

Supplementary material

1 Raw data of the conductivity

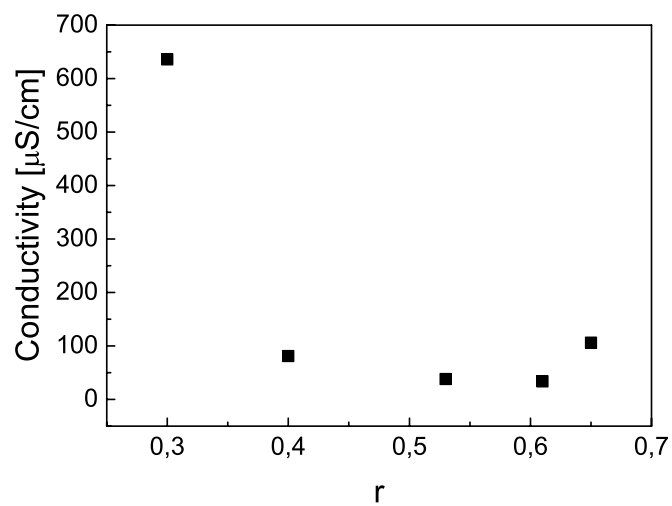


Figure 1: *Conductivity of pure cationic solutions at different ratio of anionic surfactant r .*

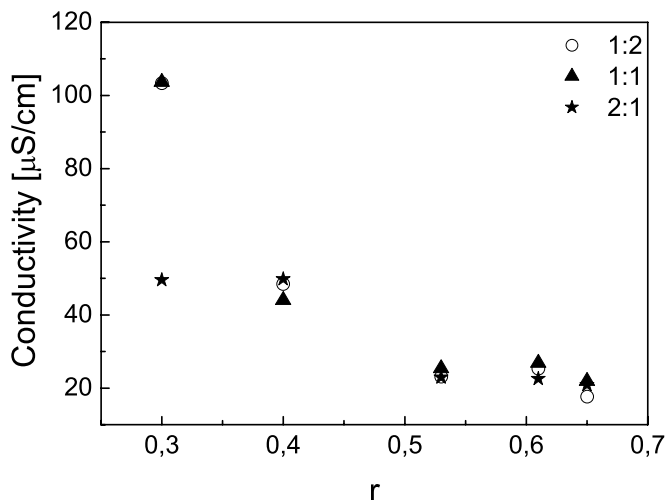


Figure 2: Conductivity of emulsions as a function of the r -ratio of the cationics at different oil/ cationic solution ratios.

2 Calculation of the surface tension for cationics adsorbed at the air-water interface

$$\sigma_{r,A} = \sigma_{r,A}(\text{dispersion}) + \sigma_{r,A}(\text{polar}) \quad (1)$$

and

$$\sigma_{r,T} = \sigma_{r,A} + \sigma_{T,A} - 2\sqrt{\sigma_{r,A}(\text{dispersion})\sigma_{T,A}(\text{dispersion})} \quad (2)$$

where $\sigma_{T,A} = \sigma_{T,A}(\text{dispersion})$ and $\sigma_{T,A}(\text{polar})=0$.

The surface tension of tetradecane is $\sigma_{T,A}=26.6$ mN/m and the interfacial tension of tetradecane-water is $\sigma_{T,W}=52.9$ mN/m.

From equation 2 it follows that $\sigma_{r=0.4,A}(\text{dispersion})=21.3\pm 1$ mN/m, $\sigma_{r=0.4,A}(\text{polar})=3.7\pm 1$ mN/m, $\sigma_{r=0.6,A}(\text{dispersion})=22.7\pm 1$ mN/m, and $\sigma_{r=0.6,A}(\text{polar})=3.3\pm 1$ mN/m.

Finally from equation 3,¹

$$\sigma_{r,W} = \sigma_{r,A} + \sigma_{W,A} - 2\sqrt{\sigma_{r,A}(\text{dispersion})\sigma_{W,A}(\text{dispersion})} - 2\sqrt{\sigma_{r,A}(\text{polar})\sigma_{W,A}(\text{polar})} \quad (3)$$

where $\sigma_{W,A}(\text{dispersion})=21.5$ mN/m and $\sigma_{W,A}(\text{polar})=50.4$ mN/m, one can calculate $\sigma_{r=0.4,W}=26.8\pm 1$ mN/m and $\sigma_{r=0.6,W}=27.9\pm 1$ mN/m.

References

- [1] Binks, B.; Clint, J. H. *Langmuir* **2002**, *18*, 1270–1273.