7 Hz), 7.21 (1H, d, J = 7 Hz), 7.45 (1H, tt, J = 7, 1 Hz), 7.52 (2H, t, J = 8 Hz), 7.67 (2H, dd, J = 8, 1 Hz); ¹H-NMR (600 MHz, CD₂Cl₂): $\delta = 6.86$ (1H, t, J = 7 Hz), 6.92 (1H, t, J = 7 Hz), 7.03 (1H, d, J = 7 Hz), 7.21 (1H, d, J = 7 Hz), 7.47 (1H, t, J = 7 Hz), 7.54 (2H, t, J = 7 Hz), 7.68 (2H, d, J = 7 Hz); ¹³C-NMR (150 MHz, CDCl₃): 121.9, 122.5, 127.4, 127.8, 128.5, 128.7, 128.8, 133.9, 135.2, 140.7, 143.1, 149.7; HR-MS (DART MS) *m/z* calcd. for C₂₈H₁₁ [MH]⁺: 355.14868, found: 355.14849; mp: 259 °C. Local maxima of UV-Vis absorption: 428, 448 nm.



Following general procedure with **1b** (124 mg, 400 μ mol). The crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/CHCl₃ as the mobile phase to afford **2b** as a red solid (99.2 mg, 79%). The product was recrystallized with CHCl₃/pentane at room temperature to obtain red needle crystal for X-ray crystallography. ¹H-NMR (600 MHz, CDCl₃): $\delta = 7.19$ (1H, s), 7.21 (1H,

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d, J = 8 Hz), 7.24 (1H, d, J = 8 Hz), 7.77 (2H, d, J = 8 Hz), 7.86 (2H, d, J = 8 Hz); C¹³ NMR (150 MHz, CDCl₃, 60 °C) 119.4 (q, J = 3 Hz), 122.2, 123.8 (q, J = 272 Hz), 125.8 (q, J = 5 Hz), 126.3 (q, J = 5 Hz), 128.7, 131.0, 131.2, 132.0 (q, J = 33 Hz), 136.2, 137.6, 141.1, 143.9, 149.6; HR-MS (DART MS) *m/z* calcd. for C₃₂H₁₅F₁₂ [MH]⁺: 627.09821, found: 627.09869; mp: not melt or decomposed. Local maxima of UV-Vis absorption: 424, 450 nm.



Following general procedure with 1c (126 mg, 400 µmol). The crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/CHCl₃ as the mobile phase to afford 2c as a red solid (65.9 mg, 53%). ¹H-NMR (600 MHz, CD₂Cl₂): δ = 7.17 (1H, d, J = 8 Hz), 7.31 (1H, d, J = 8 Hz), 7.38 (1H, s), 7.77 (1H, t, J = 8 Hz), 7.83 (1H, d, J = 8 Hz), 7.97 (1H, d, J = 1 Hz); HR-MS (DART MS) m/z calcd. for

 $[MH]^+$: 627.09821, found: 627.09830; mp: 259 °C; elemental analysis calcd. for $C_{32}H_{14}F_{12}$: C 61.35, H 2.25, F 36.39 :found C 61.18, H 2.13. Local maxima of UV-Vis absorption: 423, 446 nm.



Following general procedure with **1e** (95.6 mg, 400 μ mol). The crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/CHCl₃ as the mobile phase to afford **2e** as a brown solid (63.3 mg, 66%). The product was recrystallized with CHCl₃/pentane at room temperature to obtain brown needle crystal for X-ray crystallography. ¹H-NMR (600 MHz, CDCl₃): $\delta = 3.71$ (3H, s), 3.87 (3H, s), 6.38

(1H, dd, J = 8, 2 Hz), 6.88 (1H, d, J = 2 Hz), 6.95 (1H, d, J = 8 Hz), 6.97 (1H, ddd, J = 8, 3, 1 Hz), 7.19 (1H, dd, J = 3, 2 Hz), 7.24 (1H, ddd, J = 8, 3, 2 Hz), 7.42 (1H, t, J = 8 Hz); ¹³C NMR (150mHz, CDCl₃): 55.4 (2C), 110.0, 110.2, 113.4, 114.7, 120.8, 123.1, 129.8, 135.3 (4°), 137.1 (4°), 141.0 (4°), 141.3 (4°), 142.0 (4°), 159.8 (4°), 160.1 (4°); HR-MS (DART MS) m/z calcd. for $C_{32}H_{27}O_4$ [MH]⁺: 475.19093, found: 475.19091; mp: decomposed at 180 °C. Local maxima of UV-Vis absorption: 453, 474 nm.



Following general procedure with **1f** (120 mg, 400 μ mol). The crude product was purified by silica gel column chromatography using 2:1 (v/v) hexane/EtOAc as the mobile phase and then recrystallized with CHCl₃/pentane at room temperature to afford **2f** as a violet block crystal (18.0 mg, 15%). ¹H-NMR (600 MHz, CDCl₃): $\delta = 3.56$ (3H, s), 3.65 (3H, s), 3.81 (6H, s), 5.98 (1H, s), 6.43 (1H, s), 6.48 (1H, s),

6.76 (2H, d, J = 1 Hz); ¹³C NMR (150MHz, CDCl₃): 55.3 (2C), 55.4, 97.5, 101.1, 101.9, 106.7, 127.7 (4°), 137.4 (4°), 138.7 (4°), 140.5 (4°), 142.6 (4°), 156.0 (4°), 159.9 (4°), 161.7 (4°); HR-MS (DART MS) m/z calcd. for C₃₆H₃₅O₈ [MH]⁺: 595.23319, found: 595.23329; mp: 251-254°C. Local maxima of UV-Vis absorption: 494, 507 nm.

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Following low catalyst loading procedure with **1g** (66 μ L, 400 μ mol). The crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/EtOAc as the mobile phase to afford **2g** as a brown solid (28.7 mg, 83 μ mol). **2g** was recrystallized with CHCl₃/pentane for X-ray crystallography. ¹H-NMR (600 MHz, CDCl₃): $\delta = 1.45$ (3H, t, J = 7 Hz), 4.42 (2H, q, J = 7 Hz), 6.92 (1H, td, J = 8, 1 Hz), 6.98 (1H, td, J = 8, 1 Hz), 7.48 (1H, d,

J = 7 Hz), 7.49 (1H, d, J = 7 Hz); ¹³C-NMR (150 MHz, CDCl₃): 14.3, 61.1, 125.0, 127.4, 128.4, 130.3, 130.6, 132.5, 147.7, 153.6, 163.5; HR-MS (DART MS) m/z calcd. for C₂₂H₁₉O₄ [MH]⁺: 347.12833, found: 347.12833; mp: 140-142 °C.



Following general procedure with **1d** (62.8 mg, 200 μ mol) and **1b** (47.6 mg, 200 μ mol). The crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/CHCl₃ as the mobile phase to afford **2d** as a red solid (25.8 mg, 41 μ mol) and **2bd** as a black solid (29.1 mg, 53 μ mol). **2bd** was recrystallized with CHCl₃/pentane for X-ray crystallography. ¹H-NMR (600 MHz, CDCl₃): $\delta = 3.75$ (3H, s), 3.91 (3H, s),

6.42 (1H, d, J = 8 Hz), 6.58 (1H, s), 7.10 (4H, m), 7.19 (1H, s), 7.21 (1H, d, J = 8 Hz), 7.61 (2H, d, J = 8 Hz), 7.73 (2H, d, J = 8 Hz), 7.79 (2H, d, J = 8 Hz) ¹³C-NMR (150 MHz, CD₂Cl₂): 30.1, 55.9, 111.1, 111.9, 114.9 111.8, 121.6, 123.6, 124.8, 125.7, 126.3, 127.7, 129.4, 130.2, 131.3, 137.7, 138.7, 140.1, 143.5, 144.5, 151.0, 151.4, 161.0; HR-MS (DART MS) m/z calcd. for C₃₂H₂₁F₆O₂ [MH]⁺: 551.14457, found: 551. 14449; mp: 255-257 °C. Local maxima of UV-Vis absorption: 433, 452 nm.

3. Procedure for deuterium-labeling experiment



To a Schlenk tube containing a magnetic stirring bar were added 2-butynedioic acid and $PdCl_2(PPh_3)_2$ (40 mg, 57 µmol), 1,4-bis(diphenylphosphino)butane (49 mg, 114 µmol). Then, the Schlenk tube was sealed by septum and the contents were evacuated, then back-filled with argon gas. After dry DMSO (5 mL), 1,8-diazabicycloundec-7-ene (346 mg, 2.3 mmol) and 1-bromo-2,6-deuteriobenzene^[2,3] (364 mg, 2.3 mmol) were added via syringe through septum. The resultant mixture was stirred at 110 °C for 6 h. After the reaction mixture was cooled to room temperature, diluted with Et₂O (ca. 15 ml) and saturated aqueous solution of NH₄Cl (50 ml). Then, the mixture solution was extracted with Et₂O (20 ml × 2). The combined organic phase was subjected to silica gel column chromatography using 100% hexane as the mobile phase to afford **1h** as white solid (103 mg, 50%).

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Following general procedure with **1a**-*d* (69 mg, 400 mmol). The reaction mixture was stirred at 80 °C for 12 h. And crude product was purified by silica gel column chromatography using 10:1 (v/v) hexane/CHCl₃ as the mobile phase and to afford **2a**-*d* (10 %, 6.8 mg).



1H), 7.72 (s, 2H); HR-MS (DART MS) *m*/*z* calcd. for C₂₈H₁₃D₆ [MH]⁺: 361.18634, found: 361.18631.

4. Procedure for trapping experiment with methyl acrylate

2a-c



To a heat-dried screw-capped test tube were added a magnetic stirring bar, **1a** (35.6 mg, 200 μ mol), PdCl₂ (3.5 mg, 20 μ mol) and *o*-chloranil (97.2 mg, 400 μ mol). The screw-capped test tube was introduced inside an argon atmosphere glovebox. To the tube was added AgOTf (102 mg, 400 μ mol). The screw-capped tube was taken out of the glovebox, and to it was added anhydrous DMAc (1 mL) and methylacrylate (22 μ L, 200 μ mol). The test tube was sealed and the resulting mixture was stirred at 80 °C for 10 h. After cooling to room temperature, contents were diluted with CHCl₃. The organic layer was passed on a short silica plug, eluted with EtOAc and the solvents were evaporated under reduced pressure to afford the crude reaction mixture. The crude reaction mixture was extracted by 20 ml EtOAc in two times to remove residual DMAc and purified by silica gel column chromatography using 10:1 (v/v) Hexane/EtOAc as the

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mobile phase to afford the product as a red solid (10.6 mg, 24%). The product was recrystallized with CHCl₃/pentane at room temperature to obtained red crystal for X-ray crystallography.

¹H-NMR (600 MHz, CDCl₃): δ = 3.49 (3H, s), 5.31 (1H, d, J = 15 Hz), 5.89 (1H, d, J = 7.2 Hz), 6.80 (1H, td, J = 5.4, 1.2 Hz), 7.11 (1H, td, J = 7.8, 1.8), 7.18 (1H, d, J = 7.8 Hz), 7.22-7.25 (3H, m), 7.26-7.36 (9H, m), 7.50 (2H, d, J = 2.4 Hz), 7.51 (1H, d, J = 1.2 Hz), 7.62 (1H, d, J = 15 Hz); ¹³C-NMR (150 MHz, CDCl₃): 51.2, 120.4, 124.1, 125.2, 125.6, 127.1, 127.4, 127.9, 128.0, 128.4, 129.1, 129.4, 130.5, 134.2, 136.5, 137.7, 137.9, 138.7, 141.4,

143.3, 144.0, 144.3, 145.4, 166.8; HR-MS (DART MS) *m*/*z* calcd. for C₃₂H₂₅O₂ [MH]⁺: 441.18545, found: 441.18548; mp:190-192 °C.

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5. X-ray crystallography

Details of the crystal data and a summary of the intensity data collection parameters for obtained products are listed in below table. Suitable crystals were mounted with mineral oil on a glass fiber and transferred to the goniometer of a Rigaku Saturn CCD diffractometer. Graphite-monochromated Mo K α radiation ($\lambda = 0.71070$ Å) was used. The structures were solved by direct methods with (SIR-97)^[4] and refined by full-matrix least-squares techniques against F^2 (SHELXL-97)^[5]. The intensities were corrected for Lorentz and polarization effects. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed using AFIX instructions. (Table S1)

	2a	2b	2e	2f	2g	2bd	3
formula	$C_{28}H_{18}$	$C_{32}H_{14}F_{12}$	$C_{32}H_{25}O_4$	$C_{36}H_{34}O_8$	$C_{22}H_{18}O_4$	$C_{32}H_{20}F_6O_2$	$C_{32}H_{24}O_2$
fw	354.42	626.43	473.52	594.63	346.36	550.48	440.51
<i>T</i> (K)	103(2)	103(2)	103(2)	103(2)	103(2)	103(2)	103(2)
λ (Å)	0.71070	0.71070	0.71070	0.71070	0.7107	0.71070	0.71070
cryst syst	Monoclinic	Triclinic	Triclinic	Monoclinic	Monoclinic	Monoclinic	Triclinic
space group	C2/c	<i>P</i> -1	<i>P</i> -1	$P2_{1}/c$	$P2_{1}/n$	$P2_{1}/n$	<i>P</i> -1
<i>a</i> (Å)	23.095(10)	8.283(9)	14.463(8)	11.380(3)	11.675(3)	17.550(5)	10.094(6)
b (Å)	48.656(17)	12.769(18)	15.645(9)	8.787(2)	5.0054(8)	7.3020(18)	12.055(7)
c (Å)	18.761(13)	13.17(2)	17.602(10)	15.360(5)	14.303(3)	18.848(5)	19.805(11)
α (deg)	90	80.39(9)	107.771(10)	90	90	90	81.395(20)
β (deg)	122.671(5)	77.65(11)	98.2863(12)	111.613(3)	98.451(12)	94.759(5)	80.70(2)
γ(deg)	90	71.77(10)	105.005(10)	90	90	90	89.36(2)
$V(\text{\AA}^3)$	1774.7(13)	1285(3)	3554(3)	1428.0(7)	826.7(3)	2407.1(11)	2351.0(2)
Ζ	4	2	6	2	2	4	4
D_{calc} , (g / cm ³)	1.327	1.619	1.327	1.383	1.391	1.519	1.244
μ (mm ⁻¹)	0.075	0.155	0.087	0.097	0.095	0.125	0.076
F(000)	744	628	1494	628	364	1128	928
amust size (mms)	$0.20 \times 0.05 \times$	$0.10 \times 0.05 \times$	$0.50 \times 0.05 \times$	$0.10 \times 0.10 \times$	$0.15 \times 0.10 \times$	$0.15 \times 0.05 \times$	$0.20 \times 0.01 \times$
cryst size (mm)	0.05	0.02	0.03	0.05	0.05	0.05	0.01
2θ range, (deg)	3.54-25.00	3.19-25.00	3.01-25.00	3.01-25.00	4.22-25.00	3.02-25.00	3.03-25.00
reflns collected	5444	8312	23908	9082	4986	15510	124676
indep reflns/ $R_{\rm int}$	1550/0.0356	4401/0.0779	12266/0.0704	2502/0.1014	1433/0.0376	4231/0.1178	8052/0.0968
params	164	397	984	203	119	363	615
GOF on F^2	1.124	0.984	1.057	1.159	1.093	1.000	1.021
$R_1, wR_2 [I > 2\sigma(I)]$	0.0645, 0.1576	0.1086, 0.2672	0.1158, 0.2795	0.1478, 0.4388	0.0628, 0.1517	0.0671, 0.1506	0.1334, 0.3020
R_1, wR_2 (all data)	0.0790, 0.1690	0.1760, 0.3426	0.1944, 0.3452	0.1671, 0.4444	0.0751, 0.1671	0.1171, 0.1803	0.2815, 0.4027

Table S1. Crystallographic data and structure refinement details for products.

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Figure S1. ORTEP drawing of 2a with 50% thermal ellipsoids. All hydrogen atoms are omitted for clarity.

Half of the entire structure constitutes an asymmetric unit.



Figure S2. ORTEP drawing of 2b with 50% thermal ellipsoids. All hydrogen atoms are omitted for clarity.



Figure S3. ORTEP drawing of 2e with 50% thermal ellipsoids. Three of four independent molecules and all

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hydrogen atoms are omitted for clarity.



Figure S4. ORTEP drawing of 2f with 50% thermal ellipsoids. All hydrogen atoms are omitted for clarity.

Half of the entire structure constitutes an asymmetric unit.



Figure S5. ORTEP drawing of 2g with 50% thermal ellipsoids. All hydrogen atoms are omitted for clarity.

Half of the entire structure constitutes an asymmetric unit.

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Figure S6. ORTEP drawing of 2bd with 50% thermal ellipsoids. All hydrogen atoms are omitted for clarity.



Figure S7. ORTEP drawing of **3** with 50% thermal ellipsoids. One of two independent molecules and all hydrogen atoms are omitted for clarity.

6. Photophysical study

UV/Vis absorption spectra were recorded on a Shimadzu UV- 3510 spectrometer with a resolution of 0.5

nm.



Figure S8. Normalized absorption spectra of 2a, 2b and 2c in CHCl₃



Figure S9. Normalized absorption spectra of 2e, 2f and 2bd in CHCl₃

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7. Computational study

The Gaussian 09 program^[9] running on a SGI Altix4700 system was used for optimization (B3LYP/6-31G(d)).^[10] All structures were optimized without any symmetry assumptions. Zero-point energy, enthalpy, and Gibbs free energy at 298.15 K and 1 atm were estimated from the gas-phase studies unless otherwise noted. Harmonic vibration frequency calculations at the same level were performed to verify all stationary points as local minima (with no imaginary frequency).



Figure S10. Molecular orbital energies of 2a, 2b, 2c, 2e, 2f, and 2bd (eV).

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compound	E	E+ZPE	Н	G
2a	-1077.82546879	-1077.457127	-1077.436021	-1077.505942
2b	-2425.97440875	-2425.587410	-2425.551417	-2425.661385
2c	-2425.97440266	-2425.587341	-2425.551370	-2425.660812
2e	-1535.91597071	-1535.416890	-1535.385183	-1535.478989
2f	-1993.99276266	-1993.363882	-1993.321130	-1993.440069
2bd	-1980.94735624	-1980.504108	-1980.470304	-1980.572482

Table S2. Uncorrected and thermal-corrected (298K) energies of stationary points (Hartree).^[a]

[a] E: electronic energy; H (=E+ZPE+ $E_{vib}+E_{rot}+E_{trans}+RT$): sum of electronic and themal enthalpies; G (=H-TS): sum of electronic and thermal free energies.

Table S3. Cartesian coordinates of optimized species.

2a											
С	0.384854	0.621489	-0.056160	Н	2.960231	2.080285	0.013672	Н	-0.863423	6.044609	1.604026
С	-0.384854	-0.621489	-0.056160	С	3.069635	-1.830621	0.007862	Н	1.924283	5.366486	-1.601426
С	-1.803278	-0.259655	-0.013833	С	4.254629	-1.077662	0.032460	Н	0.833996	6.918755	0.009351
С	-2.981996	-0.995129	0.014516	Н	5.135374	0.885167	0.070918	С	0.110909	-3.130757	-0.011097
С	-1.848045	1.167462	0.002976	Н	3.113769	-2.915419	-0.011977	С	-0.849629	-3.639409	-0.904183
С	-4.210481	-0.315222	0.044995	Н	5.213511	-1.588883	0.040493	С	0.728333	-4.022875	0.884910
Н	-2.960231	-2.080285	0.013672	С	-0.458581	1.701175	-0.020614	С	-1.185841	-4.991941	-0.897384
С	-3.069635	1.830621	0.007862	С	0.458581	-1.701175	-0.020614	Н	-1.313549	-2.966357	-1.619160
С	-4.254629	1.077662	0.032460	С	-0.110909	3.130757	-0.011097	С	0.384854	-5.373933	0.895245
Н	-5.135374	-0.885167	0.070918	С	-0.728333	4.022875	0.884910	Н	1.458695	-3.645391	1.594139
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Н	-5.213511	1.588883	0.040493	С	-0.384854	5.373933	0.895245	Н	-1.924283	-5.366486	-1.601426
С	1.803278	0.259655	-0.013833	Н	-1.458695	3.645391	1.594139	Н	0.863423	-6.044609	1.604026
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С	1.848045	-1.167462	0.002976	Н	1.313549	2.966357	-1.619160				
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С	-0.582418	0.441044	-0.095351	С	-1.737149	-0.296641	-0.057176	Н	5.166042	-2.425015	-1.620209
С	0.582382	-0.441118	-0.095162	С	1.737102	0.296571	-0.056805	Н	6.087890	0.272006	1.595620
С	0.089655	-1.817846	-0.048457	С	-3.126696	0.183943	-0.043798	С	-7.207676	1.590113	-0.036602
С	0.710607	-3.060106	-0.019966	С	-4.067714	-0.342517	0.859735	С	7.207728	-1.589904	-0.035339
С	-1.335774	-1.728520	-0.028176	С	-3.544745	1.186795	-0.937047	С	2.303932	5.378100	0.096512
С	-0.078899	-4.218070	0.016815	С	-5.377643	0.127538	0.878547	С	-2.303938	-5.378233	0.096261
Н	1.792019	-3.145282	-0.026297	Н	-3.761101	-1.100461	1.573523	F	-2.633588	-5.669187	1.377731
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С	-1.470490	-4.128724	0.008988	Н	-2.837228	1.584614	-1.657967	F	-1.656228	-6.456413	-0.396557
Н	0.394867	-5.193457	0.038256	С	-5.775803	1.125241	-0.014775	F	1.656021	6.456568	-0.395377
Н	-3.195154	-2.827792	-0.041947	Н	-6.088344	-0.272252	1.594170	F	3.465732	5.254660	-0.584805

С	-0.089715	1.817772	-0.048659	Н	-5.165560	2.425493	-1.620780	F	2.634399	5.668354	1.377930
С	-0.710681	3.060022	-0.020469	С	3.126660	-0.183990	-0.043155	F	-7.784387	1.492872	1.181016
С	1.335720	1.728451	-0.027932	С	3.544989	-1.186584	-0.936557	F	-7.310072	2.876988	-0.436081
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Н	3.195089	2.827735	-0.041206	С	5.775820	-1.125130	-0.013729				
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2c											
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С	0.606969	-0.406766	-0.515156	С	1.717364	0.395267	-0.482148	Н	6.876725	-1.103432	-0.432449
С	0.192637	-1.810123	-0.472026	С	-3.132850	0.005322	-0.474353	С	-6.350841	-0.862984	1.363280
С	0.885730	-3.010259	-0.444952	С	-4.043632	-0.583730	0.418205	С	6.350999	0.862604	1.362818
С	-1.235306	-1.802418	-0.459123	С	-3.604111	0.990925	-1.360093	С	0.906739	-5.515614	-0.321617
С	0.158721	-4.213909	-0.419505	С	-5.380529	-0.186347	0.430162	С	-0.906764	5.515719	-0.319916
Н	1.969393	-3.035790	-0.448694	Н	-3.700253	-1.327153	1.128928	F	2.073157	-5.475307	-1.007741
С	-1.941948	-2.998581	-0.460292	С	-4.941372	1.381977	-1.346816	F	1.221488	-5.816436	0.960199
С	-1.235236	-4.209533	-0.436487	Н	-2.915930	1.434424	-2.073330	F	0.188678	-6.553059	-0.804285
Н	-3.026883	-3.008923	-0.486389	С	-5.836286	0.797247	-0.451453	F	-0.188791	6.553286	-0.802454
Н	-1.774687	-5.150317	-0.438486	Н	-5.288327	2.143786	-2.038653	F	-1.221273	5.816216	0.962036
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Н	-1.969444	3.035933	-0.447525	С	4.941193	-1.381661	-1.347671	F	-5.744937	-1.305411	2.486374
С	1.941893	2.998717	-0.459718	Н	2.915667	-1.433914	-2.073967	F	-7.348561	-0.033040	1.736612
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Н	1.774641	5.150448	-0.437288	С	5.836212	-0.797159	-0.452265				
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С	0.413430	0.601886	-0.305075	С	0.378844	-1.722725	-0.268301	0	5.354796	0.880171	-0.170445
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C	-2 980157	1 967284	-0 238089	н	1 485149	2 888290	-1 850405	н	-7 362682	-1 005740	-0 106758
c	-4 205038	1 279407	-0 214084	C II	0.852473	5 834810	-0.250058	л С	6 607404	0.217956	-0 136933
н	-2.982354	3.052954	-0.255748	н	2.199873	5,261690	-1.822466	н	6.762900	-0.399103	-1.032131
н	-5 128699	1 846165	-0 206280	н	1 176190	6 869173	-0 258657	н	7 362682	1 005740	-0 106758
C	1.811707	0.171955	-0.260910	C	-0.041933	-3,133361	-0.261303	н	6.706489	-0.414150	0.755664
C	3.016080	0.852946	-0.233880	C C	-1.042374	-3.584641	-1.145902	с С	0.413430	-7,555505	1.611269
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c	4 220166	0.118384	-0 202350	C C	-1 433785	-4 918220	-1 132773	ч	-0 644270	-7 677544	1 880/180
C	7.220100	0.110504	0.202330	C	1.755205	7.210220	1.154115	11	0.077217	1.011544	1.000-00

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Н	3.070456	1.935834	-0.236444	Н	-1.485149	-2.888290	-1.850405	Н	0.605374	-8.076050	0.663402
С	2.980157	-1.967284	-0.238089	С	0.142205	-5.393403	0.629820	С	-0.413430	7.555505	1.611269
С	4.205038	-1.279407	-0.214084	Н	1.303168	-3.736946	1.326722	Н	-1.036466	7.992271	2.394025
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С	2.838510	-1.307157	0.104595	С	-0.347609	-3.100314	0.025384	Н	1.314269	7.599717	-3.222574
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Н	2.711844	-2.382750	0.125205	С	0.790062	-5.054060	0.902002	0	-1.600481	5.537137	1.893180
С	3.254227	1.494763	0.100050	Н	0.887393	-3.113540	1.807658	0	1.600481	-5.537137	1.893180
С	4.351768	0.610749	0.088125	С	-0.546916	-5.236178	-1.096268	0	-3.404191	-2.849931	0.132673
Н	5.357909	1.004443	0.090186	Н	-1.510706	-3.448853	-1.771501	0	3.404191	2.849931	0.132673
С	-1.764829	0.432637	0.109926	С	0.281870	-5.840894	-0.139782	С	4.706918	3.404478	0.154395
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С	-1.949422	-0.987694	0.083239	0	-5.145798	1.694202	0.065282	Н	5.279880	3.134803	-0.743077
С	-4.137482	0.771081	0.081677	0	5.145798	-1.694202	0.065282	Н	4.565756	4.486701	0.177653
Н	-2.711844	2.382750	0.125205	0	-1.096354	-5.908726	-2.155464	С	-4.706918	-3.404478	0.154395
С	-3.254227	-1.494763	0.100050	0	1.096354	5.908726	-2.155464	Н	-5.266242	-3.092542	1.046669
С	-4.351768	-0.610749	0.088125	С	6.485473	-1.235983	0.053951	Н	-5.279880	-3.134803	-0.743077
Н	-5.357909	-1.004443	0.090186	Н	6.700086	-0.632047	-0.838281	Н	-4.565756	-4.486701	0.177653
С	0.625092	1.649288	0.097947	Н	6.720311	-0.646795	0.950973	С	1.971049	-6.903081	1.859188
С	-0.625092	-1.649288	0.097947	Н	7.109170	-2.132072	0.039669	Н	2.527619	-7.150830	0.944918
С	0.347609	3.100314	0.025384	С	-6.485473	1.235983	0.053951	Н	2.617141	-7.059914	2.725287
С	0.865032	3.880101	-1.016362	Н	-6.720311	0.646795	0.950973	Н	1.097221	-7.564613	1.937840
С	-0.484141	3.692868	0.984741	Н	-7.109170	2.132072	0.039669	С	-1.971049	6.903081	1.859188
С	0.546916	5.236178	-1.096268	Н	-6.700086	0.632047	-0.838281	Н	-2.527619	7.150830	0.944918
Н	1.510706	3.448853	-1.771501	С	-0.790062	-7.280995	-2.319213	Н	-2.617141	7.059914	2.725287
С	-0.790062	5.054060	0.902002	Н	-1.314269	-7.599717	-3.222574	Н	-1.097221	7.564613	1.937840
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С	-0.886909	0.800195	-0.023180	С	-2.015485	0.021204	0.014307	Н	4.940293	-1.850406	-1.659978
С	0.309113	-0.041384	-0.049540	С	1.439561	0.733165	-0.017484	Н	5.811642	0.837331	1.577476
С	-0.132274	-1.436312	-0.016294	С	-3.419250	0.447556	0.031686	С	6.973623	-0.961086	-0.094619

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С	-0.132274	-1.436312	-0.016294	С	-3.419250	0.447556	0.031686	С	6.973623	-0.961086	-0.094619
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С	-1.559889	-1.397994	0.017929	С	-3.868431	1.476389	-0.811020	F	-2.737626	-5.387901	1.358660
С	-0.208660	-3.842929	0.010720	С	-5.687911	0.261179	0.904805	F	-3.542678	-5.002932	-0.620202
Н	1.619108	-2.697216	-0.030381	Н	-4.049504	-0.939354	1.568377	F	-1.695258	-6.138741	-0.394250
С	-2.289916	-2.576659	0.011735	С	-5.197313	1.899610	-0.810099	F	7.119200	-2.240229	-0.507190
С	-1.601554	-3.804217	0.013602	Н	-3.167967	1.937119	-1.501352	F	7.569082	-0.852637	1.113883
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Н	-3.374159	-2.567321	-0.006108	Н	-6.411626	-0.193746	1.573742	0	-7.434987	1.622954	0.143247
С	-0.437850	2.189685	0.033583	Н	-5.503555	2.687794	-1.488378	0	1.876220	5.651633	0.135093

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С	-1.083586	3.415977	0.093118	С	2.844486	0.297902	-0.026779	С	-7.930809	2.655563	-0.696097
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С	-0.335306	4.604302	0.139419	С	3.779658	0.843887	0.871396	Н	-8.991374	2.750374	-0.457262
Н	-2.167289	3.472788	0.108304	С	4.613443	-1.097786	-0.949830	Н	-7.427389	3.611110	-0.499248
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Н	2.818566	3.320298	0.025747	С	5.527452	-0.550011	-0.045383	Н	2.102657	7.651015	0.193771

C–H Activation Route to Dibenzo[a,e]pentalenes: Annulation of Arylacetylenes Promoted by PdCl₂/AgOTf/o-chloranil

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9. ¹H and ¹³C NMR Spectra

¹H NMR spectrum of **2a** (CDCl₃, 600 MHz)



¹H NMR spectrum of 2a (CD₂Cl₂, 600 MHz)



¹³C NMR spectrum of **2a** (CDCl₃, 150 MHz)







¹H NMR spectrum of **2b** (CDCl₃, 600 MHz)



¹³C NMR spectrum of **2b** (CDCl₃, 150 MHz)



£6H

119.4



¹H NMR spectrum of 2c (CD₂Cl₂, 600 MHz)



¹H NMR spectrum of **2e** (CDCl₃, 600 MHz)



¹³C NMR spectrum of **2e** (CDCl₃, 150 MHz)



¹H NMR spectrum of **2f** (CDCl₃, 600 MHz)



¹³C NMR spectrum of **2f** (CDCl₃, 150 MHz)



¹H NMR spectrum of **2g** (CDCl₃, 600 MHz)



¹³C NMR spectrum of **2g** (CDCl₃, 150 MHz)



¹H NMR spectrum of **2bd** (CDCl₃, 600 MHz)



¹³C NMR spectrum of **2bd** (CD₂Cl₂, 150 MHz)



¹H NMR spectrum of **1a-***d* (CDCl₃, 600 MHz)



²H NMR spectrum of **1a-***d* (CHCl₃, 600 MHz)



66°Z -



¹H NMR spectrum of **2a**-d (CD₂Cl₂, 600 MHz)



²H NMR spectrum of **2a-***d* (CH₂Cl₂, 600 MHz)



9272 ------

Z. Z -----





¹H NMR spectrum of **3** (CDCl₃, 600 MHz)



¹³C NMR spectrum of **3** (CDCl₃, 150 MHz)

