

Supporting Information

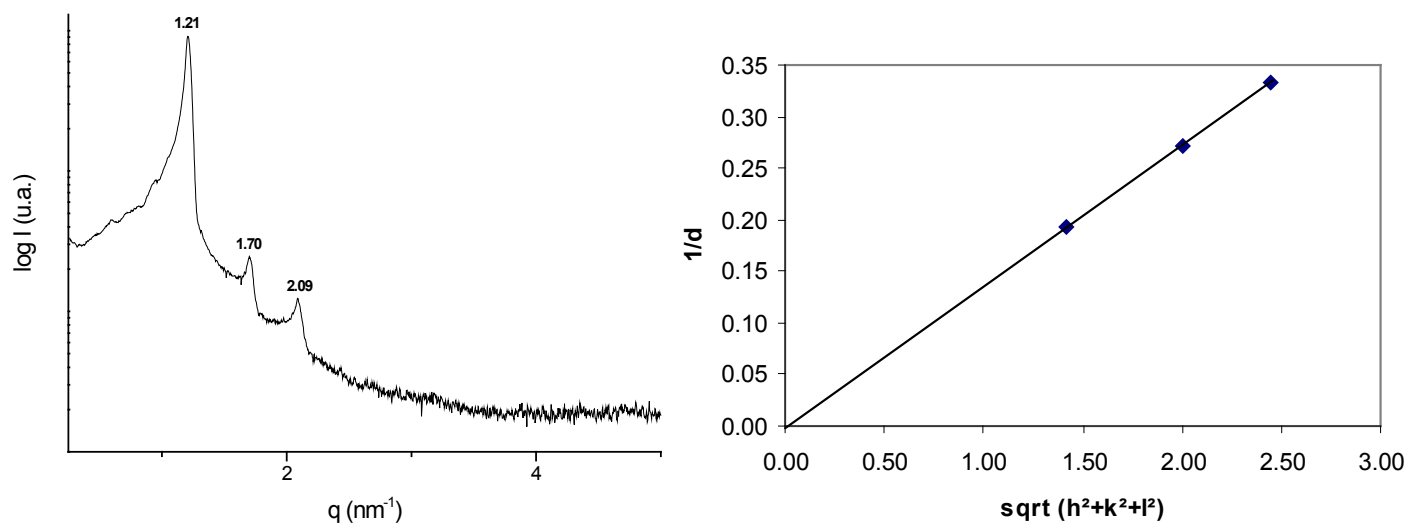
**Metastable micelles and true liquid crystal behaviour of newly designed "catanionic" surfactants**

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**Table S1.** X-Ray Diffraction Data for a 54.6% OOJ sample at 10°C indexed to the space group *Im3m* with a lattice parameter  $a = 7.4$  nm

<b>hkl</b>	<b><math>(h^2+k^2+l^2)^{1/2}</math></b>	<b><math>d_{obs}</math> (nm)</b>	<b><math>d_{cal}</math> (nm)</b>	<b><math>I_{obs}^*</math></b>
110	$\sqrt{2}$	5.2	5.2	vvs
200	$\sqrt{4}$	3.7	3.7	s
211	$\sqrt{6}$	3.0	3.0	s

\*  $I_{obs}$  are the observed intensities, which are ranged from vvs (extremely strong) to s (strong).

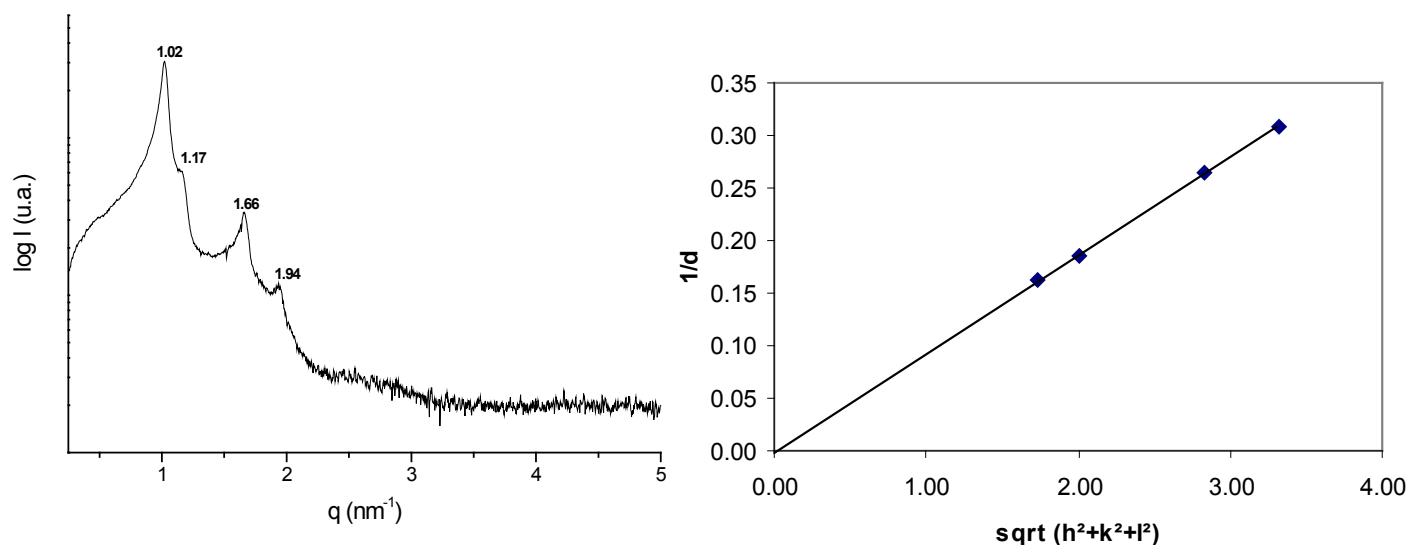


**Figure S1.** SAXS diffraction patterns obtained from a 54.6% OOJ sample at 10°C (left) and the plot of the reciprocal d spacing ( $1/d_{hkl}$ ) of the reflections observed (right).

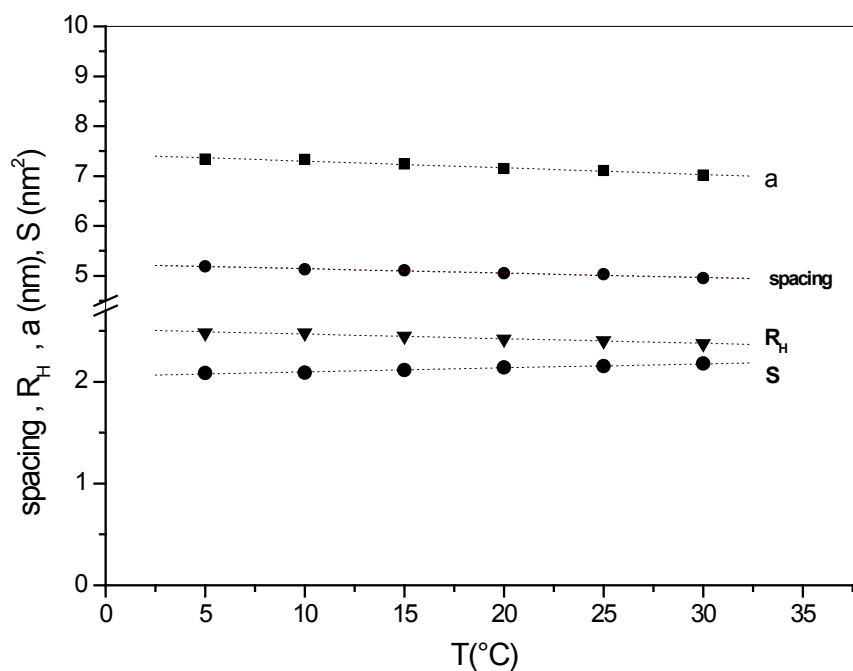
**Table S2.** X-Ray Diffraction Data for a 40.0% SSJ sample at 20°C indexed to the space group *Fm3m* with a lattice parameter  $a = 10.7$  nm

<b>hkl</b>	<b><math>(h^2+k^2+l^2)^{1/2}</math></b>	<b><math>d_{obs}</math> (nm)</b>	<b><math>d_{cal}</math> (nm)</b>	<b><math>I_{obs}^*</math></b>
111	$\sqrt{3}$	6.2	6.2	vs
200	$\sqrt{4}$	5.4	5.4	m
220	$\sqrt{8}$	3.8	3.8	s
311	$\sqrt{11}$	3.2	3.2	m

\*  $I_{obs}$  are the observed intensities, which are ranged from vs (very strong) to m (medium).



**Figure S2.** SAXS diffraction patterns obtained from a 40.0% SSJ sample at 20°C (left) and the plot of the reciprocal d spacing ( $1/d_{hkl}$ ) of the reflections observed (right).



**Figure S3.** Variation of the d spacing and of the structure parameters of the  $Im3m$  cubic phase of 47 wt% MOJ as a function of the temperature

**Table S3.** Molar volumes of the hydrophobic part of the surfactant,  $V_B$  and of the surfactant  $V_S$  for all studied systems. (considering  $\rho_{EO} = 1.13$ ,  $\rho_{alkyl} = 0.9$  and  $\rho_{PPO} = 1$ )

	$V_B$	$V_S$
LMJ	944.4	1400
MMJ	975.6	1428
LOJ	1004.4	1454
MOJ	1035.6	1482
MSJ	1037.8	1484
OOJ	1095.6	1536
SOJ	1097.8	1538
SSJ	1100.0	1540