

Supplementary Information for the Manuscript Entitled

**pH-Triggered Block Copolymer Micelles Based  
on a pH-Responsive PDPA (Poly[2-  
(diisopropylamino)ethyl Methacrylate]) Inner  
Core and a PEO (Poly(ethylene Oxide)) Outer  
Shell as a Potential Tool for the Cancer  
Therapy**

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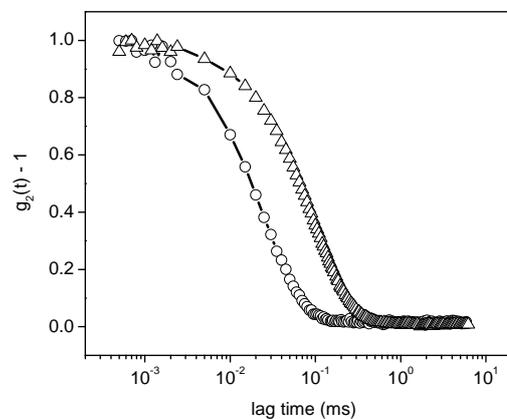
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## Dynamic Light Scattering



**Figure S1.** Autocorrelation functions for  $4.0 \text{ mg.mL}^{-1}$   $\text{PEO}_{113}\text{-}b\text{-PG2MA}_{30}\text{-}b\text{-PDPA}_{50}$  dissolved in ethanol (○) and  $0.25 \text{ mg.mL}^{-1}$   $\text{PEO}_{113}\text{-}b\text{-PG2MA}_{30}\text{-}b\text{-PDPA}_{50}$  prepared by using the nanoprecipitation protocol (Δ) in PBS (pH 7.4) both at  $37 \text{ }^\circ\text{C}$ .

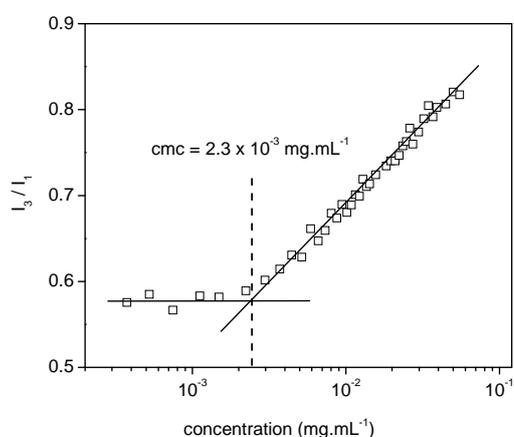
## Steady-state fluorescence

### Experimental

Steady-state fluorescence spectra of pyrene were recorded on a Hitachi F4500 Spectrofluorimeter. The block copolymer solutions over a wide range of concentration were excited at 336 nm and the emission spectra were recorded from 360 to 500 nm.

### Result

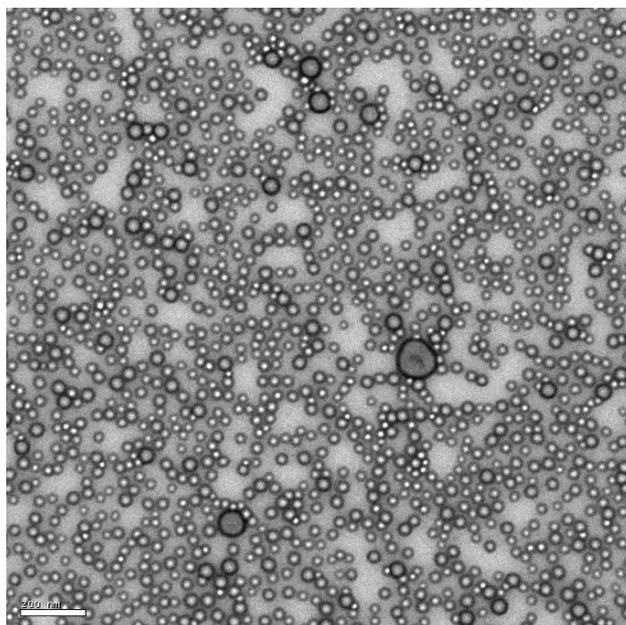
Techniques using fluorescent probes such as pyrene have been extensively used to monitor the cmc of a variety of aggregation processes. The ratio between the bands  $I_1$  (372.8 nm) and  $I_3$  (384.0 nm) of the fluorescent pyrene probe is sensitive to small changes in the environment polarity. The  $I_3/I_1$  ratio of pyrene is used as a criterion to evaluate the micropolarity of micellar aggregate microenvironments. Since the pyrene probe migrates from the polar aqueous medium to the apolar micellar cores during the micellization, the ratio  $I_3/I_1$  enables us to follow the aggregation path. The profile of  $I_3/I_1$  vs. PEO<sub>113</sub>-*b*-PG2MA<sub>30</sub>-*b*-PDPA<sub>50</sub> concentration is shown in Figure S1. In the presence of small amounts of PEO<sub>113</sub>-*b*-PG2MA<sub>30</sub>-*b*-PDPA<sub>50</sub> the  $I_3/I_1$  ratio is around 0.60 which is compatible with the water polarity. The cmc value was determined at the concentration where the plateau ends. According to this methodology, the cmc value is equal to  $\text{cmc} = 2.3 \times 10^{-3} \text{ mg.mL}^{-1}$ .



**Figure S2.**  $I_3/I_1$  vs. PEO<sub>113</sub>-*b*-PG2MA<sub>30</sub>-*b*-PDPA<sub>50</sub> concentration in PBS solution.

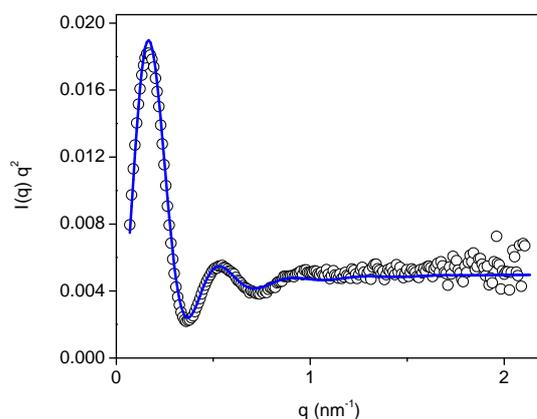
### Transmission Electron Microscopy (TEM)

The formation of mainly narrowly distributed nanosized spherical block copolymer micelles is evidenced in the TEM image given in Figure S3. The mean micelle diameter ( $2R$ ) calculated by image analysis was 35 nm, which is in good agreement with DLS results.



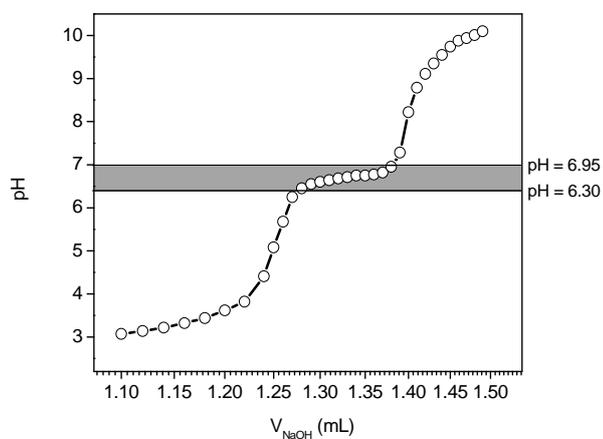
**Figure S3.** TEM image of the self-assembled block copolymer micelles.

### Small-Angle X-ray Scattering (SAXS)



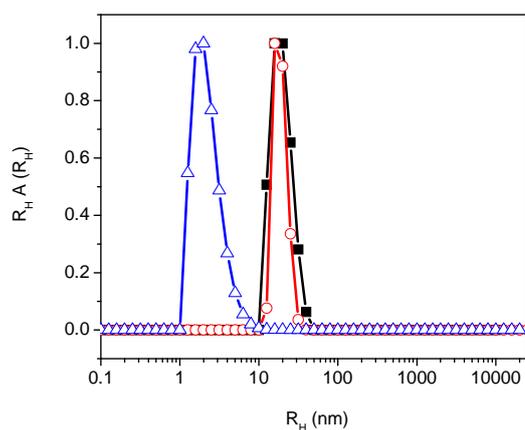
**Figure S4.** Kratky plot ( $I(q)q^2$  vs.  $q$ ) for  $5.0 \text{ mg.mL}^{-1}$  PEO<sub>113</sub>-*b*-PG2MA<sub>30</sub>-*b*-PDPA<sub>50</sub> in PBS solution (pH 7.4 at 37 °C).

## pH Titration



**Figure S5.** Potentiometric acid-base titration curve for  $0.10 \text{ mg.mL}^{-1}$   $\text{PEO}_{113}\text{-}b\text{-PG2MA}_{30}\text{-}b\text{-PDPA}_{50}$  aqueous solution at ionic strength  $I = 0.15 \text{ mol.L}^{-1}$ . The solution was titrated with  $\text{NaOH } 0.010 \text{ mol.L}^{-1}$ .

## Number-weighted Distributions



**Figure S6.** Number-weighted distributions of  $R_H$  for  $4.0 \text{ mg.mL}^{-1}$   $\text{PEO}_{113}\text{-}b\text{-PG2MA}_{30}\text{-}b\text{-PDPA}_{50}$  in PBS ( $37 \text{ }^\circ\text{C}$ ) and different pH values ( $\Delta$  - 5.5;  $\circ$  - 6.5;  $\blacksquare$  - 7.4).

