

Supplementary information for

A theory of constrained swelling of a pH-sensitive hydrogel

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Part A: Coupled nonlinear algebraic equations

The nonlinear algebra equations (3.13)-(3.15) can be solved to express the concentrations in the gel, c_{H^+} , c_+ and c_- , in terms of the concentrations in the external solution, \bar{c}_{H^+} , \bar{c}_+ and \bar{c}_- , and the swelling ratio $\det \mathbf{F}$. A combination of the three equations gives a cubic equation for νc_{H^+} , namely,

$$\left(1 + \frac{\nu \bar{c}_+}{\nu \bar{c}_{\text{H}^+}}\right) (\nu c_{\text{H}^+})^3 + N_A K_a \nu \left(1 + \frac{\nu \bar{c}_+}{\nu \bar{c}_{\text{H}^+}}\right) (\nu c_{\text{H}^+})^2 - \left(\frac{N_A K_a \nu f}{\det \mathbf{F}} + \nu \bar{c}_{\text{H}^+} \nu \bar{c}_-\right) (\nu c_{\text{H}^+}) - N_A K_a \nu \nu \bar{c}_{\text{H}^+} \nu \bar{c}_- = 0. \quad (\text{A.1})$$

The solution to this cubic equation is

$$\nu c_{\text{H}^+} = \sqrt[3]{-\frac{q}{2} + \sqrt{\left(\frac{q}{2}\right)^2 + \left(\frac{p}{3}\right)^3}} + \sqrt[3]{-\frac{q}{2} - \sqrt{\left(\frac{q}{2}\right)^2 + \left(\frac{p}{3}\right)^3}} - \frac{N_A K_a \nu}{3}, \quad (\text{A.2})$$

where

$$p = -\frac{\frac{N_A K_a \nu f}{\det \mathbf{F}} + \nu \bar{c}_{\text{H}^+} \nu \bar{c}_-}{1 + \frac{\nu \bar{c}_+}{\nu \bar{c}_{\text{H}^+}}} - \frac{(N_A K_a \nu)^2}{3}, \quad (\text{A.3})$$

$$q = -\frac{N_A K_a \nu \nu \bar{c}_{\text{H}^+} \nu \bar{c}_-}{1 + \frac{\nu \bar{c}_+}{\nu \bar{c}_{\text{H}^+}}} + \frac{(N_A K_a \nu) \left(\frac{N_A K_a \nu f}{\det \mathbf{F}} + \nu \bar{c}_{\text{H}^+} \nu \bar{c}_-\right)}{3 \left(1 + \frac{\nu \bar{c}_+}{\nu \bar{c}_{\text{H}^+}}\right)} + \frac{2(N_A K_a \nu)^3}{27}. \quad (\text{A.4})$$

Once c_{H^+} is solved, c_+ and c_- are solved from (3.13) and (3.14).

Recall that $\bar{c}_- = \bar{c}_{\text{H}^+} + \bar{c}_+$ due to electroneutrality in the external solution, and that $C_\alpha = c_\alpha \det \mathbf{F}$ in the gel. Consequently, \hat{W} defined in (5.1) can be expressed as a function of the following independent variables:

$$\hat{W} = \hat{W}(\mathbf{F}, \bar{c}_{\text{H}^+}, \bar{c}_+). \quad (\text{A.5})$$

In writing the user-supplied subroutine for ABAQUS, we also need partial derivatives of the function $\hat{W}(\mathbf{F}, \bar{c}_H, \bar{c}_+)$. These lengthy expressions can be found in the subroutine and are not given here.

Part B: A gel of spherical symmetry

Boundary-value problems of spherical symmetry have been solved for neutral gels.^{1,2} We now list the equations for a pH-sensitive gel. We name a small element of the network after the radius of the element, R , when the gel is in a state of reference. The same element of the network moves to a place of radius r when the gel is in the current state. The state of deformation of the gel is fully specified by the function $r(R)$. The stretch in each of the circumferential directions is

$$\lambda_\theta = r / R. \quad (\text{B.1})$$

The stretch in the radial direction is

$$\lambda_r = dr / dR. \quad (\text{B.2})$$

Let $s_\theta(R)$ be the nominal stress in each of the circumferential directions, and $s_r(R)$ be the nominal stress in the radial direction. Mechanical equilibrium requires that

$$\frac{ds_r}{dR} + 2 \frac{s_r - s_\theta}{R} = 0. \quad (\text{B.3})$$

Recall that the nominal stresses relate to the true stresses by $s_\theta = \sigma_\theta \lambda_r \lambda_\theta$ and $s_r = \sigma_r \lambda_\theta^2$. The stress-stretch relation (3.10) becomes that

$$s_\theta = NkT(\lambda_\theta - \lambda_\theta^{-1}) - \lambda_r \lambda_\theta (\Pi_{sol} + \Pi_{ion}), \quad (\text{B.4})$$

$$s_r = NkT(\lambda_r - \lambda_r^{-1}) - \lambda_\theta^2 (\Pi_{sol} + \Pi_{ion}). \quad (\text{B.5})$$

A combination of the above equations, together with the thermodynamic relations (3.11)-(3.15), leads to coupled first-order ordinary differential equations that govern the function $r(R)$ and $s_r(R)$.

Part C: The code of user subroutine for pH-sensitive hydrogels

```
SUBROUTINE UHYPER(BI1,BI2,AJ,U,UI1,UI2,UI3,TEMP,NOEL,
```

```
1 CMNAME,INCMPLAG,NUMSTATEV,STATEV,NUMFIELDV,FIELDV,
```

```
2 FIELDVINC,NUMPROPS,PROPS)
```

```
C=====
```

```
C      User defined hyperelastic material subroutine
```

```
C      for gel with Flory-Rehner free-energy function
```

```
C      to be used in Abaqus Standard
```

```
C-----
```

```
C      Material properties to be passed to the subroutine:
```

```
C      PROPS(1) - Nv
```

```
C      PROPS(2) - chi
```

```
C      PROPS(3) - lambda_0    initial swelling
```

```
C      PROPS(4) - vc0        ion concentration in external solution
```

```
C      PROPS(5) - pKa        dissociation constant
```

```
C      PROPS(6) - f          density of PH sensitive monomer
```

```
C      State variable:
```

```
C      TEMP      - PH0    PH in the solvent outside the gel
```

```
C      Set chemical potential of all kinds of ions at vc0 as 0.
```

C The initial value of PH0 and lambda_0 should match each other

C=====

INCLUDE 'ABA_PARAM.INC'

CHARACTER*80 CMNAME

DIMENSION U(2),UI1(3),UI2(6),UI3(6),STATEV(*),FIELDV(*),

1 FIELDVINC(*),PROPS(*)

Nv = PROPS(1)

chi = PROPS(2)

lambda0 = PROPS(3)

detF0 = lambda0**3

vCcou=PROPS(4)

vNa=0.0602

vCpro=vNa*10.0**(-TEMP)

vCco=vCcou+vCpro

Ka=10**(-PROPS(5))

f=PROPS(6)

U(1) = Nv/2 * (1/lambda0*BI1*AJ**(2.0/3.0) - 3/detF0- 2/detF0*(3*LOG(lambda0) +

LOG(AJ))-chi/detF0**2/AJ- (AJ-1/detF0)*LOG(AJ/(AJ-1/detF0))

U(2) = 0

UI1(1) = Nv/2/lambda0*AJ**(2.0/3.0)

UI1(2) = 0

$$UI1(3) = Nv/3/\lambda_0 * BI1 * AJ^{**(-1.0/3.0)} + (1-Nv)/AJ/\det F0 -$$

$$\text{LOG}(AJ/(AJ-1/\det F0)) + \chi/(\det F0 * AJ)^{**2}$$

$$UI2 = 0$$

$$UI2(3) = -Nv/9/\lambda_0 * BI1 * AJ^{**(-4.0/3.0)} - (1-Nv)/AJ^{**2}/\det F0 + 1/AJ/(AJ-1/\det F0)/\det F0 -$$

$$2 * \chi / \det F0^{**2} / AJ^{**3}$$

$$UI2(5) = Nv/3/\lambda_0 * AJ^{**(-1.0/3.0)}$$

$$UI3 = 0$$

$$UI3(4) = -Nv/9/\lambda_0 * AJ^{**(-4.0/3.0)}$$

$$UI3(6) = 4 * Nv / 27 / \lambda_0 * BI1 * AJ^{**(-7.0/3.0)} + 2 * (1-Nv) / AJ^{**3} / \det F0 -$$

$$(2 * AJ - 1 / \det F0) / (AJ * (AJ - 1 / \det F0))^{**2} / \det F0 + 6 * \chi / \det F0^{**2} / AJ^{**4}$$

$$aa = 1 + vC_{cou} / vC_{pro}$$

$$bb = Ka * (vNa)^{**aa}$$

$$cc = -(Ka * vNa * f / AJ / \det F0 + vC_{pro} * vC_{co})$$

$$dd = -Ka * vNa * vC_{pro} * vC_{co}$$

$$PP = cc / aa - (bb^{**2}) / 3 / (aa^{**2})$$

$$QQ = dd / aa - bb * cc / 3 / aa^{**2} + 2 * bb^{**3} / 27 / aa^{**3}$$

$$FORM = (QQ/2)^{**2} + (PP/3)^{**3}$$

IF(FORM.GE.0.0)THEN

$$vCH = (-QQ/2 + FORM^{**1.0/2.0})^{**1.0/3.0} + (-QQ/2 - FORM^{**1.0/2.0})^{**1.0/3.0} - bb/3/aa$$

ELSE

$$R = QQ^{**2}/4 - FORM$$

$$\theta = \text{ATAN2}(-2 * (-FORM)^{**0.5}, QQ)$$

IF(theta.LT.0.0)THEN

theta=3.1415926+theta

ENDIF

vCH=2*R**(1.0/6.0)*cos(theta/3)-bb/3/aa

ENDIF

DCDAJ=Ka*vNa*f*(AJ*detF0)**(-2)

DDCDDAJ=-2*Ka*vNa*f*(AJ*detF0)**(-3)

DDDCDDAJ=6*Ka*vNa*f*(AJ*detF0)**(-4)

DvCHDAJ=-(vCH)*DCDAJ/(3*aa*vCH**2+2*bb*vCH+cc)

DDvCHDDAJ=-(vCH*DDCDDAJ+2*DCDAJ*DvCHDAJ+2*bb*DvCHDAJ**2+6*aa*

vCH*DvCHDAJ**2)/(3*aa*vCH**2+2*bb*vCH+cc)

DDvCHDDAJ=-(6*aa*DvCHDAJ**3+18*aa*vCH*DvCHDAJ*DDvCHDDAJ+6

*bb*DvCHDAJ*DDvCHDDAJ+3*DCDAJ*DDvCHDDAJ+3*DDCDDAJ*DvCHDAJ

+vCH*DDDCDDAJ)/(3*aa*vCH**2+2*bb*vCH+cc)

vCA=aa*vCH-vCpro*vCco/vCH

DvCADAJ=(aa+vCpro*vCco/vCH**2)*DvCHDAJ

DDvCADDAJ=(aa+vCpro*vCco/vCH**2)*DDvCHDDAJ-2*DvCHDAJ**2*vCpro*vCco/vCH**3

DDvCADDAJ=DDvCHDDAJ*(aa+vCpro*vCco/vCH**2)-6*DDvCHDDAJ*

DvCHDAJ*vCpro*vCco/(vCH**3)+6*vCpro*vCco/(vCH**4)*(DvCHDAJ**3)

vCAH=f/detF0/AJ-vCA

DvCAHDAJ=-f/detF0**2/AJ**2-DvCADAJ

$$DDvCAHDDAJ=2^*f/detF0^{**3}/AJ^{**3}-DDvCADD AJ$$

$$D3vCAHD3AJ=-6^*f/detF0^{**4}/AJ^{**4}-DDDvCADD AJ$$

$$FORM11=log(vCH)+log(vCA)-log(vCAH)-log(vNa)-log(Ka)$$

$$FORM12=aa^*vCH+vCpro^*vCco/vCH$$

$$FORM21=DvCHDAJ/vCH+DvCADAJ/vCA-DvCAHDAJ/vCAH$$

$$FORM22=(aa-vCpro^*vCco/(vCH^{**2}))*DvCHDAJ$$

$$FORM31=DDvCHDDAJ/vCH-(DvCHDAJ/vCH)^{**2}+DDvCADD AJ/vCA-DvCADAJ/vCA)^{**2}-DDvCAHDD$$

$$AJ/vCAH+(DvCAHDAJ/vCAH)^{**2}$$

$$FORM32=(aa-vCpro^*vCco/(vCH^{**2}))*DDvCHDDAJ+2^*vCpro^*vCco/(vCH^{**3})*(DvCHDAJ^{**2})$$

$$FORM41=DDDvCHDDAJ/vCH-3^*DDvCHDDAJ^*DvCHDAJ/vCH^{**2}$$

$$+ 2^*(DvCHDAJ/vCH)^{**3}+DDDvCADD AJ/vCA-3^*DDvCADD AJ^*DvCADAJ$$

$$/ vCA^{**2}+2^*(DvCADAJ/vCA)^{**3}-D3vCAHD3AJ/vCAH+3^*DDvCAHDDAJ$$

$$* DvCAHDAJ/vCAH^{**2}-2^*(DvCAHDAJ/vCAH)^{**3}$$

$$FORM42=(aa-vCpro^*vCco/(vCH^{**2}))*DDDvCHDDAJ+6^*vCpro^*$$

$$vCco/(vCH^{**3})*DDvCHDDAJ^*DvCHDAJ-6^*vCpro^*vCco/(vCH^{**4})^* DvCHDAJ^{**3}$$

$$GAJ=detF0^*AJ^*vCA^*FORM11+f^*(log(detF0^*AJ)-log(f)+log(vCAH))+(detF0^*AJ-1)^*(vCpro+vCco+$$

$$vCcou)- detF0^*AJ^*FORM12$$

$$FI=vCpro+vCco+vCcou$$

$$SE=f^*(1/(detF0^*AJ)+DvCAHDAJ/vCAH)$$

$$TH=(vCA+detF0^*AJ^*DvCADAJ)^*FORM11$$

$$FO=detF0^*AJ^*vCA^*FORM21-FORM12-AJ^*detF0^*FORM22$$

$$DGDAJ=FI+SE+TH+FO$$

$$DDGDDAJ=f*(-1/(\det F0*AJ)**2+DDvCAHDDAJ/vCAH-(DvCAHDAJ/vCAH)**2)+*FORM21*(vCA+\det F0*$$

AJ

$$*DvCADAJ)+FORM11*(2*DvCADAJ+\det F0*AJ*DvCADD AJ)+\det F0*vCA*AJ*FORM31-2*FORM22-AJ*d$$

etF0*FORM32

$$DDDGD DDAJ=f*(2/(\det F0*AJ)**3+D3vCAHD3AJ/vCAH-3*DvCAHDAJ*DDvCAHDDAJ/vCAH**2+2*$$

$$(\text{DvCAHDAJ/vCAH})**3)+3*FORM21*(2*DvCADAJ+\det F0*AJ*DDvCADD AJ)+3*(vCA+\det F0*$$

AJ

$$*DvCADAJ)*FORM31+FORM11*(3*DDvCADD AJ+\det F0*AJ*DDDvCADD DDAJ)+\det F0$$

$$*AJ*vCA*FORM41-3*FORM32-\det F0*AJ*FORM42$$

$$U(1)=U(1)+GAJ/\det F0$$

$$UI1(3)=UI1(3)+DGDAJ$$

$$UI2(3)=UI2(3)+DDGDDAJ*\det F0$$

$$UI3(6)=UI3(6)+DDDGD DDAJ*\det F0**2$$

RETURN

END

References

1. S.S. Sternstein, *J. Macromol. Sci. Phys B*, 1972, **6**, 243-262.
2. X.H. Zhao, W. Hong, Z.G. Suo, *Appl. Phys. Lett.*, 2008, **92**, 051904.