

Supporting Information

Study of poly(*N,N*-diethylacrylamide) nanogel formation by aqueous dispersion polymerization of *N,N*-diethylacrylamide in the presence of poly(ethylene oxide)-*b*-poly(*N,N*-dimethylacrylamide) amphiphilic macromolecular RAFT agents.

Chloé Grazon,^{1,2} Jutta Rieger,^{1} Nicolas Sanson,² Bernadette Charleux^{1,3}*

1 Laboratoire de Chimie des Polymères, UPMC Univ. Paris 6 and CNRS, UMR 7610, 4 place Jussieu, Tour 44-54, 75252 Paris Cedex 05, France.

E-mail: jutta.rieger@upmc.fr

2 Laboratoire de Physico-chimie des Polymères et Milieux Dispersés, UMR 7615 UPMC-CNRS-ESPCI, Ecole Supérieure de Physique et de Chimie Industrielles ESPCI, 10 rue Vauquelin, 75231 Paris Cedex 05, France.

3 Université de Lyon, Univ. Lyon 1, CPE Lyon, CNRS UMR 5265, C2P2, Team LCPP Bat 308F, 43 Bd du 11 novembre 1918, 69616 Villeurbanne, France.

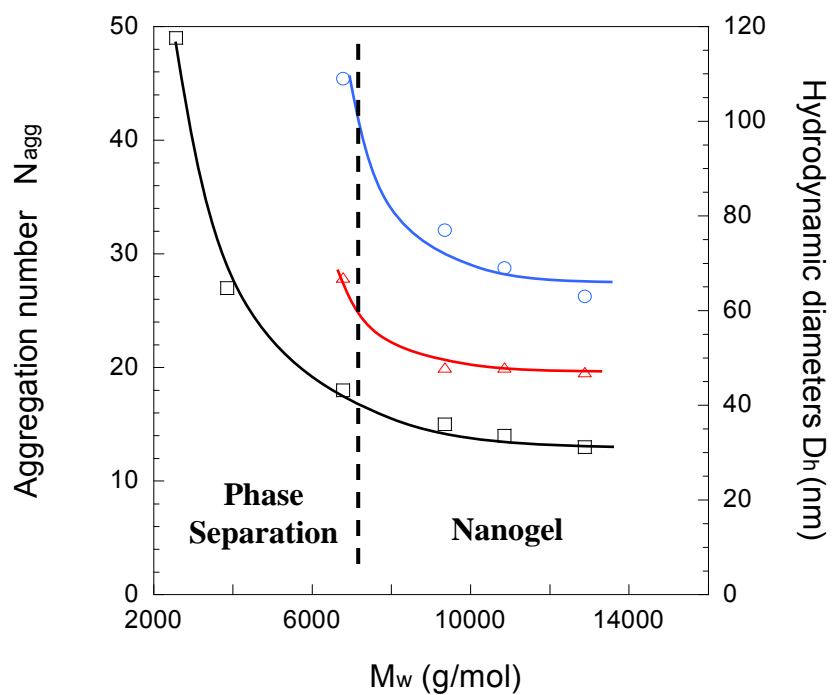


Figure SI-1. Comparison of the evolutions of the aggregation number, N_{agg} , of macroRAFT agent micelles (□) and the hydrodynamic diameters, D_h , of nanogels measured at 15°C (○) and 50°C (△) as a function of the weight-average molar mass, M_w , of the macroRAFT agent. The lines are guides to the eyes. The dotted line represents the frontier which separates stable and non stable nanogels.

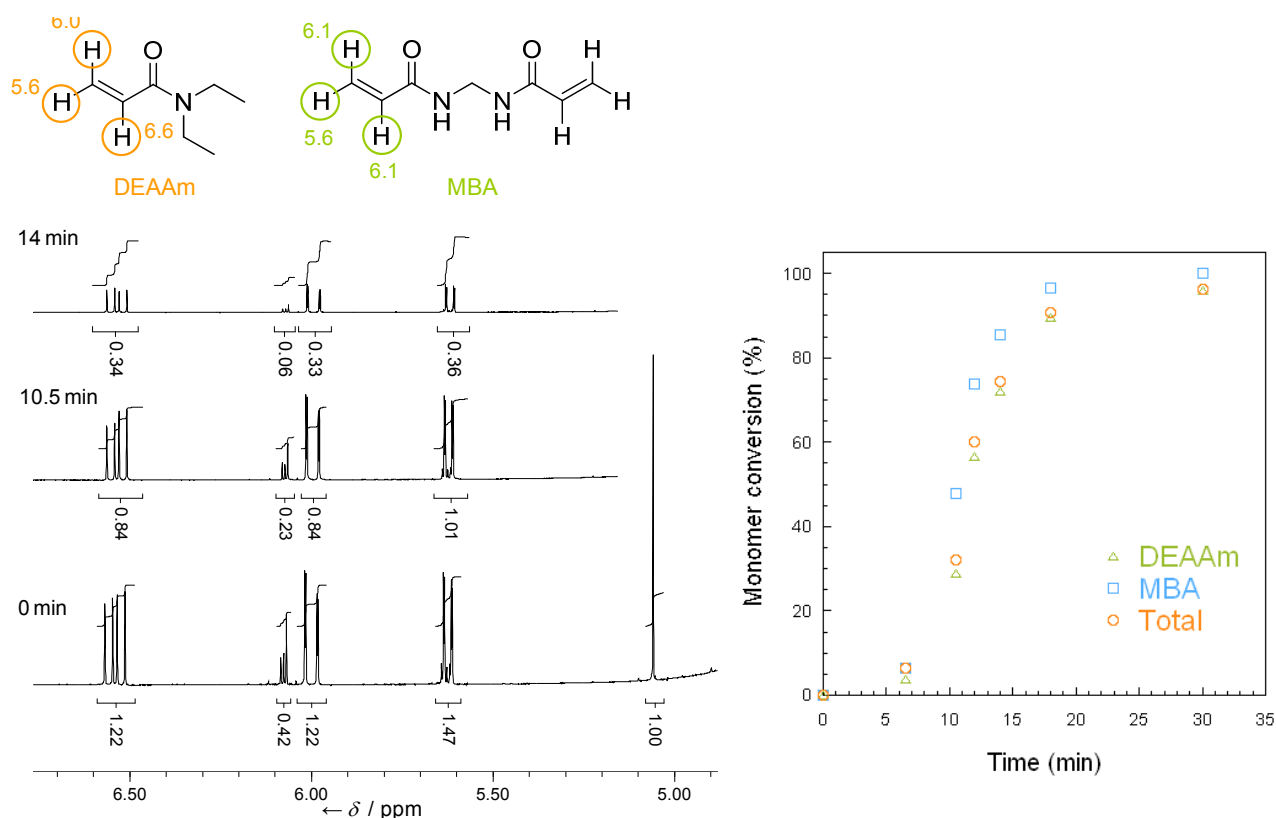


Figure SI-2. Left: Kinetic ¹H-NMR analysis (500 MHz, D₂O) of nanogel synthesis **S3CL8** (monomer concentration = 3 wt%, MBA = 8 mol%); right: Individual and global conversion of MBA and DEAAm vs. time.

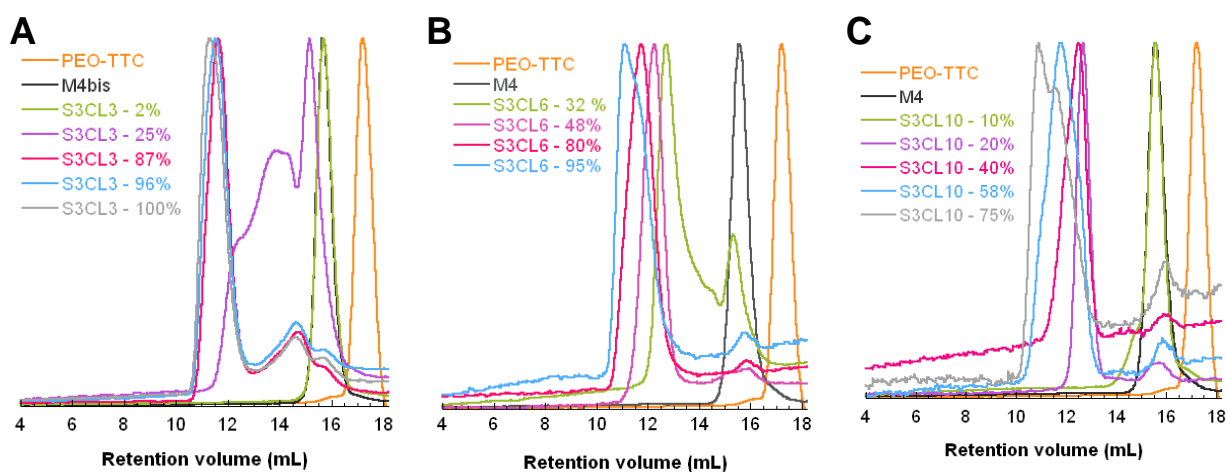


Figure SI-3. Size exclusion chromatograms in DMF (+ LiBr) at various monomer conversions for the RCC performed with 3 wt% of monomer and different crosslinker (CL) concentrations: A) **S3CL3**, B) **S3CL6**, C) **S3CL10** (cf. Table 2). The PEO-TTC and PEO-*b*-PDMAAm-TTC (**M4bis** and **M4**) macroRAFT agents are also represented.