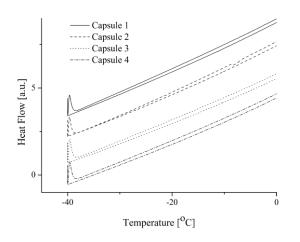
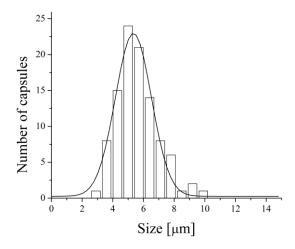
## **Supporting Information**

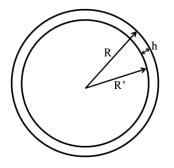


S1. DSC graph (endo up) of the four different PFO-PLLA capsules after lyophilisation.



S2 Size distribution of the capsules (based on 100 capsules counted by hand). The size of the capsules is  $5.3 \pm 2.3 \ \mu m$ .

Equation I



Calculating the shell thickness of a capsule (h) with a variable radius of the capsule (R):

$$v_{capsule} = v_{core} + V_{shell}$$

$$\frac{4}{3}\pi R^3 = \frac{4}{3}\pi (R')^3 + \frac{m_{polymer}}{\rho_{polymer}}$$

$$m_{polymer}$$

$$\frac{4}{3}\pi R^3 = \frac{4}{3}\pi (R')^3 + \frac{\overline{\rho_{\text{polymer}}}}{V_{\text{decane}}} V_{\text{core}}$$

Assuming that all the polymer ends up in the shell, this results in:

$$V_{\text{decane}} = V_{\text{core}} = \frac{4}{3}\pi (R')^3$$

$$R' = \frac{R}{\sqrt[3]{1+f}} , \text{ where } f = \frac{P \text{polymer}}{V_{decane}}$$

h=R-R'

with  $m_{polymer}$  = weighed amount of polymer [g],  $\rho_{polymer}$  = density of the polymer [g/cm<sup>3</sup>],<sup>7</sup> V<sub>decane</sub> = amount of decane [mL], R = measured diameter of the capsule/2 [µm] and h = thickness of the shell [µm].