

Supplementary Information

Hierarchical Nanoscale Multi-shell Au/CeO₂ Hollow Spheres

Pengfei, Xu, Ranbo Yu^{}, Hao Ren, Lingbo Zong, Jun Chen, Xianran Xing*

Department of Physical Chemistry, School of Metallurgical and Ecological Engineering,
University of Science and Technology Beijing, 100083, China

*Corresponding authors: ranboyu@ustb.edu.cn

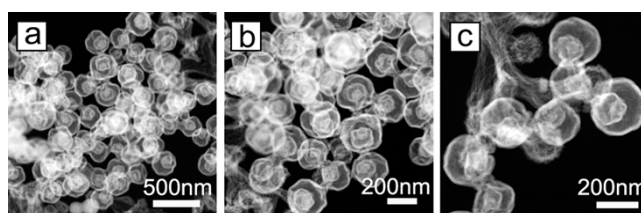


Fig. S1 HAADF STEM images of triple-shelled ceria hollow spheres. The calcination condition: at heat ramp rate of $10^{\circ}\text{C min}^{-1}$ with 60 min holding 500°C .

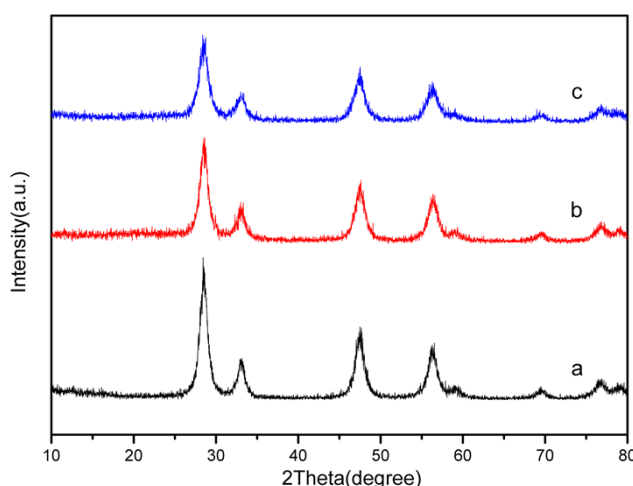


Fig. S2 XRD patterns of a) single-shell, b) double-shell and c) triple-shelled ceria hollow spheres.

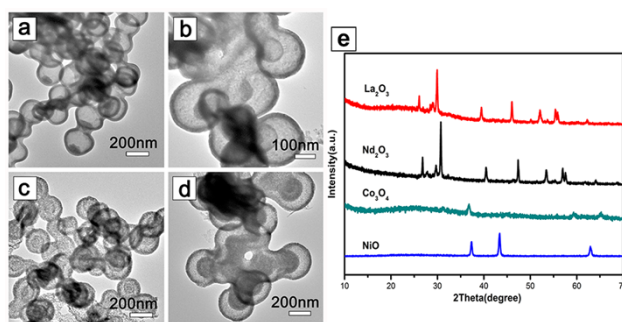


Fig. S3 a-d) Representative TEM images of various oxide multi-shelled hollow spheres: a) La_2O_3 ; b) Nd_2O_3 ; c) Co_3O_4 ; d) NiO . e) XRD patterns of the corresponding metal oxides.

When we substitute other metal ions (La^{3+} , Nd^{3+} , Co^{2+} , Ni^{2+}) for Ce^{3+} during hydrothermal process, corresponding multi-shelled metal oxides hollow spheres (La_2O_3 , Nd_2O_3 , Co_3O_4 , NiO) were generated successfully as shown in Figure S3. These results confirmed the generality of our approach for hydrothermally enhanced synthesis of multi-shelled metal oxides hollow spheres.

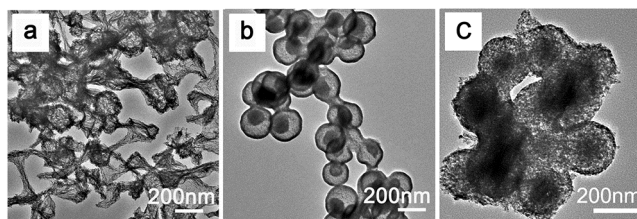


Fig. S4 TEM images of ceria obtained under different hydrothermal temperature when maintaining the same calcination condition. a) 120°C, 6h; b) 160°C, 6h; c) 180°C, 6h. The calcination condition: at heat ramp rate of 2°C min⁻¹ with 60 min holding 500°C.

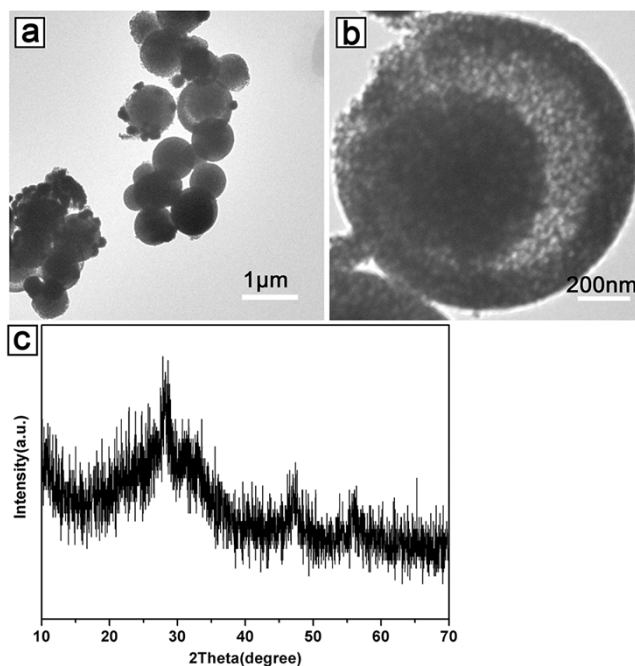


Fig. S5 a, b) TEM images and c) XRD patterns of ceria precursors obtained under the hydrothermal condition of 180°C, 3h.

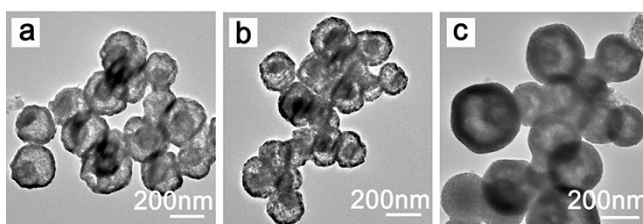


Fig. S6 TEM images of ceria obtained under the same calcination condition after the different hydrothermal time. a) 6h; b) 12h; c) 24h. The calcination condition: at heat ramp rate of 2°C min⁻¹ with 60 min holding 500°C.

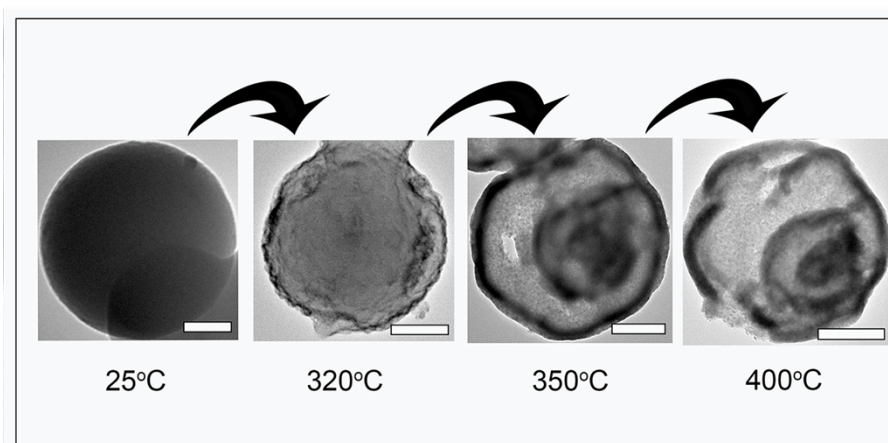


Fig. S7 The evolution of triple-shelled ceria by track of calcination process between 25°C and 400°C. The calcination condition: the heat ramp rate was set at 10°C min⁻¹. The scale bars: 50nm.

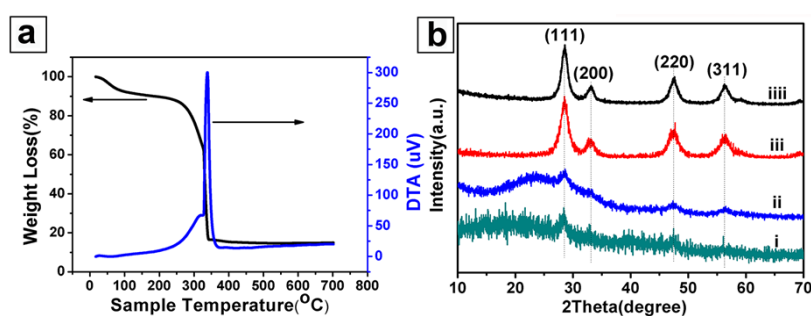


Fig. S8. a) TG-DTA curves of ACMSs; b) XRD patterns of immediate products obtained at different calcinations stage: i) before calcinations; ii) 320°C, 20min; iii) 350°C, 20min; iiiii) 400°C, 20min.

Figure S8 gave the TG-DTA curves of ACMSs and XRD patterns of immediate products obtained at different calcinations stage. The two exothermic peaks centered at 320°C and 340°C appeared. And the XRD patterns showed the crystallization of cerium precursors distinctly happened in the range of 320°C~350°C. So it was inferred that the exothermic peak centered at 320°C originated from combustion heat of organic species while crystallization heat of ceria contributed to the exothermic peak centered at 340°C.

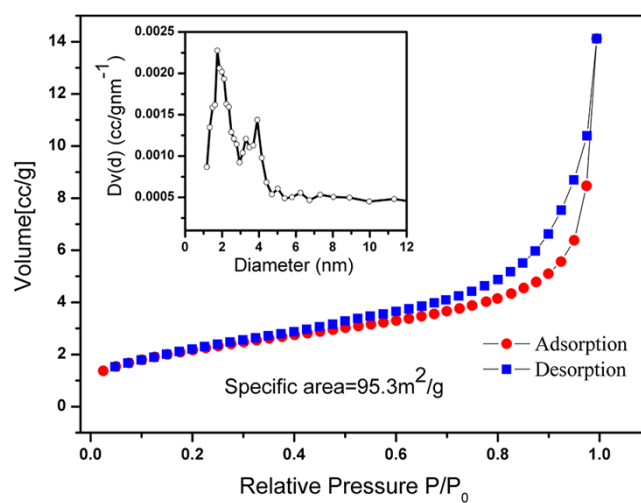


Fig. S9 The nitrogen adsorption-desorption isotherm curves of ceria hollow spheres with triple shell. Inset: the BJH pore size distribution of ceria hollow spheres with triple shell from the adsorption branch.

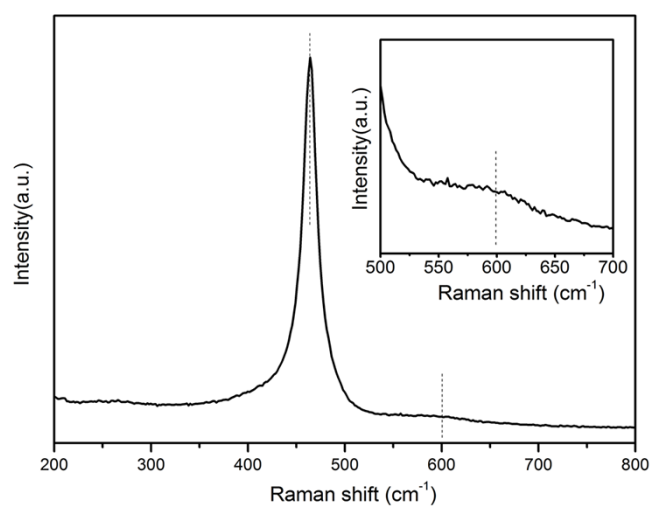


Fig. S10 Raman spectra of ceria hollow spheres with triple shell. Inset: the regional magnification.

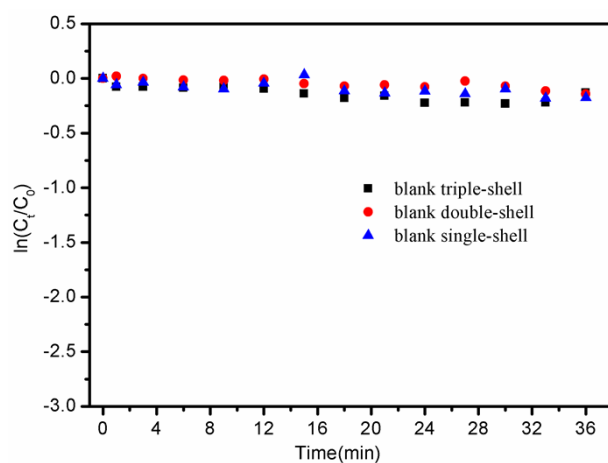


Fig. S11 Plots of $\ln(C_t/C_0)$ vs time by use of non-loaded MSCHSs (denoted as blank triple-shell, blank double-shell and blank single-shell).

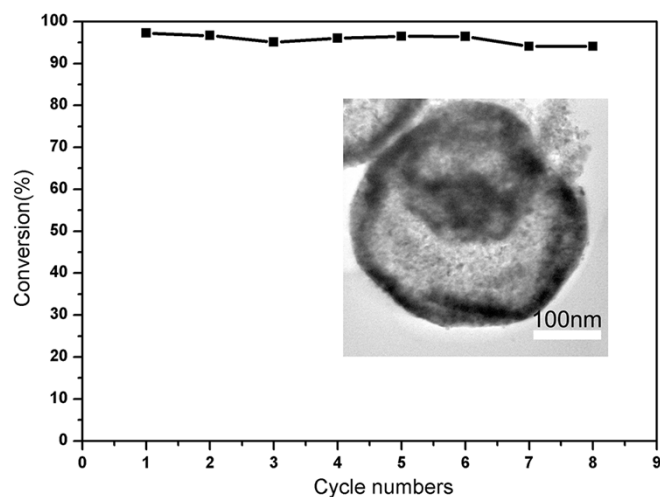


Fig. S12. Cycle stability of the triple-shell Au/CeO₂ hybrid catalyst. Inset: TEM image of the catalyst after 8 cycles.

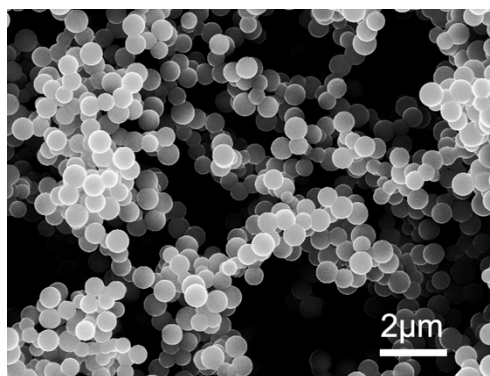


Fig. S13 Representative SEM images of as-synthesized carbonaceous spheres.