Electronic Supplementary Information

Controlled Self-Assembly of α -Helix-Decorated Peptide Nanostructures

Sung-ju Choi,^a Woo-jin Jeong,^a Tae-Hyun Kim^b and Yong-beom Lim^{*a}

 ^a Translational Research Center for Protein Function Control & Department of Materials Science & Engineering, Yonsei University, Seoul 120-749, Korea.
Fax: +82 2-312-5375; Tel: +82 2-2123-5836; E-mail: yblim@yonsei.ac.kr
^b Department of Chemistry, University of Incheon, Incheon 406-840, Korea.



PEP-1



PEP-3



PEP-2





Figure S1 Peptide building block structures.



Figure S2 Effect of acetonitrile (ACN) on the β -sheet-mediated self-assembly of PEP-2. CD spectrum of PEP-2 in water (blue) and in water:ACN (1:1, red).



Figure S3 CD spectrum of PEP-1 & PEP-2 coassembly.



Figure S4 FTIR spectrum of PEP-1 and PEP-2 coassembly. Inset: enlarged amide region.



Figure S5 Effects of acetonitrile evaporation rate on the coassembly of PEP-1 and PEP-2. In this sample, acetonitrile was rapidly evaporated (< 20 min) by using a centrifugal vacuum concentrator.



Figure S6 Effects of temperature on the stability of a) PEP-1 and b) PEP-1 and PEP-2 coassembly.



Figure S7 Dynamic light scattering data for PEP-1 nanostructures (blue) and the coassembled nanostructures of PEP-1 and PEP-2 (green).

Materials and Methods

Synthesis. Peptide was synthesized on Rink Amide MBHA resin LL (Novabiochem) using standard Fmoc protocols on a TributeTM peptide synthesizer (Protein Technologies, Inc). Standard amino acid protecting groups were employed except cysteine, in which an acid-labile methoxytrityl (Mmt) group was used. The oligoethylene glycol-based linker, N-(Fmoc-8-amino-3,6dioxaoctyl)succinamic acid (Fmoc-PEG₂-Suc-OH), was purchased from Anaspec. The peptideattached resin (20 µmol of N-terminal amine groups) was swollen in N-methyl-2-pyrrolidone (NMP) for 30 min. For cyclization, bromoacetic acid was first coupled to the N-terminal part of the resinbound peptide. Before addition to the resin, a mixture of bromoacetic acid (28 mg, 200 µmol) and $N_{\rm N}$ '-diisopropylcarbodiimide (15.5 µL, 100 µmol) in NMP was incubated for 10 min for carboxyl activation. The reaction was continued for 1 h with shaking at room temperature, in a 6 mL polypropylene tube with a frit (Restek). The resin was then washed successively with NMP and dichloromethane (DCM). For orthogonal deprotection of the Mmt group from the cysteine, the resin was treated with 1% trifluoroacetic acid (TFA) in DCM several times (1 min $\times -5$). Intramolecular cyclization reaction was performed in 3 mL of 1% diisopropylethylamine (DIPEA) in NMP overnight with shaking at room temperature. The resin was then successively washed with NMP and acetonitrile, and dried in vacuo. The dried resin was treated with cleavage cocktail (TFA: 1,2ethanedithiol: thioanisole; 95 : 2.5 : 2.5) for 3 h, and was triturated with *tert*-butyl methyl ether. The peptides were purified by reverse-phase HPLC (water-acetonitrile with 0.1% TFA). The molecular weight was confirmed by MALDI-TOF mass spectrometry. The purity of the peptides was >95% as

determined by analytical HPLC. Concentration was determined spectrophotometrically in 8 M urea using a molar extinction coefficient of tryptophan (5,500 M⁻¹cm⁻¹) at 280 nm.

Circular dichroism. CD spectra were measured using a ChirascanTM Circular Dichroism Spectrometer equipped with peltier temperature controller (Applied Photophysics., Ltd). Spectra were recorded from 250 nm to 190 nm using a 2 mm path-length cuvette. Scans were repeated five times and averaged. Molar ellipticity was calculated per amino acid residue. Peptide concentration was 5 μ M. Peptides were typically prepared in 20 mM KF solution unless described otherwise. Sample solutions were incubated at least for 2 days before measurement, and essentially the same CD spectra were obtained after prolonged incubation, indicating thermodynamic equilibrium states.

Transmission electron microscopy. Three μ L of sample was placed onto a carbon-coated copper grid and dried completely. Then 2 μ L of 1 % (w/v) uranyl acetate solution was added for 1 min and excess solution was wicked off by filter paper. Sample concentrations were typically 5 - 20 μ M in 20 mM KF. The specimen was observed with a JEOL-JEM 2010 instrument operating at 100 kV.

Infrared spectroscopy. For FT-IR measurement, 300 μ L of the sample (50 μ M in 20 mM KF) was cast from the solution onto ZnSe window. Twenty thousand scans were acquired on a Bruker Vertex 70 FT-IR spectrometer.

Dynamic light scattering (DLS). DLS experiment was performed at room temperature with LV/CGS-3 Compact Goniometer System equipped with He-Ne laser operating at 632.8 nm. The scattering angle was 90°. Before measurement, the sample was centrifuged at $16,110 \times g$ for 20 min to sediment any dust particles. Sample concentrations were typically 5 - 20 μ M in 20 mM KF. The size distribution was determined by using a constrained regularization method.