

## Supporting Information for:

# Comprehensive study of carbon dioxide adsorption in the metal-organic frameworks M<sub>2</sub>(dobdc) (M = Mg, Mn, Fe, Co, Ni, Cu, Zn)

Wendy L. Queen,\*<sup>a</sup> Matthew R. Hudson,<sup>b†</sup> Eric D. Bloch,<sup>c‡</sup> Jarad A. Mason,<sup>c</sup> Miguel Gonzalez,<sup>c</sup> Jason S. Lee,<sup>a,d</sup> David Gygi,<sup>c</sup> Joshua D. Howe,<sup>a,d</sup> Kyuho Lee,<sup>a,d</sup> Tamim A. Darwish,<sup>e</sup> Michael James,<sup>e,f</sup> Vanessa K. Peterson,<sup>g</sup> Simon J. Teat,<sup>h</sup> Berend Smit,<sup>d,i,j</sup> Jeffrey B. Neaton,<sup>a,k</sup> Jeffrey R. Long,<sup>c,l</sup> and Craig M. Brown\*<sup>b,k</sup>

<sup>a</sup>The Molecular Foundry, Lawrence Berkeley National Laboratory, Berkeley, California, 94720, USA.

<sup>b</sup>National Institute of Standards and Technology, Center for Neutron Research, Gaithersburg, Maryland, 20899, USA

<sup>c</sup>Department of Chemistry, University of California, Berkeley, California 94720, USA

<sup>d</sup>Department of Chemical and Biomolecular Engineering, University of California, Berkeley, California, 94720, USA.

<sup>e</sup>National Deuteration Facility, Australian Nuclear Science and Technology Organisation, Lucas Heights, Australia.

<sup>f</sup>Australian Synchrotron, 800 Blackburn Road, Clayton VIC 3168, Australia.

<sup>g</sup>The Bragg Institute, Australian Nuclear Science and Technology Organisation, Lucas Heights, NSW 2234, Australia.

<sup>h</sup>Advanced Light Source, Lawrence Berkeley National Laboratory, Berkeley, California, 94720, USA.

<sup>i</sup>Division of Materials Sciences, Lawrence Berkeley National Laboratory, Berkeley, California, 94720, USA.

<sup>j</sup>Department Institut des Sciences et Ingénierie Chimiques, Ecole Polytechnique Fédérale de Lausanne (EPFL), CH 1015 Lausanne, Switzerland

<sup>k</sup>Department of Physics, University of California, Berkeley, California, 94720-1462, USA.

<sup>l</sup>Department of Chemical Engineering, University of Delaware, Newark, Delaware, 19716, USA.

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## **Chemical deuteration of 2,5-dihydroxy-1,4-benzendicarboxylic-d<sub>2</sub> acid**

### **Additional Information:**

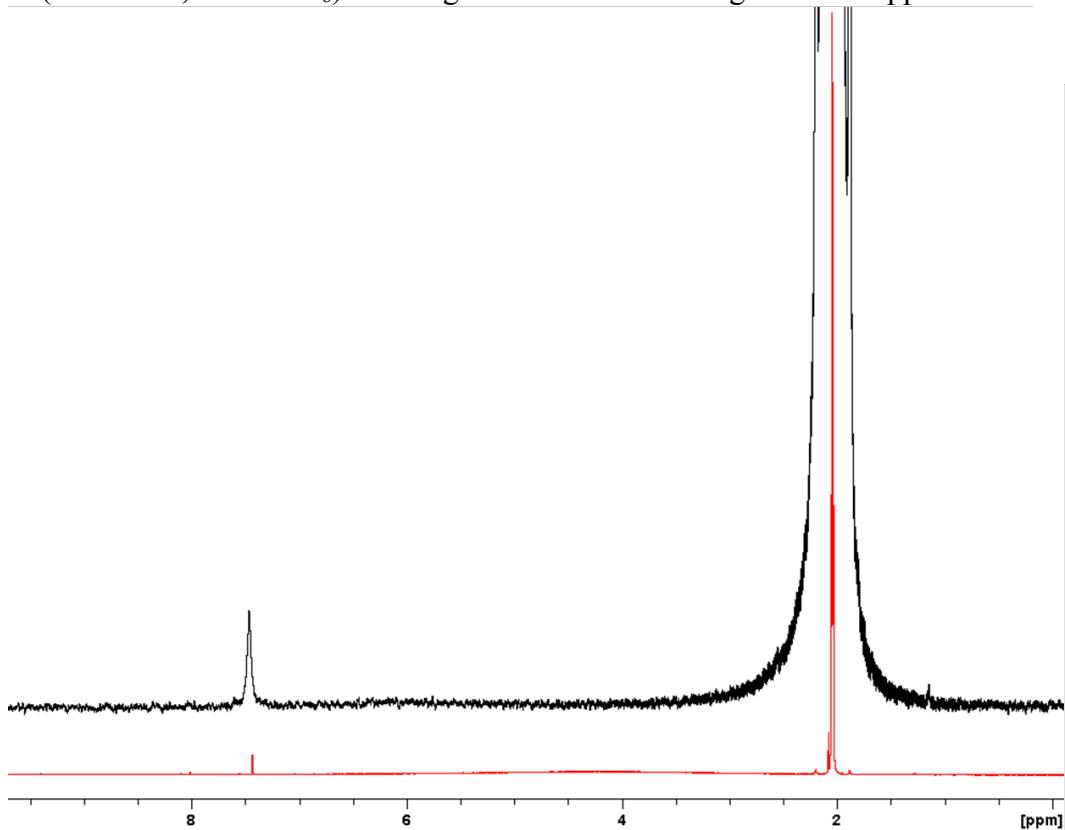
Chemicals and reagents of the highest grade were purchased from Sigma Aldrich and were used without further purification. Solvents were purchased from Sigma Aldrich and Merck (St. Louis, MO, USA) and were purified by established methods. NMR solvents were purchased from Cambridge Isotope Laboratories Inc. (MA, USA) and Sigma Aldrich and were used without further purification. D<sub>2</sub>O (99.8 %) was supplied by AECL (Ontario, Canada). Thin-layer chromatography (TLC) was performed on Fluka Analytical silica gel aluminium sheets (25 F254) (product of Sigma Aldrich, St. Louis, MO, USA). Davisil® silica gel (LC60Å 40-63 micron) (product of W. R. Grace & Co.-Conn, Columbia, USA) was used for bench-top column chromatography.\*\*

Hydrothermal reactions were performed using a Mini Benchtop 4560 Parr Reactor (600 mL vessel capacity, 3000 psi maximum pressure, 350 °C maximum temperature). Thin layer chromatography was used (referenced with the protonated compound) to estimate the purity and to develop separation protocols. <sup>1</sup>H NMR (400 MHz) and <sup>2</sup>H NMR (61.4 MHz) spectra were recorded on a Bruker 400 MHz spectrometer at 298 K. Chemical shifts, in ppm, were referenced to the residual signal of the corresponding NMR solvent. Deuterium NMR was performed using the probe's lock channel for direct observation.

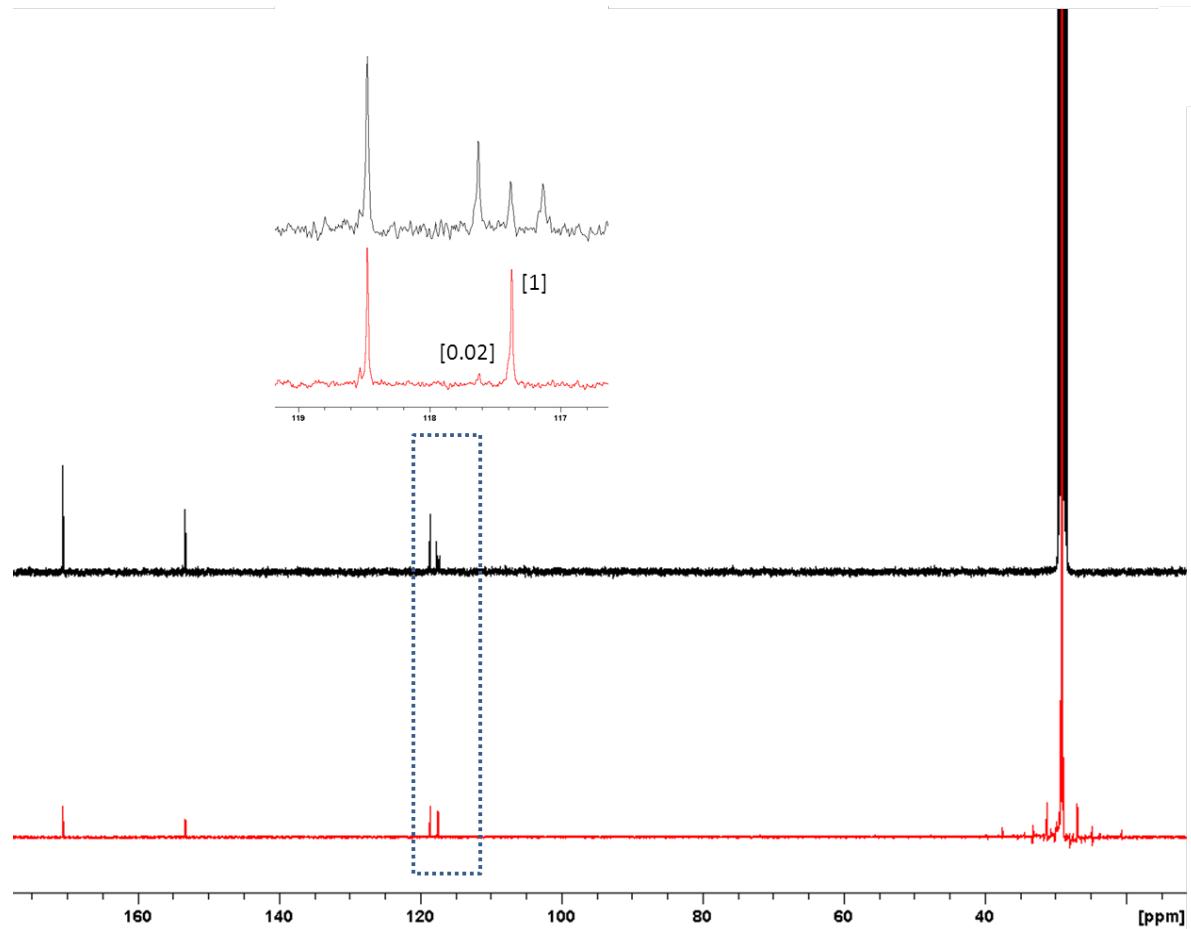
Electrospray ionization mass spectra (ESI-MS) were recorded on a 4000 QTrap AB Sciex spectrometer. The overall percentage deuteration of the molecules was calculated by NMR and MS using the isotope distribution analysis of the different isotopologues in MS.

\*\* Commercial materials and equipment are identified in this paper only to specify adequately the experimental procedure. In no case does such identification imply recommendation by NIST nor does it imply that the material or equipment identified is necessarily the best available for this purpose.

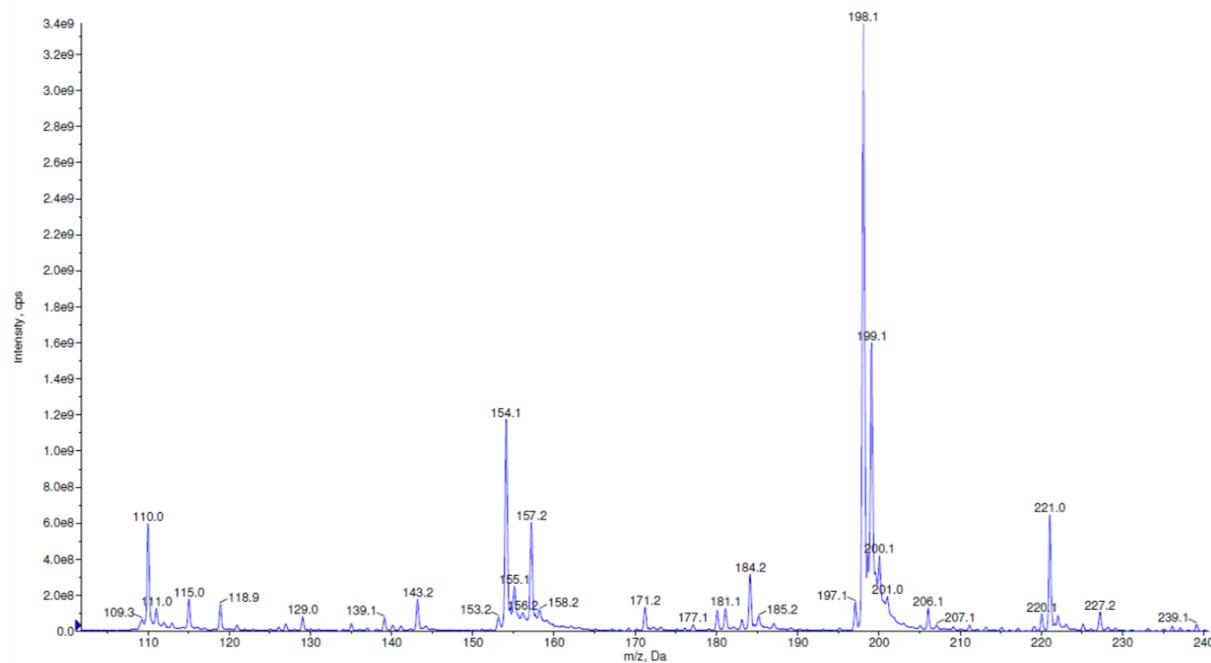
**Figure S1.** NMR Spectra of 2,5-dihydroxy-1,4-benzendicarboxylic-d<sub>2</sub> acid. Bottom (in red): <sup>1</sup>H NMR (400 MHz, acetone-d<sub>6</sub>) showing small residual proton signal at 7.43 ppm. Top (in black): <sup>2</sup>H NMR (61.4 MHz, acetone-d<sub>6</sub>) showing a broad deuterium signal at 7.46 ppm.



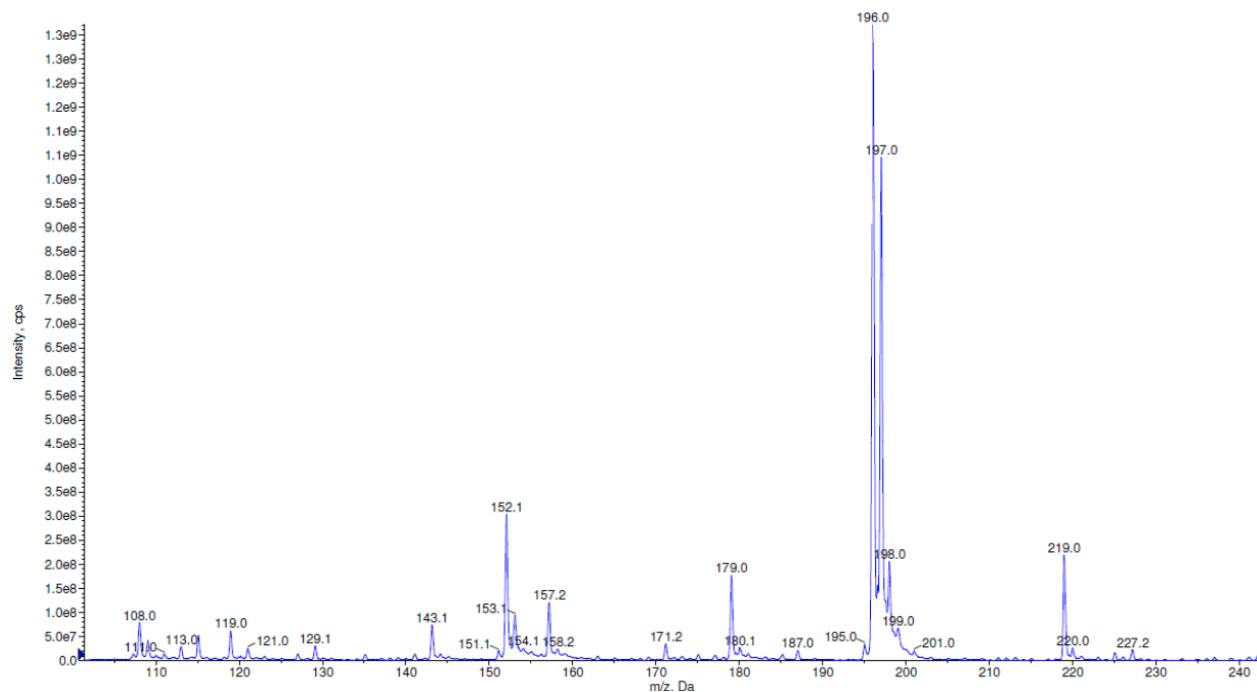
**Figure S2.**  $^{13}\text{C}$  NMR (in acetone-d<sub>6</sub>) spectra of 2,5-dihydroxy-1,4-benzendicarboxylic-d<sub>2</sub> acid. Top (in black):  $^{13}\text{C} \{^1\text{H}\}$  (proton decoupled only) spectrum showing triplet for the carbon baring D; bottom (in red):  $^{13}\text{C} \{^1\text{H}, ^2\text{H}\}$  (decoupling both  $^1\text{H}$  and  $^2\text{H}$  with D1 = 20 s) where the triplet is resolved into singlet, confirming C-D, and a residual peak integrating 1 to 0.02, respectively.



**Figure S3.** ESI-MS (negative mode) of 2,5-dihydroxy-1,4-benzenedicarboxylic-d<sub>2</sub> acid (deuterated, Mw = 200) showed two signals at m/e 199 ( $M^{-1}$ ) and 198 ( $M^{-2}$ ).



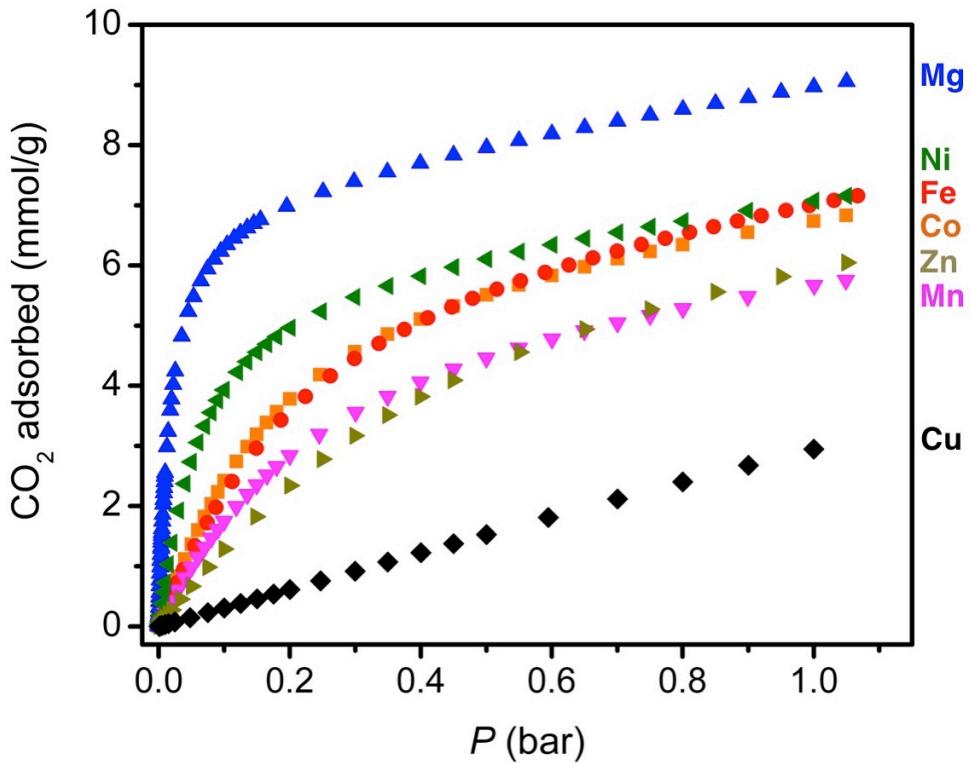
**Figure S4.** ESI-MS (negative mode) of protonated 2,5-dihydroxy-1,4-benzenedicarboxylic (protonated, Mw = 198) showed two signals at m/e 197 ( $M^{-1}$ ) and 196 ( $M^{-2}$ ).



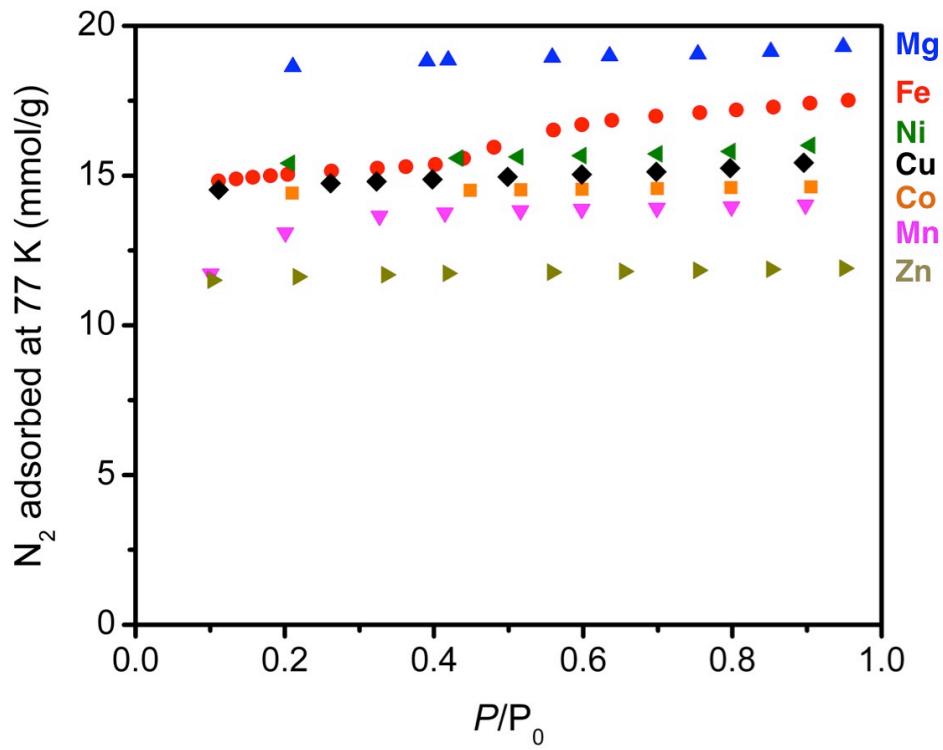
## **Low pressure gas adsorption of the M<sub>2</sub>(dobdc) series**

UHP-grade (99.999 % purity) helium, nitrogen, and carbon dioxide were used for all adsorption measurements. Gas adsorption isotherms for pressures in the range 0 – 1.1 bar were measured using a Micromeritics ASAP 2020 instrument. For standard measurements in ASAP low-pressure glass sample holders, activated samples were transferred under a N<sub>2</sub> atmosphere to preweighed analysis tubes, which were capped with a Transeal. The samples were evacuated on the ASAP until the outgas rate was less than 3  $\mu\text{bar}/\text{min}$ . The evacuated analysis tubes containing degassed samples were then carefully transferred to an electronic balance and weighed to determine the mass of sample (typically 100 – 200 mg). For cryogenic measurements, the tube was fitted with an isothermal jacket and transferred back to the analysis port of the gas adsorption instrument. The outgas rate was again confirmed to be less than 3  $\mu\text{bar}/\text{min}$ . Langmuir surface areas and pore volumes were determined by measuring N<sub>2</sub> adsorption isotherms in a 77 K liquid N<sub>2</sub> bath and calculated using the Micromeritics software, assuming a value of 16.2 Å<sup>2</sup> for the molecular cross-sectional area of N<sub>2</sub>. Adsorption isotherms between 25 and 45 °C were measured using a recirculating dewar (Micromeritics) connected to a Julabo F32-MC isothermal bath.

**Figure S5.** Excess CO<sub>2</sub> adsorption isotherms at 298 K for all M<sub>2</sub>(dobdc) frameworks.



**Figure S6.** N<sub>2</sub> adsorption at 77 K for all M<sub>2</sub>(dobdc) frameworks.



## Fitting CO<sub>2</sub> adsorption isotherms for M<sub>2</sub>(dobdc) series

The CO<sub>2</sub> adsorption isotherms were fit with a dual-site Langmuir-Freundlich model (Eqn. 1), where  $n$  is the absolute amount adsorbed in mmol/g,  $P$  is the pressure in bar,  $q_{\text{sat},i}$  is the saturation capacity in mmol/g,  $b_i$  is the Langmuir parameter in bar<sup>-1</sup>, and  $v_i$  is the Freundlich parameter for two sites 1 and 2. The fitted parameters for each adsorption isotherm can be found in Tables S1-S7. Plots of the CO<sub>2</sub> adsorption isotherms with the corresponding dual-site Langmuir-Freundlich fits can be found in Fig. S7-S13. Several recent studies have demonstrated the ability of Langmuir-type equations to successfully model adsorption in metal-organic frameworks, especially those that have well-defined adsorption sites on the pore surface.<sup>i</sup>

$$n = \frac{q_{\text{sat},1} b_1 P^{v_1}}{1 + b_1 P^{v_1}} + \frac{q_{\text{sat},2} b_2 P^{v_2}}{1 + b_2 P^{v_2}} \quad (1)$$

**Table S1.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Co<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	7.2	7.2	7.2
$b_1$	6.4	3.9	2.5
$v_1$	1.1	1.1	1.1
$q_{\text{sat},2}$	2.8	2.8	2.3
$b_2$	0.23	0.13	0.11
$v_2$	2.2	1.9	1.6

**Table S2.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Fe<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	8.3	8.3	8.3
$b_1$	4.1	2.6	1.7
$v_1$	1.1	1.1	1.1
$q_{\text{sat},2}$	3.1	3.1	3.1
$b_2$	0.12	0.086	0.064
$v_2$	2.5	2.5	2.5

**Table S3.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Mg<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	7.2	7.1	7.2
$b_1$	56.5	33.4	18.2
$v_1$	1	1	1
$q_{\text{sat},2}$	7.0	7.7	7.1
$b_2$	0.36	0.22	0.15
$v_2$	1.2	1.1	1.2

**Table S4.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Mn<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	7	7	7
$b_1$	3.3	2.1	1.4
$v_1$	1	1	1
$q_{\text{sat},2}$	10	10	10
$b_2$	0.031	0.024	0.019
$v_2$	1	1	1

**Table S5.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Ni<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	6.2	6.3	6.1
$b_1$	22.0	12.1	7.5
$v_1$	1.1	1.1	1.1
$q_{\text{sat},2}$	3.0	3.0	3.1
$b_2$	0.64	0.32	0.24
$v_2$	1.7	1.8	1.4

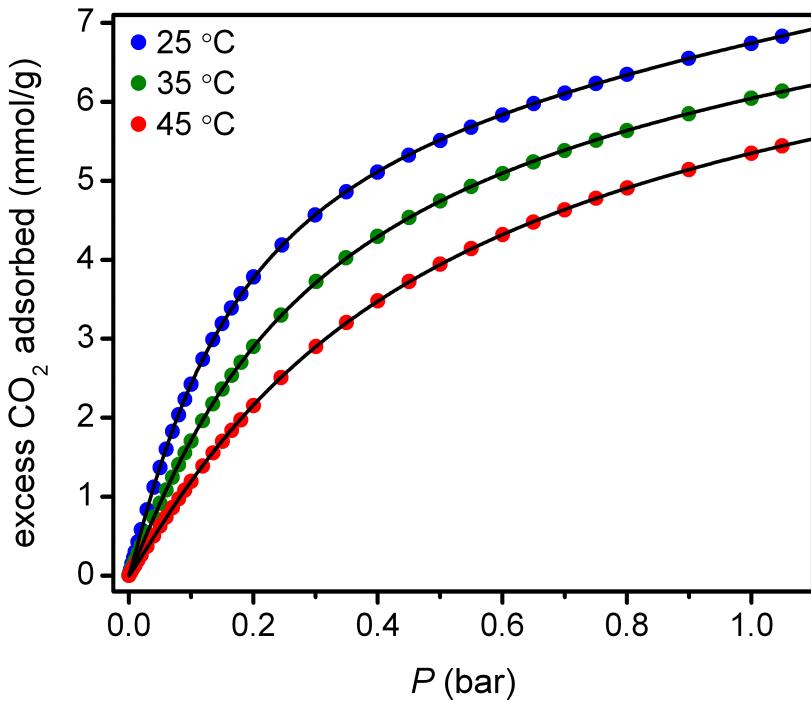
**Table S6.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Zn<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	6.7	6.7	6.7
$b_1$	2.2	1.5	1.0
$v_1$	1.1	1.1	1.1
$q_{\text{sat},2}$	3.3	3.3	3.3
$b_2$	0.68	0.51	0.39
$v_2$	0.82	0.82	0.82

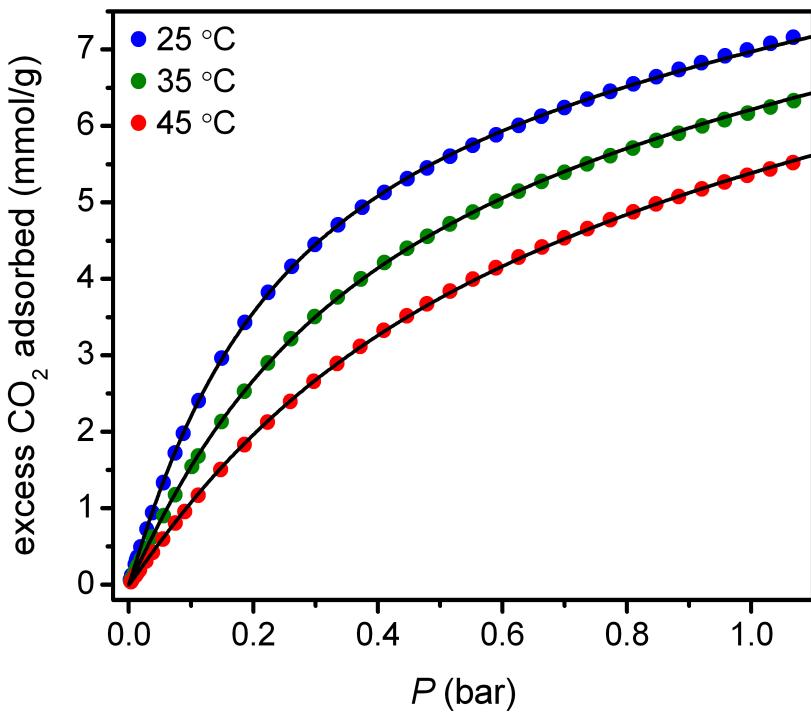
**Table S7.** Dual-site Langmuir-Freundlich parameters for CO<sub>2</sub> adsorption in Cu<sub>2</sub>(dobdc) at 25, 35, and 45 °C.

	25 °C	35 °C	45 °C
$q_{\text{sat},1}$	3.2	3.2	3.2
$b_1$	0.35	0.19	0.10
$v_1$	1.8	1.8	1.8
$q_{\text{sat},2}$	7.2	7.2	7.20
$b_2$	0.41	0.31	0.24
$v_2$	1.00	1.0	1.0

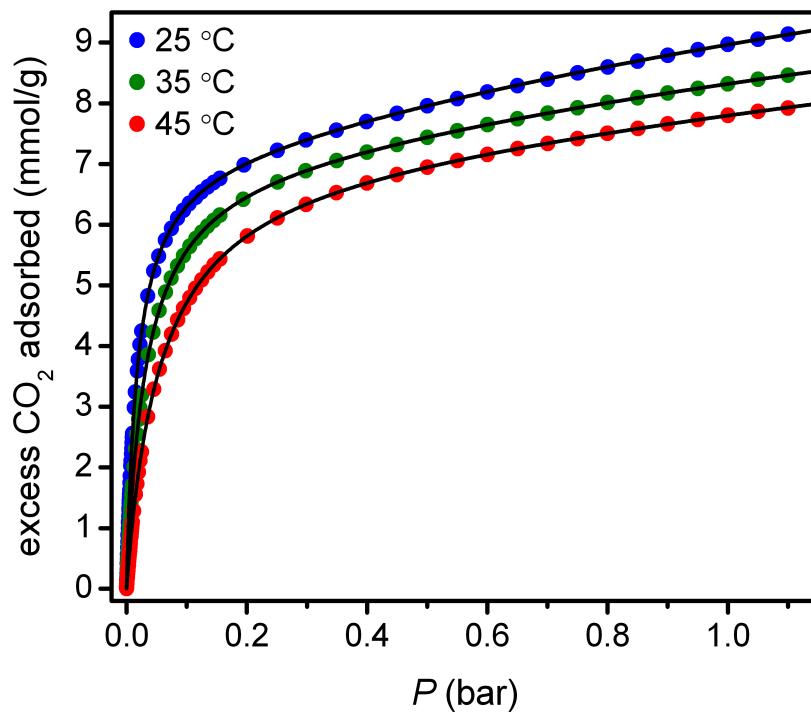
**Figure S7.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Co<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



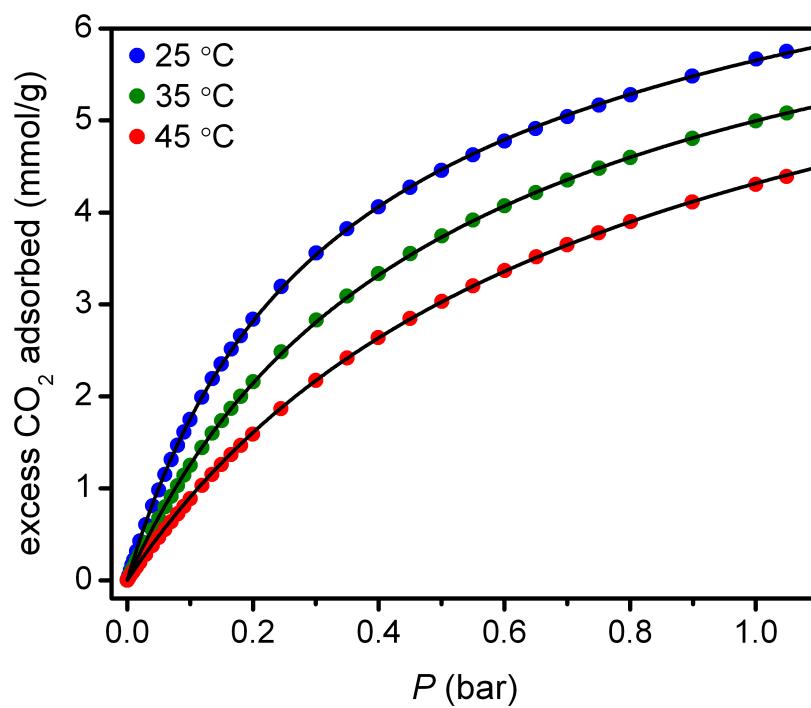
**Figure S8.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Fe<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



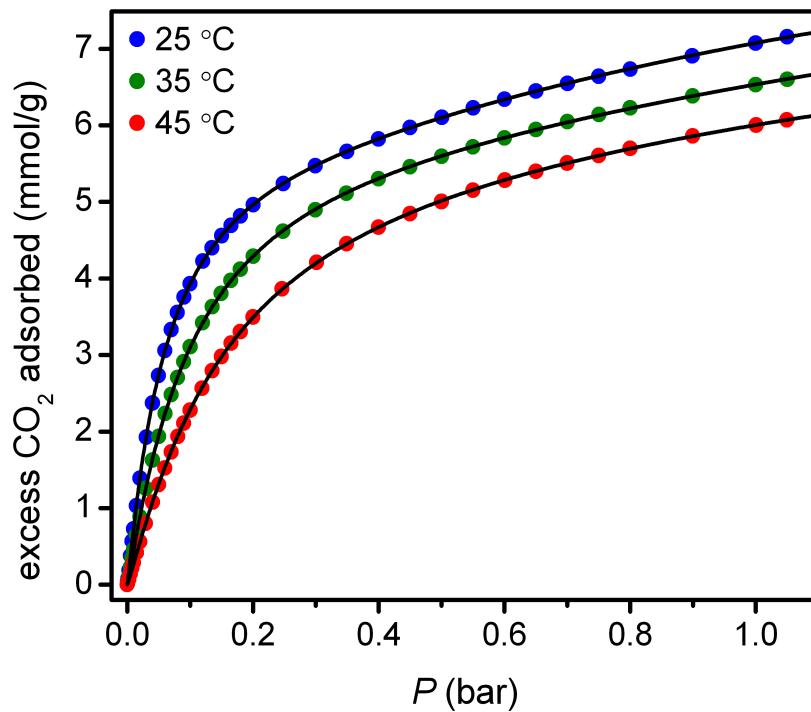
**Figure S9.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Mg<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



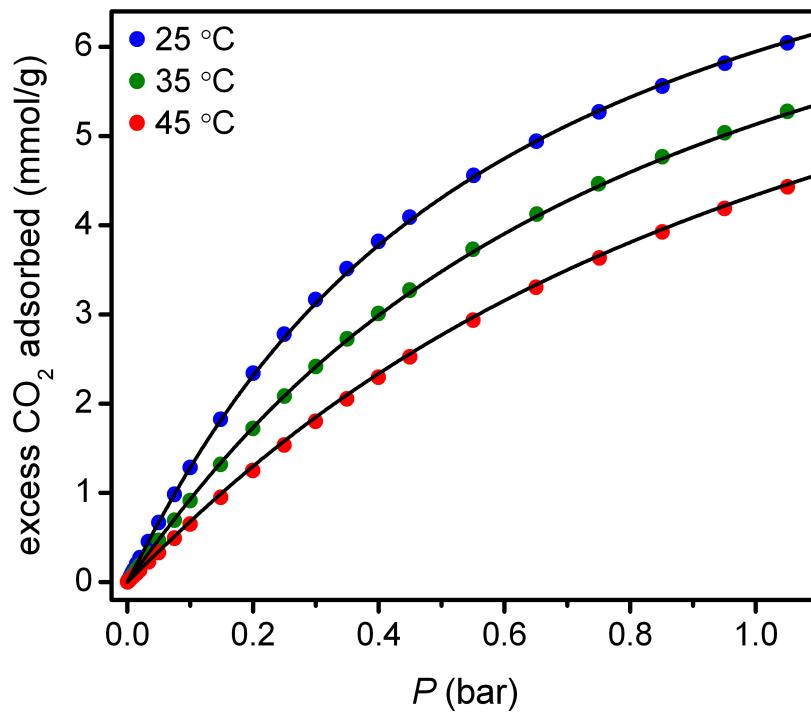
**Figure S10.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Mn<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



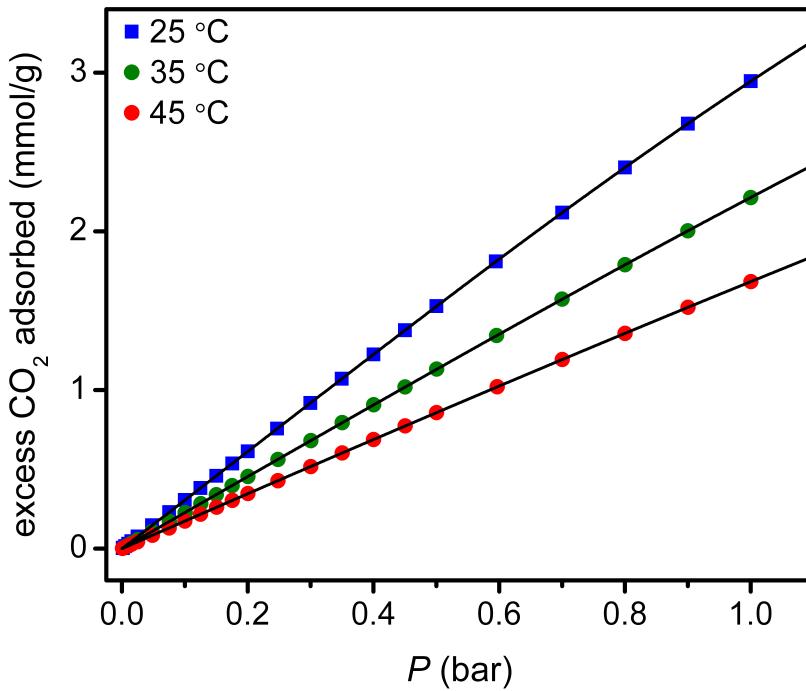
**Figure S11.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Ni<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



**Figure S12.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Zn<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



**Figure S13.** Dual-site Langmuir-Freundlich fits for CO<sub>2</sub> adsorption in Cu<sub>2</sub>(dobdc) at 25, 35, and 45 °C.



### Calculating Isosteric heats of adsorption

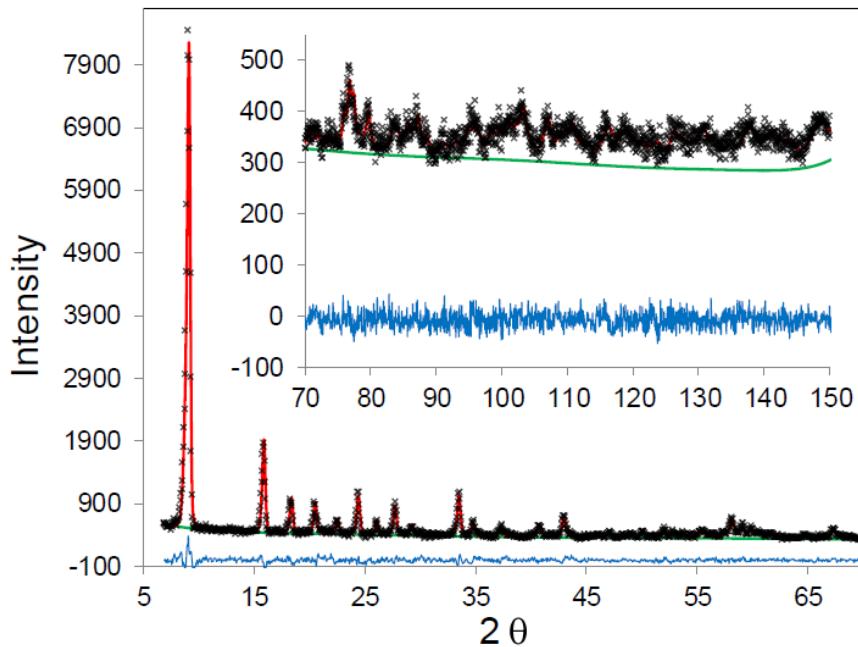
The Clausius-Clapeyron equation (Eqn. 2) was used to calculate the isosteric heats of adsorption,  $-Q_{st}$ , for CO<sub>2</sub> in each M<sub>2</sub>(dobdc) analog, using the dual-site Langmuir-Freundlich fits for each material at 25, 35, and 45 °C to calculate the pressures that correspond to a given CO<sub>2</sub> loading at each temperature.<sup>ii</sup>

$$\ln P = -\frac{Q_{st}}{R} \left( \frac{1}{T} \right) + C \quad (2)$$

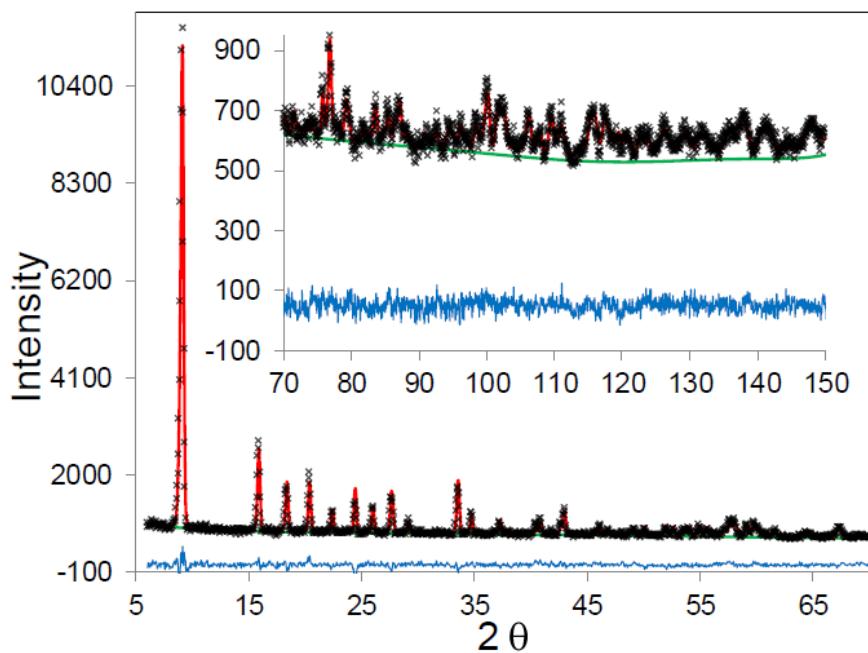
Here,  $P$  is the pressure,  $n$  is the amount adsorbed,  $T$  is the temperature,  $R$  is the universal gas constant, and  $C$  is a constant. The isosteric heat of adsorption,  $-Q_{st}$ , was obtained from the slope of plots of  $(\ln P)_n$  as a function of  $1/T$ . An error in the isosteric heat for a given loading can be calculated from the standard error in slope of the best-fit line. Fundamentally, this error describes the quality of agreement between the fitted isotherms and the Clausius-Clapeyron relation. The isosteric heats of adsorption as a function of loading for CO<sub>2</sub> in M<sub>2</sub>(dobdc) can be found in Figure 2. Note that while it is most appropriate to use absolute adsorption in thermodynamic calculations,<sup>iii</sup> the isosteric heats of adsorption reported here were calculated from experimentally measured excess adsorption isotherms since the difference between the excess and absolute CO<sub>2</sub> uptake at the highest measured pressures ( $\approx 1$  bar) is less than 0.03 mmol/g, which is within the error of the experimental measurement. As such, the isosteric of adsorption is only calculated up to the highest CO<sub>2</sub> loading that was measured at 45 °C, as the isotherm fits have only been experimentally verified in this region and difference between excess and absolute adsorption will become significant at higher loadings.

## Neutron powder diffraction data for the bare and CO<sub>2</sub> dosed M<sub>2</sub>(dobdc) series

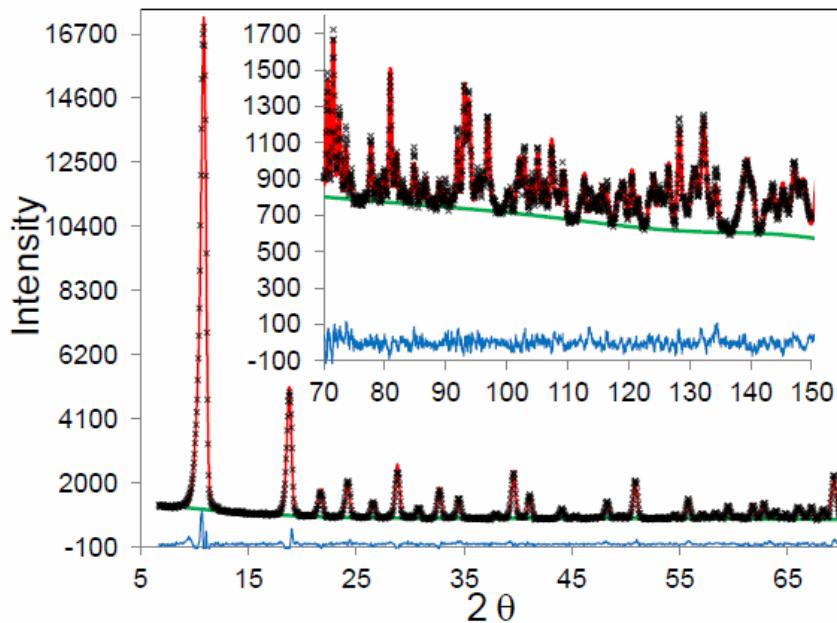
**Figure S14.** Neutron powder diffraction data for bare Zn<sub>2</sub>(dobdc), 10 K. The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



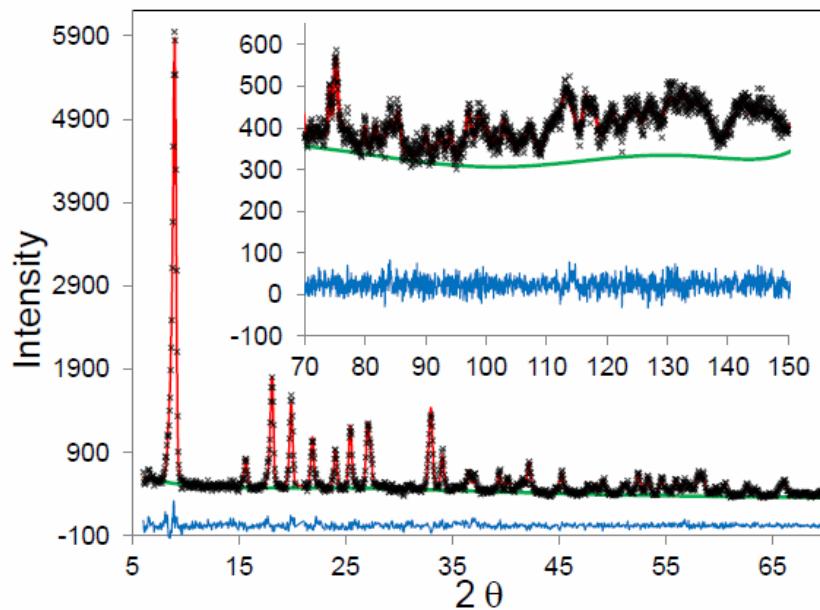
**Figure S15.** Neutron powder diffraction data for bare Co<sub>2</sub>(dobdc), 10 K. The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



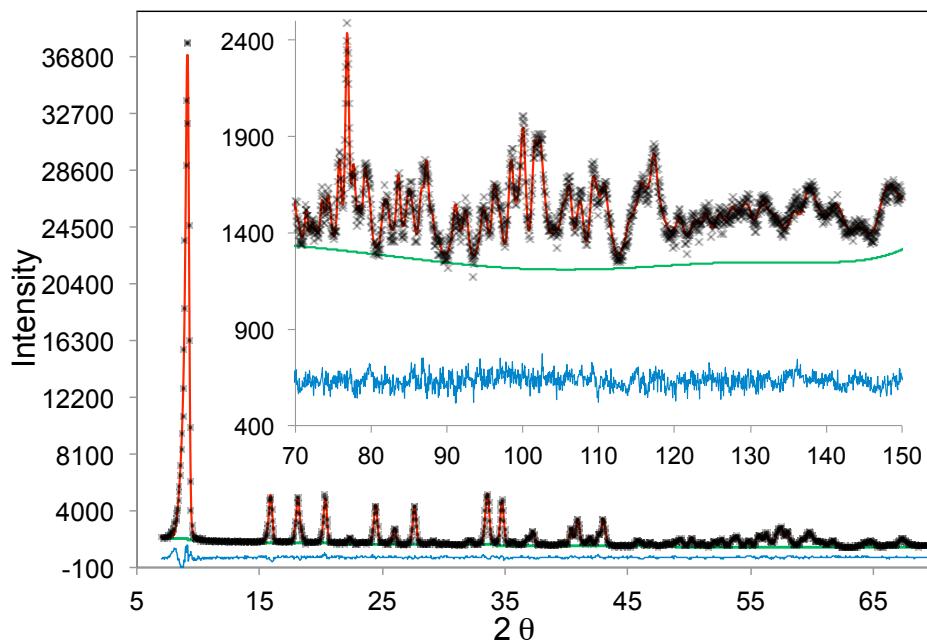
**Figure S16.** Neutron powder diffraction data for bare  $\text{Fe}_2(\text{dobdc})$ , 10 K. The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



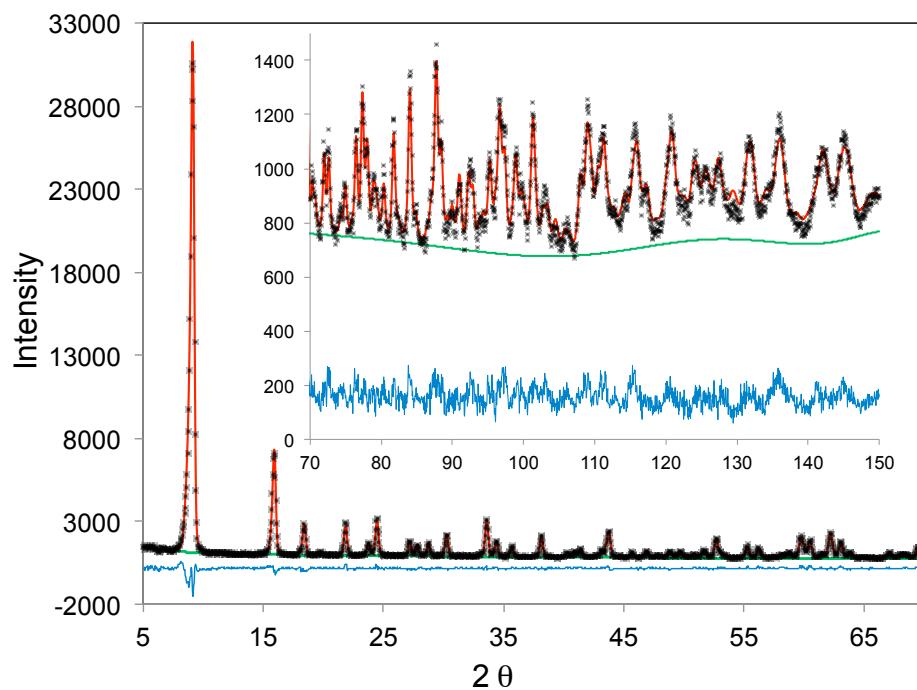
**Figure S17.** Neutron powder diffraction data for bare  $\text{Mn}_2(\text{dobdc})$ , 10 K. The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



**Figure S18.** Neutron powder diffraction data for  $\text{Mg}_2(\text{dobdc})$  with 0.75  $\text{CO}_2$  per Mg site, 10 K. The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



**Figure S19.** Neutron powder diffraction data for bare  $\text{Cu}_2(\text{dobdc})$ , 10K. The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



**Table S8.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of activated Mn<sub>2</sub>(dobdc) framework, 10 K, Trigonal, R-3,  $a = 26.333(2)$  Å,  $c = 7.0477(5)$  Å,  $V = 4232.3(5)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 0.9049$ , wRp = 0.0440, Rp = 0.0361.

Atom	X	Y	Z	Occupancy	$U_{\text{ISO}}$ (Å <sup>2</sup> )	Multiplicity
Mn	0.3854(7)	0.3606(9)	0.142(3)	1	0.007(5)	18
O1	0.3280(6)	0.2935(6)	0.360(2)	1	0.023(2)	18
O2	0.2999(7)	0.2282(7)	0.591(2)	1	0.023(2)	18
O3	0.3544(7)	0.2745(7)	0.014(3)	1	0.023(2)	18
C1	0.3186(5)	0.2441(5)	0.420(2)	1	0.014(1)	18
C2	0.3289(5)	0.2055(5)	0.288(2)	1	0.014(1)	18
C3	0.3443(6)	0.2219(5)	0.089(1)	1	0.014(1)	18
C4	0.3523(5)	0.1841(5)	-0.018(1)	1	0.014(1)	18
H	0.3599(9)	0.1921(8)	-0.157(3)	1	0.010(6)	18

**Table S9.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of activated Fe<sub>2</sub>(dobdc) framework, 10 K, Trigonal, R-3,  $a = 26.0983(4)$  Å,  $c = 6.8512(2)$  Å,  $V = 4041.3(1)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 2.936$ , wRp = 0.0359, Rp = 0.0282.

Atom	X	Y	Z	Occupancy	$U_{\text{ISO}}$ (Å <sup>2</sup> )	Multiplicity
Fe	0.3824(2)	0.3521(2)	0.1430(6)	1	0.012(2)	18
O1	0.3272(3)	0.2938(4)	0.363(1)	1	0.003(3)	18
O2	0.3010(4)	0.2272(4)	0.599(1)	1	0.024(3)	18
O3	0.3549(4)	0.2732(3)	0.007(1)	1	0.009(2)	18
C1	0.3161(4)	0.2440(3)	0.423(1)	1	0.022(2)	18
C2	0.3259(3)	0.2038(3)	0.287(1)	1	0.002(2)	18
C3	0.3430(3)	0.2226(4)	0.095(1)	1	0.012(2)	18
C4	0.3487(3)	0.1819(4)	-0.034(1)	1	0.003(2)	18
H	0.3613(6)	0.1919(6)	-0.168(2)	1	0.013(7)	18

**Table S10.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of activated Co<sub>2</sub>(dobdc) framework, 10 K, Trigonal, R-3,  $a = 25.9067(9)\text{\AA}$ ,  $c = 6.8548(5)\text{\AA}$ ,  $V = 3984.3(3) \text{ \AA}^3$ . Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.044$ , wRp = 0.0379, Rp = 0.0316.

Atom	X	Y	Z	Occupancy	$U_{(\text{ISO})} (\text{\AA}^2)$	Multiplicity
Co	0.385(1)	0.349(1)	0.136(4)	1	0.003(7)	18
O1	0.3283(6)	0.2968(6)	0.367(2)	1	0.018(3)	18
O2	0.3011(6)	0.2279(6)	0.603(2)	1	0.001(3)	18
O3	0.3547(6)	0.2749(6)	0.011(2)	1	0.001(3)	18
C1	0.3160(6)	0.2470(6)	0.424(2)	1	0.029(4)	18
C2	0.3290(6)	0.2075(5)	0.288(2)	1	0.020(4)	18
C3	0.3477(6)	0.2257(7)	0.089(2)	1	0.028(4)	18
C4	0.3503(6)	0.1810(5)	-0.026(2)	1	0.008(3)	18
H	0.3647(9)	0.1951(7)	-0.175(3)	1	0.001(4)	18

**Table S11.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of activated Zn<sub>2</sub>(dobdc) framework, 10 K, Trigonal, R-3,  $a = 25.929(2) \text{ \AA}$ ,  $c = 6.7962(9)\text{\AA}$ ,  $V = 3957.1(6) \text{ \AA}^3$ . Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 0.9577$ , wRp = 0.0484, Rp = 0.0402.

Atom	X	Y	Z	Occupancy	$U_{(\text{ISO})} (\text{\AA}^2)$	Multiplicity
Zn	0.3833(9)	0.353(1)	0.139(3)	1	0.021(6)	18
O1	0.3305(9)	0.2968(8)	0.382(3)	1	0.014(3)	18
O2	0.300(1)	0.227(1)	0.603(3)	1	0.014(3)	18
O3	0.3541(9)	0.272(1)	0.001(3)	1	0.014(3)	18
C1	0.3166(9)	0.2450(9)	0.429(3)	1	0.015(2)	18
C2	0.3267(9)	0.2058(9)	0.286(3)	1	0.015(2)	18
C3	0.3441(9)	0.224(1)	0.094(3)	1	0.015(2)	18
C4	0.3521(9)	0.185(1)	-0.029(3)	1	0.015(2)	18
H	0.365(2)	0.194(2)	-0.158(7)	1	0.05(2)	18

**Table S12.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of activated Cu<sub>2</sub>(dobdc) framework, 10 K, Trigonal, R-3,  $a = 25.8733(7)$  Å,  $c = 6.2577(2)$  Å,  $V = 3627.8(2)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.187$ , wRp = 0.0276, Rp = 0.0337.

Atom	X	Y	Z	Occupancy	$U_{\text{ISO}}$ (Å <sup>2</sup> )	Multiplicity
Cu	0.3864(2)	0.3484(2)	0.1754(8)	1	0.001(1)	18
O1	0.3551(3)	0.3026(3)	0.437(1)	1	0.012(1)	18
O2	0.2920(3)	0.2243(3)	0.628(1)	1	0.012(1)	18
O3	0.3591(3)	0.2759(3)	0.005(1)	1	0.012(1)	18
C1	0.3236(3)	0.2480(3)	0.465(1)	1	0.0106(5)	18
C2	0.3267(3)	0.2064(3)	0.303(1)	1	0.0106(5)	18
C3	0.3480(3)	0.2235(3)	0.093(1)	1	0.0106(5)	18
C4	0.3545(3)	0.1840(3)	-0.040(1)	1	0.0106(5)	18
H	0.3687(5)	0.1963(5)	-0.204(2)	1	0.021(3)	18

**Table S13.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Mg<sub>2</sub>(dobdc) framework dosed with 0.75 CO<sub>2</sub> per Mg, 10 K, Trigonal, R-3,  $a = 25.8728(8)$  Å,  $c = 6.8700(3)$  Å,  $V = 3982.7(3)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.303$ , wRp = 0.0273, Rp = 0.0223.

Atom	X	Y	Z	Occupancy	$U_{\text{ISO}}$ (Å <sup>2</sup> )	Multiplicity
Mg	0.3804(4)	0.3515(4)	0.138(1)	1	0.011(2)	18
O1	0.3252(3)	0.2956(3)	0.362(1)	1	0.012(1)	18
O2	0.3028(3)	0.2288(4)	0.598(1)	1	0.012(1)	18
O3	0.3539(4)	0.2744(4)	0.002(1)	1	0.012(1)	18
C1	0.3158(3)	0.2466(3)	0.4239(9)	1	0.0203(6)	18
C2	0.3289(3)	0.2087(3)	0.2858(9)	1	0.0203(6)	18
C3	0.3459(3)	0.2239(3)	0.092(1)	1	0.0203(6)	18
C4	0.3526(3)	0.1825(3)	-0.0272(9)	1	0.0203(6)	18
H	0.3606(5)	0.1904(5)	-0.174(2)	1	0.007(3)	18
O2b	0.2283(8)	0.7573(8)	0.602(2)	0.825(7)	0.097(6)	18
C2a	0.1773(6)	0.7214(5)	0.600(2)	0.825(7)	0.052(4)	18
O2a	0.1283(4)	0.6856(4)	0.600(1)	0.825(7)	0.011(3)	18

**Table S14.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Mn<sub>2</sub>(dobdc) framework dosed with 0.5 CO<sub>2</sub> per Mn, 10 K, Trigonal, R-3,  $a = 26.326(2)$  Å,  $c = 7.0440(5)$  Å,  $V = 4227.9(5)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 0.9695$ , wRp = 0.0436, Rp = 0.0355.

Atom	X	Y	Z	Occupancy	$U_{(ISO)}$ (Å <sup>2</sup> )	Multiplicity
Mn	0.386(1)	0.361(1)	0.142(4)	1	0.033(7)	18
O1	0.3292(6)	0.2945(6)	0.362(2)	1	0.019(2)	18
O2	0.3007(7)	0.2278(7)	0.589(2)	1	0.019(2)	18
O3	0.3572(6)	0.2768(7)	0.011(2)	1	0.019(2)	18
C1	0.3202(5)	0.2457(5)	0.419(2)	1	0.017(1)	18
C2	0.3288(6)	0.2053(5)	0.288(2)	1	0.017(1)	18
C3	0.3453(6)	0.2235(6)	0.090(2)	1	0.017(1)	18
C4	0.3524(6)	0.1832(6)	-0.021(2)	1	0.017(1)	18
H	0.361(1)	0.1919(8)	-0.161(3)	1	0.004(5)	18
O2b	0.234(2)	0.777(2)	0.470(5)	0.54(1)	0.05(1)	18
C2a	0.191(1)	0.738(1)	0.529(4)	0.54(1)	0.030(9)	18
O2a	0.146(1)	0.7000(9)	0.583(4)	0.54(1)	0.005(7)	18

**Table S15.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Fe<sub>2</sub>(dobdc) framework dosed with 0.35 CO<sub>2</sub> per Fe, 10 K, Trigonal, R-3,  $a = 26.0349(5)$  Å,  $c = 6.8508(2)$  Å,  $V = 4021.4(2)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 3.161$ , wRp = 0.0399, Rp = 0.0307.

Atom	X	Y	Z	Occupancy	$U_{(ISO)}$ (Å <sup>2</sup> )	Multiplicity
Fe	0.3844(2)	0.3531(2)	0.1400(5)	1	0.0061(7)	18
O1	0.3256(3)	0.2936(3)	0.362(1)	1	0.0050(8)	18
O2	0.3020(3)	0.2293(3)	0.603(1)	1	0.0050(8)	18
O3	0.3537(3)	0.2727(3)	0.012(1)	1	0.0050(8)	18
C1	0.3174(3)	0.2460(2)	0.4200(9)	1	0.0041(4)	18
C2	0.3278(3)	0.2040(3)	0.2893(9)	1	0.0041(4)	18
C3	0.3448(2)	0.2237(3)	0.0916(8)	1	0.0041(4)	18
C4	0.3501(3)	0.1824(3)	-0.0239(8)	1	0.0041(4)	18
H	0.3609(5)	0.1934(4)	-0.173(2)	1	0.002(3)	18
O2b	0.212(1)	0.756(1)	0.510(4)	0.350(6)	0.0434*	18
C2a	0.1686(9)	0.714(1)	0.552(4)	0.350(6)	0.0468*	18
O2a	0.1253(9)	0.6725(9)	0.592(3)	0.350(6)	0.010(6)	18

\*Uaniso O2b=[U11,U12,U13,U22,U23,U33] = [0.04(2),0.02(1),0.000(1),0.04(2),-0.00(2),0.04(2)]

Uaniso C2a=[U11,U12,U13,U22,U23,U33] = [0.05(1),0.025(9),0.00(1),0.05(1),0.00(1),0.05(2)]

**Table S16.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Co<sub>2</sub>(dobdc) framework dosed with 0.5 CO<sub>2</sub> per Co, 10 K, Trigonal, R-3,  $a = 25.8977(9)$  Å,  $c = 6.8460(4)$  Å,  $V = 3976.4(3)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 0.8989$ , wRp = 0.0345, Rp = 0.0278.

Atom	X	Y	Z	Occupancy	$U_{(ISO)}$ (Å <sup>2</sup> )	Multiplicity
Co	0.382(1)	0.349(1)	0.133(4)	1	0.004(9)	18
O1	0.3260(5)	0.2973(6)	0.372(2)	1	0.001(2)	18
O2	0.3022(6)	0.2275(6)	0.601(2)	1	0.001(2)	18
O3	0.3562(6)	0.2740(6)	0.005(2)	1	0.001(2)	18
C1	0.3165(5)	0.2470(6)	0.428(2)	1	0.008(1)	18
C2	0.3302(5)	0.2082(5)	0.288(2)	1	0.008(1)	18
C3	0.3473(5)	0.2255(6)	0.089(2)	1	0.008(1)	18
C4	0.3517(6)	0.1809(6)	-0.028(2)	1	0.008(1)	18
H	0.3616(9)	0.1928(8)	-0.174(3)	1	0.004(5)	18
O2b	0.218(2)	0.757(2)	0.504(5)	0.62(2)	0.09(2)	18
C2a	0.172(1)	0.718(1)	0.558(4)	0.62(2)	0.037(8)	18
O2a	0.127(1)	0.6848(9)	0.611(3)	0.62(2)	0.001(6)	18

**Table S17.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Zn<sub>2</sub>(dobdc) framework dosed with 0.5 CO<sub>2</sub> per Zn, 10 K, Trigonal, R-3,  $a = 25.914(2)$  Å,  $c = 6.7986(6)$  Å,  $V = 3953.8(5)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.088$ , wRp = 0.0345, Rp = 0.0279.

Atom	X	Y	Z	Occupancy	$U_{(ISO)}$ (Å <sup>2</sup> )	Multiplicity
Zn	0.3836(8)	0.3546(9)	0.144(2)	1	0.029(5)	18
O1	0.3281(6)	0.2966(7)	0.375(2)	1	0.015(3)	18
O2	0.3001(7)	0.2267(7)	0.611(2)	1	0.015(3)	18
O3	0.3537(8)	0.2743(8)	0.017(2)	1	0.015(3)	18
C1	0.3157(7)	0.2448(7)	0.434(2)	1	0.016(1)	18
C2	0.3270(7)	0.2064(6)	0.284(2)	1	0.016(1)	18
C3	0.3431(6)	0.2212(6)	0.092(2)	1	0.016(1)	18
C4	0.3525(7)	0.1841(8)	-0.036(2)	1	0.016(1)	18
H	0.361(1)	0.191(1)	-0.165(4)	1	0.032(8)	18
O2b	0.229(3)	0.766(3)	0.538(9)	0.54(2)	0.10(2)	18
C2a	0.182(2)	0.724(2)	0.577(6)	0.54(2)	0.053(1)	18
O2a	0.1356(2)	0.685(1)	0.614(4)	0.54(2)	0.01(1)	18

**Table S18.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Cu<sub>2</sub>(dobdc) framework dosed with 0.5 CO<sub>2</sub> per Cu, 10 K, Trigonal, R-3,  $a = 25.8582(7)$  Å,  $c = 6.2615(2)$  Å,  $V = 3625.8(2)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.271$ , wRp = 0.0321, Rp = 0.0266.

Atom	X	Y	Z	Occupancy	U <sub>(ISO)</sub> (Å <sup>2</sup> )	Multiplicity
Cu	0.3865(2)	0.3491(2)	0.1761(8)	1	0.002(1)	18
O1	0.3544(3)	0.3031(3)	0.434(1)	1	0.006(1)	18
O2	0.2922(3)	0.2240(3)	0.629(1)	1	0.006(1)	18
O3	0.3605(3)	0.2770(3)	0.008(1)	1	0.006(1)	18
C1	0.3246(3)	0.2478(3)	0.466(1)	1	0.0098(6)	18
C2	0.3277(3)	0.2062(3)	0.299(1)	1	0.0098(6)	18
C3	0.3481(3)	0.2247(3)	0.093(1)	1	0.0098(6)	18
C4	0.3541(3)	0.1841(3)	-0.043(1)	1	0.0098(6)	18
H	0.3672(5)	0.1955(5)	-0.198(2)	1	0.011(3)	18
O1a	0.251(2)	0.787(1)	0.468(8)	0.172(6)	0.02(2)	18
C1a	0.210(2)	0.747(2)	0.550(8)	0.172(6)	0.08(2)	18
O1b	0.170(2)	0.707(2)	0.634(7)	0.172(6)	0.008(2)	18
O2a	0.149(3)	0.045(4)	0.30(1)	0.142(5)	0.10(4)	18
C2a	0.153(2)	0.020(3)	0.449(8)	0.142(5)	0.06(3)	18
O2b	0.156(2)	-0.005(3)	0.598(8)	0.142(5)	0.02(2)	18
O3a	0.0	0.0	0.314(2)	0.152(8)	0.06(4)	6
C3a	0.0	0.0	0.5	0.152(8)	0.3(1)	3

**Table S19.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Fe<sub>2</sub>(dobdc) framework dosed with 1.3 CO<sub>2</sub> per Fe, 10 K, Trigonal, R-3,  $a = 26.0010(5)$  Å,  $c = 6.8607(2)$  Å,  $V = 4016.8(2)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.862$ , wRp = 0.0417, Rp = 0.0334.

Atom	X	Y	Z	Occupancy	U <sub>(ISO)</sub> (Å <sup>2</sup> )	Multiplicity
Fe	0.3856(2)	0.3529(2)	0.1427(6)	1	0.0055(8)	18
O1	0.3255(3)	0.2932(3)	0.352(1)	1	0.014(1)	18
O2	0.3033(3)	0.2262(4)	0.600(1)	1	0.014(1)	18
O3	0.3534(4)	0.2708(3)	0.008(1)	1	0.014(1)	18
C1	0.3161(3)	0.2447(3)	0.4228(9)	1	0.0077(5)	18
C2	0.3254(3)	0.2036(3)	0.2909(9)	1	0.0077(5)	18
C3	0.3429(3)	0.2209(3)	0.0961(9)	1	0.0077(5)	18
C4	0.3514(3)	0.1828(3)	-0.0248(9)	1	0.0077(5)	18
H	0.3606(5)	0.1954(5)	-0.171(2)	1	0.01(3)	18
O2b	0.2213(4)	0.7566(5)	0.552(2)	0.856(6)	0.142(8)	18
C2a	0.1746(3)	0.7160(4)	0.581(2)	0.856(6)	0.068(4)	18
O2a	0.1280(3)	0.6796(4)	0.615(1)	0.856(6)	0.012(2)	18
O3b	0.2006(8)	0.0157(4)	0.678(2)	0.407(6)	0.09(1)	18
C3a	0.1848(2)	0.0209(2)	0.524(1)	0.407(6)	0.028(6)	18
O3a	0.1699(7)	0.0267(4)	0.374(2)	0.407(6)	0.048(7)	18

**Table S20.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Co<sub>2</sub>(dobdc) framework dosed with 1.5 CO<sub>2</sub> per Co, 10 K, Trigonal, R-3,  $a = 25.874(1)\text{\AA}$ ,  $c = 6.8486(4)\text{\AA}$ ,  $V = 3970.5(4)\text{ \AA}^3$ . Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 0.9775$ , wRp = 0.0355, Rp = 0.0290.

Atom	X	Y	Z	Occupancy	$U_{(\text{ISO})} (\text{\AA}^2)$	Multiplicity
Co	0.379(1)	0.349(1)	0.151(4)	1	0.001(9)	18
O1	0.3287(6)	0.2959(7)	0.364(2)	1	0.005(2)	18
O2	0.3026(7)	0.2291(7)	0.599(2)	1	0.005(2)	18
O3	0.3577(7)	0.2758(7)	0.002(2)	1	0.005(2)	18
C1	0.3191(6)	0.2476(6)	0.429(2)	1	0.016(2)	18
C2	0.3282(7)	0.2069(6)	0.290(2)	1	0.016(2)	18
C3	0.3465(6)	0.2239(7)	0.091(2)	1	0.016(2)	18
C4	0.3506(6)	0.1816(7)	-0.024(2)	1	0.016(2)	18
H	0.358(1)	0.191(1)	-0.168(4)	1	0.012(7)	18
O2b	0.232(2)	0.747(2)	0.651(6)	0.73(2)	0.13(2)	18
C2a	0.179(1)	0.715(1)	0.625(3)	0.73(2)	0.028(7)	18
O2a	0.130(1)	0.6839(9)	0.605(3)	0.73(2)	0.011(7)	18
O3b	0.206(1)	0.024(1)	0.657(5)	0.78(2)	0.07(1)	18
C3a	0.184(1)	0.024(1)	0.511(5)	0.78(2)	0.07(1)	18
O3a	0.164(1)	0.029(1)	0.369(6)	0.78(2)	0.09(1)	18

**Table S21.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Zn<sub>2</sub>(dobdc) framework dosed with 1.5 CO<sub>2</sub> per Zn, 10 K, Trigonal, R-3,  $a = 25.850(2)$  Å,  $c = 6.8072(8)$  Å,  $V = 3939.4(6)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.255$ , wRp = 0.0369, Rp = 0.0309.

Atom	X	Y	Z	Occupancy	$U_{\text{(ISO)}} (\text{\AA}^2)$	Multiplicity
Zn	0.382(1)	0.351(1)	0.145(2)	1	0.007(6)	18
O1	0.3325(8)	0.2955(9)	0.372(3)	1	0.004(3)	18
O2	0.3040(8)	0.2280(9)	0.600(3)	1	0.004(3)	18
O3	0.356(1)	0.2742(9)	0.010(3)	1	0.004(3)	18
C1	0.316(1)	0.245(1)	0.429(2)	1	0.018(3)	18
C2	0.327(1)	0.2081(8)	0.288(2)	1	0.018(3)	18
C3	0.3456(8)	0.2212(9)	0.097(3)	1	0.018(3)	18
C4	0.3503(9)	0.182(1)	-0.030(3)	1	0.018(3)	18
H	0.363(2)	0.190(1)	-0.155(6)	1	0.01(1)	18
O2b	0.233(3)	0.747(3)	0.626(8)	0.69(2)	0.12(3)	18
C2a	0.183(2)	0.711(2)	0.618(5)	0.69(2)	0.05(3)	18
O2a	0.134(2)	0.680(1)	0.615(4)	0.69(2)	0.01(1)	18
O3b	0.203(2)	0.024(3)	0.690(9)	0.60(2)	0.08(2)	18
C3a	0.181(2)	0.025(2)	0.530(6)	0.60(2)	0.06(2)	18
O3a	0.160(2)	0.028(2)	0.389(9)	0.60(2)	0.07(2)	18

**Table S22.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Fe<sub>2</sub>(dobdc) framework dosed with 1.5 CO<sub>2</sub> per Fe, 298 K, 0.67 bar residual pressure, Trigonal, R-3,  $a = 26.082(1)$  Å,  $c = 6.8783(6)$  Å,  $V = 4052.1(5)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 2.435$ , wRp = 0.0225, Rp = 0.0176.

Atom	X	Y	Z	Occupancy	$U_{\text{(ISO)}}$ (Å <sup>2</sup> )	Multiplicity
Fe	0.3862(3)	0.3557(4)	0.134(1)	1	0.014(2)	18
O1	0.3256(5)	0.2966(5)	0.370(2)	1	0.020(2)	18
O2	0.3034(5)	0.2294(5)	0.590(2)	1	0.020(2)	18
O3	0.3573(6)	0.2668(5)	0.011(2)	1	0.020(2)	18
C1	0.3206(5)	0.2494(4)	0.420(2)	1	0.009(1)	18
C2	0.3157(5)	0.1994(4)	0.300(2)	1	0.009(1)	18
C3	0.3410(5)	0.2164(4)	0.117(2)	1	0.009(1)	18
C4	0.3451(5)	0.1818(4)	-0.025(2)	1	0.009(1)	18
H	.3613(8)	0.1928(8)	-0.161(3)	1	0.006(8)	18
O2b	0.210(1)	0.740(1)	0.571(5)	0.645(8)	0.14(2)	18
C2a	0.1620(9)	0.6993(8)	0.598(3)	0.645(8)	0.050(9)	18
O2a	0.1164(9)	0.660(1)	0.623(3)	0.645(8)	0.04(1)	18
O3b	0.191(4)	0.043(5)	0.66(2)	0.124(6)	0.05(5)	18
C3a	0.174(3)	0.038(4)	0.51(1)	0.124(6)	0.01(4)	18
O3a	0.159(4)	0.036(4)	0.36(2)	0.124(6)	0.01(5)	18

**Table S23.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Zn<sub>2</sub>(dobdc) framework dosed with 1.5 CO<sub>2</sub> per Zn, 298 K, 0.6 bar residual pressure, Trigonal, R-3,  $a = 25.887(2)$  Å,  $c = 6.8210(6)$  Å,  $V = 3958.6(5)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.169$ , wRp = 0.0296, Rp = 0.0246.

Atom	X	Y	Z	Occupancy	$U_{\text{(ISO)}} (\text{\AA}^2)$	Multiplicity
Zn	0.3836(9)	0.3524(9)	0.136(2)	1	0.023(5)	18
O1	0.3271(7)	0.2944(8)	0.362(3)	1	0.027(3)	18
O2	0.3033(8)	0.2283(8)	0.608(3)	1	0.027(3)	18
O3	0.3542(9)	0.2738(8)	0.003(3)	1	0.027(3)	18
C1	0.3160(8)	0.2447(8)	0.430(2)	1	0.026(2)	18
C2	0.3259(8)	0.2064(8)	0.283(2)	1	0.026(2)	18
C3	0.3412(8)	0.2202(8)	0.0939(2)	1	0.026(2)	18
C4	0.3493(8)	0.1807(9)	-0.028(2)	1	0.026(2)	18
H	0.364(1)	0.194(1)	-0.153(4)	1	0.022(9)	18
O2b	0.223(4)	0.749(4)	0.54(1)	0.46(2)	0.14(3)	18
C2a	0.177(3)	0.708(2)	0.584(8)	0.46(2)	0.09(2)	18
O2a	0.131(2)	0.669(2)	0.617(5)	0.46(2)	0.02(1)	18
O3b	0.175(7)	0.004(8)	0.75(3)	0.16(1)	0.09(5)	18
C3a	0.174(5)	0.018(5)	0.58(2)	0.16(1)	0.05(5)	18
O3a	0.172(6)	0.032(6)	0.42(3)	0.16(1)	0.07(5)	18

## Single crystal analysis of CO<sub>2</sub> in Co<sub>2</sub>(dobdc)

**Table S24.** Single crystal X-ray diffraction data and structure refinement for activated Co<sub>2</sub>(dobdc) dosed with 5 torr CO<sub>2</sub> at 150 K.

Empirical formula	Co <sub>2</sub> C <sub>10.94</sub> H <sub>2</sub> O <sub>11.88</sub>
Formula weight	441.35 g/mol
Temperature	150(2) K
Wavelength	0.7749 Å
Crystal system	Trigonal
Space group	R-3 :H
Unit cell dimensions	$a = 25.994(7)$ Å, $\alpha = 90^\circ$ $b = 25.994(7)$ Å, $\beta = 90^\circ$ $c = 6.8127(18)$ Å, $\gamma = 120^\circ$
Volume	3987(2) Å <sup>3</sup>
Z	9
Density (calculated)	1.653 Mg/m <sup>3</sup>
Absorption coefficient	2.428 mm <sup>-1</sup>
$F_{000}$	1948
Crystal color	Violet
Crystal size	0.200 x 0.050 x 0.050 mm <sup>3</sup>
$\theta$ range	3.406 to 32.075°
Index ranges	-35 ≤ h ≤ 16, 0 ≤ k ≤ 35, 0 ≤ l ≤ 8
Reflections collected	2236
Independent reflections	2236 [ $R_{int} = 0.0925$ ]
Completeness to $\theta = 27.706^\circ$	96.3 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.888 and 0.287
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	2236 / 19 / 117
Goodness-of-fit on F <sup>2a</sup>	1.130
Final R indices [I > 2σ(I)] <sup>b</sup>	$R_I = 0.0977$ , $wR_2 = 0.2437$
R indices (all data)	$R_I = 0.1284$ , $wR_2 = 0.2628$
Largest diff. peak and hole	1.564 and -1.346 e.Å <sup>-3</sup>

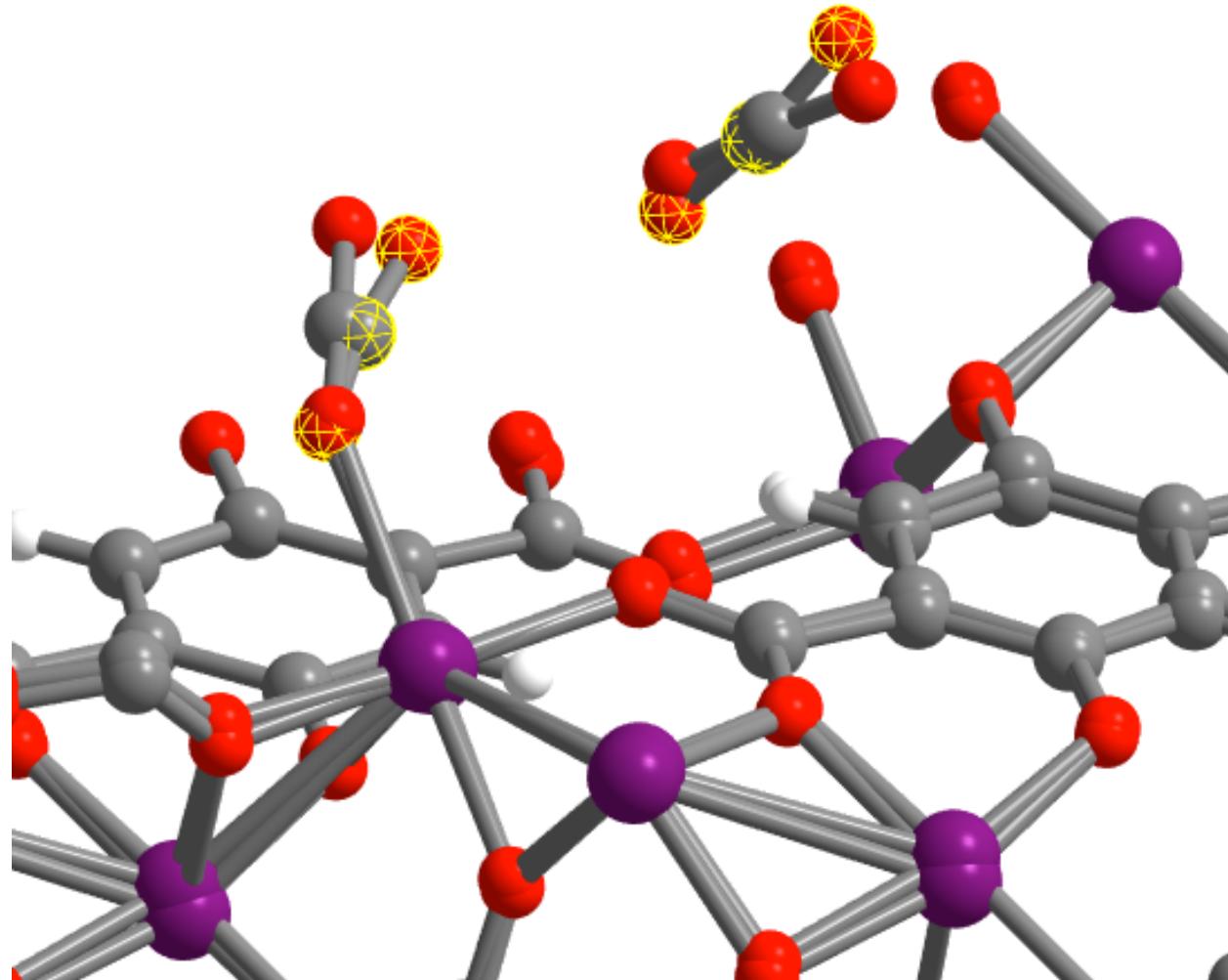
<sup>a</sup>GooF =  $\{\sum[w(F_o^2 - F_c^2)^2]/(n - p)\}^{1/2}$  where n is the number of reflections and p is the total number of parameters refinded. <sup>b</sup> $R_I = \sum|F_o| - |F_c|/\sum|F_o|$ ,  $wR_2 = \{\sum[w(F_o^2 - F_c^2)^2]/\sum[w(F_o^2)^2]\}^{1/2}$

**Table S25.** Fractional atomic coordinates, occupancies, and equivalent isotropic displacement parameters obtained from single crystal X-ray diffraction data and structure refinement of activated Co<sub>2</sub>(dobdc) dosed with 5 torr CO<sub>2</sub> at 150 K.

Atom	X	Y	Z	Occupancy	U <sub>(eq)</sub> (Å <sup>2</sup> ) <sup>a</sup>	Multiplicity
Co	0.9495(1)	0.6347(1)	0.187(2)	1	0.021(1)	18
O1	0.9407(2)	0.6902(3)	0.8309(8)	1	0.022(1)	18
O2	0.9628(3)	0.6594(3)	0.4697(9)	1	0.023(1)	18
O3	0.8953(3)	0.6335(3)	0.2350(9)	1	0.025(1)	18
C1	0.8889(4)	0.6777(4)	0.743(1)	1	0.022(2)	18
C2	0.8742(4)	0.6610(4)	0.547(1)	1	0.025(2)	18
C3	0.8191(4)	0.6502(4)	0.475(1)	1	0.024(2)	18
C4	0.9131(4)	0.6510(4)	0.408(1)	1	0.022(2)	18
O4	0.8710(5)	0.5537(4)	-0.114(2)	0.85(3)	0.055(4)	18
C5	0.820(1)	0.5347(19)	-0.106(8)	0.85(3)	0.181(15)	18
O5	0.770(2)	0.521(3)	-0.08(1)	0.85(3)	0.45(3)	18
O6	0.882(2)	0.518(2)	0.502(6)	0.62(3)	0.22(2)	18
C6	0.851(2)	0.496(2)	0.368(7)	0.62(3)	0.22(2)	18
O7	0.814(2)	0.481(2)	0.245(5)	0.62(3)	0.22(2)	18

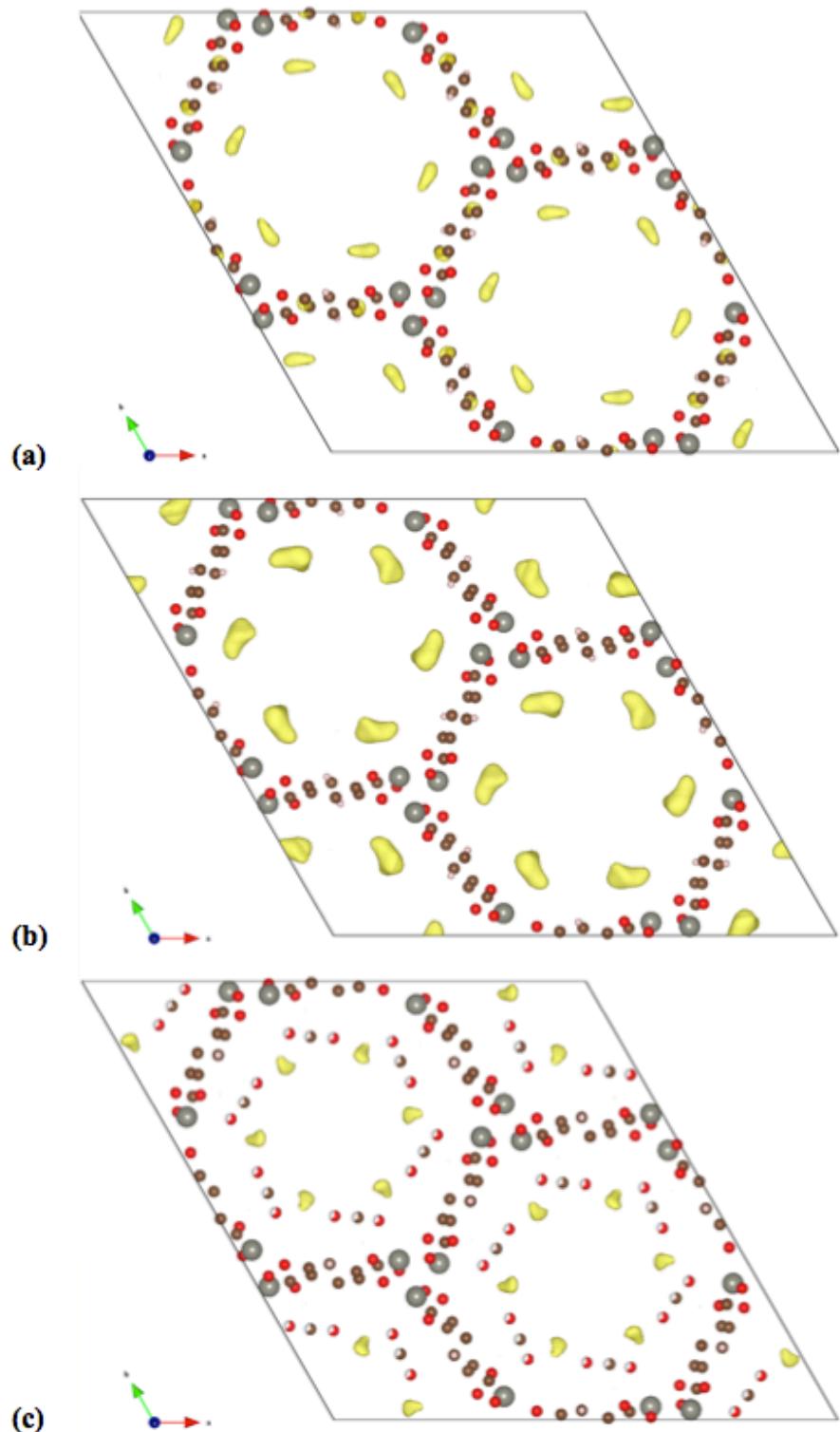
<sup>a</sup>U<sub>(eq)</sub> is defined as one third of the trace of the orthogonalized U<sup>ij</sup> tensor.

**Figure S20.** Structural overlay of the models obtained from powder neutron diffraction (10 K) and single crystal X-ray diffraction (150 K) obtained from samples of Co<sub>2</sub>(dobdc). The highlighted CO<sub>2</sub> molecules are from the single crystal model. We have shown that the CO<sub>2</sub> angles with respect to the framework surface are loading dependent. Despite differences in loading level and difference in data collection temperatures, the models show good agreement.

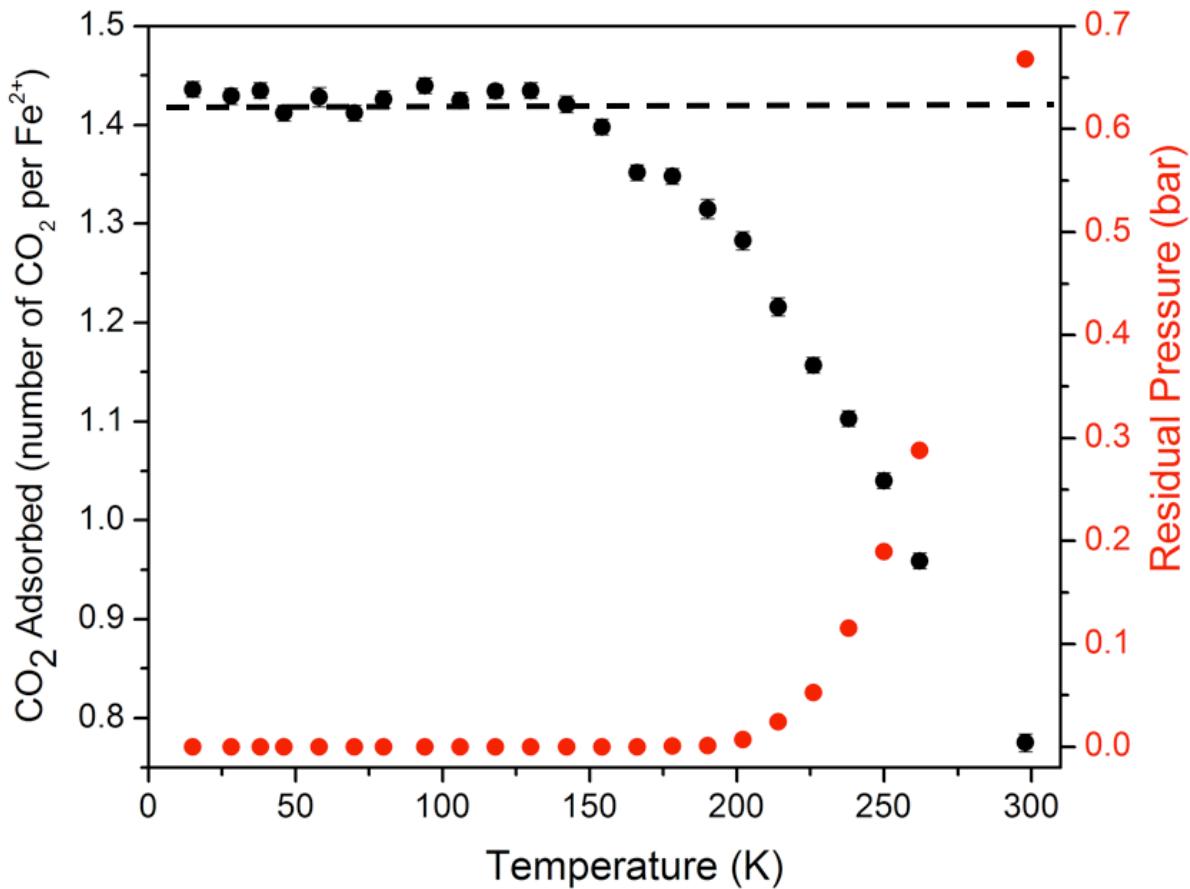


## Additional images

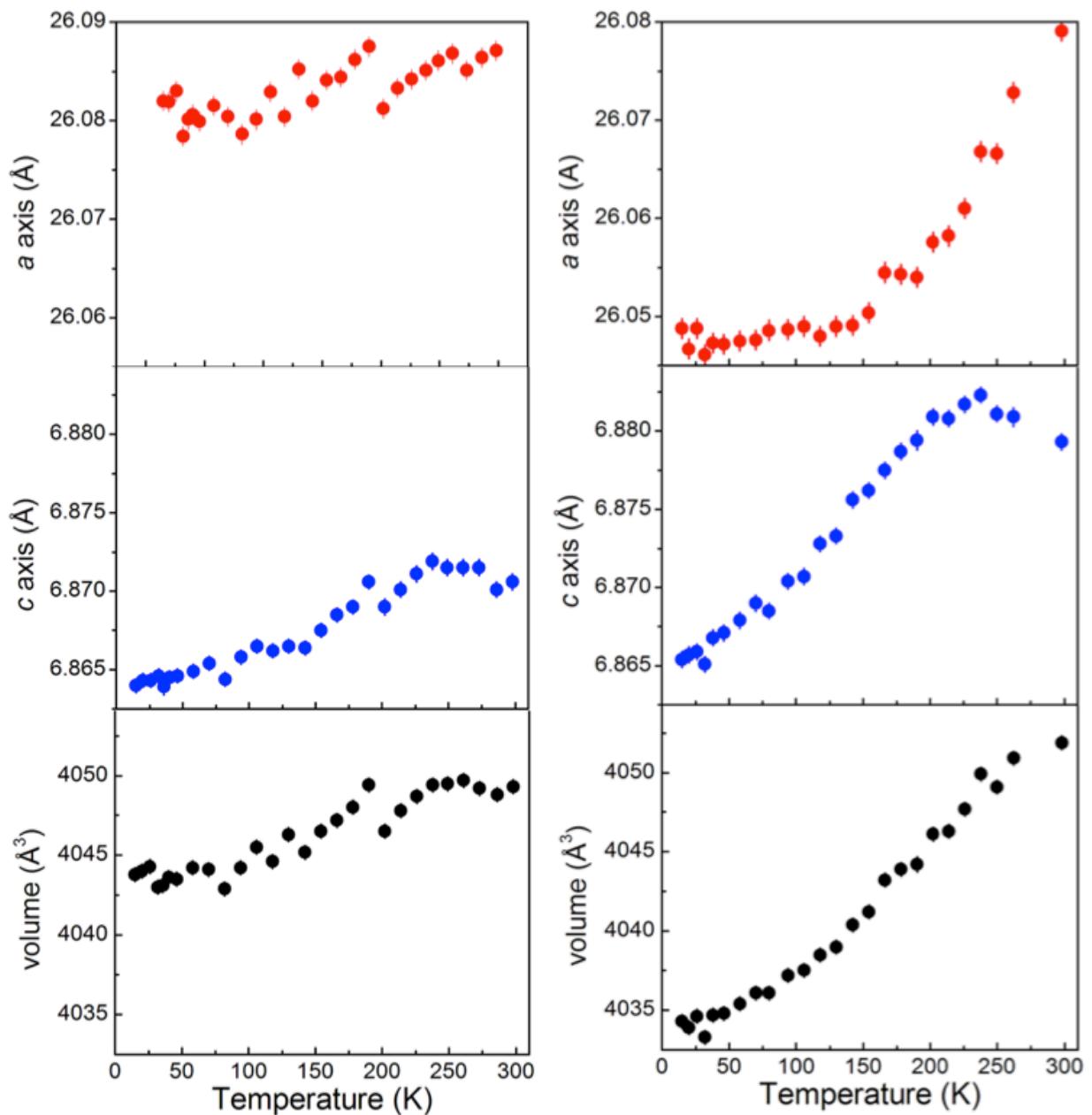
**Figure S21.** Fourier difference map showing the Zn bound CO<sub>2</sub> molecule at (a) 10 K and (b) 298 K and (c) secondary adsorption site at 10 K.



**Figure S22.** The black data points represent the total number of CO<sub>2</sub> adsorbed in sites I and II for the 1.5 CO<sub>2</sub> per iron (II) loading at temperatures ranging from 10 K to 300 K (black dashed line represents ideal adsorption). The data were acquired using a high-intensity powder neutron diffractometer, Wombat. The error bars represent a single standard deviation. The red data points show the residual pressure of CO<sub>2</sub> as a function of temperature revealing significant desorption that begins at a temperature of approximately 178 K.

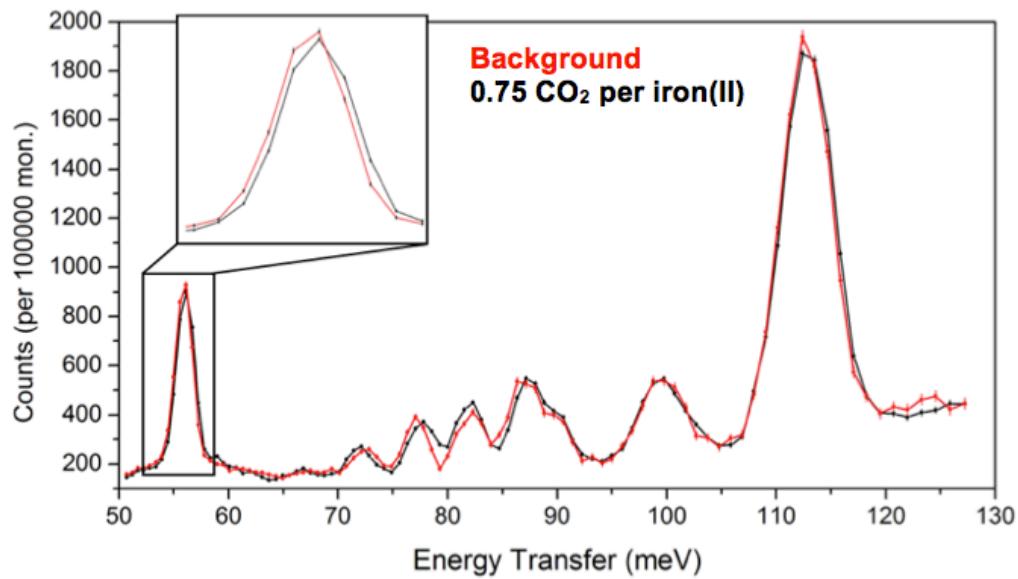


**Figure S23.** Unit cell parameters of  $\text{Fe}_2(\text{dobdc})$  loaded with 0.5  $\text{CO}_2$  per iron (II) (left) and then 1.5  $\text{CO}_2$  per iron(II) (right) plotted as a function of temperature.

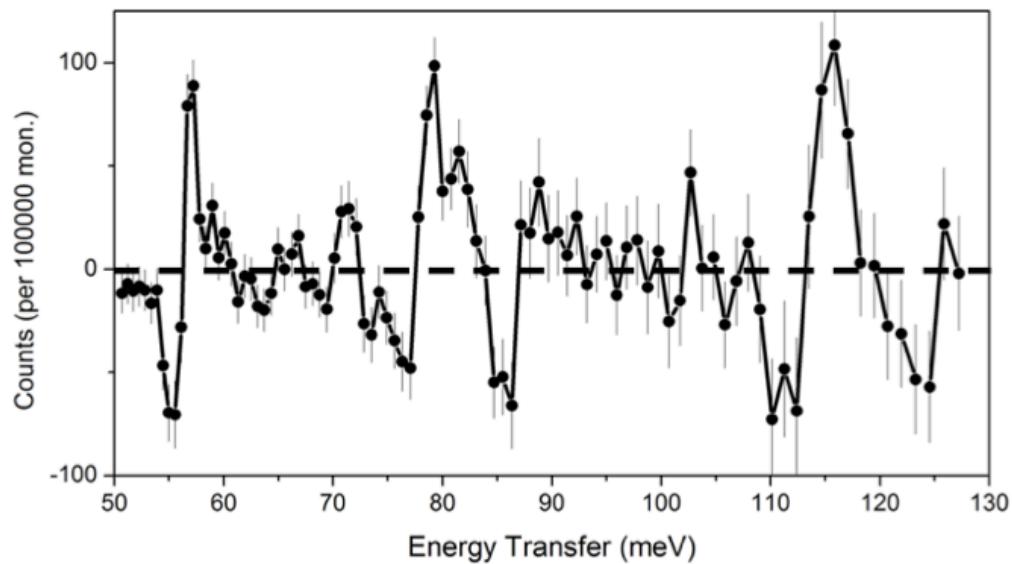


**Figure S24.** (a) Inelastic neutron scattering data of bare  $\text{Fe}_2(\text{dobdc})$  and then loaded with 0.75  $\text{CO}_2$  per iron. (b) Inelastic scattering data after the data obtained from the bare framework was subtracted from the  $\text{CO}_2$  loaded sample.

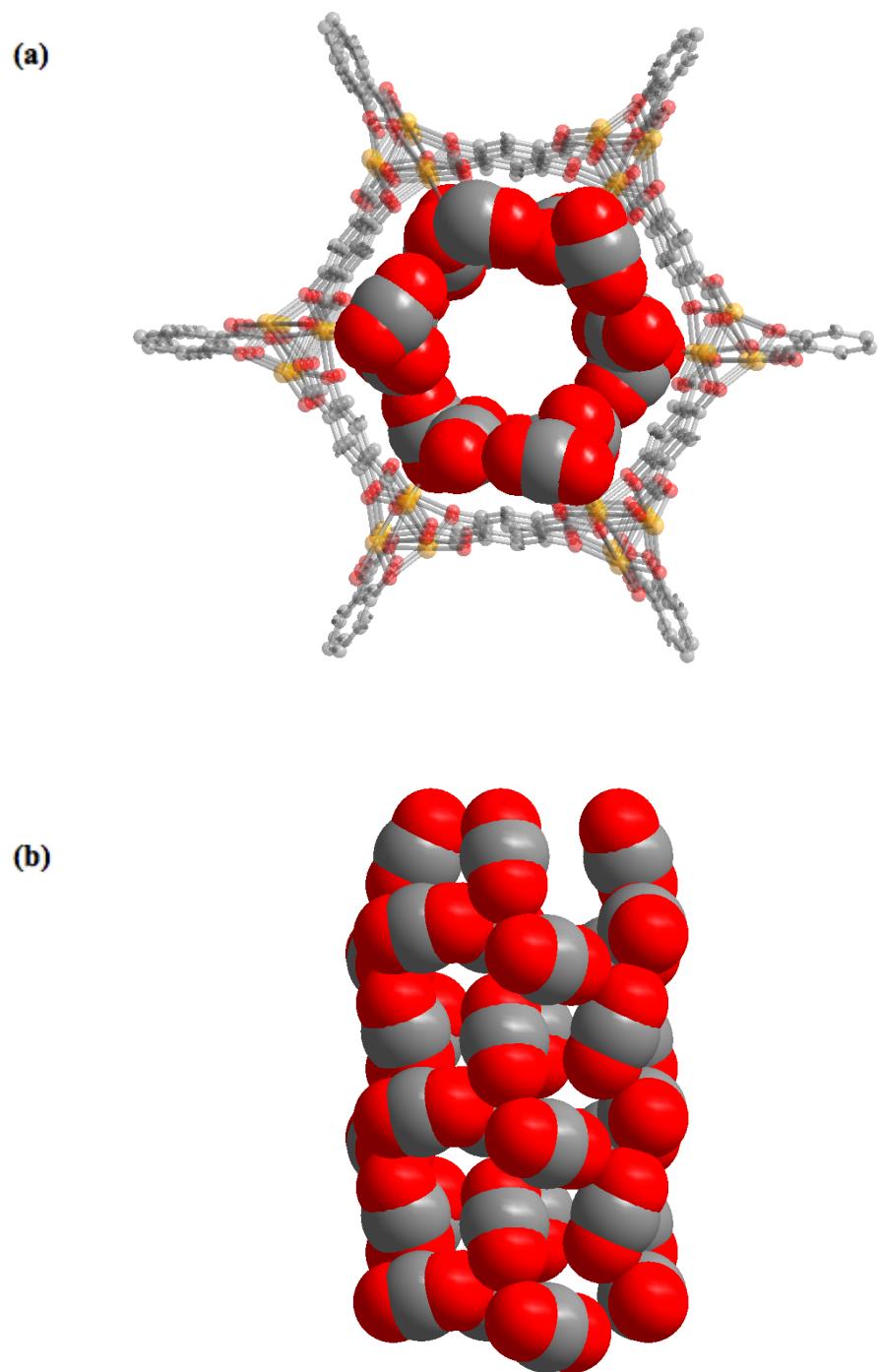
(a)



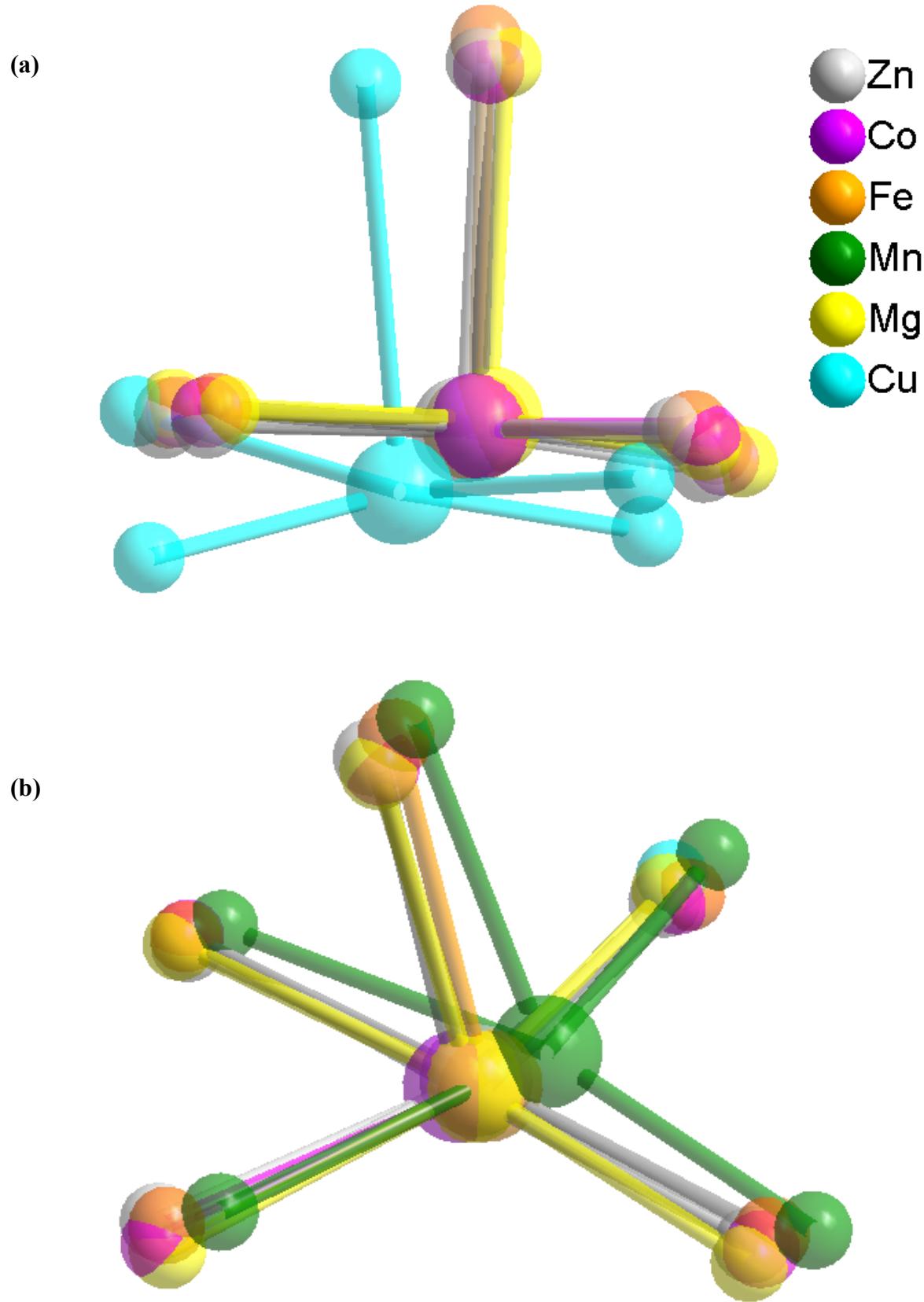
(b)



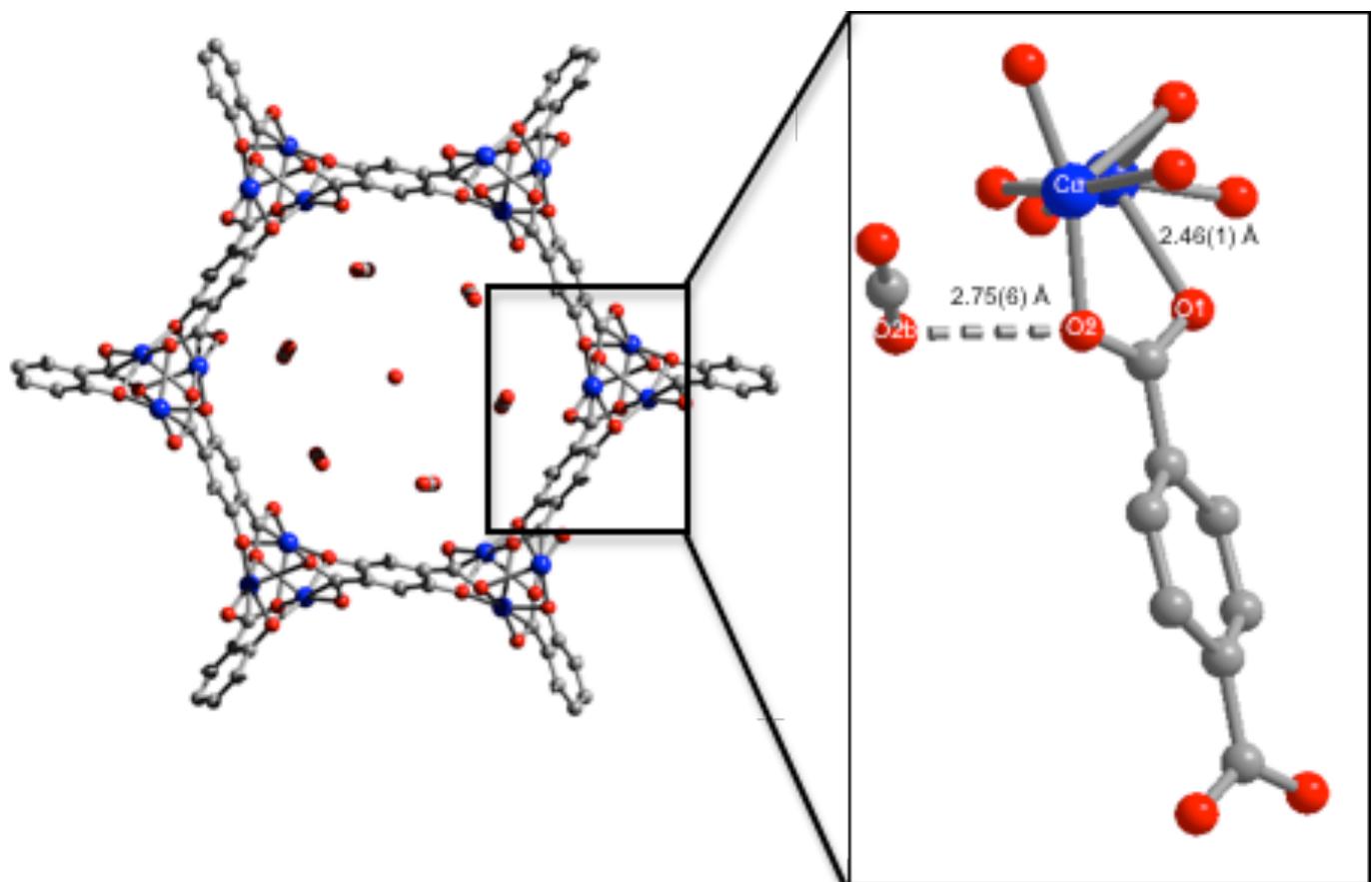
**Figure S25.** (a) Space filling model of  $\text{Fe}_2(\text{dobdc})$  loaded with 1.3  $\text{CO}_2$  per  $\text{Fe}^{2+}$  viewed along the channel axis,  $c$ , and (b) then along the  $b$ -axis, with the framework removed for clarity.



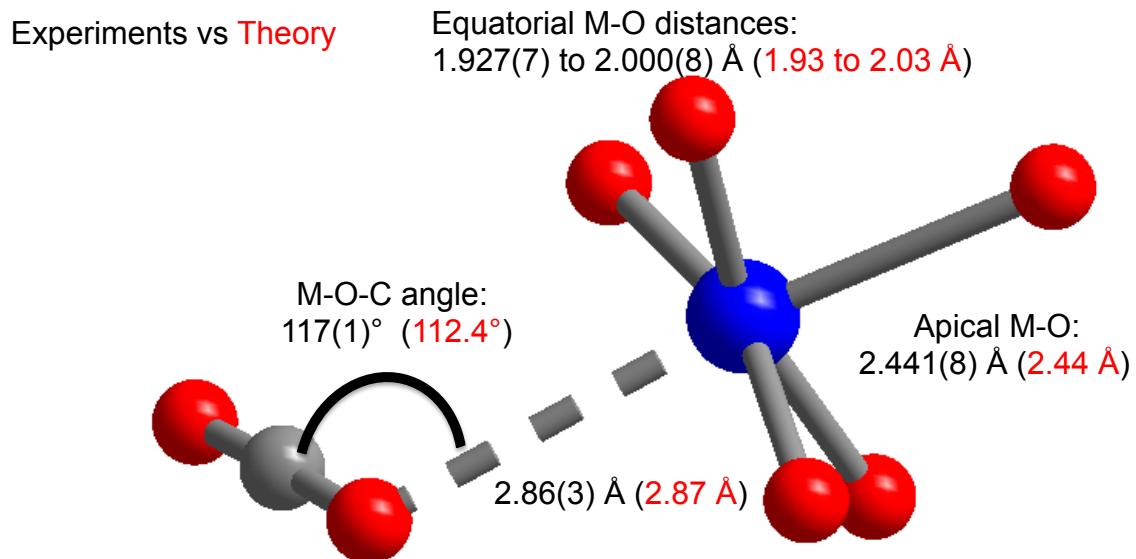
**Figure S26.** Ball and stick model showing an overlay of the  $\text{MO}_5$  polyhedra for the  $\text{M}_2(\text{dobdc})$  analogs for  $\text{M} = \text{Mg}, \text{Mn}, \text{Fe}, \text{Co}, \text{Cu}$ , and  $\text{Zn}$ . The top, (a), and bottom images, (b) reveal a significant distortion in  $\text{CuO}_5$  and  $\text{MnO}_5$  polyhedra relative to the other analogs. The two are presented separately for clarity.



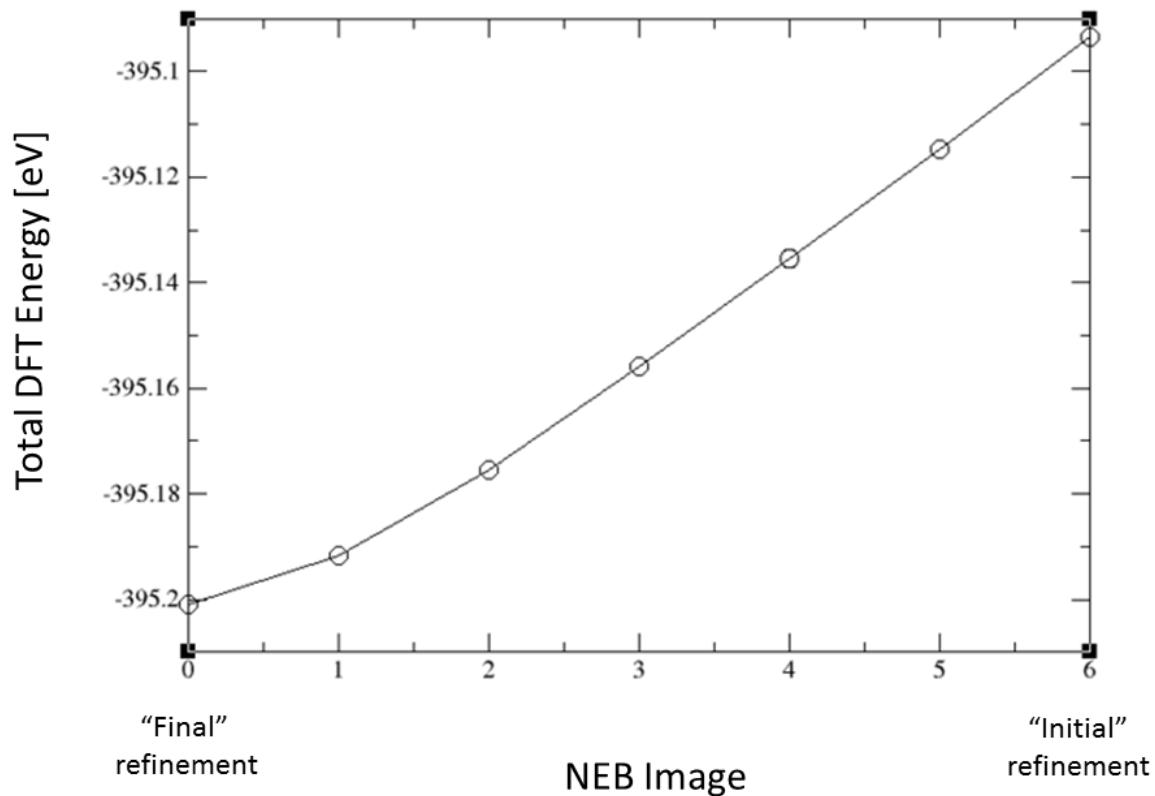
**Figure S27.** Ball and stick model showing the first structure solution of CO<sub>2</sub> adsorption in the Cu<sub>2</sub>(dobdc) framework with two CO<sub>2</sub> adsorption sites. This model was the result of a false minimum, a problem that was resolved once the third CO<sub>2</sub> adsorption site was identified.



**Figure S28.** Ball and stick model showing the Cu polyhedron with respect to the site I CO<sub>2</sub>. Comparison of experimental structural results (black) with theoretical DFT simulations (red) reveals excellent agreement.

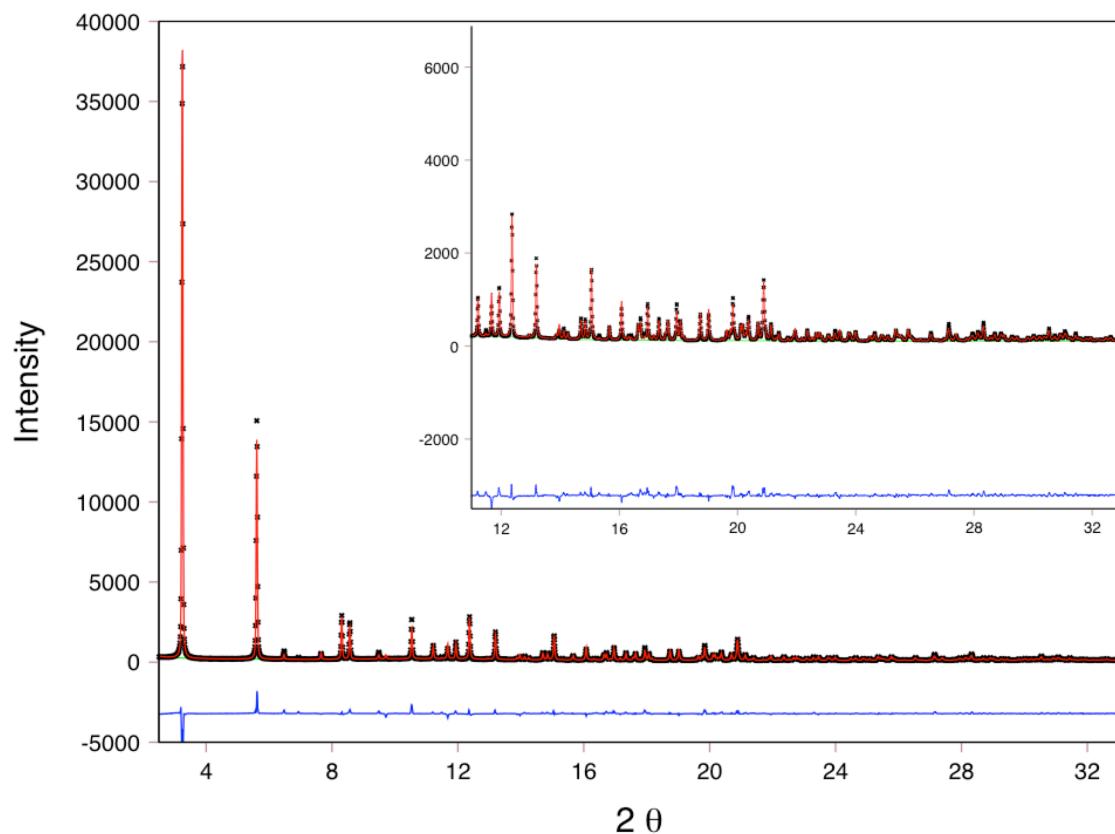


**Figure S29.** Nudged elastic band calculations, performed within DFT using vdW-DF2+*U* in line with all other calculations in this manuscript, show no barrier between the initial refinement structure and final refinement structure, supporting that the final refinement is likely the only “primary” binding site for CO<sub>2</sub> in Cu<sub>2</sub>(dobdc). Initial refinement structure is relaxed using DFT using a quasi-Newton relaxation scheme for 20 ionic steps to provide a starting image near the initial refinement; final refinement structure is taken as the vdW-DF2+*U*-predicted binding geometry, which agrees well with the final experimental refinement. Image structures for NEB are constructed using linear interpolation between the “initial” and “final” refinement structures as a starting point and allowed to relax.



## Synchrotron X-ray diffraction data for CO<sub>2</sub> dosed Cu<sub>2</sub>(dobdc)

**Figure S30.** Synchrotron X-ray powder diffraction data for CO<sub>2</sub> adsorbed in Cu<sub>2</sub>(dobdc). The green line, crosses, and red line represent the background, experimental, and calculated diffraction patterns, respectively. The blue line represents the difference between experimental and calculated patterns.



**Table S26.** Fractional atomic coordinates, occupancies, and isotropic displacement parameters obtained from Rietveld refinement of structural model of Cu<sub>2</sub>(dobdc) framework dosed with refined dose of 1.5 CO<sub>2</sub> per Cu, 150 K, Trigonal, R-3,  $a = 25.8978(3)$  Å,  $c = 6.2785(1)$  Å,  $V = 3646.76(8)$  Å<sup>3</sup>. Values in parentheses indicate one standard deviation in the refined value. Goodness-of-fit parameters:  $\chi^2 = 1.176$ , wRp = 0.0597, Rp = 0.0442

Atom	X	Y	Z	Occupancy	U <sub>(ISO)</sub> (Å <sup>2</sup> )	Multiplicity
Cu	0.3874(1)	0.3492(1)	0.1751(4)	1.0	0.0126(6)	18
O1	0.3544(4)	0.3004(4)	0.426(2)	1.0	0.014(2)	18
O2	0.2899(4)	0.2255(4)	0.633(2)	1.0	0.014(2)	18
O3	0.3624(4)	0.2781(4)	0.012(2)	1.0	0.014(2)	18
C1	0.3284(8)	0.2478(6)	0.485(2)	1.0	0.011(4)	18
C2	0.3289(8)	0.2082(8)	0.304(3)	1.0	0.011(4)	18
C3	0.3479(8)	0.2225(5)	0.091(3)	1.0	0.011(4)	18
C4	0.3546(6)	0.1838(8)	-0.024(2)	1.0	0.011(4)	18
C1a	0.271(1)	0.749(2)	0.527(6)	0.53(1)	0.18(2)	18
O1a	0.221(1)	0.727(2)	0.564(9)	0.53(1)	0.16(3)	18
O1b	0.1711(9)	0.702(1)	0.615(5)	0.53(1)	0.10(2)	18
C2a	0.148(1)	0.002(1)	0.349(4)	0.51(1)	0.12(1)	18
O2a	0.154(2)	0.001(2)	0.530(4)	0.51(1)	0.12(1)	18
O2b	0.160(1)	0.004(2)	0.711(4)	0.51(1)	0.11(2)	18
C3b	0.0	0.0	0.3163(9)	0.58(2)	0.16(2)	3
O3b	0.0	0.0	0.5	0.58(2)	0.35(9)	6

## Coordinate files for DFT structures of CO<sub>2</sub> in M<sub>2</sub>(dobdc)

Coordinates are provided in the format of VASP POSCAR files with nuclear coordinates provided as fractional (Direct) coordinates in terms of the crystallographic basis vectors. Selective Dynamics are enabled to allow only the CO<sub>2</sub> molecule to relax (with vdW-DF2+U) within the rigid (PBE+U) M<sub>2</sub>(dobdc).

```
MgMOF74-CO2
1.000000000000000
 15.2259346732869041    0.0050373306177292    0.0186555174857414
 -7.0913762758806689   13.4745115715249604    0.0182226035629158
 -7.0916724110859466  -11.7516324991570080    6.5928931085496609
C      H      O      Mg
 25       6     20       6
Selective dynamics
Direct
 0.7482245082990033  0.3534019561877599  0.1812378289473671  F  F  F
 0.6178686864118532  0.1681233715252830  0.0821353813181318  F  F  F
 0.4357333985898890  0.9695266986682114  0.8683503946494255  F  F  F
 0.3242964642812183  0.8076126796414442  0.7941651238854277  F  F  F
 0.1812417570426135  0.7482124708854769  0.3533874975116333  F  F  F
 0.0821432447687300  0.6178660181570308  0.1681074617961187  F  F  F
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 0.6322933931195180  0.1858446243288014  0.6553487757883403  F  F  F
 0.6553404514988586  0.6322932698972679  0.1857950510331037  F  F  F
 0.8141950235177688  0.3446830202774436  0.3677091014705951  F  F  F
 0.3677211439898969  0.8142100503972571  0.3446597515296901  F  F  F
 0.3446593118364092  0.3677221543508509  0.8141866785824874  F  F  F
 0.6981865926942703  0.3333917203738110  0.0692480301420986  F  F  F
 0.9043833218257902  0.5264829859982925  0.3701530491344798  F  F  F
 0.3662307749504663  0.9306573408865049  0.7366377267545516  F  F  F
 0.0692521847562730  0.6981793081050398  0.3333827932116620  F  F  F
 0.3701539145218646  0.9043621240139359  0.5264629248805193  F  F  F
 0.7366440236451126  0.3662253009791741  0.9306487282664975  F  F  F
 0.3333867003273170  0.0692587353055814  0.6981855480007439  F  F  F
 0.5264653972743858  0.3701612556937306  0.9043783722804832  F  F  F
```

0.9306492271069828	0.7366501298752794	0.3662224046151437	F	F	F
0.3018462195926972	0.6666441173467348	0.9307425348348488	F	F	F
0.0956399187615062	0.4735534469652762	0.6298350111775761	F	F	F
0.6338065183530972	0.0693844350988755	0.2633559683332010	F	F	F
0.9307653118267609	0.3018601266864636	0.6666226325414897	F	F	F
0.6298659855286601	0.0956917531533250	0.4735391504587838	F	F	F
0.2633719239128283	0.6338446853007440	0.0693546518174202	F	F	F
0.6666289119963764	0.9307756483485861	0.3018102551604684	F	F	F
0.4735433006407632	0.6298785805458209	0.0956584473233519	F	F	F
0.0693631109866644	0.2633795254480518	0.6337875662267436	F	F	F
0.279786042429261	0.8601593501446280	0.1047223474090337	T	T	T
0.2771921593065050	0.9786977964833398	0.2702874042150862	T	T	T
0.5197348673738986	0.1144740833773241	0.7873118502351915	F	F	F
0.7873126286326126	0.5197330899686392	0.1144702234295565	F	F	F
0.1144777633769891	0.7873265496067319	0.5197331478490881	F	F	F
0.4802955162322178	0.8855815579437945	0.2126891024534885	F	F	F
0.2127046167967208	0.4803123212502456	0.8855431458900682	F	F	F
0.8855577438543705	0.2127220547985900	0.4802804552218163	F	F	F

## MnMOF74-CO2

1.00000000000000  
 15.5129570444968383 0.0199324128806612 0.0317033892838443  
 -7.2060776324217928 13.6874753160057345 0.0549078577108840  
 -7.2254709284679297 -11.9118042507125974 6.7625285861417348

C H O Mn  
 25 6 20 6

## Selective dynamics

## Direct

0.0306008559147841	0.1297024176357624	0.5651282874262407	F	F	F
0.0769903566474781	0.6125416226700082	0.1623329971843503	F	F	F
0.1296233121273929	0.5645235596485776	0.0307817472222851	F	F	F
0.1624415613546546	0.0769694871596798	0.6121108053498432	F	F	F
0.1725356950349592	0.7387697072001131	0.3426765273500152	F	F	F
0.1868126950803415	0.1990470305111387	0.6711105959414070	F	F	F
0.1989897728008572	0.6709508580959849	0.1868847047478965	F	F	F
0.2611848572977635	0.6571104216220007	0.8278062181302559	F	F	F
0.3289965668931387	0.8132083359259710	0.8010643276113001	F	F	F
0.3428652600948112	0.1724988400199408	0.7380093146367699	F	F	F
0.3874293546228387	0.8375528167763093	0.9231736200095639	F	F	F
0.4354514306002031	0.9694324076567327	0.8702707416854381	F	F	F
0.5645485393997305	0.0305676113433222	0.1297292583145477	F	F	F
0.6125706453770547	0.1624471982236813	0.0768264029904628	F	F	F
0.6571347699052410	0.8275011449799052	0.2619907153632823	F	F	F
0.6710034331067476	0.1867916640740432	0.1989356873887616	F	F	F
0.7388151727022674	0.3428895783781059	0.1721938118696897	F	F	F
0.8010102271990860	0.3290491719040389	0.8131153102520798	F	F	F
0.8131872909197142	0.8009529694889324	0.3288894040586143	F	F	F
0.8274642899649862	0.2612302927998158	0.6573234426500250	F	F	F
0.8375584236453193	0.9230304978402728	0.3878891656500798	F	F	F
0.8703766878726071	0.4354764403512661	0.9692182427777070	F	F	F
0.9230096203524667	0.3874583773300415	0.8376669878156733	F	F	F
0.9693991680850544	0.8702975823641808	0.4348717125738020	F	F	F
0.2867182211167814	0.9354465838047624	0.2125142070049808	T	T	T
0.1940454050261522	0.6651274496466328	0.6443221110308102	F	F	F
0.3348912587779722	0.3558814371571231	0.8061346793294391	F	F	F
0.3557770387169938	0.8058644704015023	0.3350362596523766	F	F	F
0.6442229612829138	0.1941355145985426	0.664963703476472	F	F	F
0.6651087412220207	0.6441185628429764	0.1938653056705988	F	F	F
0.8059545949738691	0.3348725203533789	0.3556778889692609	F	F	F
0.0654193243573928	0.6917182316764610	0.3263381856069429	F	F	F
0.0706773255873543	0.2605559818555179	0.6358047063260415	F	F	F
0.110318785596183	0.4912819694054136	0.6439953899449193	F	F	F
0.2604757202878503	0.6347512241579594	0.0711149603623156	F	F	F
0.3082612291379974	0.6732228614026212	0.9350065998597827	F	F	F
0.3267432165123125	0.0653793421435935	0.6908672570424201	F	F	F
0.3652231352544320	0.9293321004963317	0.7393490240948353	F	F	F
0.3563869472215231	0.8895852114165166	0.5086274953045802	F	F	F
0.4913401911702024	0.6436544465356349	0.1113032507203258	F	F	F
0.5086598088296341	0.3563455234644337	0.8886967262797043	F	F	F
0.6436130527784982	0.1104147955834733	0.4913724746954813	F	F	F
0.6347768647455894	0.0706678845037061	0.2606509759052145	F	F	F
0.6732568134877539	0.9346206798563159	0.3091327719575716	F	F	F
0.6917388008618772	0.3267771085974118	0.0649934071401930	F	F	F
0.7395242797122776	0.3652487758420548	0.9288850396377342	F	F	F
0.8896812294034646	0.5087180605946457	0.3560046390550085	F	F	F
0.9293226964127186	0.7394440181445319	0.3641953236739610	F	F	F

0.9345806976428079	0.3082817983235628	0.6736618443929459	F	F	F
0.2922521256672325	0.9957364714278554	0.3003452131594808	T	T	T
0.2793387081643409	0.8743010856453193	0.1219770622603664	T	T	T
0.1053119502892983	0.7781768366533797	0.5154049616204972	F	F	F
0.2217813908492374	0.4840053959488202	0.8954769451737477	F	F	F
0.4839927394766903	0.8946664789977561	0.2220590064470755	F	F	F
0.5160072605233523	0.1053335140022611	0.7779409635529504	F	F	F
0.7782186241507389	0.5159946040512295	0.1045230478262980	F	F	F
0.8946880427107047	0.2218231633466061	0.4845950083794008	F	F	F

## FeMOF74-CO2

1.00000000000000  
 15.4545641719544555 0.0070457894495687 -0.0240572322099241  
 -7.1903018448809917 13.6184022160005789 0.0023475278054387  
 -7.2125667559608564 -11.8788526292561869 6.6845211247248297

C H O Fe  
 25 6 20 6

## Selective dynamics

## Direct

0.0319829952171773	0.1314413009281239	0.5662083736980463	F	F	F
0.0810330893083417	0.6177681586549113	0.1660647822298813	F	F	F
0.1313729932887568	0.5655286979596781	0.0322419307980653	F	F	F
0.1662293188708617	0.0810420954600204	0.6172554641843675	F	F	F
0.1794476037161274	0.7487115038387273	0.3495089527043120	F	F	F
0.1921438721906057	0.2049156727429065	0.6771965956873629	F	F	F
0.2048328249440345	0.6770390507102135	0.1922448081645030	F	F	F
0.2512907366009358	0.6501990799956232	0.8209418832497448	F	F	F
0.3229006585434959	0.8078568150945600	0.7952188557193978	F	F	F
0.3497837425251689	0.1794559623153589	0.7477905179389595	F	F	F
0.3822323894418105	0.8337809692271918	0.9191546289617136	F	F	F
0.4344224187409722	0.9680240952654273	0.8685000440575124	F	F	F
0.5655775512590466	0.0319759237345565	0.1314999559424663	F	F	F
0.6177676105581540	0.1662190457727988	0.0808453940382492	F	F	F
0.6502162874748336	0.8205440226846434	0.2522095120610786	F	F	F
0.677093414564614	0.1921431849054045	0.2047811592806141	F	F	F
0.7487092933990667	0.3498009200044194	0.1790581467502790	F	F	F
0.7951671750559370	0.3229609792897890	0.8077552068354947	F	F	F
0.8078561138094500	0.7950843272571575	0.3228034043126087	F	F	F
0.8205523812838535	0.2512884961612443	0.6504910172957210	F	F	F
0.8337706661290980	0.9189578895399464	0.3827445068156266	F	F	F
0.8686270067112005	0.4344713020403148	0.9677580592018984	F	F	F
0.9189668876916599	0.3822318413451171	0.8339352027701210	F	F	F
0.9680170287828602	0.8685586990718548	0.4337916263019608	F	F	F
0.2908047246193611	0.9395630015804269	0.2158201710183315	T	T	T
0.1834535634726180	0.6558912638242731	0.6342536669692151	F	F	F
0.3440709735056018	0.3659906796779779	0.8166848394496355	F	F	F
0.3658421239398990	0.8164387503069079	0.3442934229879739	F	F	F
0.6341578760600584	0.1835612346930944	0.6557066070120356	F	F	F
0.6559290264943627	0.6340093203220576	0.1833151455503739	F	F	F
0.8165464365273536	0.3441087061757528	0.3657463330307849	F	F	F
0.0691363158154772	0.7005406935938439	0.3294508408723473	F	F	F
0.0732493421441092	0.2640932822156685	0.6381365187916757	F	F	F
0.0964263242752921	0.4786175197364599	0.6329592202100969	F	F	F
0.2640208495994543	0.6369561247723965	0.0738115184107784	F	F	F
0.2995626735591799	0.6699755653385111	0.9313557165855428	F	F	F
0.3299821249769579	0.0691018490264881	0.6994129247946219	F	F	F
0.3630137044856738	0.9267370615894919	0.7357719853726792	F	F	F
0.3674508054094048	0.9034441945545026	0.5212223148496236	F	F	F
0.4786785599016099	0.6325061378083774	0.0975532677800928	F	F	F
0.5213214400983688	0.3674938321916059	0.9024467092198734	F	F	F
0.6325491945906734	0.0965558124455157	0.4787776551503882	F	F	F
0.6369862955143333	0.0732629234104820	0.2642280146273137	F	F	F
0.6700179050230304	0.9308981729734498	0.3005871042053840	F	F	F
0.7004373564408013	0.3300244046614580	0.0686442904144471	F	F	F
0.7359791504005671	0.3630438752276106	0.9261884815892429	F	F	F
0.9035736907247269	0.5213825102635425	0.3670408087898807	F	F	F
0.9267506798559495	0.7359067177843244	0.3618635112083197	F	F	F

0.9308637061845602	0.2994593364061302	0.6705491891276409	F	F	F
0.3026164101381867	0.0035544311197617	0.3100686995920389	T	T	T
0.2765200929552958	0.8742449748185608	0.1185463829756990	T	T	T
0.1080689075125250	0.7837676131637252	0.5160718035140377	F	F	F
0.2162072558736199	0.4831830759801790	0.8928891785043618	F	F	F
0.4831982614024994	0.8919137128118848	0.2165299760004942	F	F	F
0.5168017385975006	0.1080862801880969	0.7834699939995033	F	F	F
0.7837927591263707	0.5168169240197926	0.1071108144956554	F	F	F
0.8919310854874567	0.2162323868362535	0.4839281664859598	F	F	F

## CoMOF74-CO2

1.00000000000000  
 15.2999234049848614 -0.0025964635924967 -0.0152724116282011  
 -7.1265714412560213 13.5240970429364609 -0.0076107863835677  
 -7.1290723882713749 -11.7911518743697279 6.6270129221598237

C H O Co  
 25 6 20 6

## Selective dynamics

## Direct

0.0307235812173445	0.1312458441701807	0.5620598820746352	F	F	F
0.0841817790693042	0.6204342775496485	0.1698760448156378	F	F	F
0.1311721904240812	0.5619138975952040	0.0307982752504543	F	F	F
0.1699170226882671	0.0841158998703762	0.6205270926759354	F	F	F
0.1857671906184493	0.7542745520685088	0.3571364768236762	F	F	F
0.1950231400610178	0.2082580885759313	0.6766254383150709	F	F	F
0.2082051092836750	0.6763332562595892	0.1950120788918142	F	F	F
0.2455281974276744	0.6428975560187311	0.8144166959135788	F	F	F
0.3237935595509569	0.8050165920011239	0.7918770082607622	F	F	F
0.3571941110299974	0.1855937201221849	0.7543442944170948	F	F	F
0.3794938455400612	0.8301496225816294	0.9159240211438799	F	F	F
0.4382826237317232	0.9692055775168882	0.8688045163577414	F	F	F
0.5617173462682814	0.0307944414831027	0.1311954836422586	F	F	F
0.6205061544599388	0.1698503924183683	0.0840760018561184	F	F	F
0.6428059189700051	0.8144062648778103	0.2456557355829077	F	F	F
0.6762064404490360	0.1949834079988761	0.2081230067392354	F	F	F
0.7544718325723281	0.3571024439812689	0.1855833340864237	F	F	F
0.7917948907163250	0.3236667737404133	0.8049879361081906	F	F	F
0.8049768459389881	0.7917419114240758	0.3233745616849220	F	F	F
0.8142327943815459	0.2457254479314912	0.6428634931763213	F	F	F
0.8300829623117281	0.9158840851296262	0.3794728783240657	F	F	F
0.8688278095759188	0.4380861024047960	0.9692017147495449	F	F	F
0.9158181979306974	0.3795657224503444	0.8301239401843716	F	F	F
0.9692764427826575	0.8687541558298193	0.4379401179253577	F	F	F
0.2812196075268043	0.9269058190439682	0.2048837803493750	T	T	T
0.1851591590794470	0.6511083735234280	0.6284814218952164	F	F	F
0.3489323117785332	0.3717368281163047	0.8155608132923859	F	F	F
0.3716801542269721	0.8150572370820726	0.3489346412903060	F	F	F
0.6283198457730279	0.1849427479179298	0.6510653887096964	F	F	F
0.6510676882214668	0.6282631718836953	0.1844391717076093	F	F	F
0.8148408409205530	0.3488915964765766	0.3715185781047836	F	F	F
0.0772833544378813	0.7110349214399392	0.3402363262213797	F	F	F
0.0713084109012243	0.2644051961879157	0.6311257215992470	F	F	F
0.0914714003741111	0.4695233405337689	0.6236129423727874	F	F	F
0.2643709552435638	0.6309798998328588	0.0715899320038744	F	F	F
0.2883905986596247	0.6597005735624393	0.9228102054134908	F	F	F
0.3404223493790326	0.0770924929049883	0.7111422475710469	F	F	F
0.3693546415197559	0.9284002445913089	0.7355621701304287	F	F	F
0.3767015655319810	0.9085438774675296	0.5305712298972978	F	F	F
0.4694754238837220	0.6235532475533176	0.0914690978097425	F	F	F
0.5305245761162780	0.3764467224466870	0.9085308791902520	F	F	F
0.6232984344680190	0.0914561295324674	0.4694287401027069	F	F	F
0.6306453584802370	0.0715997404086863	0.2644378298695713	F	F	F
0.6595776806209699	0.9229075290950135	0.2888577814289519	F	F	F
0.7116094313403778	0.3402993964375582	0.0771898015865062	F	F	F
0.7356290447564362	0.3690201001671412	0.9284100679961256	F	F	F
0.9085286146258866	0.5304766894662336	0.3763870866272043	F	F	F
0.9286916110987775	0.7355948038120914	0.3688743084007484	F	F	F

0.9227166675621135	0.2889651085600562	0.6597637037786228	F	F	F
0.2987196771073407	0.9973874809012497	0.3044153781234868	T	T	T
0.2607767970338190	0.8546568080282415	0.1019204056595665	T	T	T
0.1193058017889683	0.7911076347622554	0.5271431754193827	F	F	F
0.2087340834764362	0.4728736100863884	0.8810431605153184	F	F	F
0.4725798471553020	0.8809420118155700	0.2088285666762175	F	F	F
0.5274201528446980	0.1190579811844330	0.7911714033237800	F	F	F
0.7912659315235615	0.5271263899136116	0.1189568324846846	F	F	F
0.8806941912110275	0.2088923652377446	0.4728567945806148	F	F	F

## NiMOF74-CO2

1.00000000000000  
 15.1665707574288984 -0.0056678327150128 -0.0154456877922552  
 -7.0674392096184029 13.4120106915082538 -0.0120535880838855  
 -7.0649648953754909 -11.6929598734154112 6.5662448745405122

C H O Ni  
 25 6 20 6

## Selective dynamics

## Direct

0.0311529640848534	0.1327372115865018	0.5592486767310945	F	F	F
0.0851849946379062	0.6236813093347635	0.1719589348439001	F	F	F
0.1327024795773113	0.5589254726416826	0.0313713097080281	F	F	F
0.1720654272343083	0.0852460438923330	0.6234008179399595	F	F	F
0.1922170318993039	0.7651698581488162	0.3641974748487371	F	F	F
0.1966744230243194	0.2095437347987428	0.6755729320798238	F	F	F
0.2094470613717263	0.6755272320352290	0.1968069828779591	F	F	F
0.2345676376142123	0.6355130849341961	0.8077835412822338	F	F	F
0.3240837200333857	0.8032251669041983	0.7904289717343431	F	F	F
0.3643734818897784	0.1923281867180293	0.7646049382074622	F	F	F
0.3760284249939119	0.8278513381072585	0.9147498406739558	F	F	F
0.4410177975149665	0.9688574841893782	0.8672611230711169	F	F	F
0.5589835262841945	0.0311439830858617	0.1327400964222889	F	F	F
0.6239705268604681	0.1721480197956424	0.0852498244490647	F	F	F
0.6356264904946869	0.8076719292202341	0.2353947326224173	F	F	F
0.6759158986586655	0.1967741930181006	0.2095706956504273	F	F	F
0.7654333324925418	0.3644876263250865	0.1922176747660558	F	F	F
0.7905539087350206	0.3244734792240465	0.8031942331703377	F	F	F
0.8033251956677390	0.7904556251235562	0.3244267353049395	F	F	F
0.8077829404851613	0.2348302577894472	0.6358021959811424	F	F	F
0.8279341914577500	0.9147533160299659	0.3765988494448038	F	F	F
0.8672984905294427	0.4410752386175858	0.9686299063402686	F	F	F
0.9148146240541521	0.3763180505875354	0.8280407325408703	F	F	F
0.9688466546072050	0.8672621483357972	0.44075090906536689	F	F	F
0.2767025602661601	0.9216083842447368	0.1988911562392541	T	T	T
0.1850500206660897	0.6474589815376532	0.6254129441446992	F	F	F
0.3523560513138762	0.3745380416843389	0.8143303035338931	F	F	F
0.3743466998090668	0.8142448865205125	0.3526264054442620	F	F	F
0.6256542702976802	0.1857558247387630	0.6473748106040276	F	F	F
0.6476435673781751	0.6254613182379600	0.1856693638508773	F	F	F
0.8149499517183756	0.3525411344006173	0.3745867266851803	F	F	F
0.0866692984104276	0.7303736335935582	0.3512202293873585	F	F	F
0.0761796359770273	0.2700048820363534	0.6288119716154696	F	F	F
0.0803810584160658	0.4610852703408526	0.6157757080486164	F	F	F
0.2698818016026934	0.628179071168521	0.0765369258877016	F	F	F
0.2698928023035592	0.6484122957759908	0.9135798352097098	F	F	F
0.3515324114222551	0.0867066663380527	0.7296785966310182	F	F	F
0.3717874067502791	0.9239244101945090	0.7300645869663711	F	F	F
0.3840961400444627	0.9190060028508711	0.5386676979118903	F	F	F
0.4613083829502571	0.6157732132911846	0.0817037550663642	F	F	F
0.5386912357417941	0.3842261466311143	0.9182959123184062	F	F	F
0.6159034786475956	0.0809933570714279	0.4613319694728730	F	F	F
0.6282125656341861	0.0760757057437544	0.2699350838635084	F	F	F
0.6484685586844918	0.9132940449212299	0.2703226194172714	F	F	F
0.7301071700809061	0.3515878201622797	0.0864198356201697	F	F	F
0.7301181707817719	0.3718202088214113	0.9234627449421779	F	F	F
0.9196185602759854	0.5389140895814464	0.3842239593361469	F	F	F
0.92381199827150311	0.7299944778859526	0.3711876957693008	F	F	F

0.9133303202816307	0.2696257263287407	0.6487794379974048	F	F	F
0.2977627224918176	0.9983284434669361	0.3012543655220274	T	T	T
0.2523835238425712	0.8428052525801015	0.0927611533721238	T	T	T
0.1256625102081657	0.8005563640885143	0.5328414664898844	F	F	F
0.1995185454526833	0.4667866508377756	0.8749607148974050	F	F	F
0.4668561668708335	0.8743289481447363	0.1996589056222007	F	F	F
0.5331434518212177	0.1256704117775627	0.8003407617625697	F	F	F
0.8004810732393750	0.5332127090845304	0.1250389524873654	F	F	F
0.8743374621762996	0.1994437518497492	0.4671582043399951	F	F	F

CuMOF74-CO<sub>2</sub>

1.000000000000000  
 15.2489455497523032 -0.0475266532617733 -0.1841801506346098  
 -7.1427936579844600 13.4679987913793848 -0.1819755501450751  
 -7.1442677751025450 -11.8272175991350359 6.4468462915043547

C H O Cu  
 25 6 20 6

## Selective dynamics

## Direct

0.0363397651295756	0.1357807304996399	0.5649584299637596	F	F	F
0.0915842000821101	0.6245426656135180	0.1776719716957516	F	F	F
0.1357749449038224	0.5649129421464352	0.0363642529427253	F	F	F
0.1776775069524561	0.0915850479358298	0.6244949243096301	F	F	F
0.2050472149976343	0.7770202428997450	0.3764212513936229	F	F	F
0.2082978788110594	0.2196895385592157	0.6816039384745309	F	F	F
0.2196733744440564	0.6815989863515242	0.2083085591967659	F	F	F
0.2229796323272879	0.6235497924860240	0.7949726935487220	F	F	F
0.3183949370542720	0.7916970755408101	0.7803309171586363	F	F	F
0.3764501841122296	0.2050606653733453	0.7769500582206561	F	F	F
0.3754647005767708	0.8223220281733816	0.9084343376162920	F	F	F
0.4350891502629324	0.9636663935061307	0.8642256210034560	F	F	F
0.5649121735362428	0.0363350737691022	0.1357755984899356	F	F	F
0.6245342512776020	0.1776773297295406	0.0915653275067143	F	F	F
0.6235497882722214	0.7949394505649110	0.2230496126092163	F	F	F
0.6816046816377721	0.2083022843815243	0.2196687502261199	F	F	F
0.7770213377794448	0.3764509187732941	0.2050285224995747	F	F	F
0.7803275956626550	0.3184017249077300	0.7916926568515734	F	F	F
0.7917017398810060	0.7803098213630832	0.3183957289101969	F	F	F
0.7949527573868309	0.2229798730385326	0.6235784194362353	F	F	F
0.8223221117396378	0.9084143119865047	0.3755047430751119	F	F	F
0.8642260252029459	0.4350877691128474	0.9636369631055572	F	F	F
0.9084154186099340	0.3754566943087312	0.8223276956890189	F	F	F
0.9636598535624827	0.8642186294226519	0.4350412374210038	F	F	F
0.2508669139064027	0.9215385322472013	0.1955184898134235	T	T	T
0.1747320071246676	0.6279375029125092	0.6078980218811765	F	F	F
0.3720555896119961	0.392139894832779	0.8252650502027379	F	F	F
0.3921183310014626	0.8252643628569984	0.3720822693107664	F	F	F
0.6078826391053269	0.1747363484022770	0.6279189467375019	F	F	F
0.6279440290800551	0.6078594650890210	0.1747346171820396	F	F	F
0.8252679652598047	0.3720626130257614	0.3921016489486959	F	F	F
0.1226732090693048	0.7819267862613515	0.3750055034155224	F	F	F
0.0829484745496529	0.2741411098580500	0.6403325365585175	F	F	F
0.0888658688150699	0.4484444548347497	0.6084807857407029	F	F	F
0.2741307534031563	0.6402506639258263	0.0829889756178304	F	F	F
0.2180649304916713	0.6249232443420851	0.8773355345173499	F	F	F
0.3750733569189393	0.1226747216912329	0.7818519650259574	F	F	F
0.3597455317276754	0.9170551551912567	0.7258588175872021	F	F	F
0.3915479670299078	0.9111345143622955	0.5515520609927478	F	F	F
0.4484481151732709	0.6084352510562141	0.0889475580104033	F	F	F
0.5515515035187875	0.3915641088660777	0.9110521093743600	F	F	F
0.6084516516621434	0.0888648455600176	0.4484476063920226	F	F	F
0.6402544406568254	0.0829449607469641	0.2741408532427130	F	F	F
0.6249276131878148	0.8773259895680567	0.2181492510223393	F	F	F
0.7819350418928153	0.3750768715962209	0.1226641363125580	F	F	F
0.7258692189813232	0.3597494520124584	0.9170106952120562	F	F	F
0.9111337498770240	0.5515549050875848	0.3915188816440249	F	F	F
0.9170511441424125	0.7258582500642135	0.3596671308262529	F	F	F

0.8773264096227464	0.2180725736609048	0.6249941639692409	F	F	F
0.2939474923103809	0.0036760334301538	0.3189373499393171	T	T	T
0.2048275427474507	0.8372978512257695	0.0695284776339858	T	T	T
0.1311401192118353	0.8208025050936243	0.5549787541826419	F	F	F
0.1791932572849930	0.4449604298710454	0.8689302597577182	F	F	F
0.4449657479187650	0.8688517549885688	0.1792063854443029	F	F	F
0.5550338707733005	0.1311476049337585	0.8207932819404817	F	F	F
0.8208063614070653	0.5550389300512606	0.1310694076270593	F	F	F
0.8688598531726655	0.1791976108446462	0.4450209166473087	F	F	F

## ZnMOF74-CO2

1.00000000000000		
15.2439154824645406	0.0096688540602981	0.0385957437582283
-7.0956567272266406	13.4918402437420184	0.0377689953887450
-7.0950190972763432	-11.7567279111508967	6.6193482706989686

C	H	O	Zn
25	6	20	6

## Selective dynamics

## Direct

0.7429966507110777	0.3527355326125701	0.1814537092950133	F	F	F
0.6143076949952260	0.1676559665477058	0.0821830680629319	F	F	F
0.4360812802994758	0.9726672922706143	0.8713626107554759	F	F	F
0.3274353813610418	0.8105584335412388	0.7960720254665432	F	F	F
0.1814398393681813	0.7429868934398840	0.3527026170229348	F	F	F
0.0822060437256198	0.6143246040535502	0.1676420111897201	F	F	F
0.8713664246340045	0.4360857529896478	0.9726477628719223	F	F	F
0.7960851053077249	0.3274451869216932	0.8105386495272811	F	F	F
0.3527131771599201	0.1814585395910413	0.7429772358697448	F	F	F
0.1676496712942921	0.0822092505958736	0.6143036733840148	F	F	F
0.9726481988238973	0.8713739904698841	0.4360644222521302	F	F	F
0.8105542365610958	0.7961003481050426	0.3274305988282222	F	F	F
0.2570353561903147	0.6473068396434911	0.8185517684852286	F	F	F
0.3857183454994413	0.8323804097263192	0.9178113509721300	F	F	F
0.5639463901302122	0.0273714078996932	0.1286346835360845	F	F	F
0.6725933993346231	0.1894803195321799	0.2039254524652563	F	F	F
0.8185794446766650	0.2570628437787406	0.6472949655914420	F	F	F
0.9178127708025841	0.3857263244493723	0.8323557694827386	F	F	F
0.1286513809080105	0.5639658750925491	0.0273525055884889	F	F	F
0.2039210895986372	0.6725962013317854	0.1894494606858004	F	F	F
0.6473033850647241	0.8185868813316333	0.2570333374332989	F	F	F
0.8323637830854409	0.9178254946963307	0.3857021687176214	F	F	F
0.0273663987250643	0.1286649130452560	0.5639432586845459	F	F	F
0.1894651711464306	0.2039413794979339	0.6725808936144659	F	F	F
0.0520446007725882	0.7608749733144781	0.7065071967601010	T	T	T
0.1913016680352797	0.6606463707992916	0.6356097262428904	F	F	F
0.6356185077589771	0.1913086837801572	0.6606297179384271	F	F	F
0.6606315764770940	0.6356253462911141	0.1912846597929345	F	F	F
0.8087261900166567	0.3393913209191410	0.3643861179391834	F	F	F
0.3644005256268485	0.8087432442175313	0.3393701427839204	F	F	F
0.3393817235467154	0.3644103209379352	0.8087207573751911	F	F	F
0.6835517681103340	0.3307110014254917	0.0678262623019492	F	F	F
0.90462494046505282	0.5234055249551801	0.3713583793226078	F	F	F
0.3655898280863639	0.9342288006152657	0.7393865861915785	F	F	F
0.0678421513085112	0.6835629808024848	0.3306951644312974	F	F	F
0.3713781553311364	0.9046372789143220	0.5233930551433588	F	F	F
0.7393966448214400	0.3655955268339852	0.9342129192867787	F	F	F
0.3307007325971156	0.0678409918196294	0.6835450462613082	F	F	F
0.5234036332396244	0.3713867293556277	0.9046234911400575	F	F	F
0.9342164803618473	0.7394022387707366	0.3655789551170514	F	F	F
0.3164741897258239	0.6693291887780575	0.9321694504412363	F	F	F
0.0953983838334693	0.4766297043285235	0.6286361880835827	F	F	F
0.6344342617643548	0.0658090658455279	0.2606088124933876	F	F	F
0.9321758580590256	0.3164873805777759	0.6693052729354960	F	F	F
0.6286369597071797	0.0954084644052102	0.4766018068896258	F	F	F
0.2606173087737531	0.6344505895493384	0.0657822039828204	F	F	F
0.6693169684394178	0.9321949773922142	0.3164623179811272	F	F	F
0.4766119594619624	0.6286520453646673	0.0953821885296833	F	F	F

0.0657977475218274	0.2606337047427232	0.6344257580178692	F	F	F
0.1235336912278654	0.8735177814466958	0.7372562217839020	T	T	T
0.9815369721584614	0.6507597754224782	0.6782742337878298	T	T	T
0.5155174752552796	0.1091377339312558	0.7800341732974516	F	F	F
0.7800434927646123	0.5155266282709832	0.1091257554190648	F	F	F
0.1091327420163424	0.7800543308086816	0.5155103126675868	F	F	F
0.4845039322343894	0.8909076460700263	0.2199650095904175	F	F	F
0.2199758544933061	0.4845175898298137	0.8908785684000975	F	F	F
0.8908900213520283	0.2199874595275233	0.4844925251477576	F	F	F

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