

## Polycatenar bows with single carbon atom elbow

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## SUPPLEMENTARY INFORMATION

Synthesis, spectroscopic and thermal characterization of compounds **6-8**

## Experimental

### General

All the solvents were *puriss p.a.* quality. All the other reagents were used as purchase from Aldrich. <sup>1</sup>H- (300 MHz) and <sup>13</sup>C- (75.5 MHz) NMR spectra were recorded on Bruker AV 300 instrument in CDCl<sub>3</sub> with SiMe<sub>4</sub> as internal standard unless stated otherwise. Infrared spectra were recorded on Bomem MB 102 spectrophotometer; absorption maxima are given in cm<sup>-1</sup>. Phase transition temperatures and textures were determined using an Olympus BX51 polarizing microscope equipped with a Linkam TH600 hot stage and PR600 temperature controller. Enthalpies of transition were measured by differential scanning calorimetry with a TA Instruments DSC-Q1000 instrument operated at a scanning rates of 2-to-10°C min<sup>-1</sup> on heating and on cooling. The apparatus was calibrated with indium (156.6°C; 28.4J.g<sup>-1</sup>) and gallium (29.8°C) as the standards. The XRD patterns were obtained with the following experimental set-up. A linear monochromatic Cu-Kα<sub>1</sub> beam ( $\lambda = 1.5405 \text{ \AA}$ ) was obtained using a sealed-tube generator (900 W) equipped with a bent quartz monochromator and a Debye-Scherrer-like geometry was used. The crude powder was filled in Lindemann capillaries of 1 mm diameter and 10 μm wall-thickness. The diffraction patterns were recorded with a curved Inel CPS 120 counter gas-filled detector linked to a data acquisition computer; periodicities up to 60 Å can be measured, and the sample temperature controlled to within ±0.05 °C from 20 to 200 °C.

### Synthesis

Alkoxybenzoic acids **1a** and **1c** were prepared by *O*-alkylation of methyl 4-hydroxybenzoate followed by hydrolysis under basic conditions while **1b** by *O*-alkylation of 4-

hydroxybenzaldehyde followed by oxidation using Jones reagent. After crystallization from 96% ethanol acids **1a-c** were obtained in 93%, 75% and 92% overall yield respectively.

**General procedure for the preparation of 4-(Alkoxybenzoyloxy)benzaldehydes **2a-c****

Aldehydes **2a-c** were prepared by direct esterification of **1a-c** with 4-hydroxybenzaldehyde employing standard dicyclohexylcarbodiimide and 4-(*N,N*-dimethylamino)pyridine method in dichloromethane.

**4-(4-Dodecyloxybenzoyloxy)benzaldehyde **2a**.** Yield, 89%; PT (°C): Cr 66 SmA 73 N 75 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88 (t, *J* = 6.5 Hz, 3H), 1.18-1.48 (m, 18 H), 1.78-1.87 (m, 2H), 4.05 (t, *J* = 6.6 Hz, 2H), 6.98 (d, *J* = 8.7 Hz, 2H Ar), 7.40 (d, *J* = 8.7 Hz, 2H Ar), 7.96 (d, *J* = 8.4 Hz, 2H Ar), 8.14(d, *J* = 8.7 Hz, 2H Ar), 10.02 (s, 1H); <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 14.0, 22.5, 25.9, 29.0, 29.2, 29.4, 29.46, 29.50, 29.52, 31.8, 68.3, 114.4, 120.8, 122.5, 131.1, 132.3, 133.8, 155.8, 163.8, 164.1, 190.8.

**4-(3,4-Didodecyloxybenzoyloxy)benzaldehyde **2b**.** Yield, 86%; mp 86-87 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88-0.92 (m, 6H), 1.29-1.56 (m, 36H), 1.82-1.94 (m, 4H), 4.07-4.13 (m, 4H), 6.96 (d, *J*=8.4 Hz, 1H Ar), 7.42 (d, *J*=8.4 Hz, 2H Ar), 7.67 (d, *J*=1.8 Hz, 1H Ar), 7.84 (dd, *J*<sub>1</sub>=1.8 Hz, *J*<sub>2</sub>=8.4 Hz, 1H Ar), 7.99 (d, *J*=8.4 Hz, 2H Ar), 10.04 (s, 1H); <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 14.11, 22.70, 25.97, 26.01, 29.04, 29.17, 29.38, 29.41, 29.63, 29.67, 29.70, 31.94, 69.11, 69.46, 112.02, 114.69, 120.85, 122.69, 124.61, 131.25, 133.92, 148.79, 154.25, 156.00, 164.40, 191.01.

**4-(3,4,5-Tridodecyloxybenzoyloxy)benzaldehyde **2c**.** Yield, 80%; mp 48-48 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88-0.92 (m, 9H), 1.29-1.56 (m, 54H), 1.74-1.90 (m, 6H), 4.05-4.11 (m, 6H), 7.41 (d, *J*=8.7 Hz, 2H Ar), 7.41 (s, 2H Ar), 7.99 (d, *J*=8.7 Hz, 2H Ar), 10.05 (s, 1H); <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 14.11, 22.69, 22.70, 26.06, 26.09, 29.30, 29.37, 29.39, 29.58, 29.64, 29.66, 29.70, 29.73, 29.76, 30.36, 31.93, 31.98, 69.32, 73.68, 108.70, 122.64, 123.22, 131.31, 134.06, 143.45, 153.05, 155.89, 164.41, 190.94

**General procedure for the preparation of 4-(Alkoxybenzoyloxy)benzoic acids **3a-c****

An oxidation was carried out by titrating a vigorously stirred solution of the appropriate aldehyde **2a-c** in acetone at r.t. with the Jones reagent until the colour of the mixture remained orange. After stirring overnight at r.t. the solvent was removed under reduced pressure and the residue was poured into H<sub>2</sub>O (100 ml) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 20 ml). The combined organic extracts were washed with H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under vacuum. The crude products were purified by crystallisation.

**4-(4-Dodecyloxybenzoyloxy)benzoic acid 3a.** Crystallised from the mixture ethyl acetate/*n*-hexane (2:1). Yield, 92%; PT (°C): Cr 121 SmC 209 N 216 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 0.92 (t, *J*=6.8 Hz, 3H), 1.27-1.52 (m, 18H), 1.77-1.87 (m, 2H), 4.05 (t, *J*=6.5 Hz, 2H), 6.98 (d, *J*=8.8 Hz, 2H Ar), 7.34 (d, *J*=8.4 Hz, 2H Ar), 8.15 (d, *J*=8.8 Hz, 2H Ar), 8.2(d, *J*=8.4 Hz, 2H, Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ 14.10, 22.70, 25.99, 29.07, 29.40, 29.57, 29.60, 29.65, 29.71, 31.91, 68.40, 114.42, 121.12, 122.08, 126.63, 131.89, 132.41, 155.63, 163.82, 164.30, 171.11.

**4-(3,4-Didodecyloxybenzoyloxy)benzoic acid 3b.** Crystallised from the mixture acetone/CH<sub>3</sub>CN (2:1). Yield, 91%; mp 165-166 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 0.87-0.92 (m, 6H), 1.22-1.54 (m, 36H), 1.83-1.93 (m, 4H), 4.07-4.13 (m, 4H), 6.96 (d, *J*=8.4 Hz, 1H Ar), 7.35(d, *J*=8.4 Hz, 2H Ar), 7.68 (s, 1 H Ar), 7.84 (d, *J*=8.4 Hz, 1H Ar), 8.22(d, *J*=8.4 Hz, 2H Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ 14.11, 22.70, 25.98, 26.02, 29.04, 29.17, 29.37, 29.41, 29.62, 29.63, 29.67, 29.70, 31.93, 69.11, 69.43, 111.97, 114.65, 120.98, 122.06, 124.58, 126.67, 131.93, 148.77, 154.14, 155.64, 164.49, 171.20

**4-(3,4,5-Tridodecyloxybenzoyloxy)benzoic acid 3c.** Crystallised from the wet acetone (80%). Yield, 89%; mp 88-89 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 0.88-0.92 (m, 9H), 1.29-1.56 (m, 54H), 1.74-1.90 (m, 6H), 4.05-4.12 (m, 6H), 7.35 (d, *J*=8.5 Hz, 2H Ar), 7.43 (s, 2H Ar), 8.22 (d, *J*=8.7 Hz, 2H Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ 14.02, 22.6, 22.61, 25.97, 25.99, 29.20, 29.28, 29.30, 29.48, 29.54, 29.57, 29.61, 29.64, 29.65, 29.67, 30.26, 31.83, 31.86, 69.22, 73.53, 108.57, 121.94, 123.21, 126.70, 131.82, 143.23, 152.93, 155.39, 164.36, 170.97.

### General procedure for the preparation of 6a-c and 7a-c

To the mixture of appropriate acid **3a-c** (2 mmol), DCC (2 mmol) and DMAP (1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 4,4'-dihydroxybenzophenone (1 mmol) or bis(4-hydroxyphenyl)methane (1 mmol) dissolved in a minimum amount of THF was added. After stirring at r.t. under Ar for 5 h. the reaction mixture was filtered and the solvent was removed under reduced pressure. The crude product was purified by column chromatography on silka gel using the mixture of sovents CH<sub>2</sub>Cl<sub>2</sub>/MeOH (100:1) for **6a-c** and CH<sub>2</sub>Cl<sub>2</sub>/MeOH (200:1) for **7a-c**.

**4,4'Bis[4-(4-dodecyloxybenzoyloxy)benzoyloxy]benzophenone 6a.** Yield, 74%; PT (°C): Cr1 11.7 Cr2 109 Cr3 171 SmC 195 N 196 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 0.90 (t, *J*=6.8 Hz, 6H), 1.22-1.55 (m, 36H), 1.80-1.89 (m, 4H), 4.06 (t, *J*=6.5 Hz, 4H), 7.01 (d, *J*=8.8 Hz, 4H Ar), 7.40 (d, *J*=8.6 Hz, 4H), 7.42 (d, *J*=8.6 Hz, 4H), 7.96 (d, *J*=8.8 Hz, 4H Ar), 8.18 (d, *J*=8.8 Hz, 4H Ar), 8.32 (d, *J*=8.8 Hz, 4H, Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ 14.12, 22.7, 26, 29.08, 29.35, 29.55, 29.63, 29.65, 31.92, 68.41, 114.44, 120.88, 121.8, 122.24, 122.32, 126.4, 131.7, 131.93, 132.44, 132.53, 135.10, 154.22, 155.66, 163.9, 164.00, 164.30, 194.40; IR (KBr):

3110, 2920, 2850, 1733, 1646, 1604, 1287, 1160, 1079, 762. Elemental analysis. Found: C, 75.89; H, 7.22. Calc. for C<sub>65</sub>H<sub>74</sub>O<sub>11</sub>: C, 75.70; H 7.23%.

**4,4'Bis[4-(3,4-didodecyloxybenzoyloxy)benzoyloxy]benzophenone 6b.** Yield, 68%; PT (°C): Cr1 79.5 Cr2 97 Cr3 113 (Col<sub>h</sub>P<sub>A</sub> 115) Col<sub>h</sub> 121 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88-0.92 (m, 12H), 1.22-1.54 (m, 72H), 1.83-1.93 (m, 8H), 4.08-4.14 (m, 8H), 6.97 (d, J=8.5 Hz, 2H Ar), 7.41 (d, J=8.5 Hz, 8H Ar), 7.69 (d, J=1.9 Hz, 2H Ar), 7.85 (dd, J<sub>1</sub>=8.5 Hz, J<sub>2</sub>=1.9 Hz, 2H Ar), 7.97 (d, J=8.6 Hz, 4H Ar), 8.32 (d, J=8.6 Hz, 4H, Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 14.03, 22.62, 25.89, 25.94, 28.96, 29.09, 29.29, 29.30, 29.33, 29.54, 29.55, 29.58, 29.62, 31.85, 69.02, 69.32, 111.88, 114.56, 120.80, 121.72, 122.19, 124.52, 126.33, 131.62, 131.85, 135.03, 148.68, 154.11, 154.14, 155.61, 163.93, 164.35, 194.32; IR (KBr): 3077, 2920, 2850, 1733, 1665, 1599, 1269, 1160, 1063, 754. Elemental analysis. Found: C, 76.20; H, 8.77. Calc. for C<sub>89</sub>H<sub>122</sub>O<sub>13</sub>: C, 76.36; H, 8.78%.

**4,4'Bis[4-(3,4,5-tridodecyloxybenzoyloxy)benzoyloxy]benzophenone 6c.** Yield, 63%; PT (°C): G 45 Cr1 69 Cr2 75 Col<sub>h</sub> 78 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88-0.92 (m, 18H) 1.22-1.57 (m, 108H), 1.75-1.91 (m, 12H), 4.06-4.12 (m, 12H), 7.41 (d, J=8.7 Hz, 4H Ar), 7.42 (d, J=8.7 Hz, 4H Ar), 7.45 (s, 4H Ar), 7.97 (d, J=8.7 Hz, 4H Ar), 8.33 (d, J=8.6 Hz, 4H Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 14.12, 22.69, 22.71, 26.07, 26.10, 29.30, 29.37, 29.41, 29.57, 29.64, 29.67, 29.7, 29.75, 29.76, 30.36, 31.93, 31.95, 69.33, 73.64, 108.70, 121.79, 122.28, 123.22, 126.57, 131.72, 131.96, 135.13, 143.41, 153.06, 154.21, 155.59, 163.99, 164.45, 194.44; IR (KBr): 3090, 2920, 2850, 1733, 1655, 1600, 1270, 1160, 1062, 756. Elemental analysis. Found: C, 76.86; H, 9.65. Calc. for C<sub>113</sub>H<sub>170</sub>O<sub>15</sub>: C, 76.74; H, 9.69%.

**Bis[4-(4-dodecyloxybenzoyloxy)benzoyloxyphenyl]methane 7a.** Yield, 60%; PT (°C): G 104 Cr1 124 Cr2 131 Col<sub>r</sub> 168.5 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.91 (t, J=6.5 Hz, 6H), 1.22-1.54 (m, 36H), 1.81-1.90 (m, 4H), 4.05-4.10 (m, 6H), 7.01 (d, J=9.0 Hz, 4H Ar), 7.19 (d, J=8.3 Hz, 4H), 7.29 (d, J=8.3 Hz, 4H), 7.39 (d, J=8.6 Hz, 4H Ar), 8.18 (d, J=9.0 Hz, 4H Ar), 8.30 (d, J=8.6 Hz, 4H, Ar); <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 14.14, 22.70, 25.99, 29.09, 29.37, 29.57, 29.60, 29.65, 29.67, 31.93, 40.37, 68.39, 114.42, 120.98, 121.70, 122.08, 126.97, 129.99, 131.80, 132.43, 138.47, 149.31, 155.35, 163.83, 164.33, 164.56; IR (KBr): 3081, 2922, 2852, 1732, 1605, 1272, 1161, 1066, 761; Elemental analysis. Found: C, 76.92; H, 7.48. Calc. for C<sub>65</sub>H<sub>76</sub>O<sub>10</sub>: C, 76.74; H, 7.53%.

**Bis[4-(3,4-didodecyloxybenzoyloxy)benzoyloxyphenyl]methane 7b.** Yield, 64%; PT (°C): Cr1 71 Cr2 122 I; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88-0.92 (m, 12H), 1.29-1.58 (m, 72H), 1.83-1.94 (m, 8H), 4.07-4.14 (m, 10H), 6.96 (d, J=8.4 Hz, 2H Ar), 7.19 (d, J=8.0 Hz, 4H Ar), 7.29 (d, J=8.0 Hz, 4H Ar), 7.38 (d, J=8.4 Hz, 4H Ar), 7.69 (s, 2H Ar), 7.86 (d, J=8.4 Hz, 2H Ar), 8.30

(d,  $J=8.4$  Hz, 4H Ar);  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 14.13, 22.70, 25.98, 26.02, 29.04, 29.17, 29.38, 29.39, 29.42, 29.63, 29.64, 29.67, 29.71, 31.94, 40.71, 69.11, 69.40, 111.96, 114.64, 120.98, 121.70, 122.11, 124.57, 126.98, 129.99, 131.80, 138.47, 148.76, 149.30, 154.14, 155.14, 164.47, 164.57; IR (KBr): 3077, 2920, 2848, 1735, 1600, 1269, 1161, 756. Elemental analysis. Found: C, 76.89; H, 9.08. Calc. for  $\text{C}_{89}\text{H}_{124}\text{O}_{12}$ : C, 77.13; H, 9.02%.

**Bis[4-(3,4,5-tridodecyloxybenzoyloxy)benzoyloxyphenyl]methane 7c.** Yield, 53%; PT (°C): G 17.5 Cr1 45 Cr2 50 Cr3 64 Col<sub>h</sub> 68 I;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 0.88-0.92 (m, 18H) 1.29-1.56 (m, 108H), 1.74-1.91 (m, 12H), 4.05-4.12 (m, 14H), 7.19 (d,  $J=8.5$  Hz, 4H Ar), 7.29 (d,  $J=8.5$  Hz, 4H Ar), 7.37 (d,  $J=8.7$  Hz, 4H Ar), 7.45 (s, 4H Ar), 8.31 (d,  $J=8.7$  Hz, 4H Ar);  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 14.12, 22.69, 22.71, 26.07, 26.09, 29.31, 29.37, 29.40, 29.57, 29.64, 29.67, 29.71, 29.74, 29.75, 29.76, 30.36, 31.93, 40.72, 69.32, 73.63, 108.68, 121.69, 122.10, 123.32, 127.13, 129.99, 131.83, 138.49, 143.35, 149.29, 153.04, 155.28, 164.49, 164.53; IR (KBr): 3070, 2919, 2848, 1732, 1595, 1274, 1162, 750. Elemental analysis. Found: C, 77.59; H, 9.95. Calc. for  $\text{C}_{113}\text{H}_{172}\text{O}_{14}$ : C, 77.35; H, 9.88%.

### General procedure for the preparation of diphenylmethanols 8a-c

Solution of  $\text{NaBH}_4$  (0.1 mmol) in 85% *i*-PrOH was added dropwise to the solution of appropriate ketone **6a-c** in THF (20 mL) at r.t. After stirring at r.t. overnight, a few drops of formic acid were added. The reaction mixture was filtered and the solvent was removed under reduced pressure. The crude product was purified by column chromatography on silika gel using the mixture of solvents  $\text{CH}_2\text{Cl}_2/\text{DIPE}$  (25:1) as eluent.

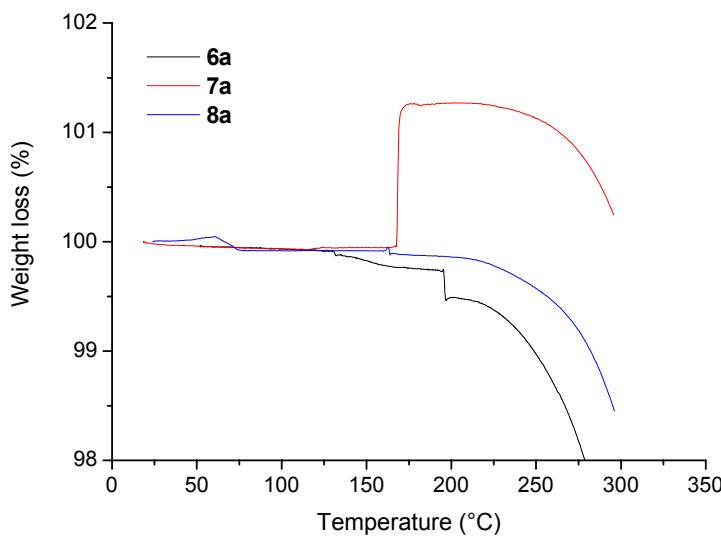
**Bis[4-(4-dodecyloxybenzoyloxy)benzoyloxyphenyl]methanol 8a.** Yield, 65%; PT (°C): Cr1 129 Col<sub>r</sub> 164 I;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 0.90 (t,  $J=6.6$  Hz, 6H), 1.29-1.55 (m, 36H), 1.81-1.90 (m, 4H), 2.37 (d,  $J=3.6$  Hz, 1H), 4.07 (t,  $J=6.6$  Hz, 4H), 5.95 (d,  $J=3.6$  Hz, 1H), 7.00 (d,  $J=9.0$  Hz, 4H Ar), 7.24 (d,  $J=8.6$  Hz, 4H), 7.39 (d,  $J=9.0$  Hz, 4H), 7.49 (d,  $J=8.6$  Hz, 4H Ar), 8.17 (d,  $J=9.0$  Hz, 4H Ar), 8.29 (d,  $J=8.6$  Hz, 4H, Ar);  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 14.12, 22.70, 25.99, 29.10, 29.36, 29.56, 29.60, 29.64, 29.66, 31.93, 68.41, 75.31, 114.43, 120.97, 121.81, 122.11, 126.85, 127.84, 131.83, 132.43, 141.23, 150.36, 155.43, 163.85, 164.34, 164.46; IR (KBr): 3518, 3077, 2919, 2852, 1732, 1605, 1275, 1198, 1165, 1074, 761. Elemental analysis. Found: C, 75.24; H, 7.50. Calc. for  $\text{C}_{65}\text{H}_{76}\text{O}_{11}$ : C, 75.55; H, 7.41%.

**Bis[4-(3,4-didodecyloxybenzoyloxy)benzoyloxyphenyl]methanol 8b.** Yield, 66%; PT (°C): Cr1 70 Cr2 139 I;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 0.85-0.90 (m, 12H), 1.16-1.54 (m, 72H), 1.81-1.91 (m, 8H), 2.34 (d,  $J=3.6$  Hz, 1H), 4.05-4.11 (m, 8H), 5.92 (d,  $J=3.6$  Hz, 1H), 6.94 (d,  $J=8.7$  Hz, 2H Ar), 7.22 (d,  $J=8.7$  Hz, 4H Ar), 7.36 (d,  $J=8.7$  Hz, 4H Ar), 7.47 (d,  $J=8.7$  Hz, 4H Ar),

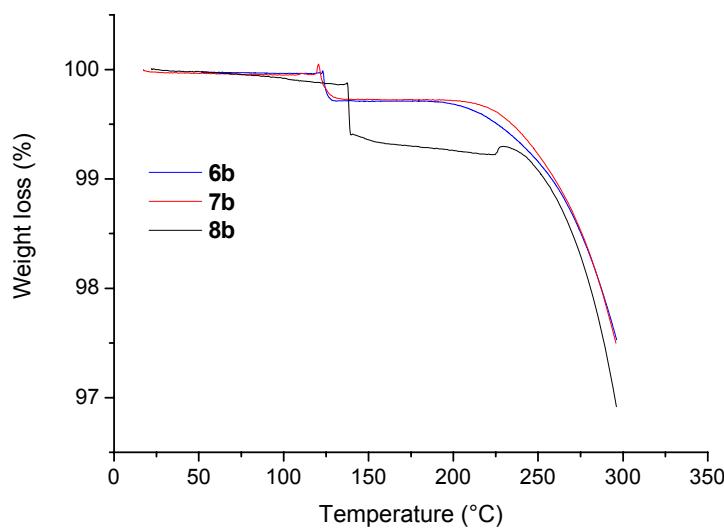
7.66 (s, 2H Ar), 7.83 (d,  $J=8.7$  Hz, 2H Ar), 8.27 (d,  $J=8.7$  Hz, 4H Ar);  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 14.12, 22.70, 25.98, 26.02, 29.04, 29.17, 29.38, 29.39, 29.42, 29.57, 29.62, 29.63, 29.67, 29.71, 31.94, 69.11, 69.40, 75.29, 111.96, 114.64, 120.96, 121.80, 122.14, 124.58, 126.85, 127.84, 131.83, 141.24, 148.76, 150.33, 154.15, 155.45, 164.45, 164.48; IR (KBr): 3510, 3082, 2920, 2852, 1735, 1600, 1271, 1200, 1161, 752. Elemental analysis. Found: C, 76.11; H, 9.09. Calc. for  $\text{C}_{89}\text{H}_{124}\text{O}_{13}$ : C, 76.25; H, 8.92%.

**Bis[4-(3,4,5-tridodecyloxybenzoyloxy)benzoyloxyphenyl]methanol 8c.** Yield, 70%; PT (°C): G 41 Cr1 62 Cr2 83 (Col<sub>h</sub> 73) I;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 0.88-0.92 (m, 18H) 1.29-1.56 (m, 108H), 1.75-1.91 (m, 12H), 2.37 (d,  $J=3.6$  Hz, 1H), 4.05-4.12 (m, 12H), 5.95 (d,  $J=3.6$  Hz, 1H), 7.25 (d,  $J=8.9$  Hz, 4H Ar), 7.37 (d,  $J=8.4$  Hz, 4H Ar), 7.44 (s, 4H Ar), 7.50 (d,  $J=8.4$  Hz, 4H Ar), 8.30 (d,  $J=8.9$  Hz, 4H Ar);  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 14.12, 22.69, 22.71, 26.07, 26.09, 29.30, 29.37, 29.40, 29.58, 29.64, 29.67, 29.70, 29.75, 29.76, 30.36, 31.93, 31.95, 69.32, 73.63, 75.27, 108.68, 121.78, 122.14, 123.29, 126.99, 127.84, 131.85, 141.26, 143.37, 150.31, 153.04, 155.34, 164.48; IR (KBr): 3521, 3070, 2922, 2852, 1737, 1595, 1270, 1191, 1161, 746. Elemental analysis. Found: C, 76.92; H, 9.80. Calc. for  $\text{C}_{113}\text{H}_{172}\text{O}_{15}$ : C, 76.65; H, 9.79%.

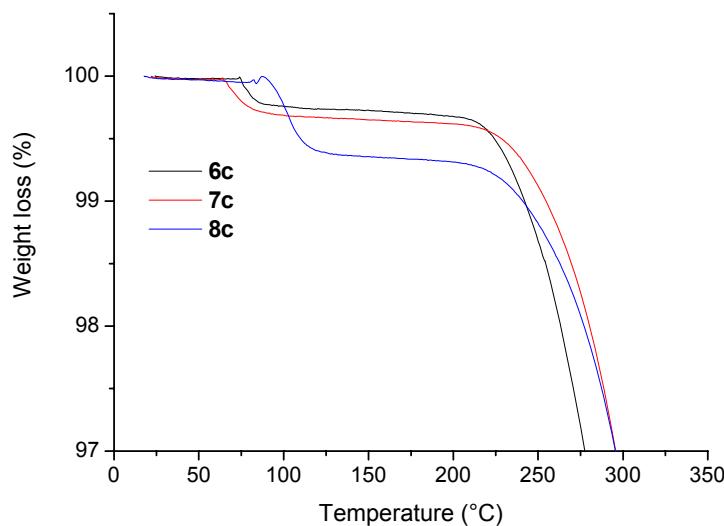
### Thermal behaviour of the compounds 6-8



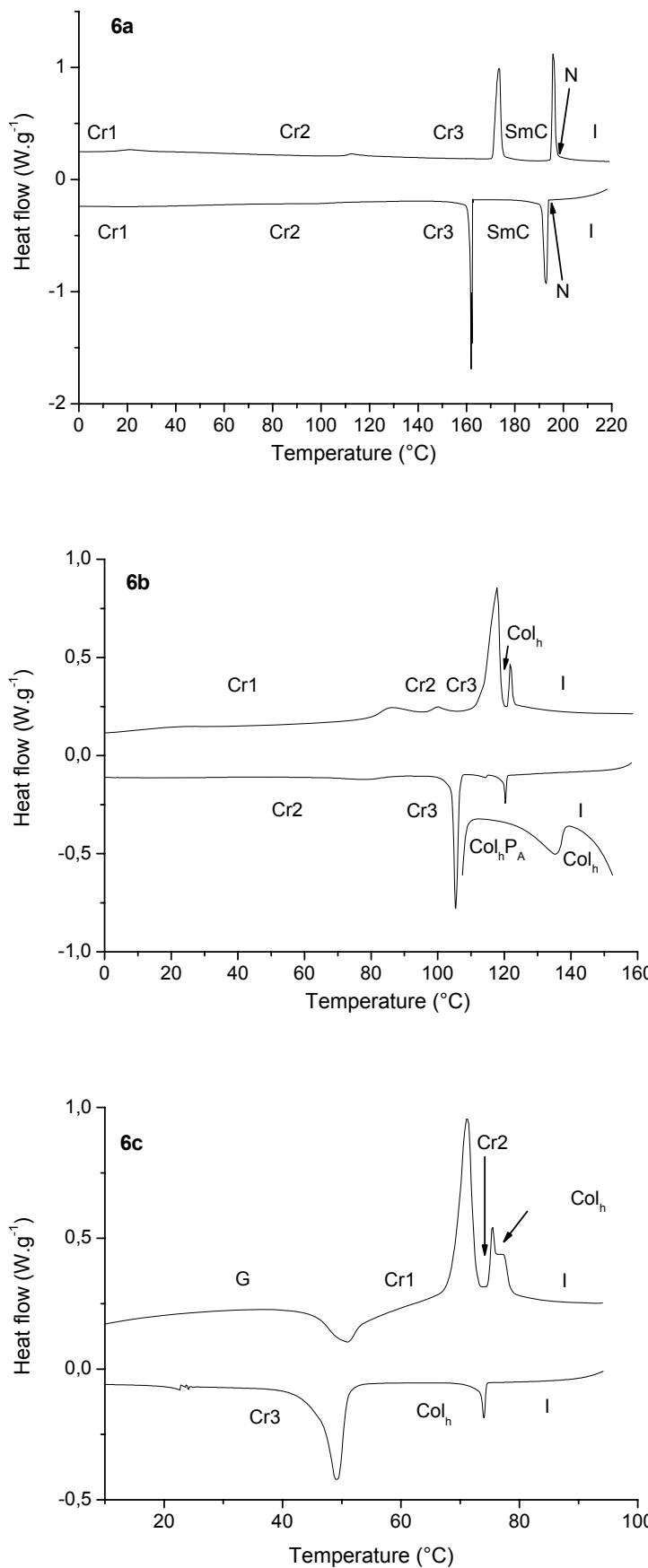
**Fig S1** TGA thermograms of the dicatenar bows



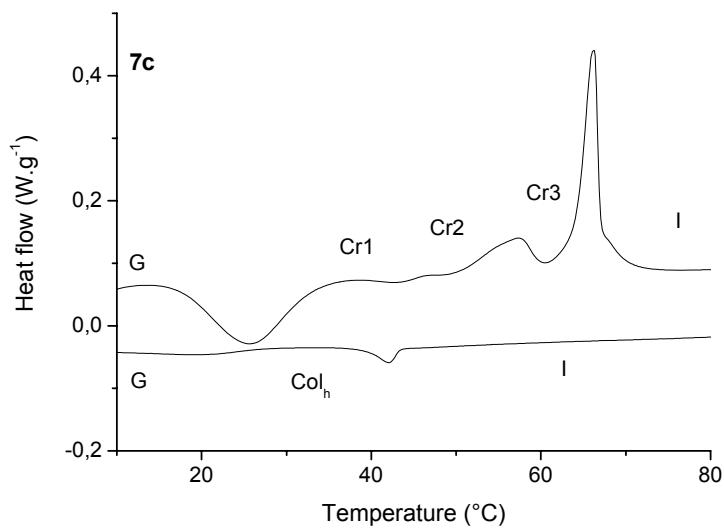
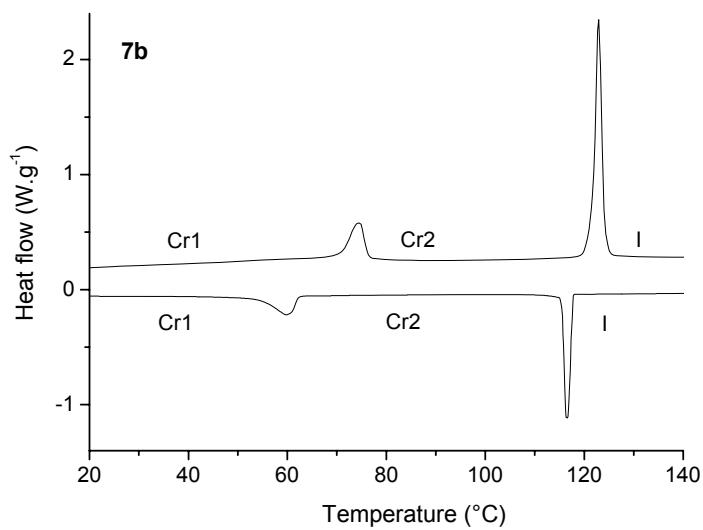
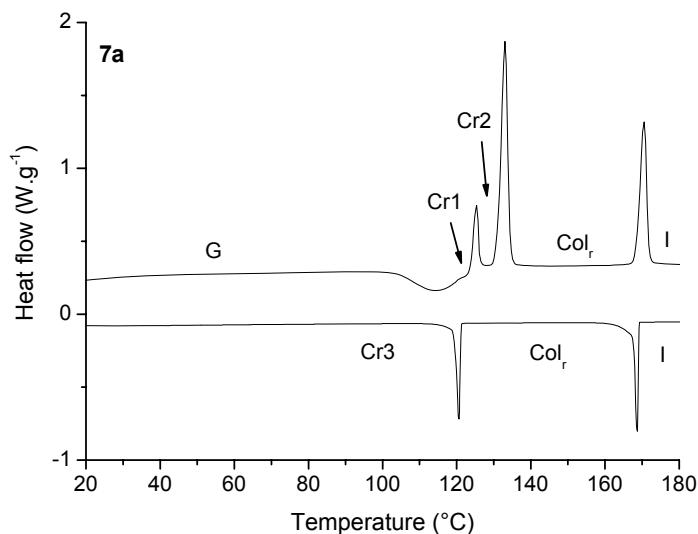
**Fig S2** TGA thermograms of the tetracatenar bows



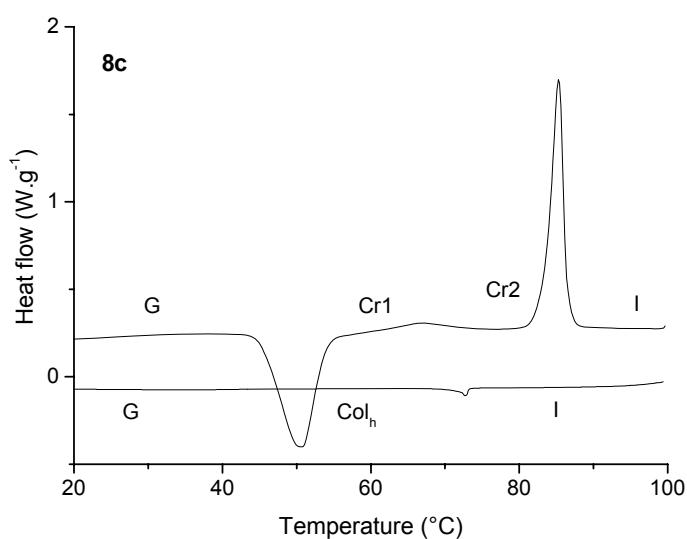
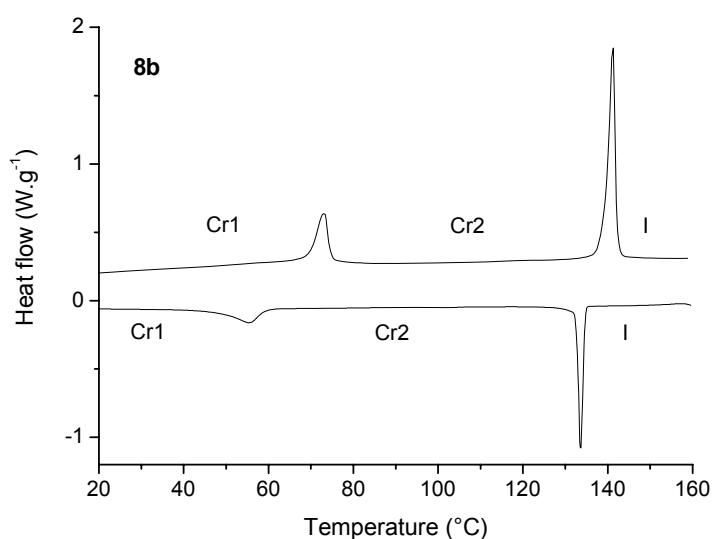
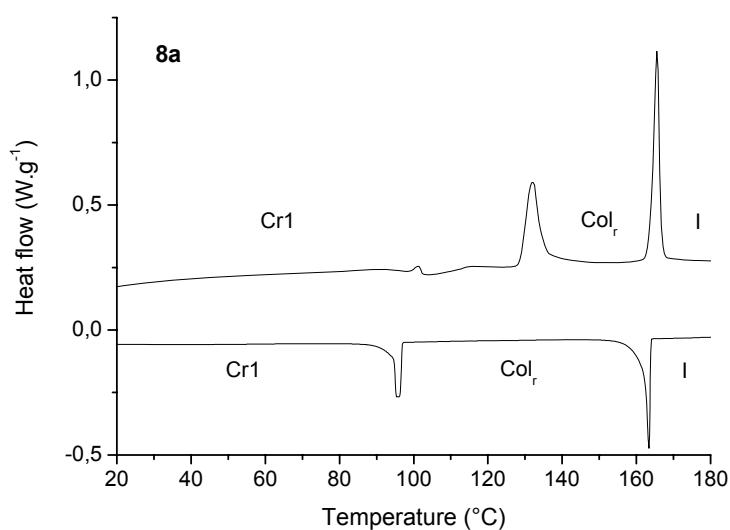
**Fig S3** TGA thermograms of the hexacatenar bows



**Fig S4** DSC thermograms of the compounds **6a-6c**



**Fig S5** DSC thermograms of the compounds **7a-7c**



**Fig S6** DSC thermograms of the compounds **8a-8c**