# Simple Design but Marvelous Performances: Molecular Gels of Superior Strength and Self-healing Properties

## Zhiyan Xu, Junxia Peng, Ni Yan, Hang Yu, Shasha Zhang, Kaiqiang Liu, Yu Fang\*

Key Laboratory of Applied Surface and Colloid Chemistry of Ministry of Education, School of Chemistry and Chemical Engineering, Shaanxi Normal University, Xi'an, 710062, P. R. China

#### 1. Gelation behaviors of the compounds

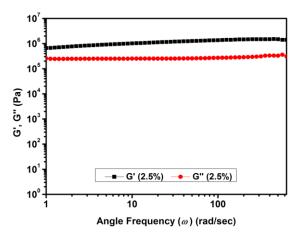
**Table S1** Gelation behaviors of compounds 1, 2, 3 and 4 in different solvents (2.5%, w/v).

solvents	1	2	3	4	solvents	1	2	3	4
methanol	P	P	P	P	kerosene	PG*	PG*	I	P
ethanol	PG*	P	P	P	ethyl ether	I	I	I	I
n-propanol	G	P	P	P	petroleum ether	PG*	G*	I	I
<i>n</i> -butanol	G	P	P	P	benzene	G	P	P	S
n-pentanol	PG*	P	P	S	toluene	G	P	P	G*
n-hexanol	PG*	P	S	S	acetic acid	I	P	P	P
n-heptanol	PG*	P	S	S	acetonitrile	I	P	P	P
n-octanol	PG*	P	S	S	TEA	PG*	I	I	VS
n-decanol	G	P	S	S	DMSO	P	G	G	S
$H_2O$	I	I	I	I	DMF	S	S	G	P
n-pentane	I	I	I	G*	THF	S	S	S	S
n-hexane	G*	G*	I	G*	acetone	G	P	S	S
n-heptane	G*	G*	I	I	ethyl acetate	G	P	S	S
n-octane	PG*	PG*	I	I	pyridine	S	S	S	S
n-nonane	PG*	PG*	I	I	CH <sub>2</sub> Cl <sub>2</sub>	P	P	S	S
n-decane	I	I	I	PG*	CHCl <sub>3</sub>	P	S	S	S
cyclohexane	PG*	I	I	PG*	CCl <sub>4</sub>	I	P	VS	G

G = Gel; PG = Partial gel; S = Solution; P = Precipitation; I = Insoluble; VS = Viscous solution.

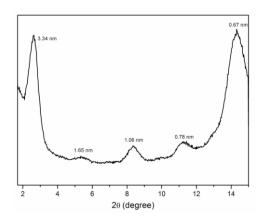
#### 2. Rheological Studies

 $G^* = Gels$  formed after sonication.



**Fig. S1** Evolution of G' and G'' as functions of the angle frequency. The applied shear stress is equal to 4 000 Pa. The sample is a gel of **1** in pyridine/methanol (v:v=4:6, 2.5%, w/v).

## 3. XRD Studies



**Fig. S2** XRD profile of 1/pyridine-methanol (v:v=4:6) recorded at room temperature.

## 4. Self-healing Behavior

Video