C–H activation: making diketopyrrolopyrrole derivatives easily accessible

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1. Instruments and materials

All ¹H and ¹³C NMR spectra were obtained in chloroform-*d*, unless otherwise noted, with Bruker DMX-500 or Bruker DMX-300. ¹³C NMR (126 MHz or 76 MHz) spectra were measured with a proton-decoupling pulse program. Chemical shifts for ¹H and ¹³C NMR were referenced to residual signals from CDCl₃ (¹H NMR δ =7.26 ppm and ¹³C NMR δ =77.23 ppm). Matrix-assisted laser desorption/ionizationtime of flight mass spectrometry (MALDI-TOF MS MS) was performed on a Voyager DE STR using 2,5-Dihydroxybenzoicacid or α -Cyano-4-hydroxycinnamicacid as the matrixes. Samples were prepared by diluting the molecules in chloroform with the matrix. Elemental analyses were conducted on a Flash EA 1112 elemental analyzer. UV-visible absorption spectra were taken on a Shimadzu UV-2450 spectrophotometer. Cyclic voltammetry (CV) was done on a CHI 660C electrochemical workstation with Pt disk, Pt plate, and standard 10 calomel electrode (SCE) as working electrode, counter electrode, and reference electrode, respectively, in a 0.1 mol/L tetrabutylammonium hexafluorophosphate (Bu₄NPF₆) CH₂Cl₂ solution.

All starting organic compounds were purchased from Aldrich, Alfa Aesar, Alladin, Energy chemical, or TCI and used without further purification. The solvent toluene and dimethyl acetamide (DMA) are anhydrous.

2. Synthetic route for the DPP 1a-1e and DPP 1-17



Scheme S1. Synthetic route of parent DPP 1a-1c



Scheme S2. Synthetic route of parent DPP 1d and 1e



Scheme S3. Synthetic route of DPP 1



Scheme S4. Synthetic route of DPP 2





Scheme S8. Synthetic route of DPP 6



Scheme S9. Synthetic route of DPP 7







Scheme S11. Synthetic route of DPP 9



Scheme S12. Synthetic route of DPP 10



Scheme S13. Synthetic route of DPP 11



Scheme S14. Synthetic route of DPP 12



Scheme S15. Synthetic route of DPP 13



Scheme S16. Synthetic route of DPP 14



Scheme S17. Synthetic route of DPP 15



Scheme S18. Synthetic route of DPP 16



Scheme S19. Synthetic route of DPP 17





Scheme S20. The comparison between our method and the reported methods

4. Fluorescence spectra of DPP 1-17



Fig. S1. Fluorescence spectra of DPP 1-17 in CHCl₃ (excitation at 467 nm)

5. ¹H & ¹³C NMR and MALDI-TOF MS spectra



Fig. S2. ¹H NMR for DPP 1a



Fig. S3. ¹H NMR for DPP 1b



Fig. S4. ¹H NMR for mono-bromo-DPP



Fig. S5. ¹H NMR for DPP 1c



Fig. S6. ¹H NMR for DPP 1d



Fig. S7. ¹H NMR for DPP 1e



Fig. S8. ¹H & ¹³C NMR for DPP 1

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Fig. S9. ¹H &¹³C NMR for DPP 2

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Fig. S10. 1 H & 13 C NMR and MALDI-TOF MS for DPP 3



Fig. S11. ¹H &¹³C NMR for DPP 4



Fig. S12. 1 H & 13 C NMR for DPP 5



Fig. S13. ¹H &¹³C NMR and MALDI-TOF MS for DPP 6



Fig. S14. ¹H &¹³C NMR and MALDI-TOF MS for **DPP 7**



Fig. S15. ¹H &¹³C NMR for DPP 8



Fig. S16. ¹H &¹³C NMR and MALDI-TOF MS for DPP 9



Fig. S17. ¹H & ¹³C NMR for **DPP 10**

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Fig. S19. ¹H &¹³C NMR and MALDI-TOF MS for DPP 12



Fig. S20. ¹H &¹³C NMR for **DPP 13**



Fig. S21. ¹H &¹³C NMR for **DPP 14**



Fig. S22. ¹H &¹³C NMR for **DPP 15**

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Fig. S23. ¹H &¹³C NMR and MALDI-TOF MS for DPP 16



Fig. S24. ¹H &¹³C NMR and MALDI-TOF MS for DPP 17