

Supplementary Information

Aggregation morphologies of a series of heterogemini surfactants with a hydroxyl headgroup in aqueous solution

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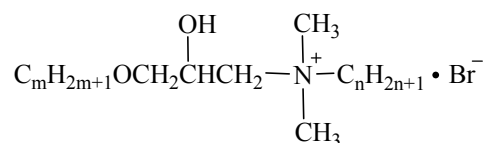
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Synthesis of C_mOhpNC_n

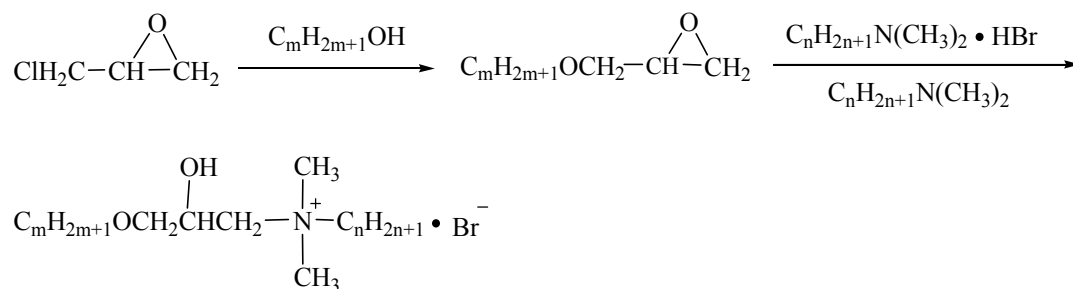
1. The molecular structure of C_mOhpNC_n is shown as follows,



$m, n = 10, 8; 10, 14; 12, 8; 12, 10; 12, 12; 12, 14; 14, 8$ and $14, 10$

Starting materials: Epichlorohydrin, *N, N*-dimethyldecylamine, *n*-decanol, *n*-dodecanol, *n*-tetradecanol, *n*-hexadecanol, hydrobromide were purchased from Sinopharm Chemical Reagent Co., Ltd (China), *N, N*-dimethyloctylamine, *N, N*-dimethyldodecylamine, *N, N*-dimethyltetradecylamine were purchased from Tokyo Chemical Industry Co., Ltd (Japan), All the reagents used were of analytical grade.

2. Synthetic routes



3. A detailed description about the synthesis

This compound was synthesized with the method reported by Tianhua Zhou¹, which is briefly described as follows: *N,N*-dimethylalkyl amine (0.01mol, 1.0 equiv) and *N,N*-dimethylalkyl amine hydrobromide (0.01mol, 1.0 equiv) were added into a mixed ethanol-water solution (80:20, 60ml). The mixture was heated to 50°C under stirring, and then alkyl glycidyl ether (0.01mmol, 1.0 equiv) prepared from epichlorohydrin^{1, 2} were added under vigorous agitation. After reacting for 4 hours, the mixture was cooled and the solvent was removed under reduced pressure. The product was recrystallized from ethyl acetate three times, followed by vacuum drying to give C_mOhpNC_n (*m, n* = 10, 8; 10, 14; 12, 8; 12, 10; 12, 12; 12, 14; 14, 8 and 14, 10) as white solids. The overall yield is from 29 to 79%.

4. ¹HNMR and elemental analysis of final products

C₁₀OhpNC₈ (400MHz, CDCl₃, TMS): δ 0.87-0.88 (*t*, 6H, 2CH₃-CH₂-), 1.26-1.34 (*m*, 24H, CH₃-(CH₂)₇-CH₂-CH₂-O- and CH₃-(CH₂)₅-CH₂-CH₂-N⁺(CH₃)₂-), 1.52-1.53 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.74 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.39-3.49 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.51-3.60 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.48 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Anal Calcd for C₂₃H₅₀BrNO₂: C, 61.04; H, 11.14; N, 3.10. Found: C, 60.66; H, 11.49; N, 2.72.

C₁₀OhpNC₁₄ (400MHz, CDCl₃, TMS): δ 0.81-0.83 (*t*, 6H, 2CH₃-CH₂-), 1.18 (*m*, 36H, CH₃-(CH₂)₇-CH₂-CH₂-O- and CH₃-(CH₂)₁₁-CH₂-CH₂-N⁺(CH₃)₂-), 1.44-1.47 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.67 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.31-3.36 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.48-3.52 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.50 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₂₉H₆₂BrNO₂: C, 64.90; H, 11.64; N, 2.61. Found: C, 63.93; H, 12.08; N, 2.09.

C₁₂OhpNC₈ (400MHz, CDCl₃, TMS): δ 0.88 (*t*, 6H, 2CH₃-CH₂-), 1.26 (*m*, 28H, CH₃-(CH₂)₉-CH₂-CH₂-O- and CH₃-(CH₂)₅-CH₂-CH₂-N⁺(CH₃)₂-), 1.52 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.74 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.39-3.42 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.57 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.49 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₂₅H₅₄BrNO₂: C, 62.48; H, 11.33; N, 2.91. Found: C, 62.17; H, 11.52; N, 2.54.

C₁₂OhpNC₁₀ (400MHz, CDCl₃, TMS): δ 0.88 (*t*, 6H, 2CH₃-CH₂-), 1.26 (*m*, 32H, CH₃-(CH₂)₉-CH₂-CH₂-O- and CH₃-(CH₂)₇-CH₂-CH₂-N⁺(CH₃)₂-), 1.52 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.74 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.39-3.43 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.56 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.49 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₂₇H₅₈BrNO₂: C, 63.75; H, 11.49; N, 2.75. Found: C, 63.01; H, 11.38; N, 2.47.

C₁₂OhpNC₁₂ (400MHz, CDCl₃, TMS): δ 0.81 (*t*, 6H, 2CH₃-CH₂-), 1.24 (*m*, 36H, CH₃-(CH₂)₉-CH₂-CH₂-O- and CH₃-(CH₂)₉-CH₂-CH₂-N⁺(CH₃)₂-), 1.45 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.67 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.32-3.37 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.51 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.41 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₂₉H₆₂BrNO₂: C, 64.90; H, 11.64; N, 2.61. Found: C, 63.86; H, 12.12; N, 2.29.

C₁₂OhpNC₁₄ (400MHz, CDCl₃, TMS): d 0.82 (*t*, 6H, 2CH₃-CH₂-), 1.24 (*m*, 40H, CH₃-(CH₂)₉-CH₂-CH₂-O- and CH₃-(CH₂)₁₁-CH₂-CH₂-N⁺(CH₃)₂-), 1.46 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.67 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.32-3.37 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.51(*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.42 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₃₁H₆₆BrNO₂: C, 65.93; H, 11.78; N, 2.48. Found: C, 63.30; H, 12.11; N, 2.32.

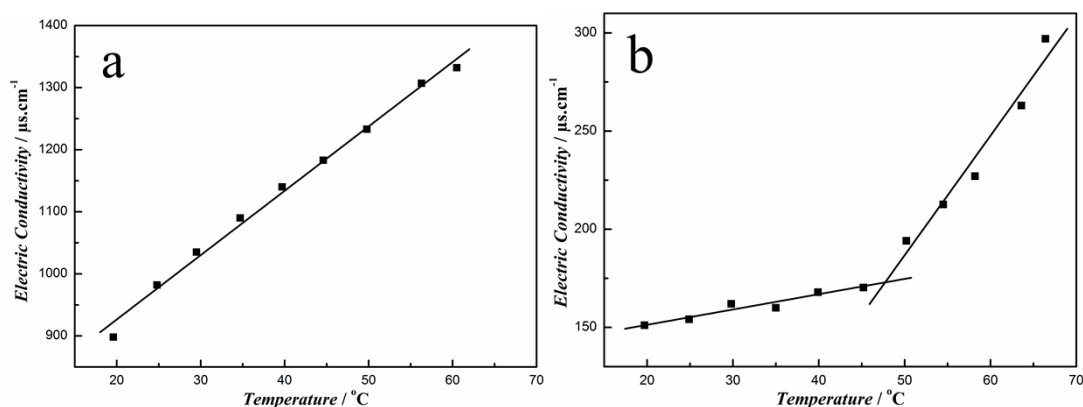
C₁₄OhpNC₈ (400MHz, CDCl₃, TMS):d 0.82 (*t*, 6H, 2CH₃-CH₂-), 1.19-1.28 (*m*, 32H, CH₃-(CH₂)₁₁-CH₂-CH₂-O- and CH₃-(CH₂)₅-CH₂-CH₂-N⁺(CH₃)₂-), 1.44-1.47 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.68 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.31-3.37 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.50-3.54 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.43 ppm (*m*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₂₇H₅₈BrNO₂: C, 63.75; H, 11.49; N, 2.75. Found: C, 63.07; H, 12.15; N, 2.59.

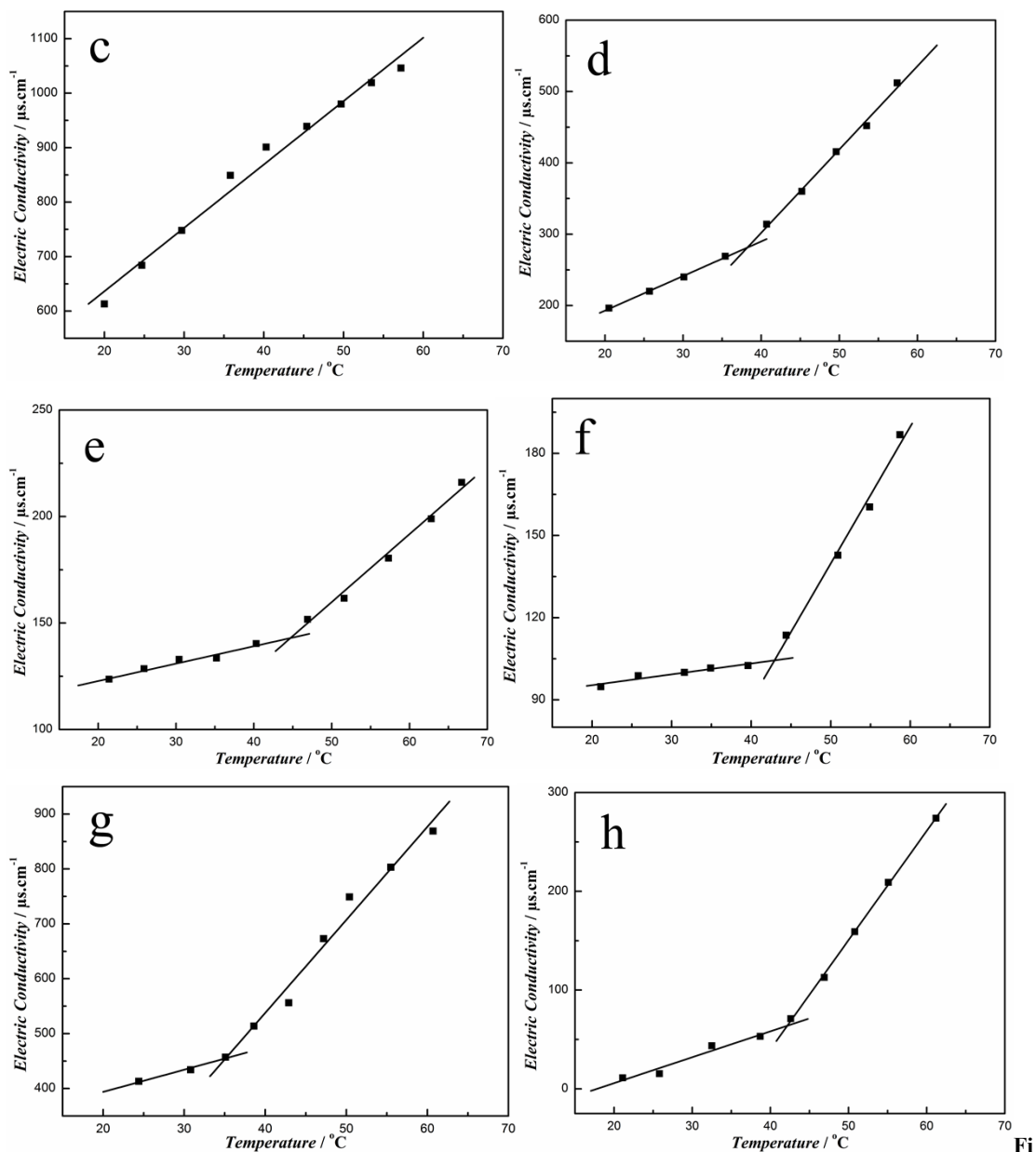
C₁₄OhpNC₁₀ (400MHz, CDCl₃, TMS): d 0.81 (*t*, 6H, 2CH₃-CH₂-), 1.19-1.28 (*m*, 36H, CH₃-(CH₂)₁₁-CH₂-CH₂-O- and CH₃-(CH₂)₇-CH₂-CH₂-N⁺(CH₃)₂-), 1.44-1.47 (*m*, 2H, CH₃-(CH₂)₇-CH₂-CH₂-O-CH₂-), 1.67 (*m*, 2H, -CH₂-CH₂-N⁺(CH₃)₂-CH₂-), 3.32-3.37 (*m*, 10H, -CH₂-N⁺(CH₃)₂-CH₂-, -CH₂-N⁺(CH₃)₂-CH₂-CH(OH)- and -CH₂-CH₂-O-CH₂-CH(OH)-), 3.50-3.53 (*m*, 4H, -CH₂-O-CH₂-CH(OH)- and -N⁺(CH₃)₂-CH₂-CH(OH)-), and 4.42 ppm (*s*, 1H, -O-CH₂-CH(OH)-CH₂-N⁺(CH₃)₂-); Calcd for C₂₉H₆₂BrNO₂: C, 64.90; H, 11.64; N, 2.61. Found: C, 63.56; H, 12.03; N, 2.28.

Measurements of C_mOhpNC_n

1 Krafft temperature

The electric conductivities of C_mOhpNC_n were measured by Model Delta 326 conductometer with the temperature ranging from 15 to 70 °C. The Krafft temperatures were determined by the break of conductivity vs. temperature plots.²





g. S1 The electric conductivities of different gemini surfactants as a function of temperature: (a) $\text{C}_{10}\text{OhpNC}_8$, (b) $\text{C}_{10}\text{OhpNC}_{14}$, (c) $\text{C}_{12}\text{OhpNC}_8$, (d) $\text{C}_{12}\text{OhpNC}_{10}$, (e) $\text{C}_{12}\text{OhpNC}_{12}$, (f) $\text{C}_{12}\text{OhpNC}_{14}$, (g) $\text{C}_{14}\text{OhpNC}_8$, (h) $\text{C}_{14}\text{OhpNC}_{10}$.

2 CMC

The electric conductivities of C_mOhpNC_n were measured by Model Delta 326 conductometer at 50°C . The CMC values were determined by the break of conductivity vs. concentration plots.

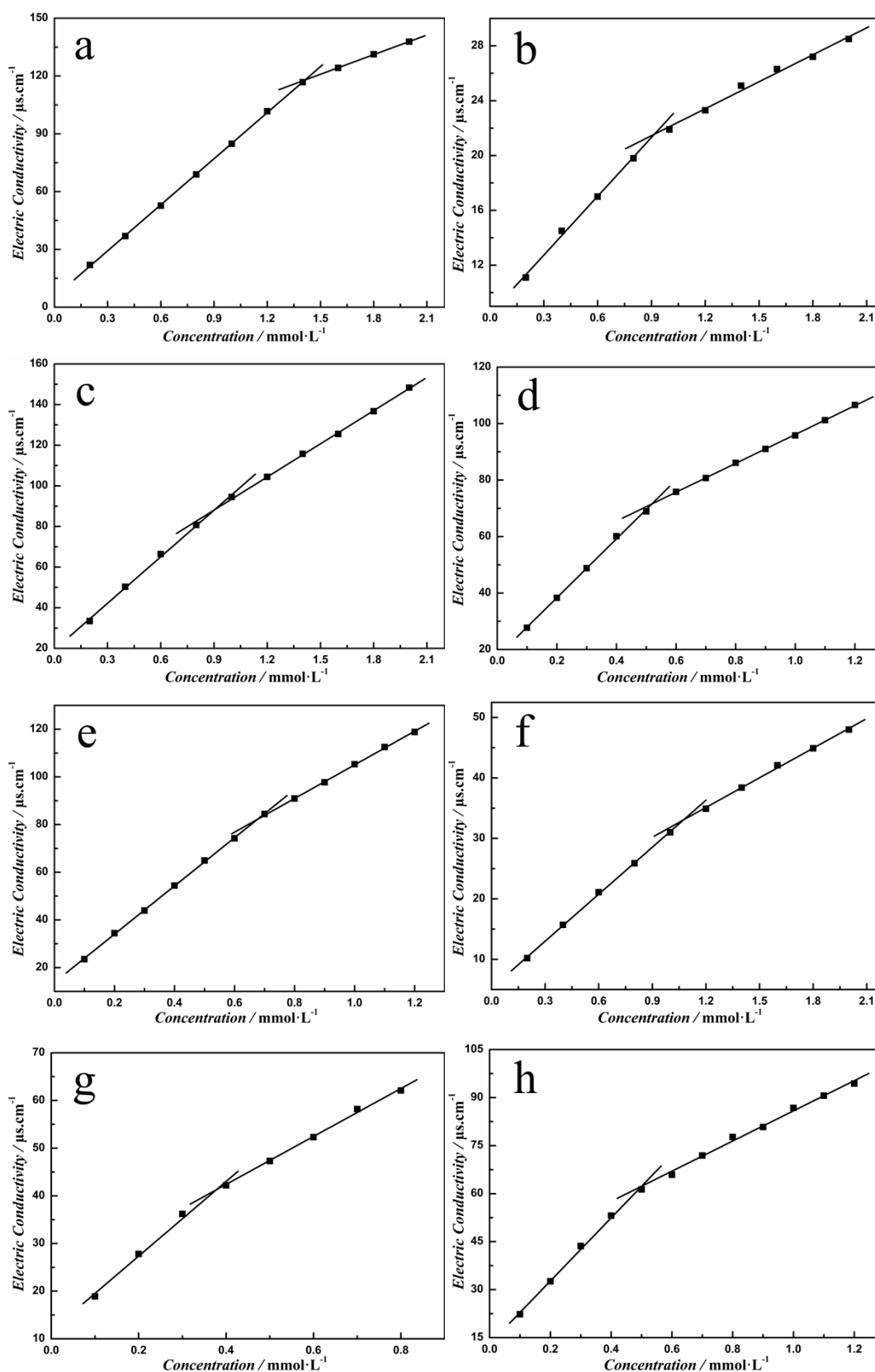


Fig. S2 The electric conductivities of different gemini surfactants as a function of concentration: (a) $\text{C}_{10}\text{OhpNC}_8$, (b) $\text{C}_{10}\text{OhpNC}_{14}$, (c) $\text{C}_{12}\text{OhpNC}_8$, (d) $\text{C}_{12}\text{OhpNC}_{10}$, (e) $\text{C}_{12}\text{OhpNC}_{12}$, (f) $\text{C}_{12}\text{OhpNC}_{14}$, (g) $\text{C}_{14}\text{OhpNC}_8$, (h) $\text{C}_{14}\text{OhpNC}_{10}$.

References

1 T. H. Zhou, J. X. Zhao, *J. Colloid Interface Sci.*, 2009, **331**, 476-483.

2 R. Zana, H. Lévy, D. Danino, Y. Talmon and K. Kwetkat, *Langmuir*, 1997, **13**, 402-408.