## Supporting Information

## A dinuclear iridium(III) complex as a visual specific phosphorescent probe for endogenous sulphite and bisulphite in living cells

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Fig. S1. Emission spetra of SP-1 (10  $\mu$ M) in the absence or presence of 100  $\mu$ M sulphite or bisulphite in a mixed solution of DMSO:HEPES buffer (3:7, 10 mM, pH 7.5),  $\lambda_{ex} = 405$  nm.



**Fig. S2.** Emission spectra of **SP-2** (10  $\mu$ M) treated with increasing concentrations of sulphite (0-30  $\mu$ M) in a mixed solution of DMSO:HEPES buffer (3:7, 10 mM, pH 7.5). Insert: the titration curve of **SP-2** reacted with sulphite. Each spectra was recorder under 405 nm excitation.



**Fig. S3.** Calibration curve of **SP-2** signal intensity *vs* sulphite (hollow square) or bisulphite (solid square) concentrations.

compounds	σ	М	$\mathbb{R}^2$	S/N	DL
sulphite	1.365	$1.740 \times 10^7 \mathrm{M}^{-1}$	0.991	3	$2.35 \times 10^{-7} \mathrm{M}$
bisulphite	1.365	$2.847 \times 10^7  \text{M}^{-1}$	0.999	3	$1.44 \times 10^{-7} \mathrm{M}$

Table S1. Calculation of the detection limit of sulphite and bisulphite with SP-2.



**Fig. S4.** Determination of the observed rate constants for reaction of **SP-2** and bisulphite (solid square) and sulphite (hollow square). 10  $\mu$ M **SP-2** reacted with 10  $\mu$ M bisulphite or sulphite in a mixed solution of DMSO:HEPES buffer (3:7, 10 mM, pH 7.5). And the emission intensity was recorded over time,  $\lambda_{ex} = 405$  nm,  $\lambda_{em} = 600$  nm.



Fig. S5. The pH effect experiments of SP-2. Phosphorescence intensity of 10.0  $\mu$ M SP-2 at 600 nm in the absence (black square) and presence of sulphite (red square, 10.0  $\mu$ M) under different pH in Britton-Robinson buffer.



Fig. S6. Potential cytotoxicity of SP-2 was evaluated by MTT assay. Tecan Infinite M200 monochromator-based multifunction microplate reader was used to measure the OD490 of each well. The cell survival rate in the control wells without SP-2 solutions was considered as 100% cell survival. The data represent the mean  $\pm$  SD of three independent experiments.



Fig. S7. HepG2 cells were incubated for 1 h with increasing concentration of bisulphite (0, 10, 50 and 250  $\mu$ M), and then replaced with PBS buffer and incubated with 5.0  $\mu$ M SP-2 for another 1 h. Confocal luminescence images were captured under excitation of 405 nm laser. Scale bar represents 50  $\mu$ m.



**Fig. S8** HepG2 cells were incubated for 1 h with: (a) control, (b) 250  $\mu$ M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, (c) 500  $\mu$ M GSH, (d) 250  $\mu$ M CN<sup>-</sup>, (e) 250  $\mu$ M SCN<sup>-</sup> and (f) 100 mM TNBS, and then replaced with PBS buffer and incubated with 5.0  $\mu$ M **SP-2** for another 1 h. Confocal luminescence images were captured under excitation of 405 nm laser. Scale bar represents 50  $\mu$ m. Line 1, 3 represent the fluorescent field, and line 2, 4 represent the bright field.



**Fig. S9.** Cyclic voltammograms for 100  $\mu$ M **SP-2** in the absence (black line) or presence of 500  $\mu$ M sulphite (red line) in 0.1 M tetrabutylammonium perchlorate (TABP) CH<sub>3</sub>CN solutions containing 5% HEPES buffer. Scan rate was 0.1 V/s.



**Fig. S10.** ESI-MS identification of **SP-2** (a) in CH<sub>3</sub>OH and its reacted product with sulphite (b).



**Fig. S11.** FT-IR spectra of **SP-2** (black line) and its reacted product with sulphite (red line).



**Fig. S12.** <sup>1</sup>H NMR of **SP-2** (black line) and its reacted product with sulphite (red line) in CD<sub>3</sub>CN, 400 MHz.



Fig. S13. <sup>13</sup>C-<sup>1</sup>H COSY spectrum of the isolated product in CD<sub>3</sub>CN, 400 MHz.



Fig. S14. <sup>1</sup>H NMR spectra of SP-2 reacted with different concentration of sulphite in 4/1 CD<sub>3</sub>CN/D<sub>2</sub>O, 400 MHz.