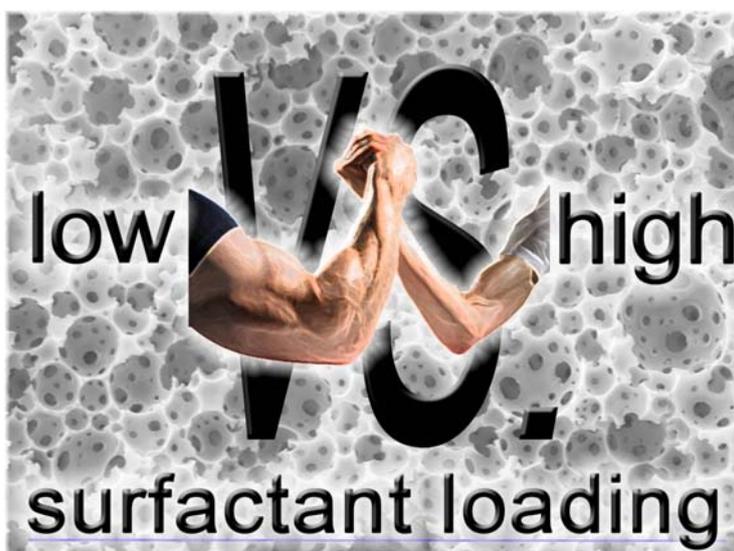


On the Mechanical Properties of HIPE Templated Macroporous poly(Dicyclopentadiene) Prepared with Low Surfactant Amounts

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Electronic Supplementary Information



1. Experimental

Dicyclopentadien (*Sigma-Aldrich*, amounts according to Table S1) and surfactants (Pluronic®L-121 (Poly(ethylene glycol)-*block*-poly(propylene glycol)-*block*-poly(ethylene glycol); MW = 4400 g•mol⁻¹); Brij®52 (Polyethylene glycol hexadecyl ether; MW= 330 g•mol⁻¹); Span™80 (Sorbitan monooleate; MW= 428 g•mol⁻¹) all *Sigma-Aldrich*) were placed in a three necked 250 mL flask and the mixture was stirred with an

overhead stirrer at 400 rpm. The corresponding amount (*cf.* Table S1) of deionised water was added drop wise under constant stirring. After addition of water the mixture was further stirred for 1 h until a uniform emulsion was produced. 250 μ L solution of the initiator (H₂IMes)(PCy₃)Cl₂Ru(3-phenyl-indenylid-1-ene) (**M2**, H₂Imes = N,N-bis(mesityl) 4,5-dihydroimidazol-2-yl) (Umicore; *cf.* Table S1) in toluene was added to the emulsion and the mixture was stirred for further 1 min. Subsequently, the emulsion was transferred to the mould (polystyrene container or steel moulds or glass vial) and was cured at 80 °C for 4 h. Resulting polymers were purified via Soxhlet extraction with acetone and dried under vacuum until constant weight was obtained

Table S1. Emulsion composition (80v% aqueous phase)

sample	m (DCPD) [g]	m (M2) [mg]	Surfactant	Surf. [g]	Surf. [vol. %]	V(H ₂ O) [mL]
⁸⁰DCPD_{0.25}	8.06	8.08	Pluronic®121	0.021	0.25	33
⁸⁰DCPD₁	8.08	8.08	Pluronic®121	0.092	1	33
⁸⁰DCPD₃	8.06	8.04	Pluronic®121	0.26	3	33
⁸⁰DCPD₅	8.03	8.11	Pluronic®121	0.44	5	33
⁸⁰DCPD₆	8.02	8.12	Pluronic®121	0.52	6	33
⁸⁰DCPD₇	8.02	8.13	Pluronic®121	0.62	7	33
⁸⁰DCPD₈	8.08	8.08	Pluronic®121	0.73	8	33
⁸⁰DCPD₉	8.01	8.04	Pluronic®121	0.82	9	33
⁸⁰DCPD₁₀	8.06	8.08	Pluronic®121	0.91	10	33
⁸⁰DCDP_{Span80}	8.04	8.10	Span™80	0.084	1	33
⁸⁰DCDP_{Span80}	8.03	8.25	Span™80	0.25	3	33
⁸⁰DCDP_{Span80}	8.01	8.31	Span™80	0.43	5	33
⁸⁰DCDP_{Brij52}	8.06	8.08	Brij® 52	0.42	5	33

Elemental Analyses

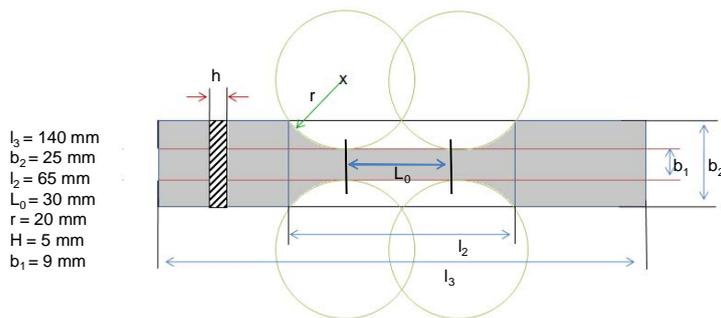
In the case of non-oxidized sample calculations from elemental analysis reveal oxygen content of the samples right after the preparation and purification (sample stored under vacuum). In each case three samples from a bigger specimen were submitted for elemental analysis and a mean value is given in the table. In the case of oxidized samples calculations from elemental analysis reveal an oxygen content (calculated according to: $O[\%] = 100 - C[\%] - H[\%]$) of the samples after four weeks of air exposure (Table S2).

Table S2. Elemental analysis data

sample	Oxidized			Un-oxidized		
	Elemental analysis			Elemental analysis		
	C[%]	H[%]	O[%]	C[%]	H[%]	O[%]
${}_{80}\text{DCDP}_{0.25}$	65	7	28	90	9	1
${}_{80}\text{DCDP}_1$	62	6.5	31.5	89.5	9	1.5
${}_{80}\text{DCDP}_3$	63	6.5	30.5	89	9	2
${}_{80}\text{DCDP}_5$	63	6	31	88.5	8.5	3
${}_{80}\text{DCDP}_6$	63	6	31	89	9	2
${}_{80}\text{DCDP}_7$	63	6	31	89.5	9	1.5
${}_{80}\text{DCDP}_8$	63	6.5	30.5	89.5	9	1.5
${}_{80}\text{DCDP}_9$	62	6.5	31.5	89	9	2
${}_{80}\text{DCDP}_{10}$	63	6.5	30.5	89.5	9	1.5

2. Mechanical properties

2.1 Mechanical testing: High internal phase emulsions were prepared as written in Experimental and HIPEs were transferred into the special stainless steel templates with following dimensions:



Samples were tested at a strain rate of 1 mm/min at ambient temperature. The Zug/Druck-Universalprüfmachine (Typ Z010, Fa. ZWICK) was equipped with a force measuring range up to 10kN. The experimental Young's modulus was determined from the initial linear slope of the stress/strain plot. The theoretical Young's modulus was calculated using Gibson's equation and was compared with actual Young's moduli determine from the stress/strain plots.

Gibson's equation,¹

$$E_f = \left(\frac{d_f}{d_p}\right)^2 E_p$$

where E_p , E_f , d_p and d_f stands for polymer modulus, foam modulus, polymer density and foam density, was used to calculate the theoretical E-modulus of DCPD polyHIPEs. Polymer modulus and polymer density were taken from a database (2 GPa and 1.04 g cm^{-3}).² In our case, calculated values were compared with experimental values assessed from stress strain plots and % of theoretical yield strength was determined (Table S3).

¹ Gibson, L. J., Ashby, M. F. *Cellular solids structure and properties*, Cambridge University Press, Cambridge, 2nd edn, **1999**.

²<http://www.matweb.com/search/DataSheet.aspx?MatGUID=16d3d6b1e32c4c368fa1dda c6afb2b93&ckck=1>

Table S3. Characterisation of pDCPD foams prepared with different surfactant amount

^a

<i>sample</i>	⁸⁰ DCPD _{0.25}	⁸⁰ DCPD ₁	⁸⁰ DCPD ₃	⁸⁰ DCPD ₅	⁸⁰ DCPD ₆	⁸⁰ DCPD ₇	⁸⁰ DCPD ₈	⁸⁰ DCPD ₉	⁸⁰ DCPD ₁₀
skeletal density [g·cm ⁻³] ^a	1.325	1.325	1.350	1.342	1.320	1.350	1.180	1.250	1.204
bulk density [g·cm ⁻³] ^a	0.261	0.256	0.274	0.250	0.240	0.250	0.240	0.230	0.251
not oxidized									
R _{p0.2} [MPa] ^b	1.5±0.3	1.3±0.2	1.8±0.2	1.5±0.3	1.0±0.2	1.5±0.2	0.2±0.07	0.2±0.01	0.1±0.03
ε @ break [%] ^b	11±2	17±4	22±1	34±2	23±17	43±9	35±6	27±6	25±7
E-module [MPa] ^b	94±5	94±10	105±8	97±4	94±2	88±2	18±2	16±1	12±3
E _{foam} [MPa] ^c	125	121	138	115	106	115	106	98	116
E _{exp.} /E _{teo.} [%] ^d	75	78	76	84	88	76	17	16	10
oxidized									
E-module [MPa] ^e	228±14	267±15	273±10	243±2	230±18	242±2	69±1	61±10	78±25
ε @ break [%] ^e	0.76±0.06	0.62±0.12	1.20±0.2	1.21±0.08	1.2±0.3	1.05±0.2	0.7±0.1	0.62±0.1	1.2±0.2

determined by helium pycnometry of oxidized samples; ^b assessed from stress strain tests at constant speed rate and ambient temperature of not oxidized samples; ^c theoretical modulus estimated from Gibson's equation; ^d % of theoretical strength; ^e determined for oxidized samples

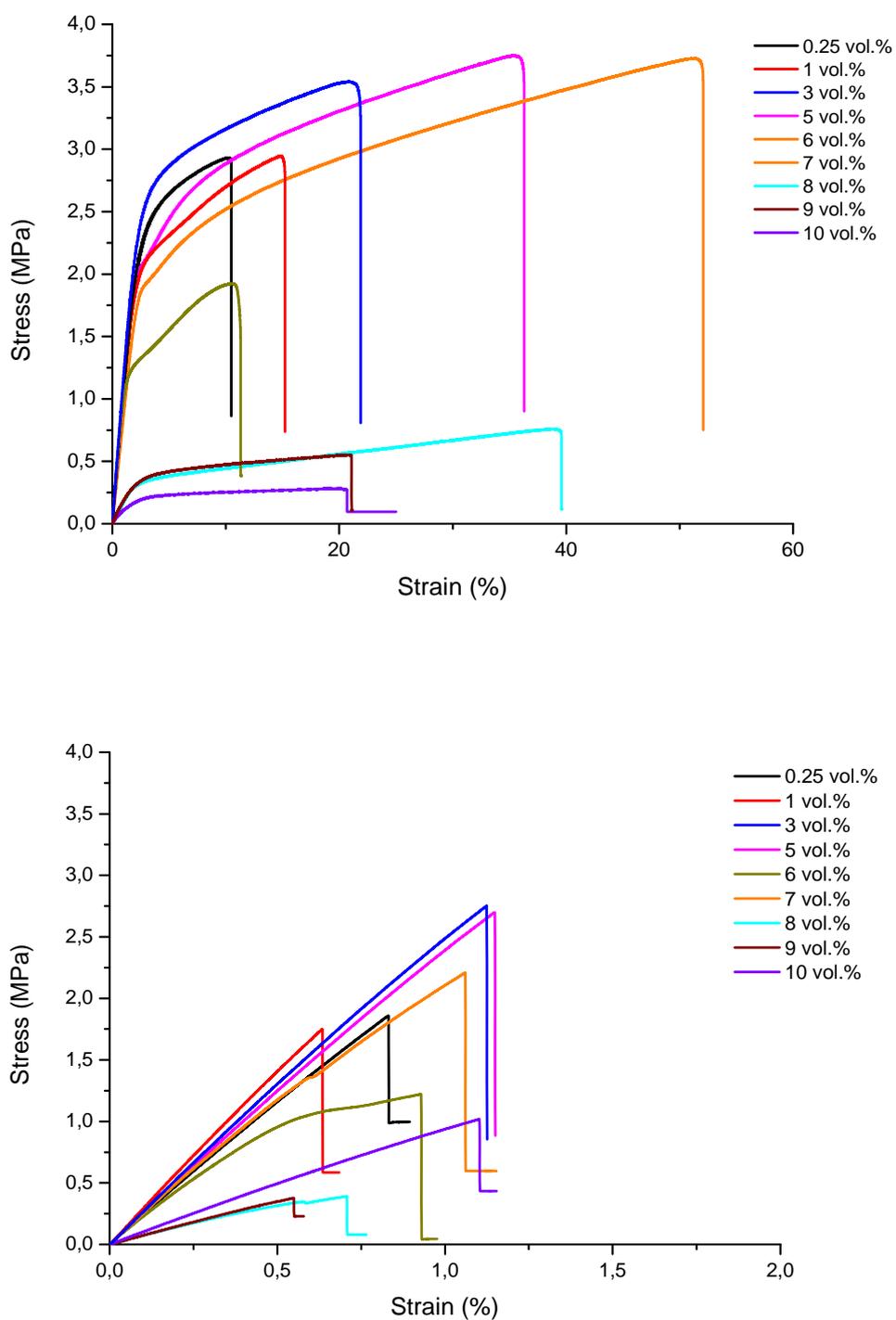


Figure S1. Strain-stress curves for not oxidized (above) and oxidized (below) **pDCPD₈₀** at different surfactant loadings

2.2 Mechanical strength vs. surfactant system

High internal phase emulsions with 5 vol. % of the surfactant used were prepared as described above. Changes made was chemistries of the surfactant used, which were in this particular case SpanTM80 and Brij[®] 52 in spite of Pluronic[®]121. Afterwards, HIPEs were transferred into the special stainless steel templates and samples for mechanical testing were prepared as described above.

Table S5. Characterisation of ${}_{80}\text{DCPD}_{\text{Span80}}$ and ${}_{80}\text{DCPD}_{\text{Brij52}}$ foams

<i>sample</i>	${}_{80}\text{DCPD}_{\text{Span80}}$	${}_{80}\text{DCPD}_{\text{Brij52}}$
$R_{p0.2}$ [MPa]	0.7 ± 0.1	1 ± 0.2
ϵ @ break [%]	13 ± 0.2	9.9 ± 0.4
E-module [MPa]	75 ± 3	50 ± 2

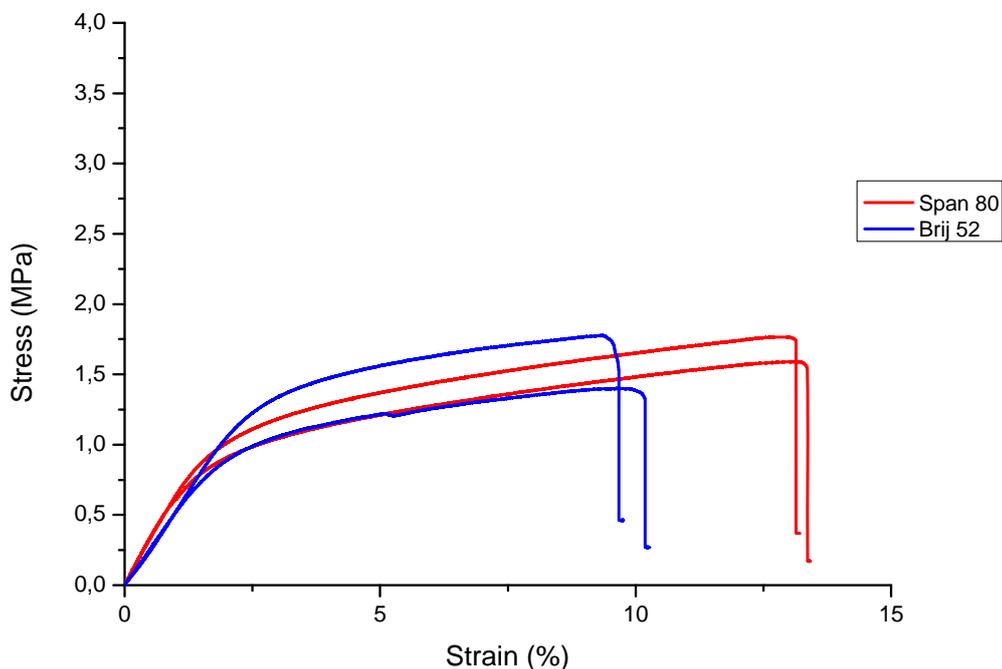


Figure S2. Strain-stress curves for un-oxidized $p_{80}\text{DCPD}_3$ with 5v% of SpanTM80 and Brij[®]52 in the HIPE system

2.3 Mechanical strength vs. porosity

Table S6. Characterisation of **pDCPD** foams prepared at 3 vol. % of the surfactant (Pluronic L121) used with porosities of 50, 60 and 70 (marked with subscripts on the left side of the abbreviation)

<i>sample</i>	p₅₀DCPD	p₆₀DCPD	p₇₀DCPD
R _{p0.2} [MPa]	5.6±0.9	3.6±0.2	2.45±0.2
ε @ break [%]	24±10	23±11	20±6
E-module [MPa]	405±15	269±9	175±6

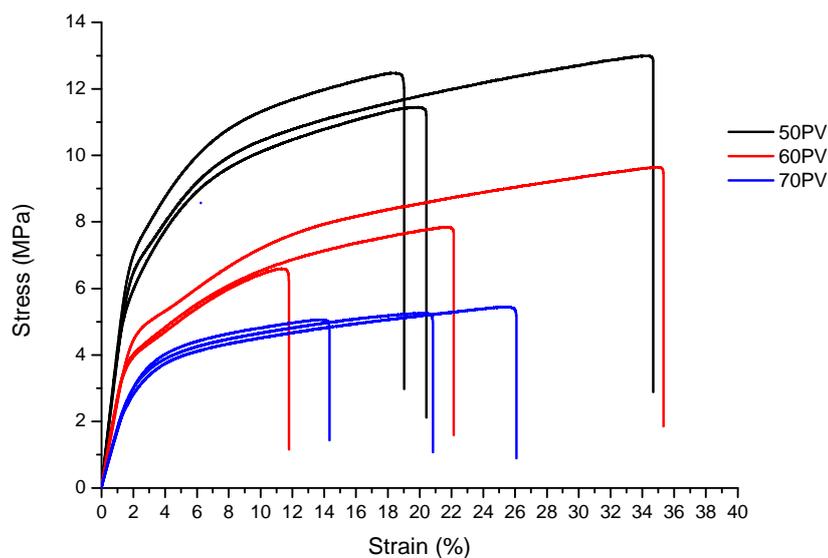
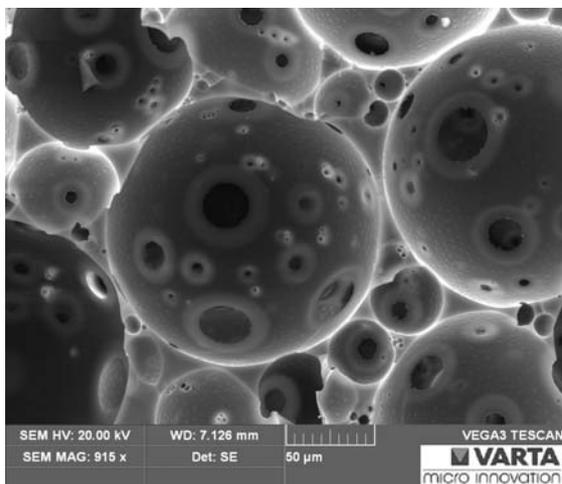


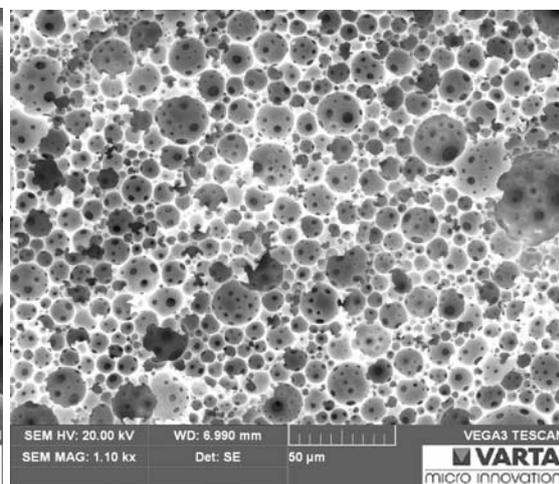
Figure S3. Strain-stress curves for un-oxidized **pDCPD₃** with different porosities at 3v% of Pluronic L121 used.

3 Scanning Electron Microscopy (SEM) investigation

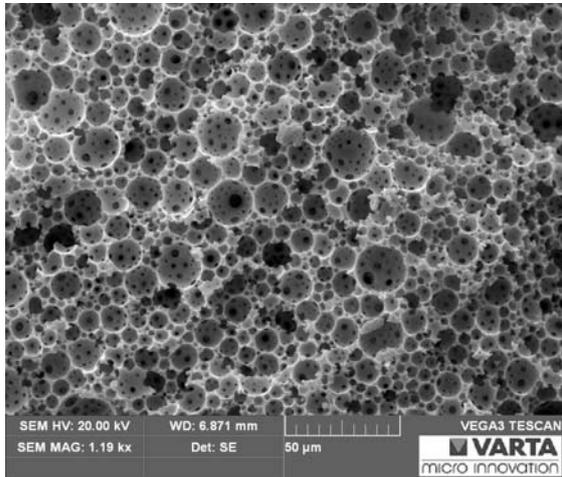
3.1 SEM images of p_{80} DCPD with different surfactant concentrations



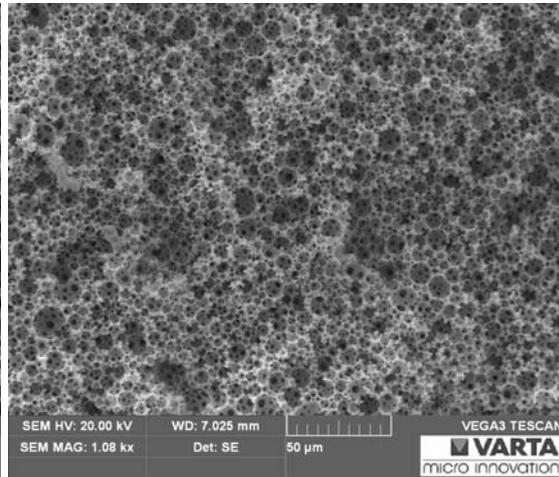
$80\text{DCDP}_{0.25}$



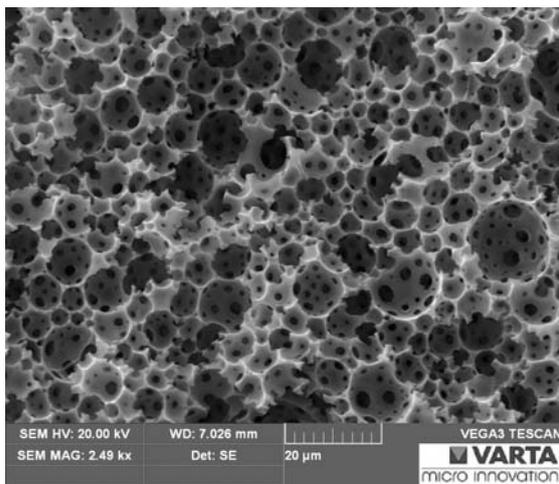
80DCDP_1



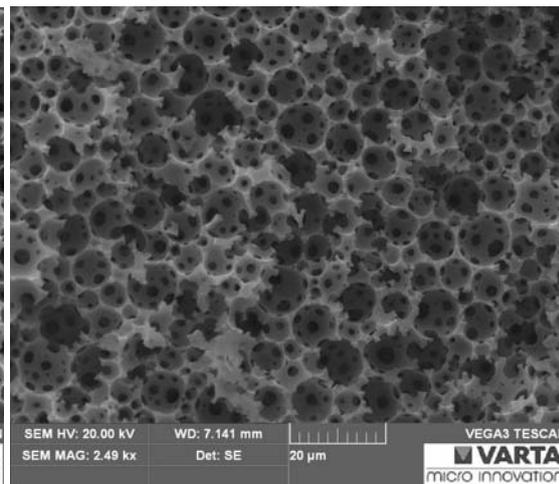
80DCDP_3



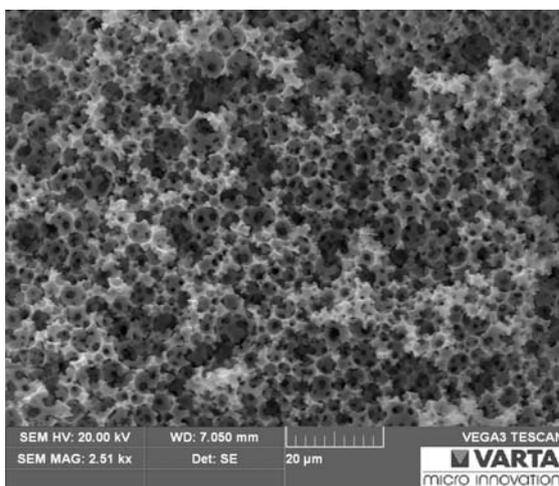
80DCDP_5



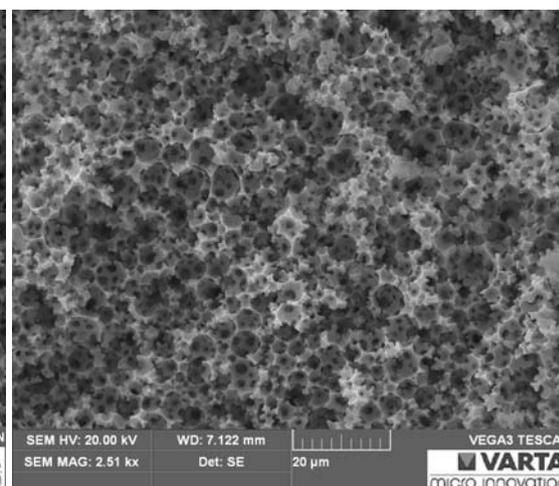
80DCDP₆



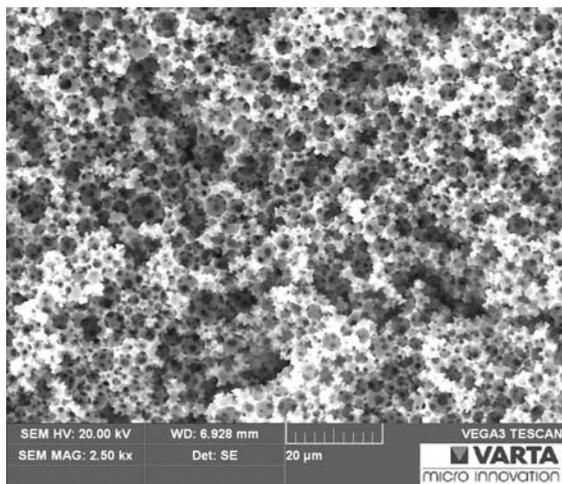
80DCDP₇



80DCDP₈



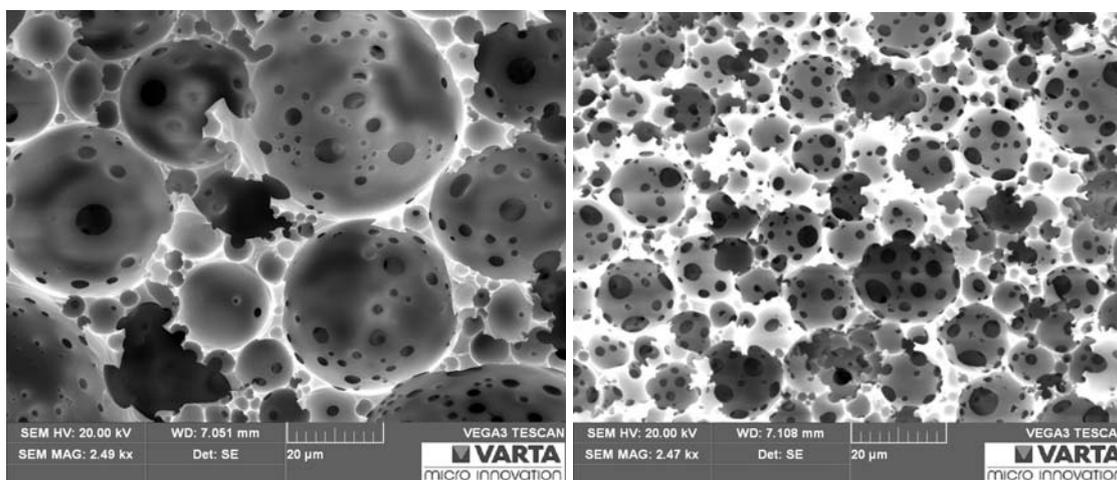
80DCDP₉



$p_{80}DCDP_{10}$

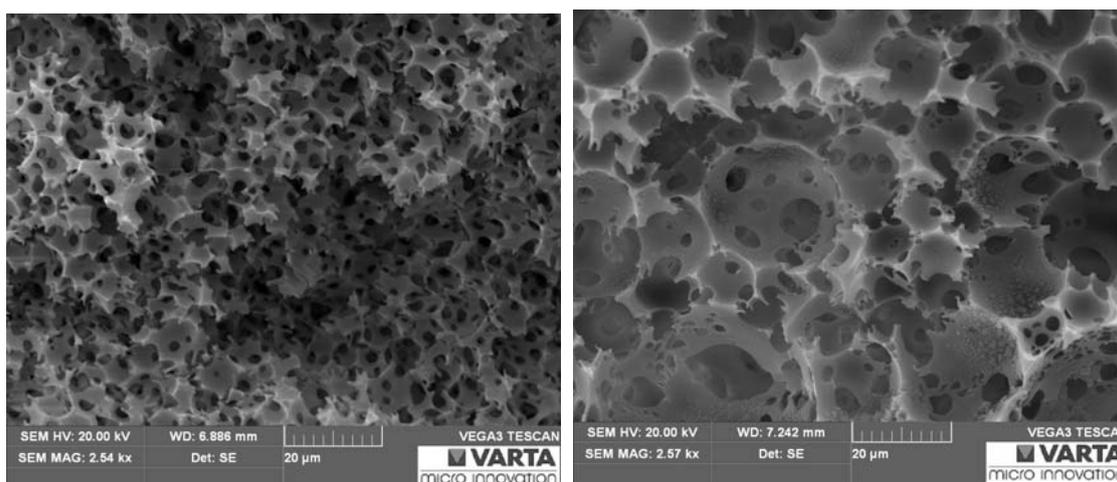
Figure S4. SEM images of **$p_{80}DCDP$** at different surfactant (Pluronic L121) concentrations (marked with subscripts 0.25, 1, 3, 5, 6, 7, 8, 9, and 10 on the right side of the abbreviation)

3.2 SEM images of p_{80} DCPD prepared with Brij®52 or Span™ 80



80DCDP_{Span80_1}

80DCDP_{Span80_3}

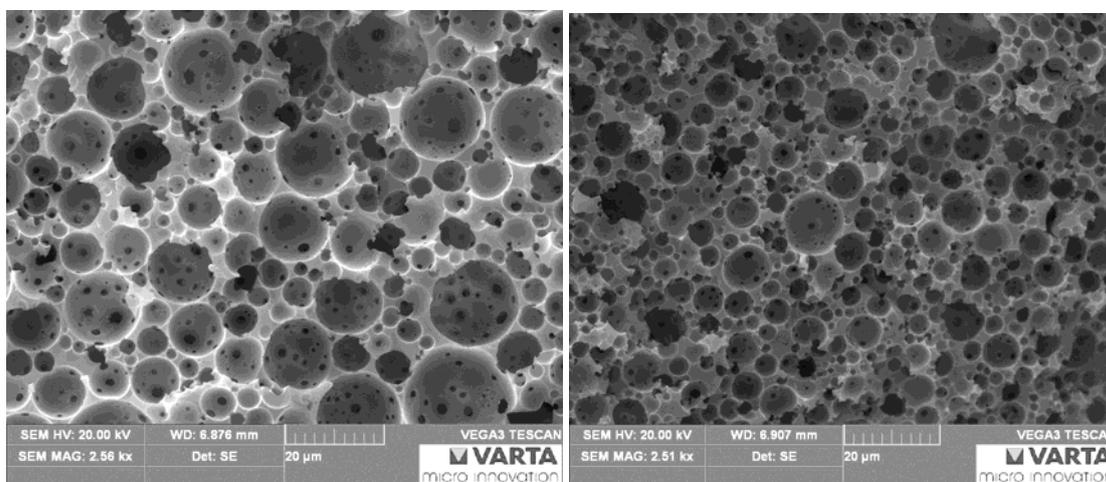


80DCDP_{Span80_5}

80DCDP_{Brij52_5}

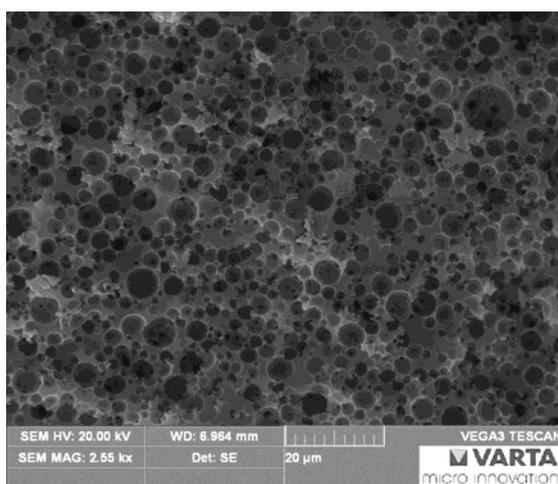
Figure S5. SEM images of p_{80} DCPD with Span 80 in the HIPE system at different surfactant concentration (marked with subscripts 1, 3 and 5 on the right side of the abbreviation) and SEM image of p_{80} DCPD with 5v% of Brij®52 used in the HIPE system

3.3 SEM images of pDCPD prepared with different porosities



70DCDP₃

60DCDP₃

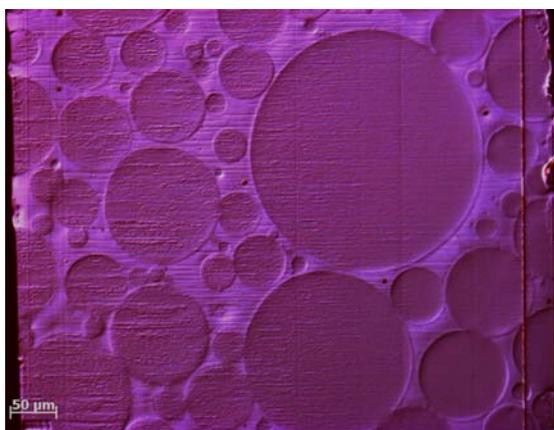


50DCDP₃

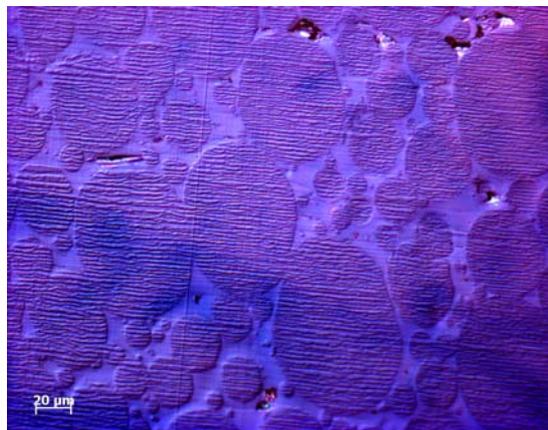
Figure S6. SEM images of pDCPD with different porosities (marked with subscripts as 70, 60 or 50 on the left side of the abbreviation) at 3 v% of Pluronic L121 used

4. Light Microscopy (LM) investigation

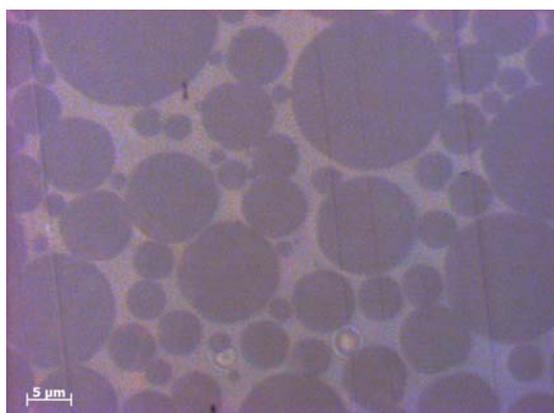
An average voids sizes were determined from LM micrographs analysis. Therefore, the mean and the standard deviation were drawn by manual measurements of diameters from a population of at least 40 voids. From SEM image analysis, it is difficult to give a correct evaluation of the void size because the voids are inside the material and during sample breaking the voids which appear are at random distance from the centre. To get a better estimation of the real void diameter by using the statistical factor $2/3^{1/2}$, void size evaluation was performed from epoxy-filled samples which were subsequently cut.



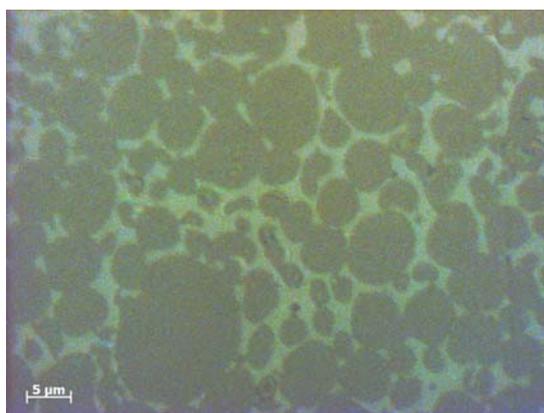
80DCDP_{0.25}



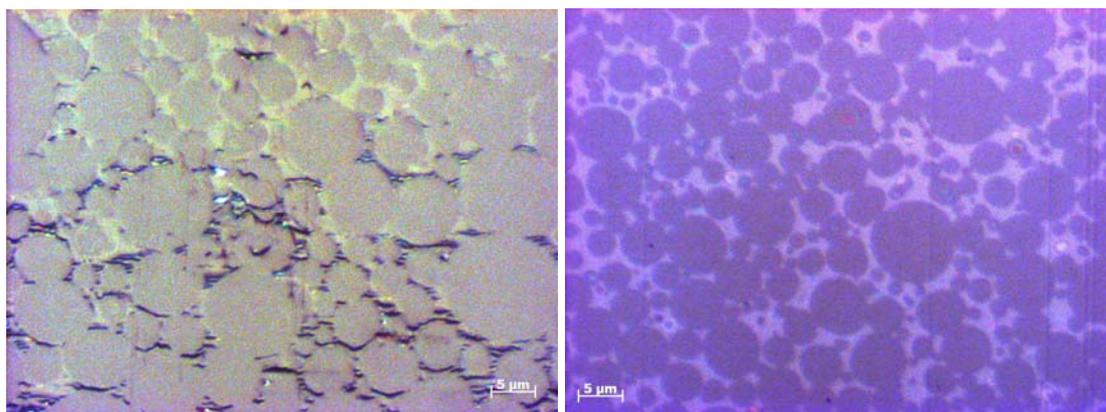
80DCDP₁



80DCDP₃

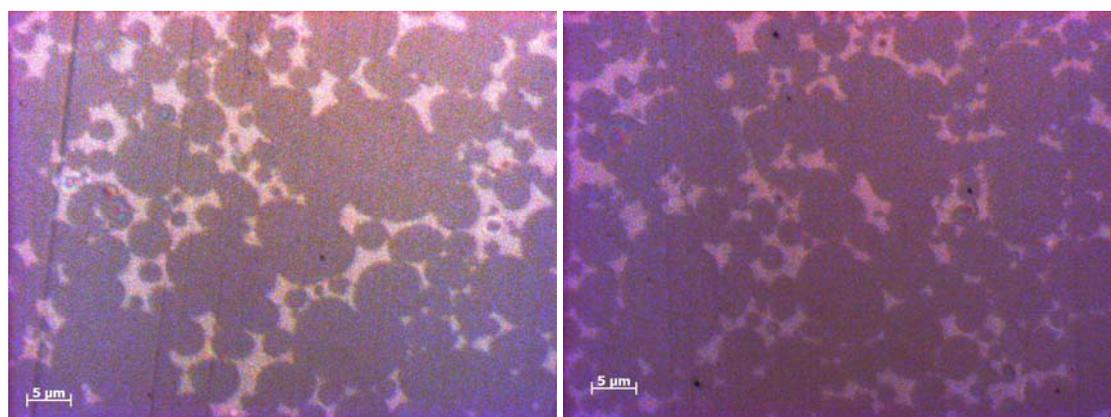


80DCDP₅



p80DCDP₆

p80DCDP₇



p80DCDP₈

p80DCDP₉



p80DCDP₁₀

Figure S7. LM images of **p₈₀DCDP** at different surfactant (Pluronic L121) concentrations (marked with subscripts 0.25, 1, 3, 5, 6, 7, 8, 9, and 10 on the right side of the abbreviation)