

Supplementary Information

Economical Hafnium Oxygen Nitride Binary/Ternary Nanocomposite Counter Electrode Catalysts for High-Efficiency Dye-Sensitized Solar Cells

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X-ray diffraction peak assignments of the as-synthesized nanocomposites

X-ray diffraction peak assignments of the as-synthesized Hf-R0-U, Hf-R5-B and Hf-R12-T samples were as follows: for HfO₂ (JCPDS:65-1142, PDF-2 Database), the diffraction peaks at 17.52°, 24.18°, 24.58°, 28.34°, 31.64°, 34.30°, 34.66°, 35.52°, 36.06°, 38.84°, 40.98°, 45.10°, 45.80°, 49.60°, 50.48°, 50.88°, 51.56°, 54.42°, 56.32°, 57.56°, 58.28°, 60.30°, 61.72°, 62.26°, 63.24°, 64.68°, 66.12°, 69.42°, 71.68°, 73.22°, 75.76°, 76.96° and 78.06° are assigned to the crystal planes (100), (011), (110), ($\bar{1}11$),

(111), (002), (020), (200), ($\bar{1}02$), (021), ($\bar{2}11$), (112), ($\bar{2}02$), (022), (220), ($\bar{1}22$), ($\bar{2}21$), (300), (130), (310), ($\bar{1}31$), (131), (113), ($\bar{2}13$), (311), ($\bar{1}23$), (222), (132), ($\bar{3}22$), (040), (041), ($\bar{4}11$), and ($\bar{1}33$), respectively. The diffraction peaks at 30.30° , 35.50° , 50.84° , 60.38° and 63.24° for $\text{Hf}_7\text{O}_8\text{N}_4$ (JCPDS:50-1173, PDF-2 Database) are assigned to the crystal planes (003), ($\bar{3}22$), (214), ($\bar{3}25$) and (006), respectively. The crystallite size of all samples can be estimated from the strongest diffraction peaks using the Scherrer equation

$$D = 0.89\lambda / (\beta \times \cos \theta) \quad (\text{S1})$$

where D is the crystallite size, λ is the wavelength of the incident X-ray ($\lambda = 0.154056$ nm), θ is the Bragg angle of the diffraction line, and β is FWHM in radians.

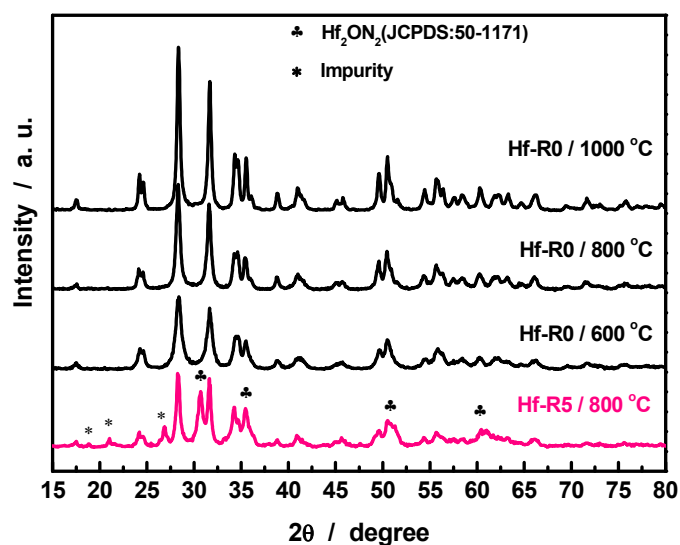


Fig.S1 XRD patterns of the as-synthesized Hf-R0 and Hf-R5 at different heating treatment temperature. The diffraction peaks are well-matched to the corresponding JCPDS card data (JCPDS:65-1142) even though the Hf-R0 samples were heated at different sintering temperature. For Hf-R5 sample, the mixed species were obtained through calcination at 800 °C for 3 h under nitrogen, and the main diffraction peaks can be indexed to HfO_2 (JCPDS:65-1142) and Hf_2ON_2 (JCPDS:50-1171). Note that there are several diffraction peaks due to the impurity are unknown, as indicated by the asterisk.

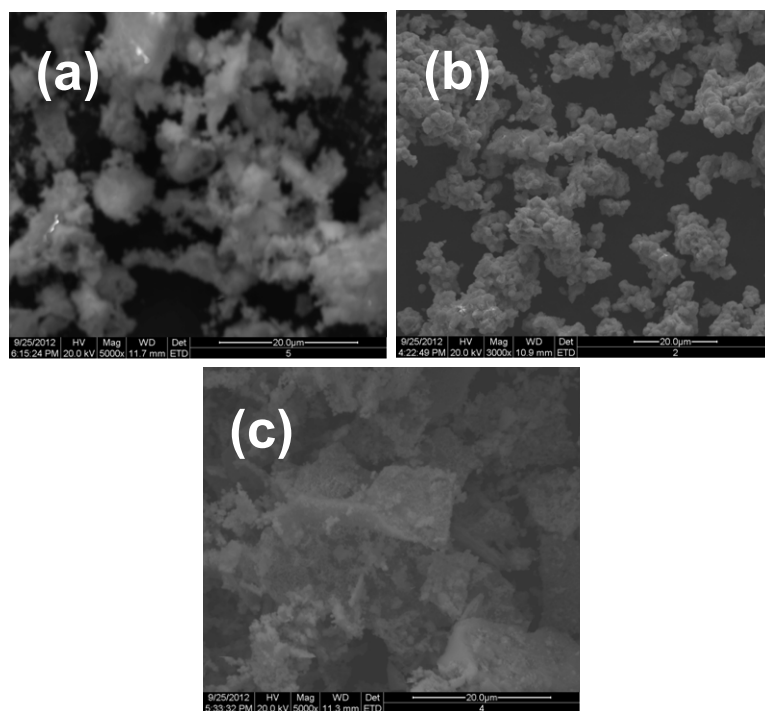


Fig.S2 SEM images for samples at various urea-metal molar ratio. (a) Hf-R0-U, (b) Hf-R5-B and (c) Hf-R12-T (scale bar 20 μm).

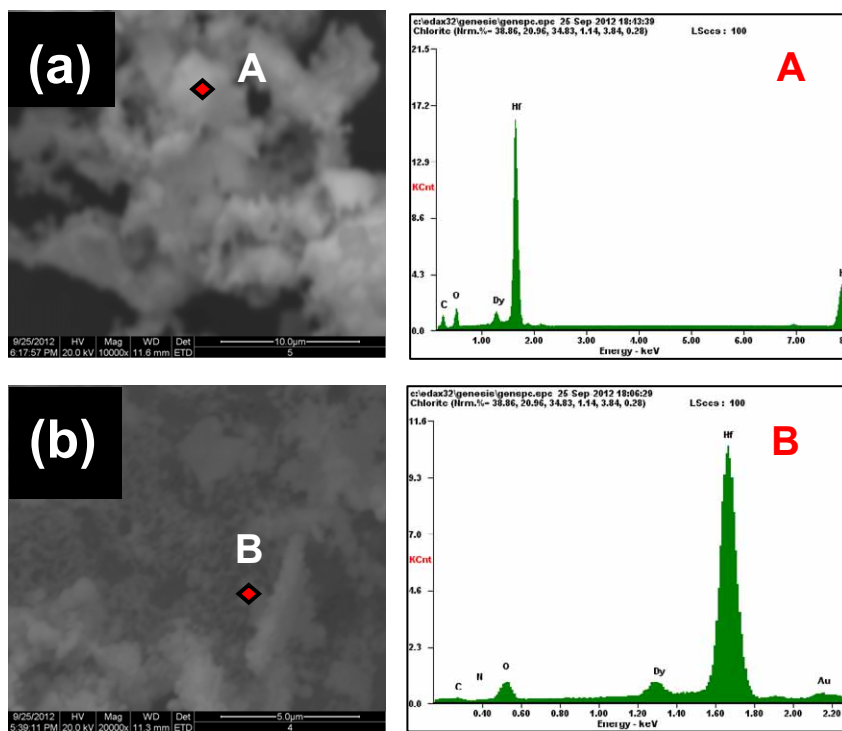


Fig.S3 Local magnified SEM images and the corresponding EDS spectra of the spots A and B for binary composites Hf-R5-B (a) and ternary composites Hf-R12-T (b), respectively.

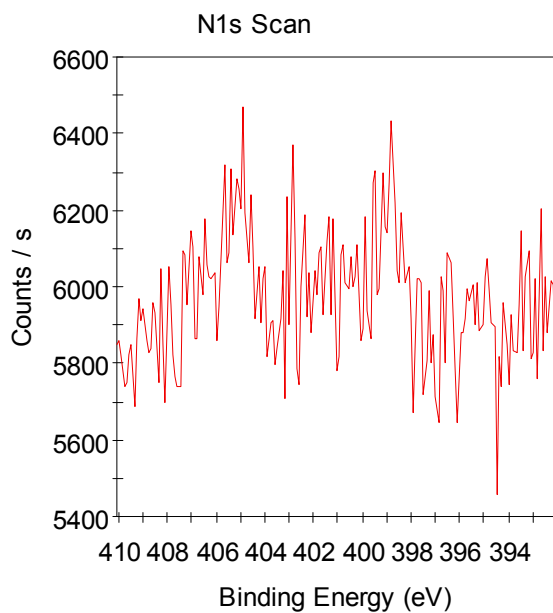


Fig. S4 N 1s XPS spectra of ternary composites Hf-R12-T sample prepared using a urea-metal route.

The equivalent circuit of the symmetrical cells

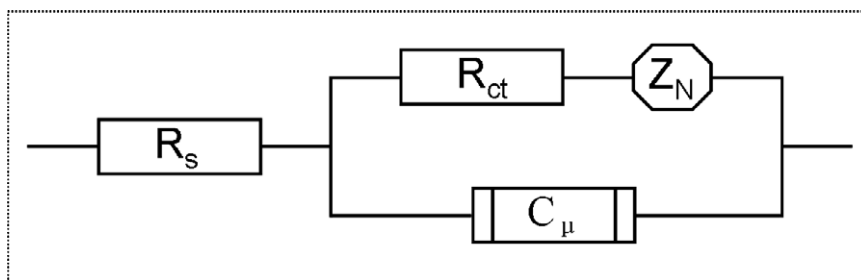


Fig. S5 The equivalent circuit of the symmetrical cells. R_s : series resistance; R_{ct} : charge-transfer resistance; C_{μ} : double layer capacitance; Z_N : Nernst diffusion impedance of the redox couple in the electrolyte.