

## Electronic Supplementary Information

for

### Self-Assembly of a Triple-Zwitterion in Polar Solutions: hierarchical Formation of Nanostructures

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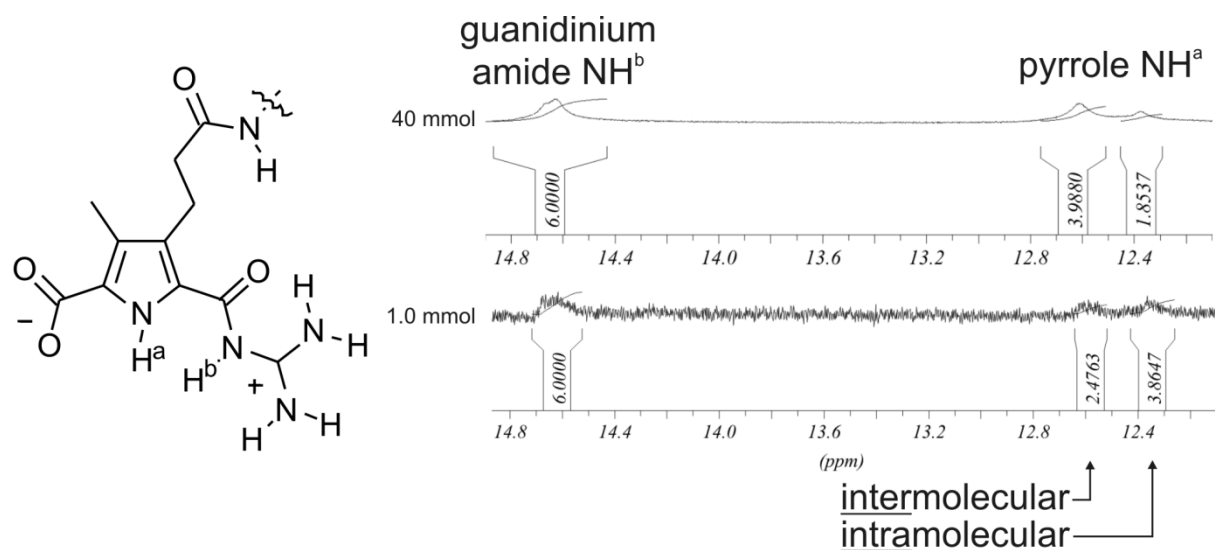
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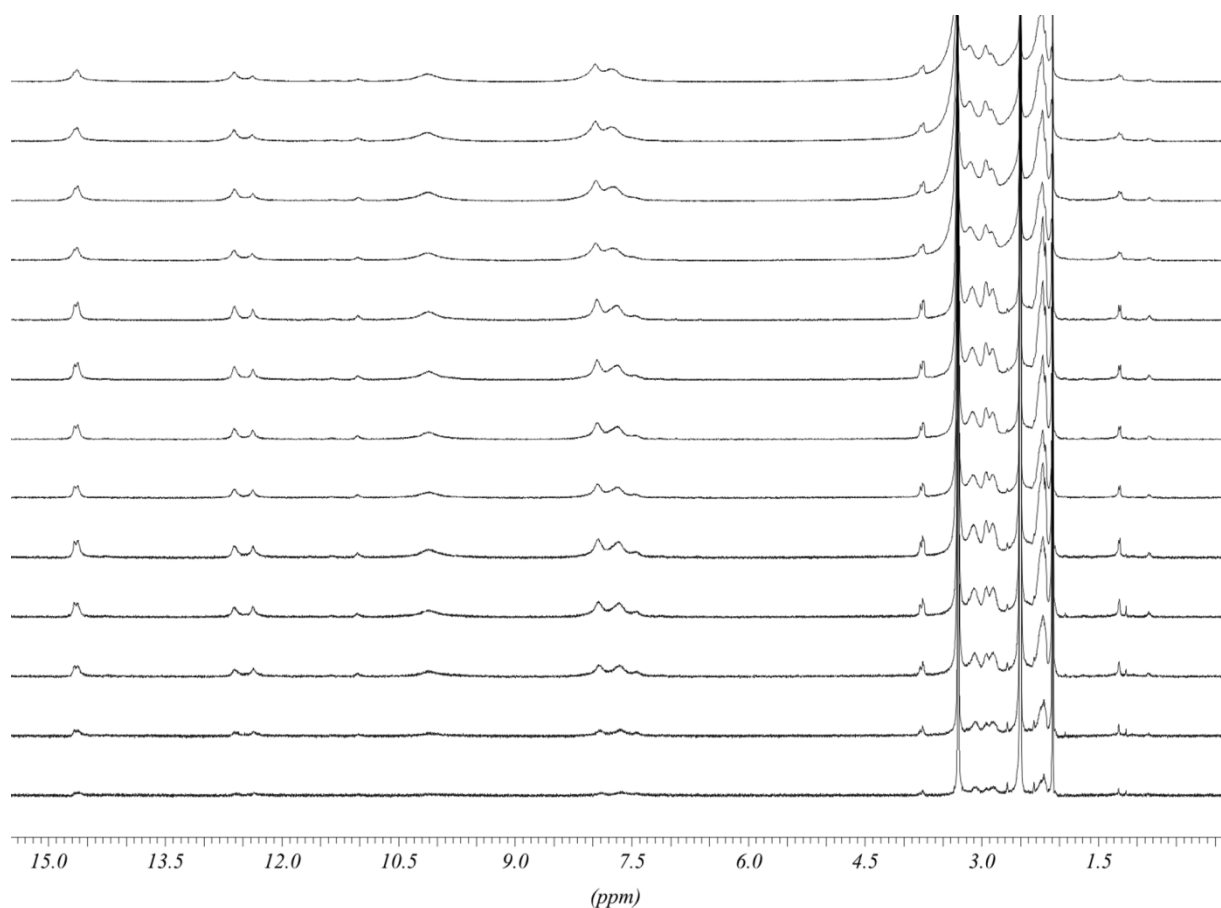
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## 1. $^1\text{H}$ NMR dilution studies



**Figure S1.** Inversion of the  $^1\text{H}$  NMR signal integral by changing binding mode from *intra*- to *intermolecular* with increasing concentration: pyrrole  $\text{NH}_{\text{intra}}$  : pyrrole  $\text{NH}_{\text{inter}} \approx 1.2 : 1.9$  (1 mM) and  $2 : 0.9$  (40 mM).



**Figure S2.**  $^1\text{H}$  NMR dilution row for triple-zwitterion **1** in a concentrations range from 1 to 40 mM in  $\text{DMSO}-d_6$  (from bottom to top).

## 2. $^1\text{H}$ DOSY NMR studies

**Table S1.** Acquisition parameters for the DOSY experiments

	<b>1 (2 mM)</b>	<b>1 (20 mM)</b>	<b>7 (2 mM)</b>
<b>PULPROG</b>	dstebpgp3s	dstebpgp3s	dstebpgp3s
<b>TD[F2]</b>	64K	64K	64K
<b>TD[F1]</b>	32	32	32
<b>NS</b>	384	128	512
<b>DS</b>	64	64	64
<b>D20</b>	50 ms	50 ms	100 ms
<b>D21</b>	5 ms	5 ms	5 ms
<b>P30</b>	5.4 ms	5.4 ms	5.4 ms
<b>P19</b>	1.1 ms	1.1 ms	1.1 ms
<b>GPZ6</b>	100 %	100 %	100 %
<b>GPZ7</b>	-13.17	-13.17	-13.17
<b>GPZ8</b>	-17.13	-17.13	-17.13

**Table S2.** **1** (2 mM,  $T = 303.4$  K)

$\delta$ [ppm]	$D \cdot 10^{-11}$ [m <sup>2</sup> /s]	$\log D$	
14.673	8.905	-10.050	
12.628	9.131	-10.039	
12.371	9.042	-10.044	
11.069	7.150	-10.146	
3.758	7.937	-10.100	
3.326	$8.068 \cdot 10^{-10}$	-9.093	water
2.868	7.643	-10.117	
2.503	$6.196 \cdot 10^{-10}$	-9.208	DMSO
2.188	7.847	-10.105	
2.096	7.725	-10.112	
0.000	$6.029 \cdot 10^{-10}$	-9.220	TMS

**Table S3.** **1** (20 mM,  $T = 303.4$  K)

$\delta$ [ppm]	$D \cdot 10^{-11}$ [m <sup>2</sup> /s]	$\log D$	
14.627	8.688	-10.061	
12.625	6.185	-10.209	
12.383	8.470	-10.072	
11.069	7.150	-10.146	
7.998	6.817	-10.166	
3.760	6.650	-10.177	
3.339	$7.820 \cdot 10^{-10}$	-9.107	water
2.870	6.320	-10.199	
2.504	$6.083 \cdot 10^{-10}$	-9.216	DMSO
2.179	6.050	-10.218	
0.000	$5.735 \cdot 10^{-10}$	-9.241	TMS

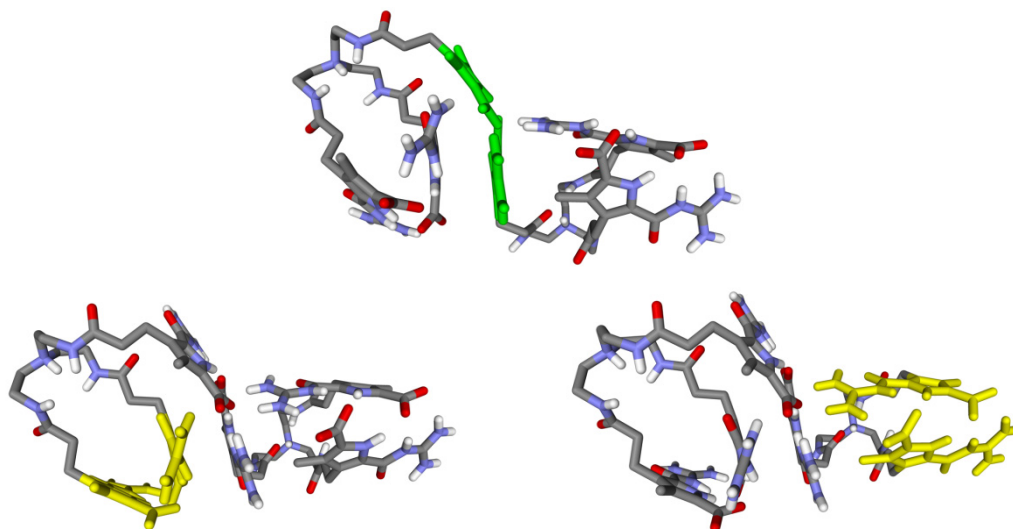
**Table S4.** **7** (2 mM,  $T = 303.4$  K)

$\delta$ [ppm]	$D \cdot 10^{-11}$ [m <sup>2</sup> /s]	$\log D$	
7.691	1.154	-9.953	
4.246	1.096	-9.960	
3.326	8.417	-9.075	water
3.057	1.098	-9.959	
2.924	1.082	-9.966	
2.505	6.535	-9.185	DMSO
2.432	1.112	-9.954	
2.262	1.089	-9.963	
2.160	1.093	-9.961	
1.469	1.091	-9.962	
1.290	1.105	-9.957	
0.003	6.213	-9.207	TMS

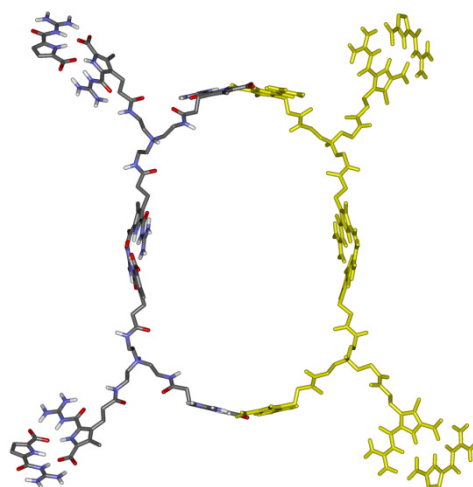
**Table S5.** Mean diffusion coefficients and corresponding hydrodynamic radii

	$\bar{D}$ [m <sup>2</sup> /s]	$r_H \cdot 10^{-9}$ [m]
<b>1</b> (2 mM)	$8.15 \cdot 10^{-11}$	1.37
<b>1</b> (20 mM)	$6.98 \cdot 10^{-11}$	1.60
<b>7</b> (2 mM)	$1.59 \cdot 10^{-10}$	0.70

### 3. Molecular Modelling



**Figure S3.** Calculated structure of dimeric **1**. The *intermolecular* binding motive is shown in green. Both *intramolecular* binding motives are shown in yellow. Non-polar hydrogen atoms are omitted for clarity.



**Figure S4.** End-capped tetramer as repeating element for the self-assembly of **1** (one capped dimer is highlighted in yellow; non-polar hydrogen atoms are omitted for clarity).

**Table S6.** Distances between the central nitrogen atoms of the *tren*-linker (A-A) and the maximum distance of the exterior binding motifs (B-B) as derived from the molecular mechanics calculations [ $\text{\AA}$ ].

A-A	B-B
23,5	34,6
22,6	33,5
24,2	33,2
22,1	20,6
19,9	32,4
24,9	29,8

24,5	29,5
20,5	27,1
24,7	28,5
22,5	26,8
24,3	29,7
26,5	27,3
22,3	33,4
23,9	33,9
25	34,6
23,2	30,5
25,4	36,4
23,7	37,2
22,8	30
24,3	35,4
18,2	27,8
24,4	26
22,2	34,8
23,2	32,9
19,1	27,1
23,4	34,4
22	34,4
24,1	22,1
22,3	31,9
20,9	30
21,7	24,3
24,5	33,4
22,7	35,7
20,9	23,9
22,8	32,8
21,1	30,1
19,4	25,5
22,9	21,7
23,5	37,5
23,6	27,8
23,3	33,8
16,5	34,7
21	27,6
24,1	32,7
23,6	36,7
23	24,7
33,4	33,5
17,6	30,3
20,5	28,9
22,6	28,9
23,5	34,9
23,1	26

22,2	34,6
19,9	30,4
23,3	31,1
23,4	32,8
19,6	28,6
22,8	24,4
23,3	33,2
24,1	31,9
<b>Ø 22,7</b>	<b>Ø 30,6</b>

10.  $^1\text{H}$  NMR spectra of the compounds 4, 5 and 7, and 1 as chloride salt or zwitterionic (10 mM) in  $\text{DMSO-}d_6$

