

**Chemo- and Diastereoselective Tandem Dual Oxidation of B(pin)-substituted Allylic Alcohols: Synthesis of B(pin)-substituted Epoxy Alcohols, 2-Keto-*anti*-1,3-diols and Dihydroxy-tetrahydrofuran-3-ones**

Nusrah Hussain, Mahmud M. Hussain, Patrick J. Carroll and Patrick J. Walsh\*

P. Roy and Diana T. Vagelos Laboratories, Department of Chemistry, University of Pennsylvania, 231 S. 34<sup>th</sup> Street, Philadelphia, PA 19104-6323  
[pwalsh@sas.upenn.edu](mailto:pwalsh@sas.upenn.edu)

**Supporting Information**

**Part 1**

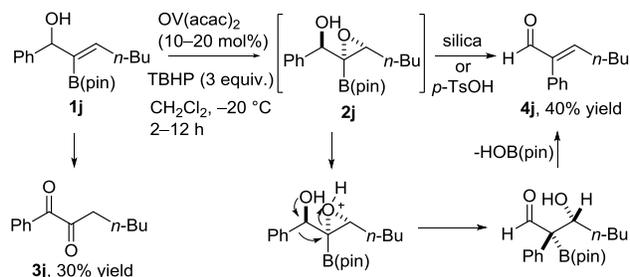
<b>Table of Contents Page</b>	<b>Page</b>
General Methods	S2
Synthesis of B(pin)-substituted Bis-allylic Alcohols	S4
Characterization of B(pin)-substituted Bis-allylic Alcohols	S4
Synthesis of Bis-epoxidation of B(pin)-substituted Bis-allylic Alcohols	S9
Characterization of B(pin)-substituted Bis-epoxides	S10
Characterization of B(pin)-substituted Mono-epoxides	S14
Oxidation of B(pin)-substituted Bis-epoxides	S20
Characterization of Epoxy-keto- <i>anti</i> -1,3-diols	S20
Synthesis of Fully Substituted Dihydroxy-tetrahydrofuran-3-ones	S23
Characterization of Dihydroxy-tetrahydrofuran-3-ones	S23
References	S26

**General Methods.** All reactions were performed under a nitrogen atmosphere with oven-dried glassware. All manipulations involving dicyclohexylborane and dimethylzinc were carried out under an inert atmosphere in a Vacuum Atmospheres drybox with an attached MO-40 DriTrain or by using standard Schlenk or vacuum line techniques. Chemicals were obtained from Aldrich, Acros, or Strem Chemicals unless otherwise specified. The oxidant *tert*-butylhydroperoxide (TBHP) was purchased from Aldrich as a ~5.5 M anhydrous solution in decane and hydrogen peroxide from Fischer as a 30% aqueous solution. Solvents were purchased from Fischer Scientific. Toluene and dichloromethane were dried through activated alumina columns. Tetrahydrofuran was distilled from sodium and benzophenone under N<sub>2</sub>. Liquid substrates were distilled prior to use. B(pin)-substituted alkynes were prepared by literature methods.<sup>1-9</sup> Neat dimethylzinc was obtained from Akzo Nobel from which 2.0 M solutions in toluene were prepared and stored in a Vacuum Atmospheres drybox. NMR spectra were obtained on Brüker 300, 360, 400 or 500 MHz Fourier transform spectrometers at the University of Pennsylvania NMR facility. <sup>1</sup>H and <sup>13</sup>C{<sup>1</sup>H} NMR spectra were referenced to residual solvent. <sup>11</sup>B{<sup>1</sup>H} NMR spectra were referenced to BF<sub>3</sub>·OEt<sub>2</sub>. The infrared spectra were obtained using a Perkin-Elmer 1600 series spectrometer. HRMS data was obtained on a Waters LC-TOF mass spectrometer (model LCT-XE Premier) using electrospray ionization in positive or negative mode, depending on analyte. Melting points were determined on a Uni-melt Thomas Hoover melting point apparatus and are uncorrected. Thin-layer chromatography was performed on Whatman precoated silica gel 60 F-254 plates and visualized by ultraviolet light or by staining with ceric ammonium molybdate, phosphomolybdic acid or potassium permanganate solutions. Silica gel (Silicaflash, P60, 40-63 μm, Silicycle) was used for airflashed chromatography, and deactivated silica gel was prepared by addition of 15 mL of Et<sub>3</sub>N to 1 L of silica gel. Full characterizations

of compounds **1a–1j**, **2b–2h**, **2j**, **3j**, **4j** and **5a–5i** were reported in our preliminary communication.<sup>8</sup>

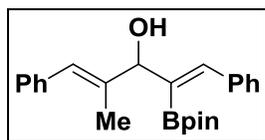
**Caution.** Dialkylzinc reagents are pyrophoric. Care must be used when handling them.

Optimization of the epoxidation of the benzylic substrate **1j** proved to be more challenging. The epoxide **2j** was observed by TLC along with the diketone **3j** (Scheme S1). Purification of the reaction mixture on silica gel resulted in decomposition of the B(pin)-substituted epoxide with formation of the  $\alpha,\beta$ -unsaturated aldehyde **4j** in ~40% yield (entries 12–13 in Table S2, and Scheme S1). We hypothesized that the enal **4j** arose via an acid or Lewis acid promoted semi-pinacol rearrangement followed by *syn*-elimination of the HO–B(pin). A similar HO–BAr<sub>2</sub> elimination takes place in the boron Wittig-type reaction.<sup>10, 11</sup> The elimination mechanism in Scheme S1 is consistent with the observed double bond geometry in the enal **4j**. The byproduct **3j** was identified as the known diketone.<sup>12</sup> Vanadium(V) catalysts are known to oxidize alcohols to the corresponding ketones in the presence of TBHP.<sup>13, 14</sup> Diketone **3j** may be formed by initial oxidation of the benzylic alcohol to the ketone followed by oxidation of the vinyl boronate ester to form the dione.



**Scheme S1.** Key Intermediates in the Proposed Mechanism of the Epoxy Alcohol Rearrangement to form Enal **4j**

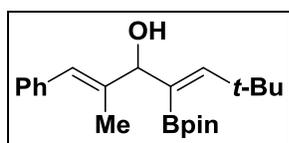
**General Procedure A: Synthesis of B(pin)-substituted Bis-allylic Alcohols.** To a suspension of HBCy<sub>2</sub> (1.2 equiv.) in toluene (2.0 mL) under N<sub>2</sub> was added alkyne-4,4,5,5-tetramethyl-[1,3,2]-dioxaborolane (1.2 equiv.) and the reaction mixture was stirred for 30 min at rt, after which it was homogeneous. The reaction vessel was cooled to -78 °C and treated with Me<sub>2</sub>Zn (1.2 equiv., 2.0 M in toluene) for 30–45 min. The solution was then warmed to -10 °C and the enal (1 equiv.) was added. The reaction mixture was stirred at -15 °C until TLC showed complete consumption of the aldehyde (8–12 h). The reaction mixture was then diluted with EtOAc and quenched with saturated NH<sub>4</sub>Cl at 0 °C. The organic layer was separated and the aqueous solution was extracted three times with 10 mL of EtOAc. The combined organic solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The crude product was purified by flash column chromatography on silica gel. The bis-allylic alcohol products are susceptible to oxidation of the B–C bond on silica under air. Rapid purification is therefore necessary to minimize oxidation to the ketones, which elute at similar R<sub>f</sub> values to the bis-allylic alcohols. The bis-allylic alcohols are stored under N<sub>2</sub> at 0 °C to preserve their purity.



**(1E,4E)-2-Methyl-1,5-diphenyl-4(4,4,5,5-tetramethyl-1,3,2-**

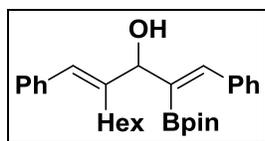
**dioxaborolane-2-yl)penta-1,4-dien-3-ol (1k).** The product was prepared by General Procedure A using  $\alpha$ -methyl cinnamaldehyde (0.42 mL, 3.0 mmol) and 4,4,5,5-tetramethyl-2-(phenylethynyl)-1,3,2-dioxaborolane (0.82 g, 3.6 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **1k** (1.04 g, 92% yield). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.44 (d,  $J$  = 7.1 Hz, 2H), 7.38 –

7.20 (m, 8H), 7.18 (s, 1H), 6.71 (s, 1H), 4.86 (d,  $J = 4.5$  Hz, 1H), 2.65 (d,  $J = 5.7$  Hz, 1H), 1.90 (s, 3H), 1.24 (s, 12H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  141.0, 139.2, 137.9, 137.7, 129.0, 128.3, 128.0, 127.9, 127.6, 126.2, 125.2, 83.9, 81.9, 24.9, 24.7, 14.8 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.5; IR (neat) 3448, 3058, 3026, 2930, 2855, 1684, 1625, 1600, 1494, 1449, 1312, 1248, 1143  $\text{cm}^{-1}$ ; HRMS  $m/z$  399.2118 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{24}\text{H}_{29}\text{BO}_3\text{Na}$ : 399.2107].



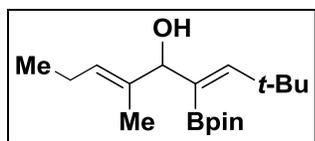
**(1E,4E)-2,6,6-Trimethyl-1-phenyl-4-(4,4,5,5-tetramethyl-1,3,2-**

**dioxaborolan-2-yl)hepta-1,4-dien-3-ol (11).** The product was prepared by General Procedure A using  $\alpha$ -methyl cinnamaldehyde (0.28 mL, 2.0 mmol) and 2-(3,3-dimethylbut-1-ynyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (0.55 g, 2.4 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **11** (0.60 g, 84% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.38 – 7.27 (m, 4H), 7.23 – 7.17 (m, 1H), 6.66 (s, 1H), 6.14 (s, 1H), 4.54 (d,  $J = 5.4$  Hz, 1H), 2.55 (d,  $J = 6.3$  Hz, 1H), 1.79 (s, 3H), 1.25 (s, 6H), 1.24 (s, 6H), 1.14 (s, 9H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  154.1, 139.9, 138.3, 129.3, 128.2, 126.3, 124.1, 84.0, 83.4, 34.3, 30.5, 25.4, 25.3, 15.5 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.0; IR (neat) 3469, 3023, 2978, 2954, 1640, 1600, 1480, 1380, 1304, 1253, 1142  $\text{cm}^{-1}$ ; HRMS  $m/z$  379.2425 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{22}\text{H}_{33}\text{BO}_3\text{Na}$ : 379.2420].



**(1*E*,4*E*)-4-Benzylidene-1-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-**

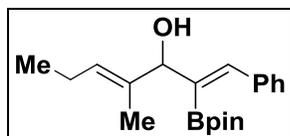
**dioxaborolan-2-yl)dec-1-en-3-ol (1m).** The product was prepared by General Procedure A using  $\alpha$ -hexyl cinnamaldehyde (0.46 mL, 2.0 mmol) and 4,4,5,5-tetramethyl-2-(phenylethynyl)-1,3,2-dioxaborolane (0.55 g, 2.4 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **1m** (0.54 g, 60% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.47 – 7.39 (m, 2H), 7.33 – 7.26 (m, 7H), 7.23 – 7.20 (m, 1H), 7.18 (s, 1H), 6.69 (s, 1H), 4.91 (s, 1H), 2.74 (s, 1H), 2.53 – 2.38 (m, 1H), 2.24 – 2.14 (m, 1H), 1.54 (p,  $J = 7.3$  Hz, 2H), 1.31 – 1.21 (m, 18H), 0.86 (t,  $J = 6.7$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  144.4, 141.8, 138.2, 138.1, 128.9, 128.7, 128.2, 128.1, 127.9, 126.5, 125.4, 84.1, 80.5, 31.7, 29.7, 29.0, 28.8, 25.2, 24.9, 22.8, 14.3 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.3; IR (neat) 3423, 3057, 3025, 2928, 2856, 1685, 1625, 1600, 1493, 1450, 1379, 1310, 1249, 1142  $\text{cm}^{-1}$ ; HRMS  $m/z$  469.2896 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{29}\text{H}_{39}\text{BO}_3\text{Na}$ : 469.2890].



**(3*E*,6*E*)-2,2,6-Trimethyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-**

**2-yl)nona-3,6-dien-5-ol (1n).** The product was prepared by General Procedure A using (*E*)-2-methylpent-2-enal (0.23 mL, 2.0 mmol) and 2-(3,3-dimethylbut-1-ynyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (0.55 g, 2.4 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **1n** (0.41 g, 67% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  6.00 (s, 1H), 5.51 – 5.40 (m, 1H), 4.37 (s, 1H), 2.33 (s, 1H), 2.01 (p,  $J = 7.6$  Hz, 2H), 1.49 (s, 3H), 1.23 (s, 6H), 1.22 (s, 6H), 1.07 (s, 9H), 0.95

(t,  $J = 7.6$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  152.5, 136.0, 126.6, 83.7, 82.5, 82.4, 34.0, 30.5, 25.2, 21.1, 14.2, 13.0 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  37.0; IR (neat) 3479, 2958, 2872, 1640, 1480, 1463, 1380, 1301, 1253, 1144  $\text{cm}^{-1}$ ; HRMS  $m/z$  331.2419 [(M+Na) $^+$ ; calcd for  $\text{C}_{18}\text{H}_{33}\text{BO}_3\text{Na}$ : 331.2420].



**(1E,4E)-4-Methyl-1-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-**

**dioxaborolan-2-yl)hepta-1,4-dien-3-ol (1o).** The product was prepared by General Procedure

A using (*E*)-2-methylpent-2-enal (0.23 mL, 2.0 mmol) and 4,4,5,5-tetramethyl-2-

(phenylethynyl)-1,3,2-dioxaborolane (0.55 g, 2.4 mmol). The crude product was purified by

flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic

alcohol **1o** (0.37 g, 56% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.39 (d,  $J = 7.7$  Hz, 2H), 7.33 –

7.17 (m, 3H), 7.09 (s, 1H), 5.56 (t,  $J = 6.9$  Hz, 1H), 4.71 (s, 1H), 2.42 (s, 1H), 2.08 (p,  $J = 7.2$

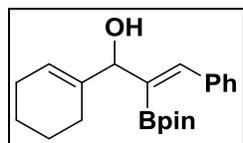
Hz, 2H), 1.64 (s, 3H), 1.24 (s, 12H), 1.00 (t,  $J = 7.5$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

$\delta$  140.1, 138.3, 135.8, 128.5, 128.12 128.1, 127.6, 84.0, 81.4, 25.0, 24.9, 21.2, 14.2, 12.8 (the

quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.0;

IR (neat) 3433, 3027, 2930, 2977, 2873, 1629, 1600, 1494, 1449, 1380, 1310, 1247, 1143  $\text{cm}^{-1}$ ;

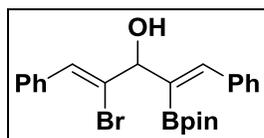
HRMS  $m/z$  351.2119 [(M+Na) $^+$ ; calcd for  $\text{C}_{20}\text{H}_{29}\text{BO}_3\text{Na}$ : 351.2107].



**(E)-1-Cyclohexenyl-3-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-**

**2-yl)prop-2-en-1-ol (1p).** The product was prepared by General Procedure A using cyclohex-1-

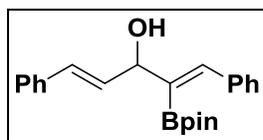
enecarbaldehyde (0.23 mL, 2.0 mmol) and 4,4,5,5-tetramethyl-2-(phenylethynyl)-1,3,2-dioxaborolane (0.55 g, 2.4 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **1p** (0.48 g, 71% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.39 (d,  $J = 7.2$  Hz, 2H), 7.28 (t,  $J = 7.1$  Hz, 2H), 7.23 (t,  $J = 7.2$  Hz, 1H), 7.08 (s, 1H), 5.84 – 5.78 (m, 1H), 4.65 (s, 1H), 2.47 (s, 1H), 2.10 – 1.96 (m, 4H), 1.69 – 1.52 (m, 4H), 1.25 (s, 12H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  140.3, 139.3, 138.2, 128.5, 128.1, 127.6, 122.8, 84.0, 80.7, 25.2, 25.1, 25.0, 24.9, 22.8, 22.7 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.3; IR (neat) 3423, 3026, 2978, 2927, 2856, 1626, 1600, 1495, 1449, 1389, 1309, 1248, 1142  $\text{cm}^{-1}$ ; HRMS  $m/z$  363.2103 [(M+Na) $^+$ ]; calcd for  $\text{C}_{21}\text{H}_{29}\text{BO}_3\text{Na}$ : 363.2107].



**(1Z,4E)-2-Bromo-1,5-diphenyl-4-(4,4,5,5-tetramethyl-1,3,2-**

**dioxaborolan-2-yl)penta-1,4-dien-3-ol (1q)**. The product was prepared by General Procedure A using  $\alpha$ -bromo cinnamaldehyde (0.42 mL, 2.0 mmol) and 4,4,5,5-tetramethyl-2-(phenylethynyl)-1,3,2-dioxaborolane (0.55 g, 2.4 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **1q** (0.81 g, 92% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.63 (d,  $J = 7.7$  Hz, 2H), 7.43 (d,  $J = 7.6$  Hz, 2H), 7.39 – 7.18 (m, 8H), 5.03 (d,  $J = 7.0$  Hz, 1H), 3.19 (d,  $J = 8.1$  Hz, 1H), 1.22 (s, 12H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  144.1, 137.6, 135.5, 129.3, 128.9, 128.8, 128.3, 128.2, 128.2, 128.1, 128.1, 84.3, 82.2, 25.1, 24.9 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.2; IR (neat) 3433, 3025, 2979, 2930, 2874,

1627, 1600, 1493, 1447, 1391, 1313, 1249, 1141  $\text{cm}^{-1}$ ; HRMS  $m/z$  463.1042  $[(M+Na)^+]$ ; calcd for  $\text{C}_{23}\text{H}_{26}\text{BBrO}_3\text{Na}$ : 463.1056].

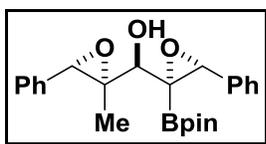


**(1E,4E)-1,5-Diphenyl-2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-**

**yl)penta-1,4-dien-3-ol (1r).** The product was prepared by General Procedure A using cinnamaldehyde (0.13 mL, 1.0 mmol) and 4,4,5,5-tetramethyl-2-(phenylethynyl)-1,3,2-dioxaborolane (0.27 g, 1.2 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the bis-allylic alcohol **1r** (0.33 g, 92% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.52 – 7.36 (m, 4H), 7.36 – 7.22 (m, 6H), 7.17 (s, 1H), 6.66 (d,  $J = 16$  Hz, 1H), 6.39 (dd,  $J = 16.8, 5.8$  Hz, 1H), 4.99 (s, 1H), 2.67 (s, 1H), 1.25 (s, 6H), 1.25 (s, 6H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  141.7, 137.9, 137.0, 131.7, 130.5, 128.7, 128.7, 128.1, 128.0, 127.7, 126.7, 84.2, 78.5, 35.7, 25.1, 24.8 (the quaternary vinyl C bearing the boron is not observed).

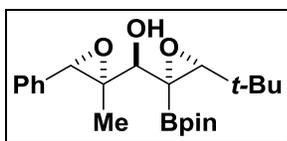
**General Procedure B: Synthesis of B(pin)-substituted Bis-epoxides.** To a Schlenk flask containing the B(pin)-substituted bis-allylic alcohol (1.0 equiv.) was added 1 mL of freshly distilled  $\text{CH}_2\text{Cl}_2$  followed by solid  $\text{OV}(\text{acac})_2$  (10 mol %) under  $\text{N}_2$ . The resulting greenish-blue solution was cooled to 0 °C and a solution of TBHP (0.7–3.0 equiv., ~5.5 M solution in decane) in 1 mL of  $\text{CH}_2\text{Cl}_2$  was added slowly to the reaction mixture over 10 min using a syringe pump at that temperature. The solution rapidly changed color to a dark brown. The reaction mixture was stirred at 0 °C until TLC showed complete consumption of the bis-allylic alcohol (30 min – 2 h). The crude reaction mixture was filtered through a short pad of silica, and the solvent was

removed under reduced pressure (>90% purity by  $^1\text{H}$  NMR). The crude product was further purified by flash column chromatography on silica gel. The epoxy boronate ester is susceptible to oxidation of the B–C bond on silica under air, and hence a rapid purification is necessary to minimize oxidation to the corresponding diol and other side products.



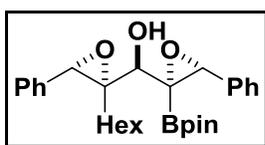
(2-Methyl-3-phenyloxiran-2-yl)(3-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-

dioxaborolan-2-yl)methanol (**2k**). The product was prepared by General Procedure B using bis-allylic alcohol **1k** (0.07 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (38.2  $\mu\text{L}$ , ~5.5 M solution in decane, 3 equiv.). The TBHP solution was added slowly to the reaction mixture over 10 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the epoxide (>90% purity by  $^1\text{H}$  NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford a single diastereomer of the bis-epoxide **2k** (19.7 mg, 69% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.43 – 7.39 (m, 2H), 7.32 – 7.27 (m, 2H), 7.26 – 7.22 (m, 1H), 4.02 (s, 1H), 3.50 (d,  $J = 3.1$  Hz, 1H), 3.38 (s, 1H), 2.72 (s, 1H), 2.14 – 2.06 (m, 1H), 1.97 – 1.82 (m, 3H), 1.51 – 1.39 (m, 2H), 1.35 – 1.25 (m, 2H), 1.00 (s, 6H), 0.93 (s, 6H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  136.0, 135.7, 128.2, 128.1, 128.0, 127.7, 126.9, 126.5, 84.8, 80.1, 65.2, 61.2, 60.1, 24.9, 24.6, 13.9 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  29.7; IR (neat) 3521, 3032, 2979, 2931, 1605, 1498, 1454, 1381, 1335, 1250, 1134  $\text{cm}^{-1}$ ; HRMS  $m/z$  431.1975 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{20}\text{H}_{29}\text{BO}_4\text{Na}$ : 431.2007].



(3-*tert*-Butyl-2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)(2-

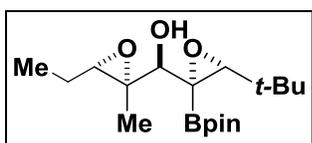
**methyl-3-phenyloxiran-2-yl)methanol (2l).** The product was prepared by General Procedure B using bis-allylic alcohol **1l** (0.10 mmol, 1 equiv.) and a solution of TBHP in 1 mL of CH<sub>2</sub>Cl<sub>2</sub> (54.5 μL, ~5.5 M solution in decane, 3 equiv.). The TBHP solution was added slowly to the reaction mixture over 10 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the bis-epoxide (>90% purity by <sup>1</sup>H NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford a single diastereomer of the bis-epoxide **2k** (28.0 mg, 72% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.45 – 7.17 (m, 5H), 4.41 (s, 1H), 3.35 (s, 1H), 2.84 (s, 2H), 1.37 (s, 6H), 1.35 (s, 6H), 1.15 (s, 3H), 1.03 (s, 9H); <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) δ 135.8, 128.1, 127.6, 126.8, 85.0, 80.8, 71.5, 65.1, 61.4, 31.7, 27.0, 25.8, 25.5, 13.9 (the quaternary vinyl C bearing the boron is not observed).



(2-Hexyl-3-phenyloxiran-2-yl)(3-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-

**dioxaborolan-2-yl)oxiran-2-yl)methanol (2m).** The product was prepared by General Procedure B using bis-allylic alcohol **1m** (0.890 mmol, 1 equiv.) and a solution of TBHP in 1 mL of CH<sub>2</sub>Cl<sub>2</sub> (0.49 mL, ~5.5 M solution in decane, 3 equiv.). The TBHP solution was added slowly to the reaction mixture over 10 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the bis-epoxide (0.37 g, 86% <sup>1</sup>H NMR yield with internal standard CH<sub>2</sub>Br<sub>2</sub>, >90% purity by <sup>1</sup>H NMR). The product was further purified by

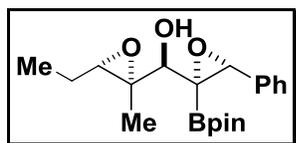
flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford a single diastereomer of the bis-epoxide **2m** (0.34 g, 80% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.45 (d,  $J = 7.4$  Hz, 2H), 7.36 – 7.26 (m, 8H), 4.48 (s, 1H), 4.10 (s, 1H), 3.66 (d,  $J = 10.7$  Hz, 1H), 2.96 (d,  $J = 10.8$  Hz, 1H), 2.02 (ddd,  $J = 13.2, 10.4, 4.8$  Hz, 1H), 1.21 – 1.14 (m, 1H), 1.13 – 1.07 (m, 2H), 1.06 – 0.97 (m, 10H), 0.96 (s, 6H), 0.76 (t,  $J = 7.2$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  136.0, 135.7, 128.1, 128.0, 128.0, 127.7, 126.9, 126.6, 84.8, 77.8, 68.0, 61.1, 61.0, 31.4, 29.3, 26.9, 24.8, 24.6, 22.5, 14.2 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  28.4; IR (neat) 3521, 3032, 2979, 2931, 1605, 1498, 1454, 1418, 1381, 1335, 1250, 1134  $\text{cm}^{-1}$ ; HRMS  $m/z$  431.1975 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{24}\text{H}_{29}\text{BO}_5\text{Na}$ : 431.2006].



(3-(*tert*-Butyl)-2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-

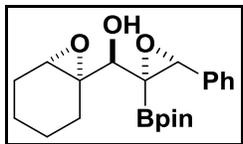
yl)oxiran-2-yl)(3-ethyl-2-methyloxiran-2-yl)methanol (**2n**). The product was prepared by General Procedure B using bis-allylic alcohol **1n** (0.89 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (0.49 mL, ~5.5 M solution in decane, 3 equiv.). The TBHP solution was added slowly to the reaction mixture over 10 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the bis-epoxide (0.27 g, 88%  $^1\text{H}$  NMR yield with internal standard  $\text{CH}_2\text{Br}_2$ , >90% purity by  $^1\text{H}$  NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford a single diastereomer of the bis-epoxide **2n** (0.24 g, 80% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  3.24 (dd,  $J = 7.5, 5.1$  Hz, 1H), 3.14 (s, 1H), 2.74 (s, 1H), 2.69 (s, 1H), 1.65 – 1.47 (m, 2H), 1.35 (s, 3H), 1.33 (s, 12H), 1.04 (t,  $J = 7.5$  Hz, 3H), 0.99 (s, 9H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  84.8, 81.4,

72.0, 63.0, 62.3, 31.6, 27.0, 25.9, 25.5, 21.8, 14.3, 10.8 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.0; IR (neat) 3525, 2977, 2933, 1411, 1381, 1334, 1250, 1135  $\text{cm}^{-1}$ .



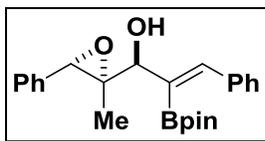
**(3-Ethyl-2-methyl-2-methyloxiran-2-yl)(3-phenyl-2-(4,4,5,5-**

**tetramethyl-1,3,2-dioxaborolan-2-yl)methanol (2o).** The product was prepared by General Procedure B using bis-allylic alcohol **1o** (0.19 g, 0.59 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (0.32 mL, ~5.5 M solution in decane, 3 equiv.). The TBHP solution was added slowly to the reaction mixture over 10 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the bis-epoxide (0.19 g, 90%  $^1\text{H}$  NMR yield with internal standard  $\text{CH}_2\text{Br}_2$ , >90% purity by  $^1\text{H}$  NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford a single diastereomer of the bis-epoxide **2o** (0.18 g, 84% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.46 – 7.39 (m, 2H), 7.33 – 7.25 (m, 3H), 4.02 (s, 1H), 3.35 (d,  $J$  = 10.1 Hz, 1H), 3.29 (dd,  $J$  = 7.2, 5.4 Hz, 1H), 2.78 (d,  $J$  = 10.3 Hz, 1H), 1.70 – 1.52 (m, 2H), 1.44 (s, 3H), 1.08 (t,  $J$  = 7.5 Hz, 3H), 1.01 (s, 6H), 0.93 (s, 6H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  136.0, 128.1, 128.0, 126.6, 84.7, 80.4, 63.1, 62.2, 61.1, 24.9, 24.5, 21.8, 14.2, 10.7 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  29.4; IR (neat) 3525, 2976, 2932, 1600, 1455, 1421, 1381, 1335, 1250, 1135  $\text{cm}^{-1}$ ; HRMS  $m/z$  383.2007 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{20}\text{H}_{29}\text{BO}_5\text{Na}$ : 383.2006].



**(7-Oxabicycloheptan-1-yl)(3-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-**

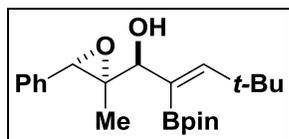
**dioxaborolan-2-yl)oxiran-2-yl)methanol (2p).** The product was prepared by General Procedure B using bis-allylic alcohol **1p** (0.12 g, 0.34 mmol, 1 equiv.) and a solution of TBHP in 1 mL of CH<sub>2</sub>Cl<sub>2</sub> (0.19 mL, ~5.5 M solution in decane, 3 equiv.). The TBHP solution was added slowly to the reaction mixture over 10 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the bis-epoxide (0.10 g, 82% <sup>1</sup>H NMR yield with internal standard CH<sub>2</sub>Br<sub>2</sub>, >90% purity by <sup>1</sup>H NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford a single diastereomer of the bis-epoxide **2p** as white solid (92.4 mg, 73% yield). M.p 104-107 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.43 – 7.39 (m, 2H), 7.32 – 7.27 (m, 2H), 7.26 – 7.22 (m, 1H), 4.02 (s, 1H), 3.50 (d, *J* = 3.1 Hz, 1H), 3.38 (s, 1H), 2.72 (s, 1H), 2.14 – 2.06 (m, 1H), 1.97 – 1.82 (m, 3H), 1.51 – 1.39 (m, 2H), 1.35 – 1.25 (m, 2H), 1.00 (s, 6H), 0.93 (s, 6H); <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) δ 136.0, 128.1, 128.0, 126.5, 84.6, 80.1, 61.9, 60.9, 56.5, 25.6, 24.9, 24.6, 24.5, 20.1, 19.7 (the quaternary vinyl C bearing the boron is not observed); <sup>11</sup>B{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 128 MHz) δ 29.7; IR (neat) 3529, 3062, 2978, 2936, 2860, 1605, 1498, 1450, 1421, 1381, 1336, 1249, 1132 cm<sup>-1</sup>; HRMS *m/z* 395.2014 [(M+Na)<sup>+</sup>; calcd for C<sub>21</sub>H<sub>29</sub>BO<sub>5</sub>Na: 395.2006].



**(E)-1-(2-Methyl-3-phenyloxiran-2-yl)-3-phenyl-2-(4,4,5,5-tetramethyl-**

**1,3,2-dioxaborolan-2-yl)prop-2-en-1-ol (3k).** The product was prepared by General Procedure B using bis-allylic alcohol **1k** (0.10 mmol, 1 equiv.) and a solution of TBHP in 1 mL of CH<sub>2</sub>Cl<sub>2</sub>

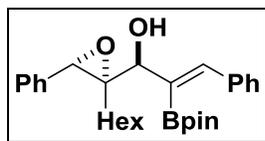
(18.2  $\mu\text{L}$ ,  $\sim 5.5$  M solution in decane, 1 equiv.). The TBHP solution was added slowly to the reaction mixture over 30 min using a syringe pump. The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **3k** (27.1 mg, 69% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ );  $\delta$  7.47 – 7.25 (m, 10H), 7.14 (s, 1H), 4.51 (s, 1H), 4.41 (s, 1H), 2.81 (s, 1H), 1.35 (s, 6H), 1.31 (s, 6H), 1.15 (s, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  142.8, 138.0, 136.4, 128.7, 128.2, 128.2, 128.1, 127.5, 126.7, 84.2, 80.1, 65.7, 60.5, 25.2, 25.1, 14.3 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.2; IR (neat) 3445, 3028, 2978, 2929, 2856, 1626, 1601, 1497, 1449, 1391, 1380, 1311, 1252, 1143  $\text{cm}^{-1}$ .



**(E)-4,4-Dimethyl-1-(2-methyl-3-phenyloxiran-2-yl)-2-(4,4,5,5-**

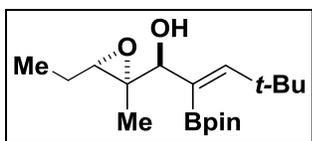
**tetramethyl-1,3,2-dioxaborolan-2-yl)pent-2-en-1-ol (3l).** The product was prepared by General Procedure B using bis-allylic alcohol **11** (0.1 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (14.6  $\mu\text{L}$ ,  $\sim 5.5$  M solution in decane, 0.8 equiv.). The TBHP solution was added slowly to the reaction mixture over 30 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the mono-epoxide (>90% purity by  $^1\text{H}$  NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **3n** (26.8 mg, 90% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.41 (m, 2H), 7.35 (t,  $J = 7.6$  Hz, 2H), 7.30 – 7.28 (m, 1H), 6.08 (s, 1H), 4.35 (s, 1H), 4.29 (d,  $J = 2.7$  Hz, 1H), 2.68 (d,  $J = 3.1$  Hz, 1H), 1.36 (s, 6H), 1.35 (s, 6H), 1.12 (s, 9H), 1.06 (s, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  154.4, 136.4, 127.8, 127.1, 126.5, 83.7, 80.6, 65.5, 60.0,

34.2, 30.1, 25.2, 25.0, 14.2 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.5.



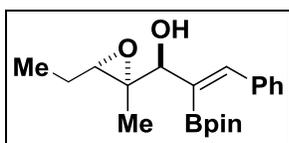
**(E)-1-(2-Hexyl-3-phenyloxiran-2-yl)-3-phenyl-2-(4,4,5,5-tetramethyl-**

**1,3,2-dioxaborolan-2-yl)prop-2-en-1-ol (3m).** The product was prepared by General Procedure B using bis-allylic alcohol **1m** (0.35 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (51.0  $\mu\text{L}$ ,  $\sim 5.5$  M solution in decane, 0.28 mmol, 0.8 equiv.). The TBHP solution was added slowly to the reaction mixture over 30 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the mono-epoxide (0.11 g, 86%  $^1\text{H}$  NMR yield with internal standard  $\text{CH}_2\text{Br}_2$ , >90% purity by  $^1\text{H}$  NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **3m** (90.6 mg, 70% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ); 7.46 – 7.27 (m, 10H), 7.15 (s, 1H), 4.67 (d,  $J = 3.9$  Hz, 1H), 4.37 (s, 1H), 2.89 (d,  $J = 4.5$  Hz, 1H), 1.50 – 1.44 (m, 2H), 1.35 (s, 6H), 1.31 (s, 6H), 1.25 – 1.06 (m, 8H), 0.82 (t,  $J = 7.1$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  143.1, 137.9, 136.4, 128.7, 128.2, 128.1, 128.0, 127.4, 126.7, 84.1, 77.7, 68.3, 60.7, 31.6, 29.6, 26.8, 25.1, 24.8, 22.6, 14.2 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.1; IR (neat) 3470, 3027, 2977, 2955, 2929, 2858, 1626, 1602, 1496, 1455, 1391, 1310, 1250, 1143  $\text{cm}^{-1}$ ; HRMS  $m/z$  485.2822 [ $(\text{M}+\text{Na})^+$ ; calcd for  $\text{C}_{29}\text{H}_{39}\text{BO}_4\text{Na}$ : 485.2839].



(*E*)-1-(3-Ethyl-2-methyloxiran-2-yl)-4,4--dimethyl-2-(4,4,5,5-

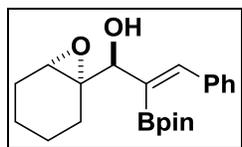
tetramethyl-1,3,2-dioxaborolan-2-yl)pent-2-en-1-ol (**3n**). The product was prepared by General Procedure B using bis-allylic alcohol **1n** (0.91 mmol, 1 equiv.) and a solution of TBHP in 1 mL of CH<sub>2</sub>Cl<sub>2</sub> (0.12 mL, ~5.5 M solution in decane, 0.7 equiv.). The TBHP solution was added slowly to the reaction mixture over 30 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the mono-epoxide (0.15 g, 75% yield <sup>1</sup>H NMR yield with internal standard CH<sub>2</sub>Br<sub>2</sub>, >90% purity by <sup>1</sup>H NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **3n** (0.13 g, 63% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>); δ 6.02 (s, 1H), 4.09 (d, *J* = 3.3 Hz, 1H), 3.08 (dd, *J* = 7.4, 5.2 Hz, 1H), 2.52 (d, *J* = 3.4 Hz, 1H), 1.65 – 1.58 (m, 1H), 1.58 – 1.50 (m, 1H), 1.28 (s, 12H), 1.25 (s, 3H), 1.09 (s, 9H), 1.04 (t, *J* = 7.5 Hz, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) δ 154.6, 83.8, 81.4, 63.4, 61.5, 34.4, 30.4, 25.5, 25.2, 21.8, 14.9, 11.0 (the quaternary vinyl C bearing the boron is not observed); <sup>11</sup>B{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 128 MHz) δ 30.2; IR (neat) 3480, 2969, 2875, 1639, 1464, 1411, 1373, 1305, 1255, 1144 cm<sup>-1</sup>; HRMS *m/z* 347.2362 [(M+Na)<sup>+</sup>; calcd for C<sub>18</sub>H<sub>33</sub>BO<sub>4</sub>Na: 347.2370].



(*E*)-1-(3-Ethyl-2-methyloxiran-2-yl)-3-phenyl-2-(4,4,5,5-tetramethyl-

1,3,2-dioxaborolan-2-yl)prop-2-en-1-ol (**3o**). The product was prepared by General Procedure B using bis-allylic alcohol **1o** (1.2 mmol, 1 equiv.) and a solution of TBHP in 1 mL of CH<sub>2</sub>Cl<sub>2</sub> (0.15 mL, ~5.5 M solution in decane, 0.7 equiv.). The TBHP solution was added slowly to the

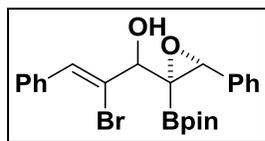
reaction mixture over 30 min using a syringe pump. The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **3o** (0.16 g, 54% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.38 (d,  $J = 6.5$  Hz, 2H), 7.32 – 7.23 (m, 3H), 7.07 (s, 1H), 4.33 (s, 1H), 3.13 (t,  $J = 6.3$  Hz, 1H), 2.75 (s, 1H), 1.68 – 1.54 (m, 2H), 1.36 (s, 3H), 1.28 (s, 6H), 1.26 (s, 6H), 1.07 (t,  $J = 7.5$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  142.9, 138.0, 128.6, 128.2, 128.0, 84.0, 80.6, 63.4, 61.6, 25.1, 24.9, 21.8, 14.7, 11.0 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128 MHz)  $\delta$  30.8; IR (neat) 3463, 3026, 2976, 2931, 2876, 1626, 1600, 1493, 1459, 1390, 1311, 1252, 1143  $\text{cm}^{-1}$ ; HRMS  $m/z$  367.2054 [(M+Na) $^+$ ; calcd for  $\text{C}_{20}\text{H}_{29}\text{BO}_4\text{Na}$ : 367.2057].



(*E*)-1-(7-Oxabicycloheptan-1-yl)-3-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-

**dioxaborolan-2-yl)prop-2-en-1-ol (3p)**. The product was prepared by General Procedure B using bis-allylic alcohol **1p** (0.07 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (9.5  $\mu\text{L}$ , ~5.5 M solution in decane, 0.7 equiv.). The TBHP solution was added slowly to the reaction mixture over 30 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the mono-epoxide (>90% purity by  $^1\text{H}$  NMR). The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **3p** (9.1 mg, 52% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.40 (d,  $J = 7.7$  Hz, 2H), 7.33 – 7.22 (m, 3H), 7.08 (s, 1H), 4.31 (s, 1H), 3.40 – 3.26 (m, 1H), 2.70 (s, 1H), 2.04 – 1.81 (m, 4H), 1.51 – 1.40 (m, 2H), 1.36 – 1.22 (m, 14H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  143.3, 138.0, 128.7, 128.1, 128.0, 84.0, 80.2, 62.4, 56.4, 25.5, 25.2, 25.1, 24.5, 20.3, 20.0 (the quaternary vinyl C bearing the boron is not observed);  $^{11}\text{B}\{^1\text{H}\}$  NMR ( $\text{CDCl}_3$ , 128

MHz)  $\delta$  30.5; IR (neat) 3479, 3025, 2974, 2932, 2875, 1627, 1600, 1494, 1461, 1389, 1311, 1253, 1143  $\text{cm}^{-1}$ ; HRMS  $m/z$  379.2052  $[(M+Na)^+]$ ; calcd for  $\text{C}_{21}\text{H}_{29}\text{BO}_4\text{Na}$ : 379.2057].



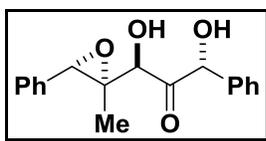
**(Z)-2-Bromo-3-phenyl-1-(3-phenyl-2-(4,4,5,5-tetramethyl-1,3,2-**

**dioxaborolan-2-yl)oxiran-2-yl)prop-2-en-1-ol (4q).** The product was prepared by General Procedure B using bis-allylic alcohol **1q** (0.14 g, 0.31 mmol, 1 equiv.) and a solution of TBHP in 1 mL of  $\text{CH}_2\text{Cl}_2$  (0.17 mL, ~5.5 M solution in decane, 3.0 equiv.). The TBHP solution was added slowly to the reaction mixture over 30 min using a syringe pump. The crude reaction mixture was filtered through a short pad of silica to afford the mono-epoxide. The product was further purified by flash column chromatography on silica gel (hexanes:EtOAc = 90:10) to afford the mono-epoxide **4q** (87.9 mg, 62% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.65 – 7.61 (m, 2H), 7.50 – 7.45 (m, 2H), 7.41 – 7.36 (m, 3H), 7.35 – 7.30 (m, 3H), 7.21 (s, 1H), 4.33 (s, 2H), 3.20 (s, 1H), 0.98 (s, 6H), 0.96 (s, 6H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  135.80, 135.28, 130.47, 129.39, 128.41, 128.33, 128.16, 128.11, 126.50, 125.94, 85.01, 79.40, 61.30, 24.73, 24.67 (the quaternary vinyl C bearing the boron is not observed); IR (neat) 3454, 3061, 3030, 2979, 2930, 1605, 1495, 1447, 1373, 1261, 1110, 1029  $\text{cm}^{-1}$ ; HRMS  $m/z$  479.0996  $[(M+Na)^+]$ ; calcd for  $\text{C}_{23}\text{H}_{26}\text{BBrO}_4\text{Na}$ : 479.1005].

**General Procedure D: Synthesis of Epoxy-2-keto-anti-1,3-diols.** To a 20 mL vial was added B(pin)-substituted bis-epoxide and 1 mL THF. The solution was cooled at 0 °C and solid  $\text{NaBO}_3 \cdot \text{H}_2\text{O}$  (3 equiv.) was added followed by 1 mL of  $\text{H}_2\text{O}$ . The reaction mixture was stirred and allowed to warm to rt. Stirring was continued until TLC showed consumption of the bis-

epoxide (4–6 h). The reaction mixture was then diluted with water (1 mL) and extracted with diethyl ether (3 x 10 mL). The combined organic layer was then washed with brine, dried over  $\text{Na}_2\text{SO}_4$ , and the solvent was removed under reduced pressure. The crude product was purified by flash chromatography on silica gel (hexanes:EtOAc = 80:20).

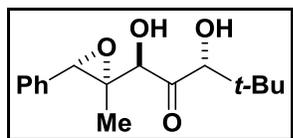
**General Procedure E: Synthesis of Epoxy-2-keto-*anti*-1,3-diols.** To a 20 mL vial was added B(pin)-substituted bis-epoxide and 2 mL THF. The solution was cooled at 0 °C and 30 %  $\text{H}_2\text{O}_2$  (3.3 equiv.) and NaOH (1.1 equiv.) were added to the solution. The reaction mixture was stirred and allowed to warm to rt. Stirring was continued until TLC showed consumption of the bis-epoxides (2–4 h). The reaction mixture was then diluted with water (1 mL) and extracted with diethyl ether (4 x 10 mL). The combined organic layer was then washed with brine, dried over  $\text{Na}_2\text{SO}_4$ , and the solvent was removed under reduced pressure. The crude was purified by flash chromatography on silica gel (hexanes:EtOAc = 80:20).



**1,3-Dihydroxyl-1-(2-methyl-3-phenyloxiran-2-yl)-3-phenylpropan-2-**

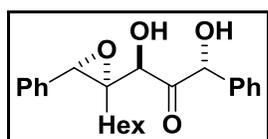
**one (5k).** The product was prepared by General Procedure D using bis-epoxide **2k** (16.3 mg, 0.04 mmol) and  $\text{NaBO}_3 \cdot \text{H}_2\text{O}$  (12.0 mg, 3 equiv., 0.12 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc = 80:20) to afford the 1,3-ketodiol **5k** (9.3 mg, 78% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.48 – 7.45 (m, 2H), 7.43 – 7.28 (m, 8H), 5.80 (s, 1H), 4.13 (s, 1H), 4.02 (s, 1H), 3.97 (s, 1H), 3.51 (s, 1H), 0.98 (s, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  209.7, 137.4, 134.5, 129.3, 129.1, 128.4, 128.3, 127.8, 126.7, 78.3, 77.8,

64.3, 63.1, 11.1; IR (neat) 3469, 3030, 2979, 2928, 2854, 1714, 1600, 1495, 1452, 1380, 1145  $\text{cm}^{-1}$ ; HRMS  $m/z$  321.1108  $[(M+Na)^+]$ ; calcd for  $\text{C}_{18}\text{H}_{18}\text{O}_4\text{Na}$ : 321.1103].



**1,3-Dihydroxy-4,4-dimethyl-1-(2-methyl-3-phenyloxiran-2-yl)pentan-**

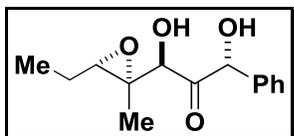
**2-one (5l).** The product was prepared by General Procedure D using bis-epoxide **2l** (23.7 mg, 0.06 mmol) and  $\text{NaBO}_3 \cdot \text{H}_2\text{O}$  (18.0 mg, 3 equiv., 0.18 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiol **5l** (13.2 mg, 78% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.53 (d,  $J = 7.2$  Hz, 2H), 7.39 (t,  $J = 7.1$  Hz, 2H), 7.34 (t,  $J = 7.2$  Hz, 1H), 5.17 (s, 1H), 3.79 (s, 1H), 3.66 (s, 1H), 3.59 (s, 1H), 2.92 (s, 1H), 1.09 (s, 9H), 1.02 (s, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  214.3, 139.0, 128.2, 128.1, 127.8, 85.1, 82.4, 76.8, 76.3, 35.2, 26.4, 18.8; IR (neat) 3427, 3062, 2979, 2928, 2855, 1712, 1600, 1480, 1409, 1380, 1304, 1144  $\text{cm}^{-1}$ ; HRMS  $m/z$  301.1411  $[(M+Na)^+]$ ; calcd for  $\text{C}_{16}\text{H}_{22}\text{O}_4\text{Na}$ : 301.1416].



**1-(2-Hexyl-3-phenyloxiran-2-yl)-1,3-dihydroxy-3-phenylpropan-2-one**

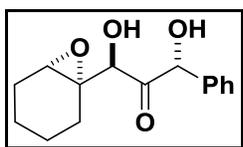
**(5m).** The product was prepared by General Procedure D using bis-epoxide **2m** (0.106 g, 0.22 mmol) and  $\text{NaBO}_3 \cdot \text{H}_2\text{O}$  (65.9 mg, 3 equiv., 0.66 mmol). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiol **5m** (51.9 mg, 64% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.47 – 7.43 (m, 2H), 7.42 – 7.31 (m, 6H), 7.26 – 7.21 (m, 2H), 5.82 (s, 1H), 4.10 (s, 1H), 4.05 (s, 1H), 3.87 (s, 1H), 3.39 (s, 1H), 1.60 – 1.51 (m, 1H), 1.43 – 1.32 (m, 1H), 1.22 – 1.14 (m, 2H), 1.13 – 1.06 (m, 6H), 0.82 (t,  $J = 7.2$

Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  210.2, 137.7, 134.6, 129.3, 129.1, 128.4, 128.3, 127.8, 126.7, 77.8, 77.6, 65.7, 64.6, 31.4, 29.5, 25.4, 24.8, 22.6, 14.2; IR (neat) 3460, 3064, 3033, 2956, 2926, 2856, 1715, 1600, 1495, 1455, 1379, 1263, 1012  $\text{cm}^{-1}$ ; HRMS  $m/z$  391.1888  $[(\text{M}+\text{Na})^+]$ ; calcd for  $\text{C}_{23}\text{H}_{28}\text{O}_4\text{Na}$ : 391.1885].



**1-(3-Ethyl-2-methyloxiran-2-yl)-1,3-dihydroxy-3-phenylpropan-2-**

**one (5o).** The product was prepared by General Procedure E using bis-epoxide **2o** (80.0 mg, 0.22 mmol), NaOH (0.24 mmol, 1.1 equiv., 60  $\mu\text{L}$ ) and 30%  $\text{H}_2\text{O}_2$  solution (0.73 mmol, 3.3 equiv., 22.5  $\mu\text{L}$ ). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiol **5o** (33.0 mg, 60% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.41 – 7.30 (m, 5H), 5.68 (d,  $J$  = 4.4 Hz, 1H), 4.13 (d,  $J$  = 6.2 Hz, 1H), 3.86 (d,  $J$  = 2.7 Hz, 1H), 3.34 (d,  $J$  = 3.8 Hz, 1H), 2.69 (t,  $J$  = 6.3 Hz, 1H), 1.62 – 1.47 (m, 2H), 1.15 (s, 3H), 1.02 (t,  $J$  = 7.5 Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  210.1, 137.5, 129.2, 128.9, 127.7, 78.8, 77.7, 65.1, 60.6, 21.9, 12.0, 10.5; IR (neat) 3460, 3064, 3033, 2955, 2926, 2856, 1715, 1603, 1495, 1455, 1379, 1263, 1012  $\text{cm}^{-1}$ ; HRMS  $m/z$  273.1092  $[(\text{M}+\text{Na})^+]$ ; calcd for  $\text{C}_{14}\text{H}_{18}\text{O}_4\text{Na}$ : 273.1103].

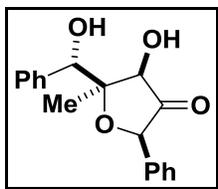


**1-(7-Oxabicycloheptan-1-yl)-1,3-dihydroxy-3-phenylpropan-2-one (5p).**

The product was prepared by General Procedure E using bis-epoxide **2p** (70.8 mg, 0.27 mmol), NaOH (0.30 mmol, 1.1 equiv., 60  $\mu\text{L}$ ) and 30%  $\text{H}_2\text{O}_2$  solution (0.89 mmol, 3.3 equiv., 27.4  $\mu\text{L}$ ).

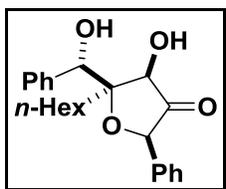
The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiols **5p** (43.2 mg, 61% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.42 – 7.31 (m, 5H), 5.68 (d,  $J$  = 5.6 Hz, 1H), 4.10 (d,  $J$  = 6.6 Hz, 1H), 3.87 (d,  $J$  = 3.5 Hz, 1H), 3.34 (d,  $J$  = 4.5 Hz, 1H), 2.95 (d,  $J$  = 2.7 Hz, 1H), 1.98 – 1.83 (m, 1H), 1.78 – 1.69 (m, 1H), 1.49 – 1.40 (m, 2H), 1.40 – 1.32 (m, 2H), 1.29 – 1.22 (m, 2H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  209.8, 137.4, 129.2, 129.0, 127.8, 78.1, (one peak overlaps with the  $\text{CDCl}_3$  peaks; two peaks are observed at 78.6, 78.0 in benzene- $d_6$ ) 60.0, 59.2, 24.6, 22.7, 20.1, 18.9; IR (neat) 3430, 2932, 2856, 1717, 1645, 1493, 1455, 1382, 1276, 1190, 1139  $\text{cm}^{-1}$ ; HRMS  $m/z$  285.1100 [(M+Na) $^+$ ; calcd for  $\text{C}_{15}\text{H}_{18}\text{O}_4\text{Na}$ : 285.1103].

**General Procedure F: Synthesis of Dihydroxy-dihydrofuran-3-(2H)-ones.** In a 20 mL vial was added the epoxide-substituted keto-*anti*-1,3-diol (1 equiv., 0.05M) followed by dry THF, and the solution was cooled to 0 °C. Either neat  $\text{BF}_3\cdot\text{OEt}_2$  or solid *p*-TsOH (1 equiv.) was added slowly to the solution. The reaction mixture was allowed to warm to rt and stirred at rt until TLC showed consumption of the epoxy keto diol (2–3 h). The reaction mixture was then diluted with water (1 mL) and extracted with diethyl ether(4 x 10 mL). The combined organic layer was then washed with brine, dried over  $\text{Na}_2\text{SO}_4$ , and the solvent was removed under reduced pressure. The crude was purified by flash chromatography on silica gel (hexanes:EtOAc = 80:20).



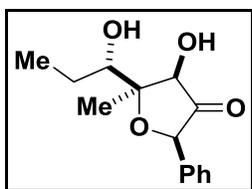
**4-Hydroxy-5-(hydroxyl(phenyl)methyl)-5-methyl-2-phenyldihydrofuran-3-(2H)-one (6k).** The product was prepared by General Procedure F using epoxide keto diol **5k**

(43.3 mg, 0.15 mmol) and  $\text{BF}_3 \cdot \text{OEt}_2$  (0.15 mmol, 19.0  $\mu\text{L}$ ). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiols **6k** (40.3 mg, 90% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.55 – 7.51 (m, 2H), 7.51 – 7.47 (m, 2H), 7.45 – 7.32 (m, 6H), 5.26 (s, 1H), 5.02 (s, 1H), 4.05 (s, 1H), 3.96 (s, 1H), 3.07 (s, 1H), 1.18 (s, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  211.4, 138.8, 136.2, 128.8, 128.7, 128.3, 128.3, 127.9, 127.0, 84.0, 79.9, 76.8, (one peak overlaps with the  $\text{CDCl}_3$  peaks), 19.3; IR (neat) 3437, 3064, 3033, 2930, 1764, 1603, 1495, 1453, 1073, 1054, 1028  $\text{cm}^{-1}$ ; HRMS  $m/z$  297.1136 [(M-H) $^-$ ]; calcd for  $\text{C}_{18}\text{H}_{17}\text{O}_4$ : 297.1127].



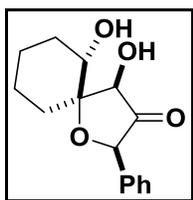
**5-Hexyl-4-hydroxy-5-(hydroxyl(phenyl)methyl)-2-phenyldihydrofuran-3-**

**2(H)-one (6m).** The product was prepared by General Procedure F using epoxide keto diol **5m** (70.0 mg, 0.19 mmol) and  $\text{BF}_3 \cdot \text{OEt}_2$  (0.19 mmol, 24.0  $\mu\text{L}$ ). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiols **6m** (63.7 mg, 91% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.58 – 7.53 (m, 2H), 7.51 – 7.46 (m, 2H), 7.45 – 7.39 (m, 2H), 7.39 – 7.30 (m, 4H), 5.22 (s, 1H), 5.00 (s, 1H), 4.79 (s, 1H), 4.10 (s, 1H), 3.45 (s, 1H), 1.49 – 1.37 (m, 2H), 1.37 – 1.12 (m, 8H), 0.88 (t,  $J = 7.0$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  211.0, 138.8, 136.4, 128.9, 128.7, 128.6, 128.4, 128.1, 127.1, 85.5, 80.0, 78.1, 76.1, 33.1, 31.8, 29.8, 22.9, 22.7, 14.2; IR (neat) 3401, 3064, 3033, 2954, 2929, 2857, 1764, 1602, 1495, 1453, 1055, 1027  $\text{cm}^{-1}$ ; HRMS  $m/z$  367.1927 [(M-H) $^-$ ]; calcd for  $\text{C}_{23}\text{H}_{27}\text{O}_4$ : 367.1909].



**4-Hydroxy-5-(1-hydroxypropyl)-5-methyl-phenyldihydrofuran-3(2H)-**

**one (6o).** The product was prepared by General Procedure F using epoxide keto diol **5o** (25.0 mg, 0.10 mmol) and *p*-TsOH (0.10 mmol, 19.4 mg). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiol **6o** (19.3 mg, 77% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.50 – 7.45 (m, 2H), 7.42 – 7.36 (m, 2H), 7.37 – 7.29 (m, 1H), 5.12 (s, 1H), 4.13 (s, 1H), 3.91 (s, 1H), 3.76 (dd,  $J = 10.6, 2.3$  Hz, 1H), 2.78 (s, 1H), 1.84 – 1.72 (m, 1H), 1.67 – 1.54 (m, 1H), 1.43 (s, 3H), 0.99 (t,  $J = 7.4$  Hz, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  211.3, 136.3, 128.8, 128.5, 126.6, 83.5, 78.8, 78.6, 77.6, 24.0, 20.2, 11.1; IR (neat) 3370, 3064, 3033, 2966, 2931, 2873, 1764, 1603, 1495, 1452, 1054  $\text{cm}^{-1}$ ; HRMS  $m/z$  273.1095 [(M+Na) $^+$ ]; calcd for  $\text{C}_{14}\text{H}_{18}\text{O}_4\text{Na}$ : 273.1103].



**4,6-Dihydroxy-2-phenyl-1-oxaspiro[4,5]decan-3-one (6p).** The product was

prepared by General Procedure F using epoxide keto diol **5p** (13.1 mg, 0.05 mmol) and *p*-TsOH (0.05 mmol, 9.7 mg). The crude product was purified by flash column chromatography on silica gel (hexanes:EtOAc =80:20) to afford the 1,3-ketodiol **6p** (8.5 mg, 65% yield).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.54 – 7.49 (m, 2H), 7.43 – 7.37 (m, 2H), 7.36 – 7.31 (m, 1H), 5.14 (s, 1H), 4.21 (s, 1H), 3.97 (dd,  $J = 8.9, 4.5$  Hz, 1H), 3.73 (s, 1H), 2.50 (s, 1H), 2.07 – 1.93 (m, 2H), 1.88 – 1.76 (m, 3H), 1.74 – 1.66 (m, 1H), 1.63 – 1.54 (m, 1H), 1.48 – 1.39 (m, 1H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  211.4, 136.5, 128.8, 128.4, 126.3, 83.3, 78.4, 78.3, (quaternary C is missing

or overlapping with the CDCl<sub>3</sub> peaks), 34.4, 30.6, 22.6, 22.4; IR (neat) 3402, 3065, 3033, 2934, 2863, 1767, 1603, 1495, 1455, 1157 cm<sup>-1</sup>; HRMS *m/z* 261.1139 [(M-H)<sup>-</sup>; calcd for C<sub>15</sub>H<sub>17</sub>O<sub>4</sub>: 261.1127].

## References:

1. H. C. Brown, N. G. Bhat and M. Srebnik, *Tetrahedron Lett.*, 1988, **29**, 2631-2634.
2. M. Kim and D. Lee, *Org. Lett.*, 2005, **7**, 1865.
3. H. C. Brown and J. A. Sinclair, *J. Organomet. Chem.*, 1977, **131**, 163-169.
4. M. W. Büttner, J. B. Nätscher, C. Burschka and R. Tacke, *Organometallics*, 2007, **26**, 4835-4838.
5. E. C. Hansen and D. Lee, *J. Am. Chem. Soc.*, 2005, **127**, 3252-3253.
6. M. M. Hussain, H. Li, N. Hussain, M. Ureña, P. J. Carroll and P. J. Walsh, *J. Am. Chem. Soc.*, 2009, **131**, 6516.
7. M. M. Hussain and P. J. Walsh, *Angew. Chem., Int. Ed.*, 2010, **49**, 1834-1837.
8. M. M. Hussain, J. H. Toribio, P. J. Carroll and P. J. Walsh, *Angew. Chem., Int. Ed.*, 2011, **50**, 6337.
9. J. Renaud, C.-D. Graf and L. Oberer, *Angew. Chem., Int. Ed. Engl.*, 2000, **39**, 3101.
10. A. Pelter, B. Singaram and J. W. Wilson, *Tetrahedron Lett.*, 1983, **24**, 635-636.
11. A. Pelter, D. Buss and A. Pitchford, *Tetrahedron Lett.*, 1985, **26**, 5093-5096.
12. A. R. Katritzky, Z. Wang, H. Lang and D. Feng, *J. Org. Chem.*, 1997, **62**, 4125-4130.
13. K. Kaneda, Y. Kawanishi, K. Jitsukawa and S. Teranishi, *Tetrahedron Lett.*, 1983, **24**, 5009-5010.
14. T. Hirao, *Chem. Rev.*, 1997, **97**, 2707-2724.