# ENTROPIC SELF-ASSEMBLY OF FREELY ROTATING POLYHEDRAL PARTICLES CONFINED TO A FLAT INTERFACE 

(SUPPLEMENTARY INFORMATION)

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## 1 Phase Behavior of TCs

As shown in Fig. S1, as number density increases the system goes from isotropic to square phase via an intermediate tetratic-like phase which is observed for $0.64<\eta^{*}<0.72$ and characterized by $\psi_{4}$ and $P_{4}$ approximately in the range $0.4-0.8$, and 0.3-0.45 respectively. Like in cubes, the intermediate phase forms diffuse $S(\mathbf{k})$ peaks and has liquid like long-range $g(r)$ behavior (see Fig. S2). Also, similar differences ensue for both cubes and TCs in comparing their confined self-assembly in 2D with the bulk self-assembly in $3 \mathrm{D}^{\mathrm{S1}}$.



Fig. S1 Equation of state for 1600 TCs obtained by expansion runs. Legend as in Fig. 1 of main text.


Fig. S2 (Top Panel) Representative snapshots and corresponding structure factors for a system of $N=3600$ TCs at (left) $P^{*}=8.1$ (tetratic-like phase) and (right) $P^{*}=$ 21.0 (square phase). (Bottom Panel) The radial distribution function for a system of $N=3600$ TCs at the same two pressures.

## 2 Two-Body potentials of mean force (PMFs)

The one-dimensional potential of mean force is defined as ${ }^{52}$

$$
\begin{equation*}
\beta P M F(r)=-\ln \left(\left\langle e^{-\beta U(r)}\right\rangle\right) \tag{1}
\end{equation*}
$$

where $\beta=1 /\left(k_{B} T\right)$ and $U(r)$ is the potential energy of interaction between two particles whose centers of mass are at a distance $r$. The procedure of calculating $\operatorname{PMF}(r)$ involves placing a particle at random position and orientation with center to center distance $r$ from another particle with random orientation. This step is followed by an overlap check using separating axes theorem ${ }^{\mathrm{S} 3}$. The above procedure is repeated for $10^{4}$ different positions for each value of $r$ and $10^{4}$ combinations of random orientations of the two particles for each position. The fraction of non-overlapping configurations at each value of $r$ corresponds to the average Boltzmann's factors in Eq. (1). Figure S3 shows obtained PMF( $r$ ) for different shapes where we set $r=1$ for the closest distance.

We also obtain the two-dimensional potential of mean force defined as

$$
\begin{equation*}
\beta P M F(x, y)=-\ln \left(\left\langle e^{-\beta U(x, y)}\right\rangle\right) \tag{2}
\end{equation*}
$$

where $U(x, y)$ is the potential energy of interaction between two particles whose centers of mass are at a distance of $x$ and $y$ from each other. These distances are defined relative to a coordinate frame whose origin is the center of mass of a particle with fixed position and orientation, and whose axes are perpendicular to the $\{100\}$ facets of the fixed particle. For convenience, the fixed particle is oriented in a way that one of the $\{100\}$ facets of a particle is parallel to the flat interface. The procedure of calculating this $P M F(x, y)$ involves placing a particle at a random orientation with position $(x, y)$ from the fixed particle followed by an overlap check. This step is repeated for $10^{4}$ different orientations for each position. Similar to $\operatorname{PMF}(r)$, the fraction of non-overlapping configurations at values of $(x, y)$ corresponds to the average Boltzmann's factors in Eq. (2). Figure S 4 shows the contour plots of $\operatorname{PMF}(x, y)$ where for each shape we make the $x$ and $y$ distances dimensionless by dividing their values by the closest distance [which was obtained from $\operatorname{PMF}(r)$ and corresponds to the particle indiameter $\left.\sigma_{i n}\right]$.


Fig. S3 One-dimensional potential of mean force, $P M F(r)$ for different shapes studied in this work.


Fig. S4 Two-dimensional potential of mean force, $P M F(x, y)$ for different shapes studied in this work. The colorbar varies from $\beta P M F(x, y)=0$ to 5 . The values of $\beta P M F(x, y)>5$ are labeled as red.

## 3 Adsorption Energy

We apply a model used in Refs. S4 and S5 to estimate the adsorption energy of a single particle on a fluid-fluid flat interface. The adsorption energy, $F$ of a truncated cube as a function of its position and orientation is given by

$$
\begin{equation*}
F(z, \delta, \omega)=\gamma_{t b}\left(A-S_{t b}\right)+\gamma_{1 t} S_{1 t}+\gamma_{2 t} S_{2 t}+\gamma_{1 b} S_{1 b}+\gamma_{2 b} S_{2 b}+\text { const. } \tag{3}
\end{equation*}
$$

Here, $\gamma_{t b}$ is the interfacial tension between the top and bottom fluid, $A$ is the total surface area of the interface and $S_{t b}$ is the interfacial area excluded due to the presence of a particle. The surface tension between particle's facets of type $i(i=1,2$ for $\{100\}$ and $\{111\}$ facets respectively) and top fluid is denoted by $\gamma_{i t}$ and the surface tension between particle facets of type $i$ and bottom fluid is denoted by $\gamma_{i b}$. The surface areas of facets of type $i$ in the bottom and top medium are denoted by $S_{i t}$ and $S_{i b}$ respectively. The value of the arbitrary additive constant is chosen such that $F=0$ when a particle is completely immersed in the bottom fluid. The position or immersion depth of a particle, $z$, is the height of its center of mass relative to an interface, which is parallel to the $x y$ plane with its normal pointing in the $z$-axis direction and positioned at $z=0$. The orientation of a truncated cube is given by two angles: the polar angle $\delta$ and the azimuthal angle $\omega$. As shown in Ref. S4, using the Young's equation, Eq. 3 is simplified to

$$
\begin{equation*}
F(z, \delta, \beta)=\gamma_{t b}\left(\cos \theta_{1} S_{1 t}+\cos \theta_{2} S_{2 t}-S_{12}\right) \tag{4}
\end{equation*}
$$

where $\theta_{i}$ is the contact angle corresponding to the three-phase contact between facet of type $i$, top fluid and bottom fluid. To obtain $F$ for (experimentally relevant) contact angle values of $\theta_{1}$ and $\theta_{2}$ for any given $(z, \delta, \omega)$, the surface areas $S_{i t}, S_{i b}$ and $S_{t b}$ are estimated using the triangular tessellation technique described in Ref. S6. The three dimensional landscape of adsorption/interfacial energy is then obtained by repeating the procedure for all possible orientations and positions, where we vary, $z$ within $(-H, H)$
where $H$ is a value slightly larger than the circumradius of a shape (for TC4 we use $H$ as the largest circumscribing radius), $\delta$ from $[0, \pi / 2]$, and $\omega$ from $[0, \pi / 4]$ using 100 equidistant steps for each of these three parameters. Due to the symmetry properties of a truncated cube, the polar angle and azimuthal angle are varied up to $\pi / 2$ and $\pi / 4$ respectively. The most stable configuration is then obtained by estimating the global minimum of this landscape and the value of $F$ corresponding to the global minimum is denoted as $F_{\text {minglob }}$. The model also provides the interfacial energy, $F_{\text {min }}(z)$ of a particle as a function of $z$ minimized with respect to angles $\delta$ and $\omega$. The estimation of $F_{\min }(z)$ at different values of $z$ is performed for a cube shaped particle of size 3 nm adsorbing on a toluene-air interface at room temperature. The value of interfacial tension, $\gamma_{t b}$ is given by $28.52 \mathrm{mN} / \mathrm{m}$ (room temperature). The profile of $F_{\min }(z)$ versus $z$ is shown for different contact angles ranging from $\cos \theta_{1}=0.0$ to $\cos \theta_{1}=0.5$ in Fig. S5.


Fig. 55 The profile of $F_{\text {min }}(z)$ versus $z$ for a cube shaped particle for different contact angles ranging from $\cos \theta_{1}=0.0$ to $\cos \theta_{1}=0.5$ (as per numbers in the inside legend).

For $\cos \theta_{1}=0.0$ and $\cos \theta_{2}=0.0$, we show the difference $\Delta F_{\text {orient }}(\delta, \omega)=F\left(z_{\text {opt }}, \delta, \omega\right)-F_{\text {minglob }}$ as a function $\delta$ and $\omega$ for different shapes in Fig. S6, where the value, $z_{\text {opt }}$ is the immersion depth of the most stable configuration. For the plots shown in Fig. S6, the cube size is 3 nm whereas the size of the other shapes is selected to have the same surface area as that of the 3 nm cube. For each shape, we also show in Fig. 6 snapshots of three different configurations whose $\Delta F_{\text {orient }}<=5 k_{B} T$. We also obtain the percentage of angular phase space, $P_{\text {orient }}$, that has $\Delta F_{\text {orient }}<=5 k_{B} T$; this is estimated by evaluating $\Delta F_{\text {orient }}$ over a grid of $(\delta, \omega)$ pairs and finding the fraction for which $\Delta F_{\text {orient }}<=5 k_{B} T$.


Fig. S6 The energy difference, $\Delta F_{\text {orient }}$ as a function of $\delta$ and $\omega$ for (a) cube, (b) TC, (c) TC4, (d) CO, (e) TO and (f) Oct at contact angle values of $\theta_{1}=90$ degrees and $\theta_{2}=90$ degrees. The snapshots of configurations at three different $(\delta, \omega)$ values with $\Delta F_{\text {orient }}<=5 k_{B} T$ are also shown for each shape (the interface plane, perpendicular to the plane of the page, appears as a horizontal line, and for the particles the $\{100\}$ facets are colored red and the $\{111\}$ facets are green).

## References:

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