

Supporting Information

Molybdenum-Molybdenum Quadruple Bonds Supported by 9,10-Anthraquinone Carboxylate Ligands. Molecular and Electronic Structures and Ground State and Photoexcited State Redox Properties

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Table S1. Crystallographic information for compound **II**

Figure S1. Emission spectra of Mo₂(DAniF₃)(O₂C-2-ThCC) and compound **II** in 2-MTHF at 77K.

Figure S2. Absorption spectra of compound **II** in toluene, CH₂Cl₂ and THF

Figure S3. ns TA spectra and associated kinetic trace of compound **I** in THF, $\lambda_{\text{exc}} = 532$ nm.

Figure S4. fs TA spectra of **II** in toluene, $\lambda_{\text{exc}} = 675$ nm

Figure S5. fs TA spectra of **II** in benzene, $\lambda_{\text{exc}} = 675$ nm

Figure S6. ns TA spectra and associated kinetic trace of compound **II** in benzene, $\lambda_{\text{exc}} = 355$ nm.

Figure S7. Kinetic traces for fs TA measurements for compounds **I** and **II**.

Figure S8. Ground-state infrared spectra of compounds **I** and **II** in THF at room temperature.

Figure S9. fs TRIR spectra of compound **II** in THF, $\lambda_{\text{exc}} = 400$ nm.

Figure 10. Comparison of long-lived features in TRIR spectra of compound **II** and Mo₂(DAniF₃)(O₂C-C≡C-C₆H₅)

Figure S11. fs TRIR spectra of compound **II** in benzene, $\lambda_{\text{exc}} = 400$ nm.

Figure S12. Kinetic traces for fs TRIR measurements for compounds **I** and **II**.

Table 1. Data Collection Parameters for **II**.

Compound	II
Chemical Formula	C ₆₀ H ₅₂ Mo ₂ N ₆ O ₁₀
Formula Weight	1208.96
Temperature (K)	150(2)
Space Group	Triclinic, P-1
<i>a</i> (Å)	9.8000(2)
<i>b</i> (Å)	12.6442(2)
<i>c</i> (Å)	23.2653(3)
α (°)	100.197(1)
β (°)	100.351(1)
γ (°)	100.798(1)
<i>V</i> (Å ³)	2718.72(8)
<i>Z</i>	2
<i>D</i> _{calcd} (Mg/m ³)	1.477
Crystal Size (mm)	0.12 X 0.15 X 0.38
Theta range for data collection	1.68 to 27.47°
μ , (Mo, K α) (mm ⁻¹)	0.527
<i>F</i> (000)	1236
Reflections collected	75386
Unique reflections	12452 [R(int)= 0.052]
Completeness to theta max	99.9%
Data/restraints/parameters	12452 / 0 / 703
R1 ^a (%) (all data)	4.49 (8.43)
wR2 ^b (%)(all data)	10.06 (11.08)
Goodness-of-fit on <i>F</i> ²	1.022
Largest diff. peak and hole (e Å ⁻³)	1.253 and -0.402

$$^aR1 = \Sigma ||F_o|-|F_c|| / \Sigma |F_o| \times 100$$

$$^b wR2 = [\Sigma w (F_o^2 - F_c^2)^2 / \Sigma (w |F_o|^2)^2]^{1/2} \times 100$$

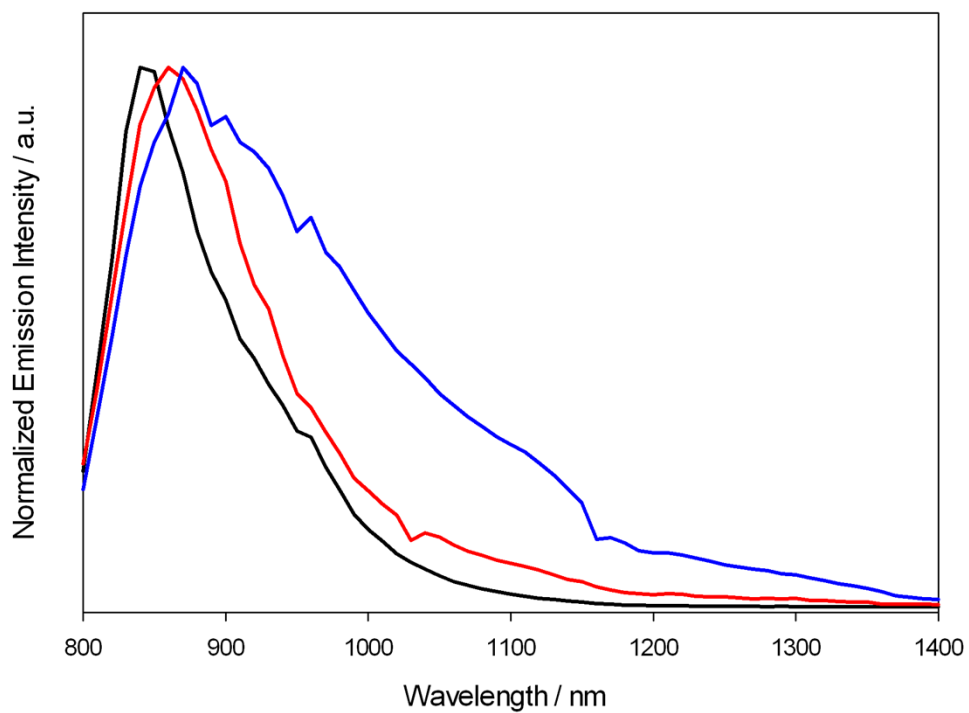


Figure S1. Emission spectra of $\text{Mo}_2(\text{DAniF}_3)(\text{O}_2\text{C-2-ThCC})$ (black, $\lambda_{\text{exc}} = 405 \text{ nm}$) and compound **II** (red, $\lambda_{\text{exc}} = 405 \text{ nm}$), (blue, $\lambda_{\text{exc}} = 658$) in 2-MTHF at 77 K.

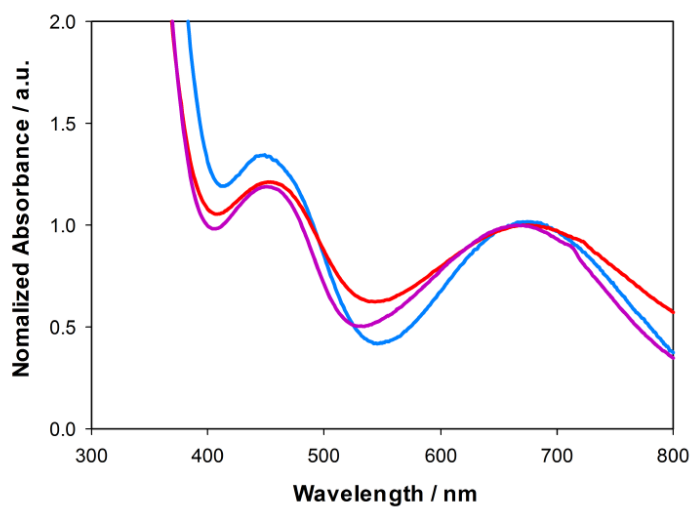


Figure S2. Absorption spectra of compound **II** in benzene (purple), CH_2Cl_2 (red) and THF (blue) at room temperature.

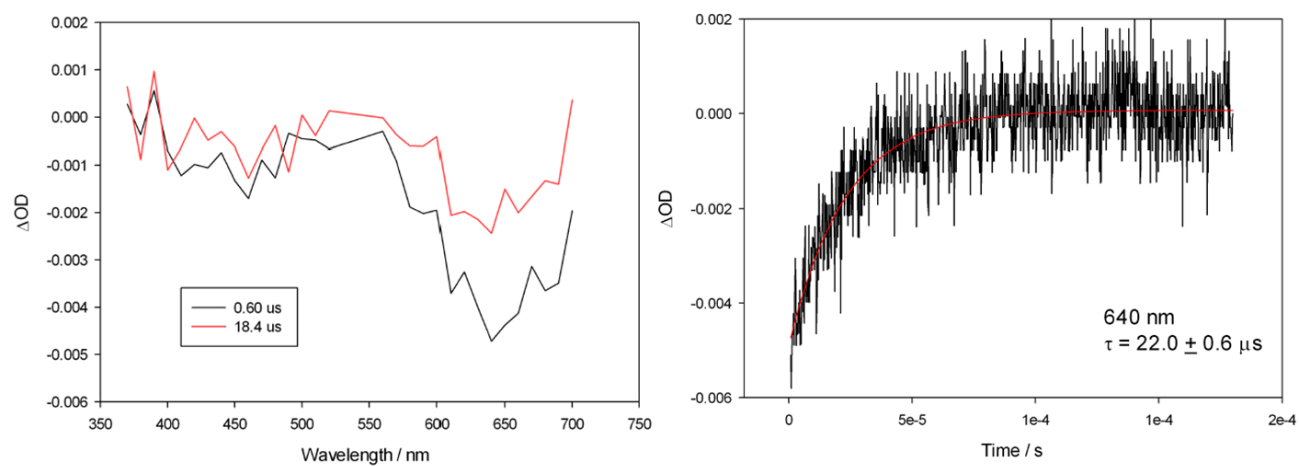


Figure S3. ns TA spectra and associated kinetic trace of compound **I** in THF, $\lambda_{exc} = 532$ nm.

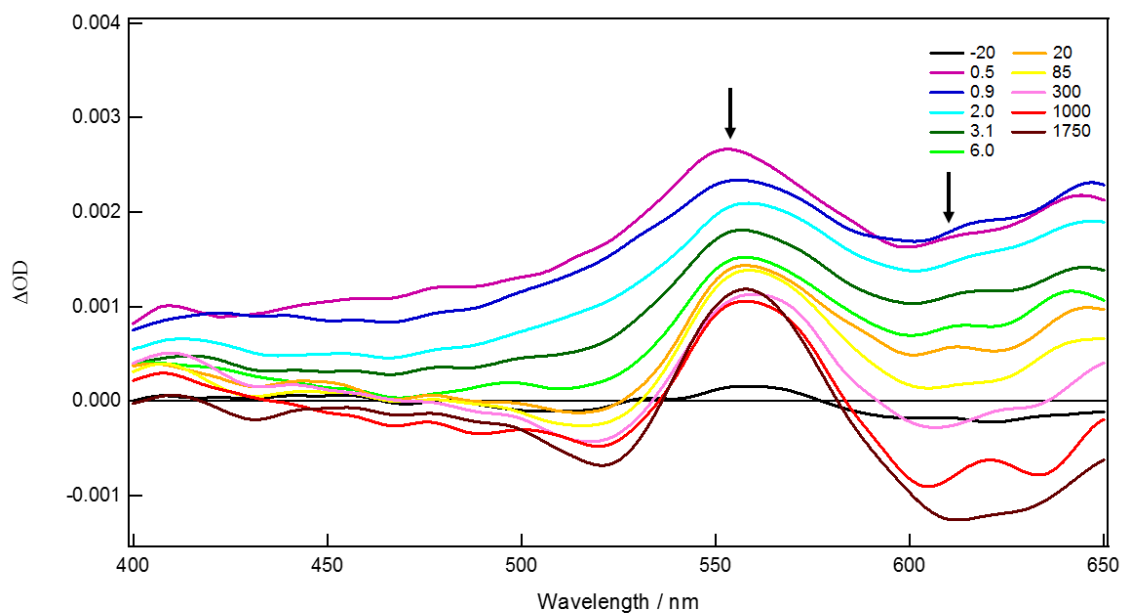


Figure S4. fs TA spectra of **II** in THF, $\lambda_{exc} = 350$ nm

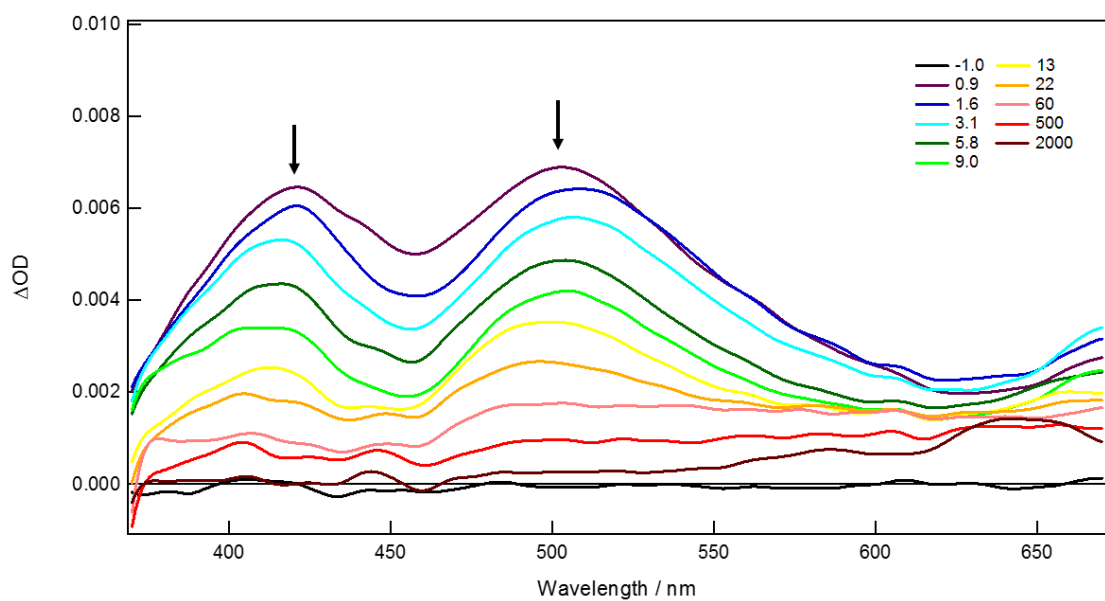


Figure S5. fs TA spectra of **II** in benzene, $\lambda_{\text{exc}} = 350 \text{ nm}$

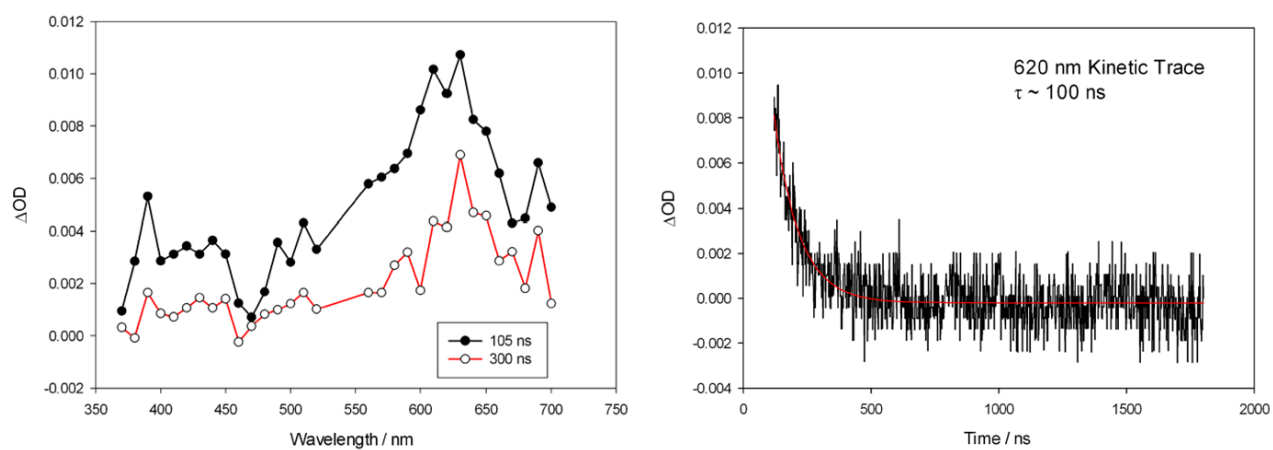


Figure S6. ns TA spectra and associated kinetic trace of compound **II** in benzene, $\lambda_{\text{exc}} = 355 \text{ nm}$

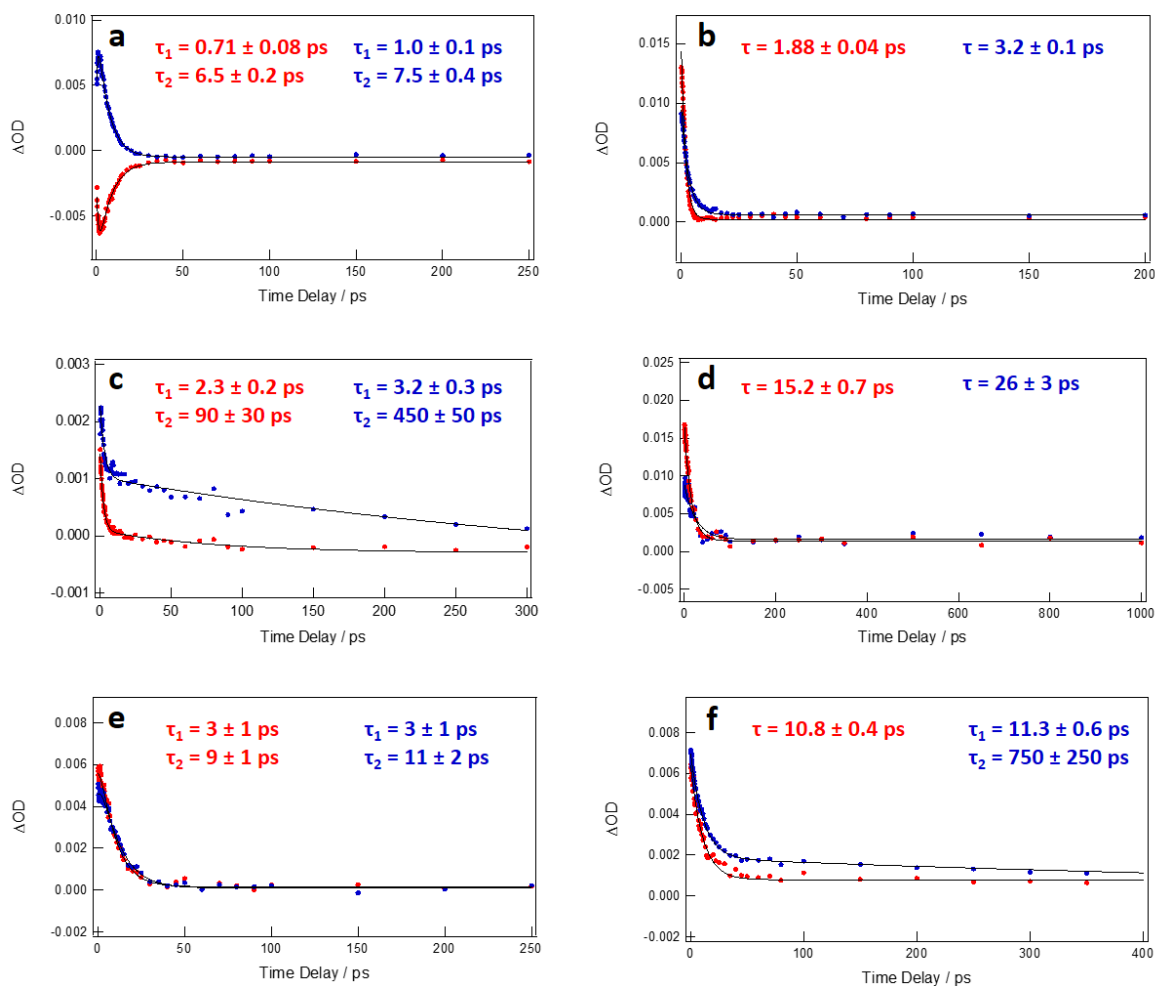


Figure S7. Kinetic traces for fs TA measurements: (a) Compound **I** in THF, λ_{exc} 675 nm, (b) Compound **II** in THF, λ_{exc} = 675 nm, (c) Compound **II** in THF, λ_{exc} = 350, (d) Compound **II** in toluene, λ_{exc} = 675, (e) and (f) Compound **II** in benzene, λ_{exc} = 675 and 350 nm respectively.

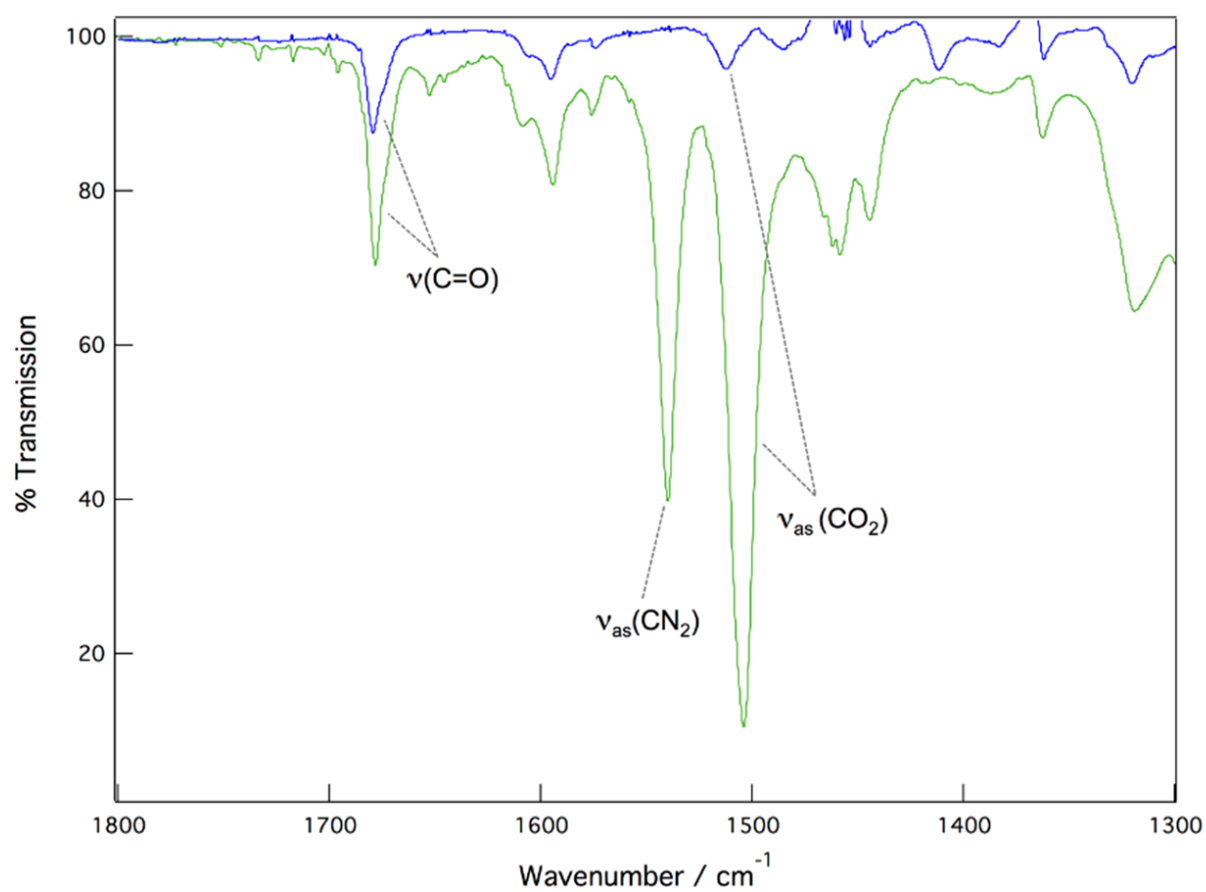


Figure S8. Ground-state infrared spectra of compound **I** (blue) and **II** (green) in THF at room temperature.

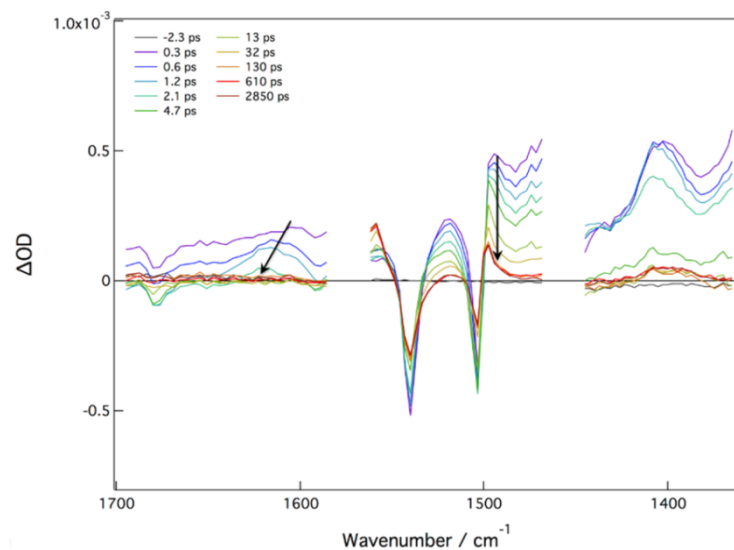


Figure S9. fs TRIR spectra of compound **II** in THF, $\lambda_{\text{exc}} = 400$ nm.

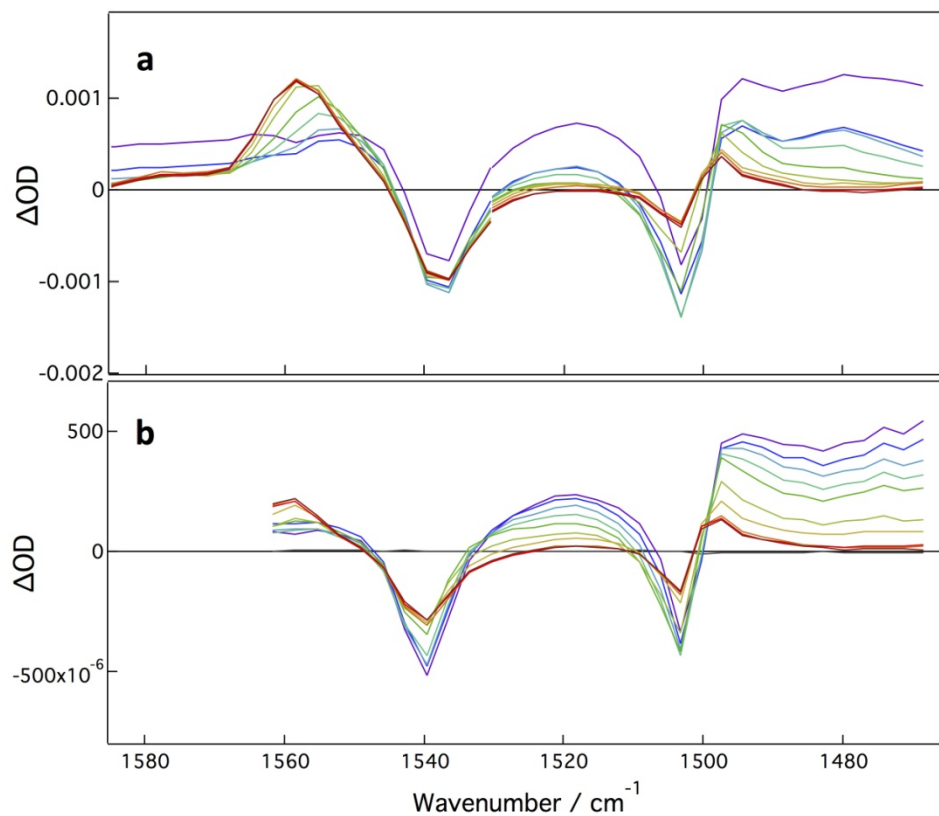


Figure S10. Comparison of long-lived features (orange trace) in TRIR spectra of $\text{Mo}_2(\text{DAniF})_3$ ($\text{O}_2\text{C}-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4$) (a) and compound **II** (b).

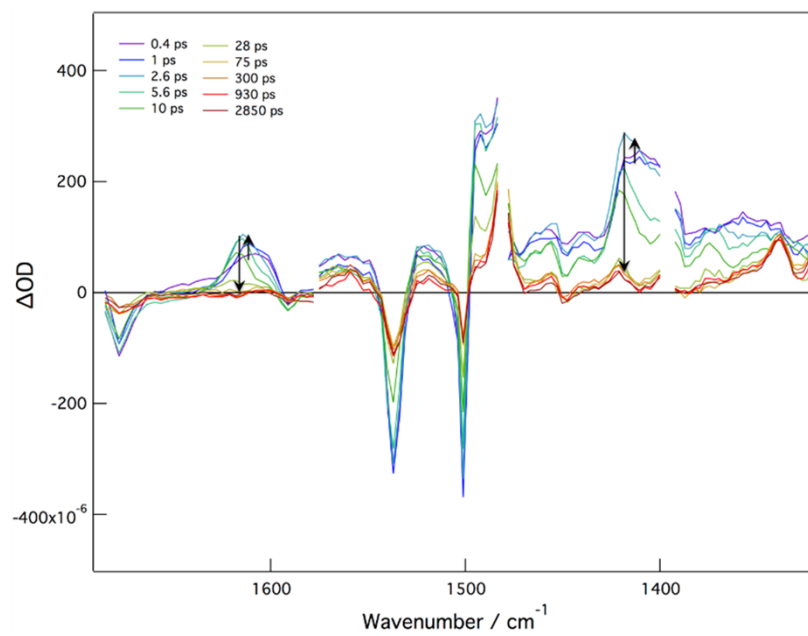


Figure S11. fs TRIR spectra of compound **II** in benzene, $\lambda_{\text{exc}} = 400$ nm.

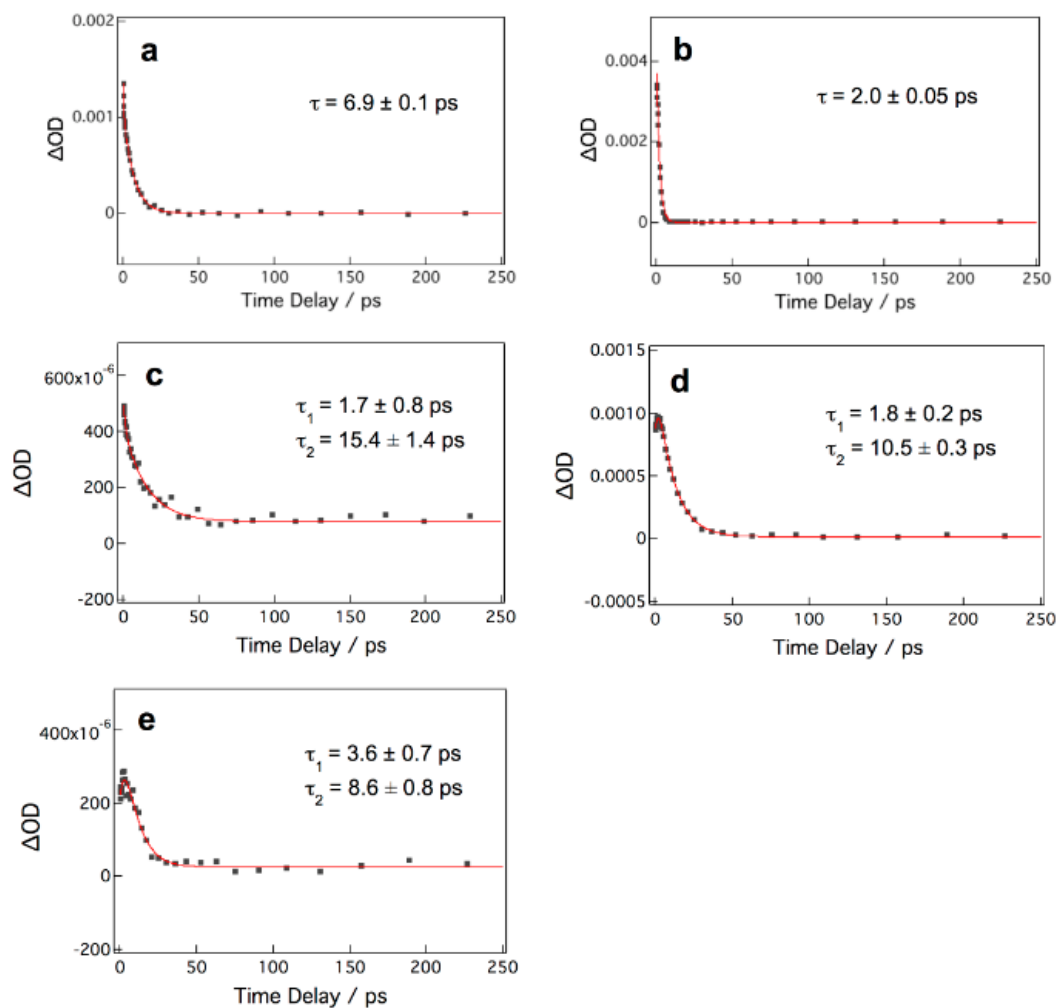


Figure S12. Kinetic traces for fs TRIR measurements: (a) Compound **I** in THF, λ_{exc} 675 nm, (b) and (c) Compound **II** in THF, λ_{exc} = 675 and 400 nm respectively, (d) and (e) Compound **II** in benzene, λ_{exc} = 675 and 400 nm respectively.