

Supplementary Information

Sensitization of Wide Band Gap Photocatalysts to Visible Light by a Molten CuCl treatment

Katsuya Iwashina,^a Akihide Iwase^{ab} and Akihiko Kudo^{*ab}

^a Department of Applied Chemistry, Faculty of Science, Tokyo University of Science, 1-3
Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan,

^b Photocatalysis International Research Center, Research Institute for Science and Technology,
Tokyo University of Science, 2641 Noda-shi, Yamazaki, Chiba-ken 278-8510, Japan.

E-mail: a-kudo@rs.kagu.tus.ac.jp

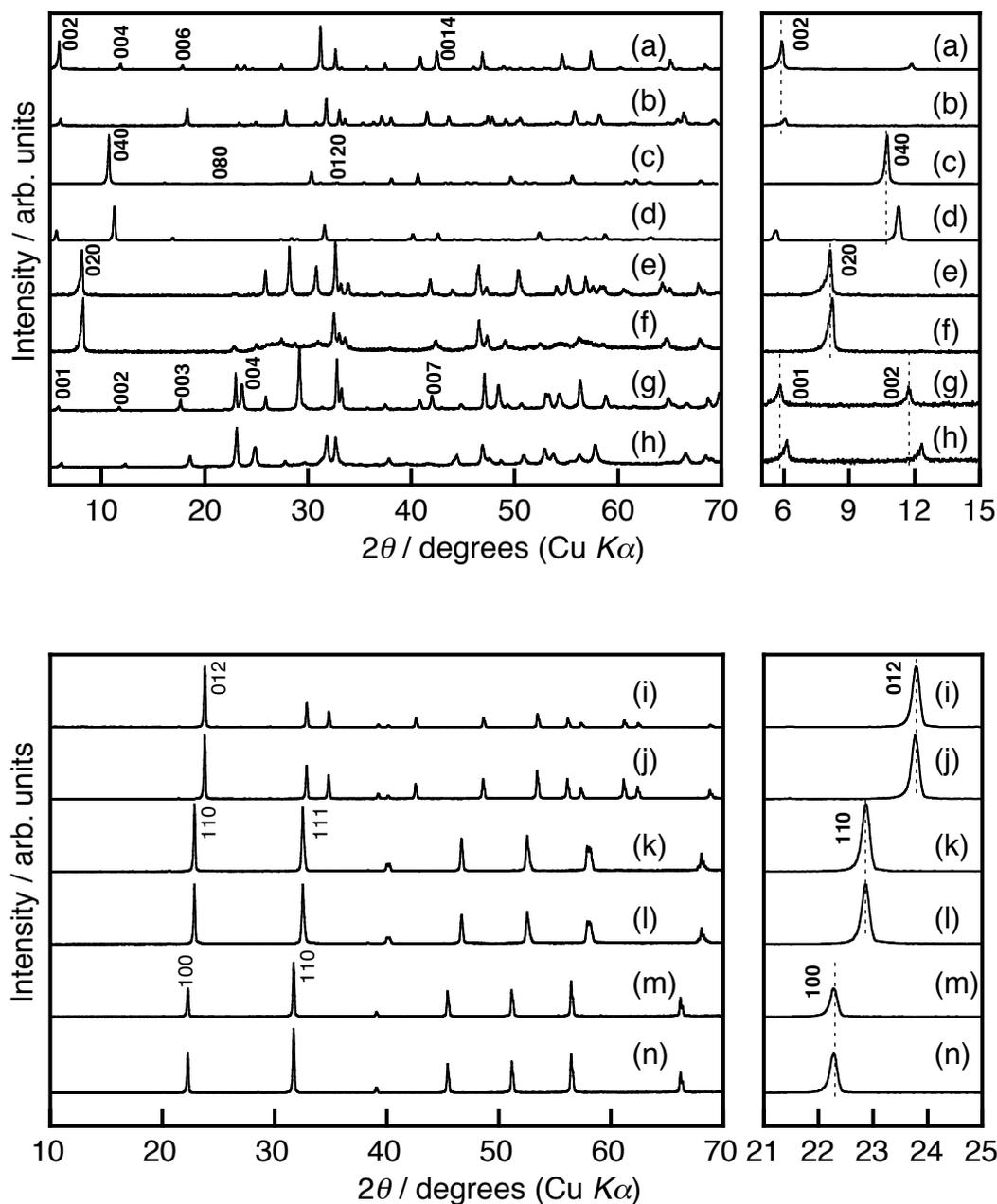


Figure S1. X-ray diffraction patterns of (a) $K_2La_2Ti_3O_{10}$, (b) $Cu(I)-K_2La_2Ti_3O_{10}$, (c) $K_4Nb_6O_{17}$, (d) $Cu(I)-K_4Nb_6O_{17}$, (e) $KLaNb_2O_7$, (f) $Cu(I)-KLaNb_2O_7$, (g) $RbCa_2Ta_3O_{10}$, (h) $Cu(I)-RbCa_2Ta_3O_{10}$, (i) $LiTaO_3$, (j) $Cu(I)-LiTaO_3$, (k) $NaTaO_3$, (l) $Cu(I)-NaTaO_3$, (m) $KTaO_3$ and (n) $Cu(I)-KTaO_3$.

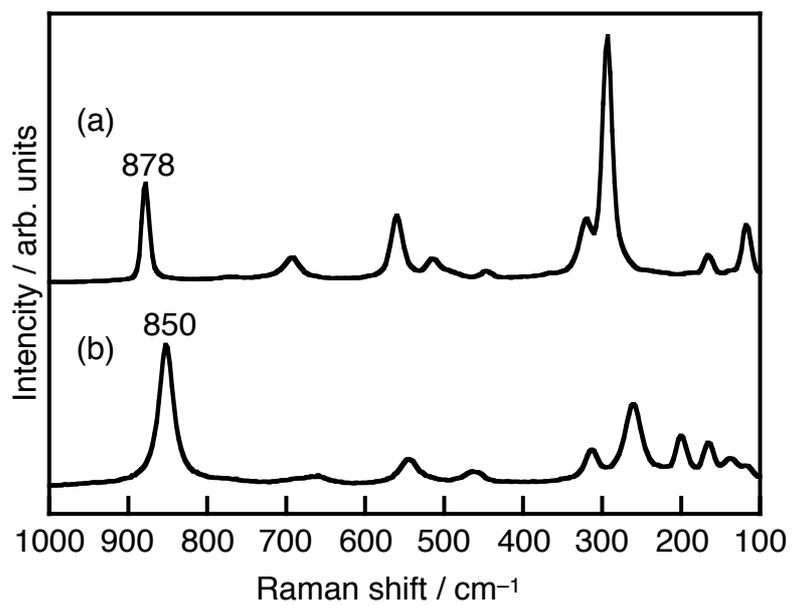


Figure S2. Raman spectra of (a) $\text{K}_2\text{La}_2\text{Ti}_3\text{O}_{10}$ and (b) $\text{Cu(I)-K}_2\text{La}_2\text{Ti}_3\text{O}_{10}$. Excitation wavelength: 532 nm.

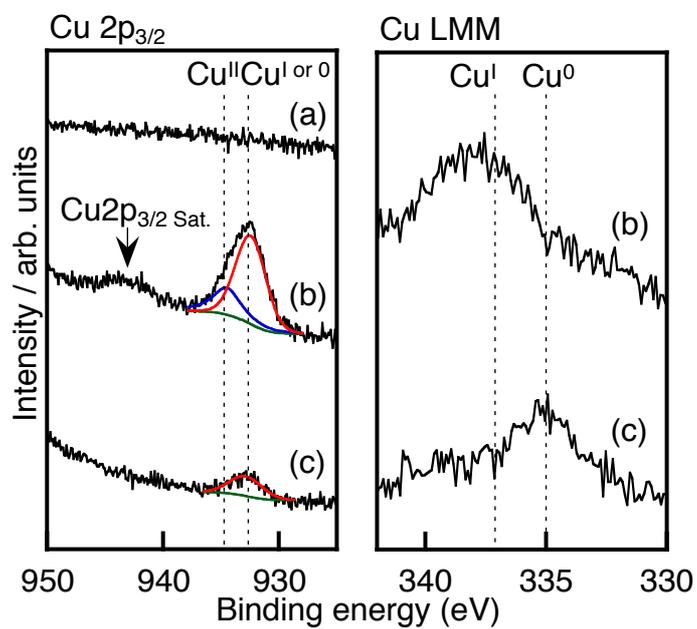


Figure S3. X-ray photoelectron and Auger spectra of (a) LiTaO₃, (b) Cu(I)-LiTaO₃ and (c) 60s Ar-etched Cu(I)-LiTaO₃. Assigned by ref. 20.

Determination of the Cu(I)-exchange ratio

ICP-AES

The calibration curves were obtained using potassium, lanthanum, titanium and copper standard solutions for ICP-AES analysis (Wako Pure Chemical). After mixing a standard solution of germanium as internal standard to correct flame intensity and the each standard solution, the mixture solutions were diluted to 100 mL with distilled water for the ICP-AES analysis. The emission wavelengths used for determinate quantity were 769.896 nm for K, 379.487 nm for La, 323.452 nm for Ti, 213.598 nm for Cu and 209.426 nm for Ge.

Figure S4 shows the calibration curves for K, La, Ti and Cu. The y-axis represents the intensity of K, La, Ti or Cu normalized by an internal standard of Ge. All calibration curves had correlation coefficients larger than 0.999.

0.02 g of $K_2La_2Ti_3O_{10}$ and $Cu(I)-K_2La_2Ti_3O_{10}$ were completely dissolved in ca. 0.2 mL of an aqua regia. The solutions were diluted to 100 mL with distilled water for the ICP-AES analysis, after adding 0.5 mL of germanium standard solution (1000 ppm).

Table S1 shows the signal intensity and calculated molar number of $K_2La_2Ti_3O_{10}$ and $Cu(I)-K_2La_2Ti_3O_{10}$. No K^+ ions were observed in $Cu(I)-K_2La_2Ti_3O_{10}$, indicating 100% of Cu(I)-exchange ratio.

XRF

Photocatalyst of $K_4Nb_6O_{17}$, $KLaNb_2O_7$, $RbCa_2Ta_3O_{10}$, $LiTaO_3$ and $NaTaO_3$ were mixed with CuO in the ratio of alkaline:Cu = 1:0, 1:0.25, 1:0.5, 1:0.75 and 1:1 to prepare calibration curves. The used emission X-ray source was Ge. The energy of X-ray used for determinate quantity was 8.094 keV for Cu $K\beta$, 16.581 keV for Nb $K\alpha$ and 9.342 keV for Ta $L\beta$.

Figure S5 shows the calibration curves of $Cu(I)-K_4Nb_6O_{17}$, $Cu(I)-KLaNb_2O_7$, $Cu(I)-RbCa_2Ta_3O_{10}$, $Cu(I)-LiTaO_3$ and $Cu(I)-NaTaO_3$. Calibration curves had correlation coefficients larger than 0.9980.

Table S2 shows the signal intensity, calculated ratio of Cu to either Nb or Ta and Cu(I)-exchange ratio of $Cu(I)-K_4Nb_6O_{17}$, $Cu(I)-KLaNb_2O_7$, $Cu(I)-RbCa_2Ta_3O_{10}$, $Cu(I)-LiTaO_3$ and $Cu(I)-NaTaO_3$. Ratio of Cu to Nb or Ta of $Cu(I)-K_4Nb_6O_{17}$, $Cu(I)-KLaNb_2O_7$, $Cu(I)-RbCa_2Ta_3O_{10}$, $Cu(I)-LiTaO_3$ and $Cu(I)-NaTaO_3$ are 46.0, 32.3, 30.0, 2.2 and 8.9%, respectively. Cu(I)-exchange ratio of $Cu(I)-K_4Nb_6O_{17}$, $Cu(I)-KLaNb_2O_7$, $Cu(I)-RbCa_2Ta_3O_{10}$, $Cu(I)-LiTaO_3$ and $Cu(I)-NaTaO_3$ can be calculated to be 69, 65, 90, 2 and 9%, respectively.

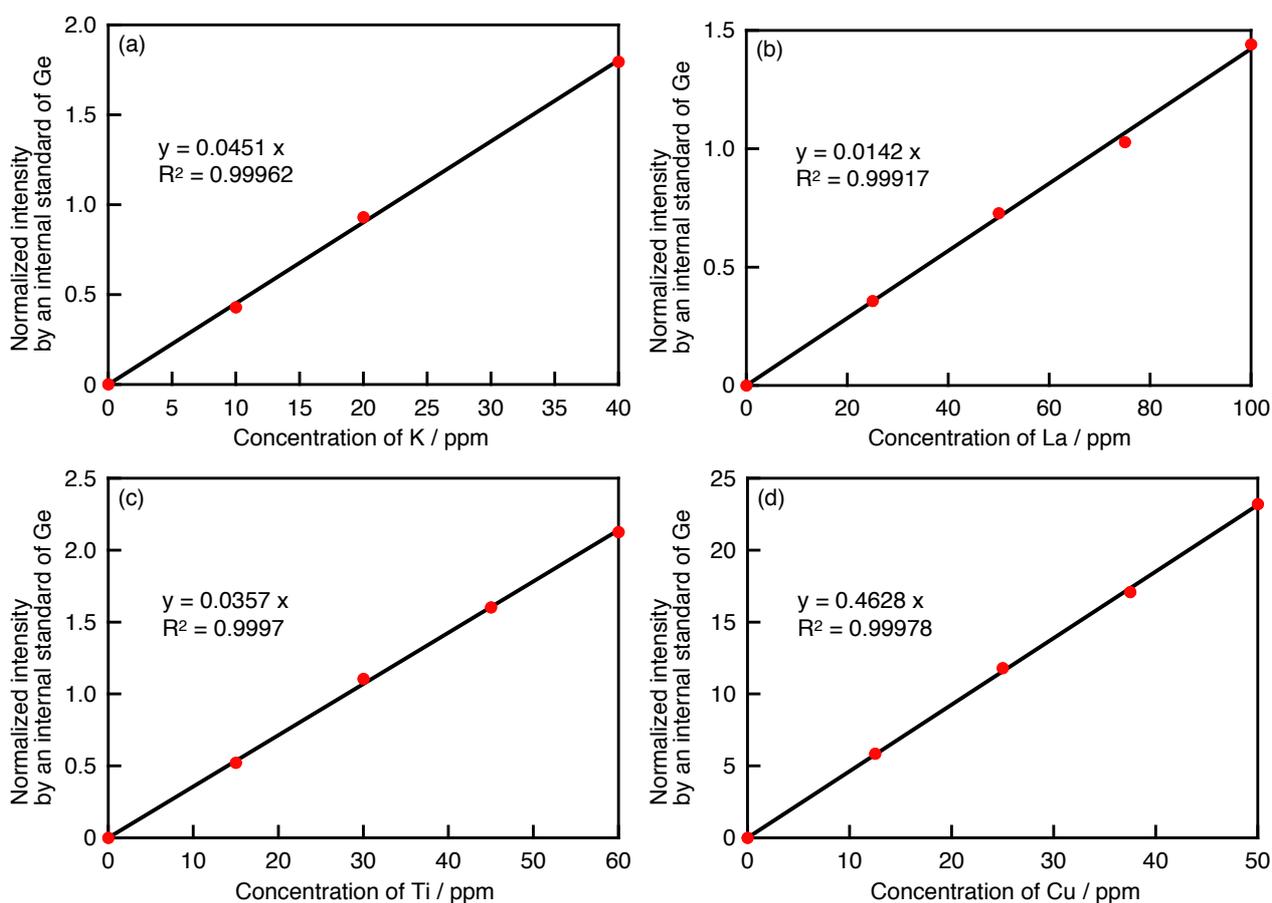


Figure S4. Calibration curves of (a) K, (b) La, (c) Ti and (d) Cu in ICP-AES analysis.

Table S1. ICP signal intensity and calculated element contents of $K_2La_2Ti_3O_{10}$ and Cu(I)- $K_2La_2Ti_3O_{10}$

Element	$K_2La_2Ti_3O_{10}$		Cu(I)- $K_2La_2Ti_3O_{10}$	
	Signal intensity	Calculated molar number/ μ mol	Signal intensity	Calculated molar number/ μ mol
K	45301	58.9	0	0
La	48992	57.0	46503	54.6
Ti	65504	87.3	61022	82.2
Cu	5796	0.07	606204	47.4
Ge	43572	–	43116	–

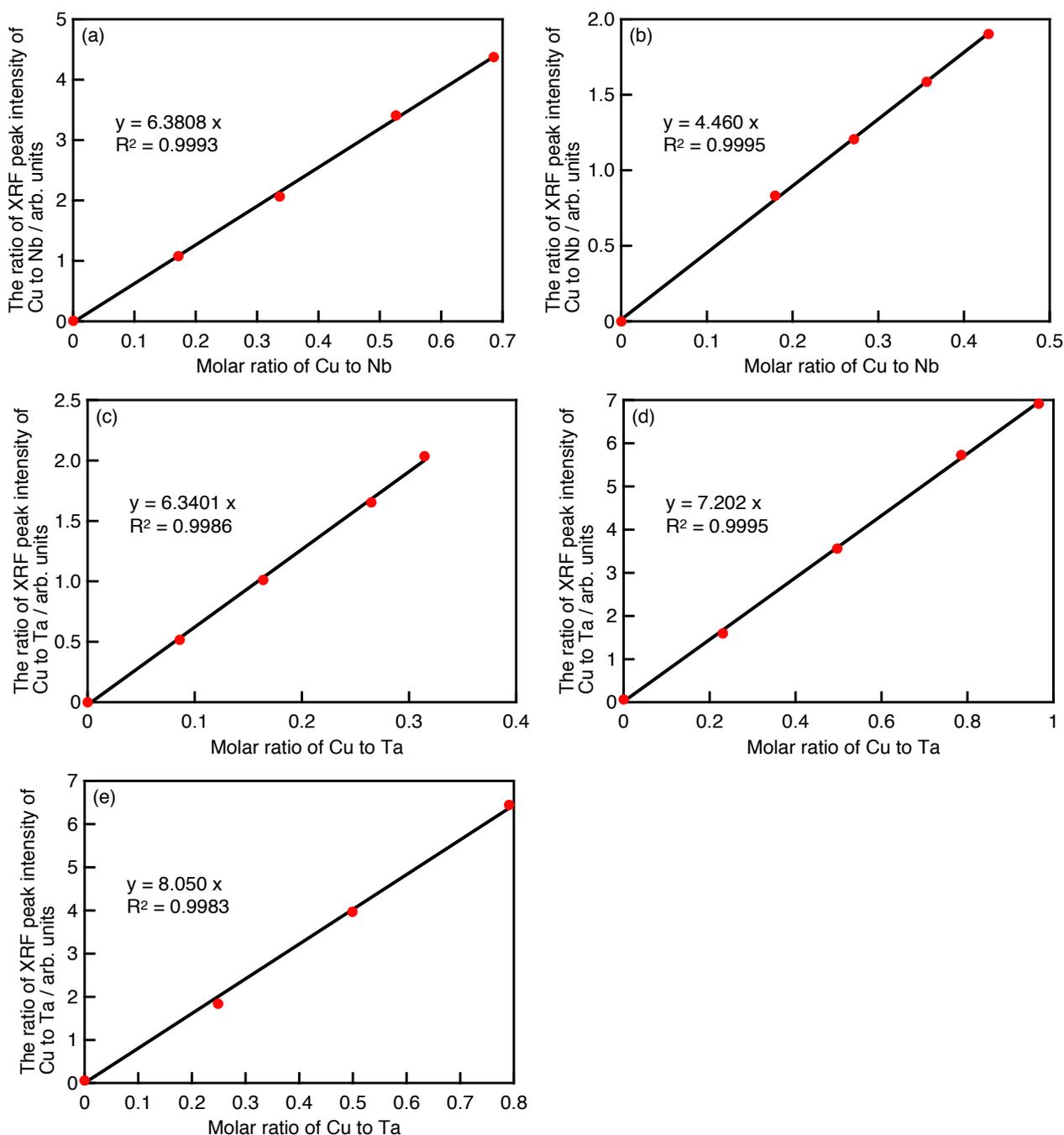


Figure S5. Calibration curves of (a) Cu(I)-K₄Nb₆O₁₇, (b) Cu(I)-KLaNb₂O₇, (c) Cu(I)-RbCa₂Ta₃O₁₀, (d) Cu(I)-LiTaO₃ and (e) Cu(I)-NaTaO₃ in XRF analysis.

Table S2. XRF signal intensity, calculated ratio of Cu to either Nb or Ta and Cu(I)-exchange ratio of Cu(I)-K₄Nb₆O₁₇, Cu(I)-KLaNb₂O₇, Cu(I)-RbCa₂Ta₃O₁₀, Cu(I)-LiTaO₃ and Cu(I)-NaTaO₃

Photocatalyst	XRF peak intensity			Ratio of Cu to either Nb or Ta %	Cu(I)-exchange ratio %
	Cu	Nb	Ta		
Cu(I)-K ₄ Nb ₆ O ₁₇	28.03	9.54	–	46.0	69
Cu(I)-KLaNb ₂ O ₇	13.03	9.04	–	32.3	65
Cu(I)-RbCa ₂ Ta ₃ O ₁₀	10.17	–	5.15	30.0	90
Cu(I)-LiTaO ₃	1.17	–	7.31	2.2	2
Cu(I)-NaTaO ₃	4.17	–	5.81	8.9	9