# A new and more powerfully activating diamine for practical and scalable enantioselective aldehyde crotylsilylation reactions 

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General Information. All reactions were carried out under an atmosphere of nitrogen in flamedried glassware with magnetic stirring unless otherwise indicated. Degassed solvents were purified by passage through an activated alumina column. Thin-layer chromatography (TLC) was carried out on glass backed silica gel TLC plates ( 250 mm ) from Silicycle; visualization by UV light and/or phosphomolybdic acid (PMA). HPLC analysis was carried out on an Agilent 1200 Series using either a Chiralpak AD-H ( $250 \times 4.5 \mathrm{~mm}$ ID) column or Chiralcel OD-H ( $250 \times 4.5 \mathrm{~mm}$ ID) column. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Bruker AVIII 300 ( 300 MHz ), AVIII 400 ( 400 MHz ), AVIII $500(500 \mathrm{MHz}$ ) or AVIII 500 Ascend $(500 \mathrm{MHz})$ spectrometer and are reported in ppm, relative to residual protonated solvent peak $\left(\mathrm{CDCl}_{3}\right.$, $7.26 \mathrm{ppm} ; \mathrm{C}_{6} \mathrm{D}_{6}, 7.16 \mathrm{ppm}$ ). Data are reported as follows: ( $\mathrm{bs}=$ broad singlet, $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{m}=$ multiplet, $\mathrm{dd}=$ doublet of doublets, $\mathrm{ddd}=$ doublet of doublet of doublets, ddt $=$ doublet of doublet of triplets, $\mathrm{td}=$ triplet of doublets; coupling constant(s) in Hz; integration). Proton decoupled ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker AVIII 400 ( 100 MHz ), AVIII 500 ( 126 MHz ) or AVIII 500 Ascend ( 126 MHz ) spectrometer and are reported in ppm from $\mathrm{CDCl}_{3}$ internal standard ( 77.23 ppm ). ${ }^{29} \mathrm{Si}$ NMR spectra were recorded on a Bruker AVIII $400(79 \mathrm{MHz})$ or AVIII $500(100 \mathrm{MHz})$ and are reported in ppm relative to TMS ( 0.00 ppm ) internal standard. ${ }^{19} \mathrm{~F}$ NMR spectra were recorded on a Bruker AVIII $400(376 \mathrm{MHz})$ and are reported in ppm relative to $\alpha, \alpha, \alpha$-trifluorotoluene ( -63.72 ppm ) internal standard. Infrared spectra were recorded on a Nicolet Avatar 370DTGS FT-IR. Optical rotations were recorded on a Jasco DIP-1000 digital polarimeter. (APCI)-MS was conducted on a JMS-LCmate LCMS (JEOL). Melting points were determined using a Stanford Research Systems DigiMelt apparatus.

## Preparation of diaminophenol 6:



Compound $\mathbf{1 3}$ was prepared using a modified literature procedure. ${ }^{1}$ To a cooled $\left(0^{\circ} \mathrm{C}\right)$ solution of conc. $\mathrm{HCl}(32.4 \mathrm{~mL}, 394 \mathrm{mmol})$ in $\mathrm{MeOH}(126 \mathrm{~mL})$ was added $(R, R)$-diaminocyclohexane ( $45.0 \mathrm{~g}, 394$ $\mathrm{mmol})$. The ice water bath was removed, and after 15 min , water $(42.0 \mathrm{~mL})$ was added. After 30 min , a solution of $\mathrm{Boc}_{2} \mathrm{O}(90.6 \mathrm{~mL}, 394 \mathrm{mmol})$ in $\mathrm{MeOH}(42 \mathrm{~mL})$ was added slowly and the resulting mixture was stirred for 12 h . The mixture was concentrated and the residue was resuspended in $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{~mL})$, and collected by filtration, rinsing with $\mathrm{Et}_{2} \mathrm{O}$ to remove any unprotected diaminocyclohexane. The resulting residue was treated with $3 \mathrm{~N} \mathrm{NaOH}\left(285 \mathrm{~mL}\right.$ ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 3 x 125 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated to give $\mathbf{1 3}$ as a beige solid ( $55.2 \mathrm{~g}, 257 \mathrm{mmol}, 65 \%$ ) that was used without further purification. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.43$ (bs, 1H), 3.23-2.95 (m, 1H), $2.31(\mathrm{td}, J=10.4,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.06-1.88(\mathrm{~m}, 2 \mathrm{H}), 1.78-1.63(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}$, $9 H)$, 1.38-0.99 (m, 6H). The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{1,2}$


Aldehyde 14 was prepared using a modified literature procedure. ${ }^{3}$ To a solution of 2-tert-butyl phenol ( $30.6 \mathrm{~mL}, 200 \mathrm{mmol}$ ) in acetonitrile ( 400 mL ) was added paraformaldehyde ( $40.4 \mathrm{~g}, 1.35 \mathrm{~mol}$ ), $\mathrm{MgCl}_{2}(28.6 \mathrm{~g}, 300 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(104.4 \mathrm{~mL}, 748 \mathrm{mmol})$. The resulting mixture was heated to reflux and stirred for 5 h . The mixture was cooled to room temperature and $5 \% \mathrm{HCl}(200 \mathrm{~mL})$ was added. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 125 \mathrm{~mL})$. The combined organic layers were concentrated and the residue was partitioned between $\mathrm{Et}_{2} \mathrm{O}(250 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(250 \mathrm{~mL})$. The layers were separated and the $\mathrm{Et}_{2} \mathrm{O}$ layer was washed with brine ( $1 \times 100 \mathrm{~mL}$ ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. The resulting oil was vacuum distilled ( $\mathrm{bp} \sim 73^{\circ} \mathrm{C} @ \sim 5 \mathrm{~mm} \mathrm{Hg}$ ) to give aldehyde 14 as a pale yellow oil $\left(20.7 \mathrm{~g}, 116 \mathrm{mmol}, 58 \%\right.$ yield). ${ }^{1} \mathbf{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 11.78(\mathrm{~s}, 1 \mathrm{H}), 9.88(\mathrm{~s}, 1 \mathrm{H}), 7.53(\mathrm{dd}, J=$ $7.7,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{dd}, J=7.7,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 9 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{3}$
(1) D. W. Lee, H. Ha and W. K. Lee, Synth. Commun., 2007, 37, 737-742.
(2) X. Zhang, T. J. Emge and K. C. Hultzsch, Angew. Chem. Int. Ed., 2012, 51, 394-398.
(3) N. Gisch, J. Balzarini and C. Meier, J. Med. Chem., 2007, 50, 1658-1667.


To a solution of compound $\mathbf{1 3}(24.9 \mathrm{~g}, 116 \mathrm{mmol})$ in $\mathrm{EtOH}(1.1 \mathrm{~L})$, was added aldehyde 14 (20.7 $\mathrm{g}, 116 \mathrm{mmol}$ ). The mixture was heated to reflux and stirred for 3 h . The mixture was cooled to room temperature and concentrated. The residue was recrystallized from minimal boiling EtOH to give imine 15 as long yellow crystals ( $39.2 \mathrm{~g}, 105 \mathrm{mmol}, 90 \%$ yield). m.p. $153-155^{\circ} \mathrm{C} ;{ }^{1} \mathbf{H} \mathbf{N M R}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) $\delta 13.73(\mathrm{bs}, 1 \mathrm{H}), 8.35(\mathrm{~s}, 1 \mathrm{H}), 7.32(\mathrm{dd}, J=7.8,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{dd}, J=7.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{t}, J=$ $7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.37(\mathrm{bs}, 1 \mathrm{H}), 3.73-3.40(\mathrm{~m}, 1 \mathrm{H}), 3.04(\mathrm{bs}, 1 \mathrm{H}), 2.18-2.02(\mathrm{~m}, 1 \mathrm{H}), 1.98-1.87(\mathrm{~m}, 1 \mathrm{H}), 1.87-$ $1.64(\mathrm{~m}, 3 \mathrm{H}), 1.46-1.30(\mathrm{~m}, 3 \mathrm{H}) 1.45(\mathrm{~s}, 9 \mathrm{H}), 1.31(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 165.02$, $160.64,155.39,137.55,129.85,129.44,118.87,117.82,79.49,77.43,72.87,54.55,35.02,33.64,32.00$, 29.57, 28.37, 25.06, 24.31; IR (thin film, $\mathrm{cm}^{-1}$ ) 3437, 3338, 2936, 2859, 2243, 1688, 1630, 1505, 1436, 1390, 1365, 1267, 733; HRMS (FAB+): calcd for $\mathrm{C}_{22} \mathrm{H}_{35} \mathrm{O}_{3} \mathrm{~N}_{2}[\mathrm{M}+\mathrm{H}]^{+}: 375.2648$, found 375.2654.


A 2-L roundbottom flask equipped with an addition funnel was charged with $\mathrm{LiAlH}_{4}(13.0 \mathrm{~g}, 343$ $\mathrm{mmol})$ and THF $(600 \mathrm{~mL})$ and the resulting mixture was cooled to $0^{\circ} \mathrm{C}$. A solution of $\mathbf{1 5}(42.8 \mathrm{~g}, 114$ mmol ) in THF ( 300 mL ) was added slowly via the addition funnel, with a THF rinse ( 100 mL ). The mixture was warmed to room temperature and stirred for 2 h . The addition funnel was replaced with a reflux condenser and the mixture was heated to reflux for 12 h . The mixture was cooled to $0{ }^{\circ} \mathrm{C}$, and the reaction was quenched by the CAREFUL and SLOW addition of water ( 100 mL ). The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}$ ( $5 \times 100 \mathrm{~mL}$ ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The resulting beige solid was purified by recrystallization from minimal boiling hexanes to give diaminophenol $(R, R)-\mathbf{6}$ as white crystals ( $30.2 \mathrm{~g}, 104 \mathrm{mmol}, 91 \%$ yield). m.p. $114-116{ }^{\circ} \mathrm{C} ;{ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.17(\mathrm{dd}, J=7.9,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.92-6.80(\mathrm{~m}, 1 \mathrm{H}), 6.70(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.01(\mathrm{~d}, J$ $=13.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{~d}, J=13.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.39(\mathrm{~s}, 3 \mathrm{H}), 2.24-2.08(\mathrm{~m}, 4 \mathrm{H}), 1.82-1.64(\mathrm{~m}, 2 \mathrm{H}), 1.42(\mathrm{~s}$, 9H), 1.31-1.11 (m, 3H), 1.03-0.85 (m, 1H); ${ }^{13} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 157.49,136.88,126.27$, 125.77, 124.47, 118.16, 62.46, 62.22, 51.02, 34.85, 33.59, 31.24, 31.16, 29.74, 25.33, 24.83; IR (thin
film, $\mathrm{cm}^{-1}$ ) $3299,3217,2927,2854,2797,2641,1590,1430,1352,1240,1084,747$; HRMS (FAB+): calcd for $\mathrm{C}_{18} \mathrm{H}_{31} \mathrm{ON}_{2}[\mathrm{M}+\mathrm{H}]^{+}: 291.2436$, found 291.2443.

## Preparation of cis-crotyltrichlorosilane 10 and trans-crotyltrichlorosilane 11:


cis-Crotyltrichlorosilane 10 was prepared using a modified literature procedure. ${ }^{4,5}$ To a cooled ($78{ }^{\circ} \mathrm{C}$ ) solution of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(0.88 \mathrm{~g}, 0.76 \mathrm{mmol})$ in THF ( 400 mL ) was added 1,3-butadiene ( $50 \mathrm{~mL}, 573$ mmol , condensed into a graduated cylinder cooled to $-78^{\circ} \mathrm{C}$ ) followed by trichlorosilane ( $38.6 \mathrm{~mL}, 382$ $\mathrm{mmol})$. After 15 min , the cooling bath was removed and the mixture was allowed to warm to room temperature. After 17 h , an aliquot was removed and ${ }^{1} \mathrm{H}$ NMR analysis showed complete consumption of the trichlorosilane. The reaction flask was fitted with a short-path distillation head and the THF was removed by distillation. The residue was transferred to a distillation-head equipped, $250-\mathrm{mL}$ roundbottom flask with a THF rinse. The THF was removed by distillation and the residue was distilled under reduced pressure ( $\mathrm{bp} \sim 60{ }^{\circ} \mathrm{C} @ \sim 30 \mathrm{~mm} \mathrm{Hg}$ ) to give 10 as a clear and colorless liquid ( $56.2 \mathrm{~g}, 297 \mathrm{mmol}, 78 \%$ yield). ${ }^{1} \mathrm{H}$ NMR spectroscopic analysis revealed that the cis to trans ratio was $\geq 99: 1 .{ }^{1} \mathbf{H} \mathbf{N M R}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 5.84-5.62(\mathrm{~m}, 1 \mathrm{H}), 5.54-5.31(\mathrm{~m}, 1 \mathrm{H}), 2.50-2.24(\mathrm{~m}, 2 \mathrm{H}), 1.76-1.58(\mathrm{~m}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{5}$

trans-Crotyltrichlorosilane 11 was prepared according to a modified literature procedure. ${ }^{5,6}$ transCrotylalcohol was purchased from Sigma Aldrich as a 19:1 (E:Z) mixture, which was confirmed by ${ }^{1} \mathrm{H}$ NMR spectroscopic analysis. To a cooled $\left(0^{\circ} \mathrm{C}\right)$ solution of trans-crotylalcohol ( $121 \mathrm{~mL}, 1.41 \mathrm{~mol}$ ) in hexachloroacetone ( 470 mL ) was added $\mathrm{PPh}_{3}(386 \mathrm{~g}, 1.47 \mathrm{~mol})$ portion-wise over 3 h . The ice/water bath was allowed to melt and warm to room temperature slowly, and after 12 h the reaction flask was fitted with a short-path distillation head. Distillation ( $\mathrm{bp} \sim 85^{\circ} \mathrm{C} @ 760 \mathrm{~mm} \mathrm{Hg}$ ) gave trans-crotylchloride as a clear and colorless liquid ( $83.2 \mathrm{~g}, 919 \mathrm{mmol}, 65 \%$ yield). ${ }^{1} \mathbf{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.88-5.73(\mathrm{~m}$, $1 \mathrm{H}), 5.69-5.52(\mathrm{~m}, 1 \mathrm{H}), 4.06-3.98(\mathrm{~m}, 2 \mathrm{H}), 1.75-1.68(\mathrm{~m}, 3 \mathrm{H})$.
(4) J. Tsuji, M. Hara and K. Ohno, Tetrahedron, 1974, 30, 2143-2146.
(5) K. Iseki, Y. Kuroki, M. Takahashi, S. Kishimoto and Y. Kobayashi, Tetrahedron, 1997, 53, 3513-3526.
(6) M. Kira, T. Hino and H. Sakurai, Tetrahedron Lett., 1989, 30, 1099-1102.

A 2 L round bottom flask equipped with an addition funnel was charged with $\mathrm{Et}_{3} \mathrm{~N}(104 \mathrm{~mL}, 1.03$ $\mathrm{mol})$ and $\mathrm{Et}_{2} \mathrm{O}(450 \mathrm{~mL})$. The solution was cooled to $0^{\circ} \mathrm{C}$ and copper(I) chloride ( $4.2 \mathrm{~g}, 43 \mathrm{mmol}$ ) was added. A solution of trichlorosilane ( $104 \mathrm{~mL}, 1.03 \mathrm{~mol}$ ) and trans-crotylchloride ( $77.4 \mathrm{~g}, 855 \mathrm{mmol}$ ) in $\mathrm{Et}_{2} \mathrm{O}(150 \mathrm{~mL})$ was prepared and transferred to the addition funnel. This solution was then added very slowly (over the course of 1 h ) to the reaction mixture, so as to maintain a reaction mixture temperature near $0{ }^{\circ} \mathrm{C}$. After the addition was complete, the mixture was allowed to warm to room temperature and after 2 h , analysis of an aliquot by ${ }^{1} \mathrm{H}$ NMR spectroscopy showed complete consumption of the transcrotylchloride. The solution was transferred to a 2 L round bottom flask by cannulation through a glass microfiber filter (Grade GF/D) equipped teflon tube (3/16 i.d.) to filter the $\mathrm{Et}_{3} \mathrm{~N} \cdot \mathrm{HCl}$ salts. The flask was fitted with a distillation head and the $\mathrm{Et}_{2} \mathrm{O}$ and $\mathrm{Et}_{3} \mathrm{~N}$ were removed by distillation. The residue was transferred to a 250 mL pear-shaped flask by cannula with an $\mathrm{Et}_{2} \mathrm{O}$ rinse. The flask was fitted with a short-path distillation head and the excess $\mathrm{Et}_{2} \mathrm{O}$ was removed by distillation and the residue was distilled (bp $\sim 140-145{ }^{\circ} \mathrm{C} @ 760 \mathrm{~mm} \mathrm{Hg}$ ) to give trans-crotyltrichlorosilane 11 as a clear and colorless liquid ( $74.4 \mathrm{~g}, 392 \mathrm{mmol}, 46 \%$ yield). ${ }^{1} \mathrm{H}$ NMR spectroscopic analysis revealed that the trans to cis ratio was 95:5. ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 5.67-5.53 ( $\mathrm{m}, 1 \mathrm{H}$ ), 5.48-5.32 (m, 1H), 2.31-2.20 (m, 2H), 1.76-1.69 $(\mathrm{m}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{5}$

## Preparation and characterization of silanes $(S, S)-3$ and $(S, S)-8$ :



To a cooled $\left(0^{\circ} \mathrm{C}\right)$ solution of $(S, S)-\mathbf{1}(1.00 \mathrm{~g}, 1.8 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(7.0 \mathrm{~mL})$ was added DBU $(270 \mu \mathrm{~L}, 1.8 \mathrm{mmol})$ followed by freshly distilled phenol ( $170 \mathrm{mg}, 1.8 \mathrm{mmol}$ ). The ice/water bath was removed and after 1.5 h , the flask was fitted with a distillation head and the volatiles were removed by distillation. The residue was treated with pentane $(10 \mathrm{~mL})$ and the mixture was stirred vigorously for 10 $\min$ to ensure complete precipitation of the $\mathrm{DBU} \cdot \mathrm{HCl}$ salts. The mixture was then filtered through an airfree filter frit, and the filtrate was concentrated to give $(S, S)-\mathbf{3}$ as a thick oil, which was analyzed without further purification. ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.31-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.25-7.19(\mathrm{~m}$, 4H), 7.08-6.97 (m, 3H), 6.90-6.84 (m, 2H), 5.04-4.84 (m, 2H), 4.09 (dd, $J=15.6,3.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.80(\mathrm{dd}, J$ $=18.8,15.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.87-2.64(\mathrm{~m}, 2 \mathrm{H}), 1.76-1.48(\mathrm{~m}, 6 \mathrm{H}), 1.22-1.03(\mathrm{~m}, 2 \mathrm{H}), 1.02-0.82(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR (101 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 154.62,141.40,141.29,132.80,131.27,131.23,129.62,129.35,122.00$,
$120.46,120.28,120.15,115.37,66.68,65.99,48.12,48.05,31.19,30.94,24.91,20.27 ;{ }^{29}$ Si NMR (79 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$-25.21.


To a cooled $\left(0{ }^{\circ} \mathrm{C}\right)$ solution of $(S, S)-6(1.00 \mathrm{~g}, 3.4 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(11 \mathrm{~mL})$ was added DBU $(1.54 \mathrm{~mL}, 10.3 \mathrm{mmol})$. Allyltrichlorosilane $(0.55 \mathrm{~mL}, 3.8 \mathrm{mmol})$ was then added slowly. The reaction mixture was allowed to warm to room temperature and after 1 h the flask was fitted with a distillation head and the volatiles were removed by distillation. The residue was treated with $\mathrm{Et}_{2} \mathrm{O}(15 \mathrm{~mL})$, and the resulting mixture was stirred vigorously for 10 min to ensure complete precipitation of the $\mathrm{DBU} \cdot \mathrm{HCl}$ salts. The mixture was then filtered through an air-free filter frit, and the filtrate was concentrated to give $(S, S)-\mathbf{8}$ as a thick oil, which was analyzed without further purification. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.23-7.18(\mathrm{~m}, 1 \mathrm{H}), 6.90-6.84(\mathrm{~m}, 1 \mathrm{H}), 6.82(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.95(\mathrm{ddt}, J=16.9,10.1,7.8 \mathrm{~Hz}, 1 \mathrm{H})$, 5.16-5.07 (m, 1H), 5.04-4.95 (m, 1H), $4.23(\mathrm{~d}, J=15.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.69(\mathrm{~d}, J=15.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.54-2.42(\mathrm{~m}$, $1 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 2.13(\mathrm{ddd}, J=10.5,9.0,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.06-1.90(\mathrm{~m}, 4 \mathrm{H}), 1.74-1.63(\mathrm{~m}, 2 \mathrm{H}), 1.39(\mathrm{~s}$, 9H), 1.26-1.15 (m, 1H), 1.14-1.03 (m, 2H), 0.86-0.73 (m, 1H); ${ }^{13} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 152.93$, 139.64, 135.86, 132.00, 130.16, 129.31, 127.17, 125.92, 121.56, 120.54, 115.69, 64.94, 62.57, 54.65, 48.97, 47.52, 38.13, 34.85, 32.33, 30.26, 30.13, 29.98, 29.94, 29.86, 29.60, 29.31, 24.96, 24.75, 24.29, 19.81, 18.88; ${ }^{29} \mathbf{S i}$ NMR ( $99 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-15.38$; HRMS (FAB+): calcd for $\mathrm{C}_{21} \mathrm{H}_{33} \mathrm{ON}_{2} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}$: 357.2362 , found 357.2351 .

The stereostructure of $\mathbf{8}$ was proved by NMR spectroscopic analysis. COSY, HSQC, and HMBC experiments allowed the unambiguous assignment of all of the relevant protons in the ${ }^{1} \mathrm{H}$ NMR spectrum, and a NOESY experiment revealed the illustrated interactions which are consistent only with the silicon stereochemistry shown in structure (S,S)-8 (Supp. Fig. 1). Copies of the COSY, HSQC, HMBC, and NOESY spectra are provided below.


Supplementary Figure 1. NOESY data confims the stereochemical assignment at silicon in ( $S, S$ )-8.

## General Procedure for the one-pot allylation or crotylation of aldehydes with $(\boldsymbol{R}, \boldsymbol{R})-6$ :



To a cooled $\left(0^{\circ} \mathrm{C}\right)$ solution of $(R, R)-6(1.45 \mathrm{~g}, 5.0 \mathrm{mmol}, 1.1$ equiv. $)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(16.5 \mathrm{~mL})$ is added DBU ( $2.24 \mathrm{~mL}, 15.0 \mathrm{mmol}, 3.3$ equiv). The allyltrichlorosilane $\mathbf{9}$ or cis-crotyltrichlorosilane $\mathbf{1 0}$ or transcrotyltrichlorosilane $\mathbf{1 1}$ ( $\mathbf{9}: 0.80 \mathrm{~mL}, \mathbf{1 0}$ or 11: $0.84 \mathrm{~mL}, 5.5 \mathrm{mmol}, 1.2$ equiv) is then added slowly. The ice/water bath is removed and after 1 h the mixture is recooled to $0^{\circ} \mathrm{C}$. The aldehyde ( $4.5 \mathrm{mmol}, 1.0$ equiv) is added and the solution is maintained at $0^{\circ} \mathrm{C}$ for 1 h . The progress of the reaction may be monitored by TLC, 1 h is generally sufficient for full conversion.
(A) General procedure for the acidic workup with recovery of 6: The mixture is concentrated and the residue is suspended in $\mathrm{Et}_{2} \mathrm{O}(27.0 \mathrm{~mL})$. The mixture is stirred vigorously for 20 min to ensure complete precipitation of the $\mathrm{DBU} \cdot \mathrm{HCl}$ salts. The mixture is then filtered through a frit, and the filtrate is treated with $n-\mathrm{Bu}_{4} \mathrm{NF}(5.0 \mathrm{~mL}, 1 \mathrm{M}$ in THF, 5.0 mmol , 1.1 equiv). After $2 \mathrm{~h}, 1 \mathrm{M}$ aqueous $\mathrm{HCl}(25.0 \mathrm{~mL}$, 25.0 mmol , 5.5 equiv) is added and the mixture is transferred to a separatory funnel. The layers are separated and the aqueous layer is extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50.0 \mathrm{~mL})$. The combined organic layers are washed with $\mathrm{H}_{2} \mathrm{O}(2 \times 25.0 \mathrm{~mL})$ and saturated aqueous $\mathrm{NaHCO}_{3}(1 \times 25.0 \mathrm{~mL})$, dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue is purified by chromatography on silica gel to provide the allylation or crotylation product, and we have found that a simple filtration through a pad of silica is generally sufficient.

Recovery of 6: The combined aqueous layers from above are treated with 1 M aqueous NaOH ( 50 $\mathrm{mL}, 50 \mathrm{mmol})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 50.0 \mathrm{~mL})$. The combined organic layers are washed with water ( $2 \times 25.0 \mathrm{~mL}$ ), dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue is dissolved in minimal hot 9:1 MeOH: $\mathrm{H}_{2} \mathrm{O}$ (during the dissolution process the temperature should not be allowed to exceed $80{ }^{\circ} \mathrm{C}$, as the ligand may start to undergo decomposition at higher temperatures). The hot saturated solution is allowed to cool to room temperature, and distilled water ( 10.0 mL ) was added to ensure complete crystallization of 6 . The white solid is collected by filtration through a frit with a cold $1: 1 \mathrm{MeOH}: \mathrm{H}_{2} \mathrm{O}$ rinse and then dried in vacuo (overnight, with gentle warming with an oil bath set to $40{ }^{\circ} \mathrm{C}$ ) to give recovered diaminophenol 6 .

## (B) General procedure for the simplified workup for acid-sensitive substrates without

 recovery of 6: The reaction mixture is treated with $n$ - $\mathrm{Bu}_{4} \mathrm{NF}(4.5 \mathrm{~mL}, 1 \mathrm{M}$ in THF, $4.5 \mathrm{mmol}, 1.0$ equiv) and the ice/water bath is removed. After 30 min the mixture is concentrated and the residue is purified by chromatography on silica gel to provide the allylation or crotylation product.
## Aldehydes employed in this study:

The six aldehydes employed in Table 1 are all known compounds and were prepared according to the literature procedures. ${ }^{7,8,9,10,11,12}$


## Data for the products in Table 1:



Workup procedure A was used. Product 16 was isolated as a colorless oil ( $83 \%$ yield, $>99 \% \mathrm{ee}$ ). ${ }^{1} \mathbf{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.42-7.27(\mathrm{~m}, 5 \mathrm{H}), 5.83(\mathrm{ddt}, J=17.2,10.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.20-5.04(\mathrm{~m}$, $2 \mathrm{H}), 4.56(\mathrm{~s}, 2 \mathrm{H}), 3.92-3.85(\mathrm{~m}, J=10.0,6.7,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.52(\mathrm{dd}, J=9.5,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{dd}, J=$ 9.5, 7.4 Hz, 1H), $2.31(\mathrm{~d}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.29-2.25(\mathrm{~m}, 2 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{13}$ The enantiomeric excess was determined by chiral HPLC analysis: OD-H column, 98.5:1.5 hexanes: $i \mathrm{PrOH}, 1 \mathrm{~mL} / \mathrm{min}, 254 \mathrm{~nm}$. The absolute configuration was determined by optical rotation and comparison to the literature value: $[\alpha]_{\mathrm{D}}{ }^{22}+2.3^{\circ}\left(\mathrm{CHCl}_{3}, c\right.$ 2.0 $)$; lit: $[\alpha]_{\mathrm{D}}^{23}+2.0^{\circ}\left(\mathrm{CHCl}_{3}\right.$, c 2.3$)$ for the $(S)$ enantiomer with $94 \%$ ee. ${ }^{13}$

[^0]



Workup procedure B was used and product 17 was isolated as a colorless oil ( $81 \%$ yield, $98 \% \mathrm{ee}$ ). ${ }^{1} \mathbf{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.84(\mathrm{ddt}, J=17.2,10.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.17-5.03(\mathrm{~m}, 2 \mathrm{H}), 3.74-3.67(\mathrm{~m}$, 1 H ), 3.63 (dd, $J=9.9,3.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.46 (dd, $J=9.9,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.39(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.30-2.19$ (m, $2 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.08(\mathrm{~s}, 6 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{14}$ The enantiomeric excess was determined by ${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) analysis of the Mosher ester. ${ }^{15}$ The absolute configuration was determined by optical rotation and comparison to the literature value: $[\alpha]_{\mathrm{D}}{ }^{22}+2.4^{\circ}\left(\mathrm{CHCl}_{3}, c 2.0\right)$; lit: $[\alpha]_{\mathrm{D}}{ }^{20}+1.7^{\circ}\left(\mathrm{CHCl}_{3}, c 0.24\right)$ for the $(S)$ enantiomer with $59 \%$ ee. ${ }^{14}$

(14) W. R. Roush, L. K. Hoong, M. A. J. Palmer, J. A. Straub and A. D. Palkowitz, J. Org. Chem. 1990, 55, 4117-4126. (15) (a) J. A. Dale, D. L. Dull and H. S. Mosher, J. Org. Chem., 1969, 34, 2543-2549; (b) T. R. Hoye, C. S. Jeffrey and F. Shao, Nature protocols, 2007, 2, 2451-2458.


Workup procedure $\mathbf{B}$ was used to prepare product $\mathbf{1 2}$ (for convenience on small scale, not because it is acid sensitive), which was isolated as a colorless oil (94\% yield, 96\% ee, 98:2 dr). ${ }^{1}$ H NMR ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43-7.27(\mathrm{~m}, 5 \mathrm{H}), 5.74(\mathrm{ddd}, J=17.2,10.3,7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.14-4.98(\mathrm{~m}, 2 \mathrm{H}), 4.55(\mathrm{~s}$, 2 H ), 3.65-3.62 (m, 1H), $3.56(\mathrm{dd}, J=9.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.40(\mathrm{dd}, J=9.5,7.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.40-2.27(\mathrm{~m}, 2 \mathrm{H})$, $1.08(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{16}$ The enantiomeric excess and diastereomer ratio were determined by chiral HPLC analysis: OD-H column, $98: 2$ hexanes: $i \operatorname{PrOH}, 1 \mathrm{~mL} / \mathrm{min}, 254 \mathrm{~nm}$. The assay was developed using a $3: 1$ anti:syn mixture of racemic diastereomers. The absolute configuration was determined by optical rotation and comparison to the literature value: $[\alpha]_{\mathrm{D}}{ }^{22}+23.8^{\circ}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, c 2.0\right)$; lit: $[\alpha]_{\mathrm{D}}{ }^{20}+24.6^{\circ}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, c 1.8\right)$ for the $(S, R)$ enantiomer with $96 \%$ ee. ${ }^{16}$

Workup procedure $\mathbf{A}$ was used to prepare product $\mathbf{1 8}$, which was isolated as a colorless oil $(88 \%$ yield, $97 \%$ ee, $94: 6 \mathrm{dr}) .{ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.44-7.27(\mathrm{~m}, 5 \mathrm{H}), 5.90-5.74(\mathrm{~m}, 1 \mathrm{H}), 5.13-5.03$ (m, 2H), 4.59-4.50 (s, 2H), 3.71-3.63 (m, 1H), $3.55(\mathrm{dd}, J=9.6,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{dd}, J=9.6,7.6 \mathrm{~Hz}$, $1 \mathrm{H}), 2.43-2.29(\mathrm{~m}, 1 \mathrm{H}), 2.25(\mathrm{~d}, J=3.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.04(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{17}$ The enantiomeric excess and diastereomer ratio were determined by chiral HPLC analysis: OD-H column, $98: 2$ hexanes: $i \mathrm{PrOH}, 1 \mathrm{~mL} / \mathrm{min}, 254 \mathrm{~nm}$. The assay was developed using a $\sim 3: 1$ anti:syn mixture of racemic diastereomers. The absolute configuration was determined by optical rotation and comparison to the literature value: $[\alpha]_{\mathrm{D}}{ }^{22}-7.1^{\circ}\left(\mathrm{CHCl}_{3}, c 2.0\right)$; lit: $[\alpha]_{\mathrm{D}}{ }^{20}-5.6^{\circ}\left(\mathrm{CHCl}_{3}, c 1.2\right)$ for the $(S, S)$ enantiomer with $99 \%$ ee. ${ }^{16}$
(16) B. M. Hackman, P. J. Lombardi and J. L. Leighton, Org. Lett., 2004, 6, 4375-4377.
(17) D. K. Mohapatra, P. P. Das, M. R. Pattanayak and J. S. Yadav, Chemistry (Weinheim an der Bergstrasse, Germany), 2010, 16, 2072-2078.


| $\begin{gathered} \text { Peak } \\ \# \end{gathered}$ | $\begin{gathered} \text { RetTime } \\ {[\mathrm{min}]} \end{gathered}$ |  | Type | $\begin{gathered} \text { Area } \\ {\left[m A U^{\star} s\right]} \end{gathered}$ | Height <br> [mAU] | $\begin{gathered} \text { Area } \\ \% \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 13.160 | 1 | MF | 559.68024 | 32.48430 | 12.4638 |
| 2 | 13.923 | 1 | FM | 1688.06042 | 81.23139 | 37.5922 |
| 3 | 15.872 | 1 | MM | 557.01508 | 24.84939 | 12.4044 |
| 4 | 22.323 | 1 | MM | 1685.69287 | 46.69843 | 37.5395 |







Workup procedure $\mathbf{A}$ was used and product 19 was isolated as a pale yellow oil ( $92 \%$ yield, $93 \%$ ee, $>99: 1 \mathrm{dr}$ ). ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26-7.14(\mathrm{~m}, 2 \mathrm{H}), 6.94-6.78(\mathrm{~m}, 2 \mathrm{H}), 5.78$ (ddd, $J=17.2$, $10.4,7.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.13-4.98(\mathrm{~m}, 2 \mathrm{H}), 4.45(\mathrm{~s}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.70(\mathrm{ddd}, J=9.2,5.7,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.66$ (m, 1H), 3.61 (ddd, $J=9.3,8.2,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.87(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.30-2.21(\mathrm{~m}, 1 \mathrm{H}), 1.81-1.64$ (m, $2 \mathrm{H}), 1.05(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{16}$ The enantiomeric excess and diastereomeric ratio were determined by chiral HPLC analysis of the 3,5-dinitrobenzoate ester derivative: AD-H column, 98:1:1 hexanes:EtOH:MeOH, $1 \mathrm{~mL} / \mathrm{min}, 254$ nm . The assay was developed using a $\sim 3: 1$ anti:syn mixture of racemic diastereomers. We previously determined that the $(R, R)$-diastereomer elutes third. ${ }^{16}$

## Electronic Supplementary Material (ESI) for Chemical Science

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Workup procedure $\mathbf{A}$ was used and product 20 was isolated as a pale yellow oil ( $82 \%$ yield, $98: 2$ dr). ${ }^{1} \mathbf{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 7.26-7.18 (m, 2H), 7.00-6.79 (m, 2H), 6.03-5.79 (m, 1H), 5.23-5.01 $(\mathrm{m}, 2 \mathrm{H}), 4.45(\mathrm{~s}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.65-3.52(\mathrm{~m}, 2 \mathrm{H}), 3.46(\mathrm{dd}, J=9.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.24(\mathrm{bs}, 1 \mathrm{H}), 2.43-$ $2.27(\mathrm{~m}, 1 \mathrm{H}), 2.26-2.10(\mathrm{~m}, 1 \mathrm{H}), 1.96-1.79(\mathrm{~m}, 1 \mathrm{H}), 0.91(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{18}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400 \mathrm{MHz}\right)$.

(18) T. M. Trygstad, Y. Pang and C. J. Forsyth, J. Org. Chem., 2009, 74, 910-913.


Workup procedure $\mathbf{A}$ was used and product 21 was isolated as a pale yellow oil ( $82 \%$ yield, $92: 8$ dr. ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26-7.20(\mathrm{~m}, 2 \mathrm{H}), 6.95-6.79(\mathrm{~m}, 2 \mathrm{H}), 5.85(\mathrm{ddd}, J=17.5,10.5,7.3$ $\mathrm{Hz}, 1 \mathrm{H}), 5.10-4.97(\mathrm{~m}, 2 \mathrm{H}), 4.44(\mathrm{~s}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.63(\mathrm{dd}, J=9.1,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.46(\mathrm{dd}, J=9.1$, $6.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.22(\mathrm{bs}, 1 \mathrm{H}), 2.39-2.24(\mathrm{~m}, 1 \mathrm{H}), 1.99-1.85(\mathrm{~m}, 1 \mathrm{H}), 1.04(\mathrm{~d}, J=6.7$ $\mathrm{Hz}, 3 \mathrm{H}), 0.95(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{19}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400 \mathrm{MHz}\right)$.

(19) K. Tanaka, Y. Fujimori, Y. Saikawa and M. Nakata, J. Org. Chem., 2008, 73, 6292-6298.


Workup procedure $\mathbf{A}$ was used and product 22 was isolated as a pale yellow oil ( $80 \%$ yield, 93:7 dr). ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26-7.22(\mathrm{~m}, 2 \mathrm{H}), 6.90-6.85(\mathrm{~m}, 2 \mathrm{H}), 5.90(\mathrm{ddd}, J=16.6,11.0,8.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.10-5.00(\mathrm{~m}, 2 \mathrm{H}), 4.44(\mathrm{~s}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.55(\mathrm{dd}, J=9.2,4.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.48(\mathrm{dd}, J=9.2$, $7.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.39(\mathrm{bs}, 1 \mathrm{H}), 3.35(\mathrm{dd}, J=8.0,3.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.41-2.29(\mathrm{~m}, 1 \mathrm{H}), 1.97-1.84(\mathrm{~m}, 1 \mathrm{H}), 1.10(\mathrm{~d}, J$ $=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{20}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400\right.$ MHz ).

(20) S. R. Chemler and W. R. Roush, J. Org. Chem., 2003, 68, 1319-1333.


Workup procedure $\mathbf{A}$ was used and product 23 was isolated as a pale yellow oil ( $80 \%$ yield, $98: 2$ dr). ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26-7.22(\mathrm{~m}, 2 \mathrm{H}), 6.90-6.85(\mathrm{~m}, 2 \mathrm{H}), 5.84(\mathrm{ddt}, J=17.2,10.1,7.0$ $\mathrm{Hz}, 1 \mathrm{H}), 5.16-5.02(\mathrm{~m}, 2 \mathrm{H}), 4.44(\mathrm{~s}, 2 \mathrm{H}), 3.84-3.79(\mathrm{~m}, 1 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 3.49(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.52$ $(\mathrm{d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.29-2.15(\mathrm{~m}, 2 \mathrm{H}), 1.94-1.81(\mathrm{~m}, 1 \mathrm{H}), 0.95(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{21}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400 \mathrm{MHz}\right)$.

(21) K. C. Nicolaou, A. P. Patron, K. Ajito, P. K. Richter, H. Khatuya, P. Bertinato, R. A. Miller and M. J. Tomaszewski, Chem. Eur.J., 1996, 2, 847-868.


Workup procedure $\mathbf{A}$ was used and product 24 was isolated as a pale yellow oil ( $80 \%$ yield, $94: 6$ dr). ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26-7.20(\mathrm{~m}, 2 \mathrm{H}), 6.90-6.85(\mathrm{~m}, 2 \mathrm{H}), 5.63(\mathrm{ddd}, J=17.2,10.3,8.7$ $\mathrm{Hz}, 1 \mathrm{H}), 5.10-4.92(\mathrm{~m}, 2 \mathrm{H}), 4.49-4.39(\mathrm{~m}, 2 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 3.57-3.46(\mathrm{~m}, 3 \mathrm{H}), 2.62(\mathrm{bs}, 1 \mathrm{H}), 2.29$ (ddt, $J$ $=15.5,8.8,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.92(\mathrm{ddt}, J=7.0,4.8,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.09(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 0.95(\mathrm{~d}, J=7.0 \mathrm{~Hz}$, $3 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{22}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400 \mathrm{MHz}\right)$.

(22) E. de Lemos, F.-H. Porée, A. Bourin, J. Barbion, E. Agouridas, M.-I. Lannou, A. Commerçon, J.-F. Betzer, A. Pancrazi and J. Ardisson, Chemistry (Weinheim an der Bergstrasse, Germany), 2008, 14, 11092-11112.


Workup procedure $\mathbf{A}$ was used and product 25 was isolated as a pale yellow oil ( $80 \%$ yield, $95: 5$ dr). ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26-7.23(\mathrm{~m}, 2 \mathrm{H}), 6.91-6.84(\mathrm{~m}, 2 \mathrm{H}), 5.79(\mathrm{ddd}, J=17.3,10.3,8.4$ Hz, 1H), 5.14-5.05 (m, 2H), 4.45 (s, 2H), 3.80 (s, 3H), 3.59-3.42 (m, 3H), 2.33-2.19 (m, 2H), 2.01-1.88 $(\mathrm{m}, 1 \mathrm{H}), 1.00-0.93(\mathrm{~m}, 6 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature. ${ }^{23}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400 \mathrm{MHz}\right)$.

(23) W.-H. Jung, C. Harrison, Y. Shin, J.-H. Fournier, R. Balachandran, B. S. Raccor, R. P. Sikorski, A. Vogt, D. P. Curran and B. W. Day, J. Med. Chem., 2007, 50, 2951-2966.


Workup procedure B was used and product 26 was isolated as a colorless oil ( $87 \%$ yield, $96: 4 \mathrm{dr}$ ). ${ }^{1} \mathbf{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.91-5.75(\mathrm{~m}, 1 \mathrm{H}), 5.20-5.10(\mathrm{~m}, 2 \mathrm{H}), 4.37-4.28(\mathrm{~m}, 1 \mathrm{H}), 4.09(\mathrm{dd}, J=$ $7.9,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.94-3.86(\mathrm{~m}, 1 \mathrm{H}), 3.53(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.36-2.20(\mathrm{~m}, 3 \mathrm{H}), 1.78$ (ddd, $J=14.2,7.4$, $3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.72-1.67(\mathrm{~m}, 1 \mathrm{H}), 1.67-1.59(\mathrm{~m}, 4 \mathrm{H}), 0.90(\mathrm{td}, J=7.5,3.0 \mathrm{~Hz}, 6 \mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data is in agreement with data reported in literature and comparison of the optical rotation to the literature value of the enantiomer verifies the absolute configuration of 26: $[\alpha]_{\mathrm{D}}{ }^{21}+8.0^{\circ}\left(\mathrm{CHCl}_{3}, c\right.$ 2.0); lit: $[\alpha]_{\mathrm{D}}{ }^{20}-6.8^{\circ}\left(\mathrm{CHCl}_{3}, ~ c ~ 1.5\right) .{ }^{12}$ The diastereomer ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 400 \mathrm{MHz}\right)$.


## Summary of Computational Method

All computations were performed using Jaguar, version 7.8, Schrodinger, LLC, New York, NY, 2010.

The geometries of the two silanes were optimized using the B3LYP functional and the 6-31G** basis set. Effective potentials (LACVP) were used for the Br atoms. At the optimized geometries the wavefunctions were recalculated using the cc-pVTZ basis. We include the Cartesian coordinates and final total energies for both optimized geometries below.

## Summary of Results for (S,S)-3:

| B3LYP/6-31G**/LACVP | final total energy: -1625.001295 hartrees |
| :--- | :--- |
| B3LYP/cc-pVTZ | final total energy: -6747.602235 hartrees |


final geometry:

| atom | x | y |  |
| :--- | ---: | ---: | ---: |
| C1 | -0.4982009010 | -0.5640678668 | 0.8474852299 |
| C2 | -0.1576222468 | -0.1958625197 | 2.1534727283 |
| C3 | 1.0727613256 | 0.4195061570 | 2.4103339770 |
| C4 | 1.9480308209 | 0.6792080770 | 1.3545975611 |
| C5 | 1.6100992271 | 0.3226800174 | 0.0485496634 |
| C6 | 0.3863010488 | -0.3031291056 | -0.1978759771 |
| O7 | -1.0334436822 | -0.4956775657 | 3.1641396989 |
| Si8 | -1.7700455138 | 0.4549984989 | 4.3390466916 |
| C9 | -2.0418846886 | 2.1770231235 | 3.6001871698 |
| C10 | -2.9453582217 | 2.2023827073 | 2.3981006494 |
| C11 | -4.1739416179 | 2.7254940397 | 2.3657413176 |
| N12 | -0.9690280043 | 0.5010854285 | 5.8820652026 |
| C13 | -0.0182402632 | 1.5211069219 | 6.2959736263 |


| C14 | -1.8832521928 | -0.0303815375 | 6.9120198075 |
| :---: | :---: | :---: | :---: |
| C15 | -1.2458034339 | -0.6978712367 | 8.1379147312 |
| C16 | -2.3613932726 | -1.1446182504 | 9.1043421478 |
| C17 | -3.4133951723 | -2.0357962735 | 8.4171237440 |
| C18 | -3.9781515781 | -1.4018151944 | 7.1292920414 |
| C19 | -2.8147551025 | -1.0311506027 | 6.2005926724 |
| N20 | -3.1562436694 | -0.4342479901 | 4.8919526346 |
| C21 | -3.9444448181 | -1.2624966286 | 3.9820337703 |
| H22 | -1.4504213607 | -1.0553405725 | 0.6723215627 |
| H23 | 0.1161663285 | -0.5909943474 | -1.2102681409 |
| H24 | 2.2960357665 | 0.5248909082 | -0.7686712377 |
| H25 | 2.9026480627 | 1.1563089897 | 1.5595912761 |
| H26 | 1.3492056702 | 0.6661535884 | 3.4314868324 |
| H27 | -2.5132331221 | 0.7994705880 | 7.2880968237 |
| H28 | -2.2340289456 | -1.9584575091 | 6.0269068021 |
| C29 | -5.4539340515 | -1.1295751911 | 4.1239484780 |
| H30 | -3.6779750782 | -0.9929979915 | 2.9521214853 |
| H31 | -3.6650948037 | -2.3234562540 | 4.0869491238 |
| C32 | 1.3531507227 | 1.0026834989 | 6.7192227862 |
| H33 | 0.1362551364 | 2.2254684592 | 5.4658514117 |
| H34 | -0.4373671572 | 2.1283356211 | 7.1145561899 |
| H35 | -2.4427923510 | 2.8145502182 | 4.3979522928 |
| H36 | -1.0552203807 | 2.5812033345 | 3.3295028244 |
| H37 | -4.6593087801 | -2.1025063641 | 6.6351534134 |
| H38 | -4.5639154299 | -0.5031553137 | 7.3621598592 |
| H39 | -4.2305463587 | -2.2563653252 | 9.1137291298 |
| H40 | -2.9544201580 | -3.0019942718 | 8.1640183474 |
| H41 | -1.9244707066 | -1.6715447829 | 9.9606146801 |
| H42 | -2.8566432472 | -0.2515898447 | 9.5111943039 |
| H43 | -0.5640965570 | -0.0063261404 | 8.6456683819 |
| H44 | -0.6432242736 | -1.5575338546 | 7.8176766809 |
| H45 | -2.5441732871 | 1.7428642439 | 1.4944631117 |
| H46 | -4.7804541868 | 2.6966756196 | 1.4655286638 |
| H47 | -4.6152696545 | 3.2073877397 | 3.2355323212 |
| C48 | 2.1729130327 | 1.7853923151 | 7.5408644804 |
| C49 | 3.4568765913 | 1.3668567354 | 7.8907550777 |
| C50 | 3.9176293209 | 0.1458346876 | 7.4075155358 |
| C51 | 3.1271196752 | -0.6574073858 | 6.5934063563 |
| C52 | 1.8445041272 | -0.2208140995 | 6.2541827064 |
| H53 | 1.8076188861 | 2.7385413777 | 7.9182807815 |
| H54 | 4.0833437751 | 1.9788909381 | 8.5303746046 |
| Br55 | 5.7236649228 | -0.4575323063 | 7.8929799115 |
| H56 | 3.4999892846 | -1.6081282321 | 6.2279784707 |
| H57 | 1.2090494514 | -0.8403520809 | 5.6287183110 |
| C58 | -6.2844097134 | -2.2189132765 | 3.8395641986 |
| C59 | -7.6746544207 | -2.1053454618 | 3.8977312124 |
| C60 | -8.2296587197 | -0.8796223542 | 4.2485675590 |
| C61 | -7.4324588634 | 0.2231670223 | 4.5390098918 |
| C62 | -6.0453303708 | 0.0890537655 | 4.4721939767 |
| H63 | -5.8440929298 | -3.1764730047 | 3.5689194189 |
| H64 | -8.3093067094 | -2.9572251584 | 3.6794077329 |
| Br 65 | -10.1857988199 | -0.7034761173 | 4.3440632248 |
| H66 | -7.8834537248 | 1.1710570300 | 4.8123383374 |
| H67 | -5.4074055684 | 0.9367597894 | 4.6996228566 |

## Summary of Results for (S,S)-8:

```
B3LYP/6-31G** final total energy: -1293.855729 hartrees
B3LYP/cc-pVTZ final total energy: -1294.193939 hartrees
```


final geometry:

| atom | x | y | z |
| :--- | ---: | ---: | ---: |
| C1 | 0.0076574797 | -0.0149841146 | -0.0145660255 |
| C2 | 0.0154663137 | 0.0001894247 | 1.5174014492 |
| C3 | 1.4664736155 | -0.0149028333 | 2.0378823512 |
| C4 | 2.1427474578 | -1.3289760735 | 1.6257535011 |
| C5 | 2.1300149303 | -1.4461415240 | 0.0879261775 |
| C6 | 0.7193075933 | -1.2895544495 | -0.5104417204 |
| N7 | -0.6535704584 | 1.1496187770 | 2.1631059257 |
| Si8 | 0.0148064139 | 1.2823197407 | 3.7681395841 |
| C9 | 0.3494102496 | 3.0215884364 | 4.4215168904 |
| C10 | 0.7041654423 | 3.1204446970 | 5.8792273811 |
| C11 | -0.1026466933 | 3.5624936228 | 6.8463533031 |
| N12 | 1.3908804297 | 0.2666592124 | 3.4834484720 |
| C13 | 2.6442184381 | 0.2908123302 | 4.2189722303 |
| C14 | -2.1269295348 | 1.1061770323 | 2.1790008072 |
| C15 | -2.7025329133 | 0.0197466015 | 3.0784920780 |
| C16 | -2.1781927671 | -0.1643640914 | 4.3752459941 |
| C17 | -2.6797326889 | -1.1543016174 | 5.2506787558 |


| C18 | -3.7359368018 | -1.9387187543 | 4.7633448001 |
| :---: | :---: | :---: | :---: |
| C19 | -4.2671304311 | -1.7736681485 | 3.4859143046 |
| C20 | -3.7441677001 | -0.7990217049 | 2.6442002246 |
| 021 | -1.1723470184 | 0.6815643042 | 4.8068671488 |
| H22 | -0.5597997914 | 3.6015111839 | 4.2175758882 |
| H23 | 1.1452926756 | 3.4498737189 | 3.7972997043 |
| H24 | 2.4653468474 | 0.5904070252 | 5.2558088189 |
| H25 | 3.3798589514 | 0.9908582780 | 3.7872866141 |
| H26 | 3.1113423419 | -0.7008760411 | 4.2500822614 |
| H27 | 2.0179625364 | 0.8042535971 | 1.5350100017 |
| H28 | -0.4553702844 | -0.9400544726 | 1.8583843837 |
| H29 | 3.1764615814 | -1.3676654657 | 1.9880786055 |
| H30 | 1.6098807236 | -2.1721938379 | 2.0843766903 |
| H31 | 0.5107076061 | 0.8850668941 | -0.3909743618 |
| H32 | -1.0195164161 | 0.0111993006 | -0.3969239148 |
| H33 | 2.7849928761 | -0.6689301452 | -0.3299192143 |
| H34 | 2.5609537545 | -2.4068966077 | -0.2166670475 |
| H35 | 0.7769920619 | -1.2886706512 | -1.6050991610 |
| H36 | 0.1131075912 | -2.1633214233 | -0.2340656024 |
| H37 | -2.5131551600 | 0.9811633042 | 1.1611089035 |
| H38 | -2.4810649071 | 2.0913835608 | 2.5142894264 |
| H39 | -4.1436649129 | -0.6633007225 | 1.6417144348 |
| H40 | -5.0813952376 | -2.4101347327 | 3.1521976933 |
| H41 | -4.1587253179 | -2.7098327117 | 5.3953217189 |
| C42 | -2.1042793903 | -1.3731072697 | 6.6678066022 |
| H43 | 1.7080958302 | 2.7919087048 | 6.1499752962 |
| H44 | 0.2177804776 | 3.5917021445 | 7.8831528841 |
| H45 | -1.1118728091 | 3.9076309732 | 6.6361154590 |
| C46 | -2.8378701491 | -2.5048449880 | 7.4177769426 |
| C47 | -0.6127836486 | -1.7773563136 | 6.5749118321 |
| C48 | -2.2537589989 | -0.0852128572 | 7.5138818867 |
| H49 | -2.3940299514 | -2.6243878589 | 8.4115503416 |
| H50 | -3.9015873540 | -2.2854964171 | 7.5580759102 |
| H51 | -2.7499093123 | -3.4670950135 | 6.9023985113 |
| H52 | -0.2003329190 | -1.9230108576 | 7.5803351467 |
| H53 | -0.5016386947 | -2.7194293044 | 6.0269092448 |
| H54 | -0.0175976963 | -1.0181465801 | 6.0671461427 |
| H55 | -1.8774750342 | -0.2613684505 | 8.5286061093 |
| H56 | -1.7005493751 | 0.7500186722 | 7.0835386033 |
| H57 | -3.3079356503 | 0.2025631881 | 7.5938412162 |

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2.0 f1 (ppm) 0.0

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C-BrC


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${ }^{13} \mathrm{C}$ NMR


|  | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $) 0$ | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | ( |



|  | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | T | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 0 | 45 | 40 | 35 | 30 | 25 | 20 | 15 | 10 | 5 | $\begin{gathered} 0 \\ \mathrm{f} 1(\mathrm{ppm}) \end{gathered}$ | －5 | －10 | －15 | －20 | －25 | －30 | －35 | －40 | －45 | －5 |

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${ }^{1} \mathrm{H}$ NMR

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～${ }^{13} \mathrm{C} \mathrm{NMR}$

$\stackrel{\sim}{i}$

2S：Lt
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[^1]11



$$
\begin{array}{llllllllllll}
210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\
\mathrm{f} 1(\mathrm{ppm})
\end{array}
$$

$-10$

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HSQC


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[^0]:    (7) S. Sano, Y. Kobayashi, T. Kondo, M. Takebayashi, S. Maruyama, T. Fujita and Y. Nagao, Tetrahedron Lett., 1995, 36, 2097-2100.
    (8) S. F. Vanier, G. Larouche, R. P. Wurz and A. B. Charette, Org. Lett., 2010, 12, 672-675.
    (9) C. L. Flowers and P. Vogel, Chemistry (Weinheim an der Bergstrasse, Germany), 2010, 16, 14074-14082.
    (10) A. B. Smith, T. J. Beauchamp, M. J. LaMarche, M. D. Kaufman, Y. Qiu, H. Arimoto, D. R. Jones and K. Kobayashi, J. Am. Chem. Soc., 2000, 122, 8654-8664.
    (11) Kaugars, S. J. Nelson, F. E. Dutton and S. E. Martin, Synth. Commun., 1993, 23, 797-809.
    (12) A. B. Smith, S. S.-Y. Chen, F. C. Nelson, J. M. Reichert and B. A. Salvatore, J. Am. Chem. Soc., 1995, 117, 12013-12014.
    (13) G. E. Keck and D. Krishnamurthy, Organic Syntheses, 1998, 75, 12-18.

[^1]:    $\stackrel{\infty}{\infty}$

