

Supporting Information

One Step Route to the Fabrication of Arrays of TiO₂ Nanobowls via Complementary Block Copolymer Templating and Sol-Gel Process

Xue Li,¹ Juan Peng,² Joo-Hee Kang,⁴ Jin-Ho Choy,⁴ Martin Steinhart,³ Wolfgang Knoll² and Dong Ha Kim^{4*}

¹*School of Chemistry and Chemical Engineering, University of Jinan, 106 Jiwei Road, Jinan 250022, People's Republic of China*

²*Max Planck Institute for Polymer Research, Ackermannweg 10, 55128 Mainz, Germany*

³*Max Planck Institute for Microstructure Physics, Weinberg 2, D-06120 Halle, Germany*

⁴*Division of Nano Sciences and Department of Chemistry, 11-1 Daehyun-Dong, Seodaemun-Gu, Seoul 120-750, Korea*

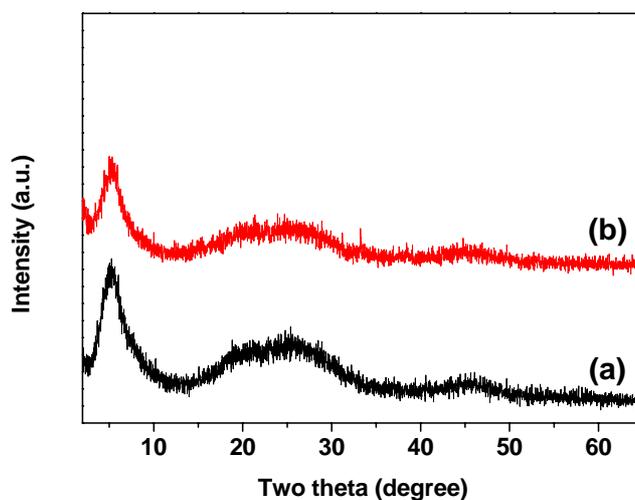


Figure S1. XRD patterns for the initial hybrid SEO/TiO₂ (a) and SEO/HAuCl₄(0.1)/TiO₂ (b) films.

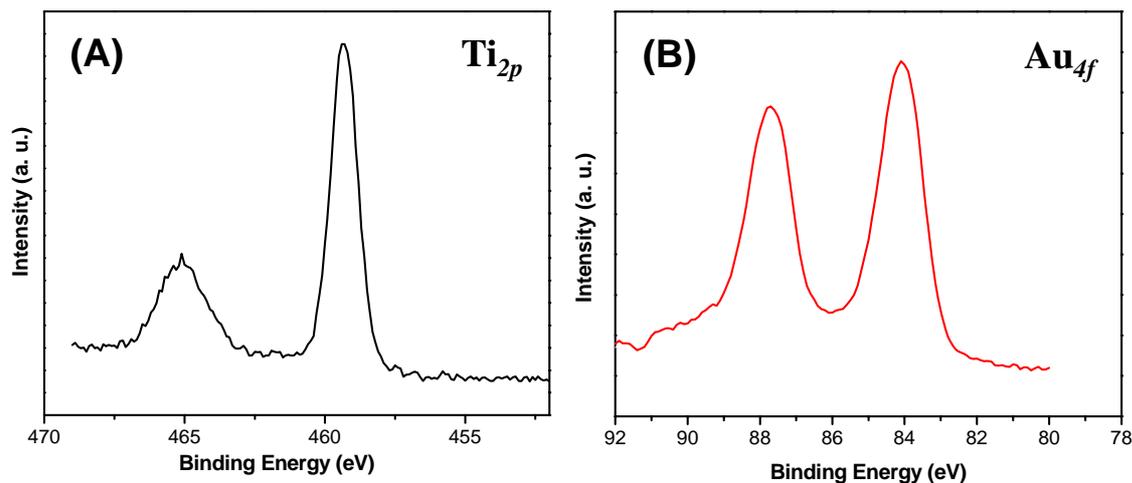


Figure S2. (a) High-resolution XPS Ti_{2p} spectrum of the surface of a porous hybrid PS-*b*-PEO/SG film. (b) High-resolution XPS Au_{4f} spectrum of the surface of Au/TiO₂ nanobowls obtained from porous hybrid PS-*b*-PEO/HAuCl₄/SG films after removal of the block copolymers.

The high resolution Ti_{2p} spectrum (Figure S2(A)) exhibits characteristic peaks of $Ti_{2p_{3/2}}$ and $Ti_{2p_{1/2}}$ in TiO₂ at 459.4 eV and 465.2, respectively.^{33,34} From the composite Au/titania nanobowls by introducing HAuCl₄ precursor to the PS-*b*-PEO/SG system followed by the removal the block copolymer with deep UV irradiation, characteristic peaks of Au⁰ at 87.6 eV ($Au_{4f_{5/2}}$) and 84.1 eV ($Au_{4f_{7/2}}$) on the surfaces of the nanobowls are observed (Figure B).³⁴ Such hybrid nanostructures can help prevent the coalesce of Au nanoparticles on TiO₂ surface.³⁵

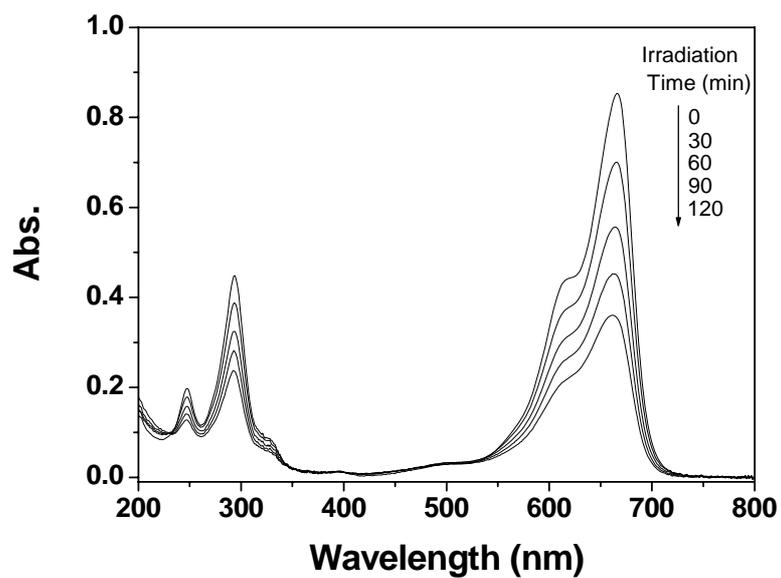


Figure S3. Photocatalytic activity of an PS-*b*-PEO/Au/TiO₂ film in terms of the decomposition of methylene blue as a function of UV irradiation time.