

Assembly of the anisotropic microcapsules in aqueous dispersions

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Table S1. Free energy components of liquid and solid surfaces used in calculations.

Material	Surface free energy (mJ/m ²)			
	γ_s	γ_s^{LW}	γ_s^+	γ_s^-
water ¹	72.8	21.8	25.5	25.5
PMAA ^{2,3}	41	40	0	9

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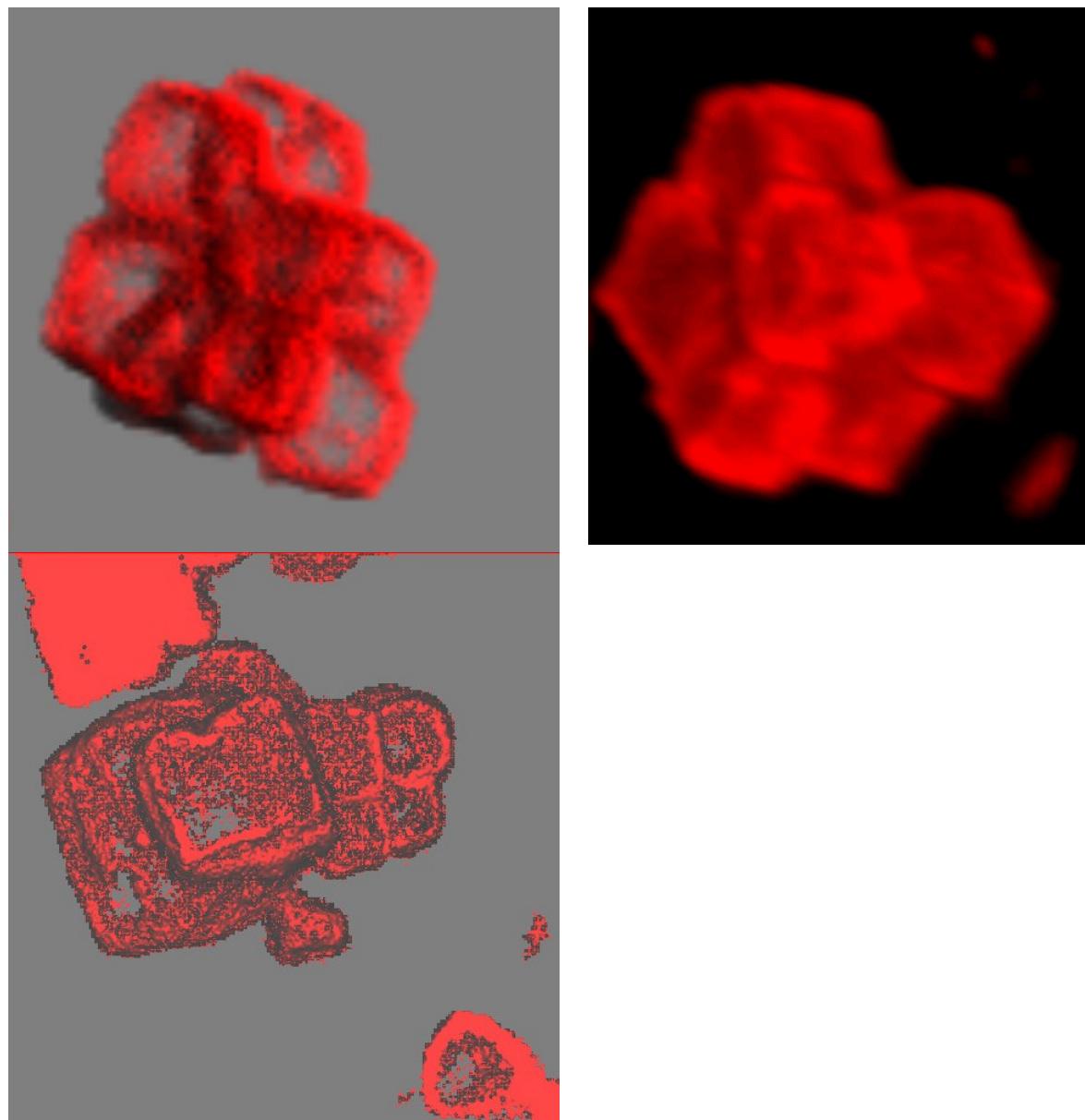


Figure S1. Confocal images of three dimensional (3D) assemblies of cubic $(\text{PVPON}/\text{PMAA})_4$ microcapsules in buffer pH 4.5, $C=0.1 \text{ M}$.

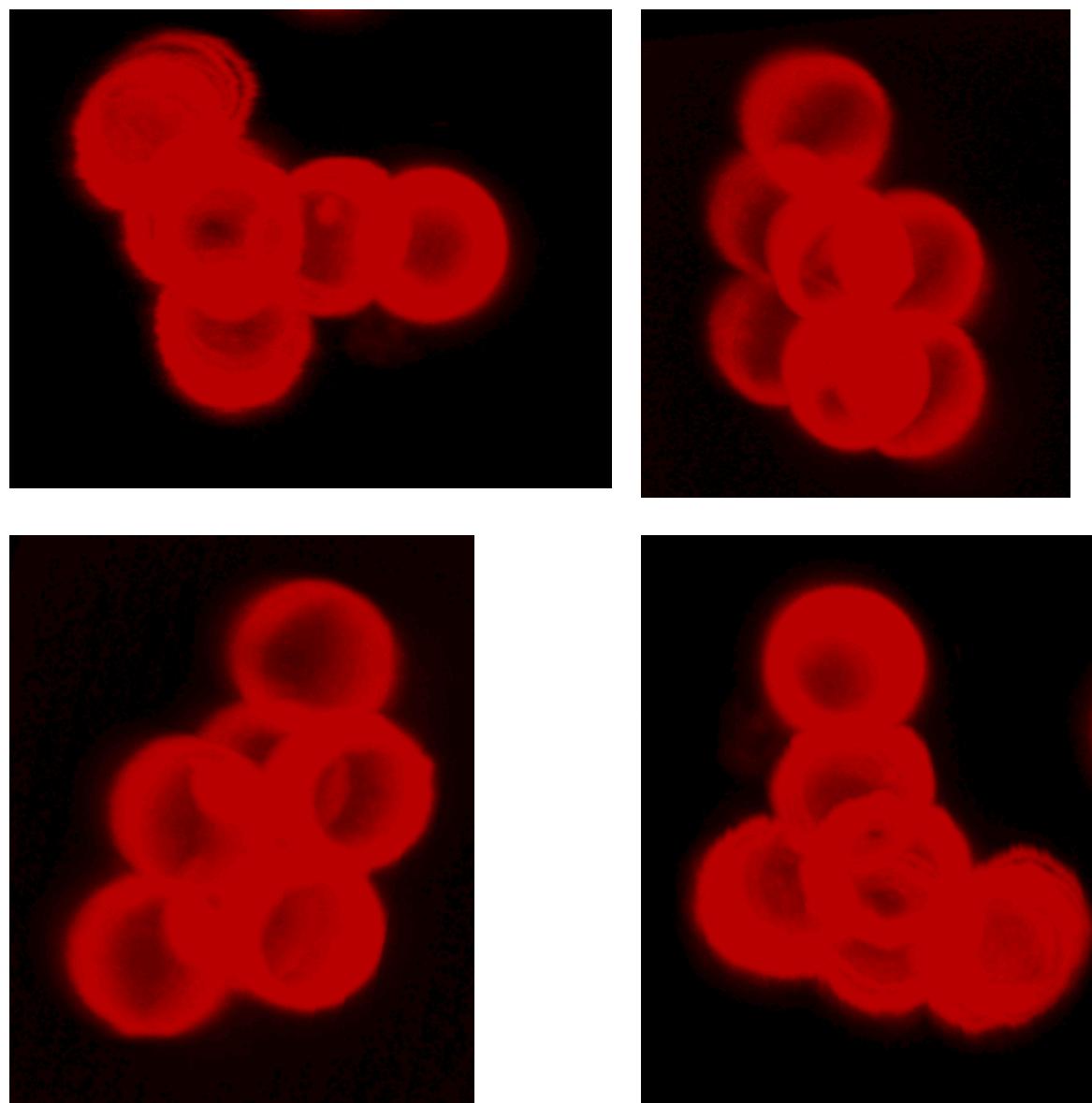


Figure S2. Confocal images of three dimensional (3D) assemblies of spherical $(\text{PVPON/PMAA})_4$ microcapsules in buffer pH 4.5, C=0.1M NaCl

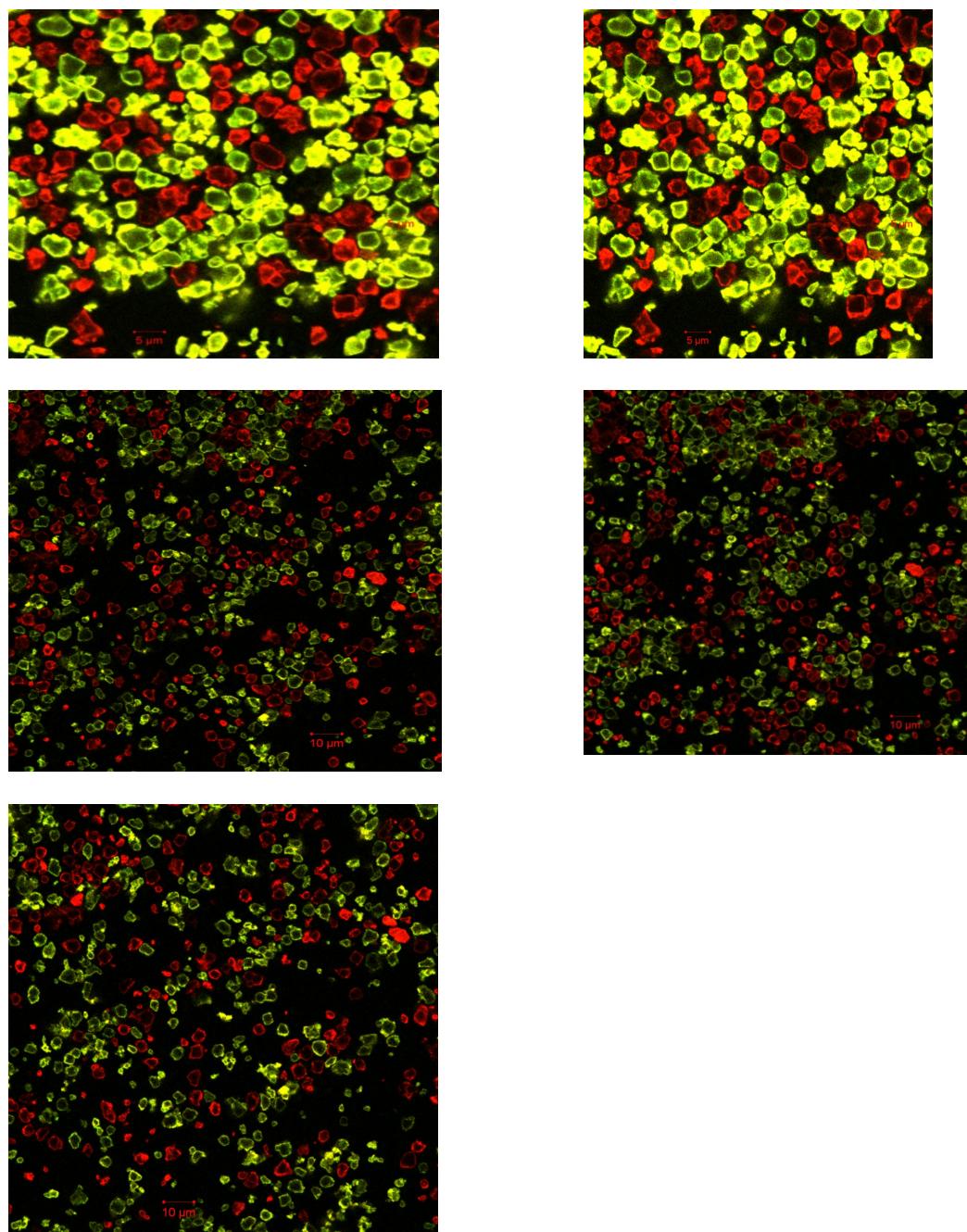


Figure S3. Confocal images of assembled cubic $(\text{PVPON/PMAA})_4$ and $(\text{PVPON/PMAA})_4\text{PPE}$ microcapsules in buffer pH 3.5, C= 0.01 M NaCl

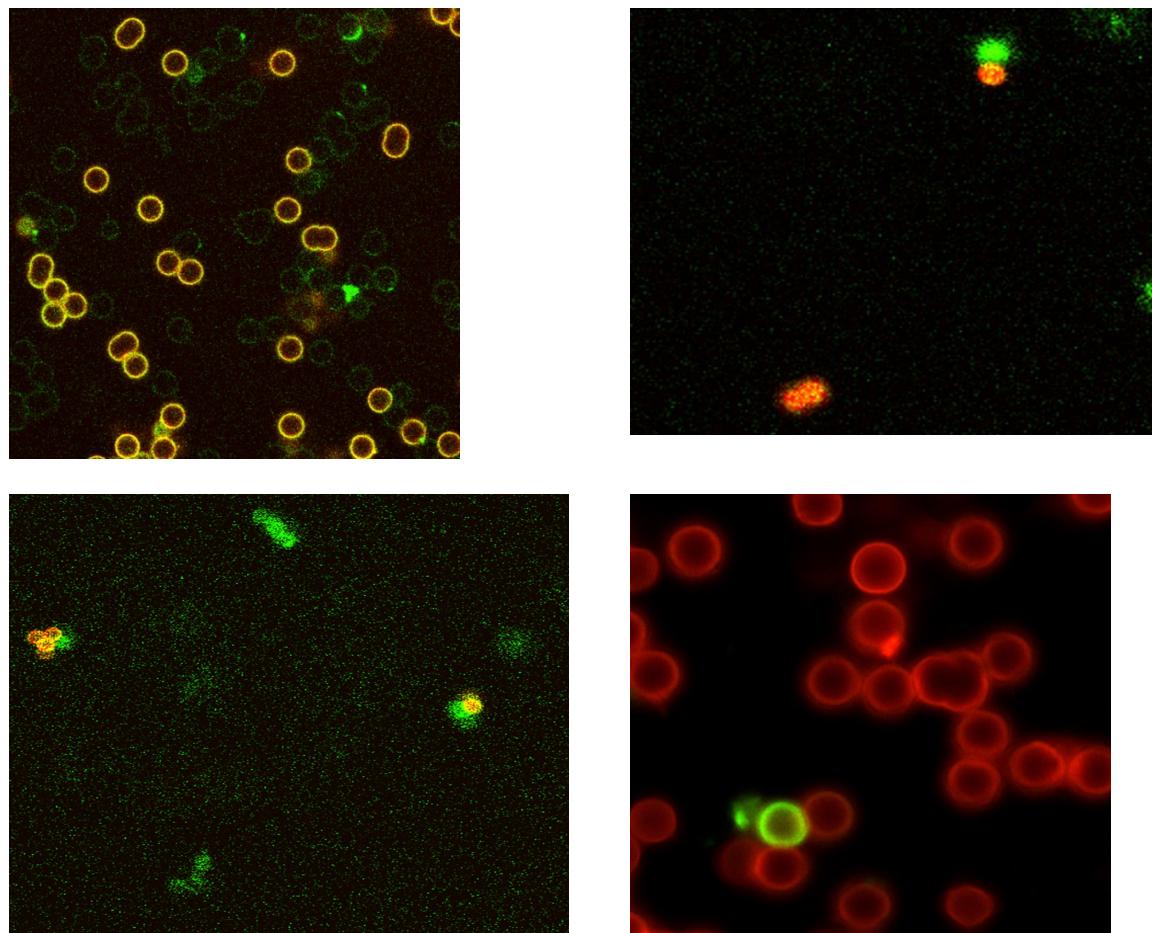


Figure S4. Confocal images of assembled spherical $(\text{PVPON/PMAA})_4$ and $(\text{PVPON/PMAA})_4\text{PPE}$ microcapsules in buffer pH 3.5, C= 0.01 M NaCl

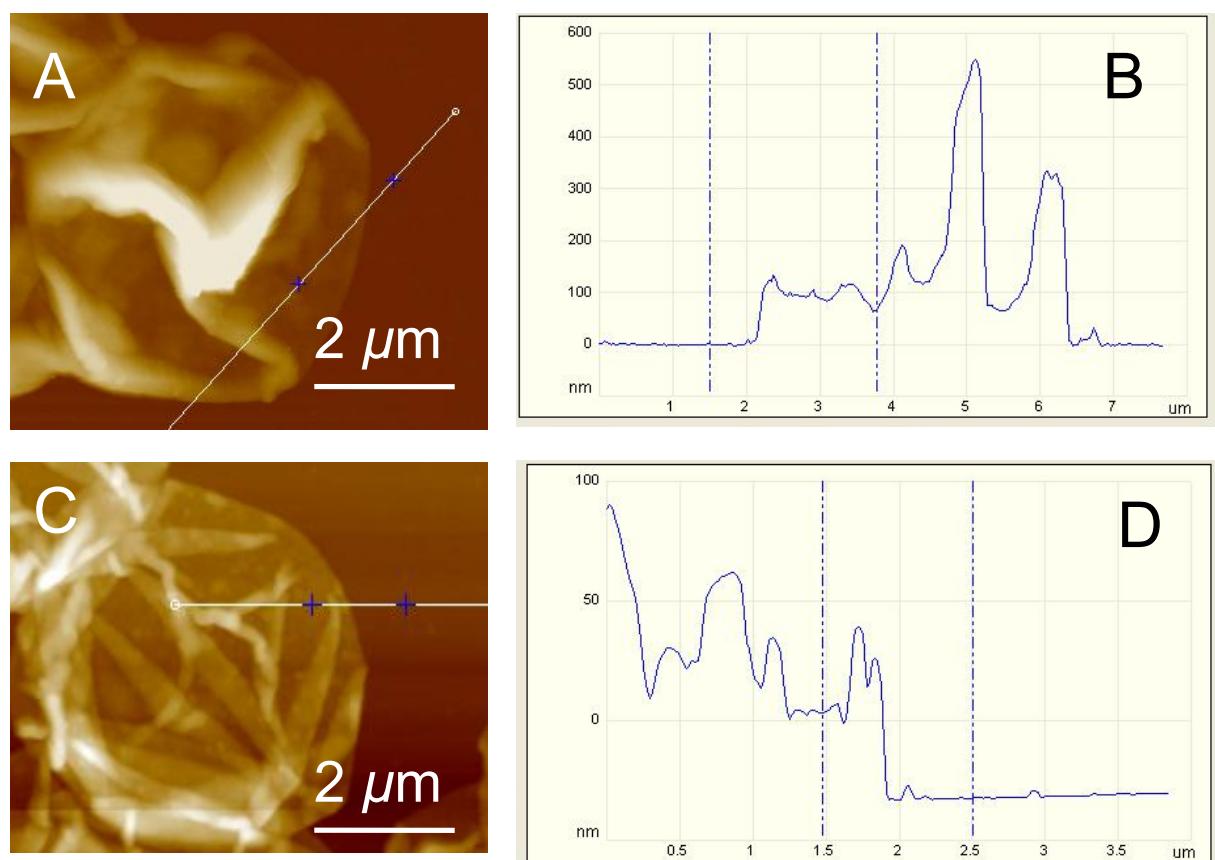


Figure S5. AFM cross section analysis of cubic (A, B) and spherical (C, D) ($\text{PVPO}_\text{N}/\text{PMAA}_\text{A}$)₄ microcapsules. Thickness of bilayer was estimated from flat areas in cross-sections.

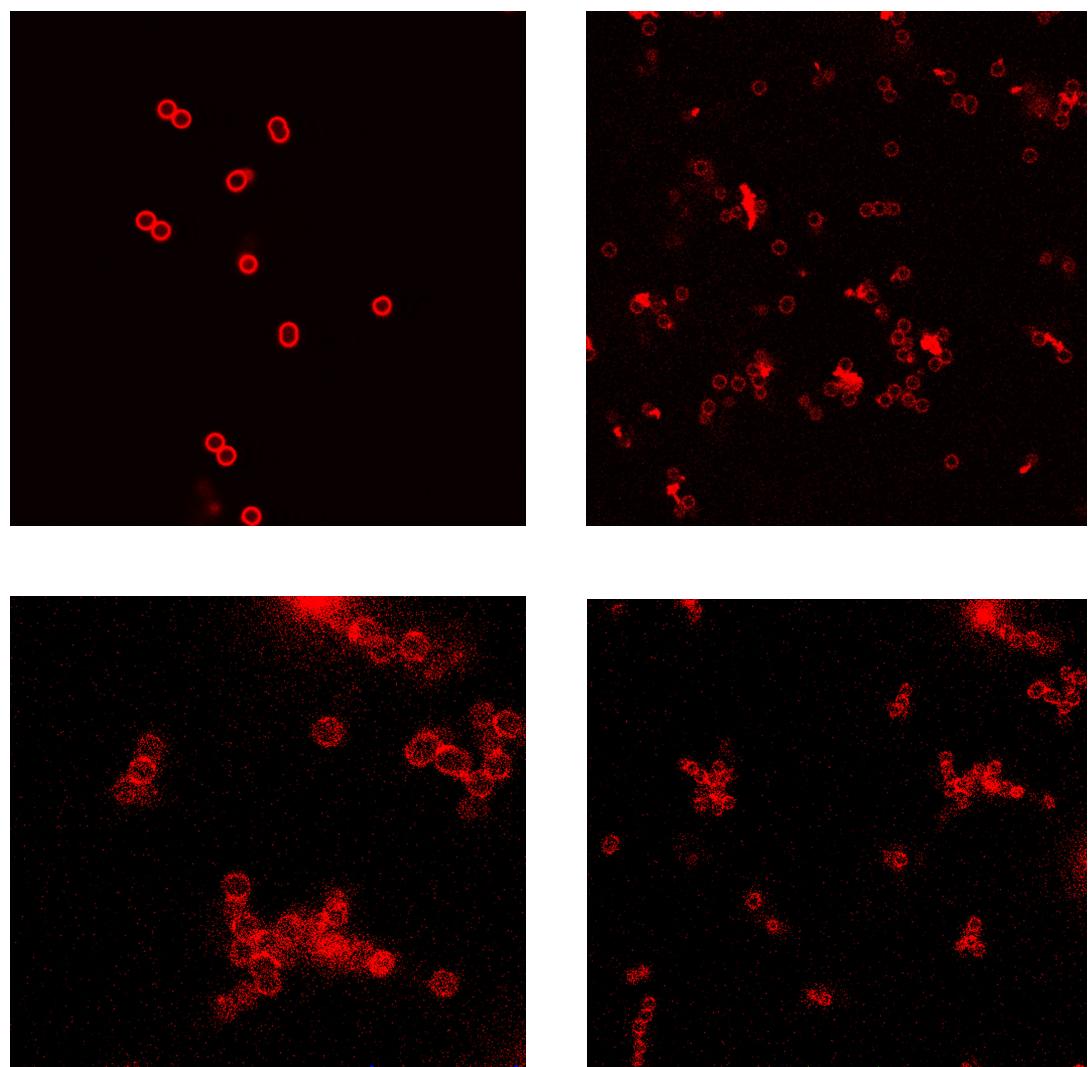


Figure S6. Confocal images of spherical $(\text{PVPON}/\text{PMAA})_4$ microcapsules assemblies in buffer pH 4.5 C=0.01 M NaCl (top) and C=1 M NaCl (bottom).

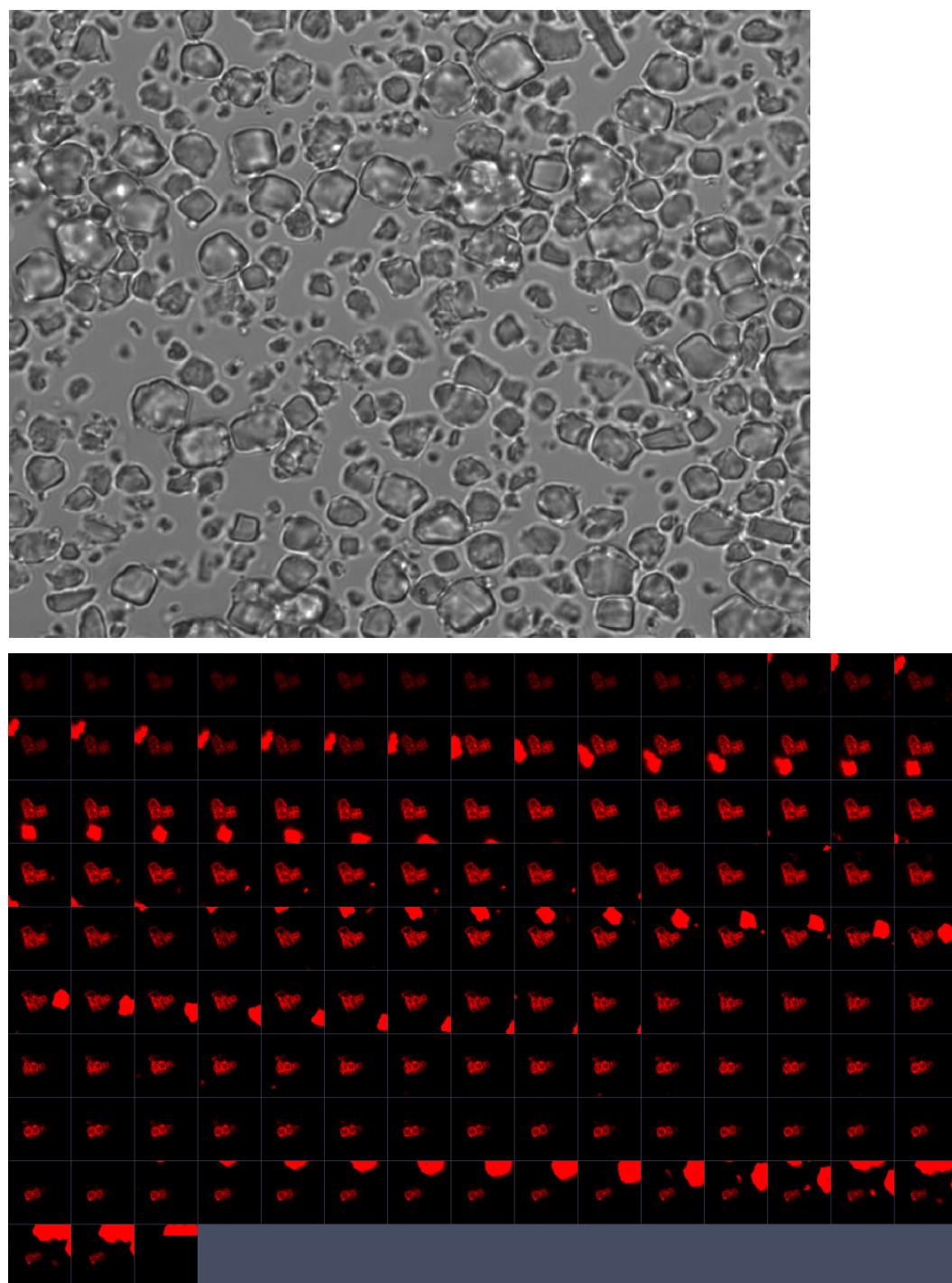


Figure S7. Confocal images of cubic $(\text{PVPON}/\text{PMAA})_4$ microcapsules assemblies in buffer pH 4.5 C=0.01 M NaCl (top) and C=1 M NaCl (bottom). Cross sections of three dimensional (3D) assemblies of cubic $(\text{PVPON}/\text{PMAA})_4$ microcapsules in buffer pH 4.5, C=1M at sequential time lapses (bottom).

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- ¹ Good, R. J. *Contact Angle, Wettability and Adhesion*; K.L. Mittal, ed.; V.S.P. Utrecht: The Netherlands, 1993; pp. 3-36.
- ² Lee, L. H. in *Preprints, Div. Organic Coatings and Plastics Chemistry, 154th National Meeting of ACS*, Chicago, IL, 1967, p. 61
- ³ Wu, S. *Org. Coat. Plast. Chem.*, 1971, **31**, 27.